

OC/EC Analysis with Thermal-Optical Methods

Effects of temperature protocol and non-carbonaceous compounds

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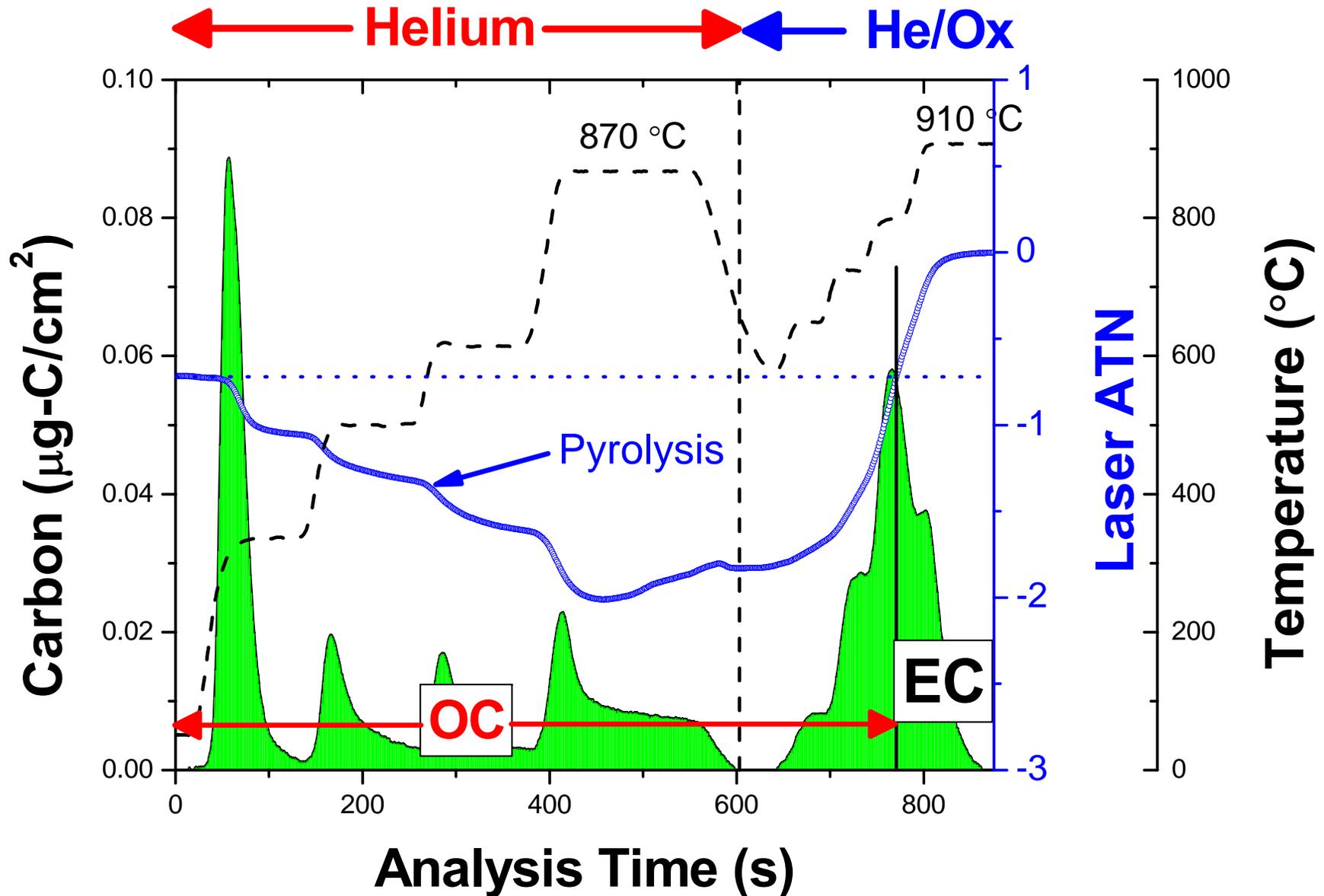
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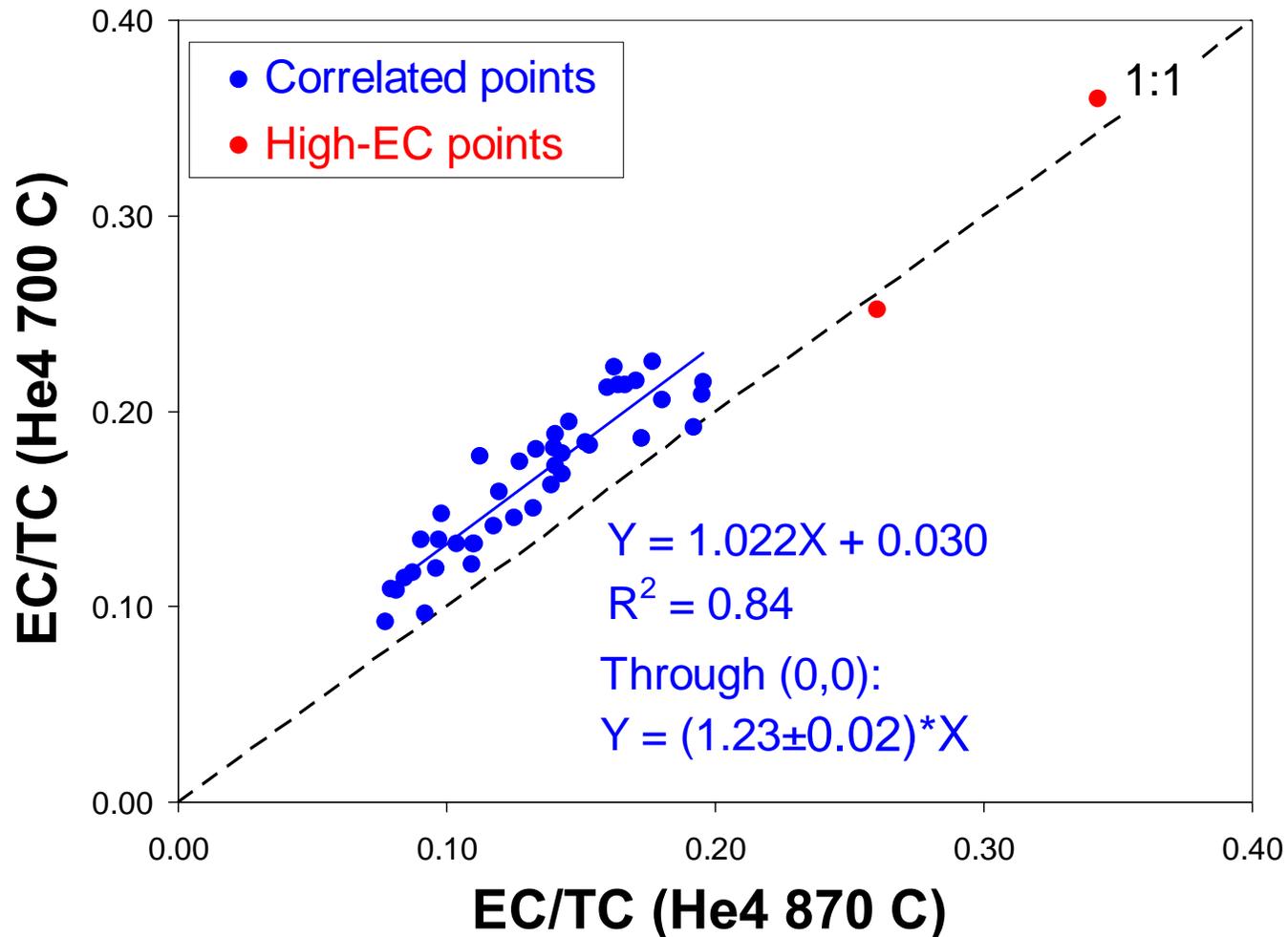
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Thermal-Optical Analysis for OC/EC

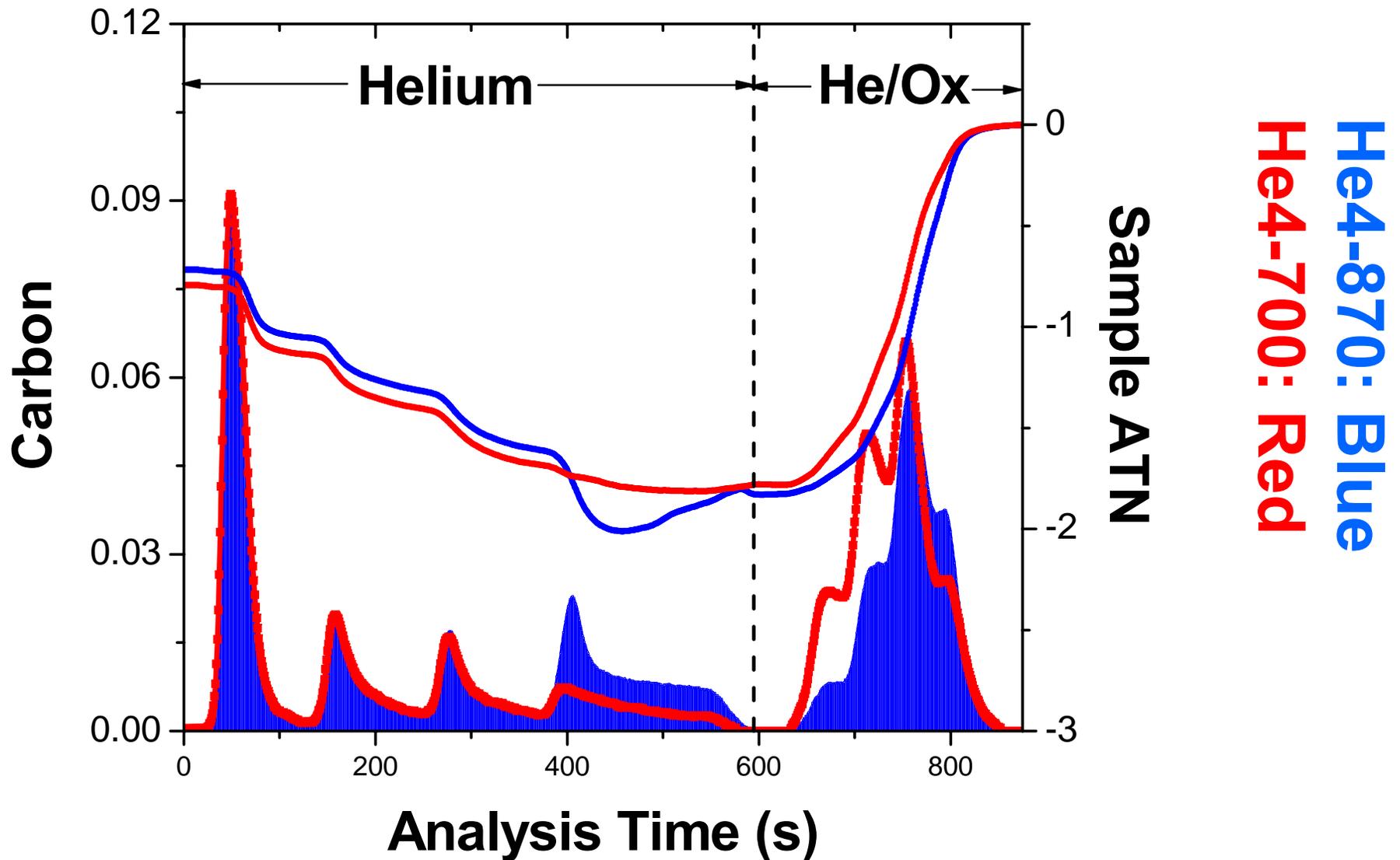


Temperature protocol affects EC



Also seen by Chow et al. (2001); Schauer et al. (2003); Conny et al. (2004)

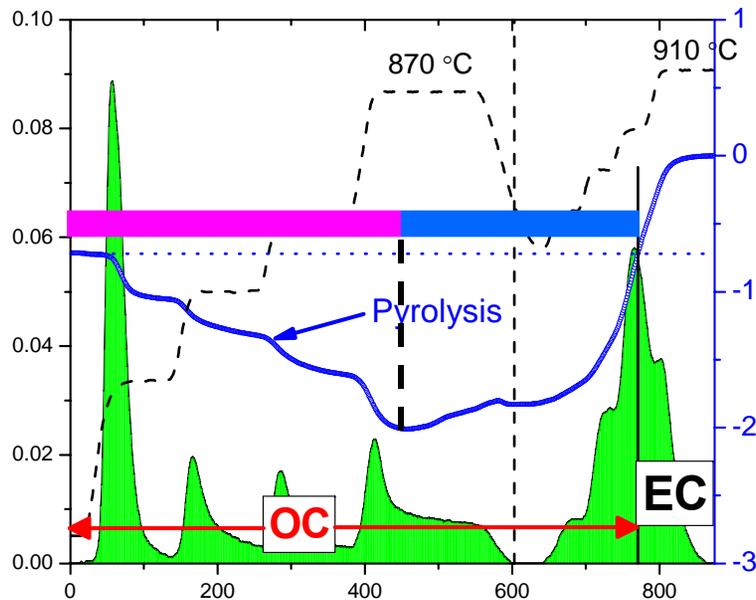
Differences: He4-870 v/s He4-700



Objectives of our research

- Develop a mechanistic understanding of the thermal/optical method:
 - Evaluate the assumptions
 - Carbon evolution pattern
 - Optical properties of evolving carbon
- Estimate the potential uncertainties
- Examine possible improvements to the EPA thermal/optical method

Method assumptions



$$PC_{\text{total}} * k_{\text{PC}} \sim PC_{\text{pre-split}} * k_{\text{PC}} + EC_{\text{pre-split}} * k_{\text{EC}}$$

Assumption 1: $EC_{\text{pre-split}} = 0$

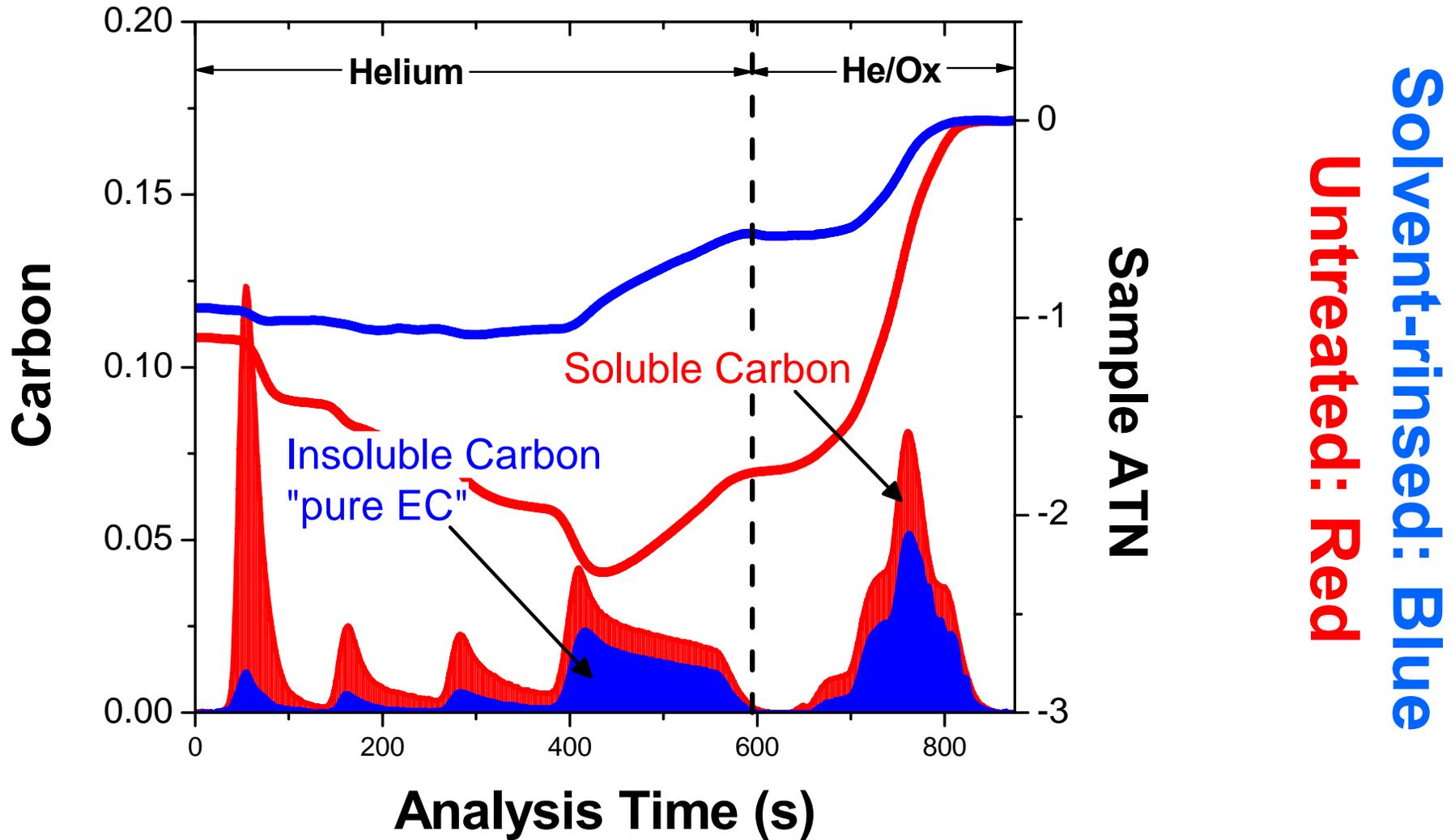
(PC evolves completely before EC)

or

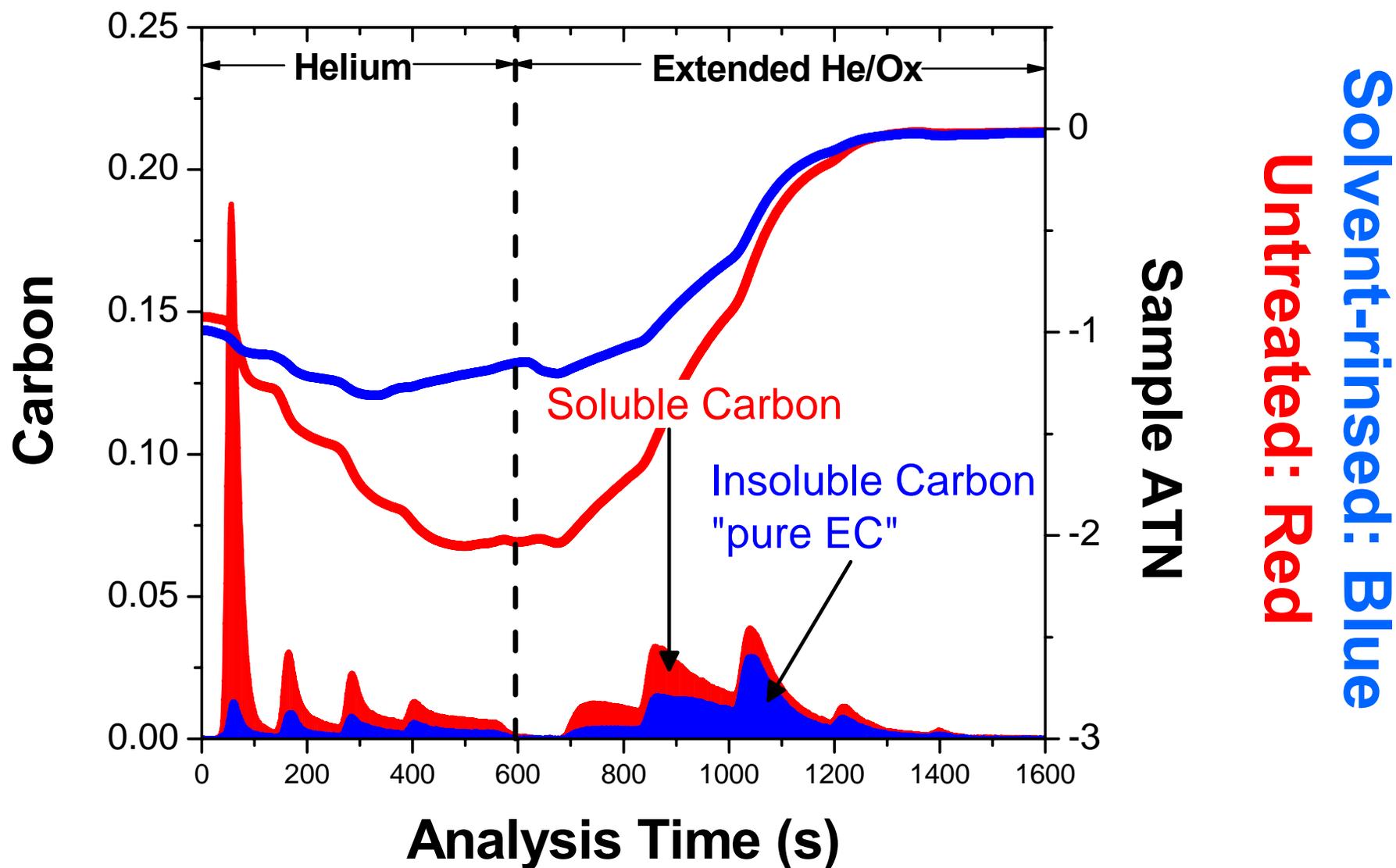
Assumption 2: $k_{\text{PC}} = k_{\text{EC}}$

(Optical properties of PC and EC are similar)

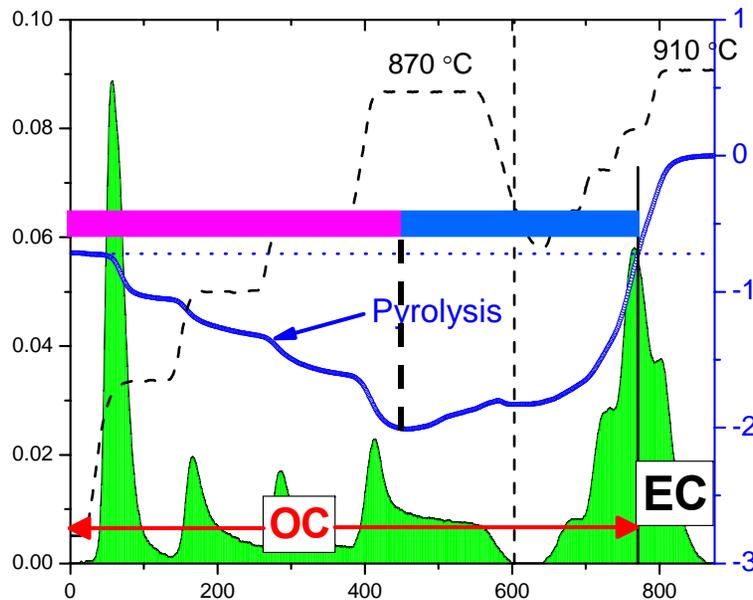
He4-870: Co-evolution of PC and EC



He4-700: Co-evolution of PC and EC



Assumption 1 not valid



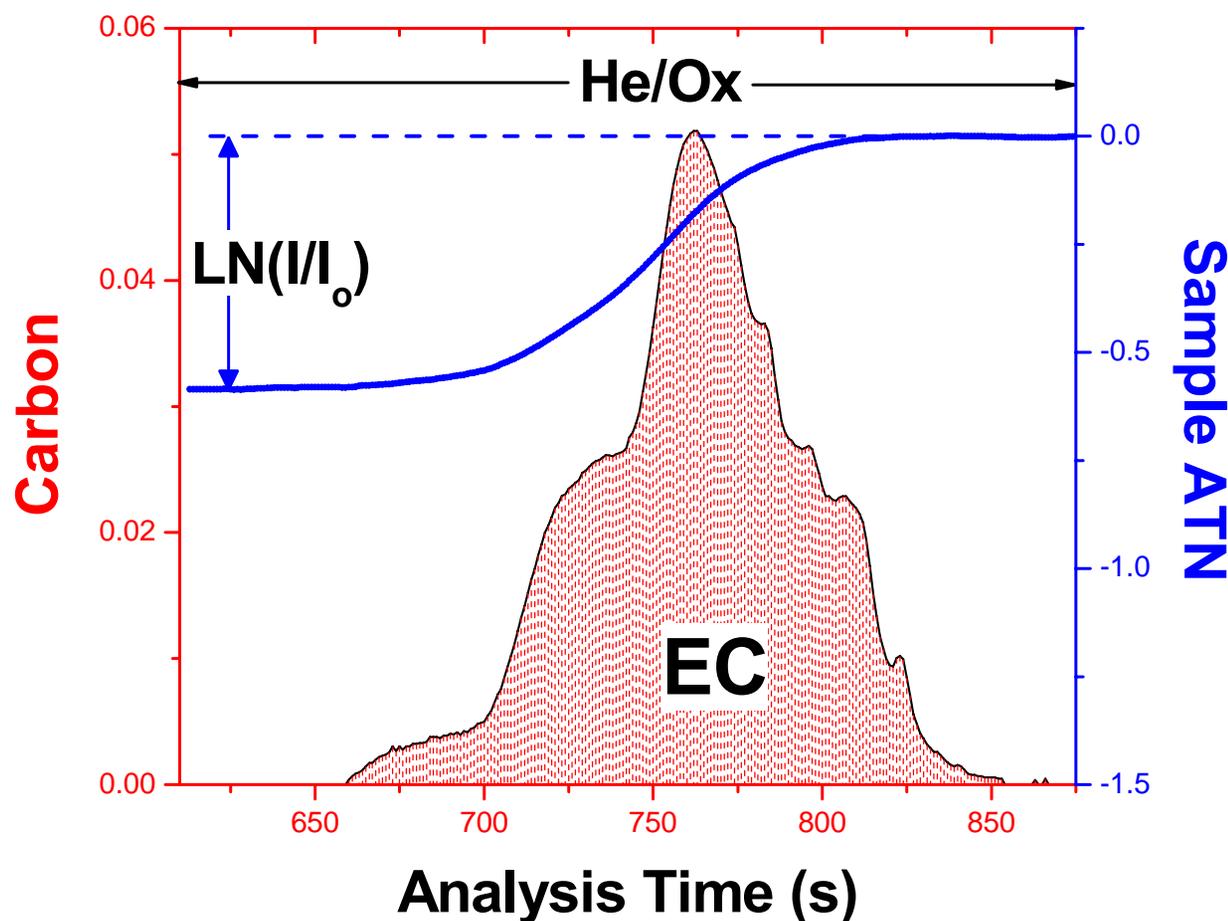
$$PC_{\text{total}} * k_{\text{PC}} \sim PC_{\text{pre-split}} * k_{\text{PC}} + EC_{\text{pre-split}} * k_{\text{EC}}$$

~~Assumption 1: $EC_{\text{pre-split}} \neq 0$~~
(PC co-evolves with EC)

or

Assumption 2: $k_{\text{PC}} = k_{\text{EC}}$
(Optical properties of PC and EC are similar)

Calculating k_{EC} with the solvent-rinsed filters



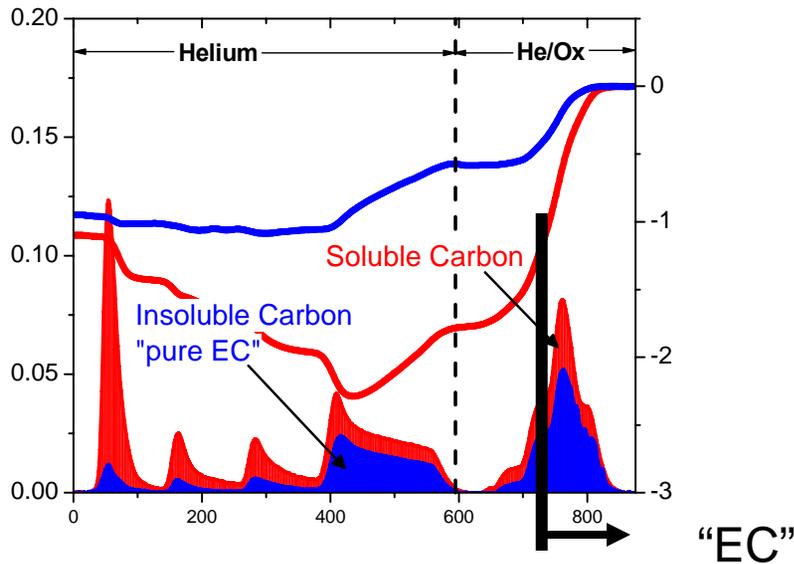
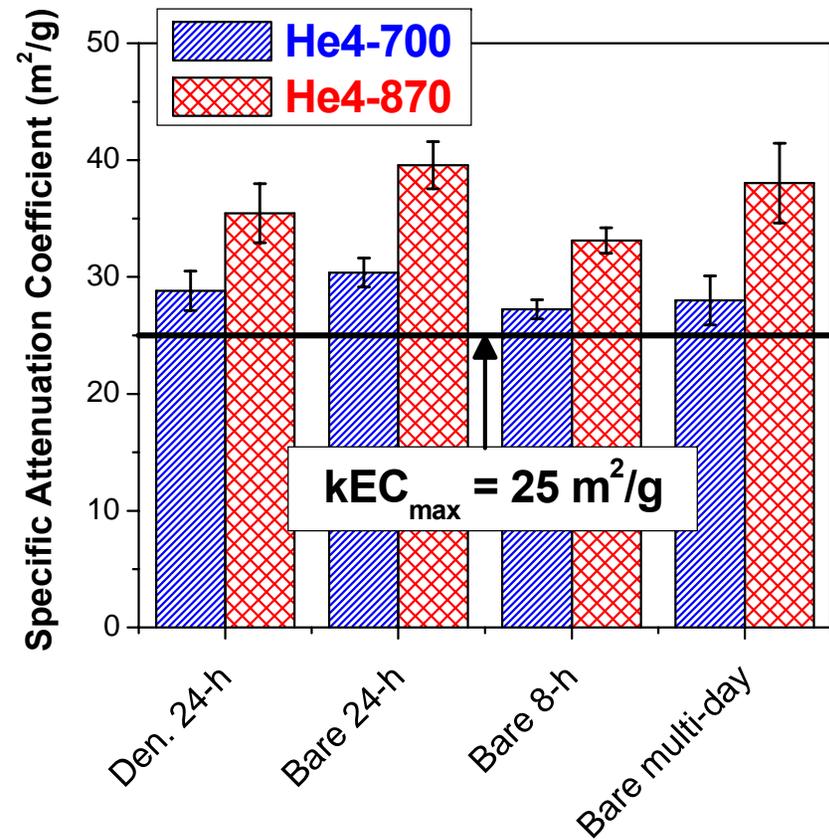
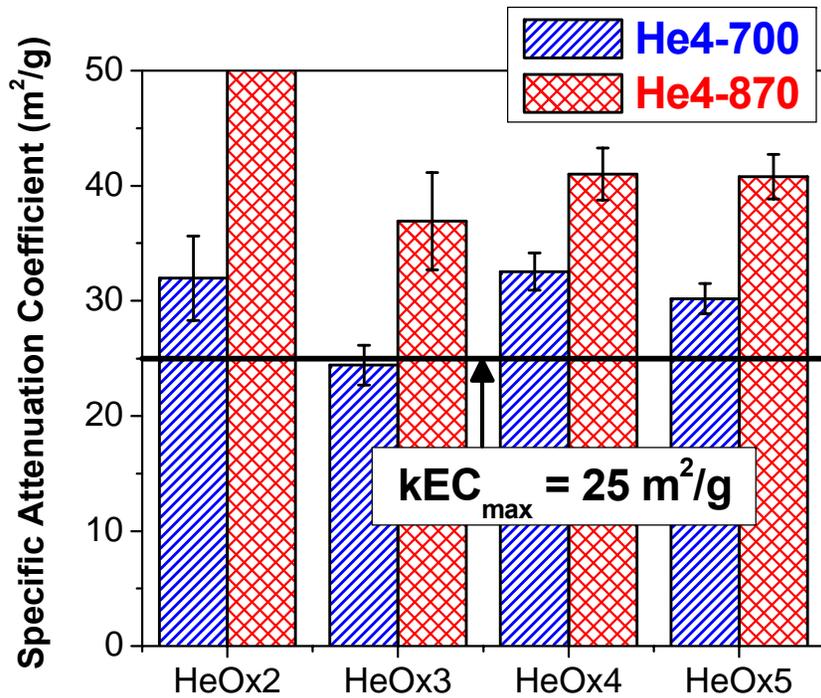
Lambert-Beer law:
 $LN(I/I_0) = EC * k_{EC}$

$$k_{EC_{rinsed}} = 25.8 \pm 2.8 \text{ m}^2/\text{g-C}$$

k_{EC} similar to the findings of Gundel et al. (1984)

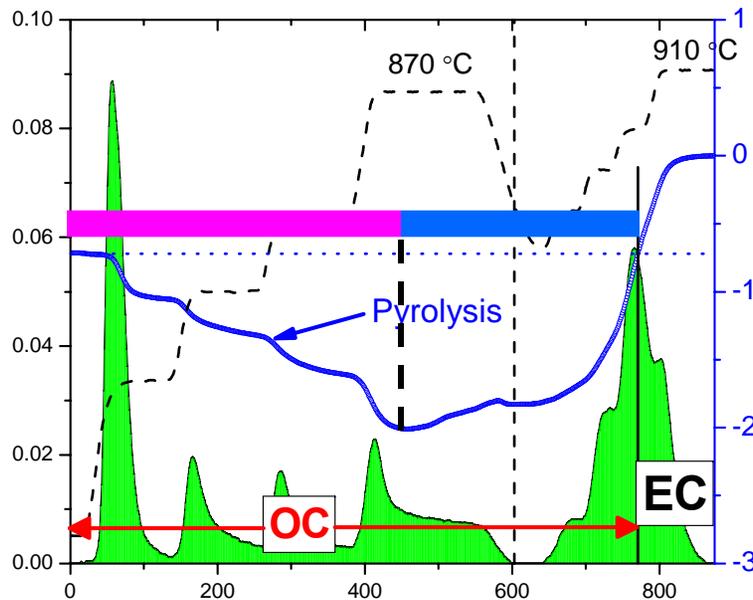
Squirrel Hill tunnel samples: $k_{EC} \sim k_{He/Ox} = 20.6 \pm 1.8 \text{ m}^2/\text{g}$

Specific attenuation of untreated carbon



Instrument-defined EC (k_{EC})

Assumption 2: also not valid!



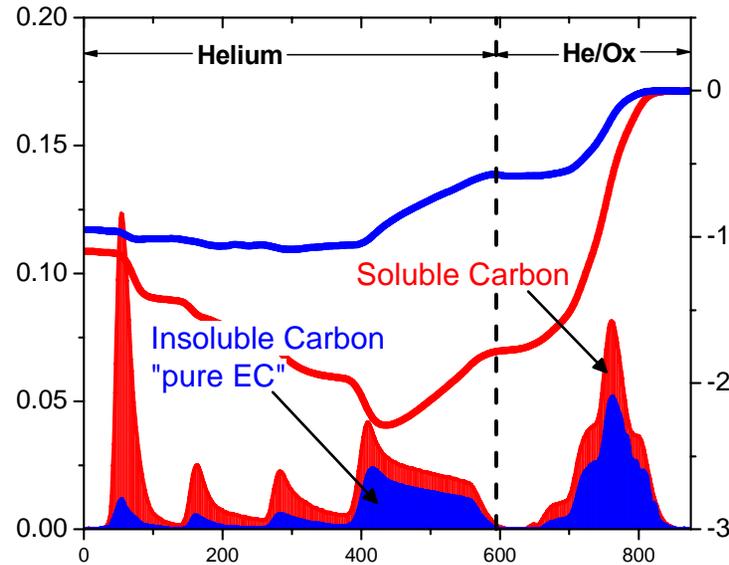
$$PC_{\text{total}} * k_{PC} \sim PC_{\text{pre-split}} * k_{PC} + EC_{\text{pre-split}} * k_{EC}$$

~~Assumption 1: $EC_{\text{pre-split}} \neq 0$~~
(PC co-evolves with EC)

or

~~Assumption 2: $k_{PC} \gg k_{EC}$~~
(Optical attenuation by PC is much greater than that by EC)

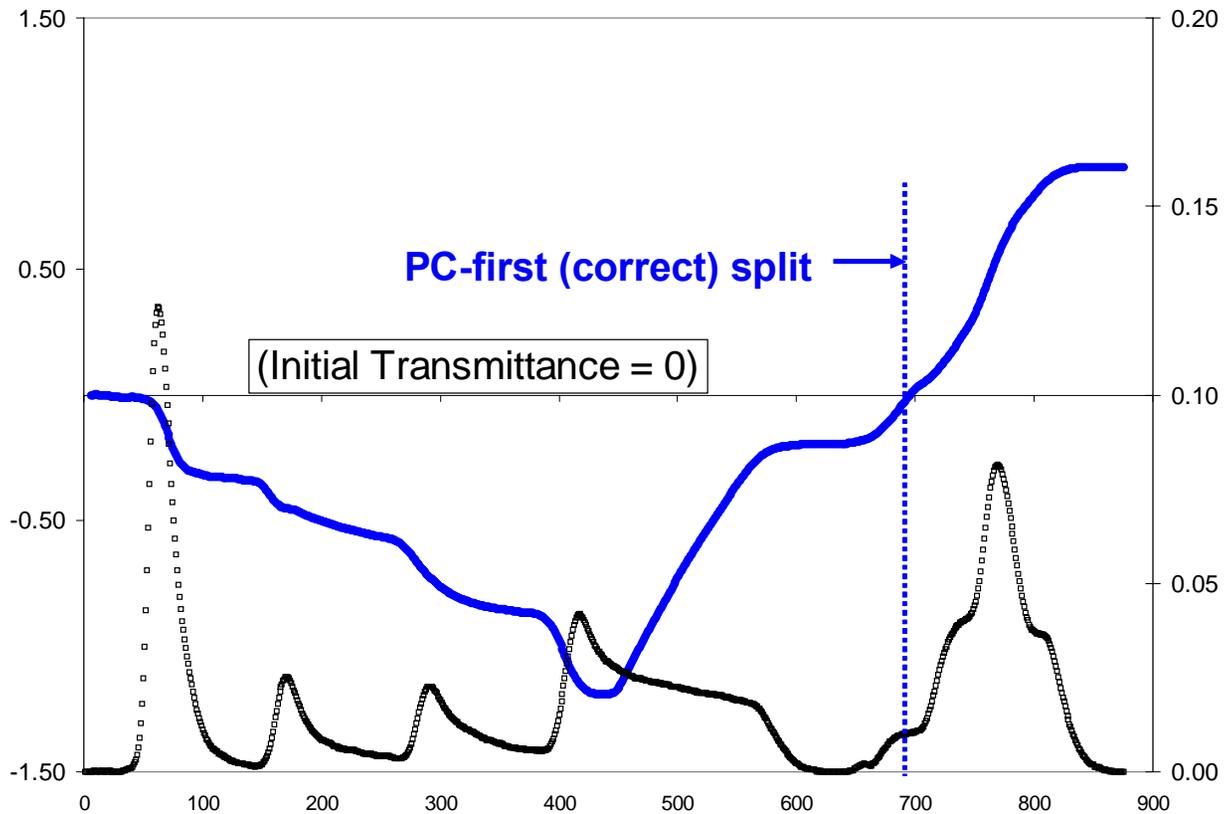
Uncertainties due to different optical properties



$$PC_{\text{total}} * 55.8 = PC_{\text{pre-split}} * 55.8 + EC_{\text{pre-split}} * 15.7$$

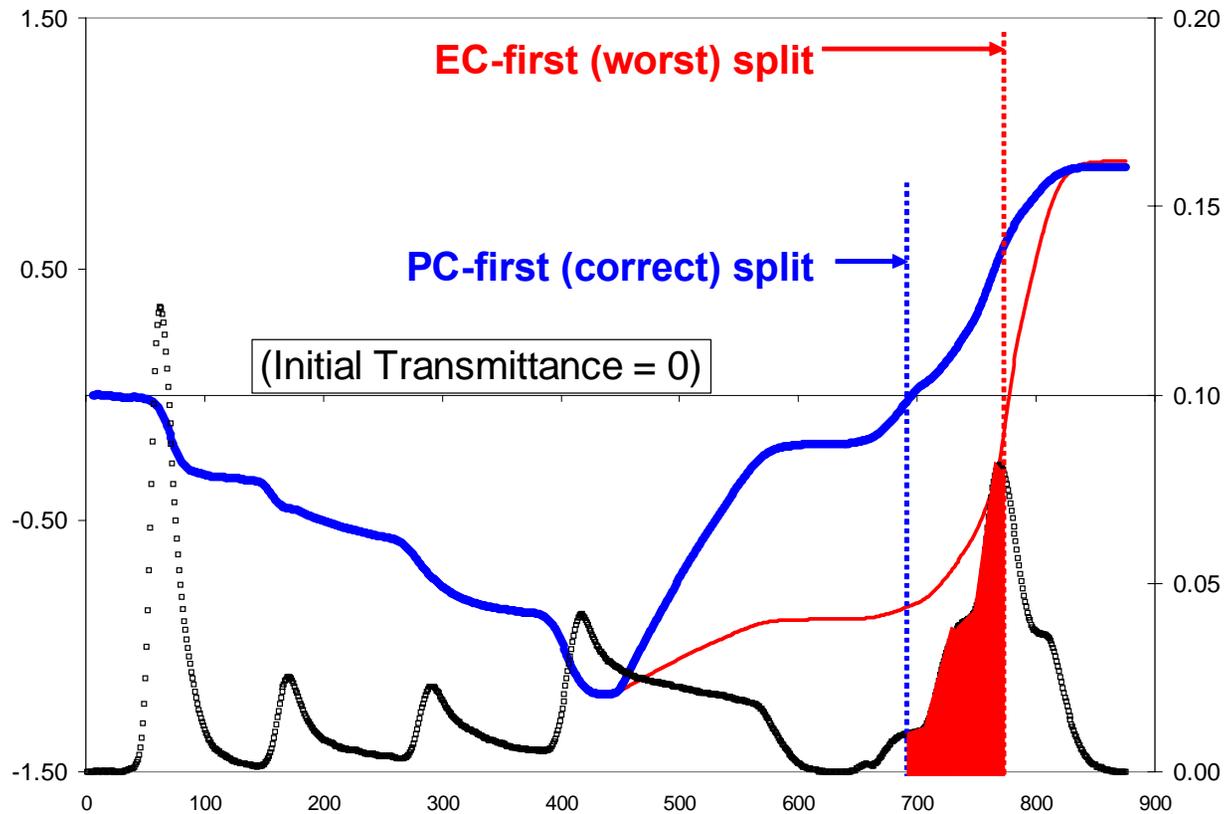
- (a) Ideal: PC evolves completely before EC
- (b) Worst case: EC evolves completely before PC
- (c) Actual: Co-evolution of PC and EC

Ideal: PC evolves before EC



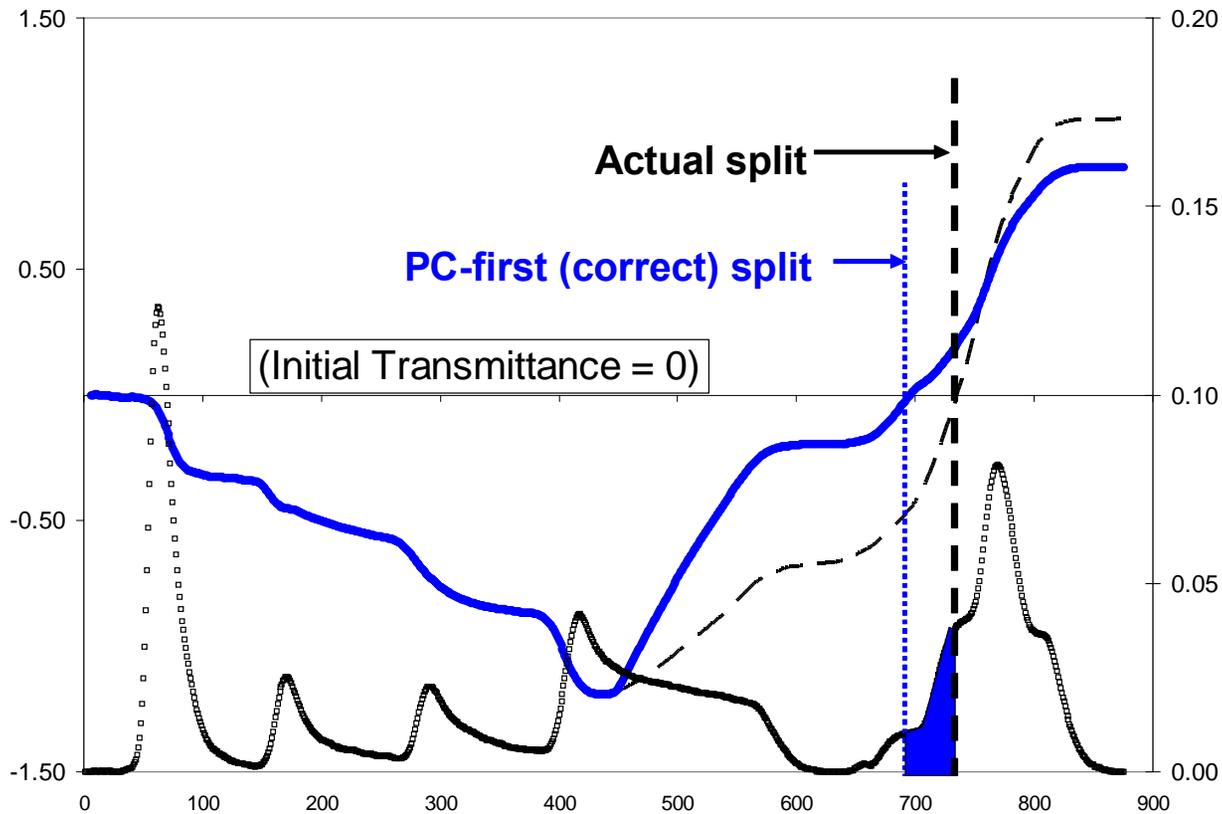
Actual EC = $5.5 \mu\text{g-C}/\text{cm}^2$

Worst case: EC evolves completely before PC



Measured EC = $1.7 \mu\text{g-C/cm}^2$
Error = - 68%

Actual under-estimation by He4-870 protocol

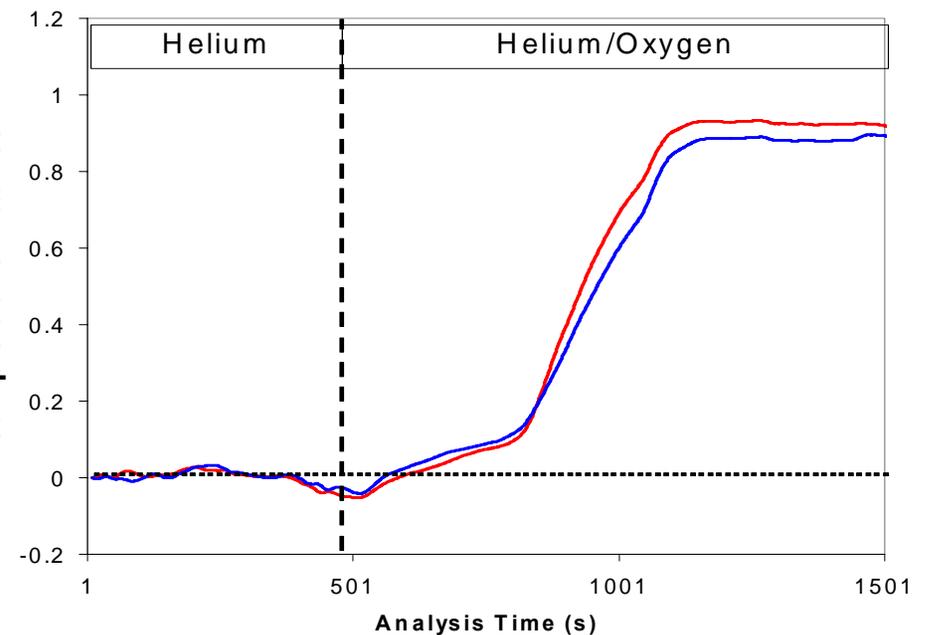
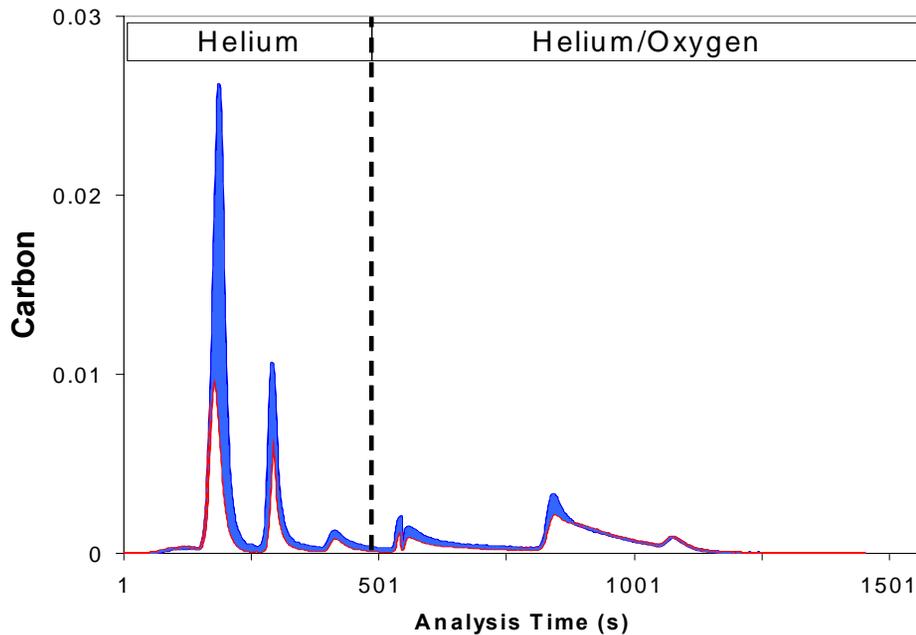


Measured EC = $4.4 \mu\text{g-C}/\text{cm}^2$
Error = - 20%

Improving the EPA thermal/optical method

- Reducing the peak Helium temperature reduces the negative bias
- He4-700 increases the EC compared to the He4-870
- What about reducing the peak Helium temperature further?
 - e.g. IMPROVE has a peak Helium temperature of 550 °C
 - Potential positive bias with unpyrolyzed organics?

Improving the EPA thermal/optical technique: He4-550?



Vehicular exhaust:

TC $14.2 \mu\text{g-C}/\text{cm}^2$

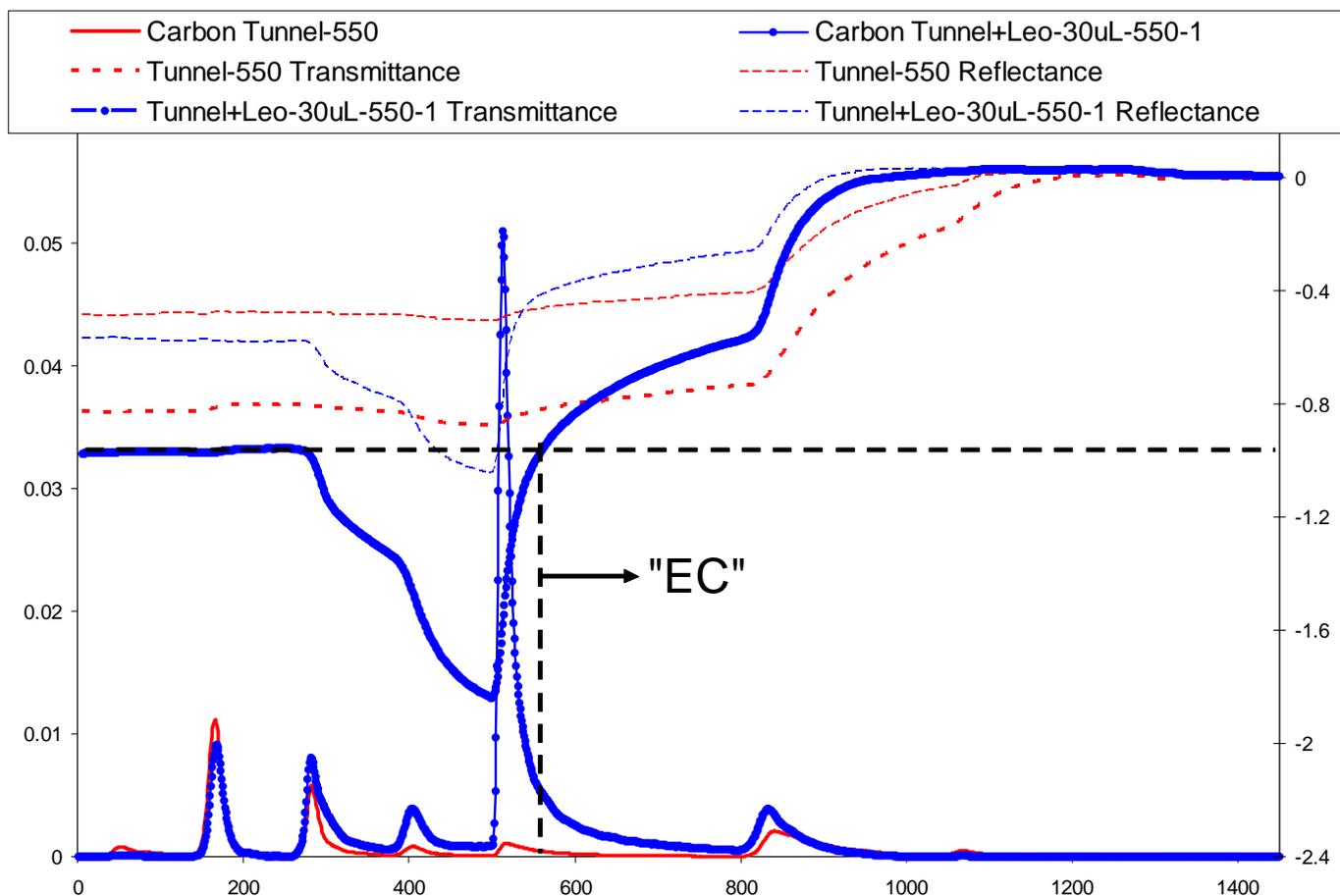
EC $5.4 \mu\text{g-C}/\text{cm}^2$

Veh. exhaust + Levoglucosan:

TC $23.9 \mu\text{g-C}/\text{cm}^2$

EC $7.4 \mu\text{g-C}/\text{cm}^2$

He4-550: Positive bias with leonardite?



Vehicular exhaust:
TC 6.83 $\mu\text{g-C/punch}$
EC 2.28 $\mu\text{g-C/punch}$

Veh. exhaust + leonardite:
TC 22.2 $\mu\text{g-C/punch}$
EC 5.89 $\mu\text{g-C/punch}$

Conclusions

- The thermal/optical method assumptions are not valid
 - PC and EC co-evolve; $k_{PC} \gg k_{EC}$ for Pittsburgh samples
- He4-870 under-estimates EC
 - Premature EC evolution coupled with different k_{EC} and k_{PC}
- He4-700 EC is 23 ± 2 % higher than He4-870 EC
 - Reduces premature EC loss
- He4-550 may be too low, possibly subject to positive bias
 - Need to investigate large polymeric and colored organic compounds

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