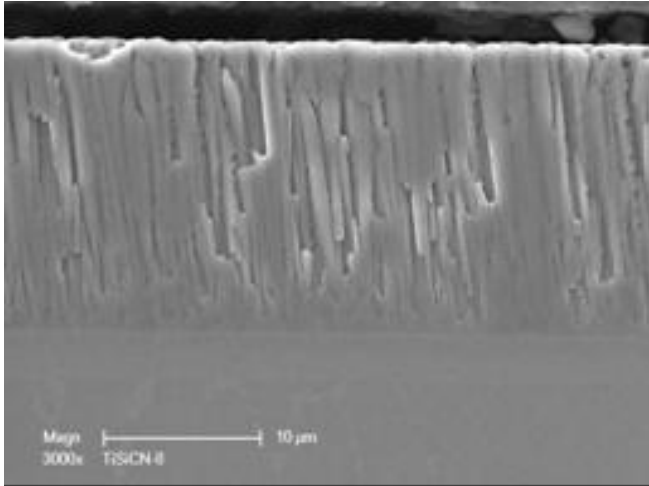




ELECTRIC POWER  
RESEARCH INSTITUTE



## Computational Modeling and Assessment of Nanocoatings for USC Boilers



Presented by:

D. Gandy (EPRI), [davgandy@epri.com](mailto:davgandy@epri.com)

S. Cheruvu (SWRI), [Scheruvu@swri.org](mailto:Scheruvu@swri.org)

J. Shingledecker (EPRI), [Jshingledecker@epri.com](mailto:Jshingledecker@epri.com)

**DOE Sponsored Program: DE-FC26-07NT43096**

# Background

- Fireside corrosion of boiler waterwalls continues to be the #1 issue resulting in forced outages and boiler unavailability for “conventional” coal-fired fossil boiler power plants.



Equivalent Availability Loss from Boiler Tube Failures in US is 2.5-3.0%

# Background

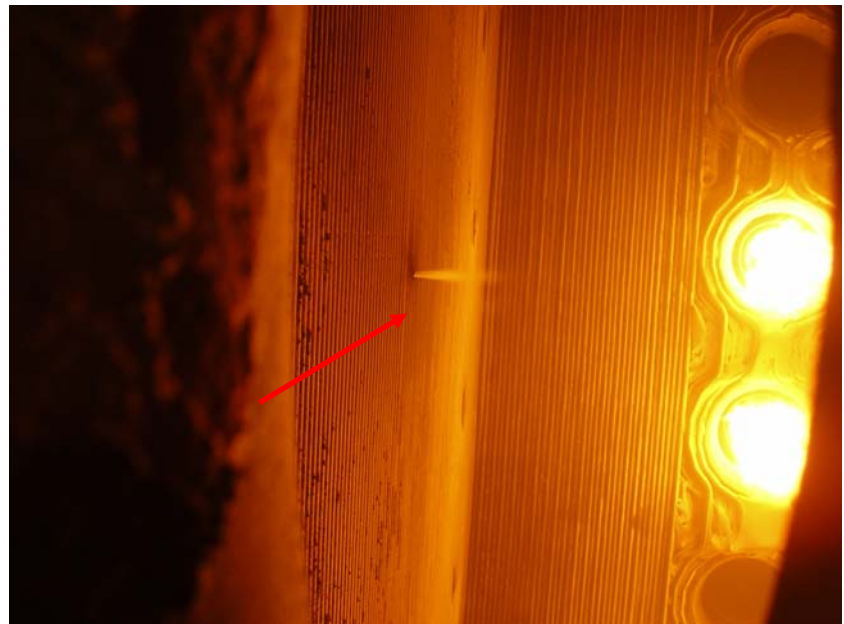
## --Annual Corrosion Costs

Corrosion Problem	O&M Corrosion Cost \$ Millions	Depreciation Corrosion Cost \$ Millions	Total Corrosion Cost \$ Millions
<b>Waterside/Steamside Corrosion of Boiler Tubes</b>	916.0	228.4	1,144.4
Turbine CF & SCC	458.0	142.8	600.8
Oxide Particle Erosion of Turbines	274.8	85.7	360.5
Heat Exchanger Corrosion	274.8	85.7	360.5
<b>Fireside Corrosion of Waterwall Tubes</b>	183.2	142.8	326.0
Generator Clip to Strand Corrosion	183.2	28.6	211.8
Copper Deposition in Turbines	91.6	57.1	148.7
<b>Fireside Corrosion of SH &amp; RH tubes</b>	91.6	57.1	148.7

Source: Syrett, et al. Low Temperature  
Corrosion, EPRI

# Background

- Typical boiler wastage rates are:
  - Subcritical – 20 mils (0.5mm)/year
  - Supercritical – 40-100 mils (1.0-2.5mm)/year
- Corrosion rates tend to increase with increasing temperatures.
- Higher operating metal temperature of supercritical boiler tubes tend to increase corrosion rates by 2-5X



# Background

- USC boiler metal temperatures will approach 1400F (760C).
  - Accelerated corrosion rates are anticipated at these temperatures
- Alloys such as P91, Super 304H (austenitic SS), and Alloy 230 (nickel-based) alloys will be required.
- Advanced austenitic SS typically exhibit poor sulfidization or coal ash resistance.
  - Reliable sulfidization and oxidation resistant nanocoatings are required for improved durability of USC boiler tubes.



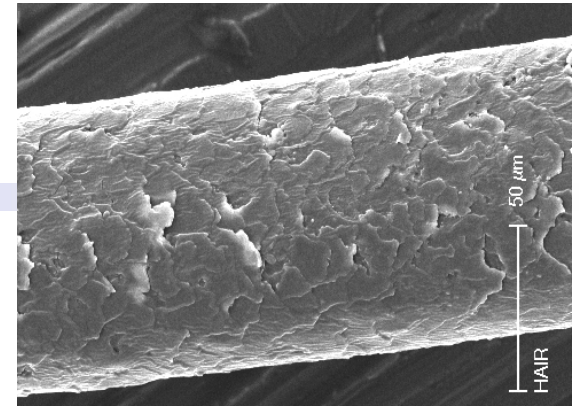
# DOE Nanocoatings Project Objectives

1. Develop/demonstrate *nano-structured coatings* using computational modeling methods to improve corrosion/erosion performance of tubing in USC boiler applications.
2. Improve the reliability and availability of USC fossil-fired boilers and oxy-fuel advanced combustion systems by developing advanced nano-structured coatings:
  - optimized utilizing science-based computational methodologies
  - validated via experimental verification and testing in simulated boiler environments

# What is a Nanocoating?

## Nanocrystalline materials/coatings:

- Single or multi-phase
- Grain size **<100 nm** (1nm = 10 Angstroms)
- Interfaces cover significant portion of the microstructure
- Dense
- Very hard and tough
- Offer excellent corrosion/oxidation/erosion properties compared to conventional materials/coatings

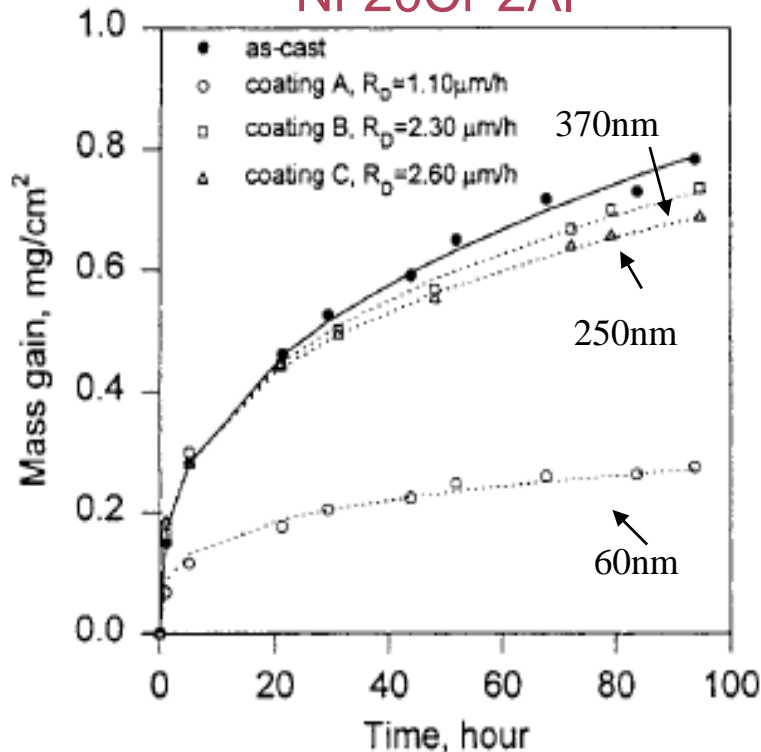


**Human Hair = 85μm = 85000nm**

# Oxidation Behavior of Nanocoatings

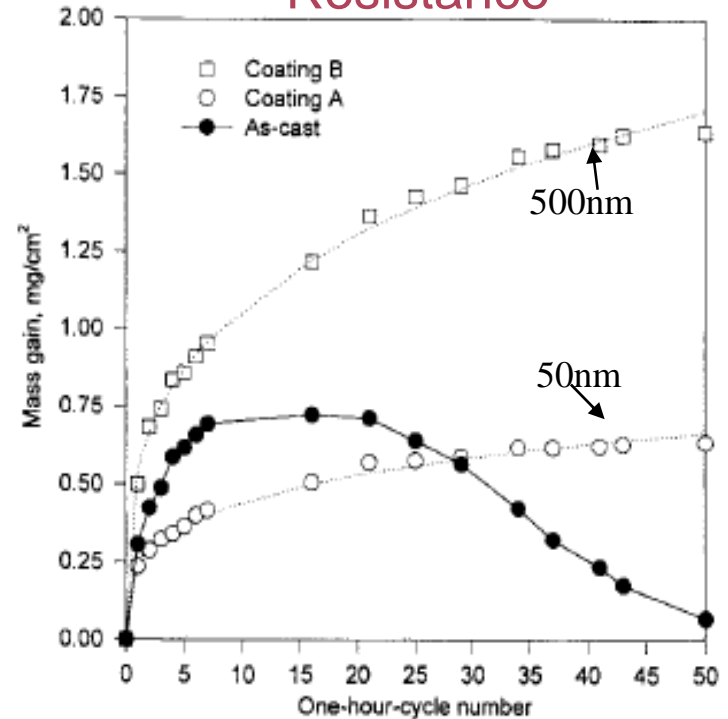
--@ 1000°C (1832°F)

## Effect of Grain Size Ni-20Cr-2Al



Lower oxidation kinetics

## Spallation Resistance



Improved scale spallation resistance

Similar results were reported for CoCrAlY/FeNiCrAl microcrystalline coatings

- 1) Gao et al., *Advanced Materials*, 2001, 13, pp. 1001, 2) Liu et al., *Oxidation of Metals*, 1999, 51, pp.403, and 3) Wang et al., *Oxidation of Metals*, 1996, 45, pp. 39

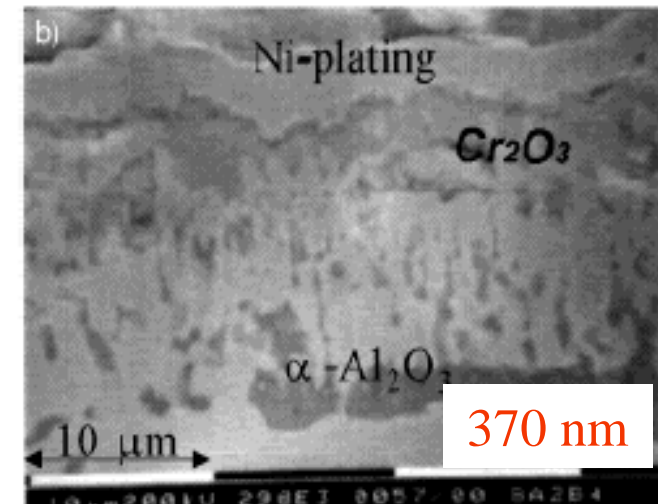
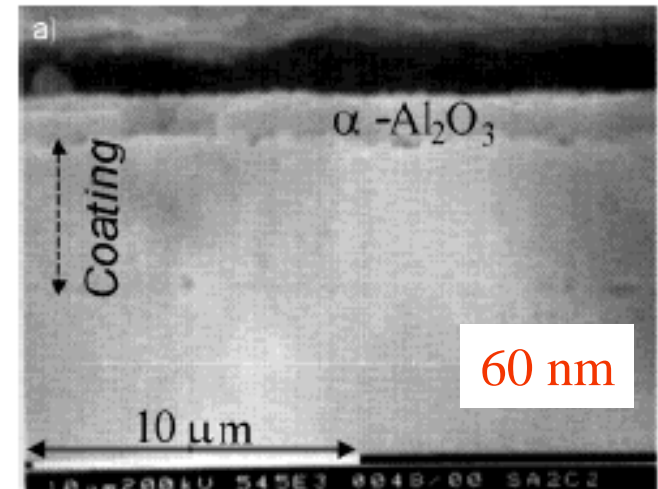


# Finer Grain Size Promotes Formation of Stable Alumina Scale

- Coating Ni-20Cr-2 Al
- Exposed @ 1000°C for 100 hrs
- Fine grained coating (60nm) exhibited continuous protective alumina scale
- Coarse grained coating severely oxidized

Finer grain size lowers minimum  
Al requirement!

**This would increase life by 3 to 4X**



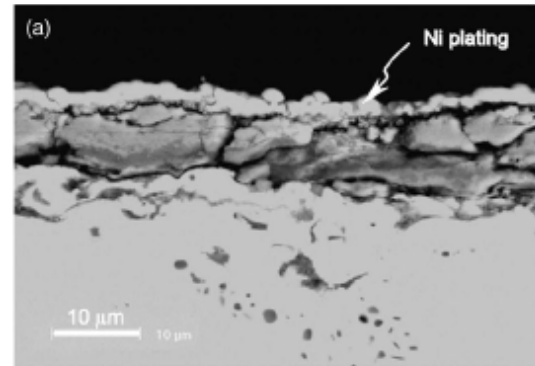
Gao et al., Nano and Microcrystal Coatings, *Advanced Materials*, 2001, 13, pp. 1001

# Corrosion Resistance of Nanocoatings

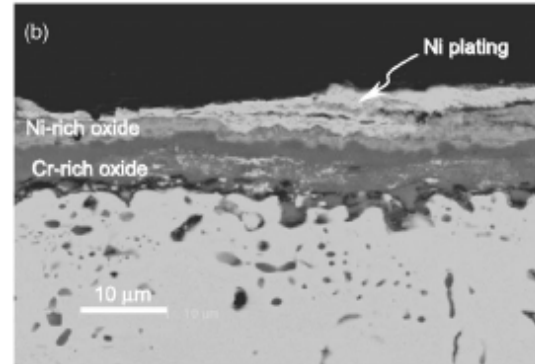
--Finer grain size also lowers minimum Cr requirement!!

For corrosion resistance formation of chromia scale is crucial  
Typically 20% Cr is required for protective scale formation

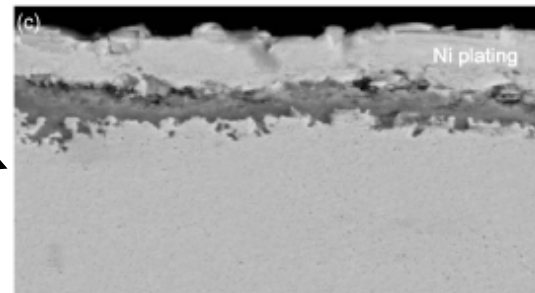
- Ni-10% conventional coating was severely corroded (no continuous chromia scale!)
- Continuous chromia scale was formed under the Ni-rich oxide in Ni-20% Cr coating - showed minor internal sulfidation
- Nanostructured Ni-11% coating showed continuous chromia scale (no sulfidation)



Ni-10% Cr  
Conventional



Ni-20% Cr  
Conventional



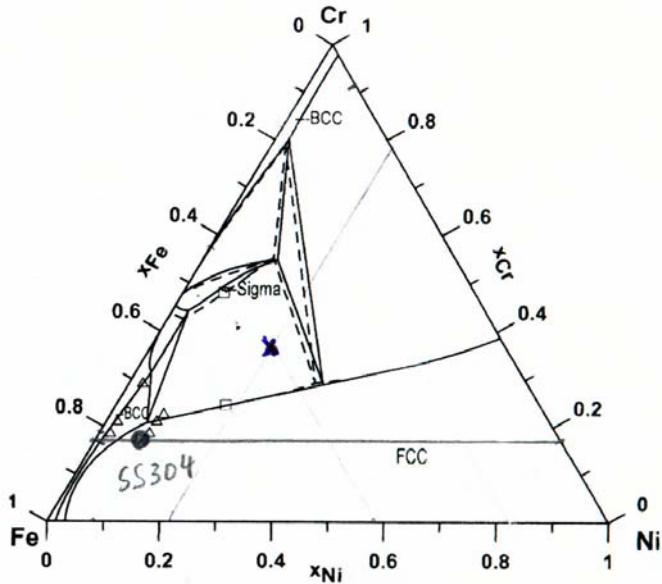
Ni-11% Cr  
Nanostructure

K<sub>2</sub>SO<sub>4</sub>-20%NaCl-10K<sub>2</sub>SO<sub>4</sub> @ 700°C for 50 hrs

# Project Tasks

---

- Task 1: Computational Modeling of MCrAl Systems
- Task 2: Establishment of Baseline Coating Data
- Task 3: Process Advanced MCrAl Nanocoatings
- Task 4: Fire-Side Corrosion Testing
- Task 5: Computational Modeling & Validation
- Task 6: Mockup Demonstration
- Task 7: Project Management & Reporting



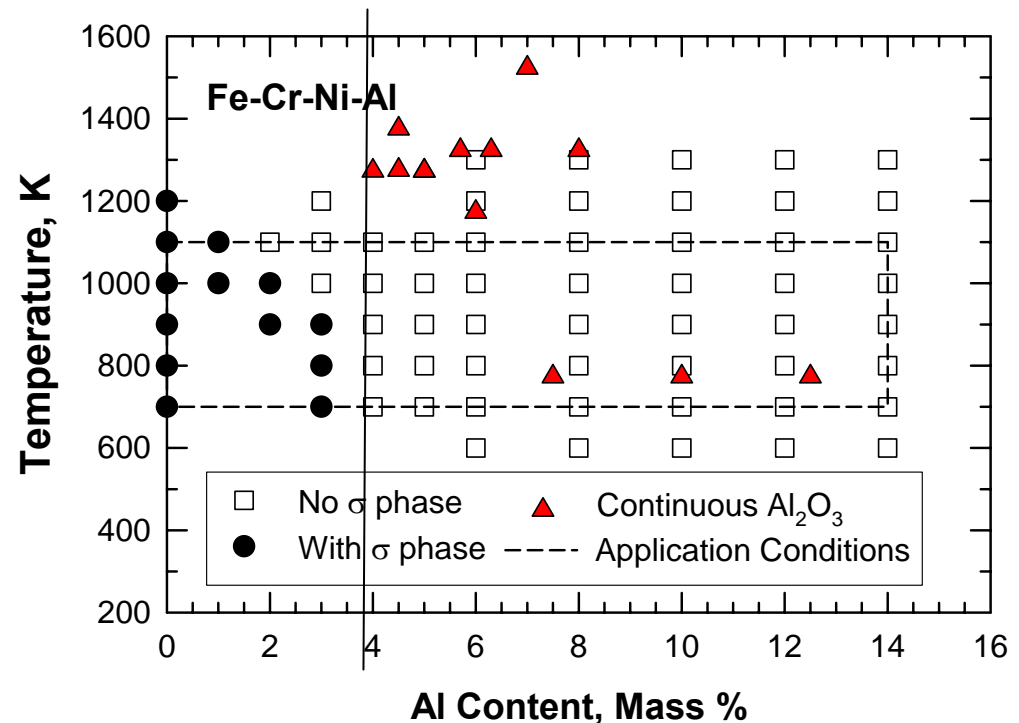
## Task 1: Computational Modeling of MCrAl Systems

--completed

Objective: Select potential MCrAl nanostructured coating compositions using computational modeling

# Task 1- Computational Modeling Of MCrAl Composition Selection

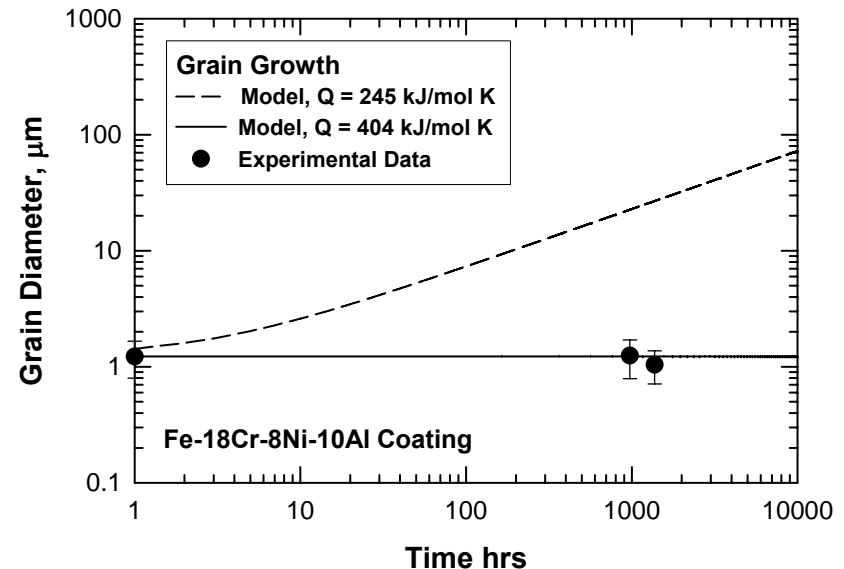
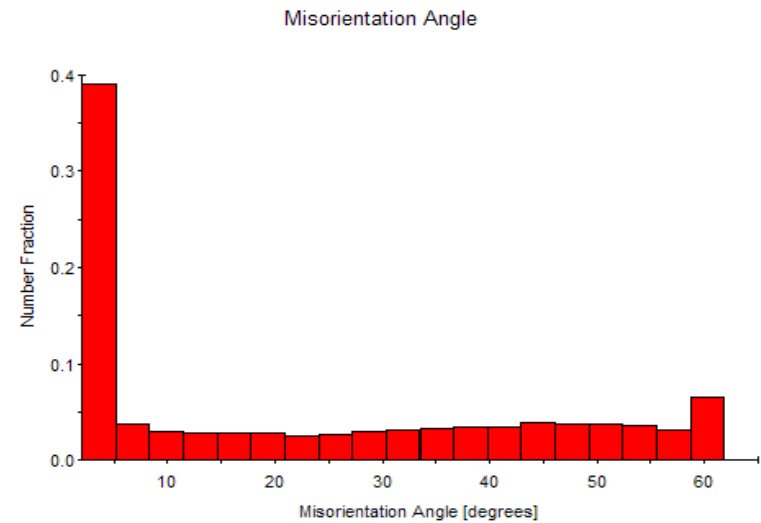
- Completed computations of FeNiCrAl<sub>x</sub> phase diagrams
- Al additions suppress sigma phase formation, while Mo and Co promotes.
- 4-5%Al is required to form a continuous Alumina scale.
- To ensure sufficient Al source, 10% Al is selected.
- Suggested nano-coating systems for evaluation:
  - 310 +10%Al
  - 310 +30-35Ni +10%Al
  - Fe-35-40Ni-25Cr-10%Al



# Task 1 - Computational Modeling

## --Grain Growth Modeling

- Grain growth model results showed that the presence of high concentration low angle boundaries in the coating **prevents grain growth**
- Thermal exposure at 750°C leads to sintering
- Coating/interface toughness was determined
- The sputtered coatings exhibited good toughness
- Indentation testing showed no coating delamination





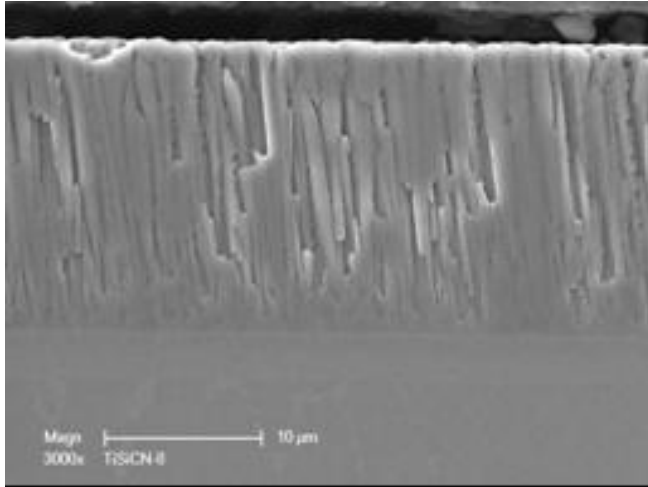
# Candidate Nanocoatings for Evaluation

## Model Recommendation

- Coating composition: Fe-25Cr- 30/40Ni-10 Al

## Candidate Nanocoatings Selected For Evaluation

- 310 + Al (Fe-25Cr-20Ni-10Al)
- Haynes 120+Al (Fe-37Ni-25Cr-Al)
- Haynes 160+Al (Ni-29Cr-28Cr-3Si-Al)
- Haynes 188+Al (Co-22Ni-22Cr-14W-10Al)



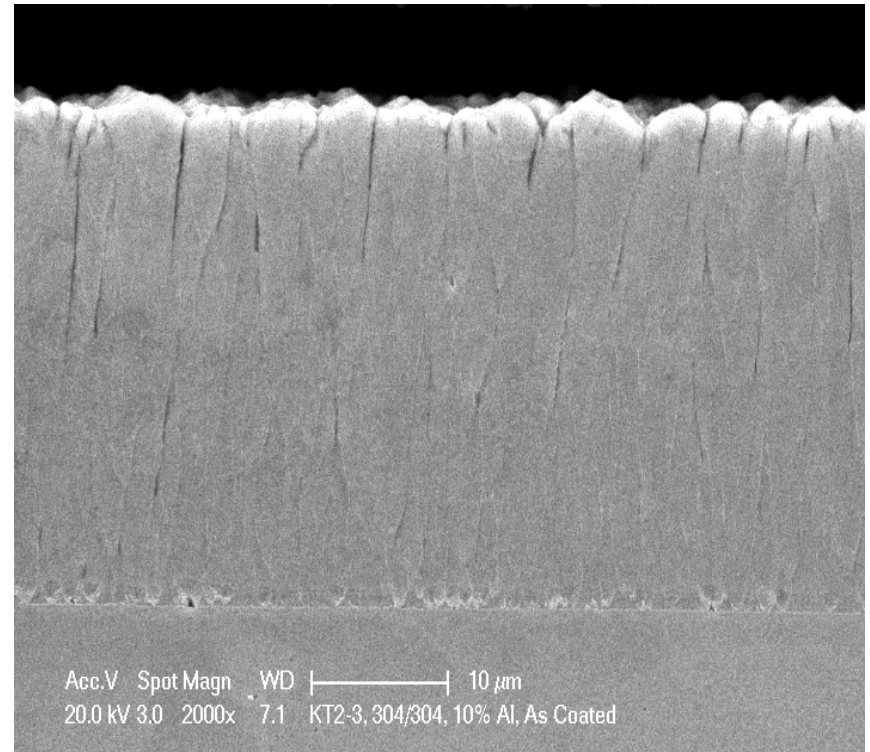
## Task 2: Establishment of Baseline Coating Data

--completed

**Objective: Evaluate conventional-  
and available vendor  
nanocoatings to assess properties**

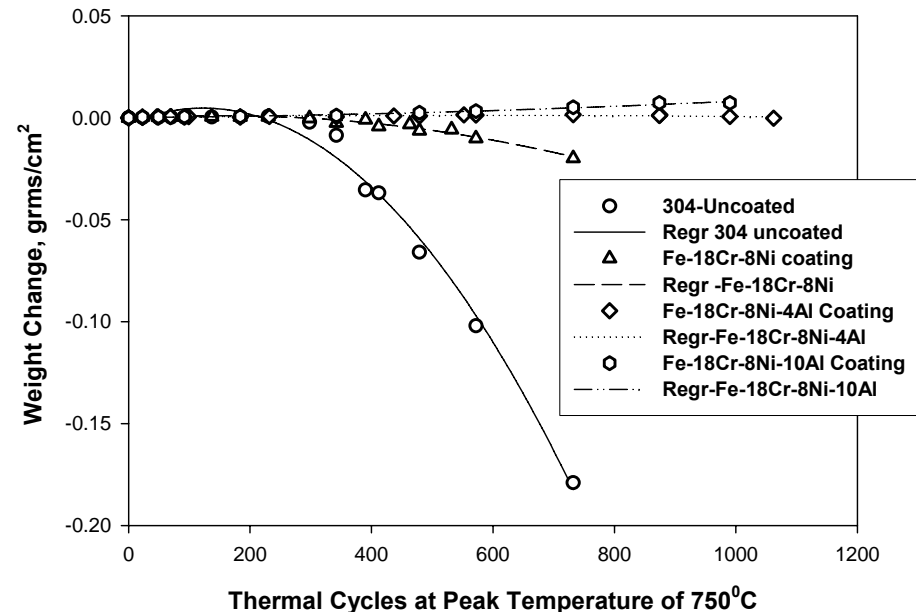
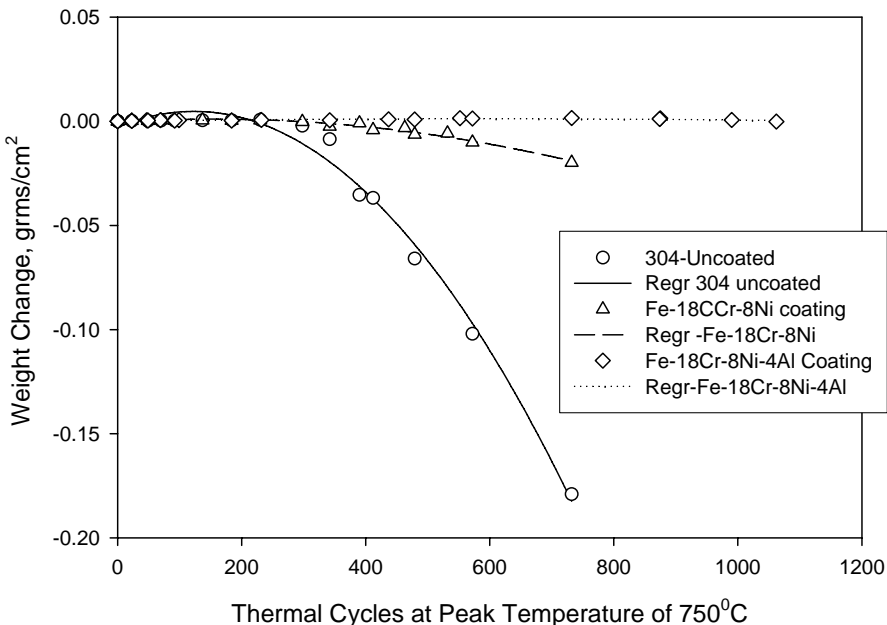
## Task 2. Establishment of Baseline Coating Data

- Fe-18Cr-8Ni-xAl Nano-coatings were deposited on 304SS and P91 samples
- Ni-20Cr-xAl Nano-coatings were deposited on Haynes 230 and P91 samples
- Long term cyclic oxidation tests were conducted on the coated samples



# Task 2. Baseline Coating Data

## --Cyclic Oxidation Behavior of Fe-18Cr-8Ni-xAl Coatings

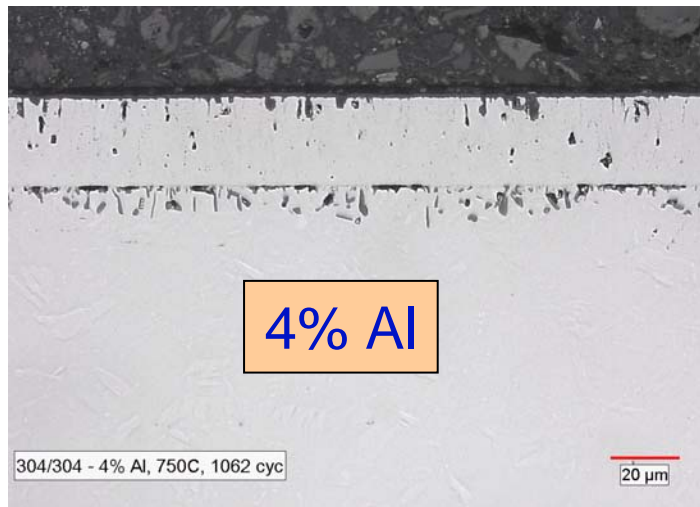
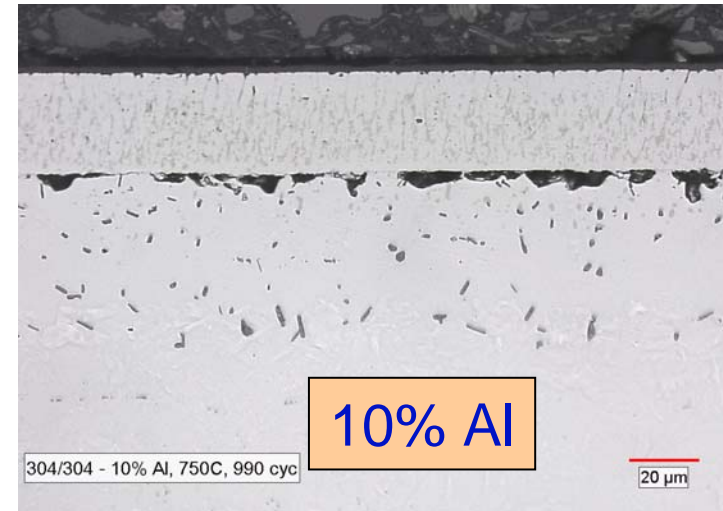
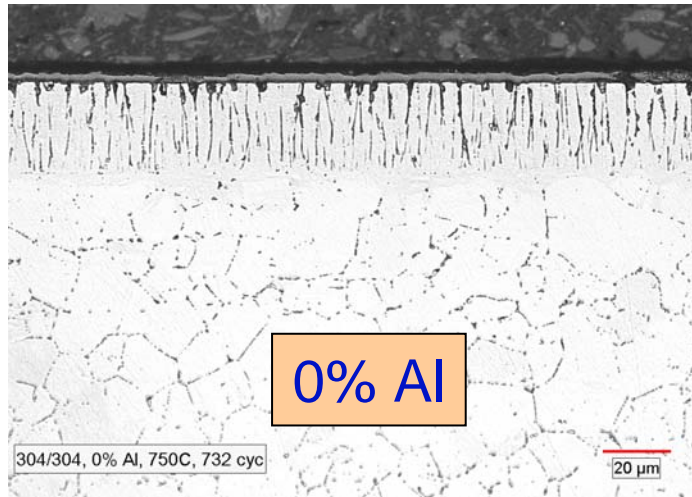


- Fe-18Cr-8Ni-xAl Nano-Coating Improved Oxidation Resistance **By 2X.**
- The **Addition Of Al** Improved
  - Oxide-Scale Spallation Resistance
  - And Oxidation Resistance

## Task 2. Baseline Coating Data

### -- Coating Oxidation Characterization

#### Fe-18Cr-8Ni-xAl



- 1). The protective oxide layer
  - 0% Al coating  $\text{Cr}_2\text{O}_3$
  - 4 and 10%Al coatings  $\text{Al}_2\text{O}_3$
- 2). 4% Al coating was oxidized
- 3). 10% Al coating was free from internal oxidation
- 4). Inward diffusion of Al led to formation of inter-diffusion zone

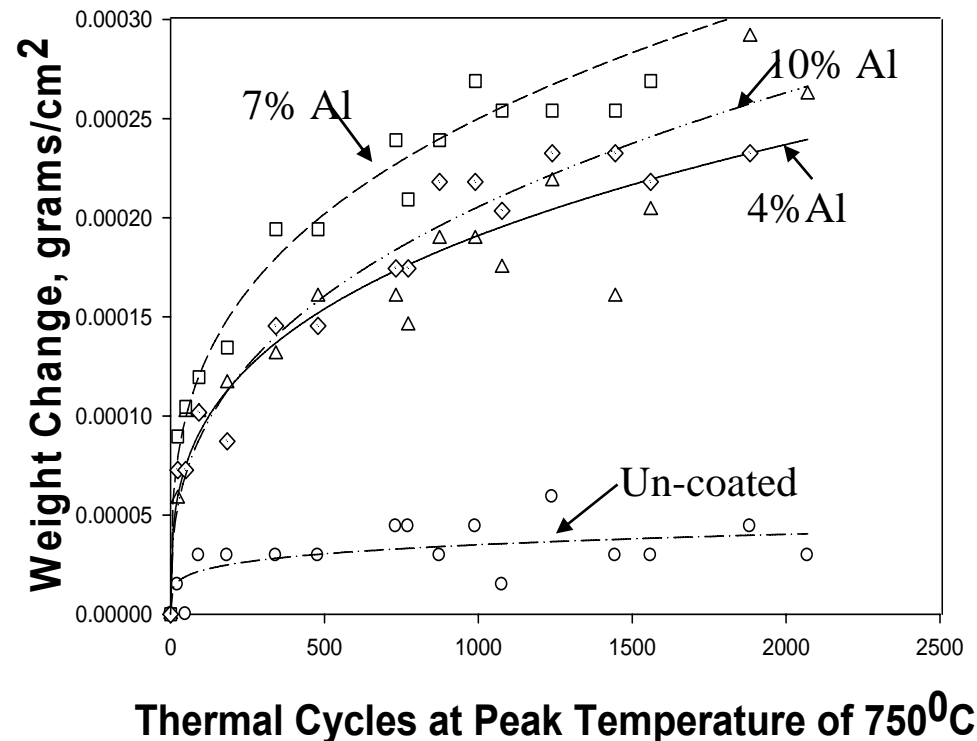
## Task 2. Baseline Coating Data

### --Cyclic Oxidation Behavior of Ni-20Cr-xAl Nanocoated and Uncoated Samples at 750°C

Continuous mass gain of the coated samples indicates the outward growth of oxide scale

Scale is highly resistant to spallation

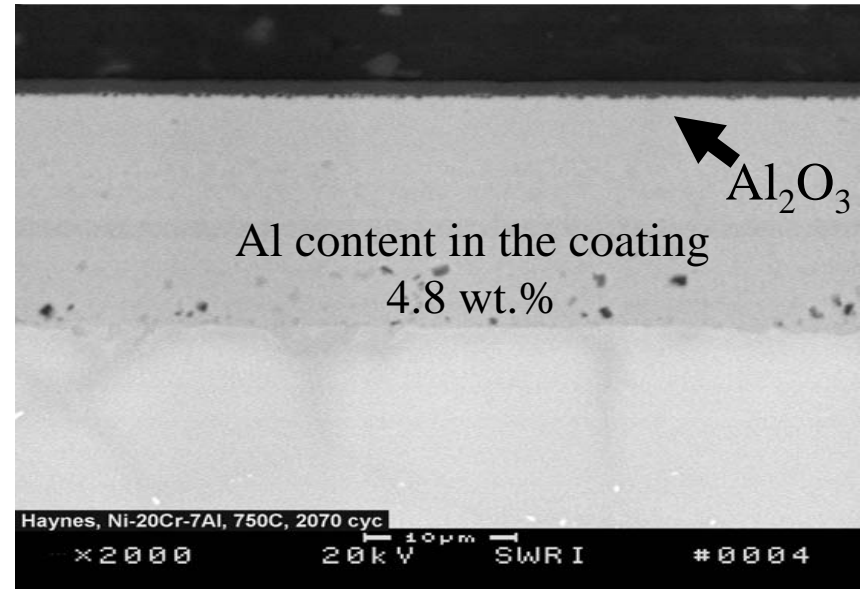
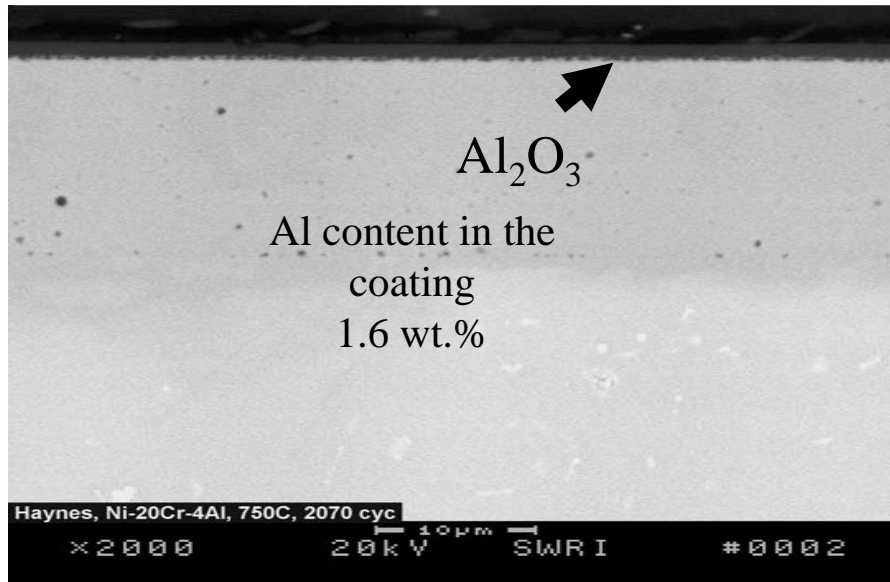
Haynes 230 may not need a coating at 750C



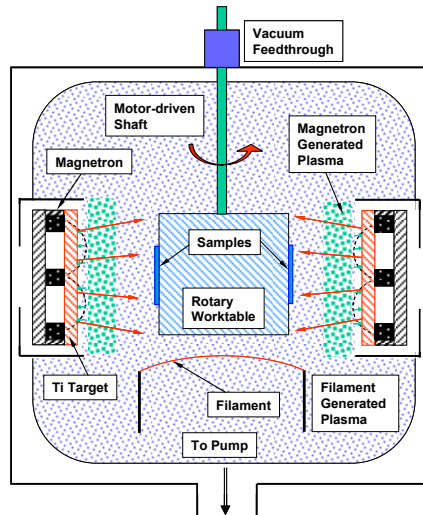


## Task 2. Baseline Coating Data

### --Oxide Scale on Ni-20Cr-xAl NanoCoating After Exposure at 750°C



- Protective external oxide scale is dense, free from cracks
- Highly resistant to spallation and coating is free from internal oxidation
- Continuous mass gain is due to outward growth of the scale
- Al consumption rate is high
- The coating with 4% Al may not be durable for long-term service
- Established need for at least 7% Al for long-term durability



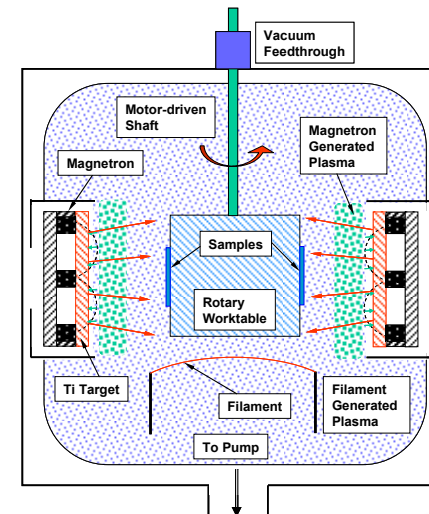
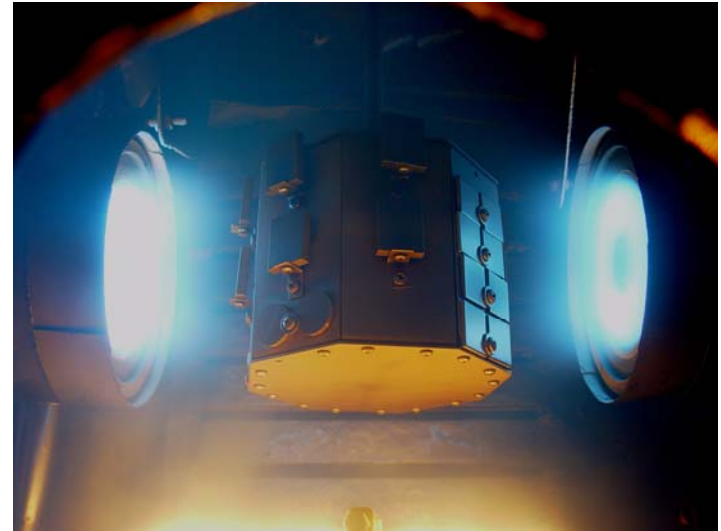
# Task 3: Processing of Advanced NanoCoatings

--in progress

**Objective: Apply advanced nanocoatings on USC substrates and evaluate properties**

# Task 3 - Processing of Advanced NanoCoatings

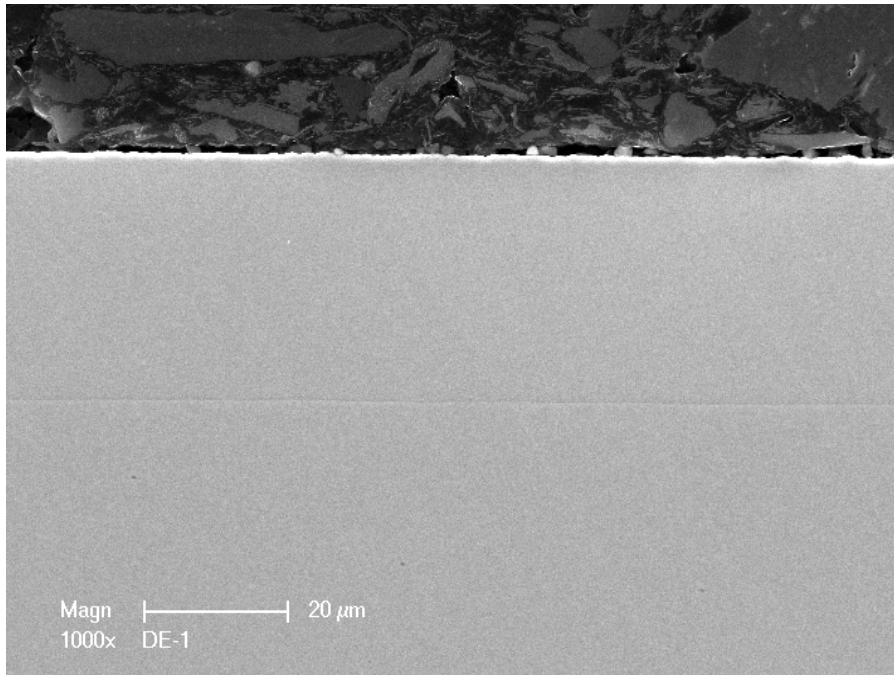
- Plasma Enhanced Magnetron Sputtering
- Process optimization study was conducted to improve the density of the coating
  - power applied to targets and bias voltage were varied
- Several trial coatings: 304+10Al, 310+10Al and Ni-20-10Al were deposited
- Coated samples were destructively examined



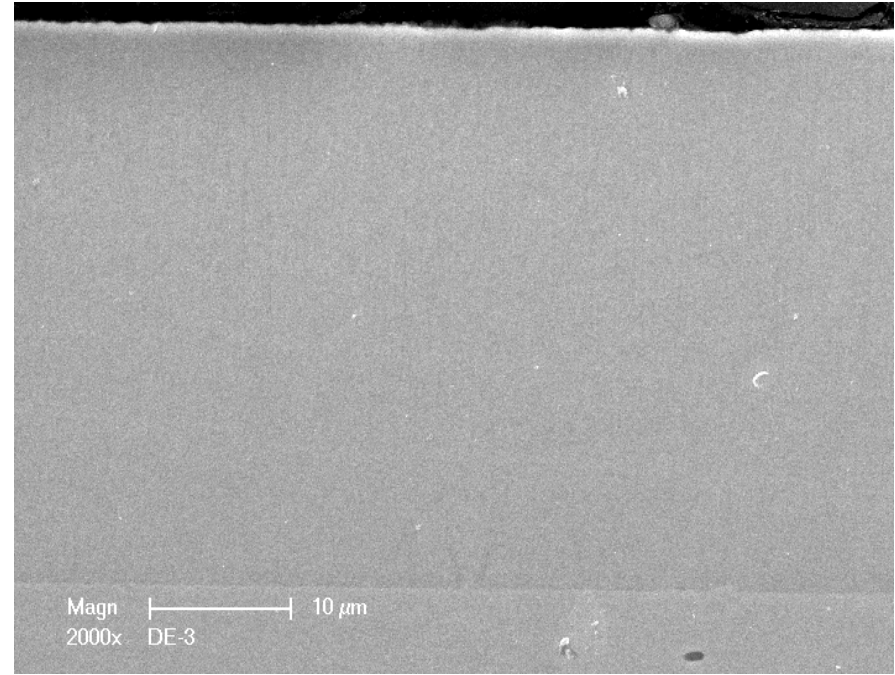
# Task 3 - Processing of Advanced NanoCoatings

## --Metallographic Examination

Result: Dense Nanocoating Microstructures



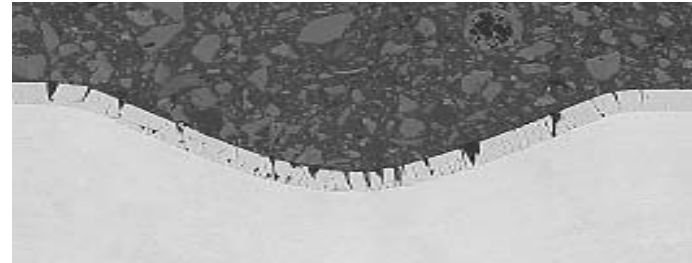
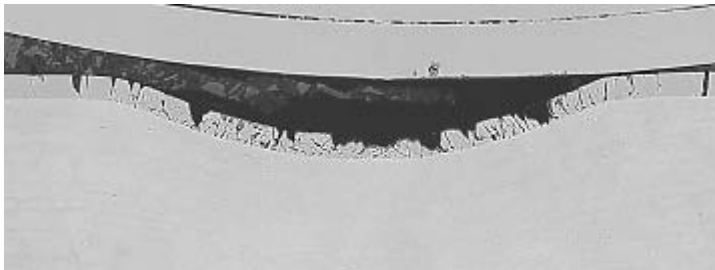
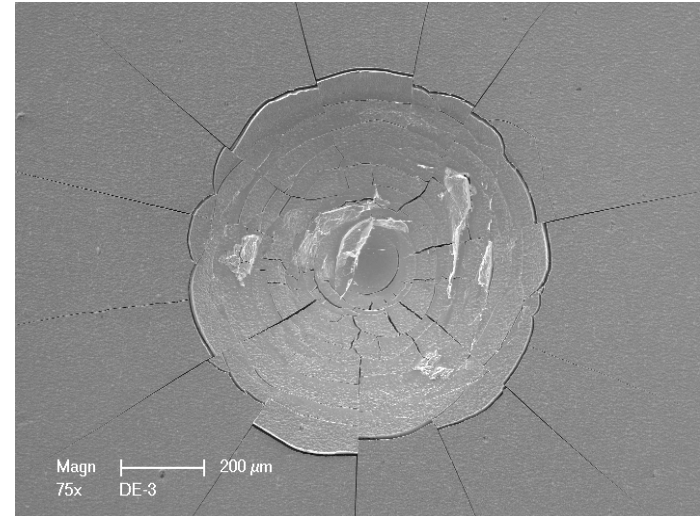
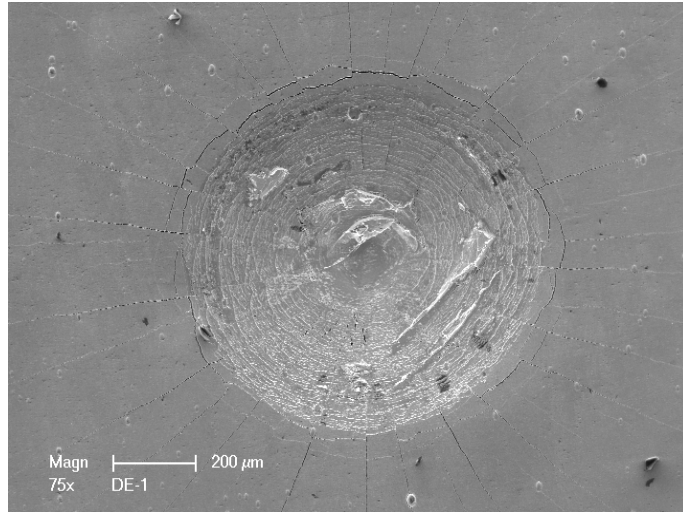
1000X



2000X



# Task 3 - Processing of Advanced NanoCoatings -- Coating Adhesion Results



**No coating delamination,  
DE 3 parameters were selected**

## Task 3 - Processing of Advanced NanoCoatings -- Selection of Process

- Process parameters of DE 3 produced good quality coating
- Four advanced coatings (next slide) were deposited on P91, 304 SS and Haynes 230 samples for corrosion testing using DE 3 parameters

### Concerns:

- ✓ Accelerated Al consumption rate
- ✓ Repeatability of Coating Quality
- ✓ Coating Cracking

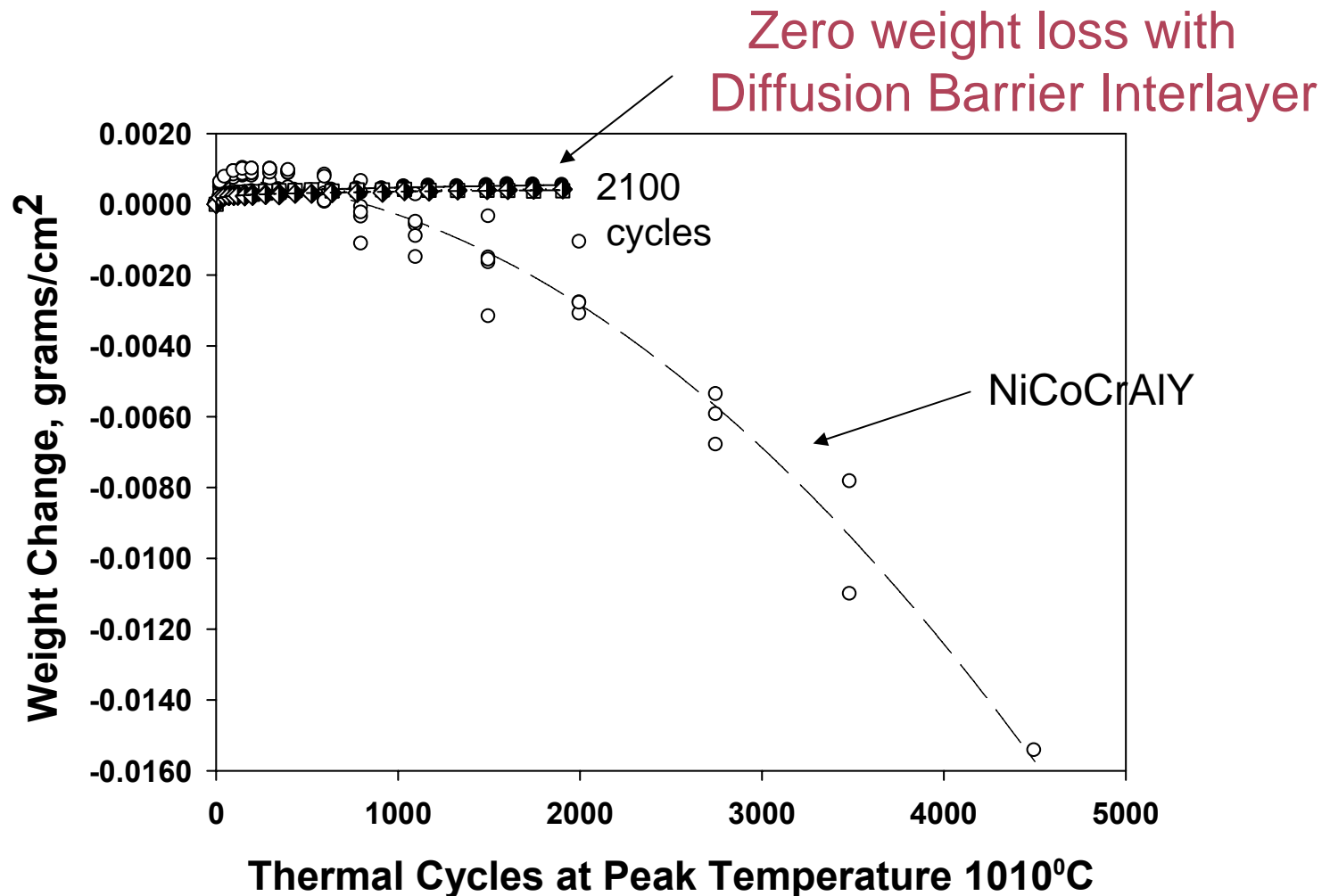


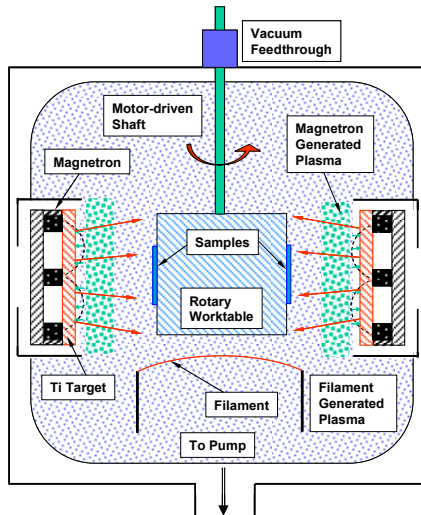
## Task 3 - Processing of Advanced NanoCoatings

- Four advanced coatings were deposited:
  - 310 + Al (Fe-25Cr-20Ni-10Al)
  - Haynes 120+Al (Fe-37Ni-25Cr-Al)
  - Haynes 160+Al (Ni-29Cr-28Cr-3Si-Al)
  - Haynes 188+Al (Co-22Ni-22Cr-14W-10Al)
- Coated P91, 304SS and Haynes 230 samples were shipped to Foster Wheeler for Corrosion Testing
- Process optimization trials for depositing interlayer and advanced coatings are in progress

# Task 3 - Processing of Advanced

## --Comparison of Nanocoatings with Conventional Plasma-Sprayed NiCoCrAlY





# Task 4: Fireside Corrosion Testing

--in progress

**Objective: Conduct accelerated  
fireside corrosion tests**

## Task 4 - Corrosion Testing

- In addition to the 4 newly developed test nanocoatings, two vendors supplied nanocoatings for corrosion testing:
  - N-TECH Inc.
  - NanoSteel



# Task 4 - Corrosion Testing

## --Testing Conditions

### Waterwall Testing

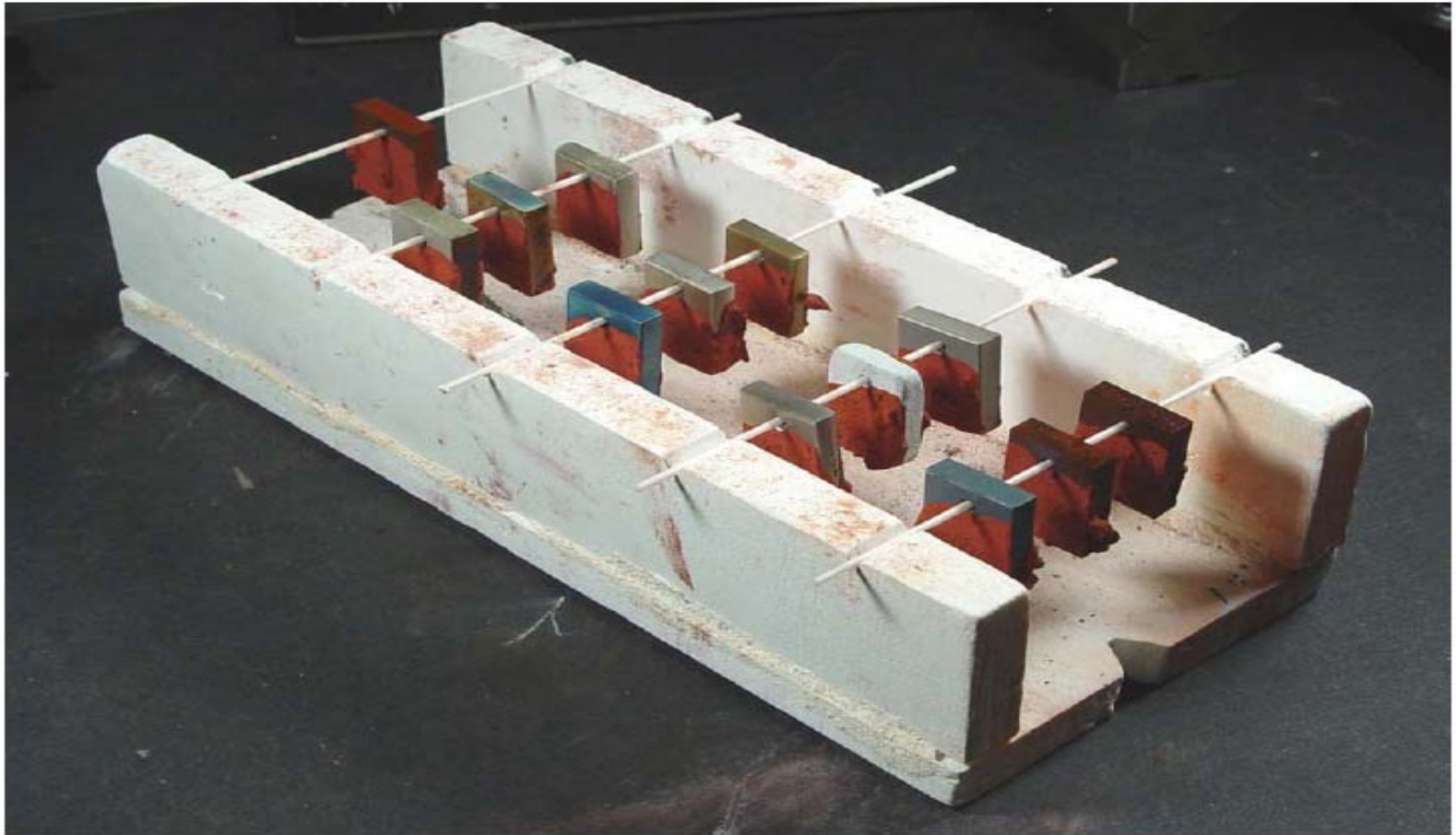
- 850F (454C), 975F (524C), and 1100F (593C)
- 40 percent FeS and 0.2 percent chlorides (same as USC program)

### Superheater/Reheater Testing

- 1100F (593C), 1300F (704C), and 1500F (816C).
  - 5 percent alkali sulfates (same as USC program)
  - simulate Eastern bituminous coal compositions.
- Perform 1000hr tests (at 100hr intervals).

## Task 4 - Corrosion Testing

### -- Following first 100hrs of testing



Waterwall Test Rack at 850F (100 hrs)



## Task 4 - Corrosion Testing

### -- Following 500hrs of testing



- A few nanocoatings still in good shape after 500hrs exposure.
- Many samples exhibited some level of blistering.
  - Suggests coating processing issues.
- Metallography just starting
- Will require recoating (process improvement warranted) and additional corrosion testing.

# Conclusions To Date

- For long-term term durability, nano-coatings should contain ~10%Al.
  - A continuous, Al-rich protective oxide scale can be achieved with this level of aluminum.
- 4 nanocoatings selected using computational thermodynamics.
- Good cyclic oxidation performance exhibited for baseline nanocoatings (Fe-18Cr-8Ni-xAl, Ni-20Cr-xAl)
- Early nanocoatings corrosion results suggest processing concerns—more work to do!
- A diffusion barrier layer slows Al consumption rate dramatically.

# What's Next

- Complete fire-side corrosion testing (1000 hr) of nanocoatings at Foster-Wheeler (Task 4)
- Repeat coating processing trials (Task 3) and additional corrosion work.
- Validate computational model (Task 5)
- Nanocoating mockup demonstration (Task 6)

# Together...Shaping the Future of Electricity