## Interconnect Alloys Metallurgy and Manufacturing

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### Introduction

- Environmental exposure conditions can lead to oxidation/functional degradation of interconnects and adjacent components
- Alloy and process development to reduce contribution to SOFC performance degradation



### Overview

- Phase I results and timeline
- Current focus of Phase II
  - Silicon removal trials
  - Oxidation of commercially available ferritic stainless steels
  - Production of novel ferritic stainless steels
- Ongoing work



## Phase I Review



### Results to Date - Phase I

- Timeline
  - 12 month period
  - Calendar year 2006
- Proof of concept for solid-state silicon removal
- Melting, processing, and testing of concept alloys
- Testing/analysis of commercially available stainless steels
- Testing/analysis of various oxidation-resistant coatings on commercially available stainless steels

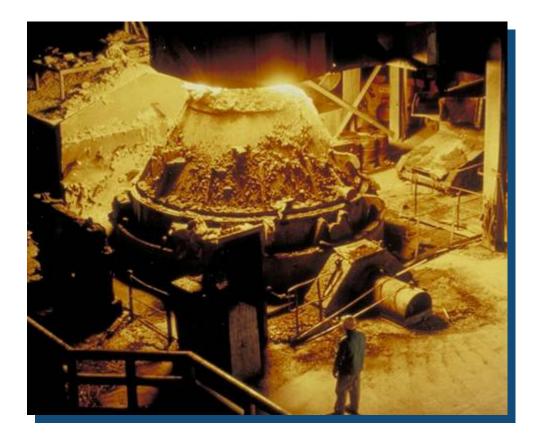


## Phase II Status



### Silicon in Stainless Steels

- Silicon is present in most readily available stainless steels
- By-product of the AOD steelmaking process
- Commercially available stainless steels generally contain approximately 0.5 wt. % silicon



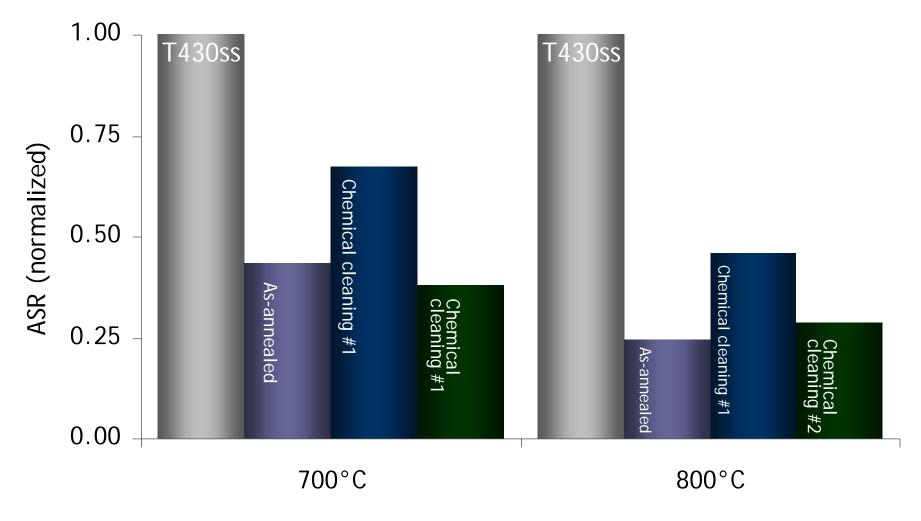


### Silicon Removal Trials

- High-temperature pre-treatment with optional chemical component
- Tested using a variety of Fe-Cr stainless steels (T430, T439, T441HP™ alloys)
- Formation/removal of an SiO<sub>2</sub> surface film



### Effect on ASR - Prior Work





ASR measurement temperature (°C)

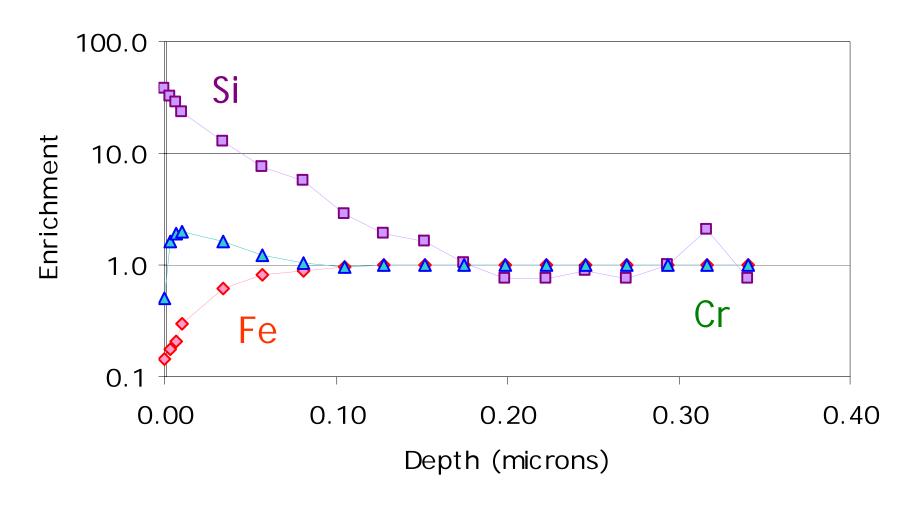
samples pre-oxidized in air for 500 hours at 800°C

### Silicon Removal

- Production of larger test panels
  - Third party testing and analysis
  - Internal evaluation
- Characterization of treated surface
- Oxidation testing
- Long-term electrical evaluation



#### Post-Treatment Characterization





T430 post-treatment samples AES analysis with sputter depth-profiling

- Thermally activated process
- More effective for thinner samples
  - Absolute quantity of Si removed is a function of temperature, surface area
  - Amount of silicon available for removal (reservoir) is a function of substrate thickness
- Evaluation
  - Rolling trial (0.08-0.15 mm thick T430 samples)
  - Calculations



starting with an "infinite sheet"



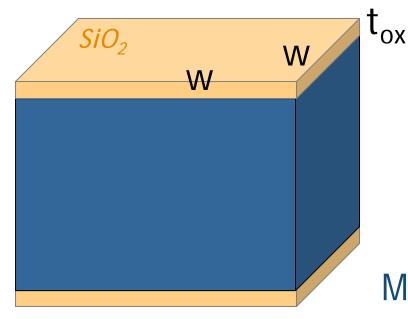
Establish volume element

Set surface area = 1

$$2w^2 = 1$$



$$V_{M} = W^{2}X_{M} = 0.5X = M_{M}/\rho_{M}$$

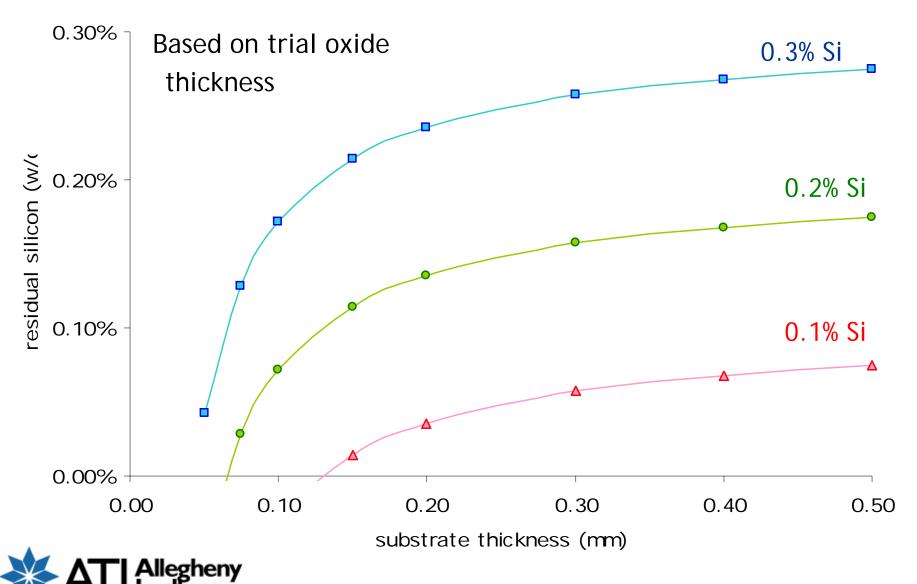


Allow the sample to oxidize on both sides

$$V_{ox} = 2w^2t_{ox} = M_{ox}/\rho_{ox}$$

 $M_{ox}$  allows for the determination of Si consumption / residual  $M_{Si}$  in the volume element after the formation of an oxide layer of thickness  $t_{ox}$ 





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### Oxidation Testing

- Oxidation testing carried out in a variety of environments
  - Ambient air
  - Air + 10% water vapor
  - 4%H<sub>2</sub> + 10% water vapor in argon carrier
- Testing using duplicate specimens at 800°C
  - T430, T430 De-Si
  - T439, T439 De-Si
  - T441HP™ alloy, T441HP De-Si
  - E-BRITE® alloy



# **Alloy Compositions**

Element	T430 \$43000	T439 alloy \$43035	T441HP™ alloy S44100	E-BRITE® alloy S44627
С	0.015	0.015	0.015	0.001
Mn	0.4	0.4	0.3	0.05
Si	0.4	0.7	0.5	0.2
Cr	16.5	17.5	17.5	26.0
Al	0.05	0.05	0.05	0.05
Мо	-	-	-	1.0
Cb	-	-	0.46	0.15
Ti	-	0.4	0.2	-
N	0.015	0.015	0.015	0.005



## Oxidation Testing

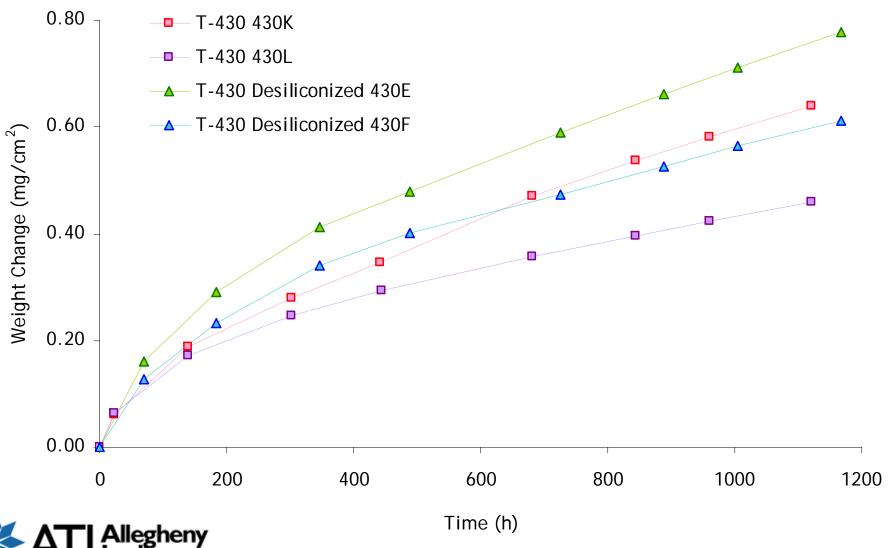
- Type 430 exhibited good oxidation resistance in ambient air but one sample (desiliconized) exhibited breakaway oxidation in humidified air.
- Type 439 exhibited a consistent tendency for spallation in the as-received condition. Spallation did not occur in desiliconized Type 439.
- T441HP™ alloy exhibited the best general resistance to oxidation.



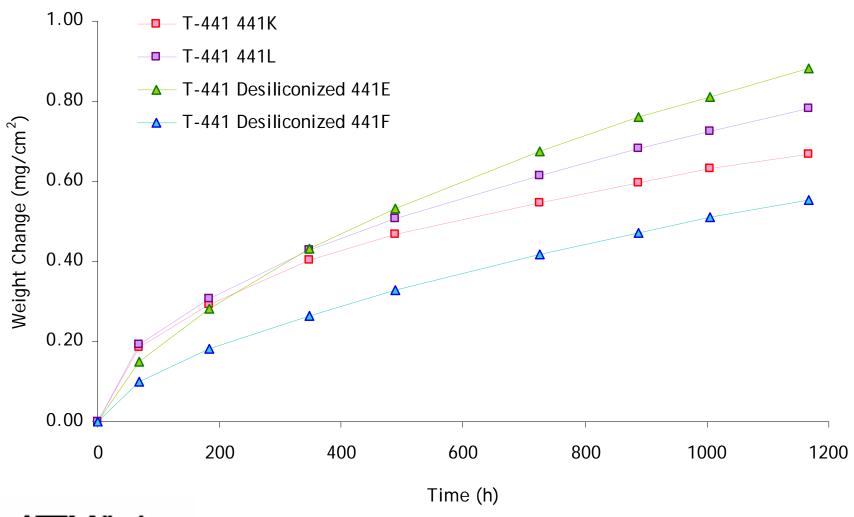
## Oxidation Testing

- E-BRITE® alloy exhibited a tendency towards weight loss in humidified air, which is consistent with past work (this alloy was not tested in ambient air).
- The exposure to simulated anode gas (Ar-H<sub>2</sub>-H<sub>2</sub>O) was the least aggressive of all of the test exposures. De-siliconization was uniformly beneficial in the SAG composition tested.

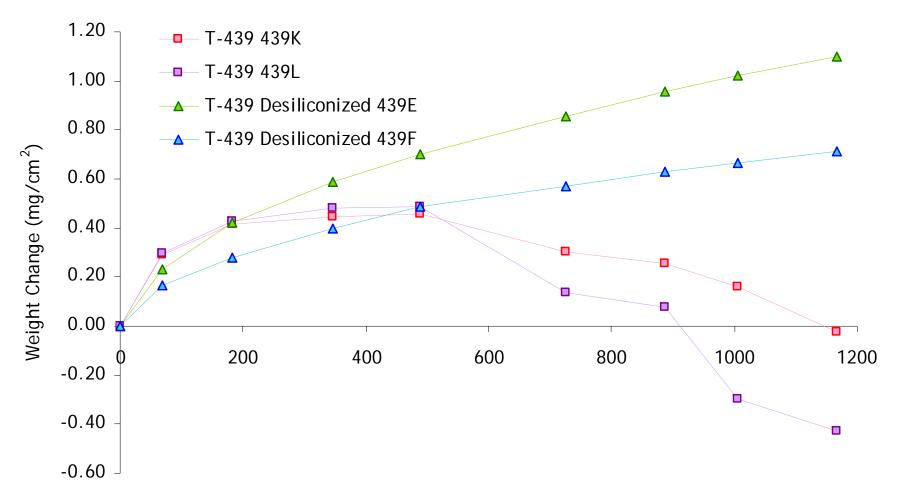






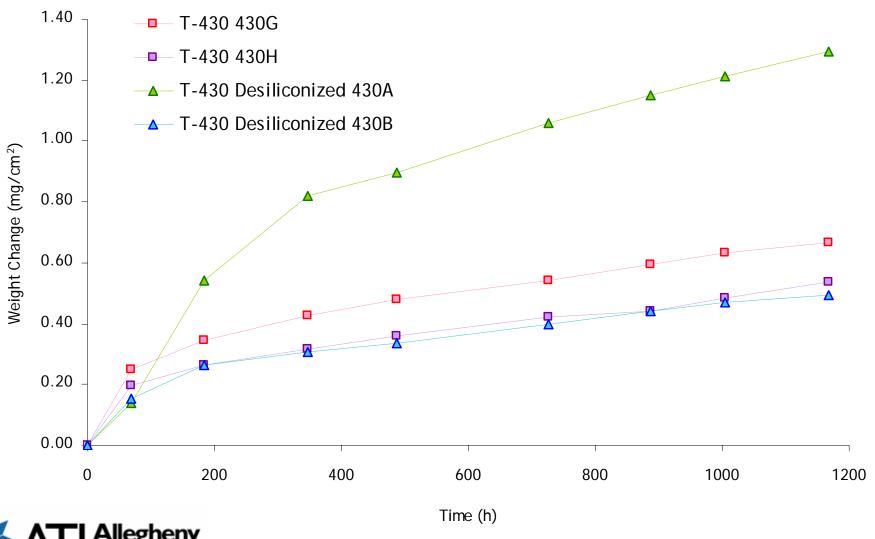




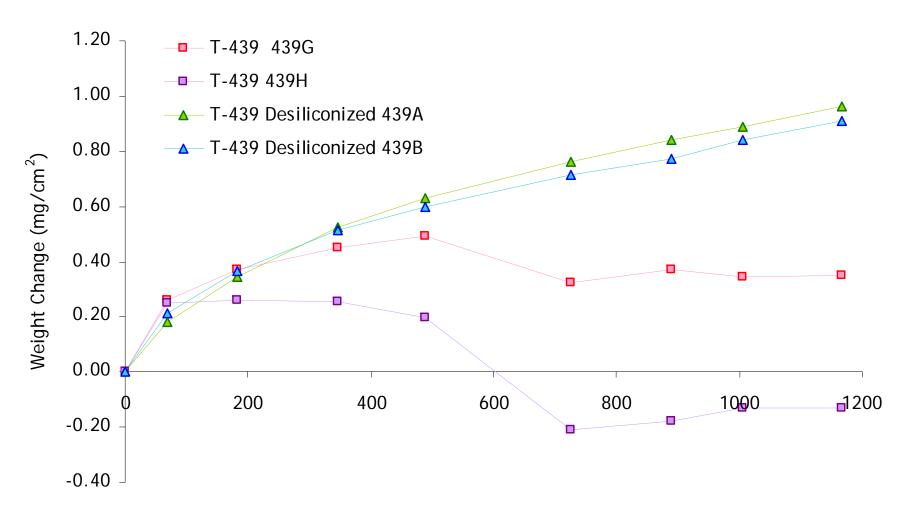




Time (h)

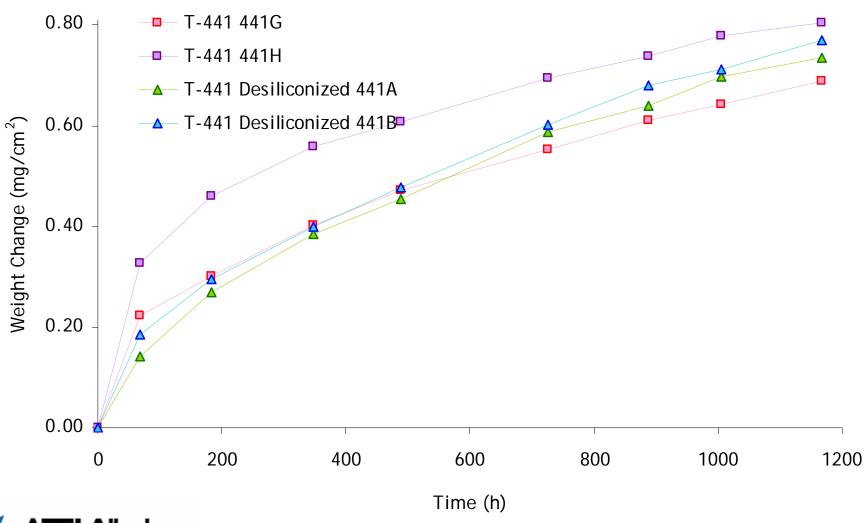




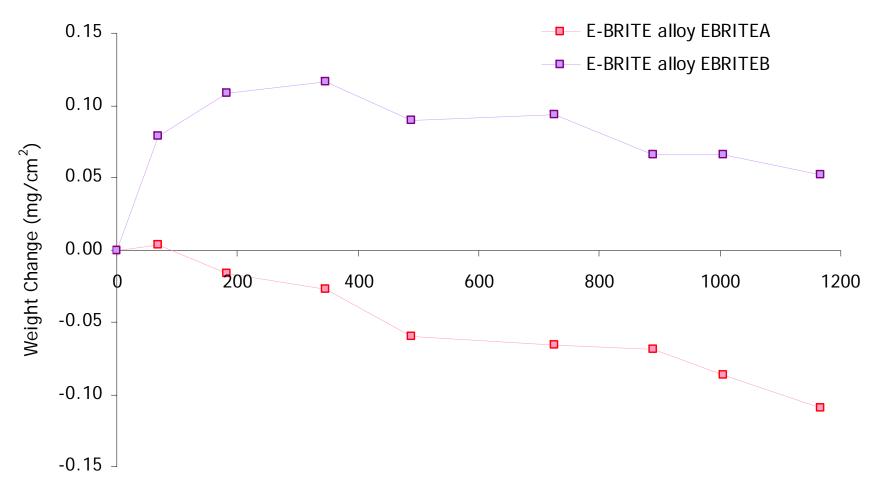




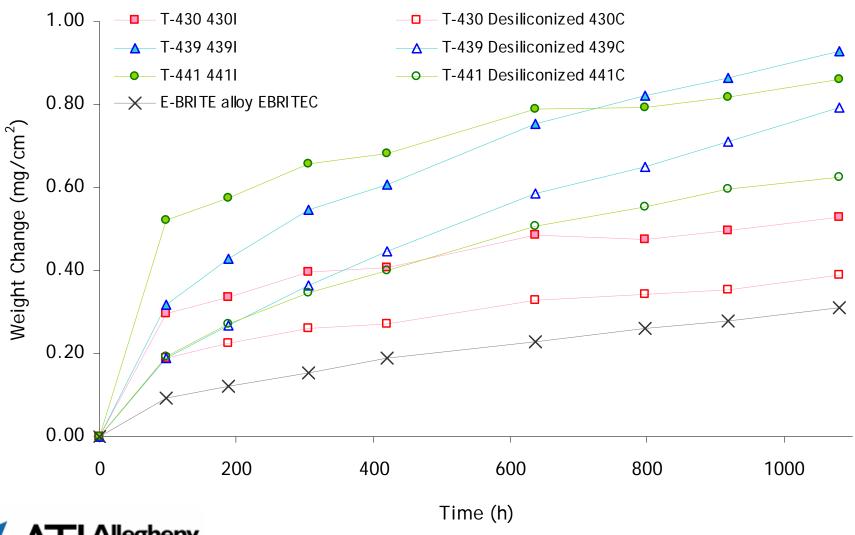
Time (h)







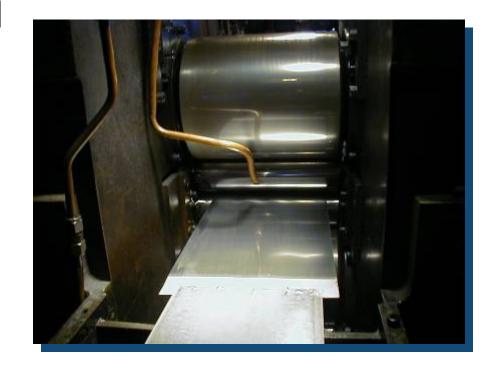






### Novel Ferritic Stainless Steels

- Currently processing a set of lab-melted VIM heats
- Optimized compositions
  - High- and low-chromium variants
  - Minor element control
  - Advanced processing





### Summary - Alloy Characterization

- Oxidation screening testing is complete
- Long-term oxidation testing ongoing
- ASR evaluation of a matrix of alloys has been initiated



### Summary - Current Development

- Compositional matrix melted and processed to sheet
- Silicon removal trials ongoing



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