

# Novel and Improved Electrode Structures Through Infiltration

Steve Visco

Mike Tucker, Craig Jacobson,  
Tal Sholklapper, Grace Lau,  
Lutgard De Jonghe

Lawrence Berkeley National Laboratory  
Berkeley, California USA



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

Presented at 8<sup>th</sup> Annual SECA Workshop and Peer Review  
*Core Technology Program – Electrodes: Performance and Degradation*

August 7<sup>th</sup> – 9<sup>th</sup> 2007

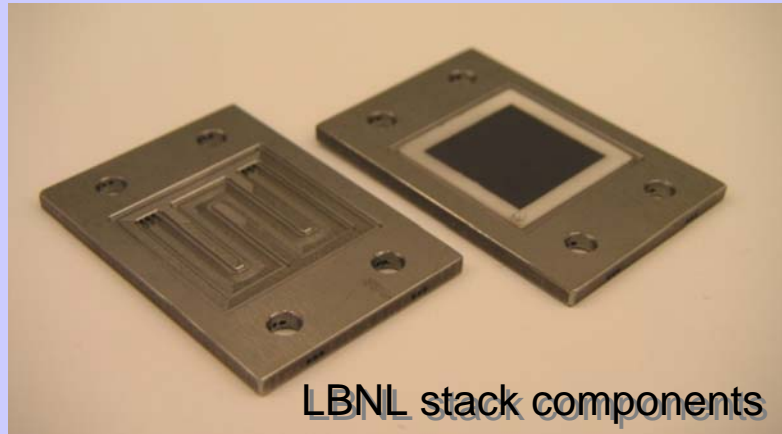
# LBL SECA Core Program

In FY07 the LBNL core effort was focused on the following areas:

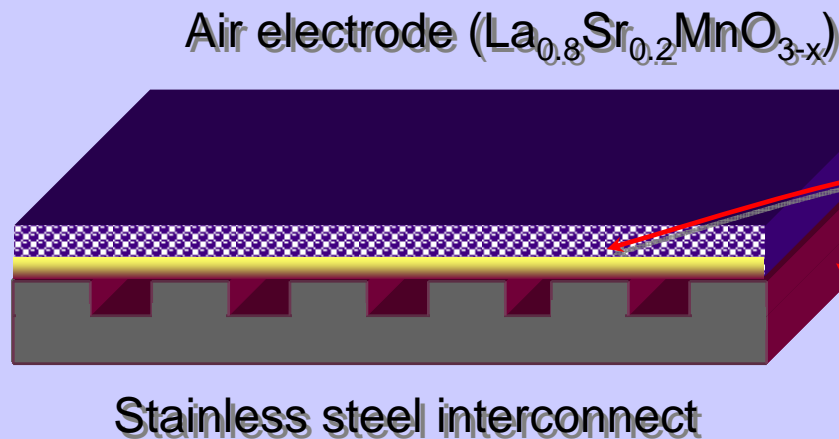
- 1) Infiltration of perovskites and other appropriate catalysts into composite cathodes to form a interconnected network of nanoparticulate coating;
- 2) Infiltration of ceria and other appropriate materials into Ni-YSZ anodes to improve sulfur tolerance;
- 3) Determination of baseline performance and long term stability of infiltrated and non-infiltrated electrodes;
- 4) Design and fabrication of 2-cell stack for national labs and industrial teams as a standard platform for testing electrodes, interconnects, contact paste, and seals in a manner that allows reliable comparison across research teams;
- 5) Continued optimization of interconnect coating technology and elucidation of the mechanism of chromium migration through protective coatings.



# Metal Stability & Interactions

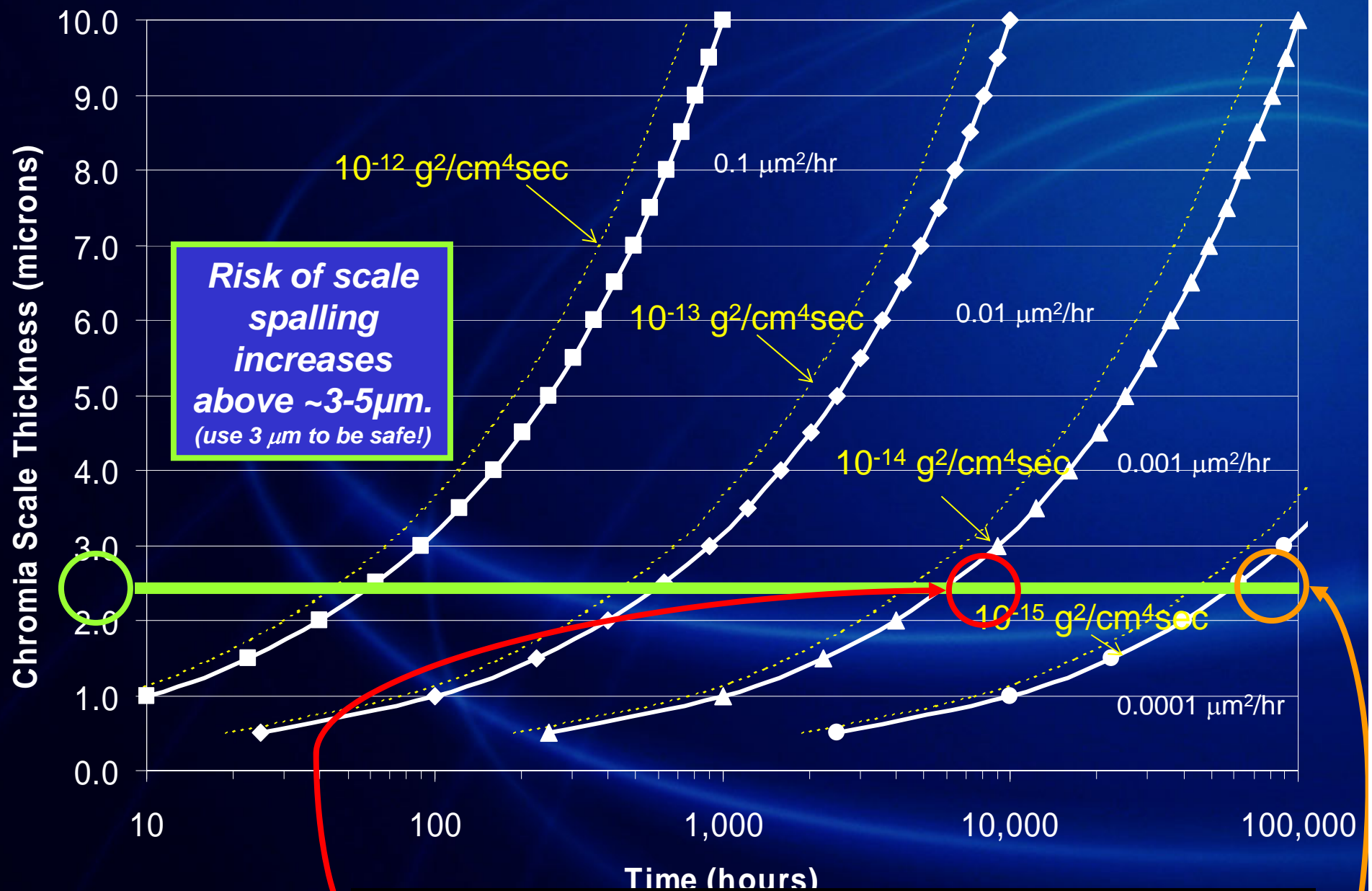


- Oxidation behavior
- Oxide spallation
- Area specific resistance
- Chromium migration



## Transpiration

- Vapor chromium transport
- Bulk & grain boundary Cr transport
- Surface migration



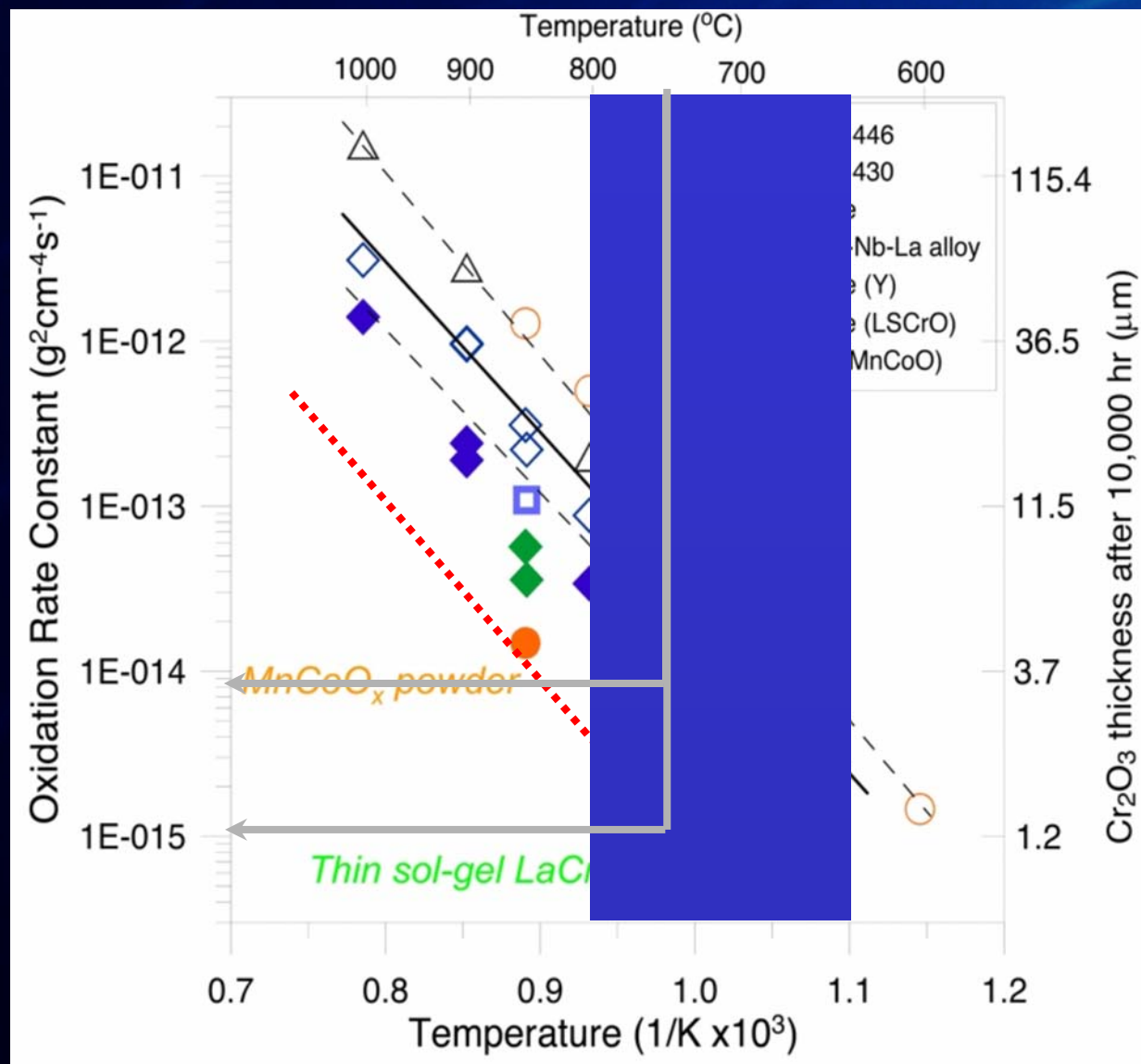
ERNEST ORLAND  
BERKELEY NATION

$\sim 10^{-14} \text{ g}^2/\text{cm}^4\text{sec}$  for transportation 5,000 – 10,000 hrs

$\sim 10^{-15} \text{ g}^2/\text{cm}^4\text{sec}$  for stationary 50,000 – 100,000 hrs

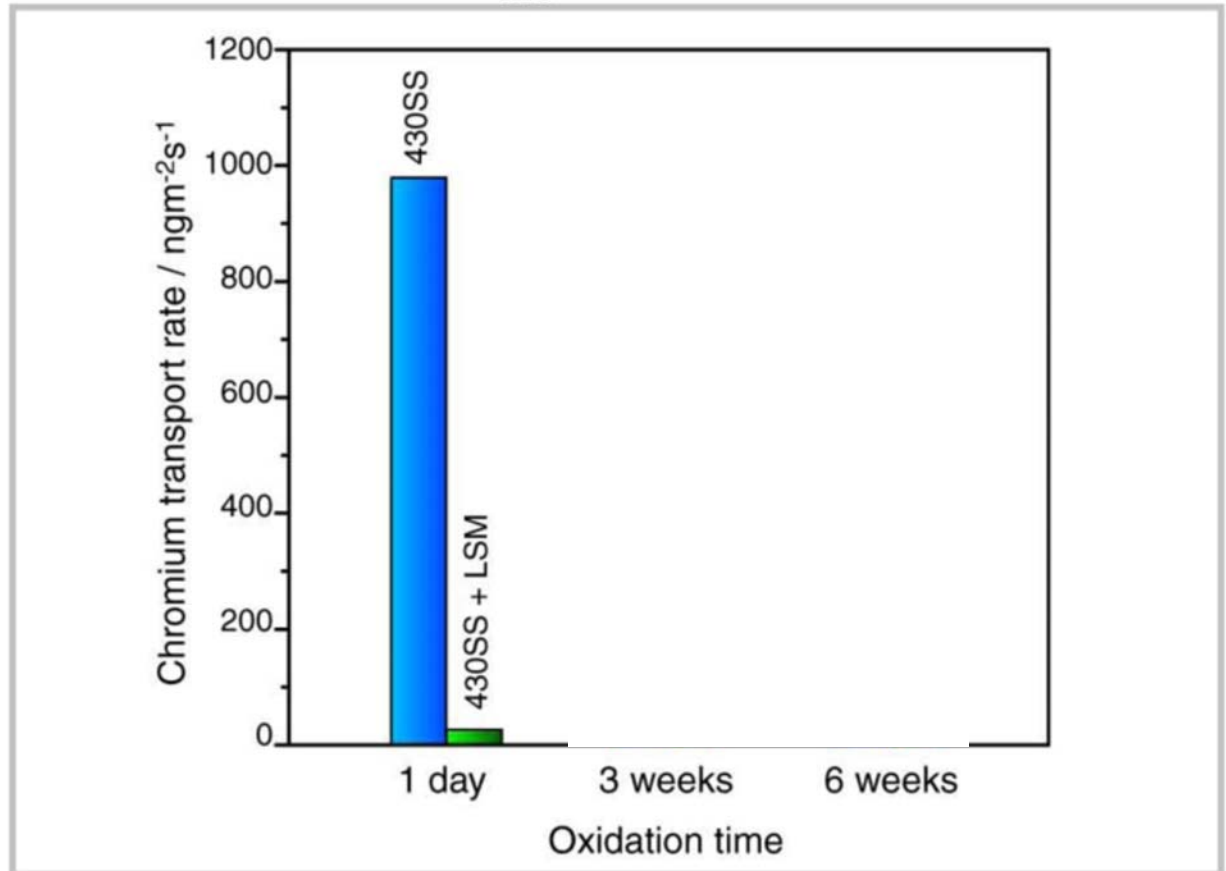
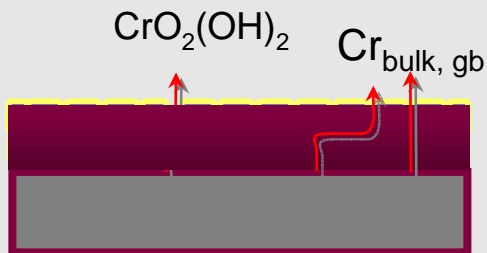


# High Temperature Oxidation of Metal Components



# Long-term Stability of Coatings for Preventing Cr Loss

- Oxidation: 1073 K,  $P_{\text{H}_2\text{O}} = 2.0 \times 10^3 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3\text{s}^{-1}$  (200ml/min)
- Cr test: 1073 K, 86.4 ks (24 hrs),  $P_{\text{H}_2\text{O}} = 1.0 \times 10^4 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3\text{s}^{-1}$  (200ml/min)



# Condition for minimum spallation of (~1%) scales on 430ss after isothermal oxidation and fast cooling to RT

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

The lower the operating temperature the thicker the scale can be

$\text{Cr}_2\text{O}_3$  not only grows slower but also can be thicker before failure

RE slow scale growth and increase adhesion/thickness

Reducing atmosphere treatment also increase adhesion

Sweet spot between 650–750 C

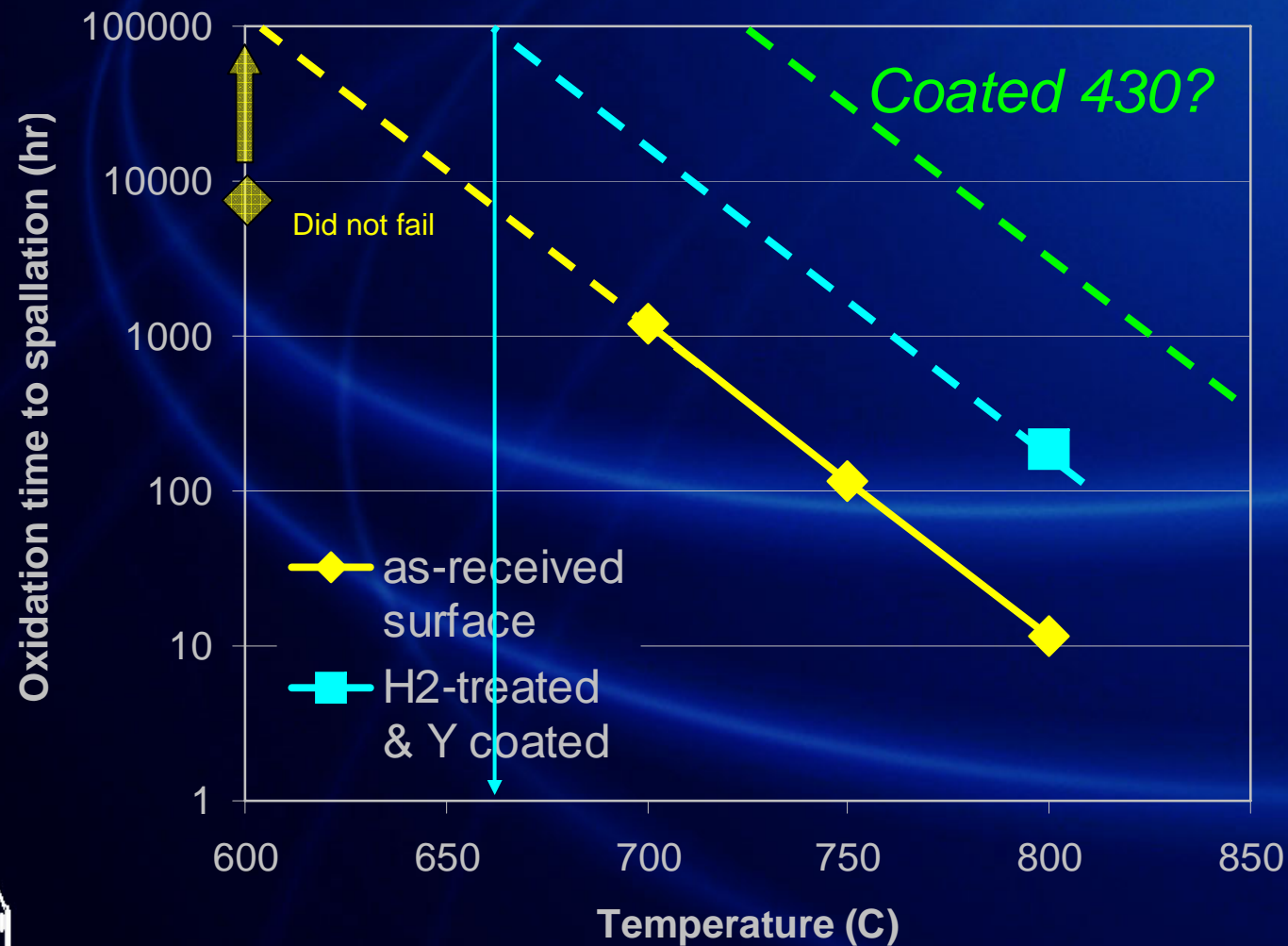
*Scale thickness decrease because of higher thermal stresses and/or more defect formation at high oxidation temperatures*



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# Conditions to reach ~1% Spallation in static air after isothermal oxidation and fast cooling to RT

## Time to Minimum Spallation

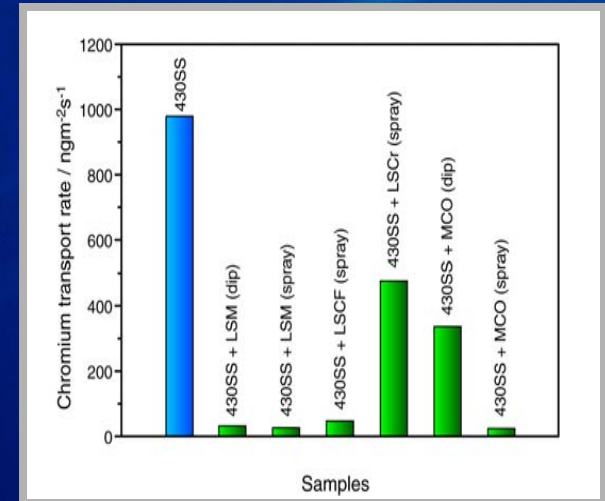




# What have we done to solve the Cr problem?

## Cr Evaporation

- Coat steel to prevent Cr diffusion to electrodes
- Density of coating seems more important than coating material



## Cr Deposition

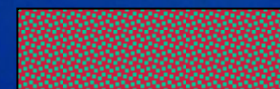
- Pairwise MOx-Cr interactions suggest Cr tolerant catalysts
- Enhance Cr tolerance of commercially available electrodes by infiltration

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

## Rxn Couples



LSM +  $\text{Cr}_2\text{O}_3$

LNF +  $\text{Cr}_2\text{O}_3$



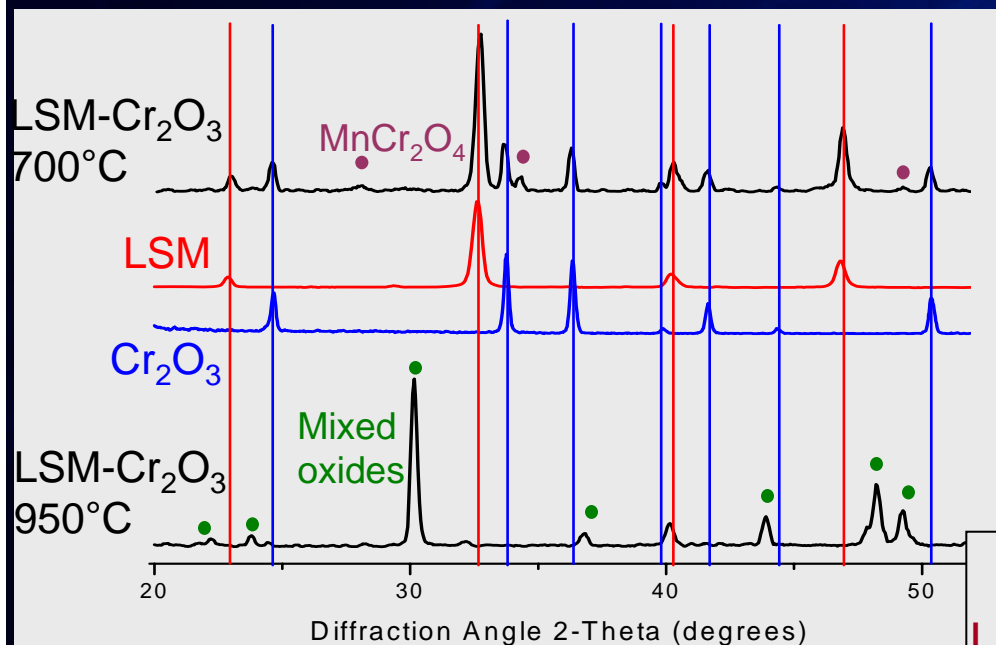
QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are required to see this picture.



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

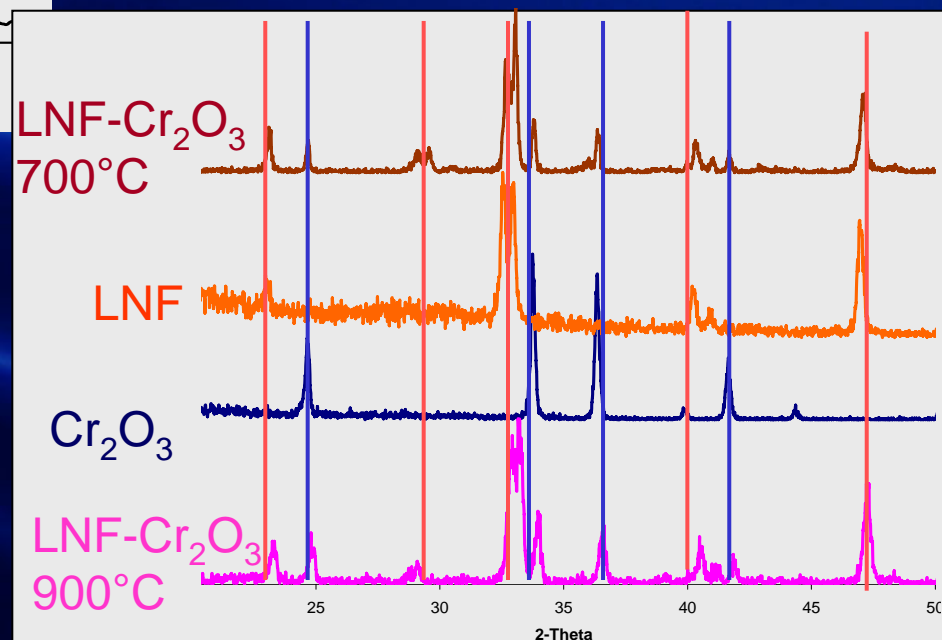
# LNF Does Not React With $\text{Cr}_2\text{O}_3$

Pellets of  $\text{LNF-Cr}_2\text{O}_3$  and  $\text{LSM-Cr}_2\text{O}_3$  powder mixtures reacted for 150h at  $700^\circ\text{C}$  and  $900\text{--}950^\circ\text{C}$



$\text{LSM-Cr}_2\text{O}_3$   
 $700^\circ\text{C}$  minor reaction  
 $950^\circ\text{C}$  complete reaction

$\text{LNF-Cr}_2\text{O}_3$   
 $700^\circ\text{C}$  NO reaction  
 $900^\circ\text{C}$  NO reaction



# LBNL Infiltration Core Program

- Improve existing structures (and novel electrode design)
  - Improve cathode performance at low temperature
  - Improve tolerance to Cr
  - Sulfur tolerant anodes
- Novel Electrode Design
  - Infiltration technology allows flexibility in SOFC design and processing
  - Enables mSOFCs



# Infiltration Structures & Challenges



Electrolyte supported: porous electrodes - straightforward



Anode supported: cathode is straightforward, anode may be too dense in unreduced state



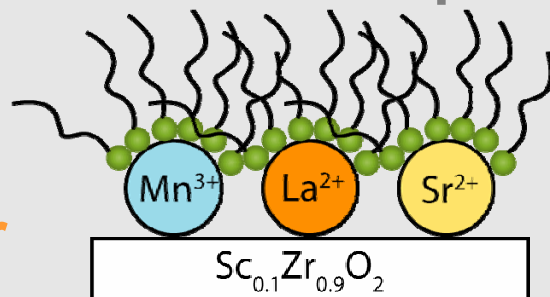
Metal supported: engineered for infiltration - entire electrode structure is infiltrated



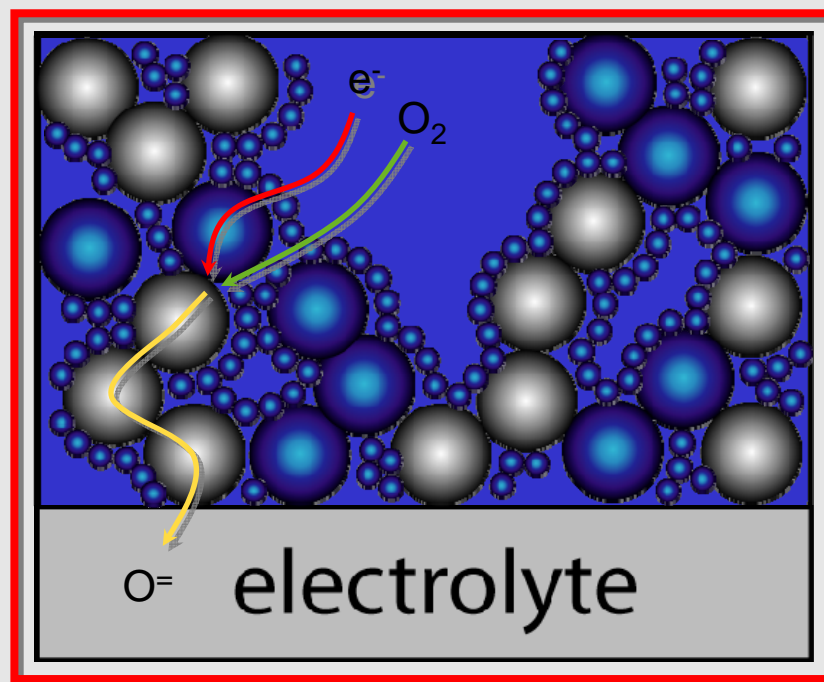
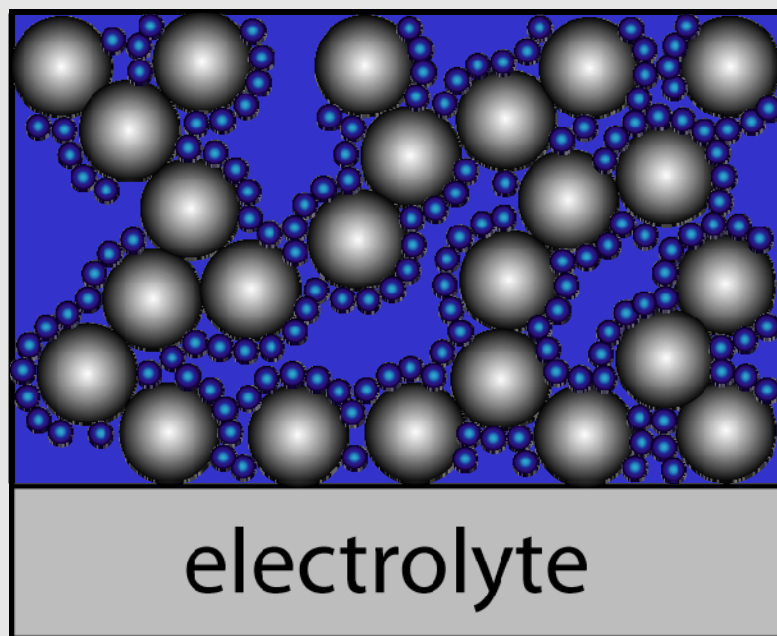


# Infiltration Step

Nitrate-Surfactant  
Concentrated Precursor



Surfactant dispersed Electrode Precursors

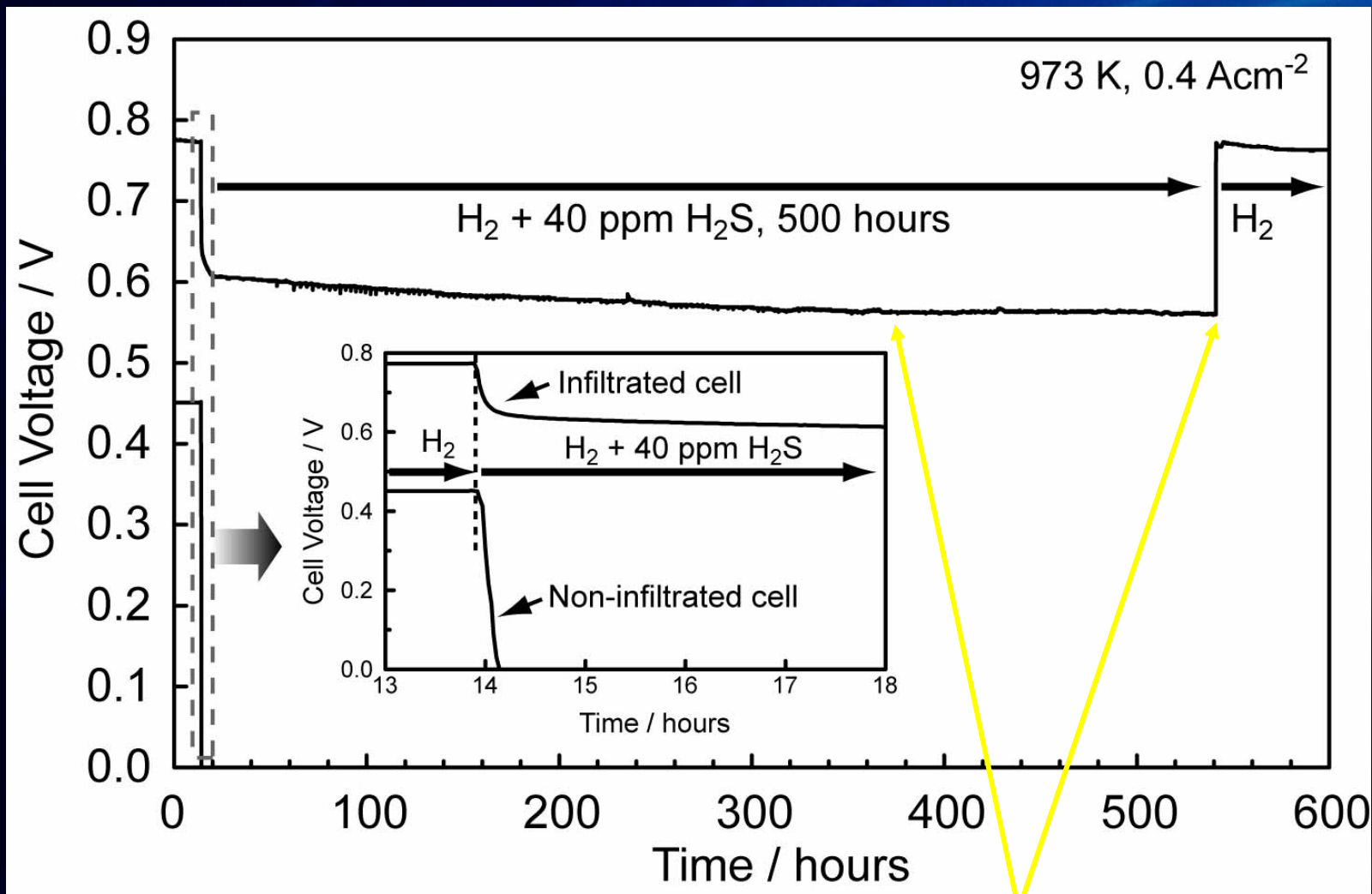


Porous electrolyte matrix

Composite Commercial  
electrodes (YSZ-LSM)

● electronic conductor    ● ionic conductor

# Sulfur Tolerant Ni-YSZ



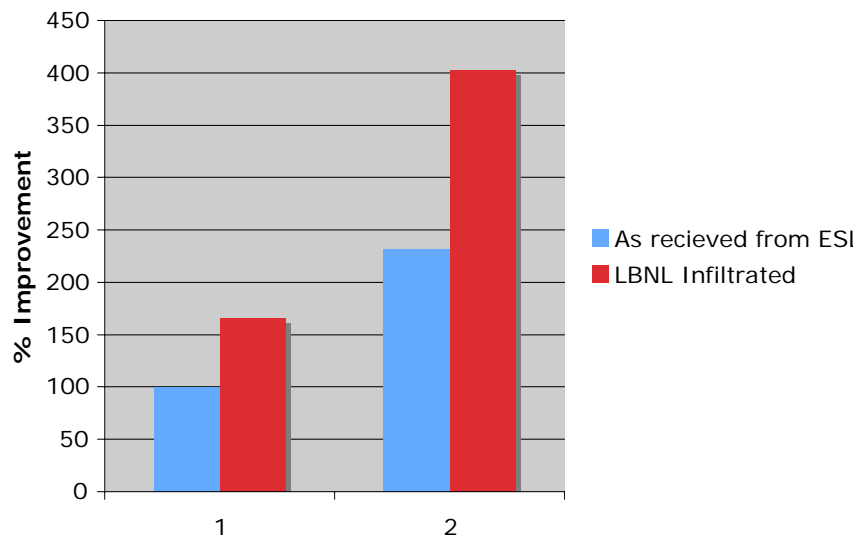
0% degradation over 180 hrs



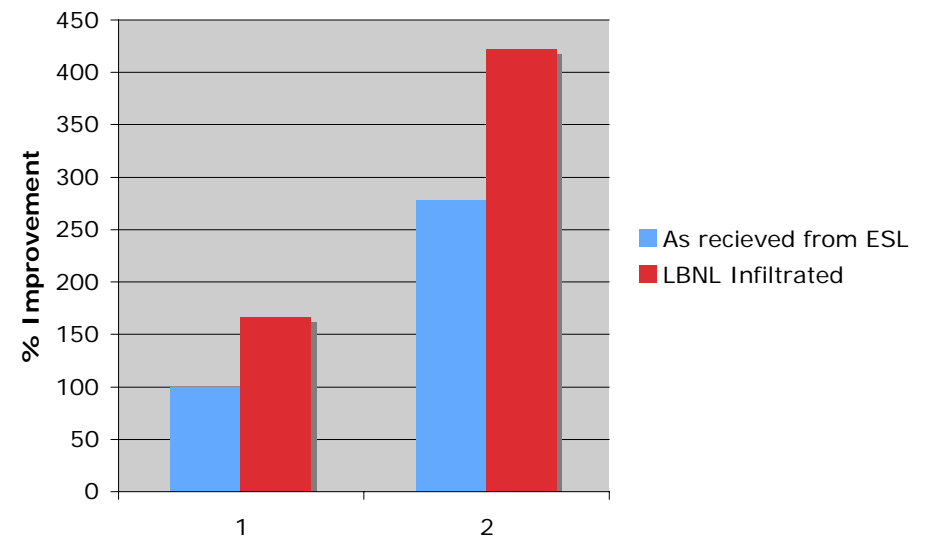


## LBNL Collaboration with Electro Sciences Lab to Improve Performance of ESL SOFC product

Performance @ 0.7'



Performance @ Peak Power



Working on standard cell for **700 C** operation - available to industrial teams, Universities, and National Labs - US supplier

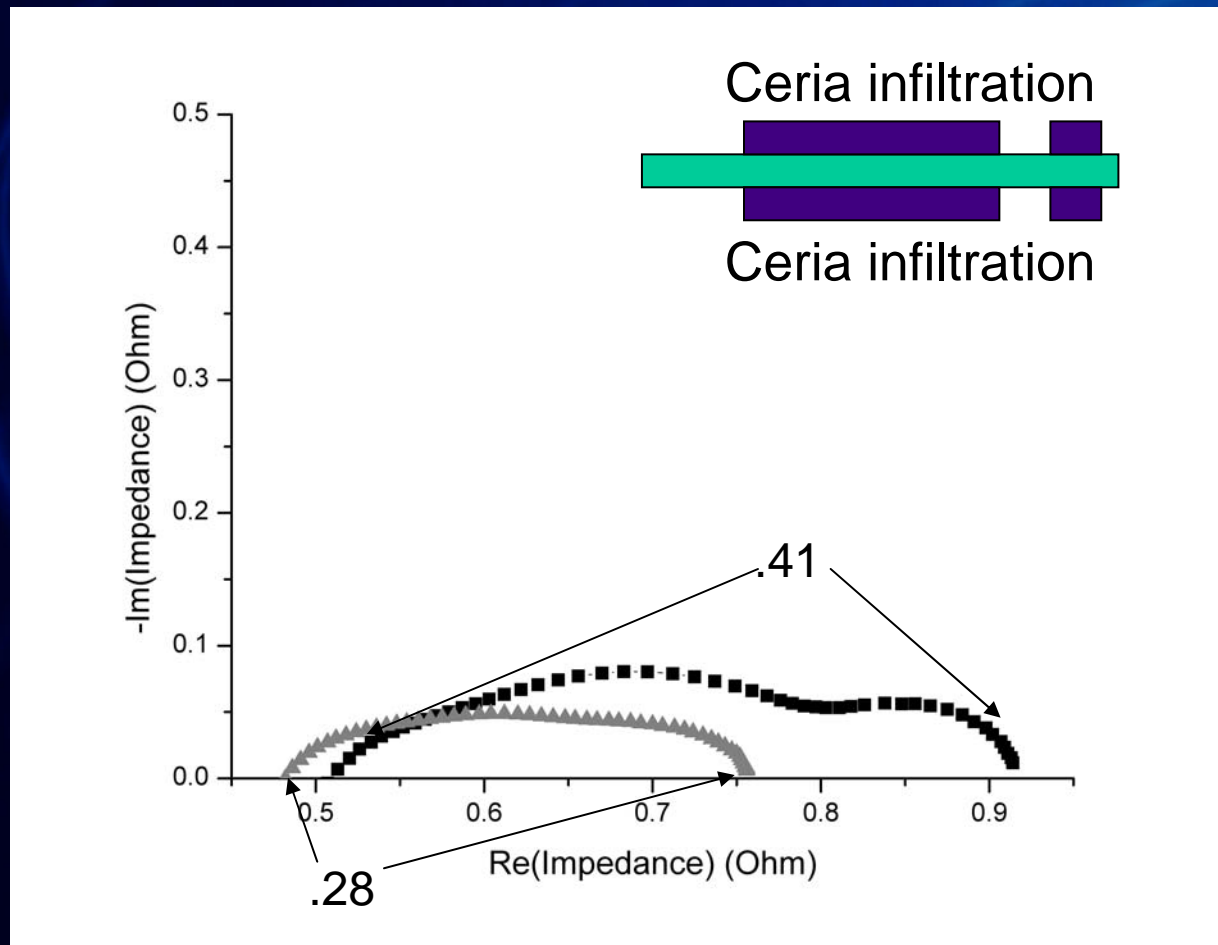


ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# Commercial Symmetric Electrolyte Supported LSCF Cell from INDEC LSCF-YDC/TZ3Y/YDC-LSCF



# HC Starck LSCF/LSCF Cell



=>45% improvement in cell resistance

Electrolyte supported cell: electrode Impedance  
before and after infiltration 700 °C



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY



# Core Technology Program Technology Transfer

- Infiltration workshop
- Transfer technology to companies to U.S. companies and labs
- Guidance to manufacturers of cell stack components (ESL) to enhance U.S. competitiveness



# Infiltration Workshop: February 16th, 2007

- Argonne National Laboratory
- Pacific Northwest National Lab
- Georgia Tech
- Instructional DVD from Workshop available

QuickTime™ and a  
DV/DVCPRO - NTSC decompressor  
are needed to see this picture.



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# LBNL 2-cell Standard Stack Core Effort

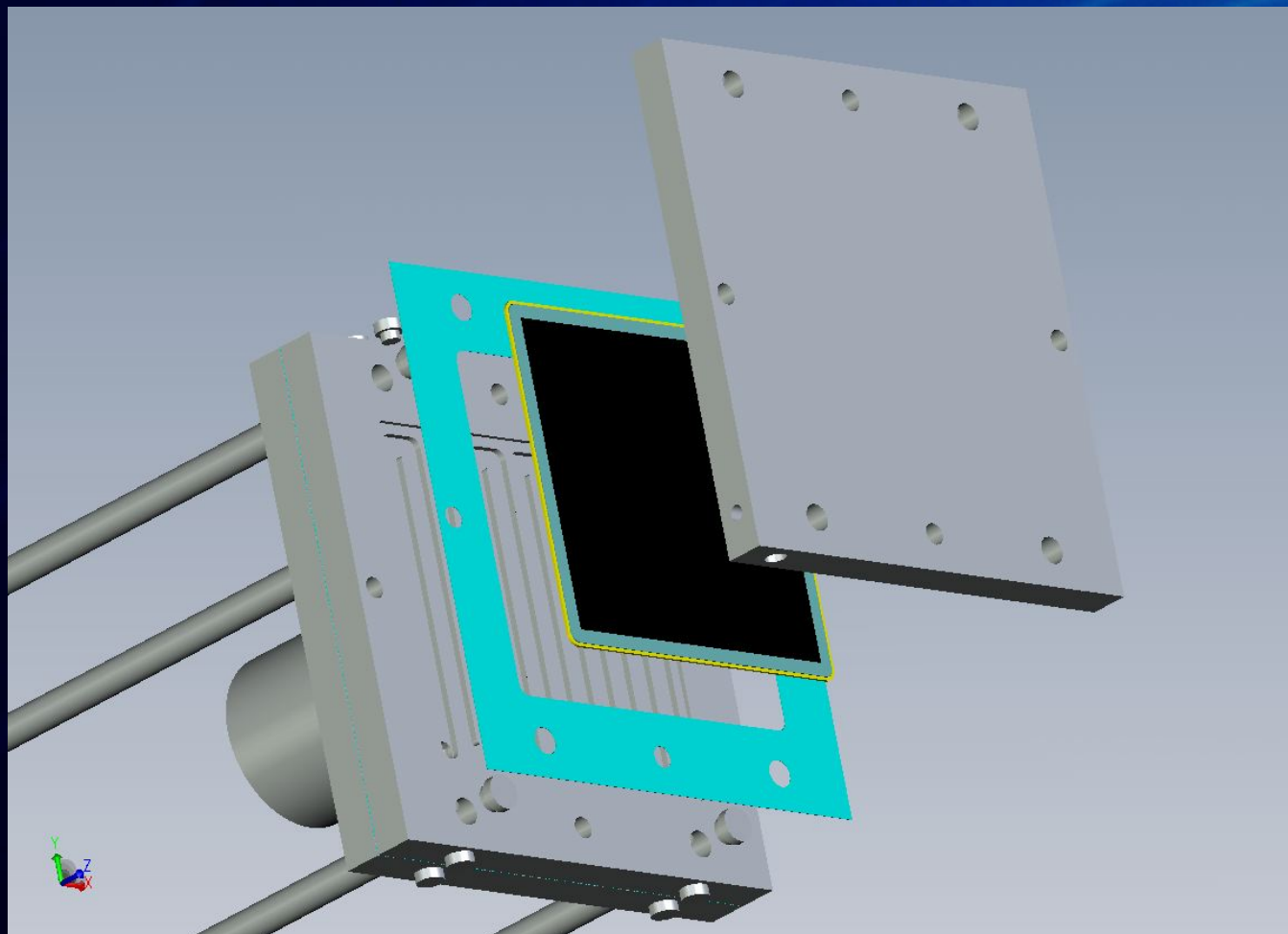


Based on 2.5 cm x 2.5 cm SOFC plates for 2" bore furnace  
Original design by Hideto Kurokawa



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

## Scaled-up Standard Stack: LBNL lead with Lane Wilson & Wayne Surdoval



5 cm x 5 cm SOFC plate design to fit into 3" bore furnace (\$1500)



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY



## Quotation 4418

### McAllister Technical Services

West 280 Prairie Avenue  
Coeur d'Alene, ID 83815

Ph: 208-772-9527

Fax: 208-772-3384

Email: [solutions@mcallister.com](mailto:solutions@mcallister.com)

URL: [www.mcallister.com](http://www.mcallister.com)

Date: 02-Feb-07

This estimate is good for 60 days from the date shown above. Prices quoted are for quantities shown.

To

Steven J. Visco

LBL Materials Sciences Division

1 Cyclotron Road

Berkeley, CA 94720

Ph: 510.486.5821

Fax: 510.486.4881

[sjvisco@lbl.gov](mailto:sjvisco@lbl.gov)

Terms Offered: Net-30

Delivery: 12 Wks. ARO

(based on current workload)

FOB: Factory, Coeur d'Alene, Idaho

Item	Quantity	Description	Model #	Unit Price	Amount U.S.D.
1	1	SOFC Cell Assembly, CNC Program		750.00	\$750.00
2	1	Design	SOFC 5X5	760.00	\$760.00
5	5	Additional Center Plates (if ordered at same time)		\$135.00	\$675.00
Thank you for the opportunity to quote on this project!				Sales Tax	N/A
				Shipping	at cost

Estimate authorized by

Name:

Title: Robert McAllister, President



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY



# McAllister Build of 2-cell 5 x 5 cm SOFC Plate Stack



- Standardized test platform
- Allows testing of electrodes, seals, contact pastes, in a uniform manner
- Allows comparison of results between labs, universities, and industry
- Fits in inexpensive furnaces
- Is not intended as a precursor to commercial device
- ~ \$800/ea. after initial build



# LBNL Work on mSOFCs

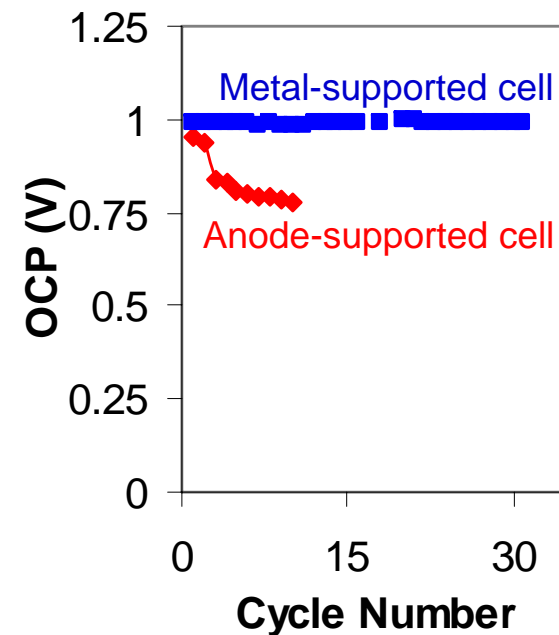
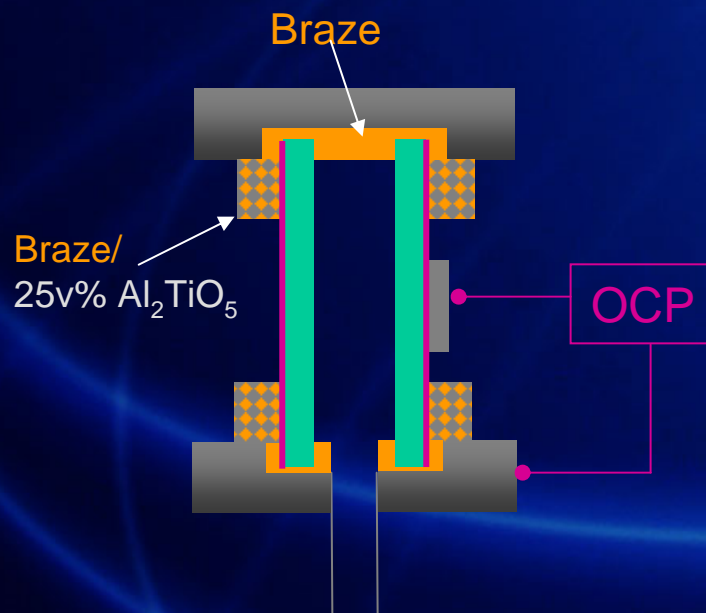
- Build structure from low cost materials
- Obtain performance similar to anode supported cells
- Show long term stability (rapid progress)
- Work with cell manufacturers (licensing & sponsored research)



# Rapid Thermal Cycling – Braze-Sealed Cell

150-735°C, ~500°C/min

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



Anode supported tubular cell cannot tolerate rapid thermal cycling  
Cell failed, joint did not

Metal-supported cell/brazed joint is robust to thermal cycling  
Unexpected shutdowns, redox cycles



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# 650-700°C Performance

Moist hydrogen fuel, air

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

650°C

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

700°C

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# Infiltrated Electrodes Support High Power Density

Moist hydrogen fuel, pure oxygen (removes gas transport limitation)

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

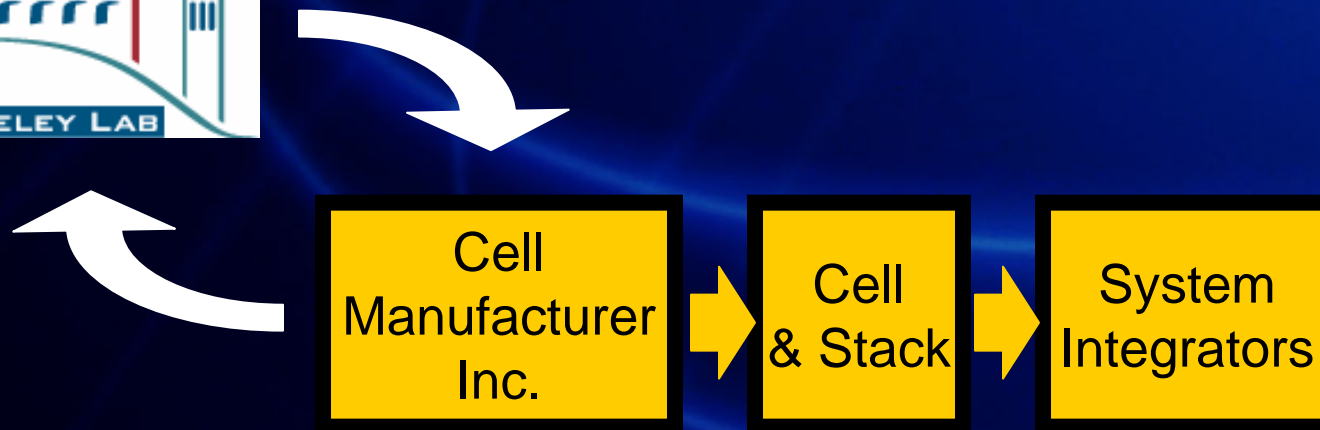
Temperature	Max Power (mW/cm <sup>2</sup> )	Power at 0.7V (mW/cm <sup>2</sup> )
650°C	982	726
700°C	>1300	993
750°C	>1300	>1300



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY



# Work with manufacturer to ensure manufacturability as continue cell development



- Military
- APU
- Residential
- CHP
- Portable
- RV/Marine
- Aerospace
- Etc.



# High Volume Porous Metal Media

Coal: kW to MW?



QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



ERNEST ORLANDO LAWRENCE  
BERKELEY NATIONAL LABORATORY

# Transitioning Technology to Private Sector

- LBNL is in discussions with cell/stack manufacturers for licensing infiltration and mSOFC technology for both planar and tubular configurations
- Wide range of IP being negotiated for SOFC, an coating for filtration (including spin-off applications for for coal gasification)
- Commercial interest in infiltration and mSOFC technology is rising quickly



# Future Work

- Continued focus on infiltration technology as a means of improving cathode (and anode) performance at reduced cell temperatures
- Emphasis on baseline degradation studies on commercial cells as a metric of infiltration performance over time
- Continuing activities in technology transfer
- New stuff





# Acknowledgements

---

This work was supported by the U.S. Department of Energy under Contract No. DE-AC02-05CH11231.

Thanks to Lane Wilson and Wayne Surdoval for their input to the LBNL program

Good luck to Lane at BES

