

Mass Balance Closure and the Federal Reference Method for PM_{2.5} in Pittsburgh

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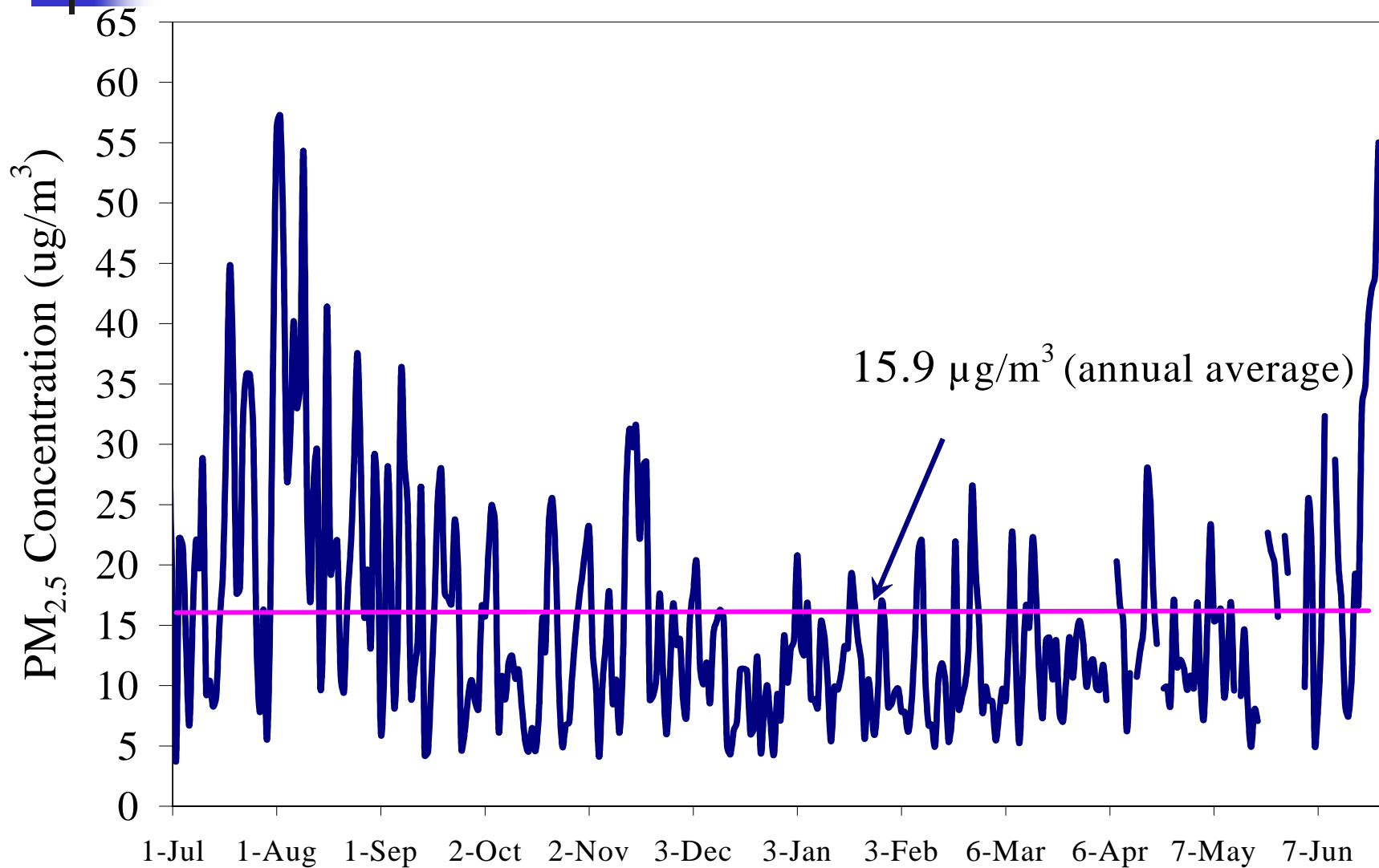
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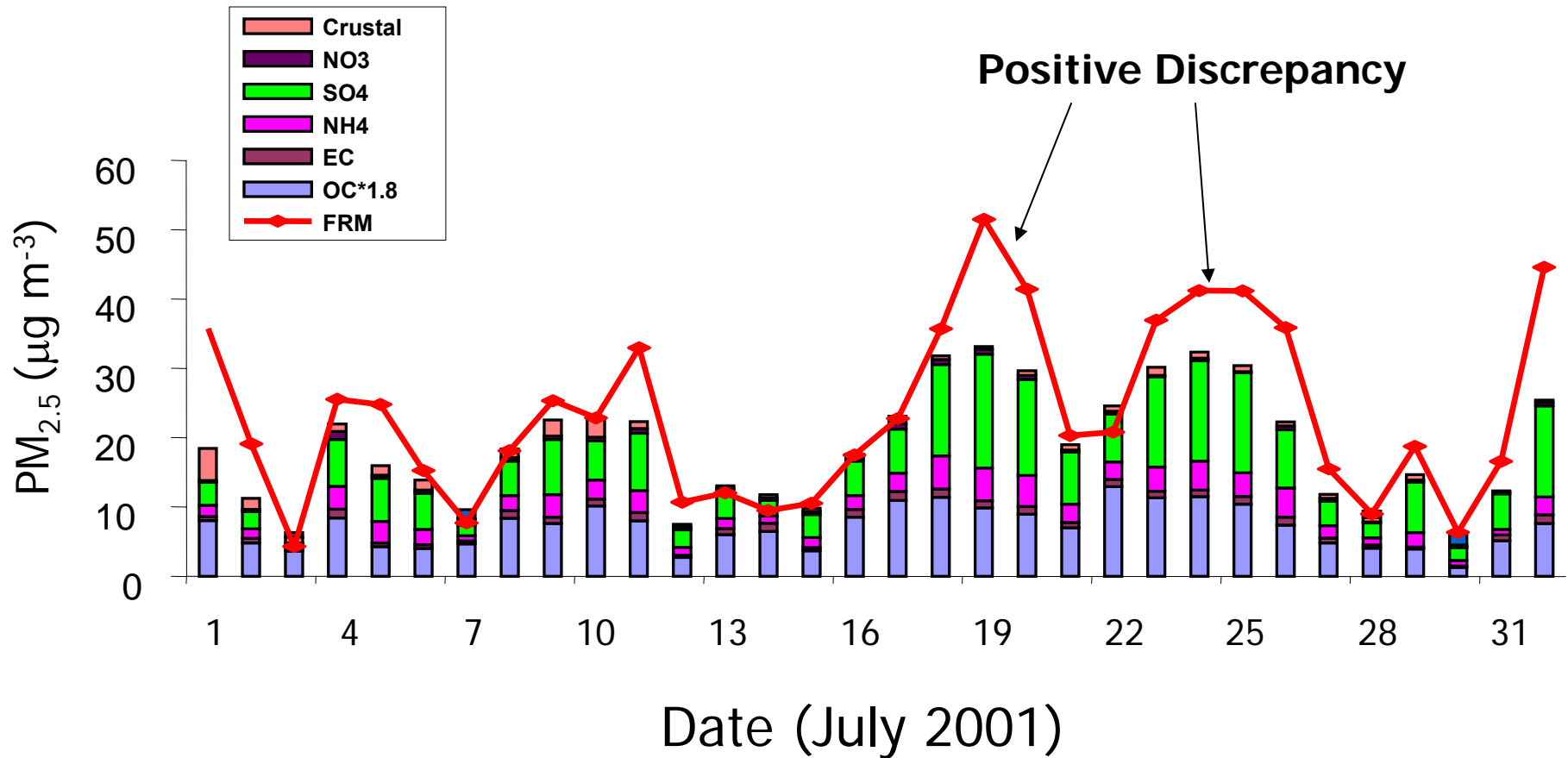
Overview

- Compare FRM PM_{2.5} mass to the sum of aerosol chemical components
- Mass balance discrepancy
 - Positive: FRM > Σ chemical components
 - Negative: FRM < Σ chemical components
- Water
- Volatilization of Organics and Nitrate

FRM Mass July 2001 – June 2002

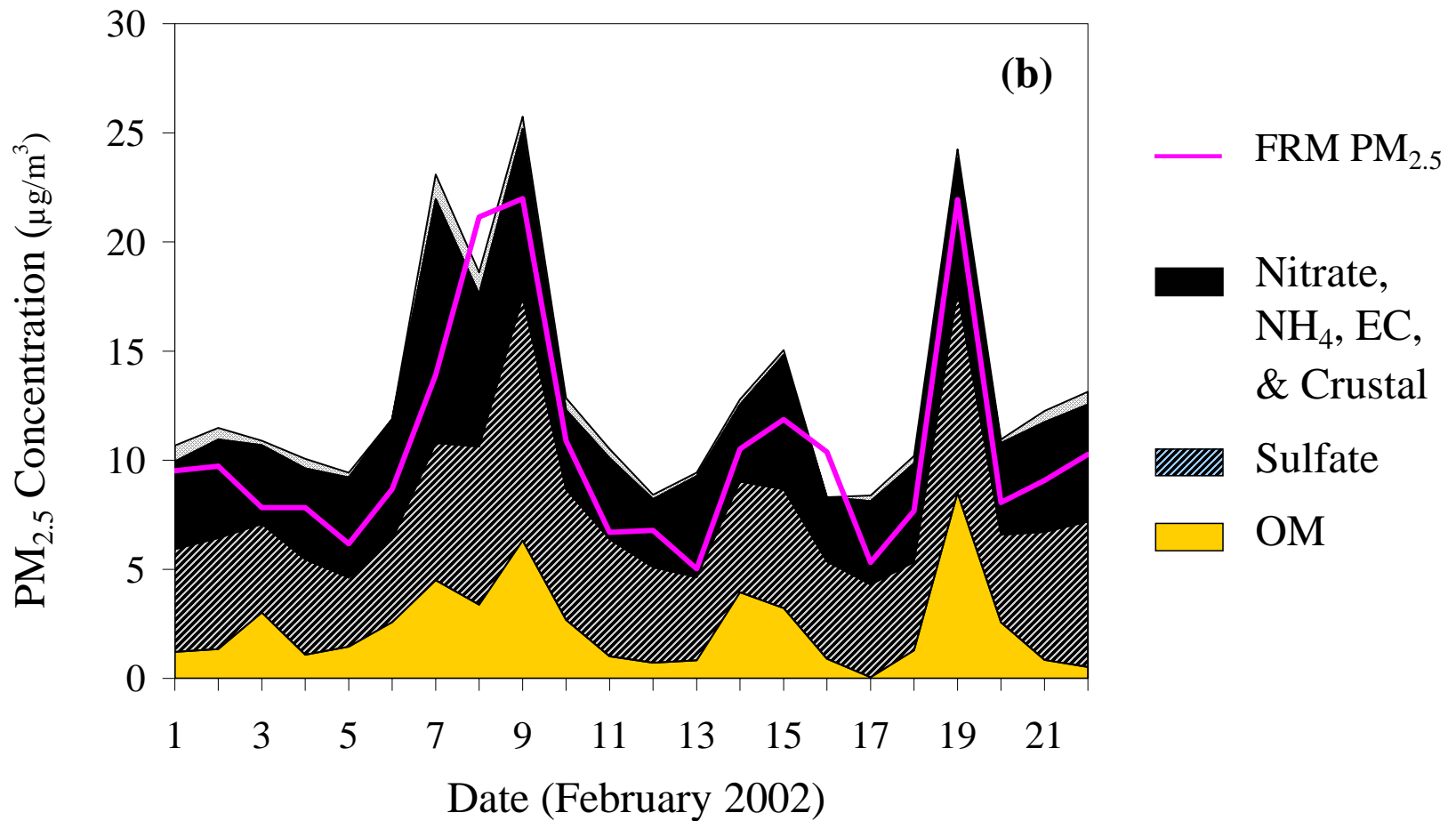


Mass Balance – July 2001

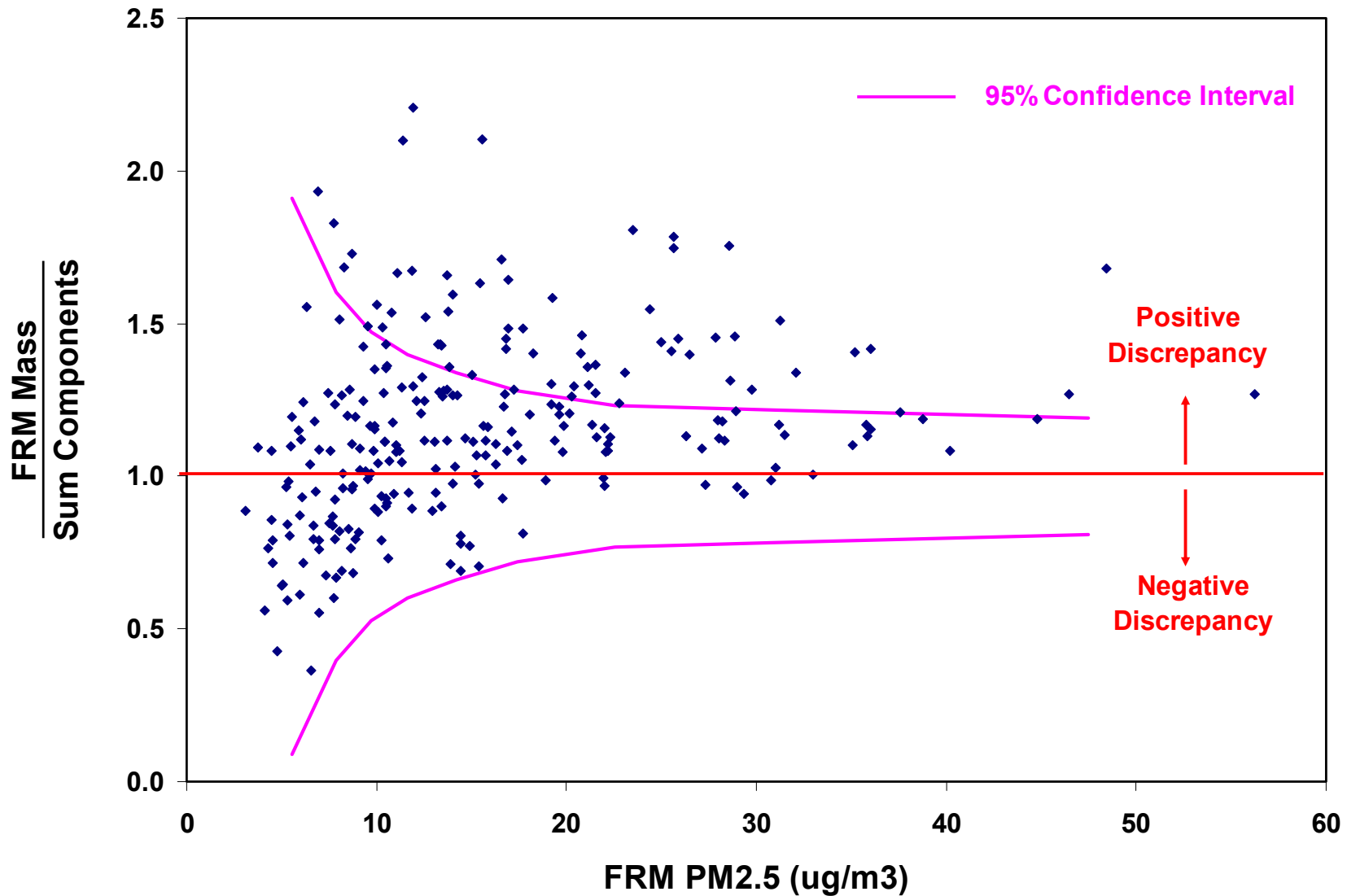


OM = OC x 1.8

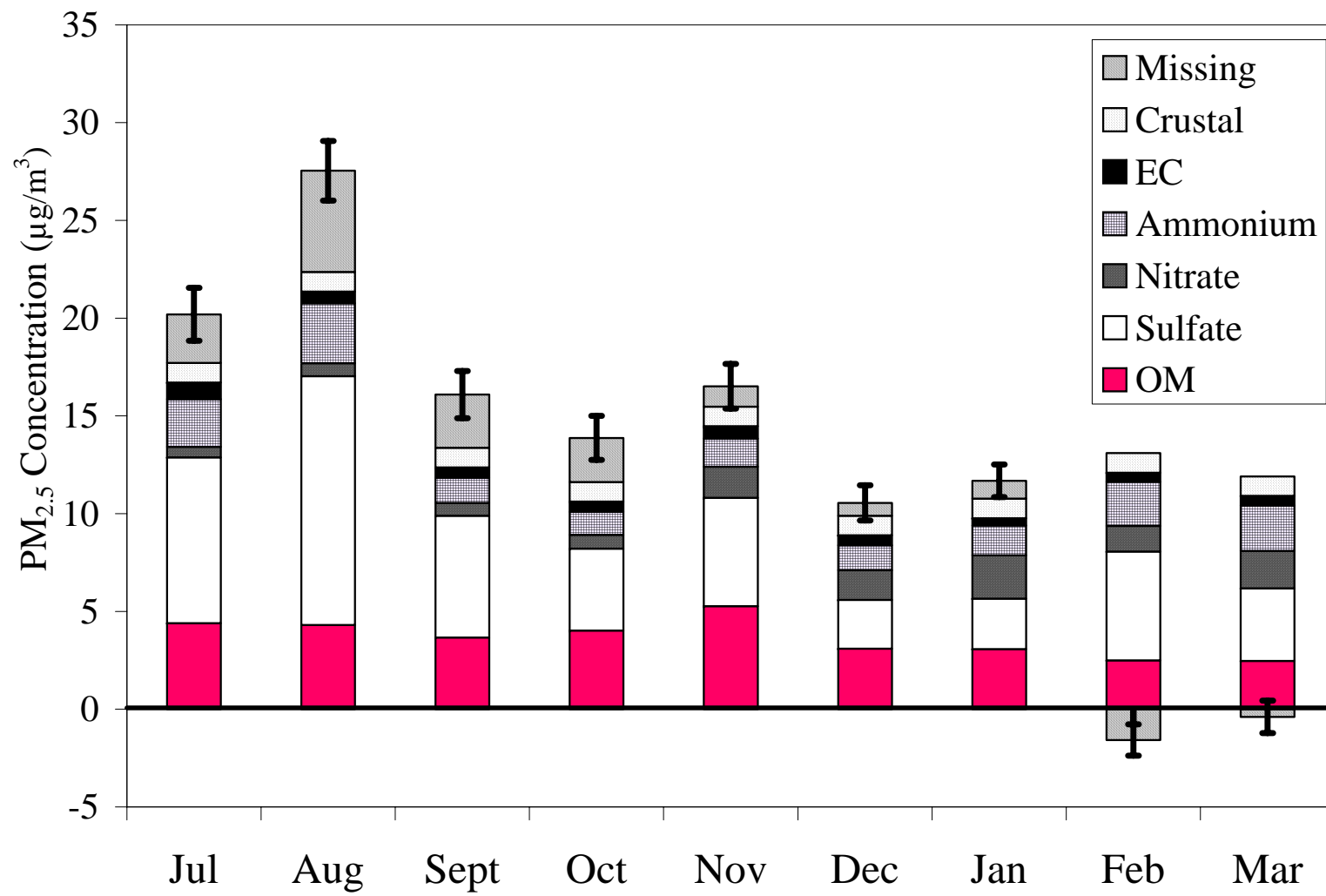
Mass Balance - February 2002



Daily Mass Balance Discrepancy



Monthly Mass Balance



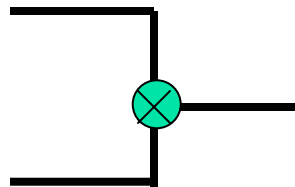


Hypotheses to Explain Mass Discrepancy

- Water: Positive mass discrepancy
(FRM > components)
- Volatilization: Negative mass discrepancy (FRM < sum of components)
 - Organic volatilization
 - Nitrate volatilization

Measurements of aerosol water content

Dry inlet



Wet inlet

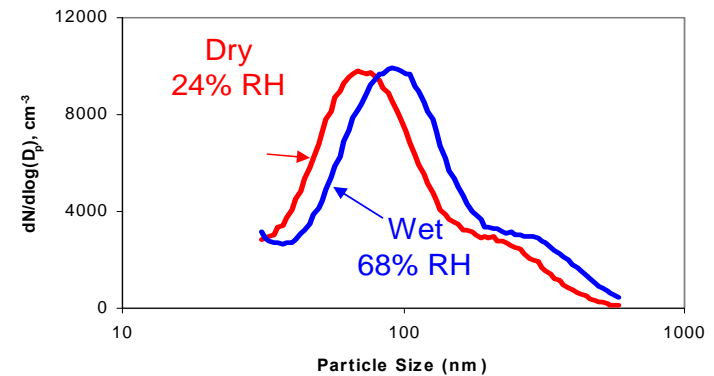
Particle Sizers

Nano-SMPS

SMPS

APS

DAASS: Dry/Ambient Aerosol Size Spectrometer
12PD-14



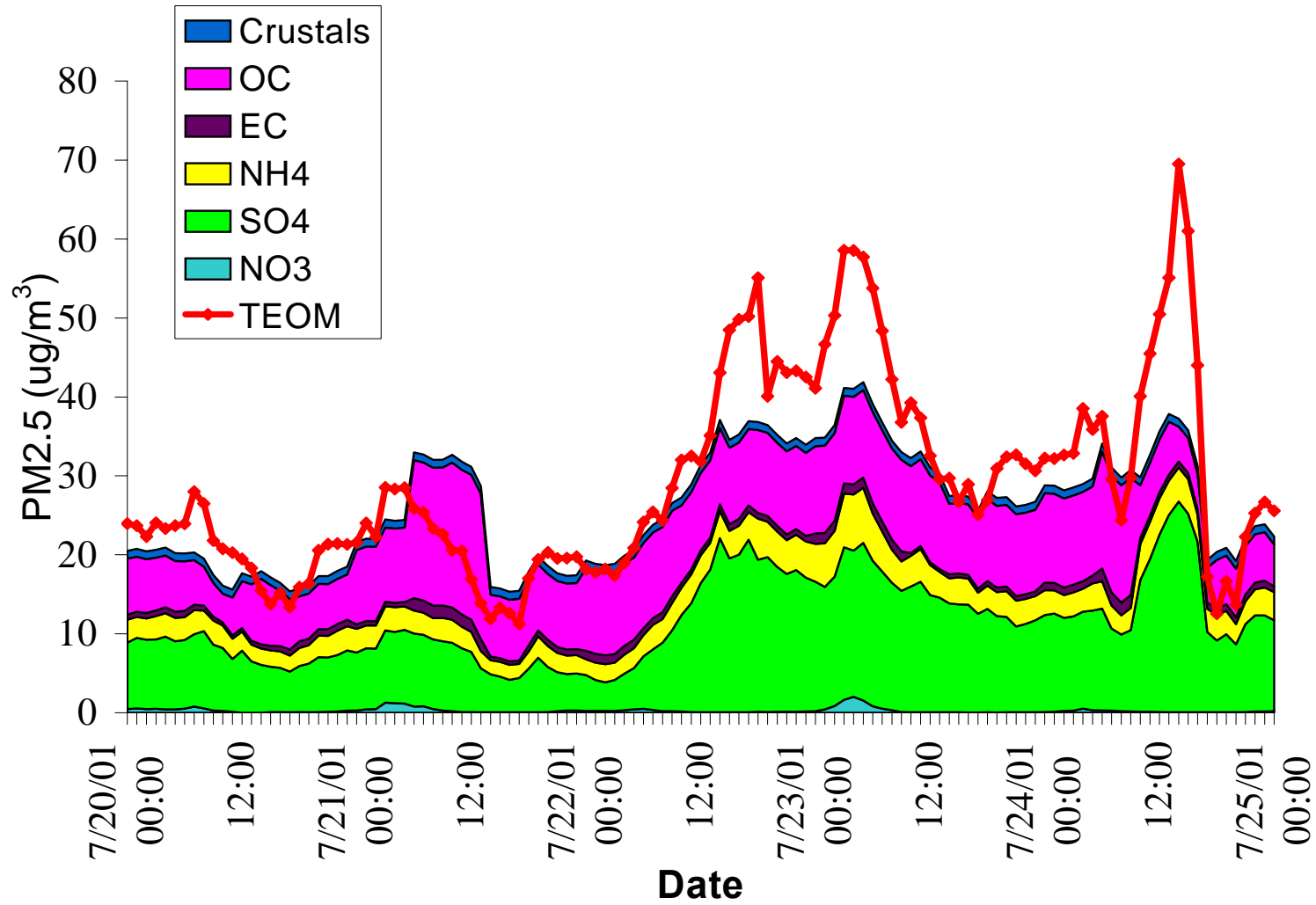
Estimating Aerosol Water at 35% RH

$$M_{H_2O} = \left(V_{wet} - \frac{M_{dry}}{\rho_{dry}} \right) \rho_{H_2O}$$

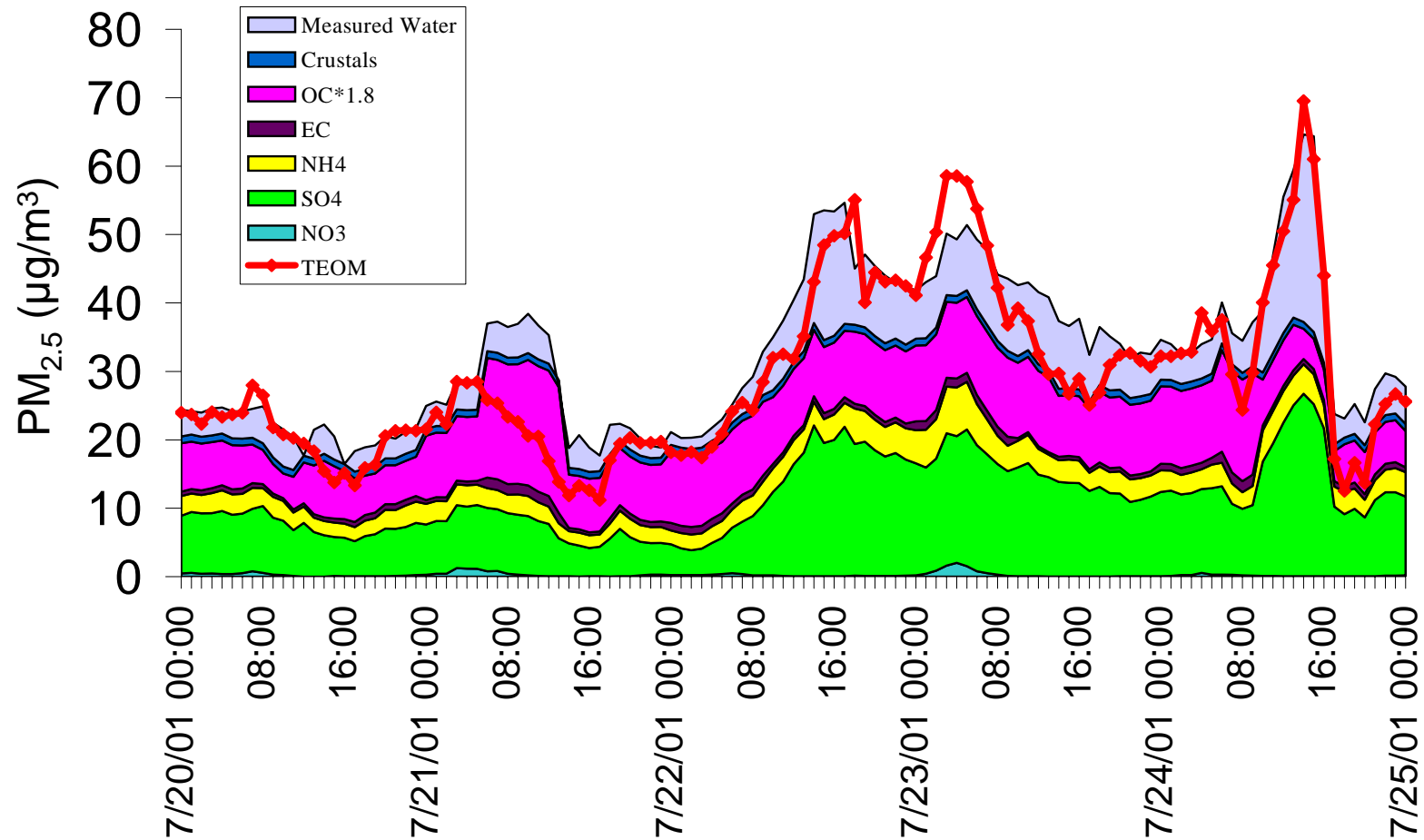
$$\frac{1}{RH} = 1 + \mathbf{B} \frac{M_{dry}}{M_{water}}$$

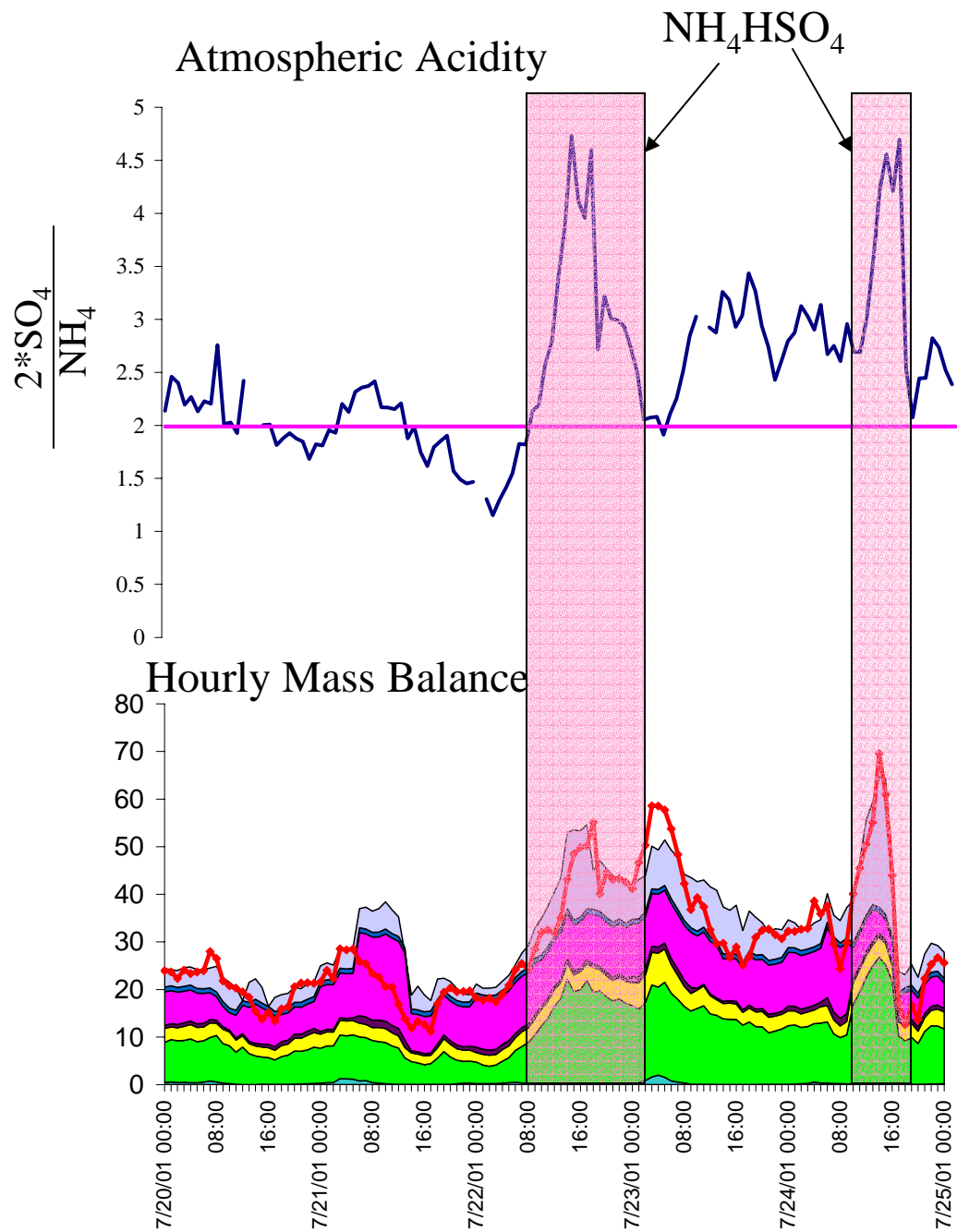
Where V_{wet} is $PM_{2.5}$ volume, M_{dry} is dry mass – sum of components, ρ_{dry} is dry density, and \mathbf{B} is $f(RH, \text{composition})$.

Hourly Mass Balance



Hourly Mass Balance With Water

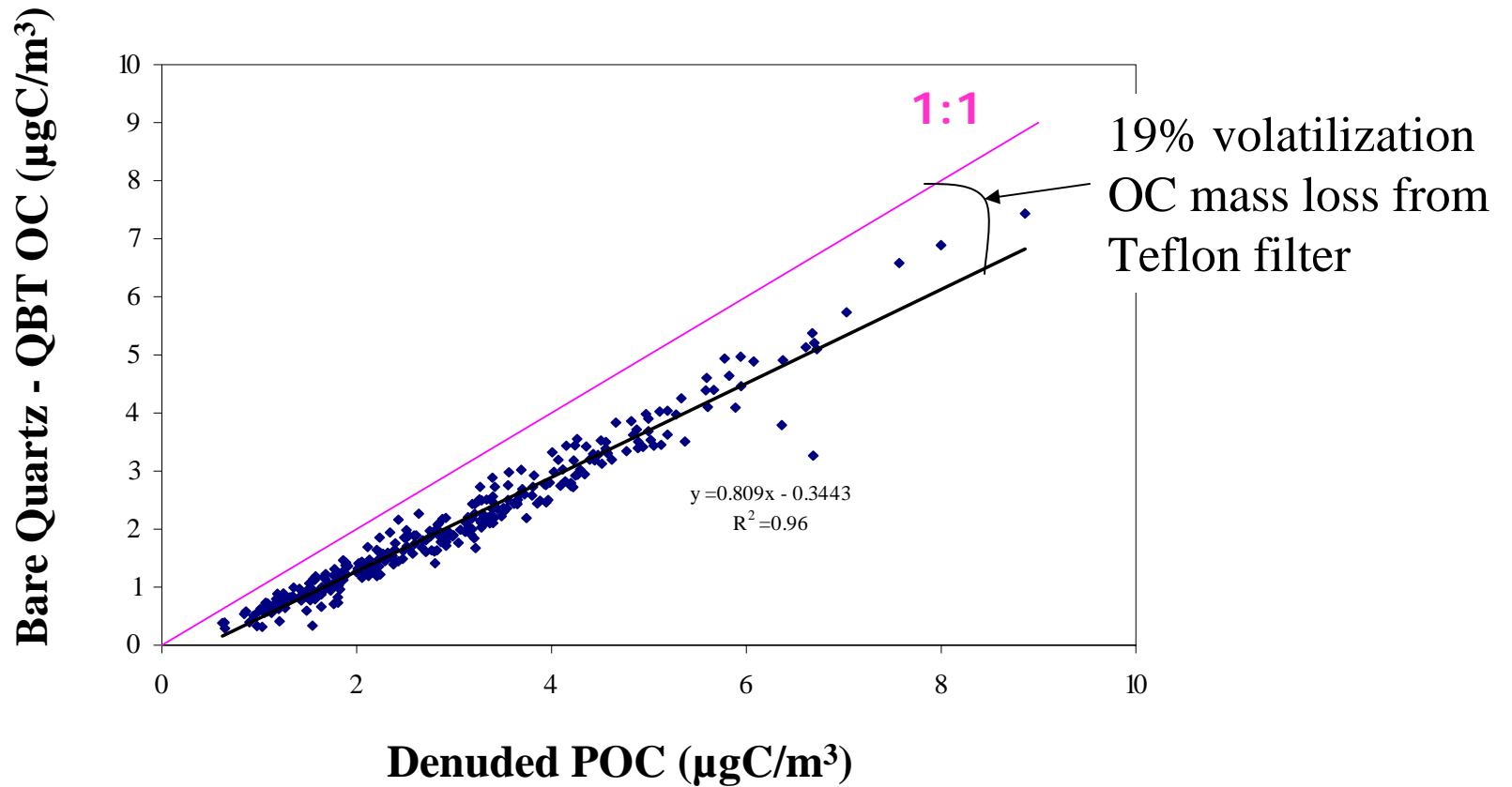




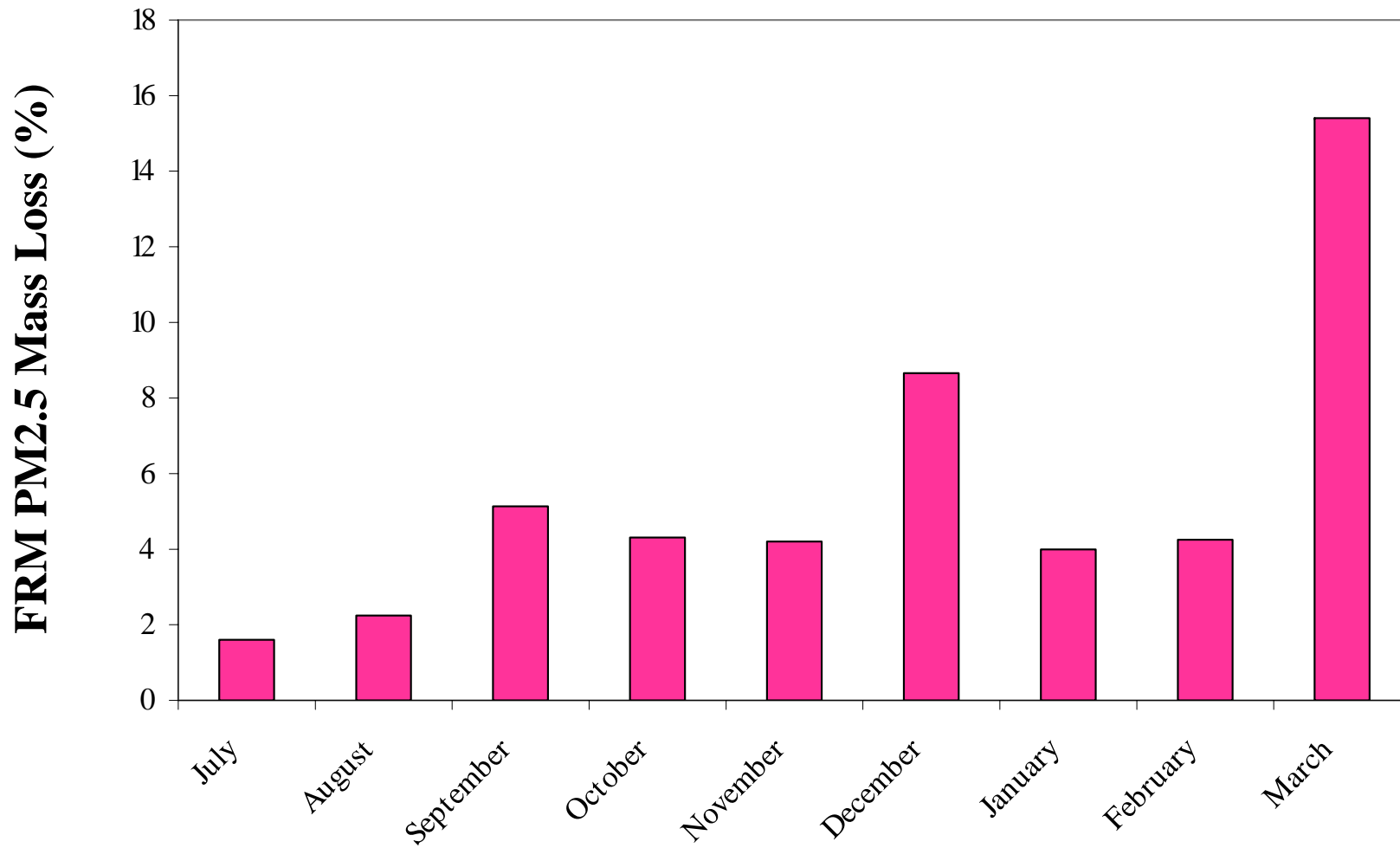
Water content seems impacted by acidity; the more acidic the atmosphere, the greater the aerosol water

Corresponds to presence of NH_4HSO_4 , a species with hygroscopic properties.

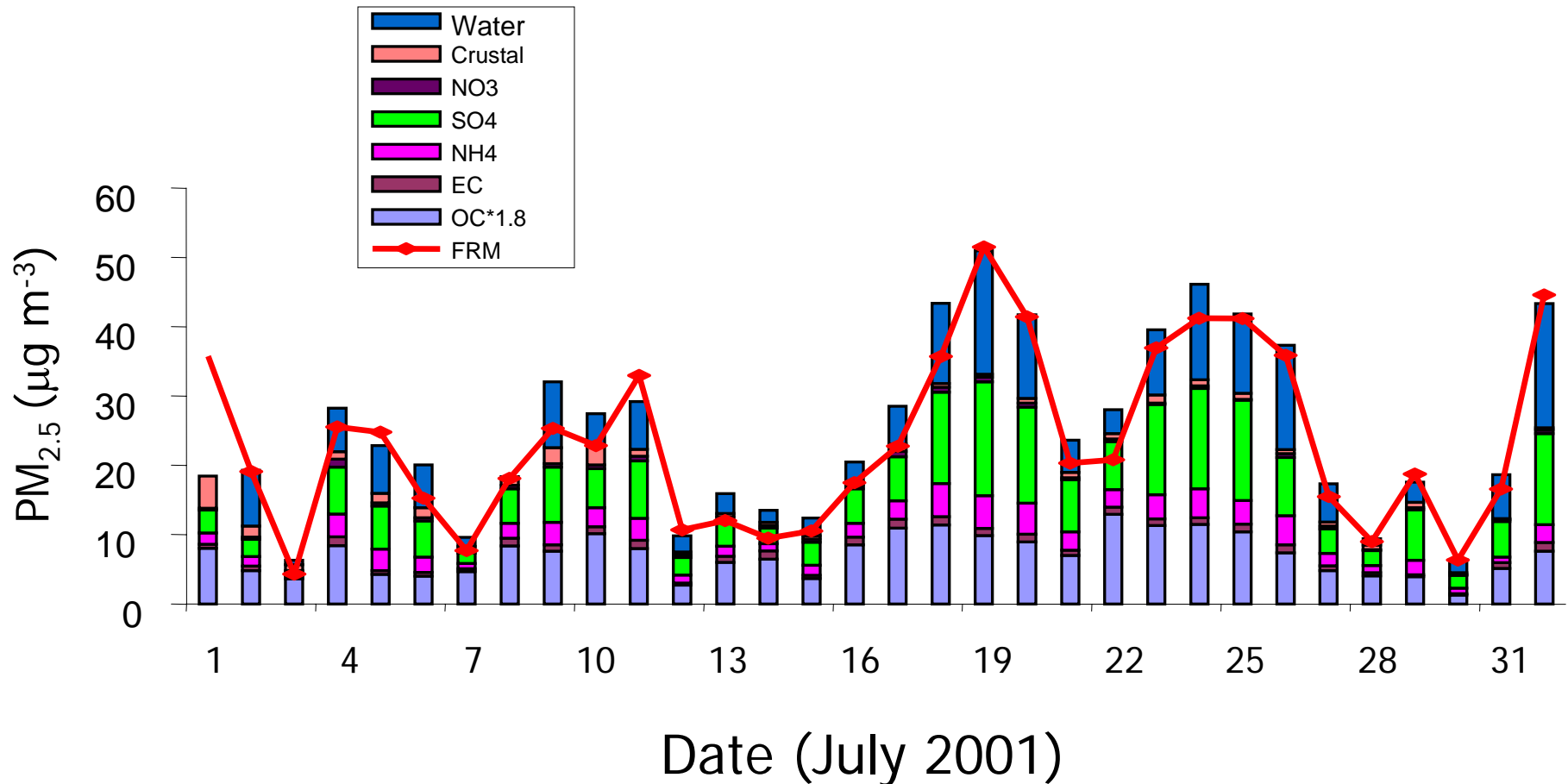
Volatilization of Organics



Mass loss from volatilization of Nitrates

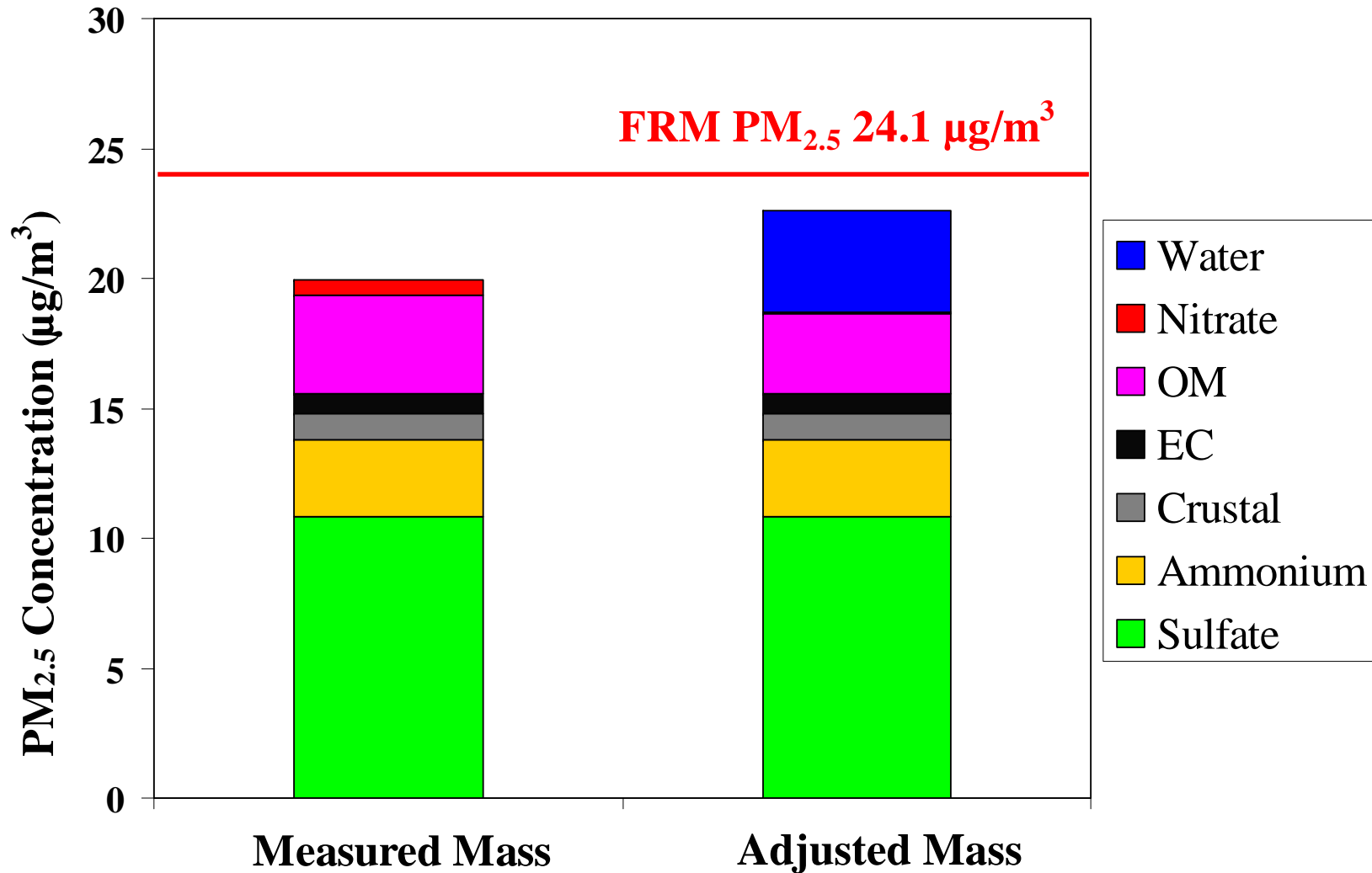


Mass Balance Closure – July 2001

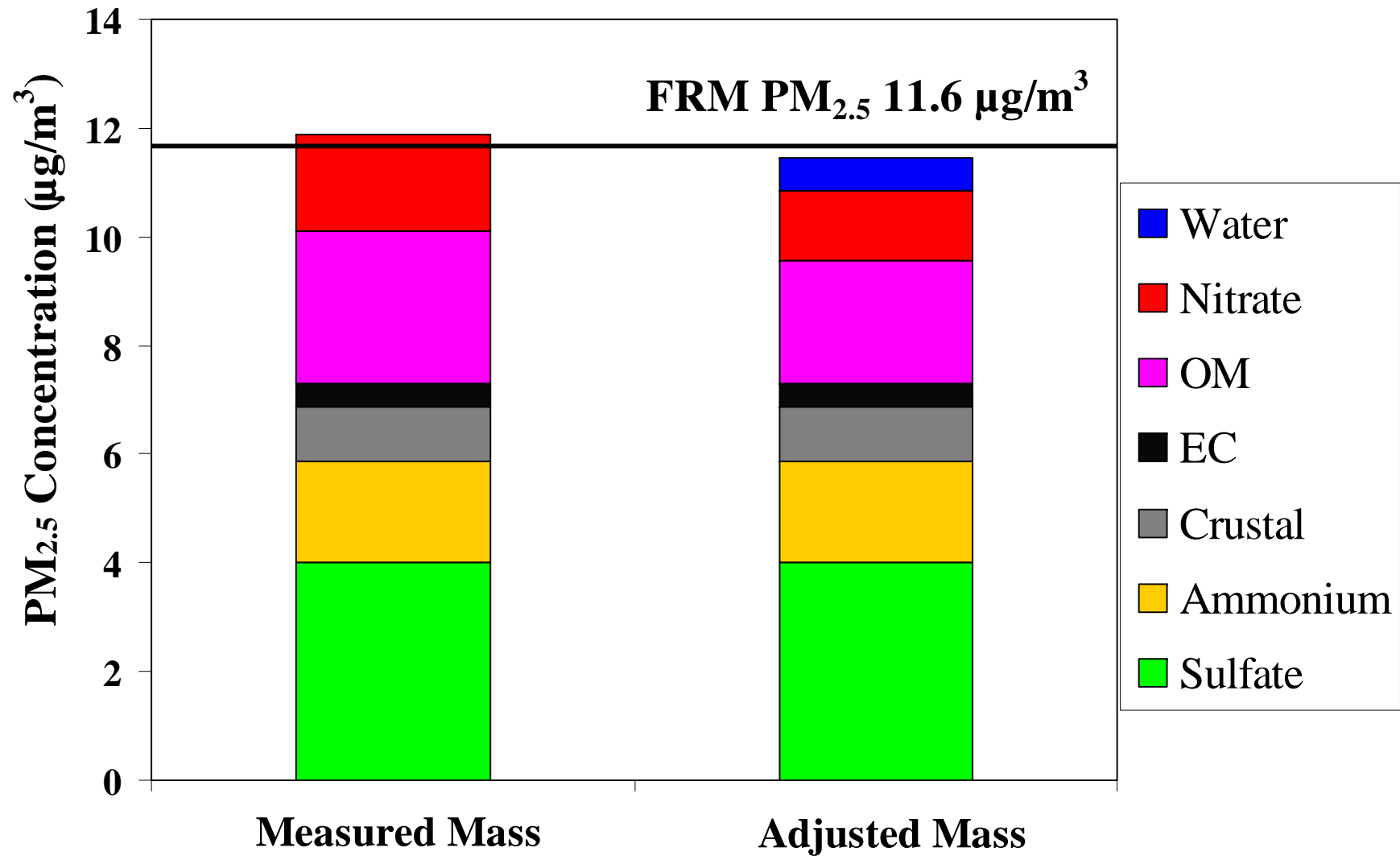


Good mass balance was achieved for the winter months

Mass Balance Closure Summer 2001



Mass Balance Closure Winter 2002





Conclusions

- Accounting for water and volatilization we can account for FRM mass in Pittsburgh
 - Water retention significant on acidic high PM days
 - Volatilization losses in winter corresponding to higher nitrate

S.L. Rees, A.L. Robinson, A.Y. Khlystov, C.O. Stanier, and S.N. Pandis, "Mass Balance Closure and the Federal Reference Method for PM_{2.5} in Pittsburgh, Pennsylvania," *Atmospheric Environment*, 38: 3305-3318, 2004.



Acknowledgements

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