

# Interconnects

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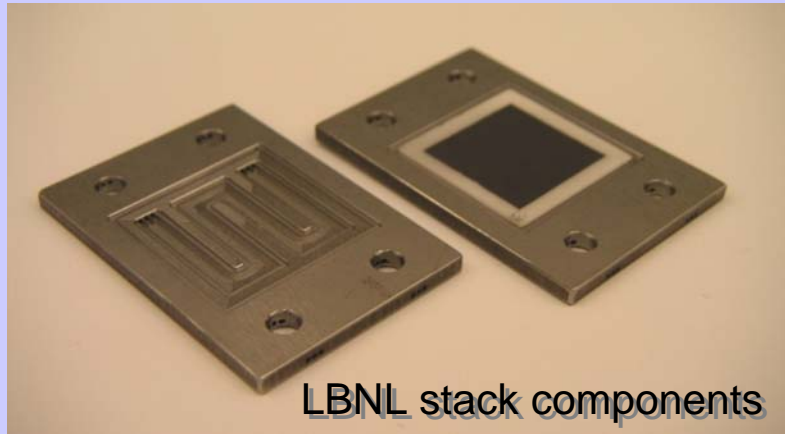
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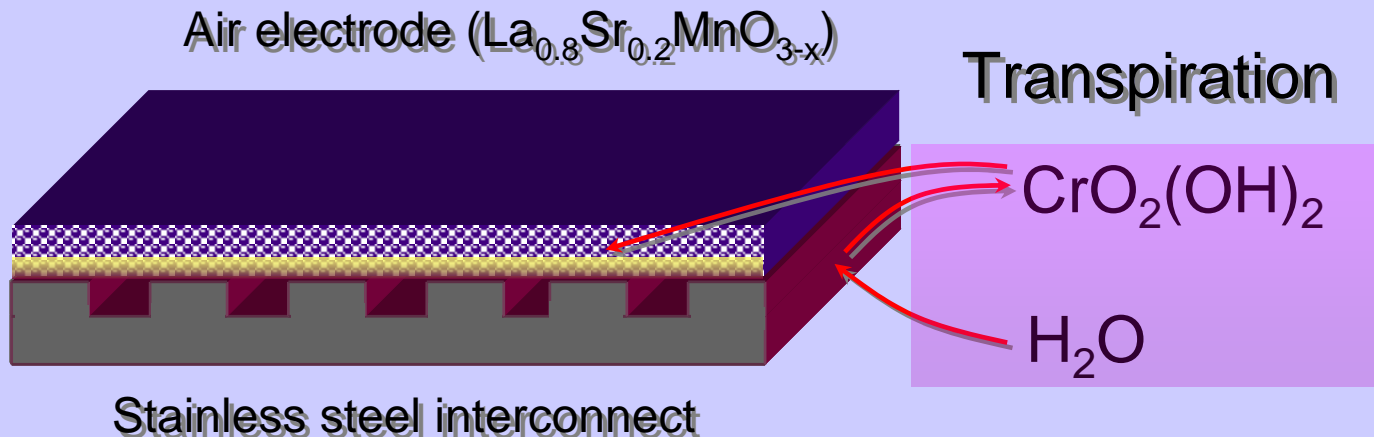


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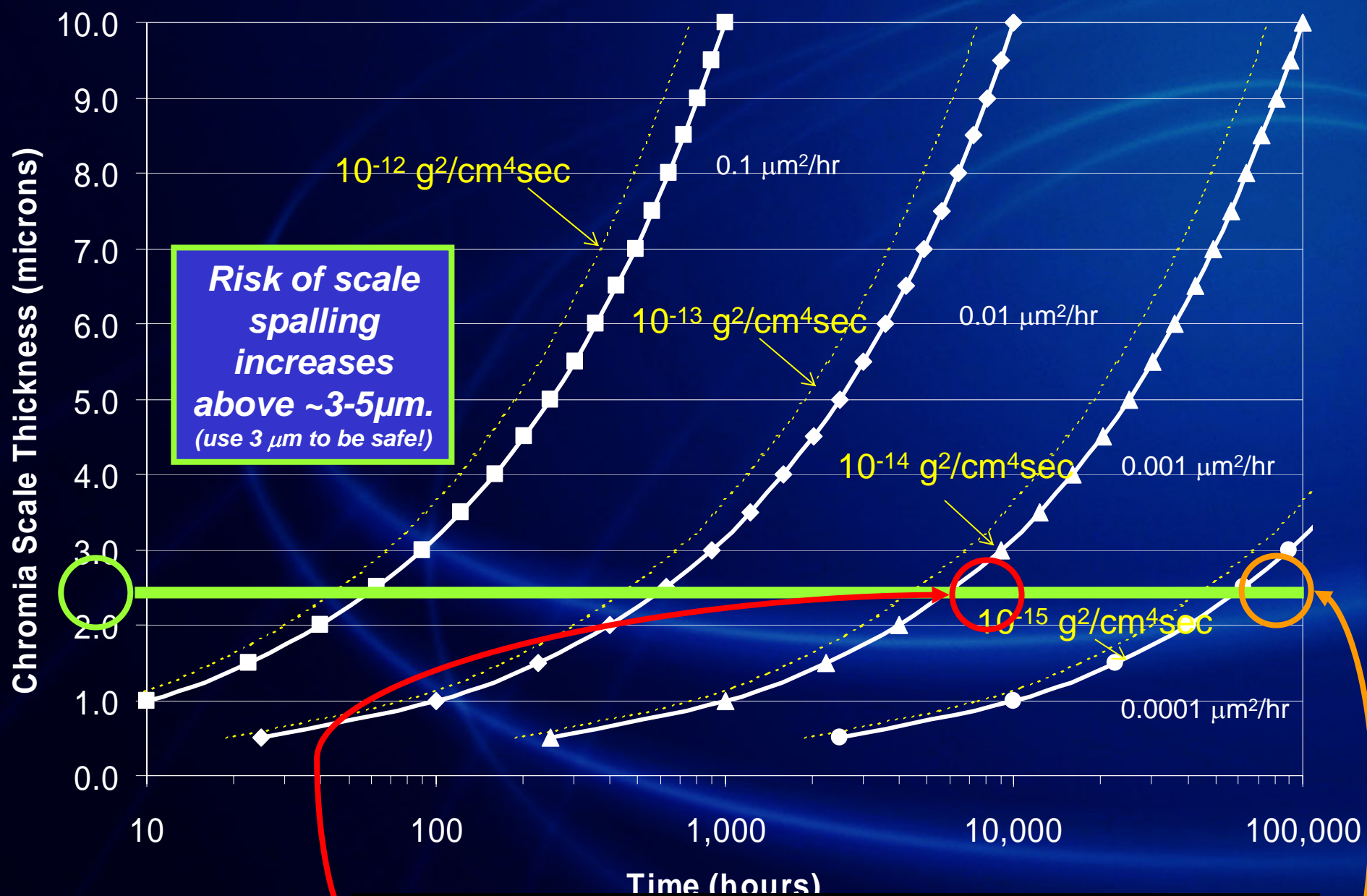
# Metal Stability & Interactions



- Oxidation behavior
- Oxide spallation
- Area specific resistance
- Chromium migration



- Vapor chromium transport
- Bulk & grain boundary Cr transport
- Surface migration



**$\sim 10^{-14} \text{ g}^2/\text{cm}^4\text{sec}$  for transportation 5,000 – 10,000 hrs**

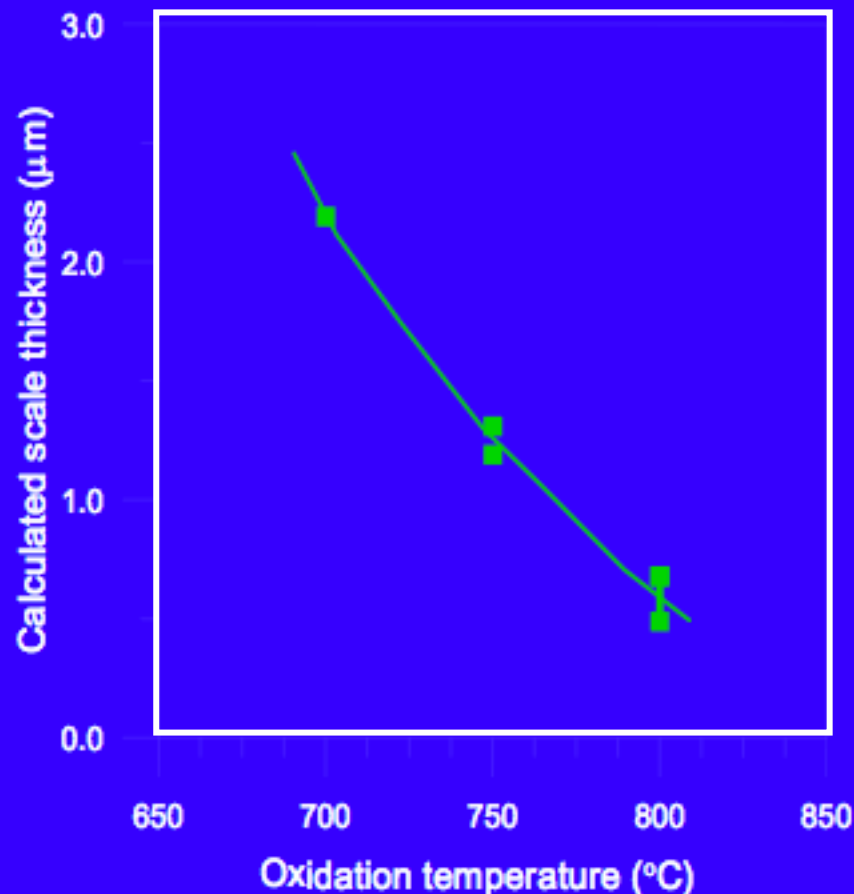
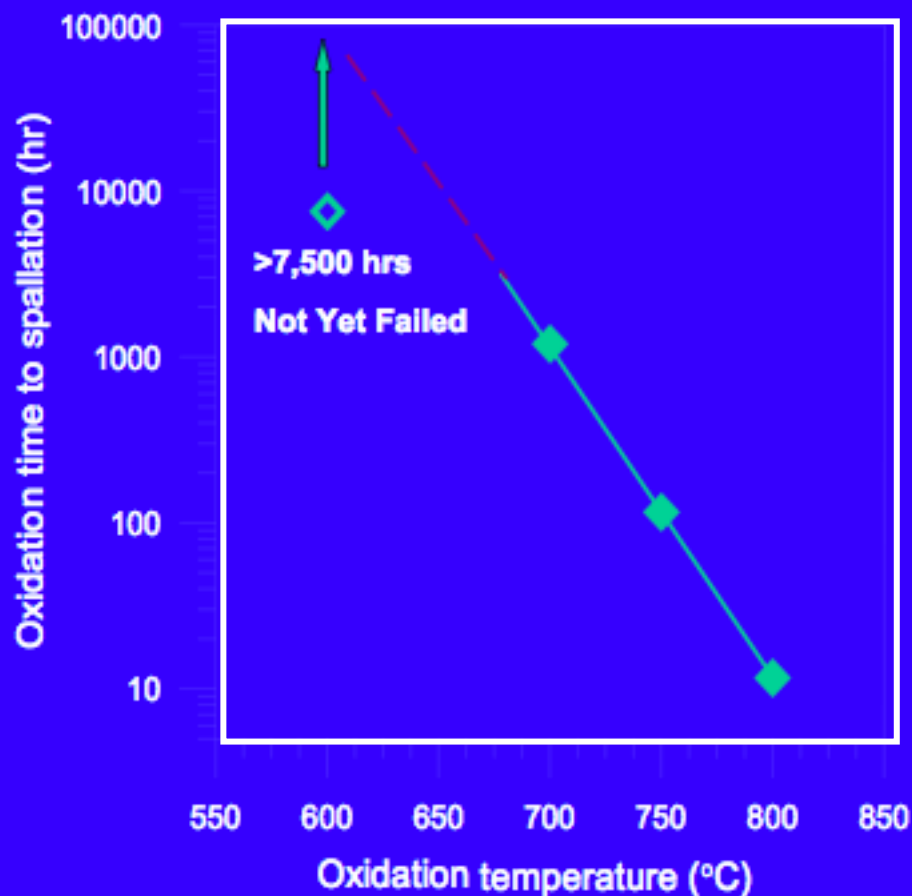
**$\sim 10^{-15} \text{ g}^2/\text{cm}^4\text{sec}$  for stationary 50,000 – 100,000 hrs**



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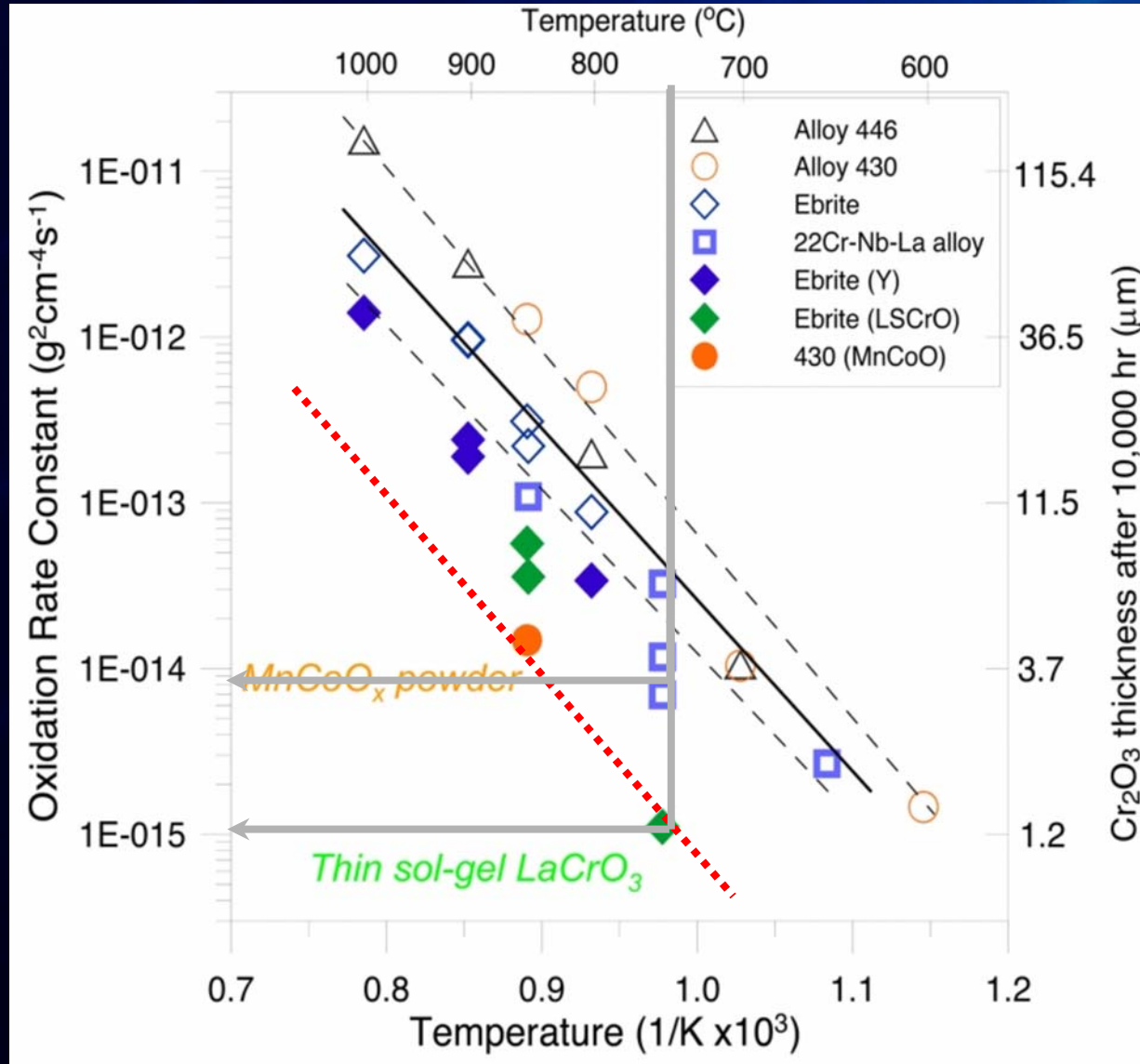
# Condition for minimum spallation of scales on 430ss after isothermal oxidation and fast cooling to RT

These are small and sporadic (1%?)



*Scale thickness decrease because of higher thermal stresses and/or more defect formation at high oxidation temperatures*

# High Temperature Oxidation of Metal Components

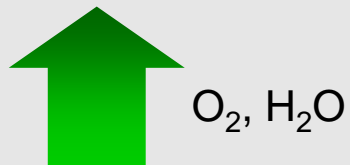




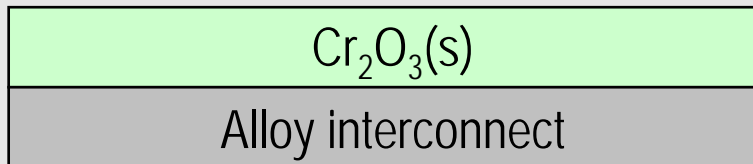
# Chromium Vaporization

- Cr gas species

Cr	CrOH	CrOOH
CrO	Cr(OH) <sub>2</sub>	CrO(OH) <sub>2</sub>
CrO <sub>2</sub>	Cr(OH) <sub>3</sub>	CrO(OH) <sub>3</sub>
CrO <sub>3</sub>	Cr(OH) <sub>4</sub>	CrO(OH) <sub>4</sub>
	Cr(OH) <sub>5</sub>	CrO <sub>2</sub> (OH)
	Cr(OH) <sub>6</sub>	CrO <sub>2</sub> (OH) <sub>2</sub>



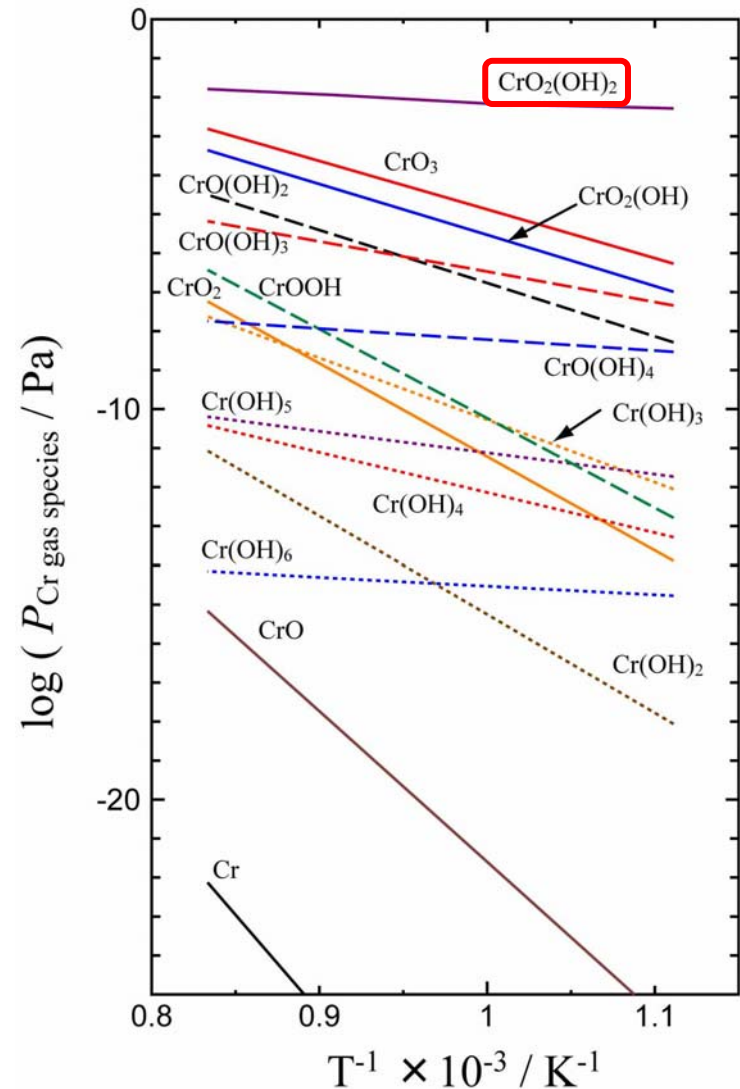
O<sub>2</sub>, H<sub>2</sub>O



B.B. Ebbinghaus, *Combust. Flame.*, **93** (1993) 119

- Partial pressures of Cr gas species

Cathode side:  $P_{O_2} = 2 \times 10^4$  Pa,  $P_{H_2O} = 2 \times 10^3$  Pa

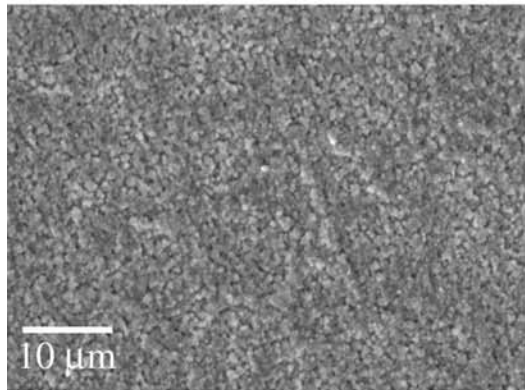


# Results – uncoated and coated 430 -SS

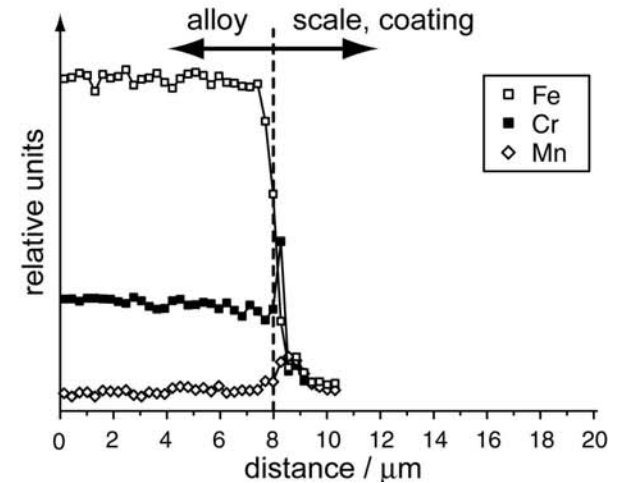
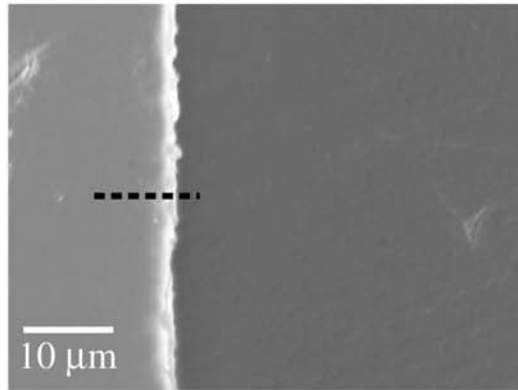
- 1073 K, 86.4 ks (24 hrs),  $P_{\text{H}_2\text{O}} = 1.0 \times 10^4 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3 \text{ s}^{-1}$  (200ml/s)

430SS

(a) surface

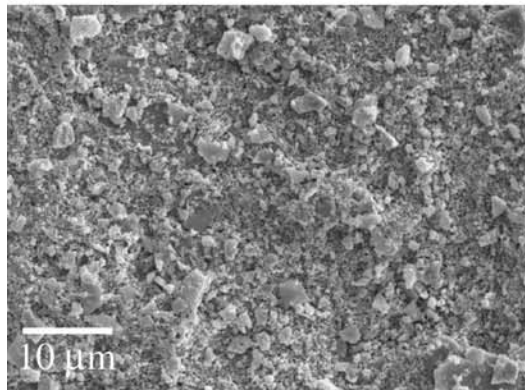


(b) cross section

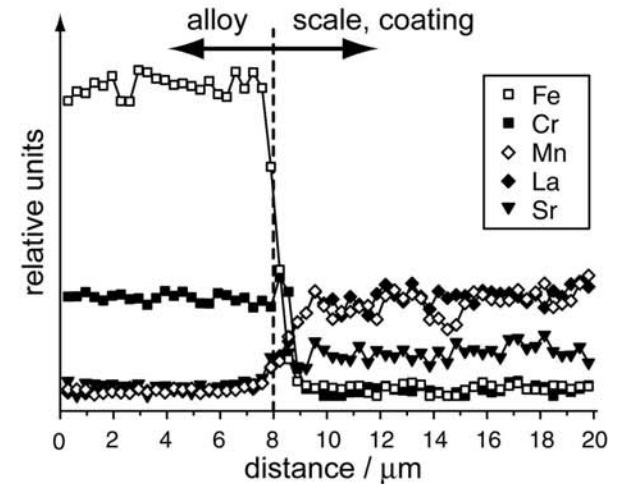
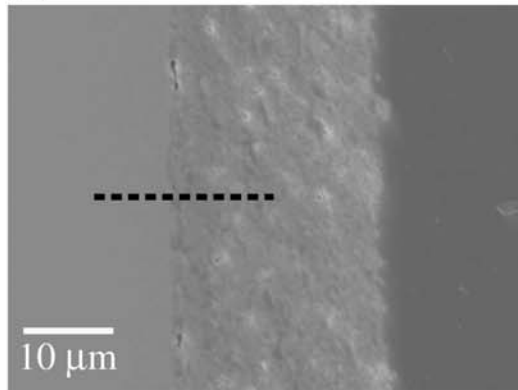


430SS + LSM (spray)

(a) surface



(b) cross section



# Experimental

## Sample preparation

Pre-oxidation (for uncoated and coated 430SS)

1073 K, 172.8 ks, in Air

Cr vaporization test

873~1073 K (main test )

873~1273 K (preliminary test with  $\text{Cr}_2\text{O}_3$  )

86.4~259.2 ks (1~3 days) in Air + 10%  $\text{H}_2\text{O}$

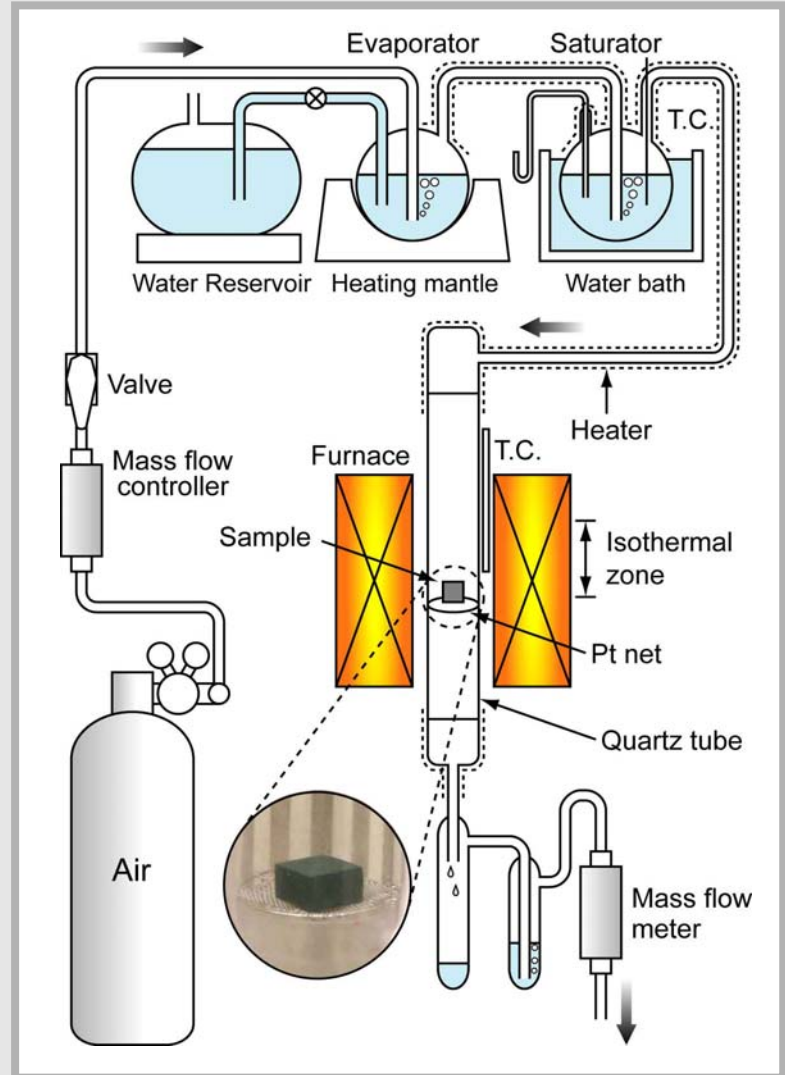
## Analysis method

ICPMS (Inductively coupled plasma mass spectrometry)

SEM (Secondary electron spectroscopy)

EDS (Energy dispersive spectroscopy)

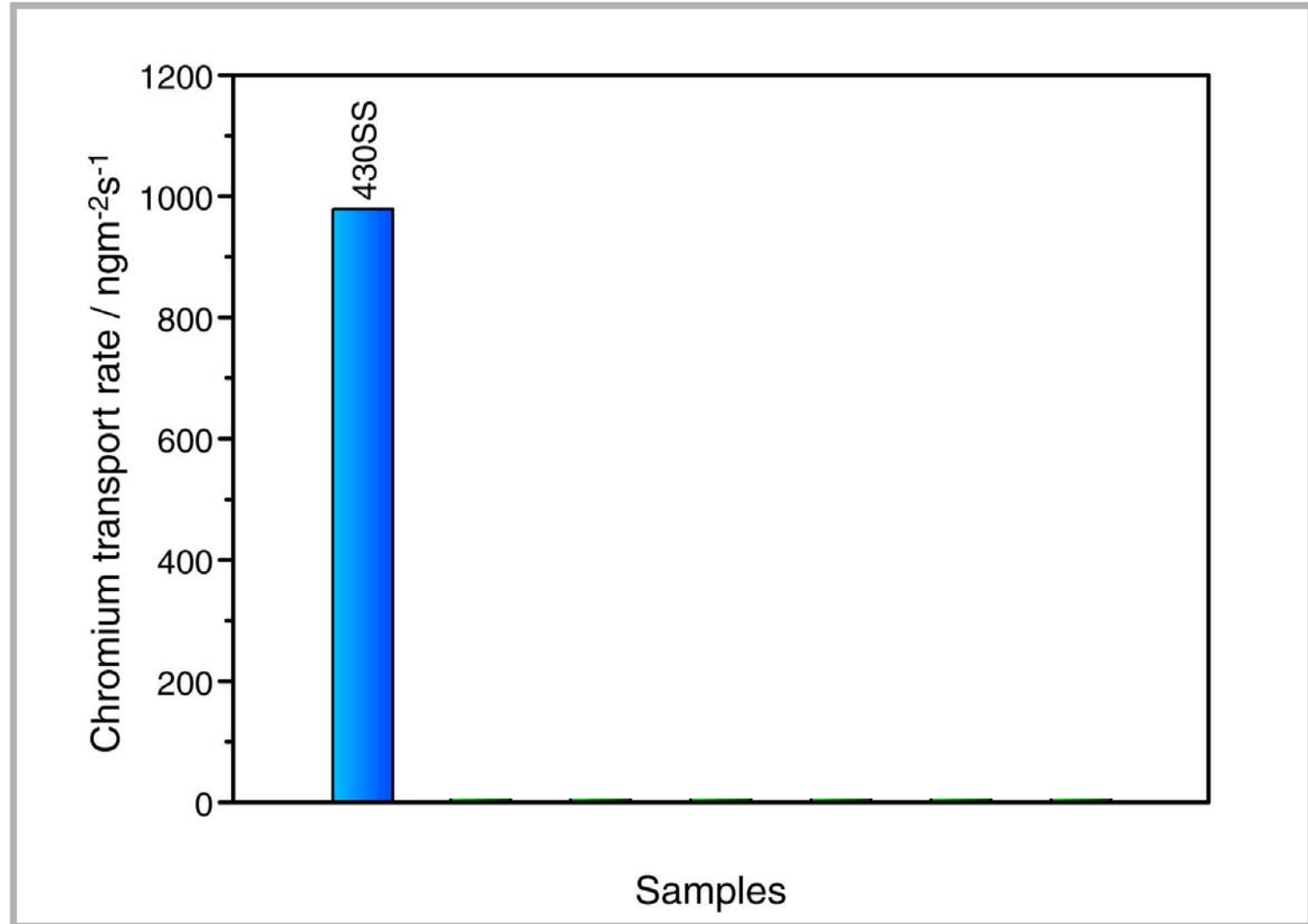
## Apparatus





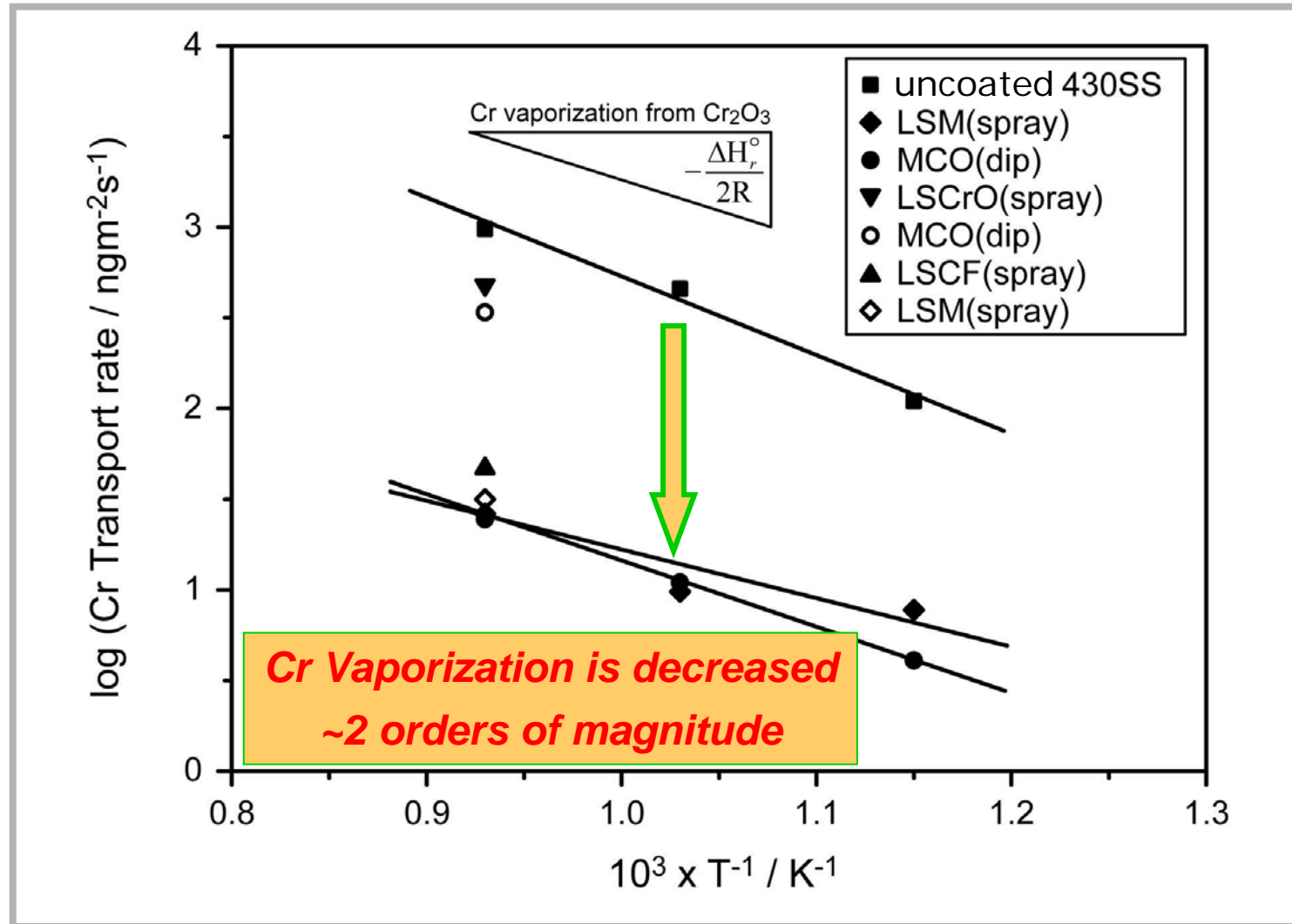
# Results – uncoated and coated 430 -SS

● 800°C, 86.4 ks (24 hrs),  $P_{\text{H}_2\text{O}} = 1.0 \times 10^4 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3 \text{s}^{-1}$  (200ml/min)



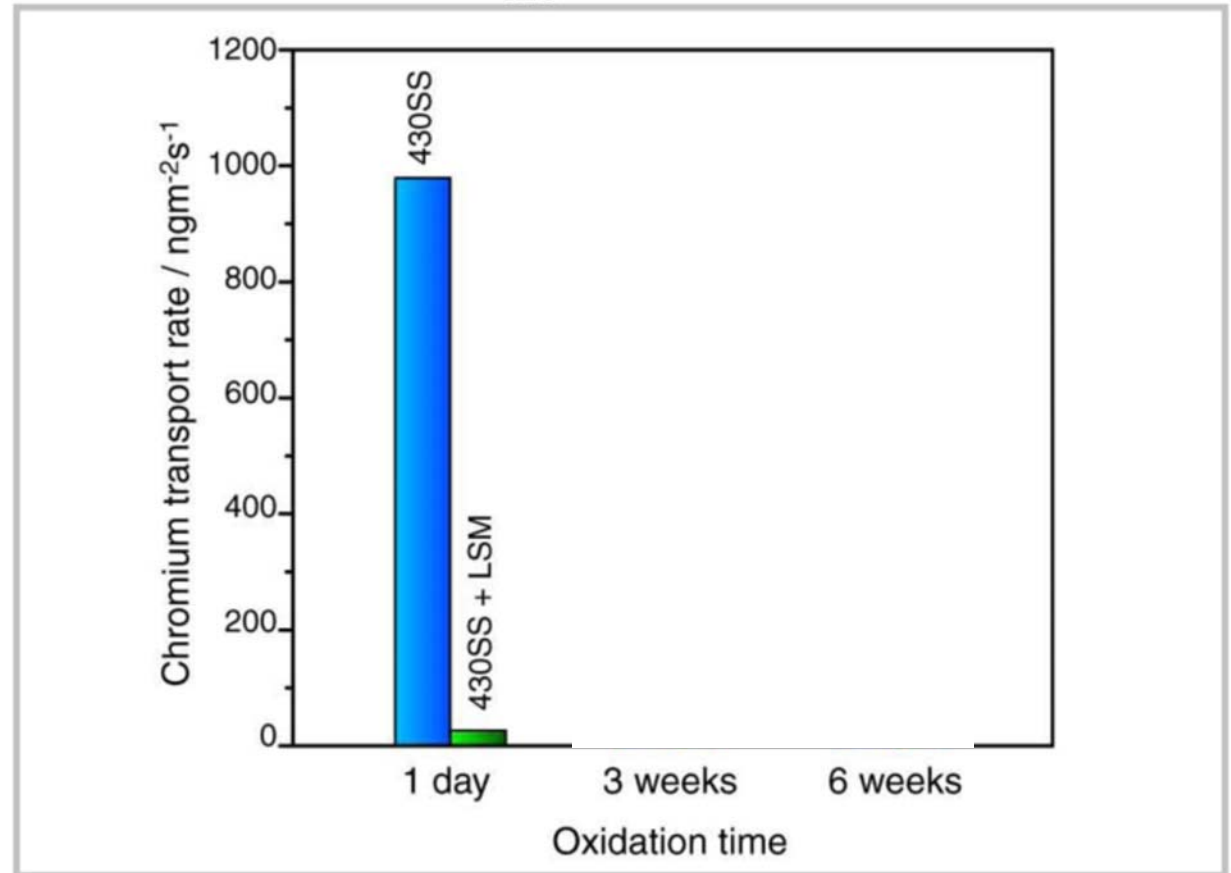
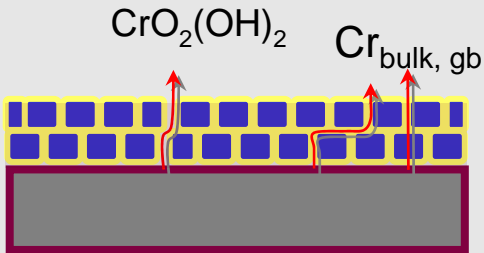
# Results – uncoated and coated 430 -SS

●  $P_{\text{H}_2\text{O}} = 1.0 \times 10^4 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3 \text{s}^{-1}$  (200ml/s)



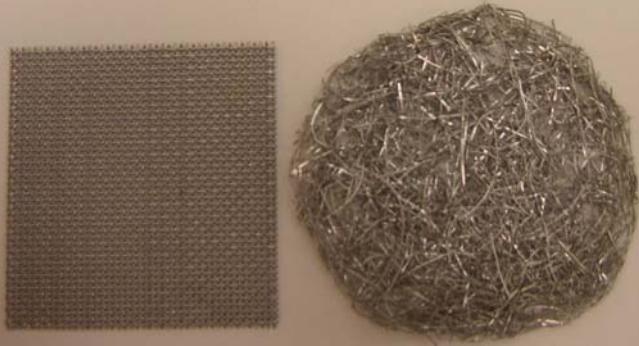
# Long-term Stability of Coatings for Preventing Cr Loss

- Oxidation: 1073 K,  $P_{\text{H}_2\text{O}} = 2.0 \times 10^3 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3\text{s}^{-1}$  (200ml/min)
- Cr test: 1073 K, 86.4 ks (24 hrs),  $P_{\text{H}_2\text{O}} = 1.0 \times 10^4 \text{ Pa}$ ,  $3.33 \times 10^{-6} \text{ m}^3\text{s}^{-1}$  (200ml/min)

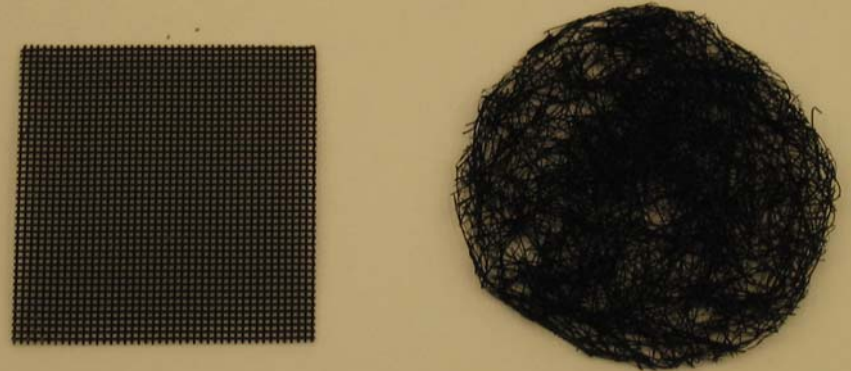


# Protected LSM Spray Coat on 434 Stainless Steel Mesh and Wool (sintered at 800°C for 1hr)

Before Treatment

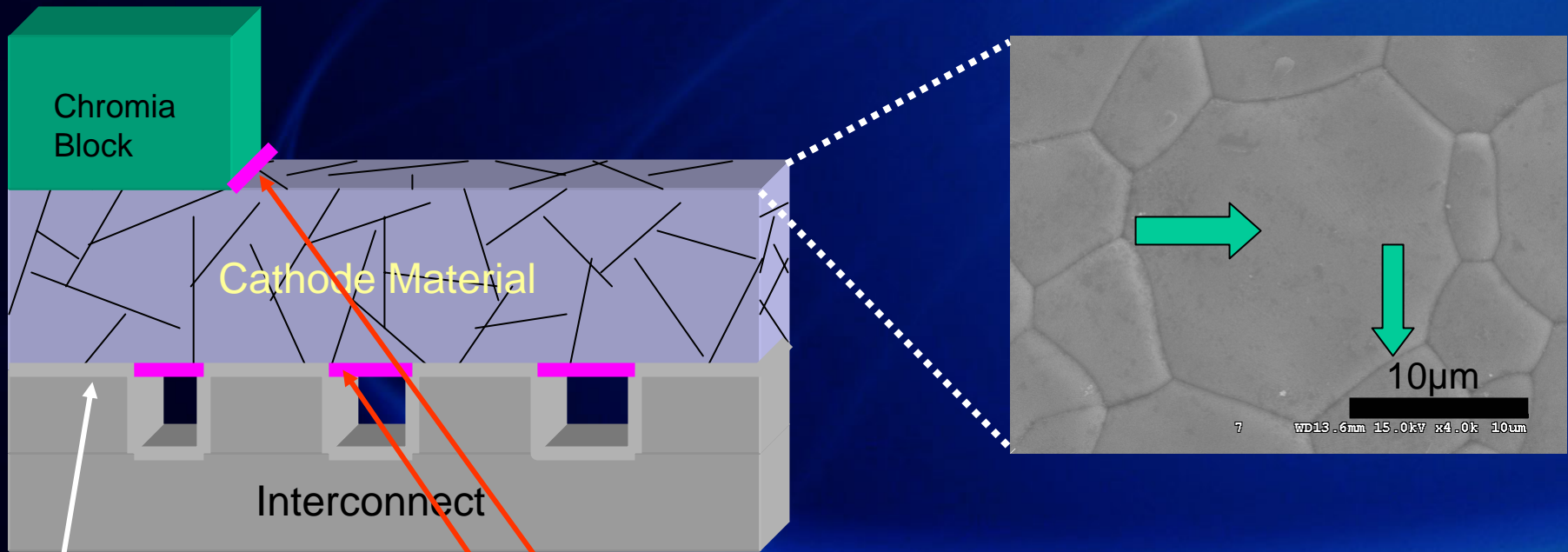


After Spray and Sinter



Oxidation resistant, compliant current connects

# Solid State Chromium Transport on Cathode Materials

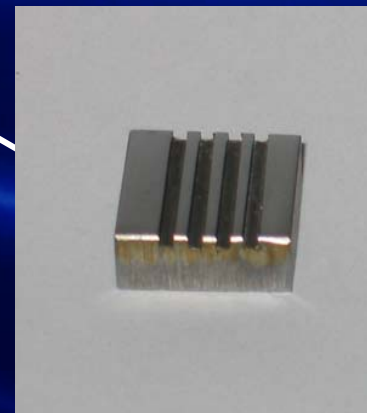
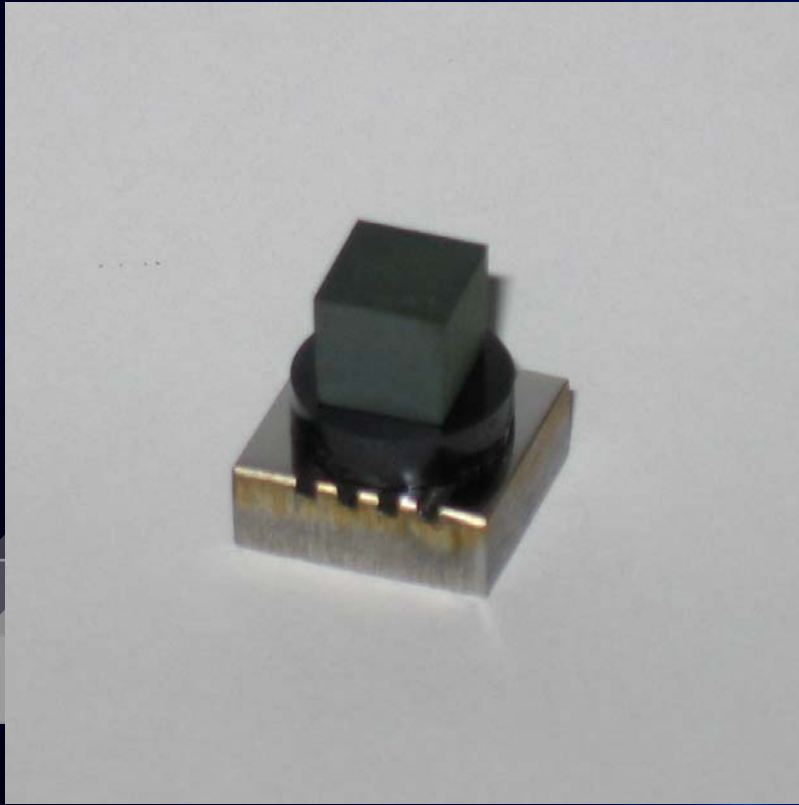


Bulk or grain boundary diffusion at the edge of chromia block and interconnect.

Suppression of Cr Transport by coating with Ag? Pt? Sacrificial Material?



# Solid State Diffusion Transport of Chromium on Cathode Materials Experimental Setup



Sinter in the furnace at 700°C, 750°C, 800°C respectively for 500 hr.

# Benefits of Coatings

- Decrease oxidation rate  $>10 \times$
- Decrease Cr vaporization by 10-100  $\times$
- Decreases spallation (thermal cycling)
  - RE improve adhesion
  - Outer layer prevents buckling
- Decreases contact resistance

But...

Cost effective?

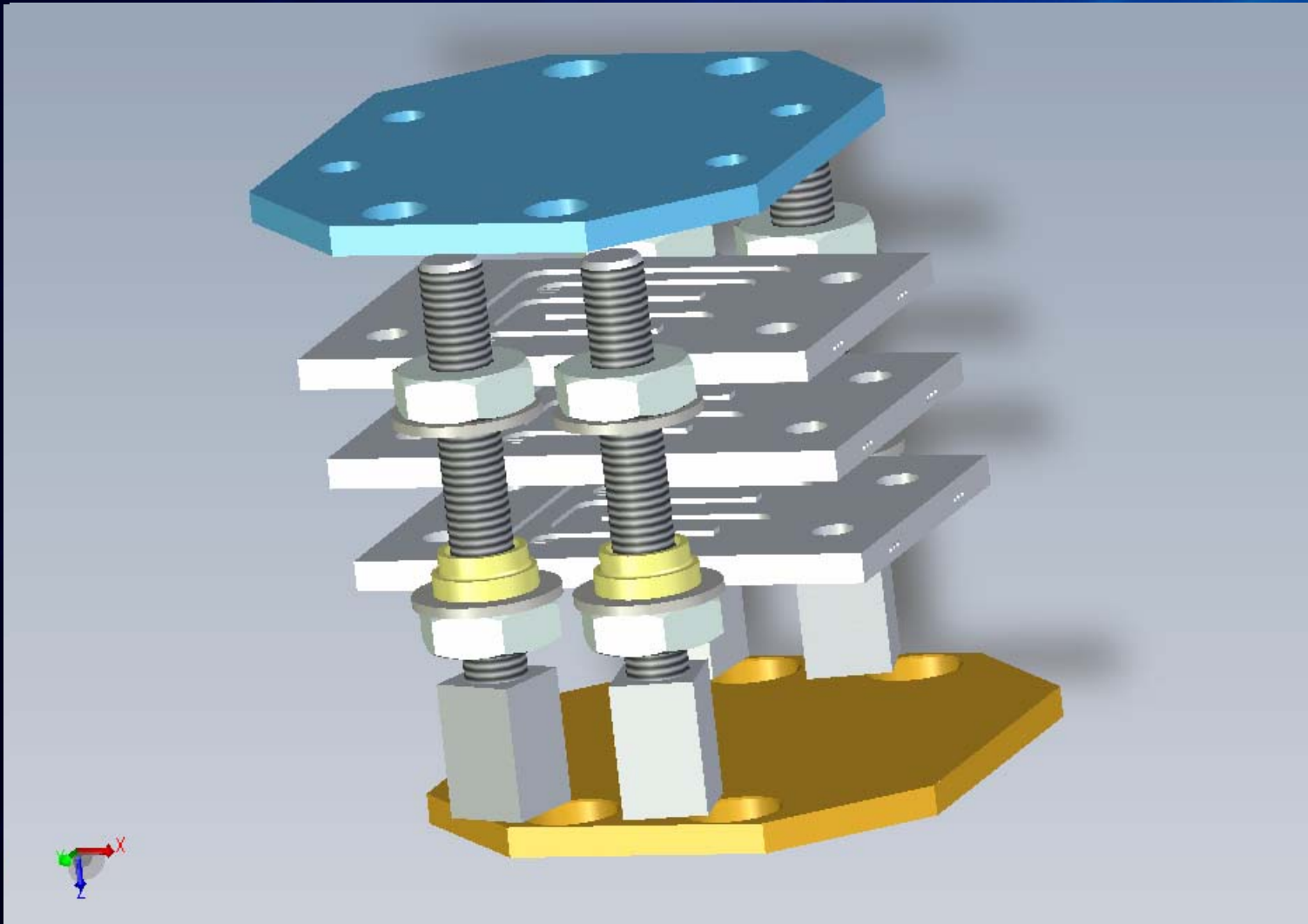
Complete coverage?

In-situ repair if fails?

Long-term stability?

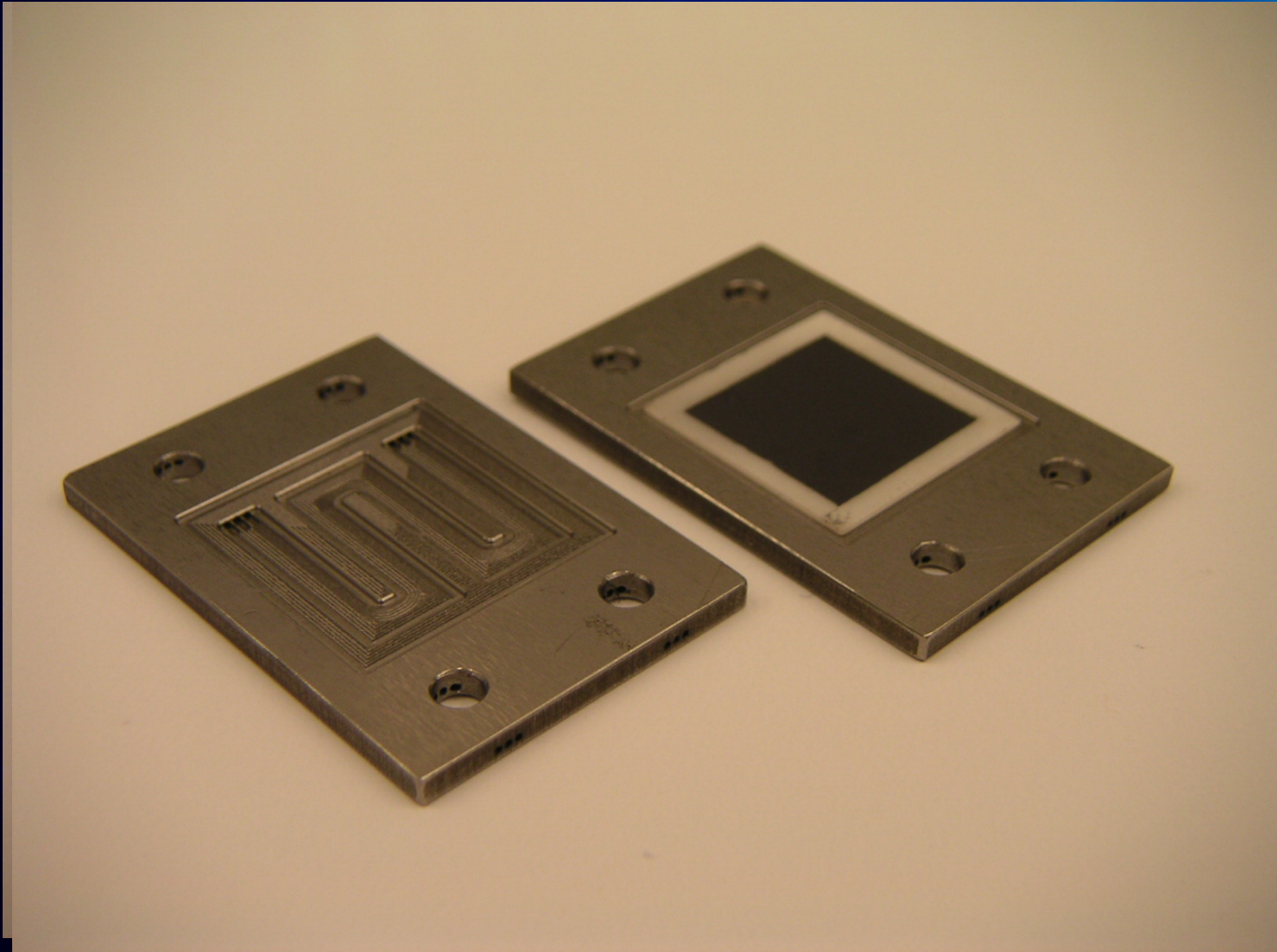


# National Laboratory Stack Test Platform

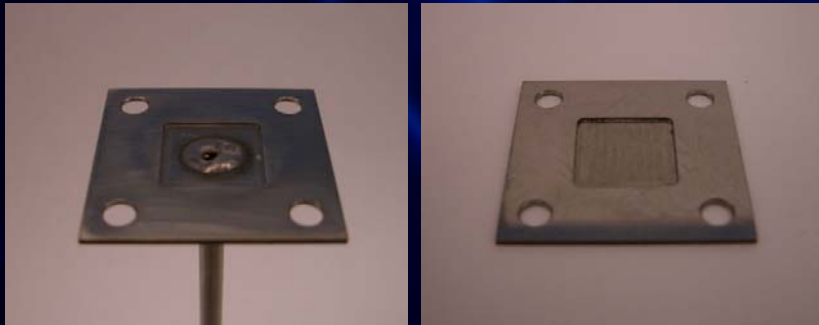




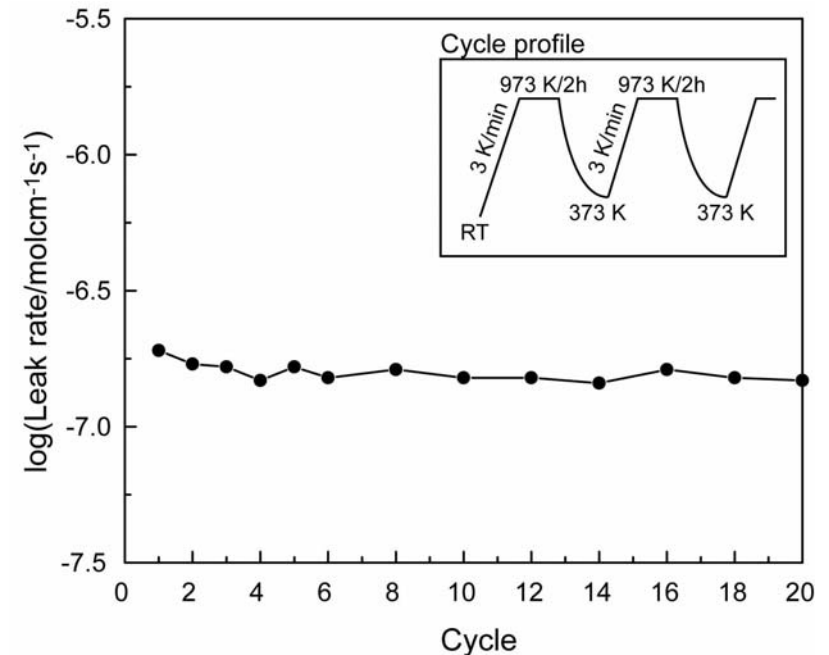
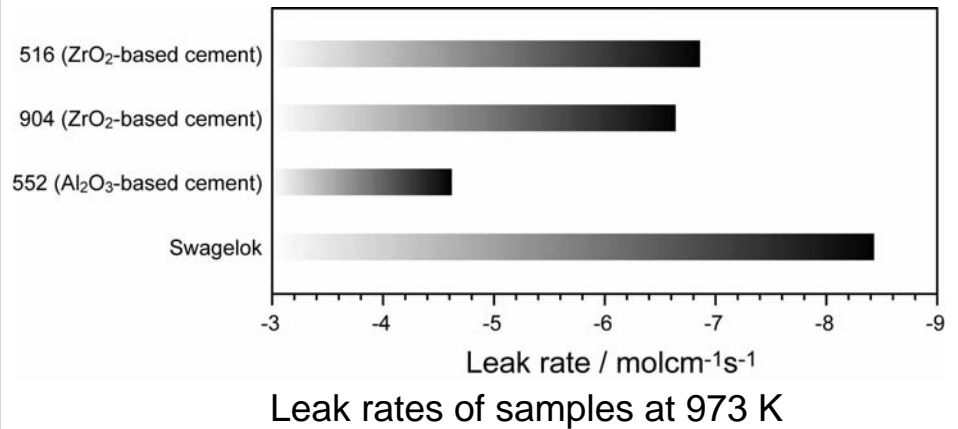
# National Laboratory Stack Test Platform



# ● Vacuum leak test of ceramic cements for sealing

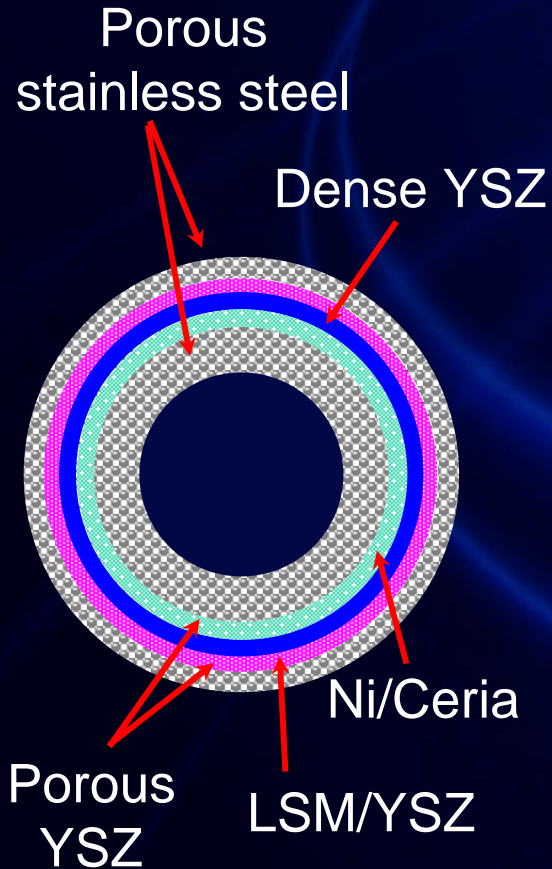


Test rig for vacuum leak test  
(plate: 430, tube: 316, welded with Nicro)

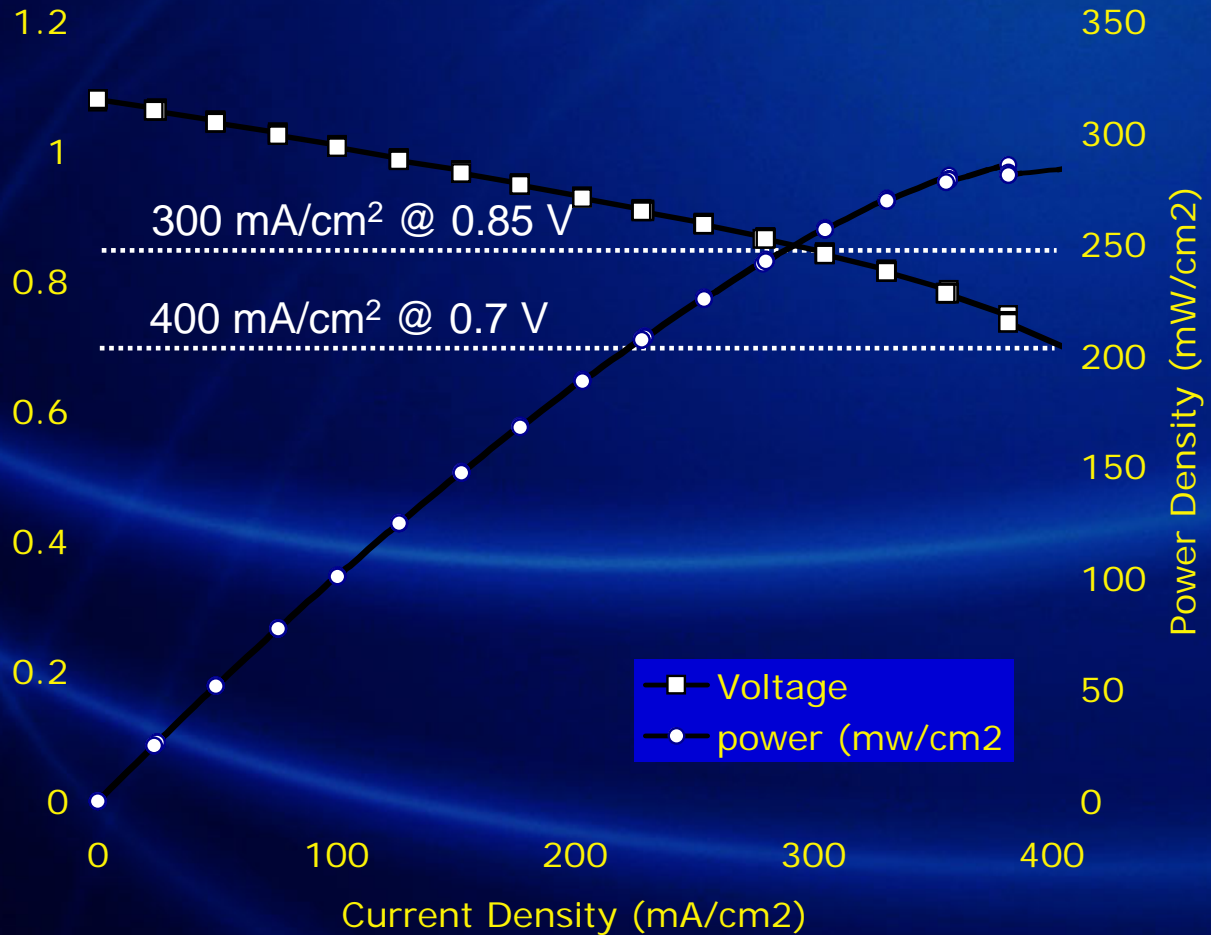




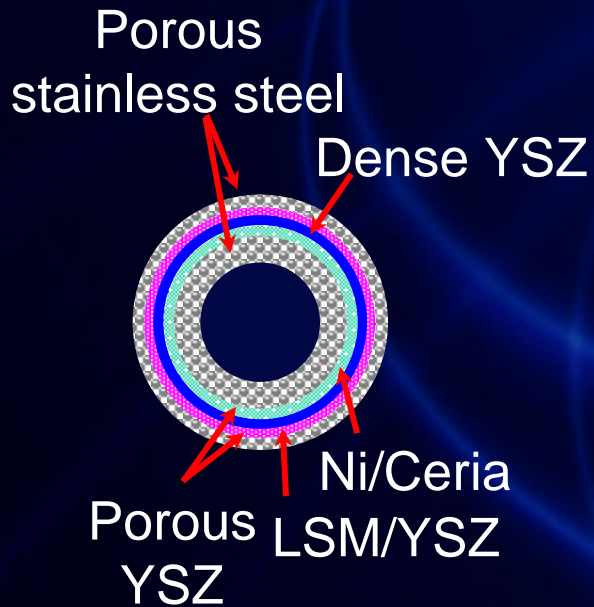
# Metal Supported Tubular SOFC w/Infiltrated Electrodes



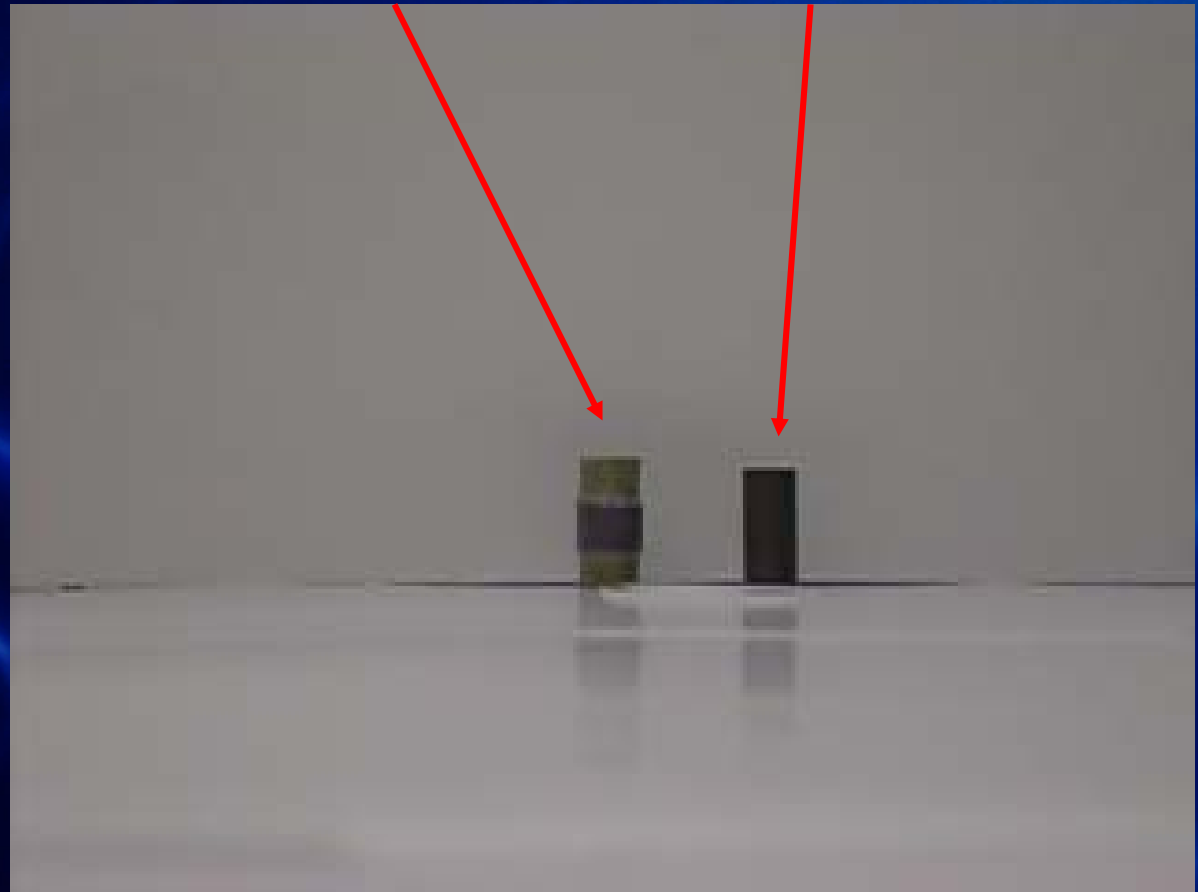
$\text{H}_2 + 3\% \text{H}_2\text{O} / \text{Air}$  at 700 °C



# Metal Supported vs. Ceramic SOFCs



Metal SOFC    Ceramic SOFC



# Summary

- SOFC Applications
  - Degradation rates of cells/stacks continue to decrease but further work is needed for stationary power applications
  - Transportation is near term but issues with thermal cycling must be addressed.
  - Feedback on failure mechanisms in stacks is needed
- Corrosion
  - The SOFC community is accumulating the necessary knowledge and understanding
  - More work on long term oxidation (effect of Si, etc) is needed
  - Basic research on interdiffusion, effects of g.b. etc is needed
- Protection
  - Coatings appear to resolve many issues with interconnect, if they can be cost effective



# Team:

## Investigators:

Steven Visco  
Lutgard De Jonghe

P.I. Program Lead  
Co-PI

## Scientists:

Peggy Hou  
Velimir Radmilovic

High temperature corrosion  
NCEM FIB/SEM/TEM

## Post Doc:

Hideto Kurokawa  
Ken Lux

Cr transport phenomena  
Air electrode stability

## Senior Technical Staff:

Craig Jacobson  
Mike Tucker  
Grace Lau  
Inna Belogolovsky

Processing and characterization  
Metal supported SOFC development  
Processing and analysis  
Processing and testing

## Graduate Students:

Tal Sholklapper  
Liming Yang  
Xuan Chen

Nano-particulate catalysts  
Novel anode catalysts  
Infiltration of cathode catalysts





# Acknowledgements

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- This work was supported by SECA through NETL
- Thanks to Lane Wilson for helpful discussions on Cr issues and model stack construction





