# Chromium Vapor Sensor for Monitoring Solid Oxide Fuel Cell Systems



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#### **Outline**

- Project Team Introduction/Description
- Background
- Technical approach
- Project objective
- Project structure
- Project schedule
- Project budget
- Risk Management
- Technology Readiness Level (TRL)



## **Project Team**

- Phase I
  - PI: Jeffrey Fergus
  - Graduate student: Moaiz Shahzad
  - Undergraduate student: TBD
- Phase II
  - Fuel Cell Energy, Hossein Ghezel-Ayagh
  - Naval Research Lab, Fritz Kub
  - University of Connecticut, Prabhakar Singh

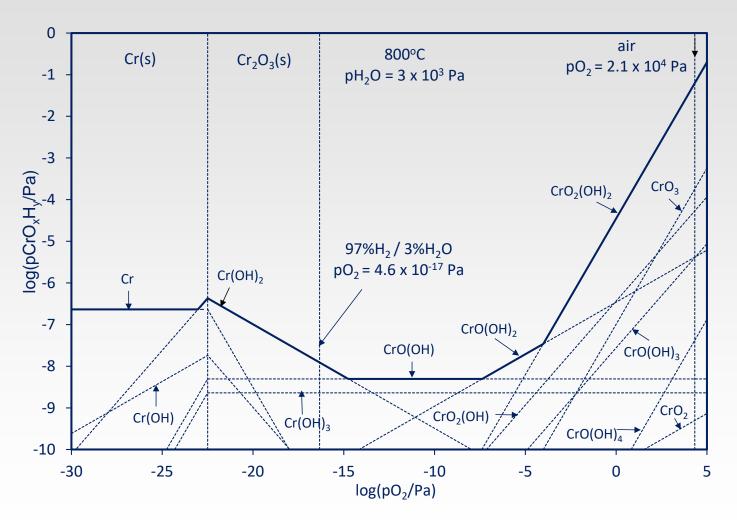


## **Background**

- Source of Chromium
  - Chromia formers used for interconnect due to high electronic conductivity of Cr<sub>2</sub>O<sub>3</sub> relative to Al<sub>2</sub>O<sub>3</sub> and SiO<sub>2</sub>
  - Oxidation of chromia scale (interconnect or balance of plant)
- Chromium Deposition
  - Cr<sup>6+</sup> reduced to Cr<sup>3+</sup> (*i.e.* Cr<sub>2</sub>O<sub>3</sub>) on cathode



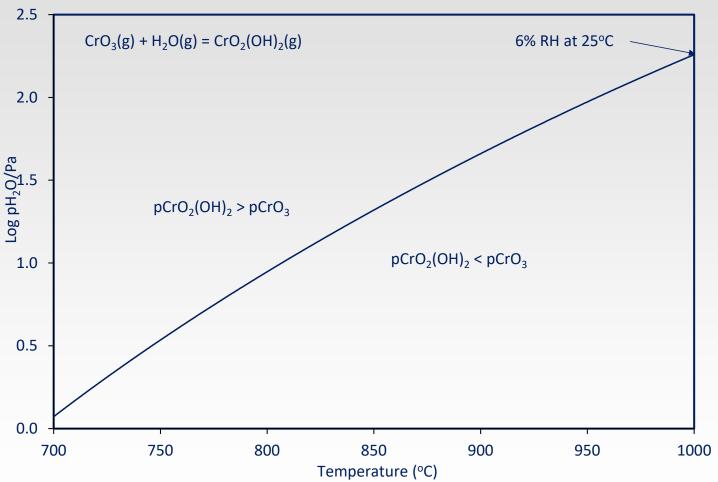
#### Cr-O-H Vapor Pressures



Vapor pressures higher in oxidizing conditions



# Stability of CrO<sub>3</sub> / CrO<sub>2</sub>(OH)<sub>2</sub>

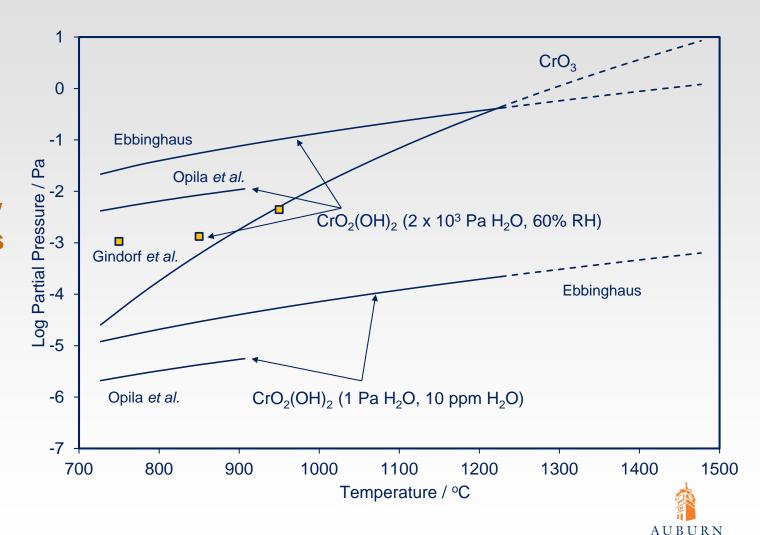


CrO<sub>2</sub>(OH)<sub>2</sub> predominant even in relatively dry conditions



# Vapor Pressure of CrO<sub>3</sub> / CrO<sub>2</sub>(OH)<sub>2</sub>

Vapor pressure of  $CrO_2(OH)_2$  high at relatively low temperatures



## **Objective**

#### Phase I

 To design, fabricate and test a chromium sensor for monitoring the chromium vapor produced during the operation of an SOFC

#### Phase II

- Evaluate the sensors in an operating fuel cell system in collaboration with FuelCell Energy
- Evaluate sensor in chromium getter system developed at the University of Connecticut.
- Develop of smaller sensors based on thin-film deposition techniques will involve collaboration with the Naval Research Laboratory.

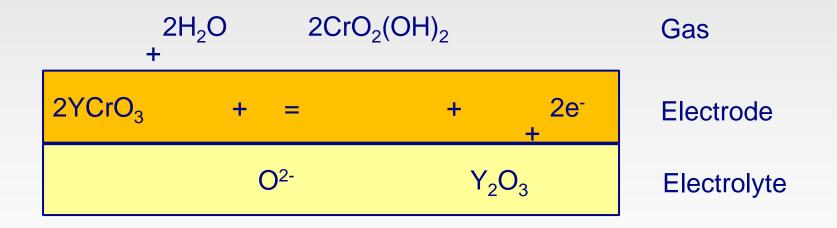
#### **Technical Approach**

- Potentiometric Chemical Sensors
  - Solid electrolyte based
  - Thermodynamic not kinetic
    - Stable
    - Not microstructure dependent
- Auxiliary Electrode
  - Relate activity of target (Cr) to that of the mobile species (O<sup>2-</sup> or Na<sup>+</sup>)
    - $Cr / O^{2-}$ :  $2Cr + 3O^{2-} = Cr_2O_3 + 6e^{-}$
    - Cr / Na<sup>+</sup>: 5Cr + 3Na<sub>2</sub>CrO<sub>4</sub> = 6Na<sup>+</sup> + 4Cr<sub>2</sub>O<sub>3</sub> + 6e<sup>-</sup>



## YSZ Auxiliary Electrode Reaction

$$2YCrO_3 + 2H_2O + O^{2-} = 2CrO_2(OH)_2 + Y_2O_3 + 2e^{-}$$





## Beta Alumina Auxiliary Electrode Reaction

$$2Na_2CrO_4 + 2H_2O = 4Na^+ + 2CrO_2(OH)_2 + O_2 + 2e^-$$

$$2H_2O$$
  $2CrO_2(OH)_2 O_2 + +$ 
 $2Na_2CrO_4 = +$   $2e^ Na^+$ 

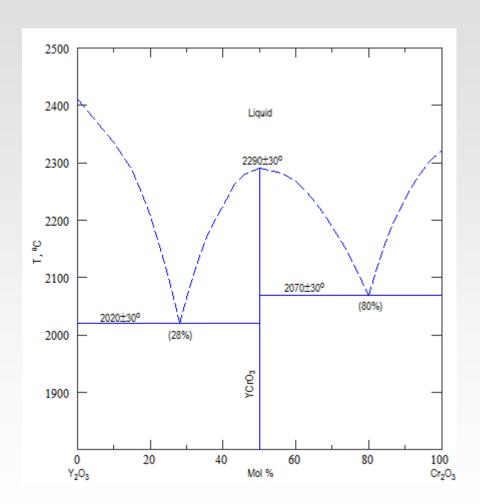
Gas

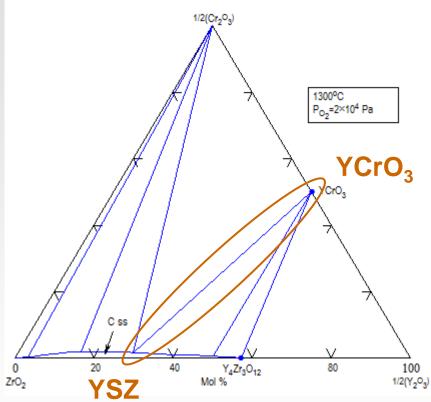
Electrode

**Electrolyte** 



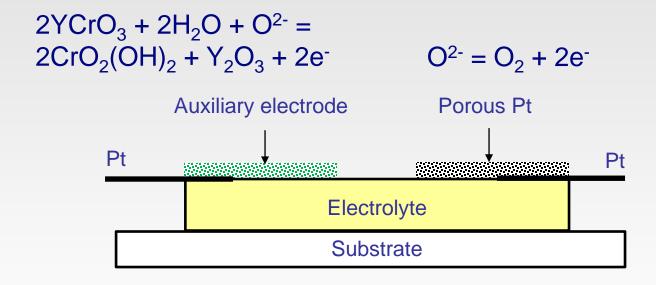
## Zr-Y-Cr-O Phase Equilibria





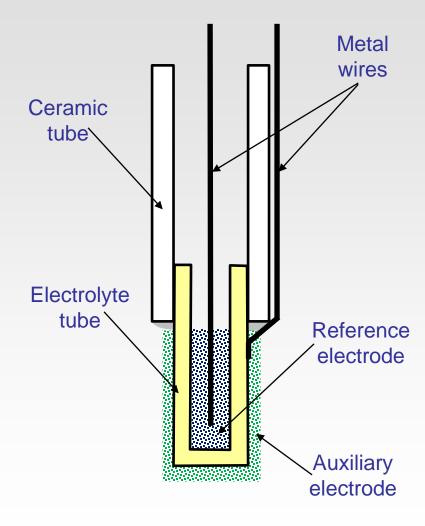


#### Sensor Schematic - Planar



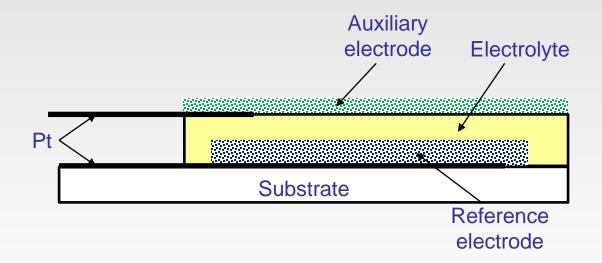


#### Sensor Schematic - Tubular





## Sensor Schematic - Thin Film (Phase II)





#### Structure - Phase I

- Development of Chromium Sensor
  - Solid electrolytes
    - Yttria-stabilized zirconia
    - Beta" alumina
  - Electrodes
    - Pt
    - YCrO<sub>3</sub>
    - Na<sub>2</sub>Cr<sub>2</sub>O<sub>4</sub>
  - Geometries
    - Tubular
    - Planar



#### Structure - Phase II

- Evaluation in SOFC System
  - Fuel Cell Energy
- Integration into chromium capture system
  - University of Connecticut
- Miniaturization using thin films
  - Naval Research Lab



#### Schedule

Activity	2016			2017												2018		
	0	N	D	J	F	M	Α									J		M
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
2 YSZ based planar sensor																		
2.1 Auxiliary electrode																		
2.2 Planar cell																		
3 YSZ based tubular sensor																		
3.1 Reference electrode																		
3.2 Tubular cell																		
4 Beta alumina -based sensor																		
4.1 Auxiliary electrode																		
4.2 Reference electrode																		
4.3 Tubular cell																		
1 Demonstrate planar sensor with YSZ electrolyte																		
2 Demonstrate tubular sensor with YSZ electrolyte																		
3 Demonstrate tubular sensor with beta alumina electrolyte																		

## Risk Management

- Suitability for application low risk
  - Solid electrolyte sensors have been demonstrated in aggressive environments
    - Oxygen dissolved in molten steel
    - Oxygen in exhaust gas
- Selective response to chromium higher risk
  - Reactions involve O<sub>2</sub> / H<sub>2</sub>O
  - Mitigation
    - Screen auxiliary electrodes for any chromium response
    - Evaluate oxygen and sodium ion conductors



## **Technology Readiness Level**

- Design similar to commercially available sensors
  - Oxygen sensor for molten steel
  - Automotive exhaust gas sensor
- Collaboration with company will be needed for scale up of fabrication and commercialization



# Questions, Comments, Suggestions?

