



# AOI 2: Modularization of Ceramic Hollow Fiber Membrane Technology for Air Separation

#### DE-FE0031473

PI: Xingjian (Chris) Xue University of South Carolina Columbia SC 29208 Email: <u>Xue@cec.sc.edu</u>

PO: Diane R. Madden National Energy Technology Laboratory U.S. Department of Energy





#### **Objective of project**

 Develop membrane stack and module for air separation and oxygen production using ceramic hollow fiber membrane technology

#### Strategic alignment of project to Fossil Energy objectives

- Cost of Energy and Carbon Dioxide (CO2) Capture
  - Pure oxygen instead of air for combustion of power plant produces CO2, no need to separate nitrogen from down stream;
  - Reduce the cost and simplify the system for CO2 capture.
- Power Plant Efficiency Improvements
  - Pure oxygen instead of air increases efficiency of power plant;
  - Cost-effective, reliable technologies to improve the efficiency of coal-fired power plants.





## Status at beginning of project

- Single membrane fabrication and performance testing;
- Single membrane design with traditional architecture, material system, and microstructure;
- No stack/module designs with traditional single membrane cells.

Technology benchmarking (for air separation and oxygen production)

- Cryogenic distillation;
- Pressure swing adsorption;
- Ceramic permeation membrane;
  - Simple system: dense mixed conducting membranes;
  - Producing high purity oxygen from air;
  - Economically competitive technology.





- New membrane design with novel architecture, material system, and microstructure;
- Significantly reduce capital cost of membrane cell and operating cost;
- Potentially improve reliability, durability, and endurance;
- Potentially enhance performance;
- Enabling flexible up-scaling for stack/module.
- No change of project goal/objectives;
- Market need: in addition to coal-fired power plants, oxygen has wide applications in industries





- "Fabrication and characterization of an asymmetrical hollow fiber membrane for air separation and oxygen production", 4<sup>th</sup> Global Congress & Expo on Materials Science and Nanoscience, Amsterdam, Netherlands, Oct. 2018. (invited talk)
- "An asymmetrical hollow fiber membrane for oxygen permeation", Collaborative Conference on Materials Science and Technology, Beijing, China, Sept. 2018. (invited talk)
- Journal of The Electrochemical Society, 165 (13) F1032-F1042 (2018).
- Journal of Solid State Electrochemistry, 2018, 22:2929-2943.







- Perovskite  $BaCo_{0.7}Fe_{0.3-x}Yb_xO_{3-\delta}$   $\delta$  (x =0, 0.05, 0.10 and 0.15) BCF, BCFYb5, BCFYb10 and BCFYb15 (a); details of the selected 2 $\theta$  range of 30-32° (b).
- XRD pattern and Rietveld refinement of BCFYb10

	x=0.05	x=0.10	x=0.15
a (Å)	4.1042(7)	4.1153(5)	4.1367(7)
V(ų)	69.136(56)	69.698(00)	70.791(99)
GOF (χ²)	4.59	4.93	4.92
R <sub>F</sub> (%)	3.15	2.57	2.56
R <sub>wp</sub> (%)	3.59	3.77	3.82

- impurity phase  $Fe_3O_4$  generated in  $BaCo_{0.7}Fe_{0.3}O_{3-}$  $_{\delta}$  sample due to large ionic size mismatch between Ba and Co/Fe;
- Very small amount of Yb-doping effectively stabilized the cubic structure of BaCo<sub>0.7</sub>Fe<sub>0.3</sub>O<sub>3-δ</sub> to room temperature;
- Yb B-site doping shifted peak to lower angles, increased lattice parameter and cell volume.





- Sintering ability of bulk materials
  - Surface (a, b and c) and cross-section (d, e and f) SEM images of bulk BCFYb5 (a, d), BCFYb10 (b, e) and BCFYb15 (c, f) sintered at 1190 °C in air for 6 h.
  - Measurement results: Relative densities of BCFYb5, BCFYb10 and BCFYb15 pellets reached 95.72%, 93.58% and 89.21% respectively;
  - Increasing Yb content, the pellets became harder to densify and average grain size decreased.





- Sintering ability of bulk materials
  - Surface (a, b and c) and cross-section (d, e and f) SEM images of bulk BCFYb5 (a, d) sintered at 1190 °C in air for 6 h, bulk BCFYb10 (b, e) sintered at 1220 °C in air for 6 h and bulk BCFYb15 (c, f) sintered at 1260 °C in air for 6 h;
  - Yb dopant is a sintering inhibitor;
  - The competing effect of Yb inhibiting and sintering temperature leads to increased average grain size.



- Temperature dependent electrical conductivity of bulk in air
  - 150-450 °C, conductivity increased exponentially with temperatures; beyond 450 °C, increased in a little bit low rate. Arrhenius plot showed two regions with different Ea;
  - Mixed conductor: co-presence of electron holes and oxygen vacancies; high temp loss lattice oxygen and partial annihilation of electron holes; lead to observable conductivity change@450 °C;
  - Charge carriers conducted through route of strongly overlapped B-O-B bond, and Zerner double exchange process of  $B^{n+}-O^{2-}-B^{(n+1)+} \rightarrow B^{(n+1)+}-O^{-}-B^{(n+1)+} \rightarrow B^{(n+1)+}-O^{2-}-B^{n+}$ .







- Sintering temperature effect on porous surface electro-catalytic property
  - Symmetrical cells BCFYb10|SDC|BCFYb10 sintered at 1050 °C (a), 1100 °C (b) and 1150 °C (c) in air for 2 h;
  - Electrochemical impedance spectra at different temperatures in air;
  - Arrhenius plots of polarization resistance measured at 600-750 °C.



- Sintering temperature effect on porous surface electro-catalytic property
  - Symmetrical cells sintered at 1100 °C in air for 2 h: BCFYb5 (a), BCFYb10 (b) and BCFYb15 (c) cathode on SDC electrolyte;
  - Arrhenius plots of polarization resistance measured at 600-750 °C.
  - BCFYb10 demonstrated better performance: lower Rp and Ea







- Dissociation: $O_{2,ad} \leftrightarrow 2O_{ad}$ ;
- Charge transfer:  $O_{ad} + 2e^- + V_0^{..} \leftrightarrow O_0^{\times}$ ;
- Reaction order: (a) close to 0.5, primarily contributed by dissociation;
- Reaction order: (b) close to 0.25, charge transfer process;
- Surface exchange processes
  - polarization resistance vs. applied oxygen partial pressure at different temperatures, and corresponding reaction orders;
  - (a) polarization resistance associated with high frequency arc;
  - (b) polarization associated with low frequency arc;







- Surface exchange processes: dominant process for BCFYb10
  - Arrhenius plot of polarization resistance under different oxygen partial pressures;
  - (a) Ea in range of 1.04~1.07 eV for surface oxygen dissociation process;
  - (b) Ea in range of 0.089~0.092 eV for charge transfer process;
  - Oxygen dissociation is a dominant process;



- Durability test for BCFYb10 in air at 700 °C for over 120 h
  - EIS was measured intermittently during the test;
  - Ohmic resistance slightly decreased probably due to thermal aging of various bonding;
  - Polarization resistance remained relatively constant, indicating good stability of BCFYb10.

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# Solution preparation for functional layer coatings





















- Fabrications of multiple functional layers for single membranes
  - Thin functional layers;
  - Porous layer and dense layer exist alternatively.







- Fabrications of multiple functional layers for single membranes
  - Optimizations for fabrication processes;
  - Challenge: fabrication of thin film dense layer on porous functional layer.







- Fabrications of multiple functional layers for single membranes
  - Optimizations for fabrication processes;
  - Challenge: fabrication of thin film dense layer on porous functional layer.





- Further screening materials for membrane applications;
- Finish up optimizations of fabrications for multiple functional layers and single membrane cells;
- Testing and characterization of single membranes;
- Assembly of stacks with single membranes;
- Stack testing and characterizations;
- Modeling and analysis.





- Oxygen has wide applications in industries:
  - Energy (oxygen combustion/gasification, improve efficiency, enable CO2 capture, etc.);
  - Manufacturing (metal production, glass production, welding, plasma cutting, pulp and paper production, refining)
  - Environmental (water and wastewater treatment);
  - Healthcare
  - Others (chemicals, pharmaceutical and biotechnology, etc.)
- Oxygen needs are/will be intensive in these industries.
  - Technology advancement and/or Innovations are needed to fulfill these needs.
- The technology studied in this project:
  - Low cost, reliable technology for high purity oxygen production from air;
  - Has up-scaling flexibility for oxygen production at different scales.





- The technology, if successful, can be directly integrated into gasification based power plant system to achieve FE goals/objectives:
  - As an oxygen supply module integrated into the system (replace air supply unit);
  - Improve efficiency of power plant system (no nitrogen involvement);
  - Enable cost-effective, efficient, and reliable CO2 separation and capture.
- The technology can also be a stand-alone oxygen production unit/system
  - Can be scaled for oxygen production at different scales (directly transferred to market);
  - Relevant companies (Praxair, Airgas) might be interested in this technology (integrated into their oxygen production systems);
- Remaining technology challenges:
  - Fabrication process optimization for single membranes;
  - Stack assembly, testing, and characterization;
  - Modeling and analysis.



## **Concluding Remarks**



### • Applicability to Fossil Energy and alignment to strategic goals

- Low cost technology for pure oxygen production from air;
- Up-scaling flexibility (stack, module);
- Can be used as oxygen supply unit, incorporated into gasification based power plant system; (replace air supply unit)
- Improve efficiency of power plant system;
- No nitrogen involved in the system, enable cost-effective, efficient, and reliable CO2 separation/capture.

#### • Project's next steps and current technical challenges

- Keep doing what were planned in the project;
  - Single membranes: fabrication, testing, characterization;
  - Stack assembly, testing, and characterization;
  - Modeling and analysis
- Current technical challenges;
  - Technical challenges could pop-up during the course;
  - E.g., fabrication process optimization: takes longer time than planned due to complexity of process





Acknowledgments

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### Post-docs and Graduates