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In Situ X-Ray Studies of Segregation and Activity in SOFC Cathode Materials

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Principal Results

A-site segregation in SOFC cathode materials including LSM, LSC and LSCF was measured as a function of pO_2 , temperature and electrochemical state.

- Strontium surface segregation occurs in (001)-oriented $La_{0.7}Sr_{0.3}MnO_3$ thin films over a wide range of temperatures (25–900°C) and oxygen partial pressures ($pO_2=0.15$ –150 Torr).
 - The strontium surface concentration increases with decreasing pO_2 .
 - A cathodic potential reduces strontium segregation in (110)-oriented LSM films.
- Strong segregation is also found in (001)-oriented LSC and LSCF films
 - little dependence on pO_2 .
- The different pO_2 dependence is consistent with much smaller oxygen vacancy gradients in materials with high vacancy mobilities and concentrations.

Overview

■ High Level View

- Motivation and Approach
- Overview of In Situ X-Ray Techniques

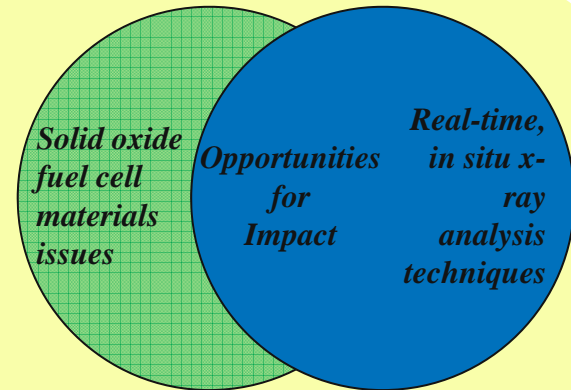
■ Film Structure, Composition versus Environment

■ Chemical and Electronic Structure

■ Conclusions

Synchrotron Studies - Goals and Objectives

Develop molecular-level models of SOFC cathode materials to stimulate rational design and development of high-performance cathode materials.



■ In Situ Controlled Atmosphere Studies

- Equilibrium structure in controlled atmosphere (e.g. variable pO_2).
- Identify driving forces for structural and chemical rearrangement

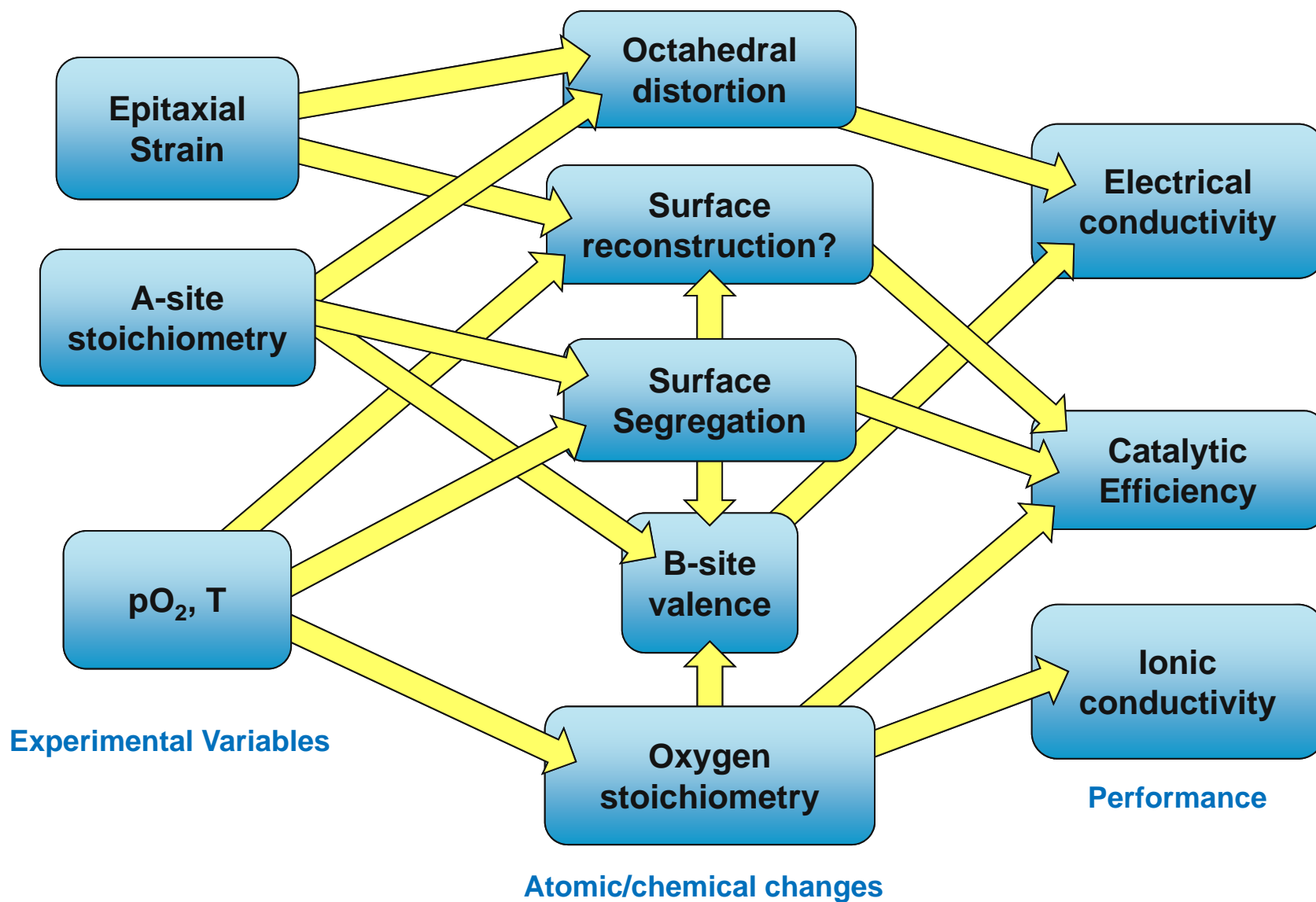
■ In Situ Electrochemical Studies (Hoydoo You - poster yesterday)

- Determine dynamic changes of cathode occurring in SOFC half-cell
- Correlate with equilibrium structures and ex situ measurements

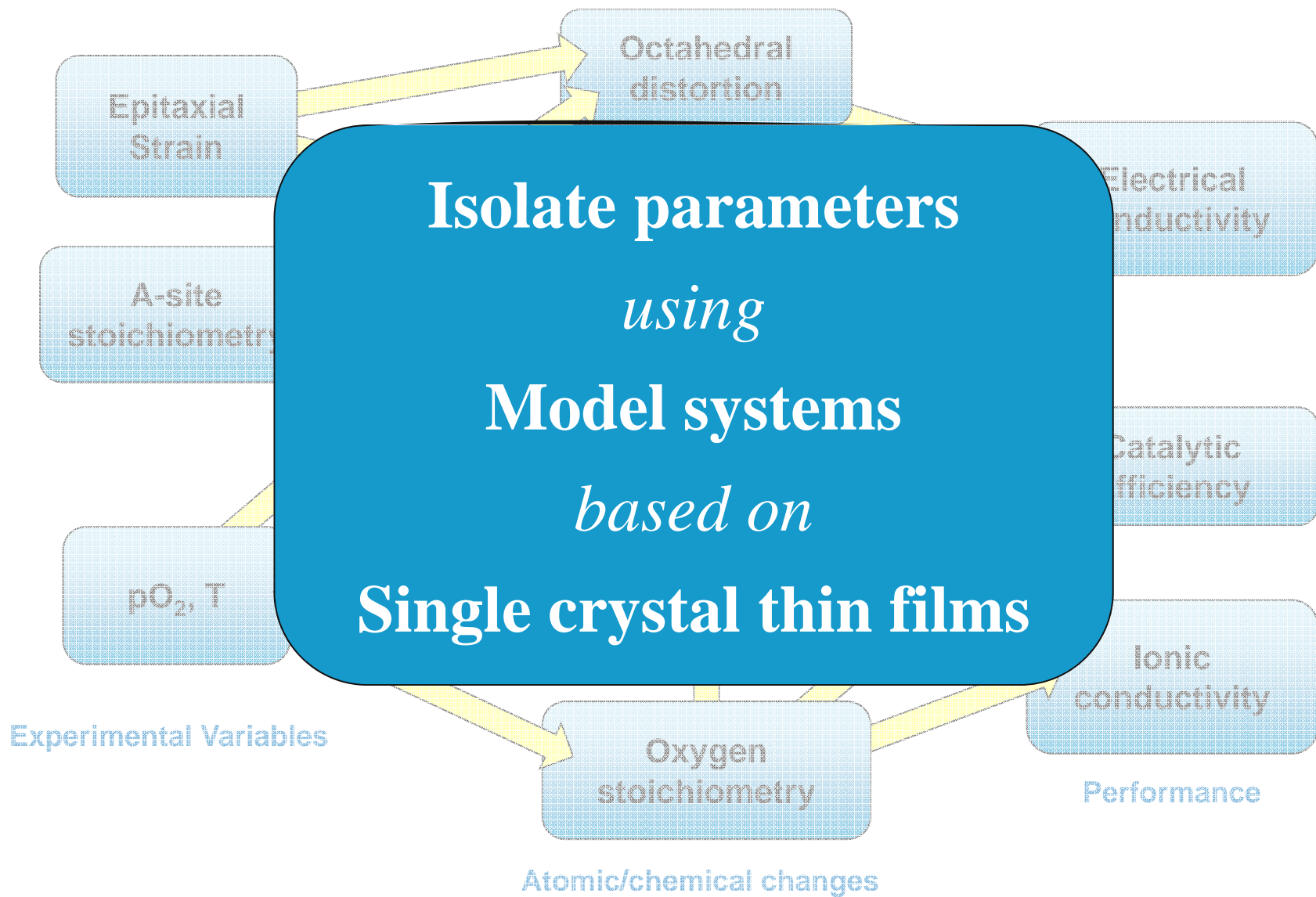
■ In Situ Studies of Operating Fuel Cells

- *Focus on cathode side of fuel cell*
- *Examine atomic structure and chemical state of individual constituents*
- *Correlate with ex situ measurements and performance data*

LSM, LSC, LSCF: Complicated Interactions



LSM, LSC, LSCF: Complicated Interactions



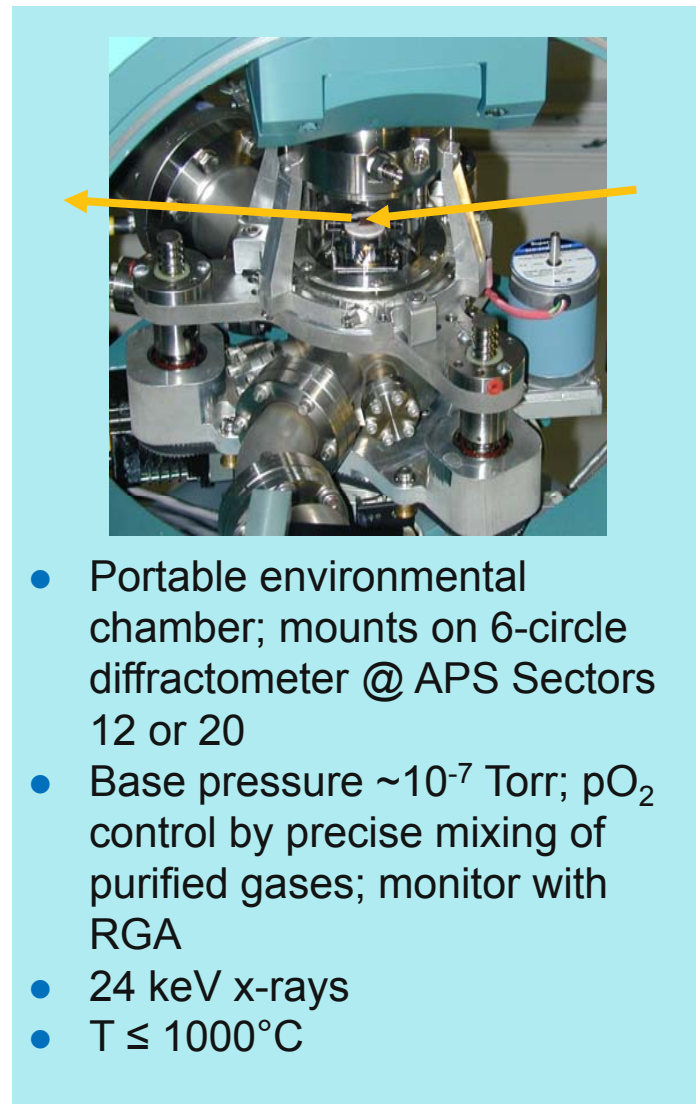
Approach

■ LSM, LSC and LSCF epitaxial films grown by Pulsed Laser Deposition (PLD) at Carnegie Mellon University

- Growth: 750°C, 50 mTorr O₂, La_{0.7}Sr_{0.3}MnO₃, La_{0.7}Sr_{0.3}CoO₃ and La_{0.6}Sr_{0.4}Co_{0.2}Fe_{0.8}O₃
- Cooled in 300 Torr pO₂
- (001) SrTiO₃ (STO), (110) NdGaO₃ (NGO) & DyScO₃ (DSO) substrates provide different epitaxial strain conditions
- Yttria-Stabilized Zirconia (YSZ) (111) and (001) single crystal substrates for electrochemical measurements

■ In situ synchrotron x-ray studies

- Probes atomic-scale processes during realistic SOFC conditions
- Studies performed at the Advanced Photon Source
- Total reflection x-ray fluorescence (TXRF) to determine surface composition
- Grazing incidence & high angle diffraction to determine surface and film structure



Synchrotrons Have Revolutionized X-Ray Analysis

- **The Advanced Photon Source is nine orders of magnitude brighter than laboratory sources.**
- **Brightness has enabled:**
 - Scattering from single layers of atoms
 - Nanometer resolution imaging
 - Realtime, *in situ* measurements from all types of surfaces and ultra-thin films
 - Structure determination of buried interfaces
- **Great potential for advancing understanding of complex industrial processes.**

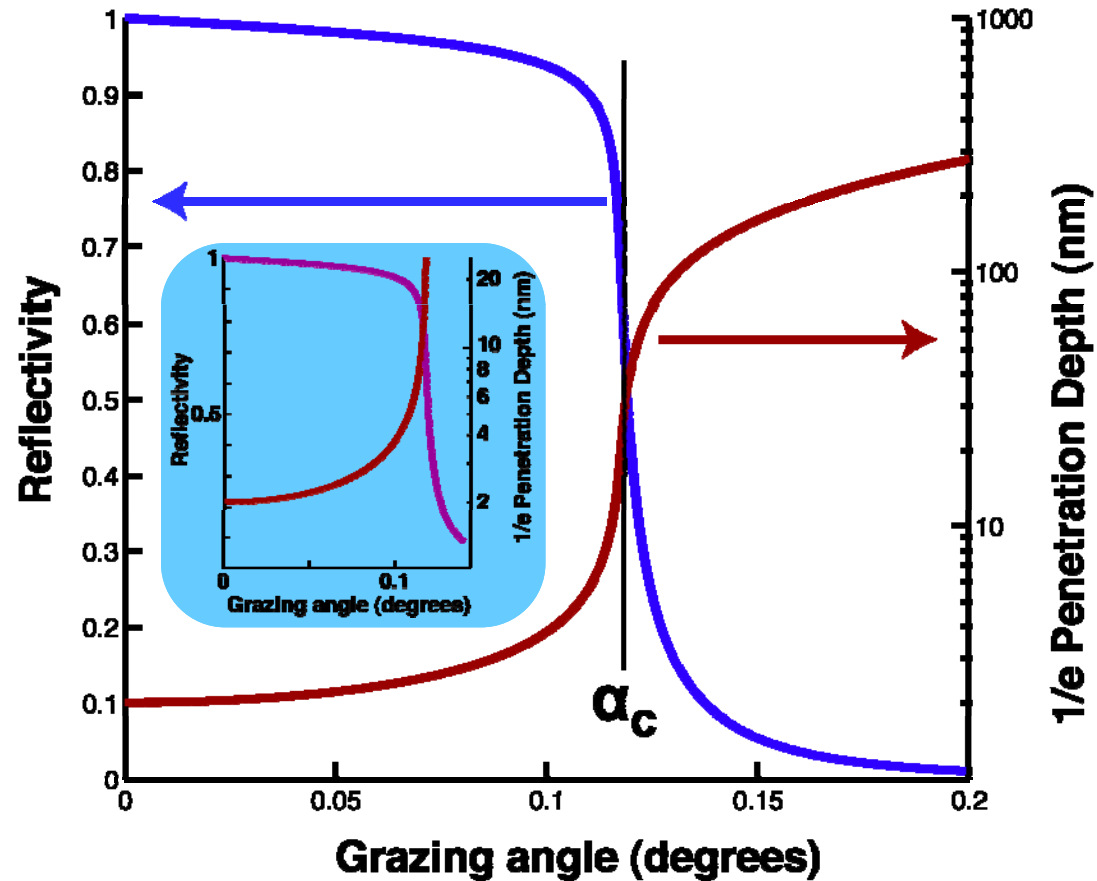
Typical X-Ray Measurements

Example from $\text{La}_{0.6}\text{Sr}_{0.4}\text{Co}_{0.2}\text{Fe}_{0.8}\text{O}_3$ Studies

- **Composition fluctuations: Strontium surface segregation?**
 - Total reflection x-ray fluorescence (TXRF)
- **Chemistry induced ordering: Surface reconstructions?**
 - Grazing incidence x-ray diffraction
- **Are there structural changes with pO_2 ?**
 - Diffraction, reflectivity
- **Chemical changes with pO_2 ?**
 - Resonant scattering techniques
 - X-ray absorption spectroscopy (XANES)

Total Reflection - Making X-rays Surface Sensitive

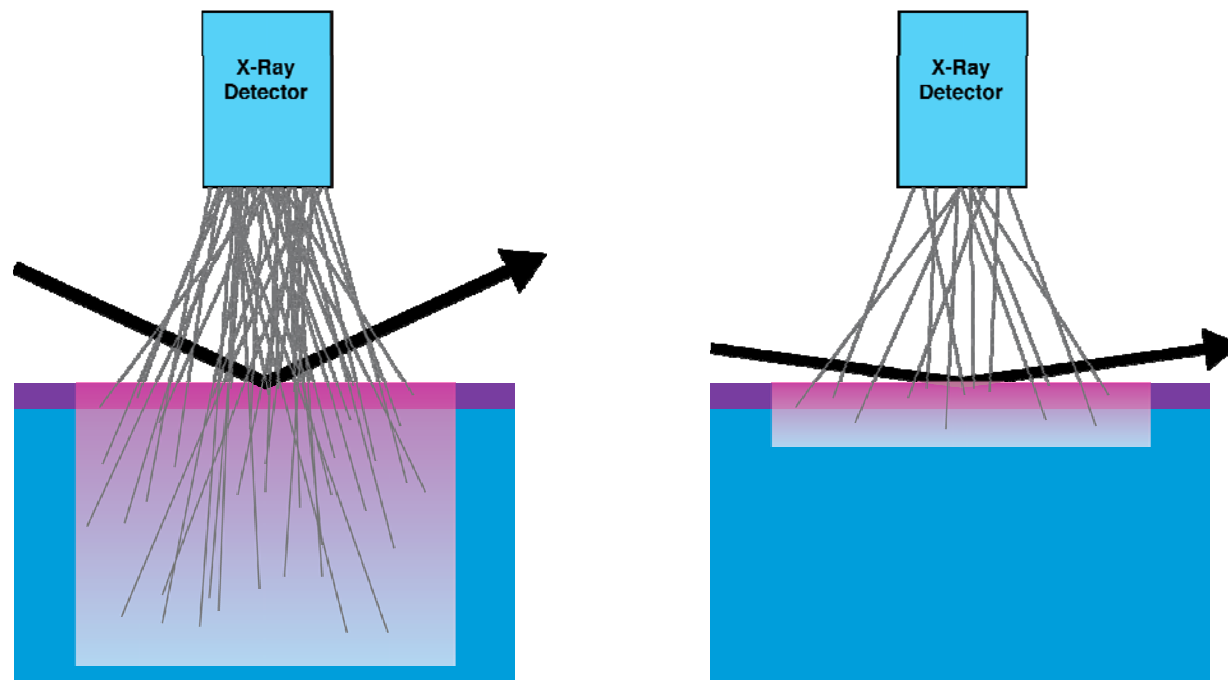
$\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ using 24 keV Photons



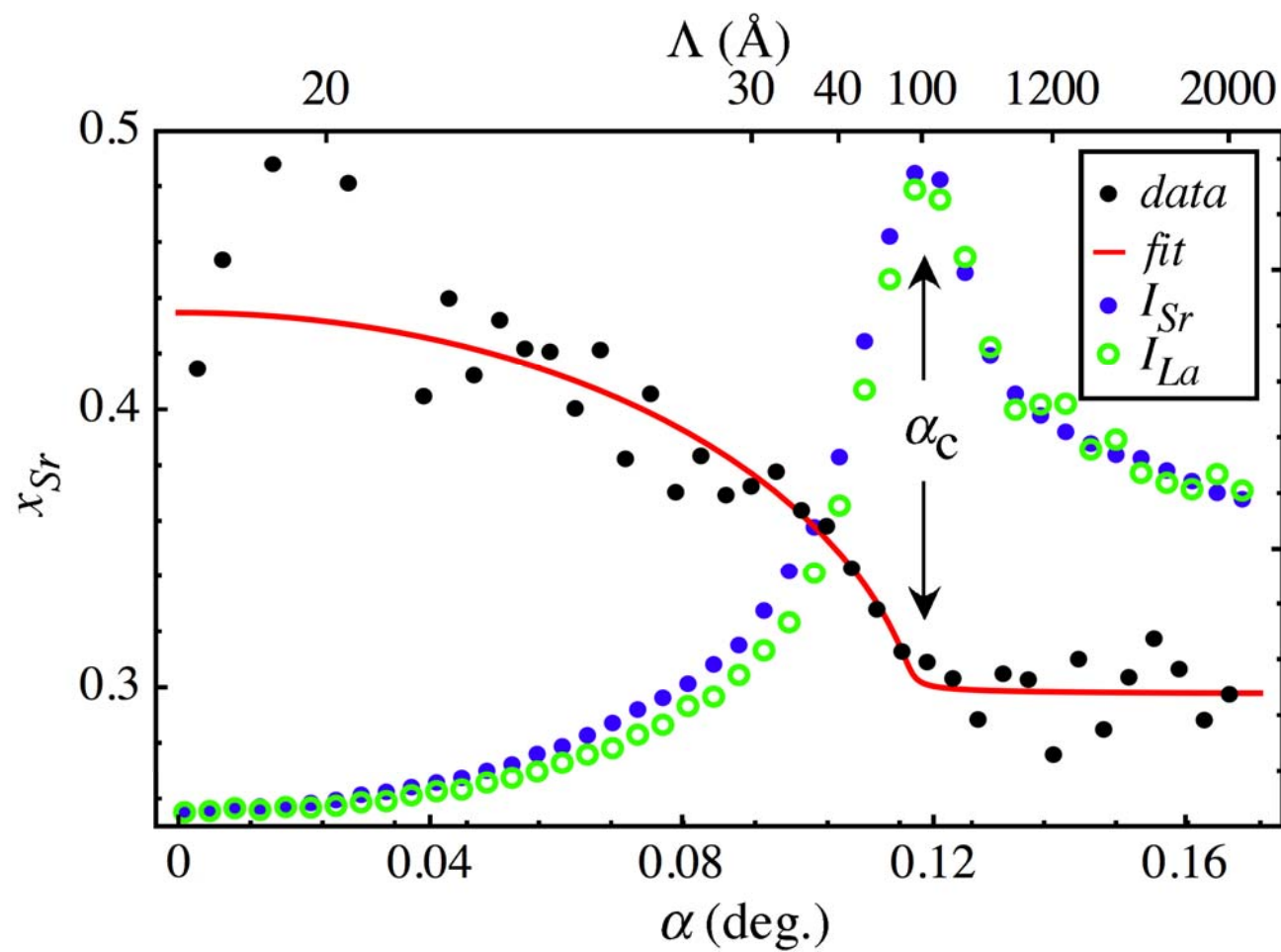
Total Reflection X-Ray Fluorescence (TXRF)

TXRF is a standard technique for analyzing impurities on semiconductor substrates since each element has a standard spectra.

We've extended it to quantitative studies of nanometer composition gradients at surfaces and buried interfaces.



Typical Analysis of TXRF

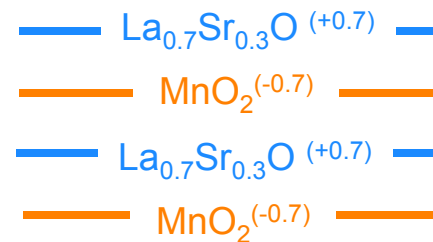
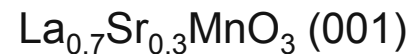
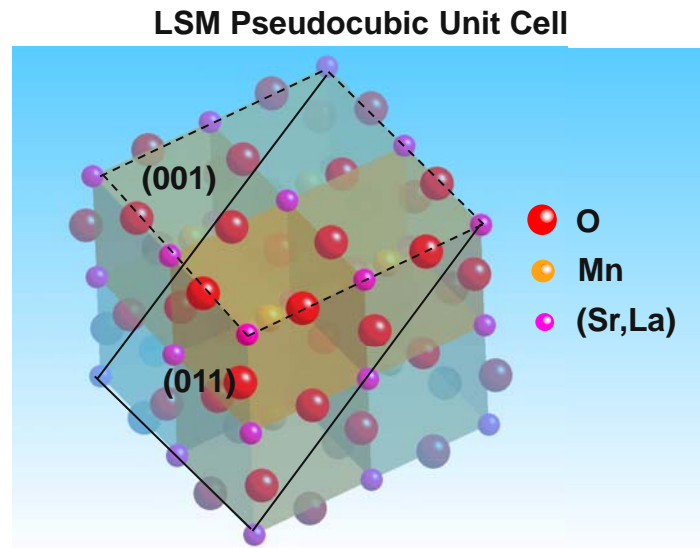


Overview

- High Level View
- **Film Structure, Composition versus Environment**
 - TXRF Measurements of Surface Segregation
 - Surface Structure
- Chemical and Electronic Structure
- Conclusions and Future Directions

Why look for segregation?

- Structure of surfaces are crucial for determining catalytic performance.
- $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$ surfaces tend to be polar.

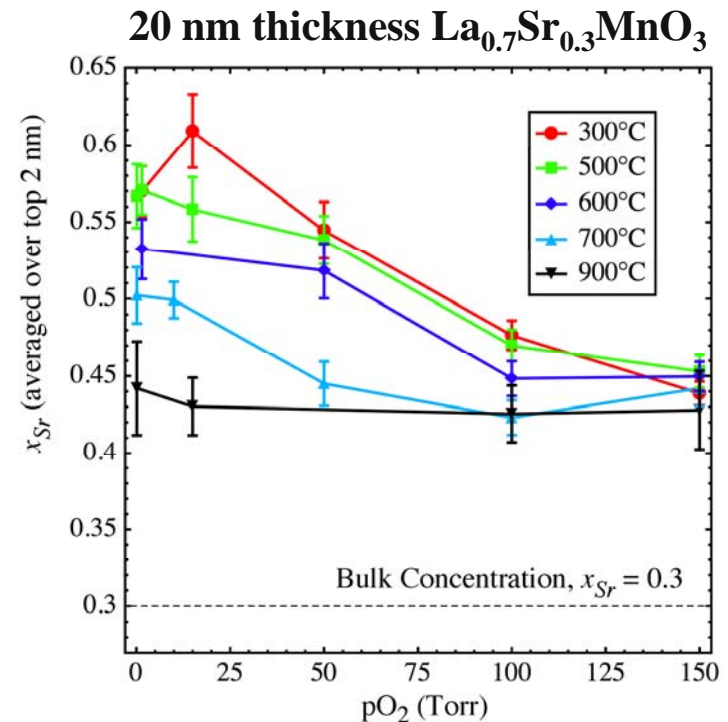
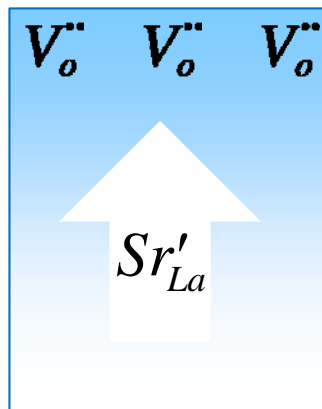


Polar surfaces are not stable!

LSM on DSO: Where we were last year

■ Applied Physics Letters 93, 151904 (2008):

- pO_2 -dependence in Sr surface segregation
- Possibly driven by surface oxygen vacancies
- Possible implications for growth, SOFC performance

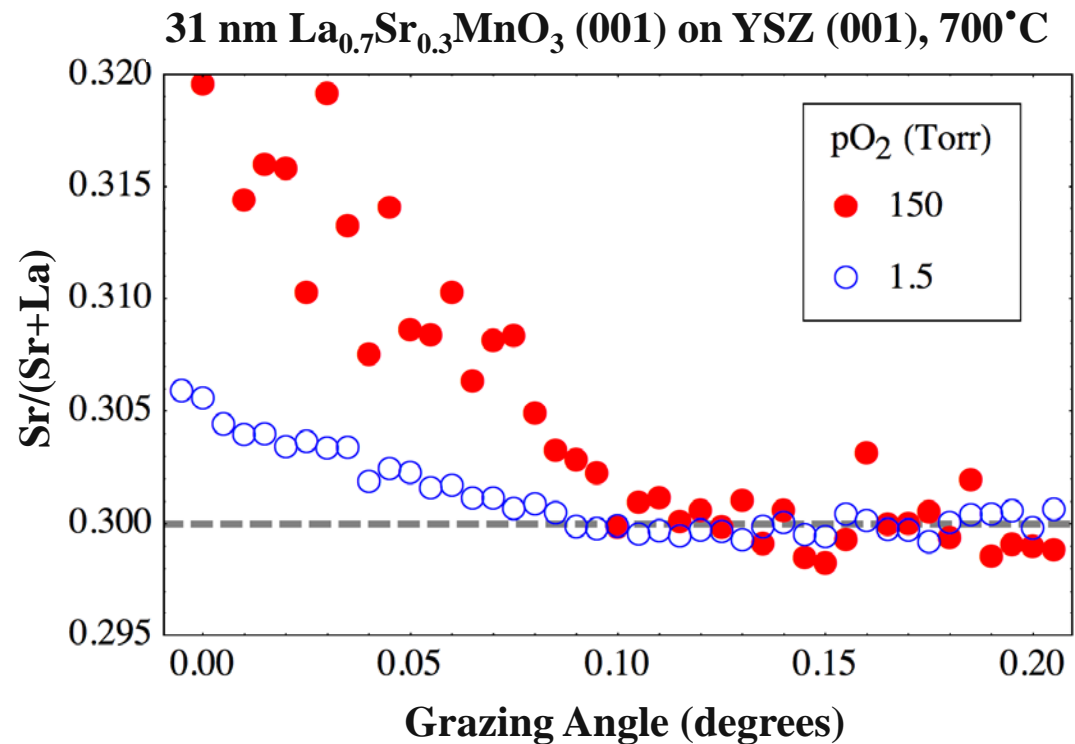


Change in Sr concentration from bulk

	Operating T (700-1000 C)	Low T (300 C)
Low pO_2 (mTorr)	+35%	+50%
Operating pO_2 (atmospheric)	+21%	+25%

$\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ on YSZ

- Reduced segregation compared with LSM/DSO
- Grain boundary segregation may limit surface concentration
- pO_2 -dependence is opposite



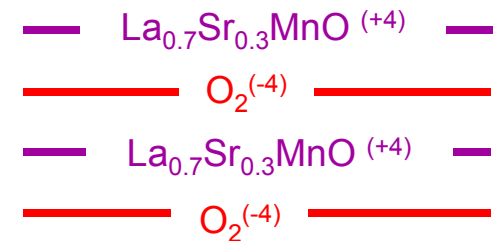
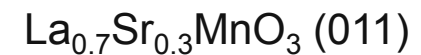
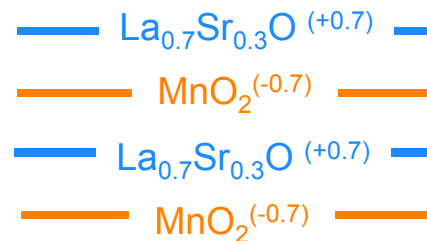
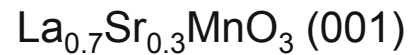
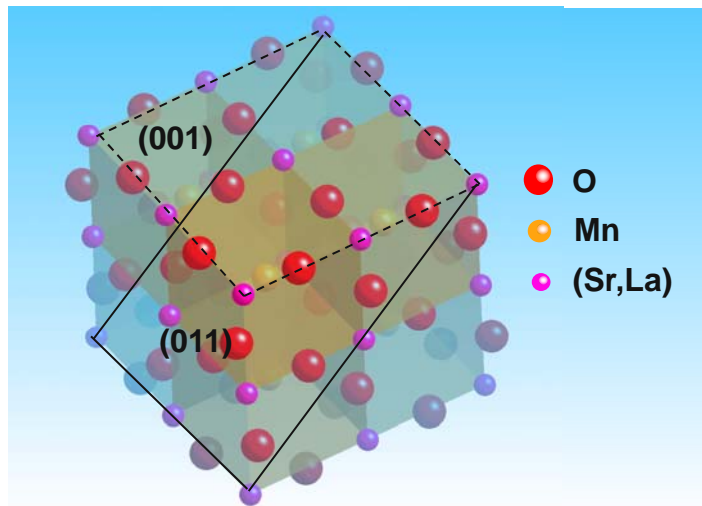
Do more oxygen vacancies in YSZ increase Sr segregation at the YSZ interface?

Orientation Dependence of Segregation

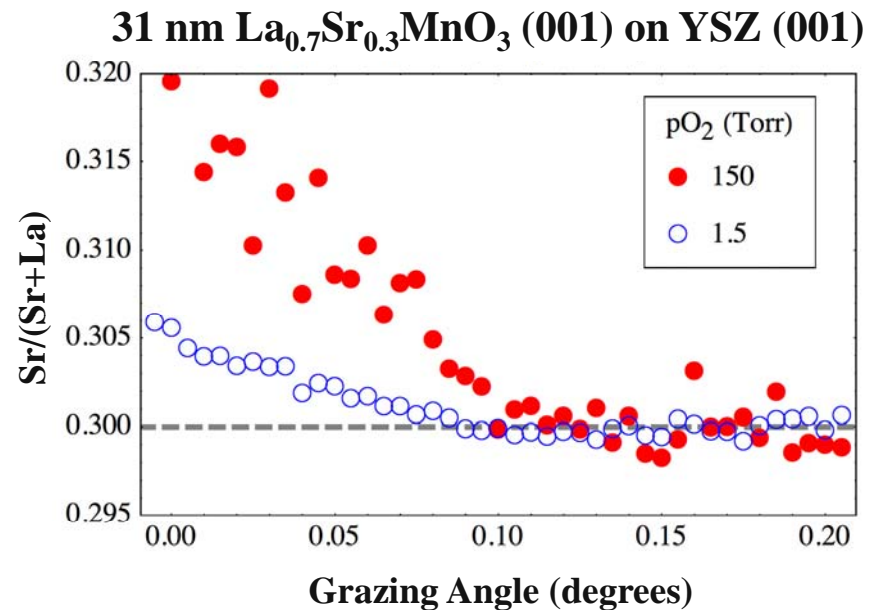
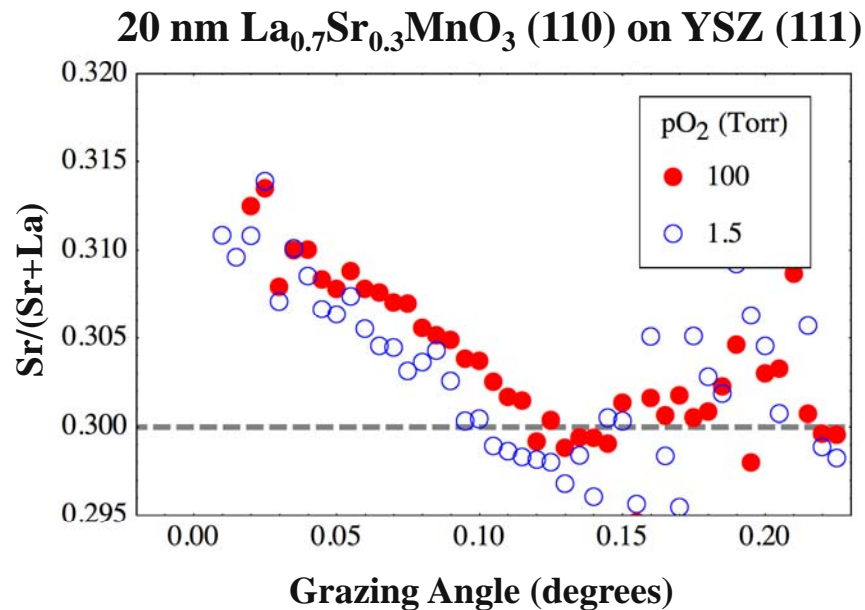
Substrate orientation changes

- film orientation
 - LSM(011) and LSC(011) grow on YSZ(111)
- the degree of epitaxy
- magnitude of surface polarity.

LSM Pseudocubic Unit Cell



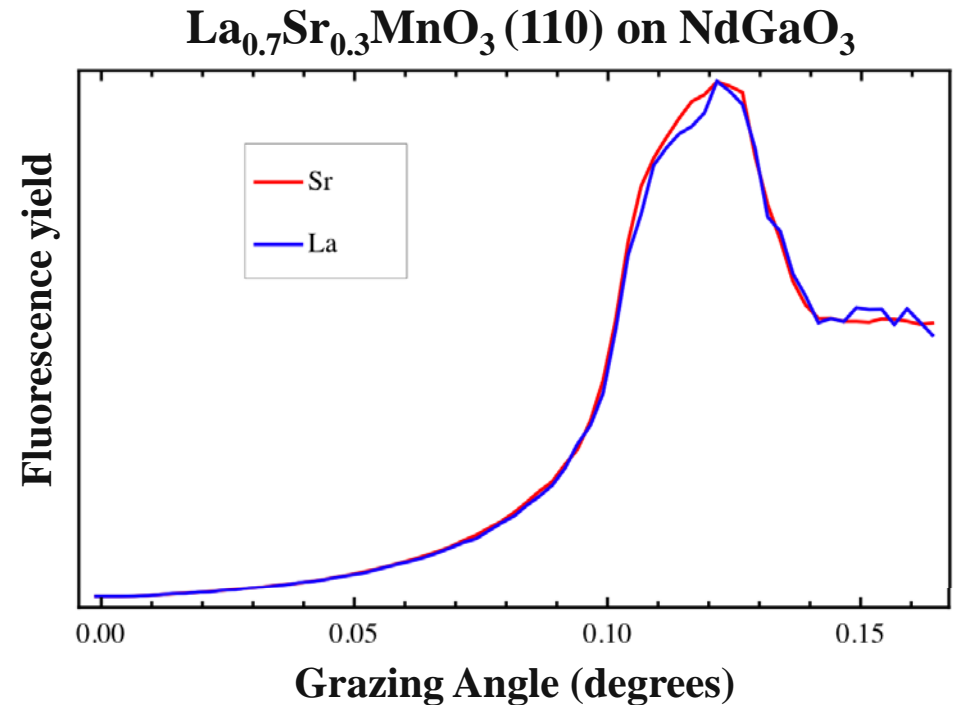
$\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ on YSZ: Orientation Dependence



Segregation is observed for both orientations but is not significantly stronger for (110) surface.

Segregation: (110)-oriented $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$

- We found Sr segregation for (100) $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ on NdGaO_3
 - at 700°C and 300°C
 - $p\text{O}_2 = 0.15 - 150$ Torr.
- No evidence for Sr segregation at similar conditions for (110) $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ on NdGaO_3

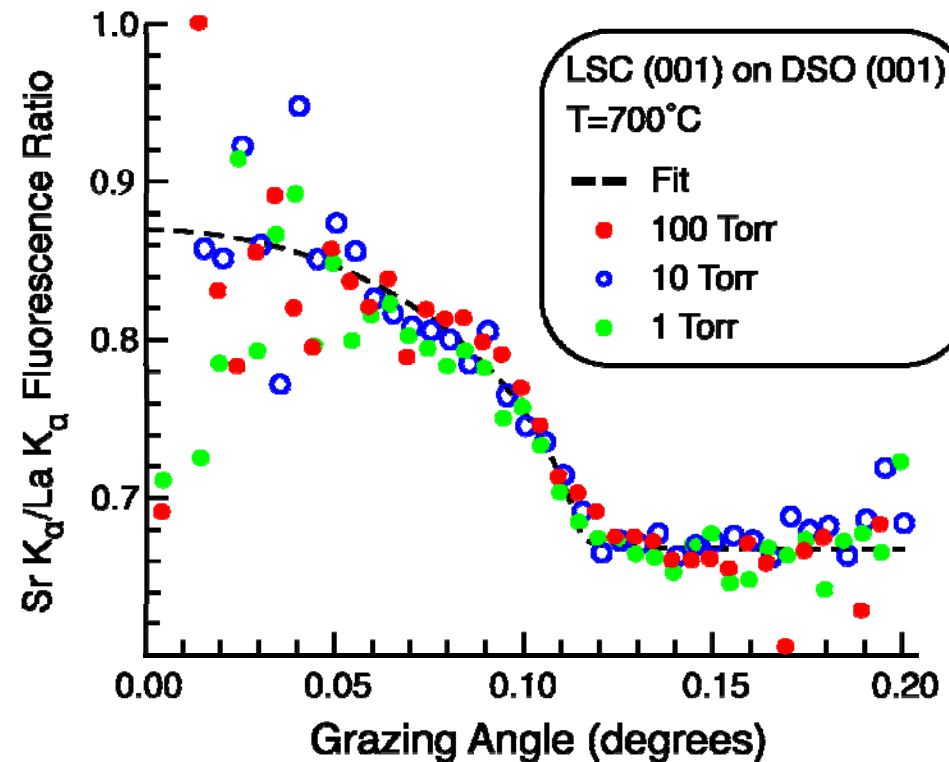


Do other mechanisms (perhaps reconstructions) start compensating the polarity?

$\text{La}_{0.6}\text{Sr}_{0.4}\text{CoO}_3$ (001) on DyScO_3

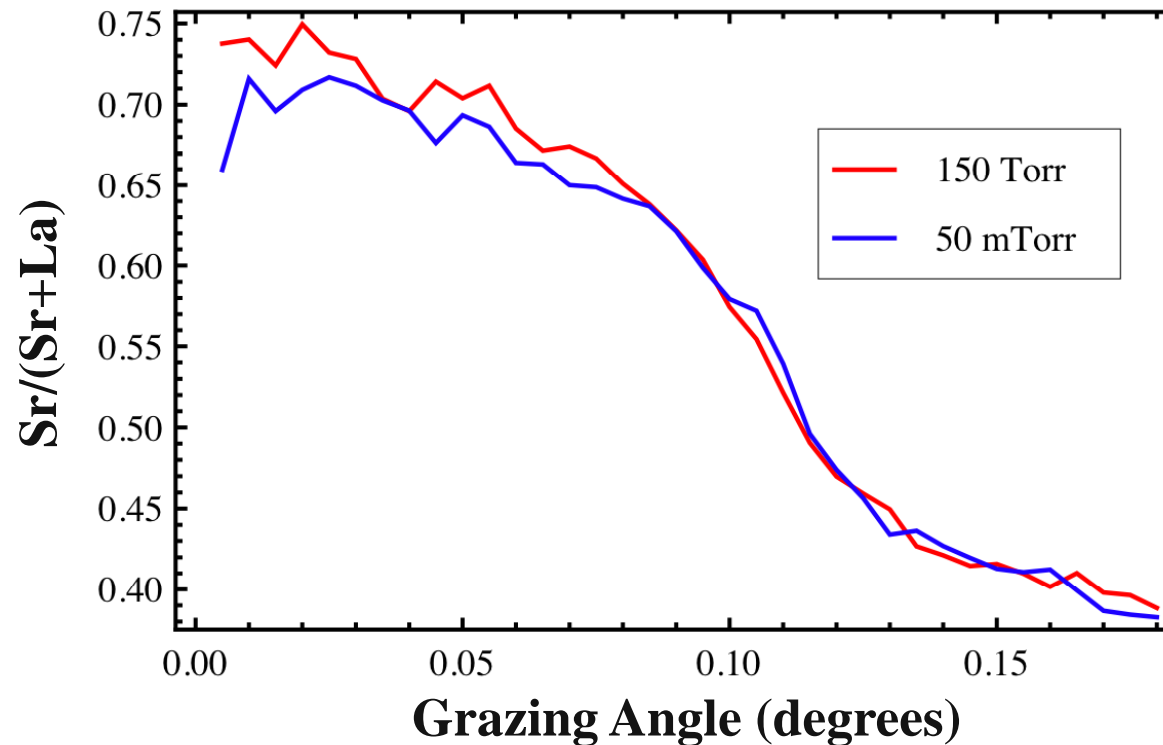
Strontium segregation

- Similar to LSM/DSO in magnitude
- No pO_2 dependence at this relatively high temperature.



Strontium concentration approaches 50% at the near surface. Will this high concentration drive oxygen vacancy ordering?

LSCF (001) Surface Segregation



- Segregation larger than LSC/DSO measured at 700°C
- No apparent pO_2 -dependence

Summary so far

- **Strontium segregation is usually found**
 - but there is evidence of other mechanisms suppressing segregation in a few cases
- **The pO_2 dependence is correlated with the ability of the film to support an oxygen vacancy gradient**

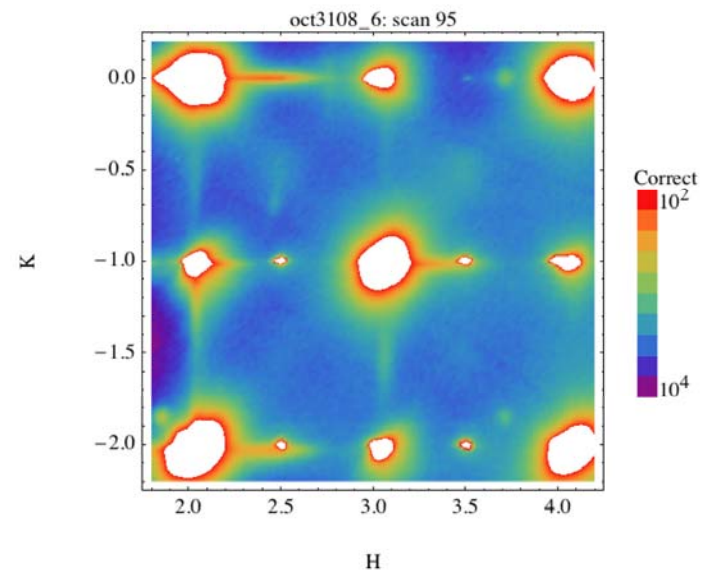
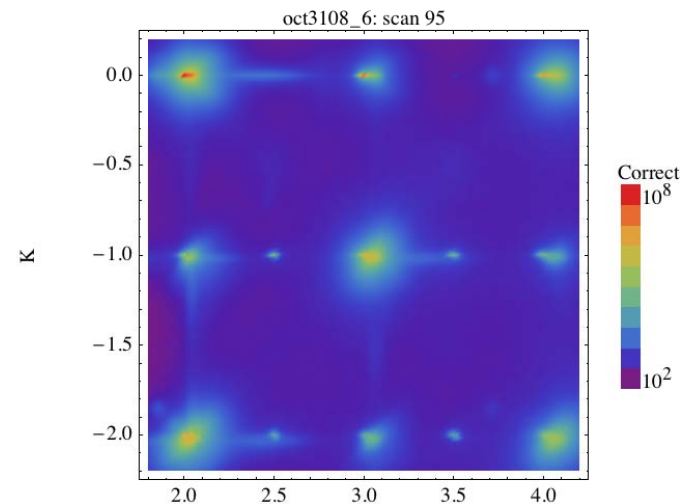
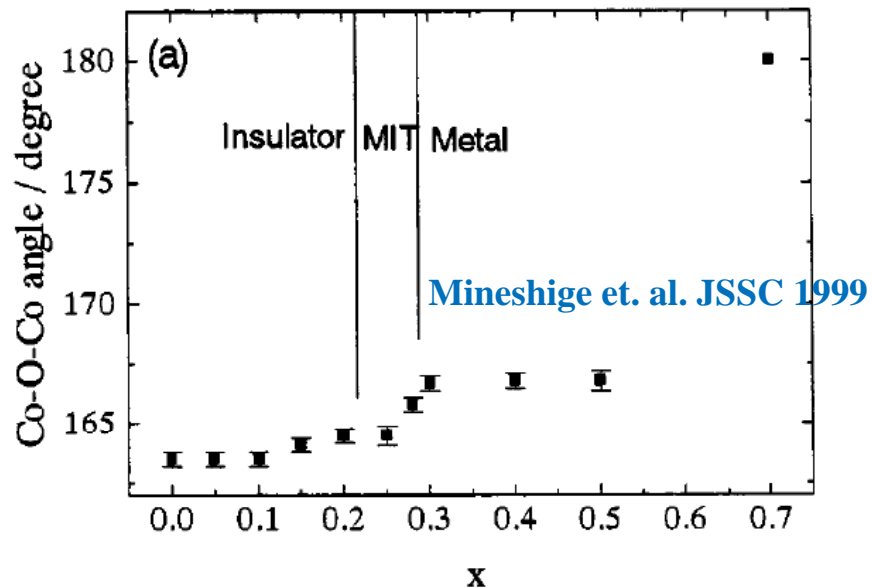
Are there other changes in surface structure that play an important role?

Overview

- High Level View
- **Film Structure, Composition versus Environment**
 - TXRF Measurements of Surface Segregation
 - Surface Structure
- Chemical and Electronic Structure
- Next Steps and Directions

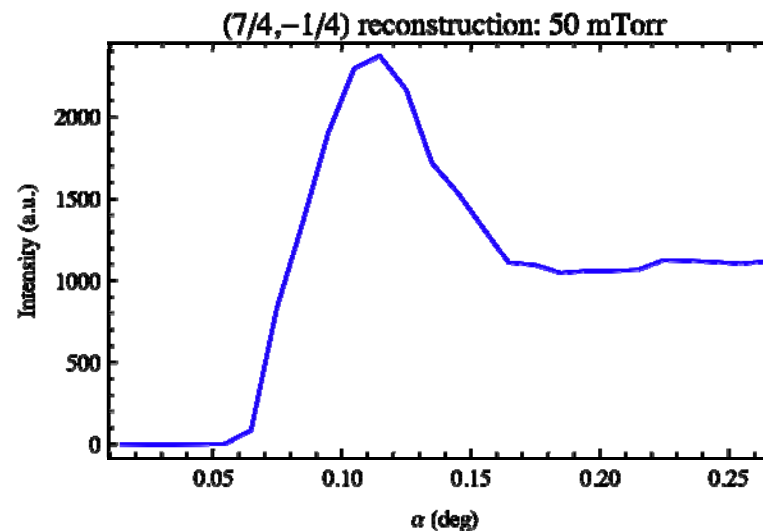
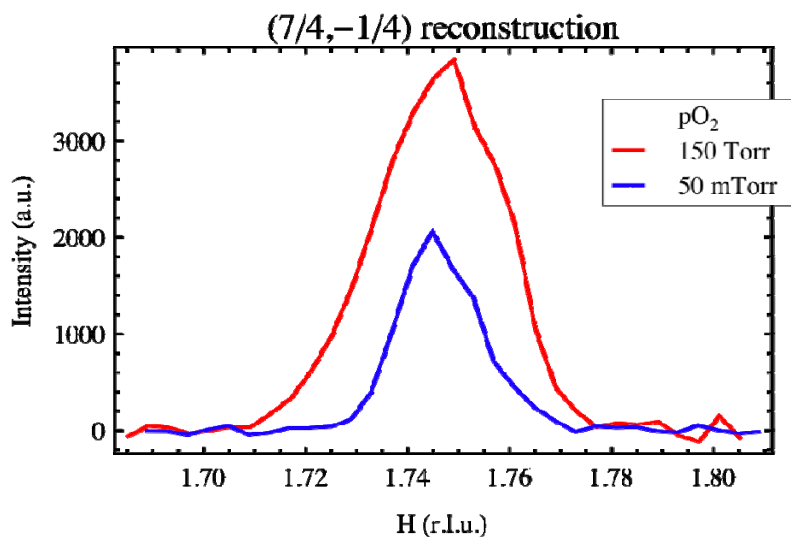
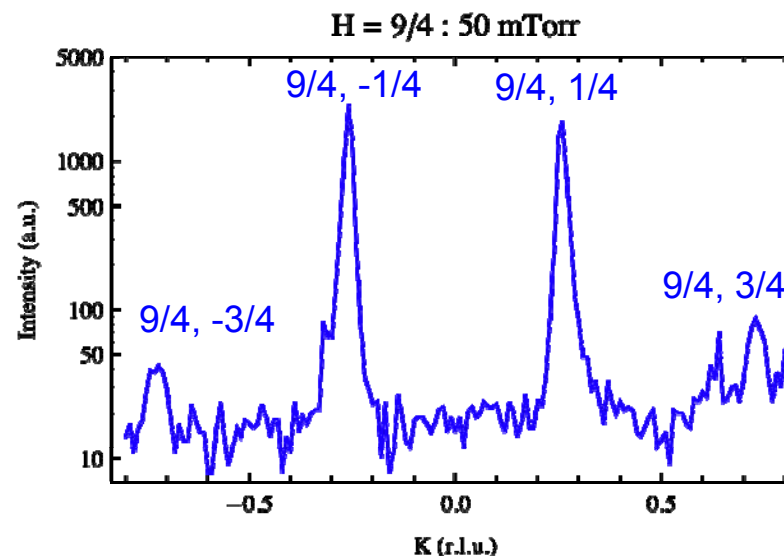
$La_{0.6}Sr_{0.4}CoO_3$ on $DyScO_3$ (110)

- Search for reconstructions
- Strength of $\frac{1}{2}$ order peaks related to degree of CoO_6 octahedral tilting.
(related to O 2p/Co 3d overlap).
 - Octahedral tilting in LSCO corresponds to electronic conductivity



LSCF surface structure: reconstructions

- Superlattice peaks present at quarter order positions (but not half)
- Intensities of superlattice peaks are dependent on $p\text{O}_2$.
- Preliminary results, more data needed to determine structural parameters.



Overview

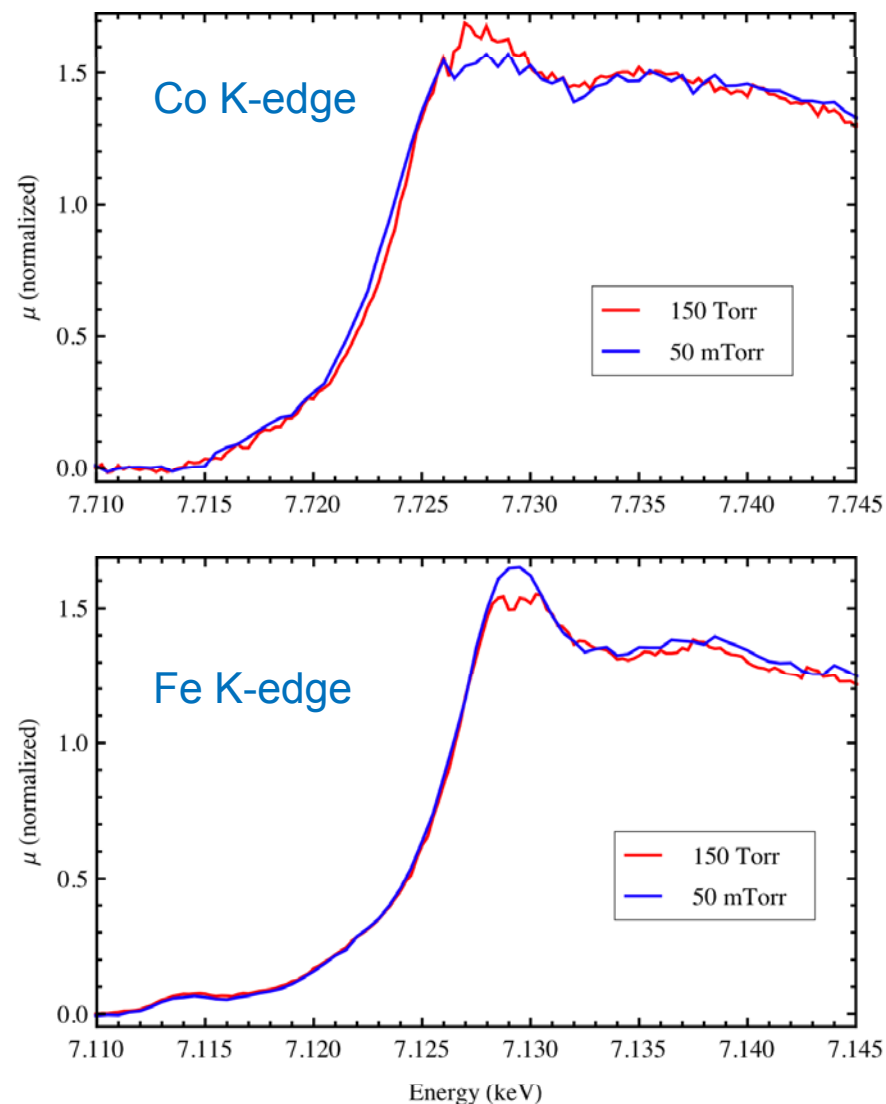
- High Level View
- Film Structure, Composition versus Environment
- Chemical and Electronic Structure
 - X-Ray Absorption Near Edge Structure (XANES)
 - ~~Total Reflection Inelastic X-Ray Scattering (TRIXS)~~
 - ~~Resonant Anomalous X-Ray Reflectivity (RAXR)~~
- Summary and Conclusions

Measuring the local defect chemistry of LSCF

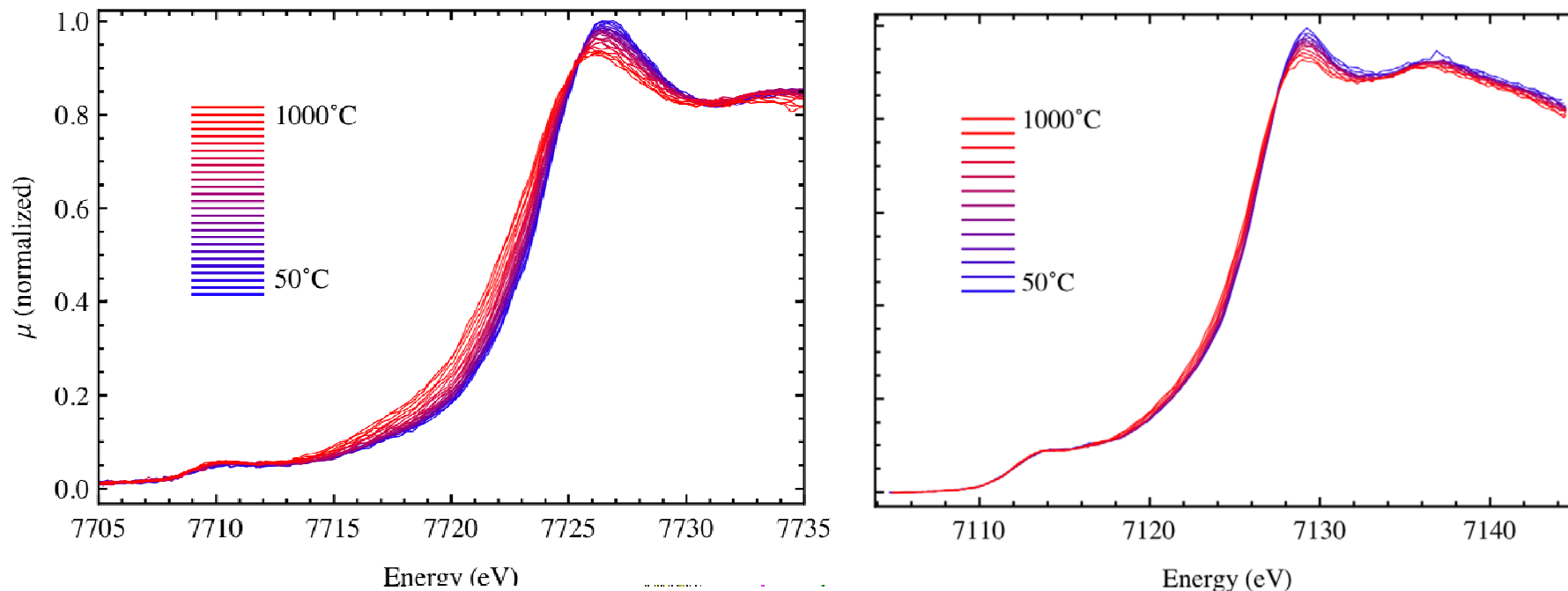
■ Goal: Correlate B-site oxidation state with vacancy concentrations

- $T = 25\text{-}1000^\circ\text{C}$. Measure bulk $\text{La}_{0.6}\text{Sr}_{0.4}\text{Co}_{0.2}\text{Fe}_{0.8}\text{O}_3$ in air and flowing argon
- Measure changes in Fe and Co oxidation state with XANES
- Measure local oxygen vacancy concentration by changes in oxygen coordination from Co and Fe EXAFS (number of nearest neighbors)

■ Question: Do oxygen vacancies prefer to sit adjacent to cobalt sites?

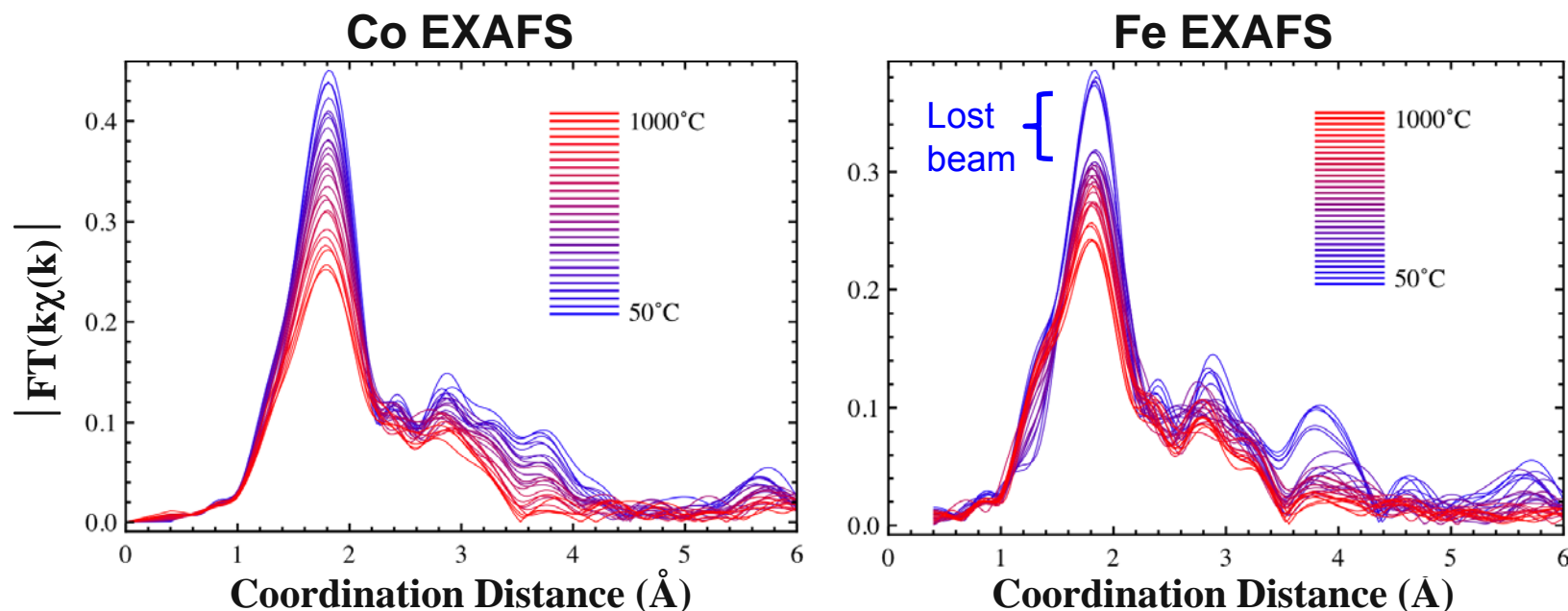


B-site reduction: Co vs. Fe XANES



- Shown: 9 hour ramp from 50 to 1000°C in flowing argon.
- Negative edge shift indicates increasing reduction during heating.
- Much larger edge shift in Co edge than Fe K-edge
 - Co preferentially reduces

B-site coordination



- **Decrease in first shell amplitude with temperature**
 - Due thermal disorder and, possibly, drop in coordination
 - No significant change in (Co,Fe)-O bond length
 - Thermal Debye Waller factors can be fit using T-dependence
- **Assuming similar thermal Debye-Waller factors, cobalt sees a larger drop in coordination**
 - consistent with higher local oxygen vacancy concentration at Co site

Summary of Principal Results

- **Strontium surface segregation occurs in (001)-oriented $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ thin films over a wide range of temperatures (25–900°C) and oxygen partial pressures ($p\text{O}_2=0.15\text{--}150$ Torr).**
 - The strontium surface concentration increases with decreasing $p\text{O}_2$.
 - A cathodic potential reduces strontium segregation in (110)-oriented LSM films.
- **Strong segregation is also found in (001)-oriented LSC and LSCF films**
 - little dependence on $p\text{O}_2$.
- **The different $p\text{O}_2$ dependence is consistent with much smaller oxygen vacancy gradients in materials with high vacancy mobilities and concentrations.**
- **Oxygen vacancies in LSCF appear to be preferentially associated with cobalt sites.**
 - defect models should account for this possibility

Acknowledgements

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The End