

HIGH TEMPERATURE GAS SENSOR FOR COAL COMBUSTION SYSTEM

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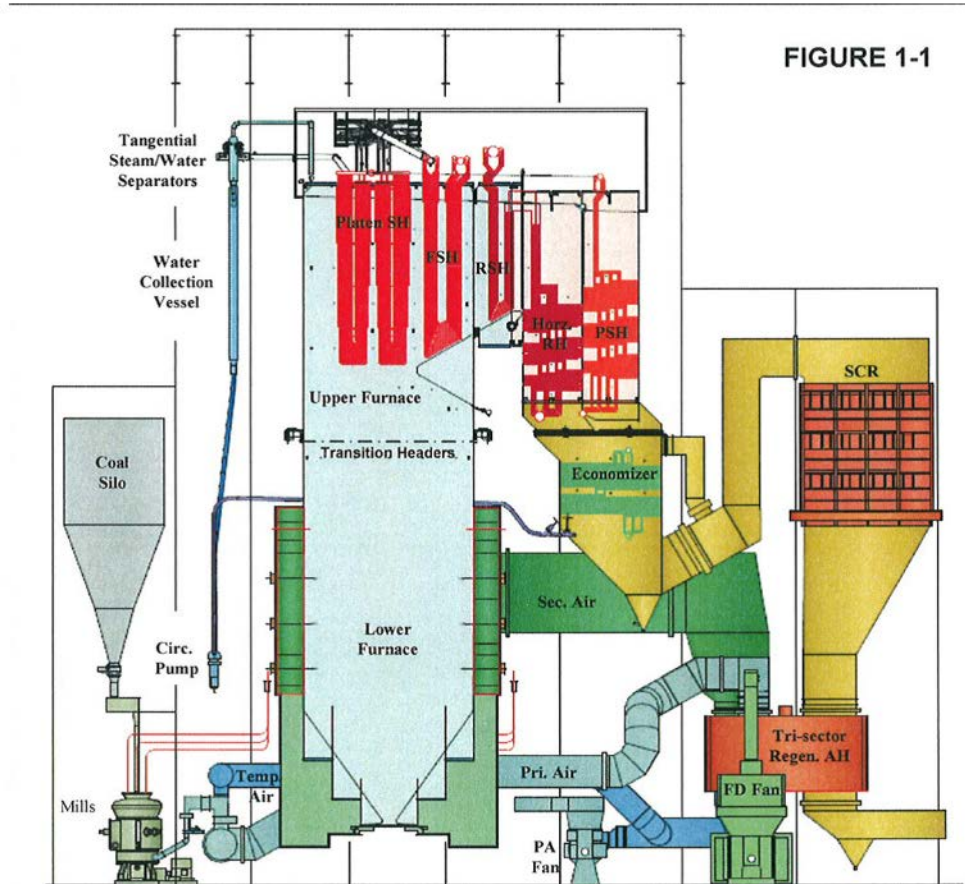
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OUTLINE

- Background & Scientific Approaches
- Project Objectives
- Project Team
- Planned Tasks & Milestones
- Research & Development Progress
- Summary

Background



Efficiency =

39%



Optimization of
combustion
process



Sensor

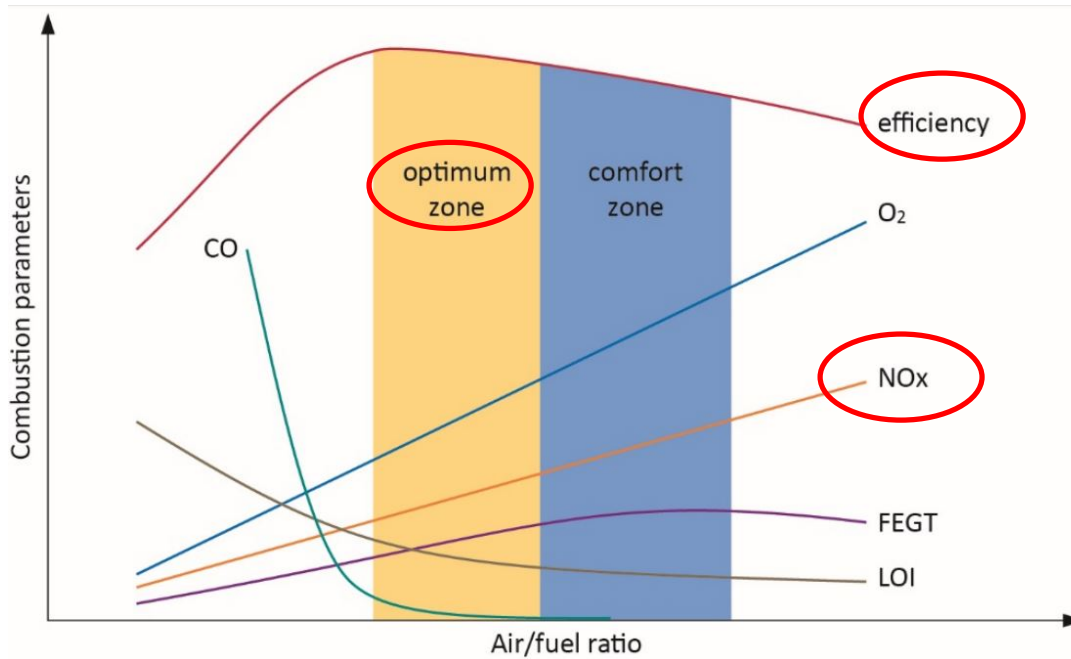


Background

Furnace	Coal flow Combustion air flow Temperature Oxygen CO Presence/quality of flame Heat flux	Electrostatic or microwave-based Pitot tubes, Venturis, thermal mass flow meters Thermocouple, IR or acoustic pyrometry, TDLAS Electrochemical cell, paramagnetic NDIR, catalytic bead, TDLAS UV/vis/IR detector, optical imaging Heat flux sensors (thermocouple or RTD-based)
Emissions monitoring and pollutant control	NO and NO ₂ SO ₂ Hydrocarbons CO Particulates NH ₃ slip H ₂ /CO ₂ /CH ₄ Limestone slurry pH Mercury Carbon-in-ash	CLD, UV photometry, electrochemical cell NDIR, UV photometer, FTIR Flame ionisation detector NDIR, catalytic bead Optical opacity UV photometry, diode laser/mid-IR absorption Thermal conductivity detector Electrochemical UV absorption Microwave-based



Background



The variation of key combustion parameters with air/fuel ratio



Background

CO₂ or NO_x sensing?

Rate of change of CO₂ is rather small at the point of optimum excess air.

CO₂ is not a very sensitive measurement.

NO₂ and NO give opposite signal to mixed-potential type sensors

CO sensing

CO is a direct measure of the completeness of combustion, unaffected by air infiltration.

Maximum boiler efficiency when the CO is between 100 and 400 ppm.

CO is a very sensitive indicator of improperly adjusted burners.

O₂ sensing?

It uses the probe should be installed close to the combustion Zone

The flow should be turbulent probe cannot distinguish leakage from excess oxygen left over after combustion.

A relatively insensitive measurement.



High Temperature Gas Sensors

GC/MS, Infrared spectroscopy, Chemiluminescent etc.

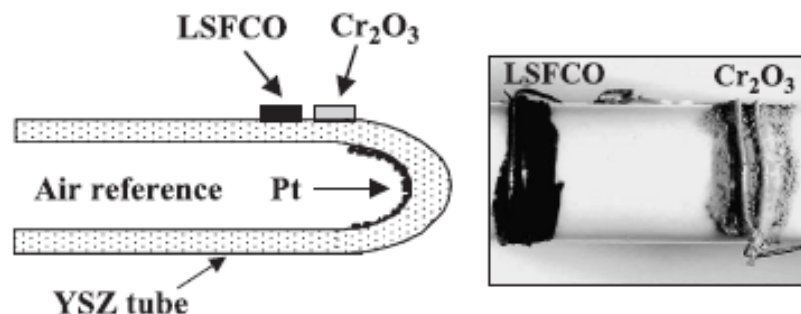
SiC-base (Schottky diode) sensors – Silicide formation

Physical properties based sensing (mass, dielectric constant, temp, surface stress etc.)

Electro-Chemical Sensors

Potentiometric

Amperometric



Requirements

In situ, online sensors

Accurate

Robust

Low cost

**Miniature and easy for
deployment**

Challenges

High temperature

Corrosive conditions

Poisoning gases

Local disturbance



PROJECT OBJECTIVES

- (1) To develop an accurate, robust, high temperature oxygen sensor based on refractory, reliable, catalytically inactive materials capable of monitoring combustion in a coal-fired plant in real time to improve combustion performance;
- (2) To investigate the feasibility and sensitivity of a new catalytic/non-catalytic sensor design to detect “oxidizable” target gases at high temperatures where other electrochemical sensors have failed;
- (3) To integrate and test the basic components of the proposed sensor in a commercial, 700 MW power plant consistent with TRL-5.

West Virginia University -

- In-depth understanding/modeling of electrochemical reaction – more accurate, better prediction & sensor material selection
- Characterization of electrochemical kinetics toward oxygen and target gas reactions.
- Testing in lab- and industrial environments

Los Alamos National Lab: Experimental Development of High-temperature Sensor

KWJ Engineering – High Temperature Sensor Packaging

Longview Power – Testing site for the High Temperature Gas Sensor

Project Team Member – Longview Power

Location	Monongalia County, near Madsville, WV
Status	Operational
Commission date	2011
Owner(s)	Longview Power
Thermal power station	
Primary fuel	Coal and natural gas
Type	Steam turbine
Power generation	
Nameplate capacity	700 MW



- Officially a "zero discharge" power plant in WV
- Includes a new air pollution control system that results in emissions that are Among the lowest in the nation for coal plants.
- Emits less CO₂ than most other coal plants because of its [fuel efficiency](#)

PLANNED TASKS & MILESTONES

I.D.	Task								
		Q1	Q2	Q3	Q4	Q5	Q6	Q7	Q8
1.0	Project Management	■	■	■	■	■	■	■	■
2.0	Sensor Development	■	■	■	■	■	■	■	■
3.0	Sensor Packaging	■	■	■	■	■	■	■	■
4.0	Lab-scale Sensor Testing	■	■	■	■	■	■	■	■
5.0	Post-mortem Characterization	■	■	■	■	■	■	■	■
6.0	Electrochemical Mechanisms Investigation	■	■	■	■	■	■	■	■
7.0	Sensor Testing in Utility Boiler	■	■	■	■	■	■	■	■

Task 1.0 Quarterly, annual, and final reports

Task 2.0 High temperature gas sensor with the temperature capability up to 1300°C

Task 3.0 Packaging for the sensor developed by LANL in Task 2

Task 4.0 Library of performance matrix for the sensor in lab-scale power plant simulator

Task 5.0 Microstructures of high temperature gas sensor after lab-scale testing

Task 6.0 Verification of the electrochemical mechanisms of high temperature gas sensing on maximum reading and temperature-proportional signal.

Task 7.0 Library of performance matrix for the sensor in utility boiler & microstructures of high temperature gas sensor after testing

DECISION POINTS:

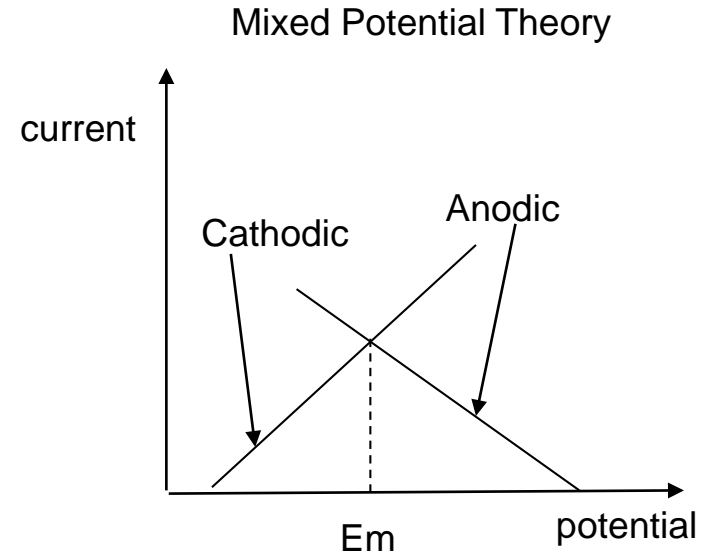
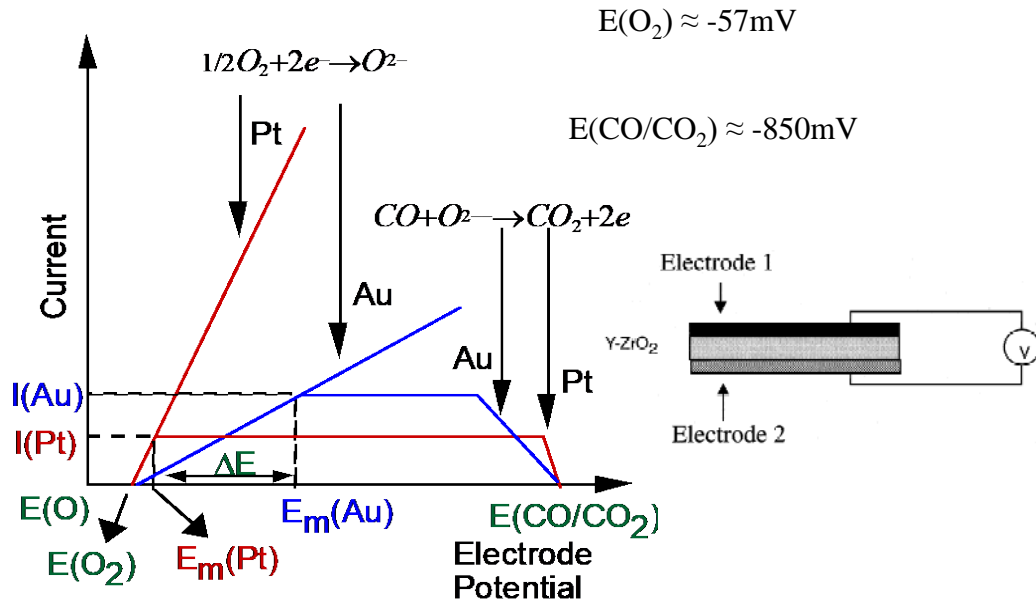
1.Q1 – Finish PMP

2.Q3 - Sensing ability (lab) ≤ 800 ppm CO concentration in a P_{O_2} range of 0.5-2% @ 1000C

3.Q7 – Sensing ability (lab) ≤ 400 ppm CO concentration in a P_{O_2} range of 1-3% @ 1000 C



Current Mixed Potential Sensors



CO electrochemical oxidation kinetics

$$i_{CO} = 2FAD_{CO} \frac{C_{CO}}{\delta}$$

Oxygen reduction kinetics

$$i_{O_2} = i_{O_2}^o \frac{4F}{RT} (E_m - E_{O_2})$$

Heterogeneous catalysis decrease CO available for electrochemical oxidation

High Temperature Electrochemical Gas Sensors

Determine reducing gas composition in a background of oxygen
Working principle



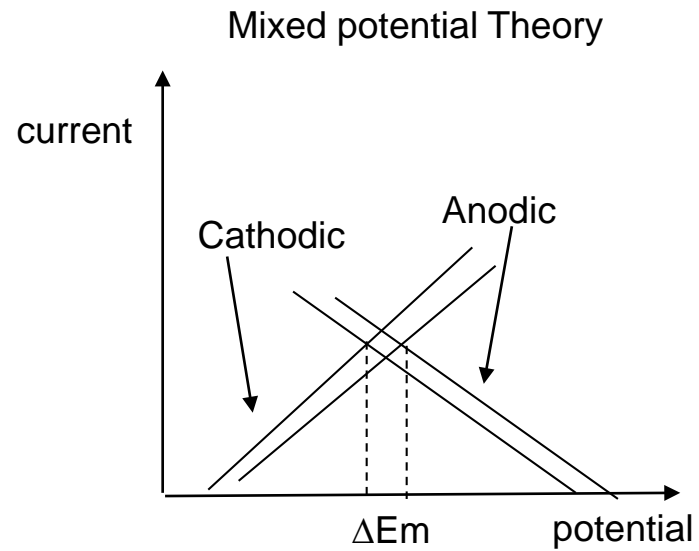
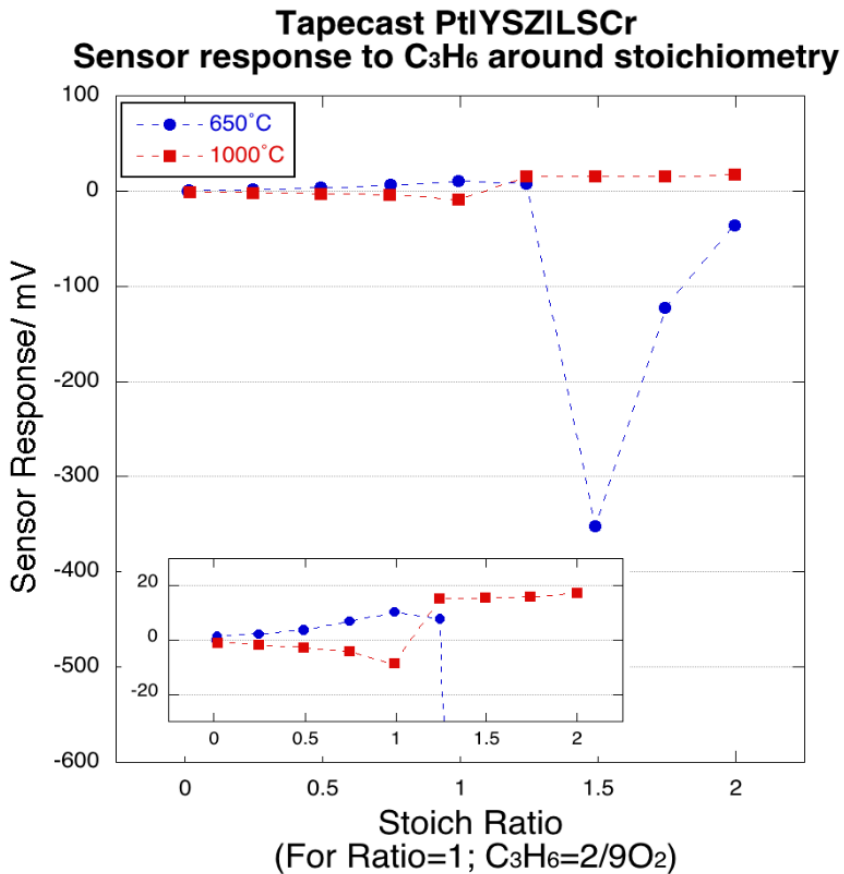
Current Mixed-potential sensors

Proposed HT sensors

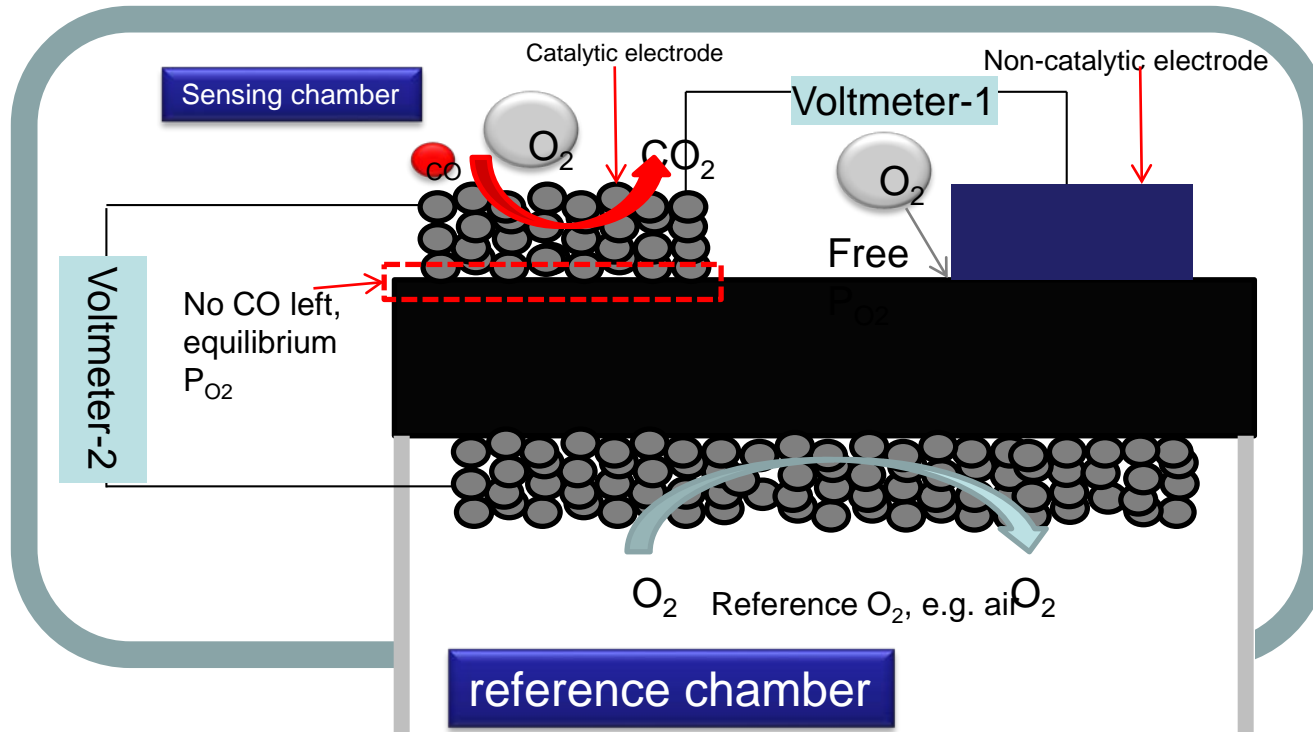
- Mixed potential sensors
- $T_{op} < 600 \text{ }^\circ\text{C}$
- High sensitivity to CO/HCs/NOx
- High durability
- Dense electrodes/Porous electrolyte

- Oxygen (Free vs Equilibrium)
- T_{op} up to $1500 \text{ }^\circ\text{C}$
- Higher sensitivity as $T \uparrow$ and $P_{O_2} \downarrow$
- High durability
- One dense and one porous electrode

Sensor Development



Sensor Development



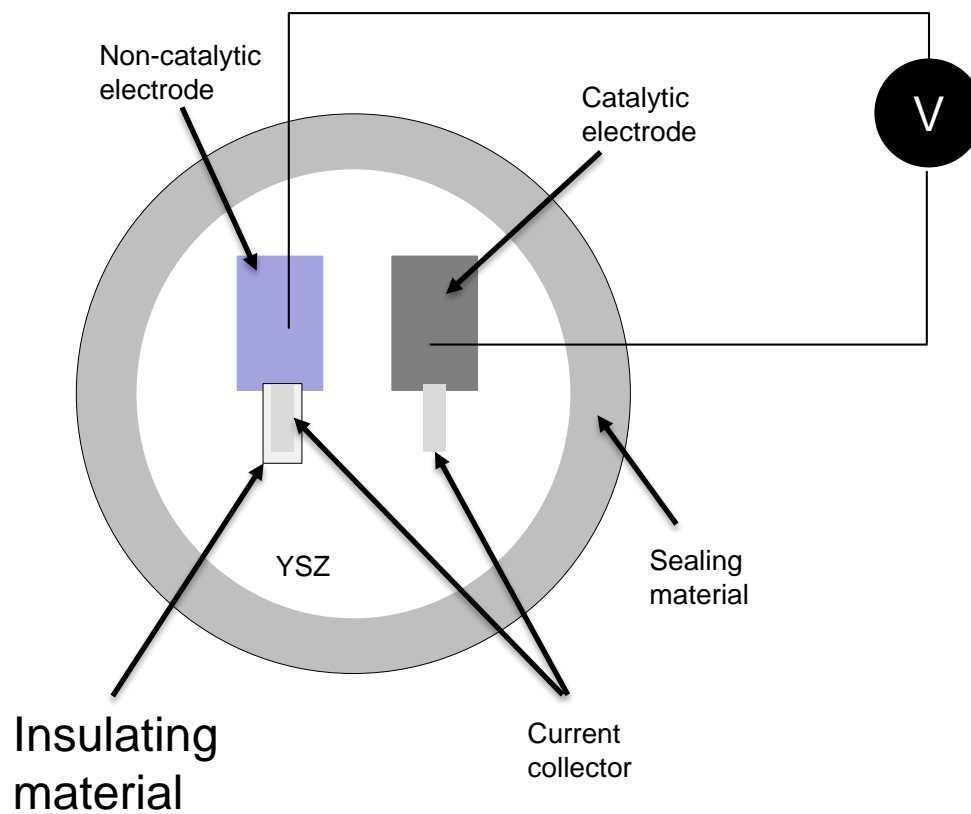
$$\Delta E_1 = \frac{RT}{nF} \ln\left(\frac{P_{O_2, Free}}{P_{O_2, Eq}}\right)$$

$$\Delta E_2 = \frac{RT}{nF} \ln\left(\frac{P_{O_2, Eq}}{P_{O_2, ref}}\right)$$



Sensor Development

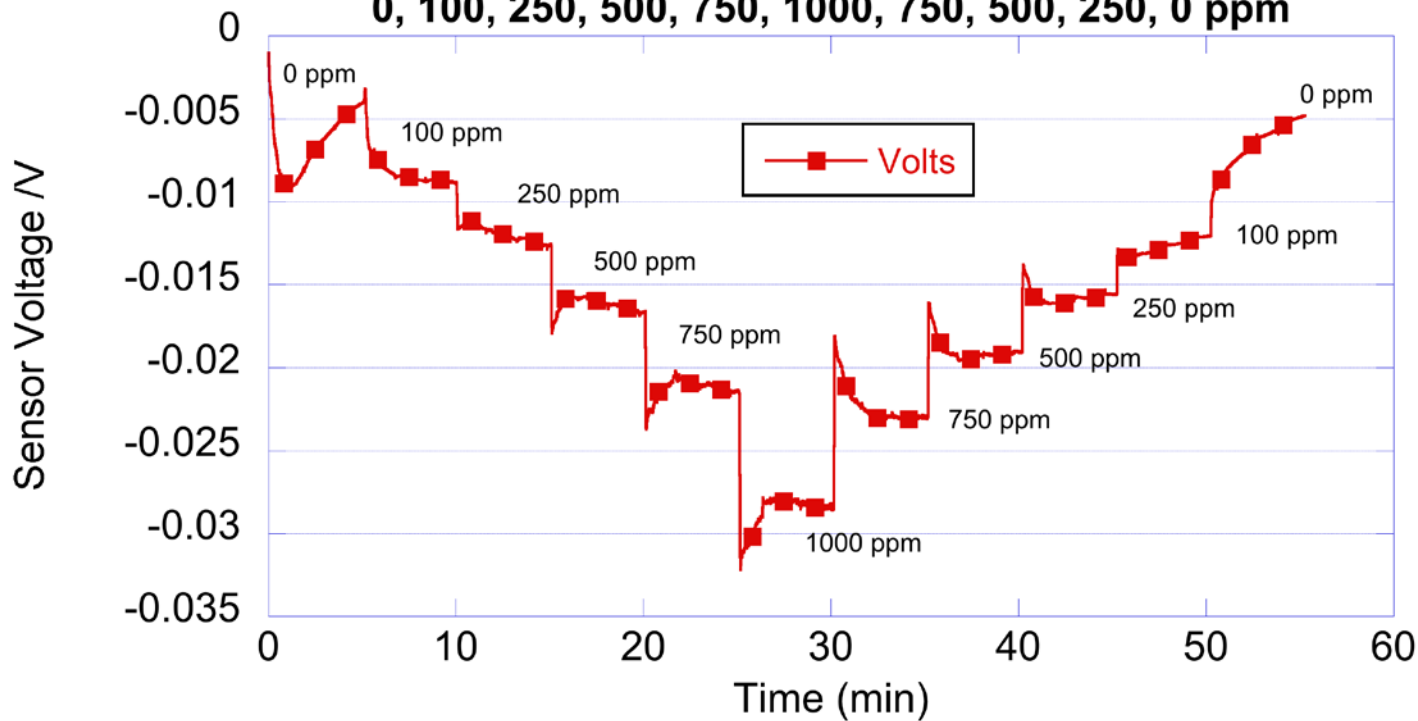
One-chamber design



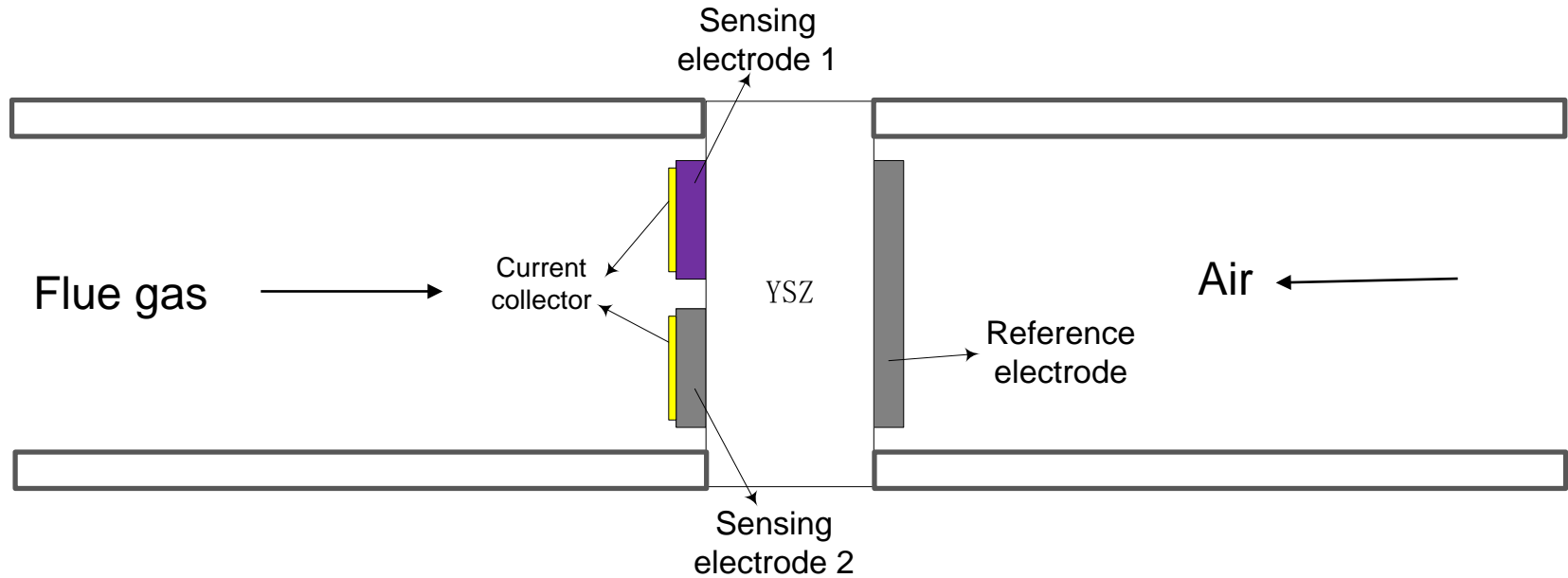
Sensor Testing – Lab-Scale

One-chamber design sensor

D-3315-65 annealed 900°C, 3 days
2.5%O₂, 50% RH, 5%CO₂, 500 sccm total flow
0, 100, 250, 500, 750, 1000, 750, 500, 250, 0 ppm



Two-chamber design



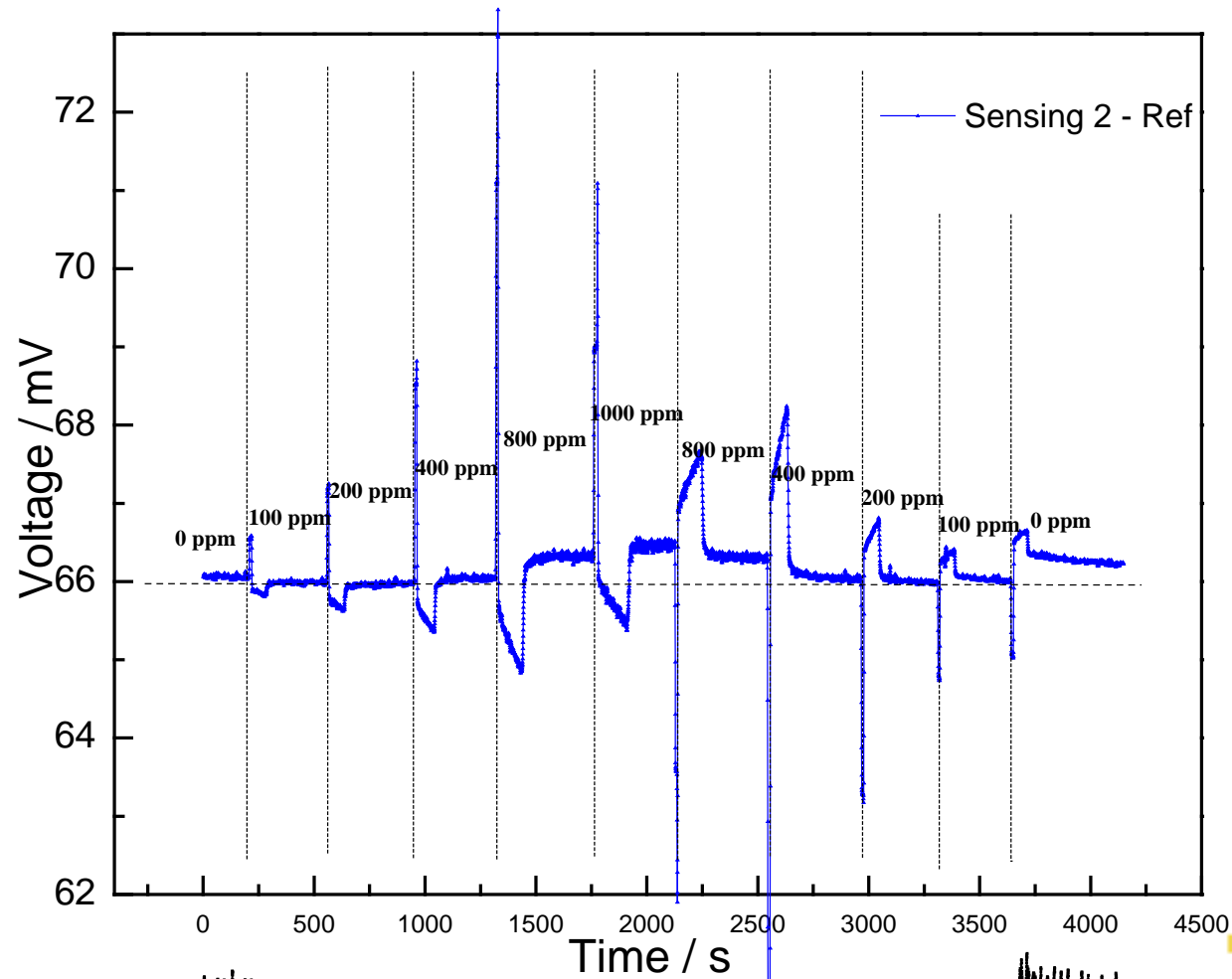
Sensor Testing – Lab-Scale

Two-chamber design sensor

– 1000 °C

- Sensing side:
- CO/N₂ Bal. || N₂
|| air
 - 200 sccm_{Total}
 - P_{O₂} = 2.1%

- Ref side:
- Air
 - 100 sccm



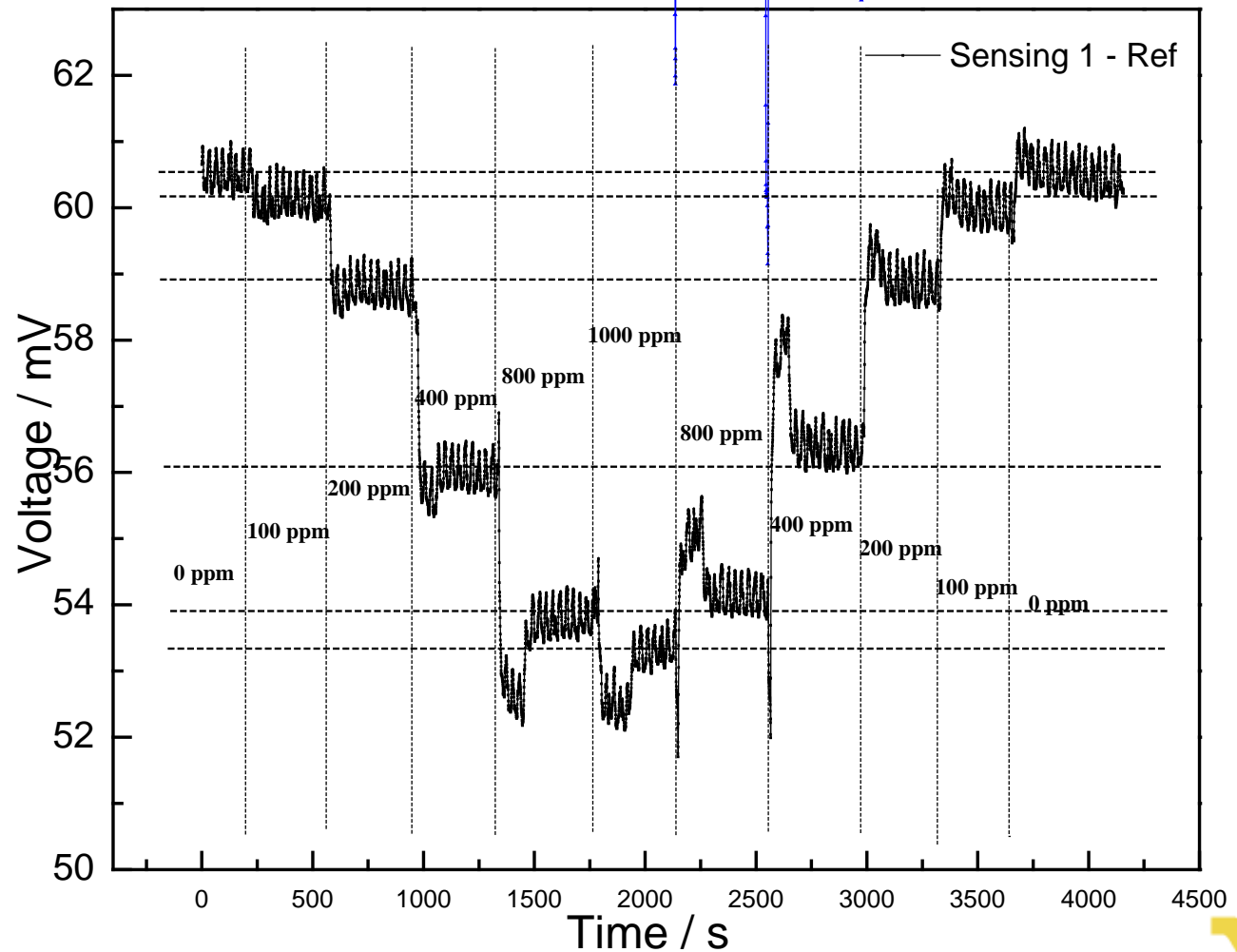
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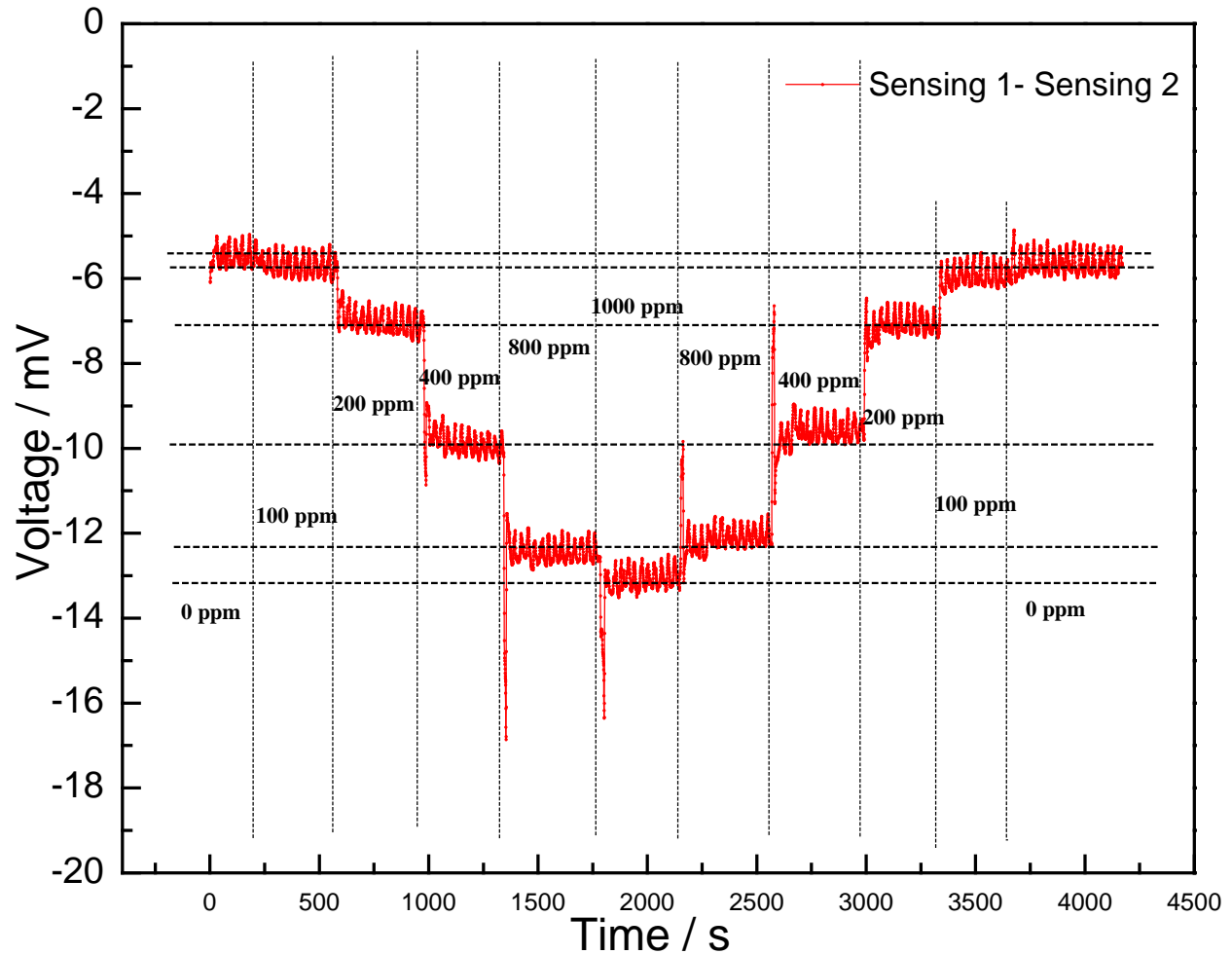
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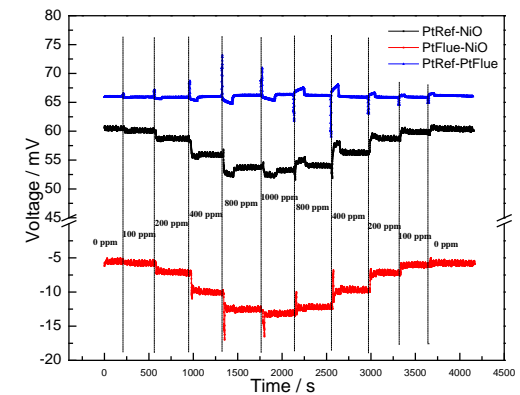
SUMMARY & FUTURE WORK

Major Progress To-Date

- Two complementary approaches have been developed
- Clearly met the 1st Go/No-Go Target

Future Work

- Continue R&D to meet the 2nd Go/No-Go Target
- Mechanisms Investigations
- Packaging Development
- Prepare for Installation in Longview Boiler



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