

*The Advanced Combustion Engineering
Research Center (ACERC)*



High Pressure Coal Combustion

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Objectives

Measure coal combustion characteristics at:

- High pressure (1-30 atm)
- High temperature (up to 1300°C)
- High heating rate (10^5 K/s)

Rationale

- No reliable data at high pressures and high temperatures
(lots of TGA data, but not much at practical conditions)
- Form of rate expression is still debated in literature
(nth order vs. Langmuir-type expressions)

Industrial Relevance

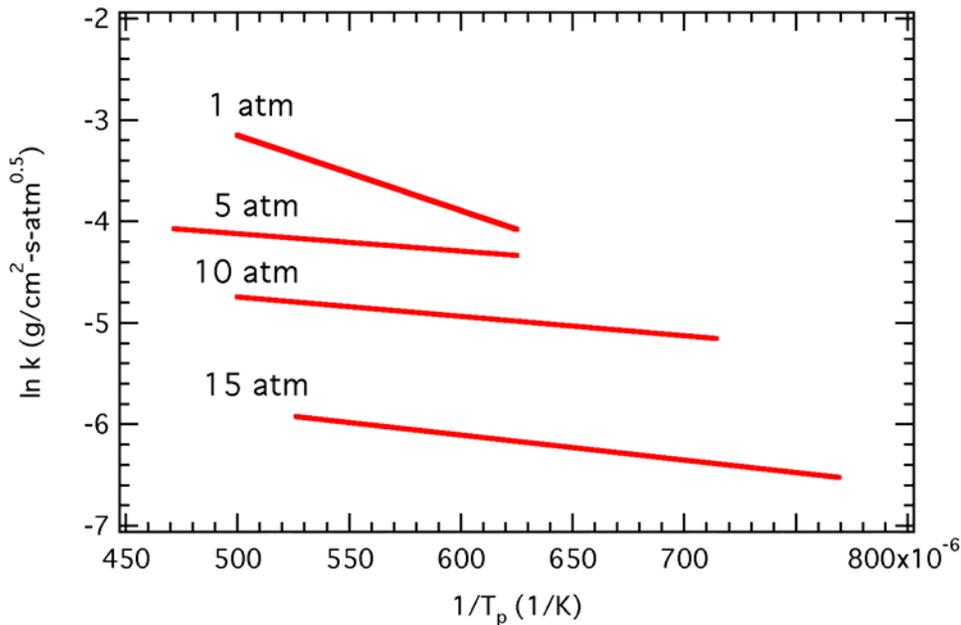
- Oxygen-blown gasifiers/combustors (IGCC and PFBC)
- Pulverized Coal Injection (PCI) into tuyeres of blast furnaces
- Enhanced oxygen combustion systems

Previous High Pressure Work at BYU

- Monson and coworkers
 - High pressure drop tube
 - Chars generated at 1 atm
 - Char reaction rates at 1-15 atm in 5-21% O₂
 - Activation energy as a function of pressure
- Hecker and coworkers
 - Chars generated in FFB at 1 atm
 - High pressure TGA work ($n = 0.7$)
- Hong and coworkers
 - Simple Langmuir rate expression with Thiele modulus
 - Explained several sets of data, including Monson's burnout data (but not measured T_p 's)

Monson Data

(Data fit with $n=0.5$)

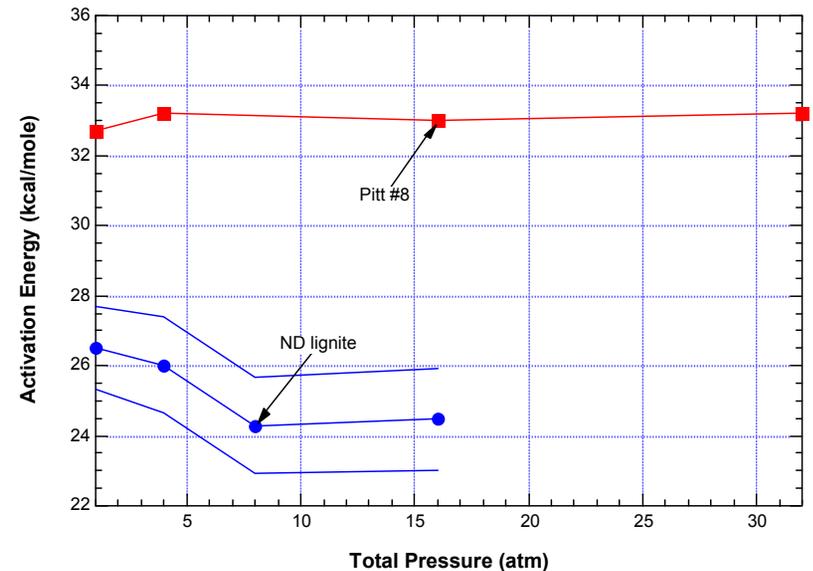
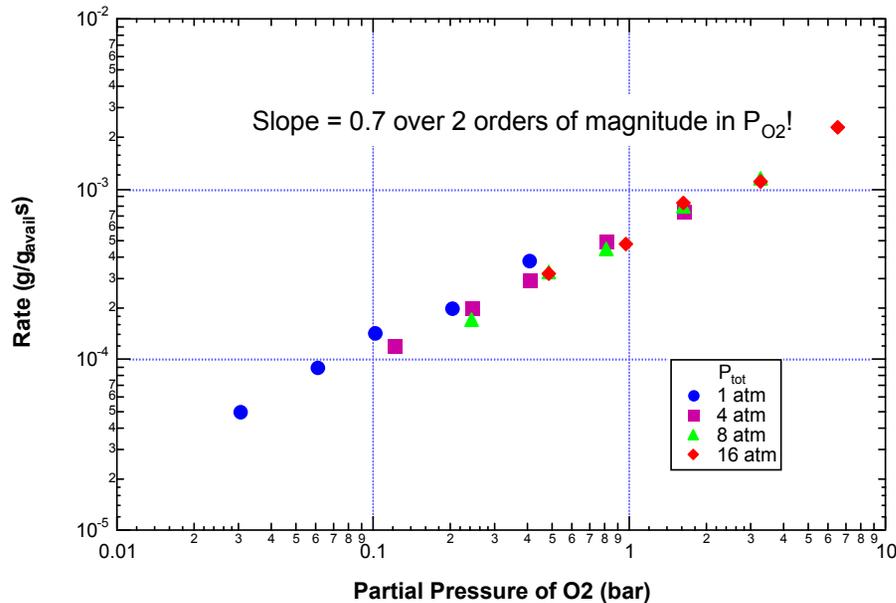


Pressure (atm)	E (kcal/mole)
1	14.8
5	3.4
10	3.8
15	4.9

The decrease in E with P_{tot} does not seem reasonable

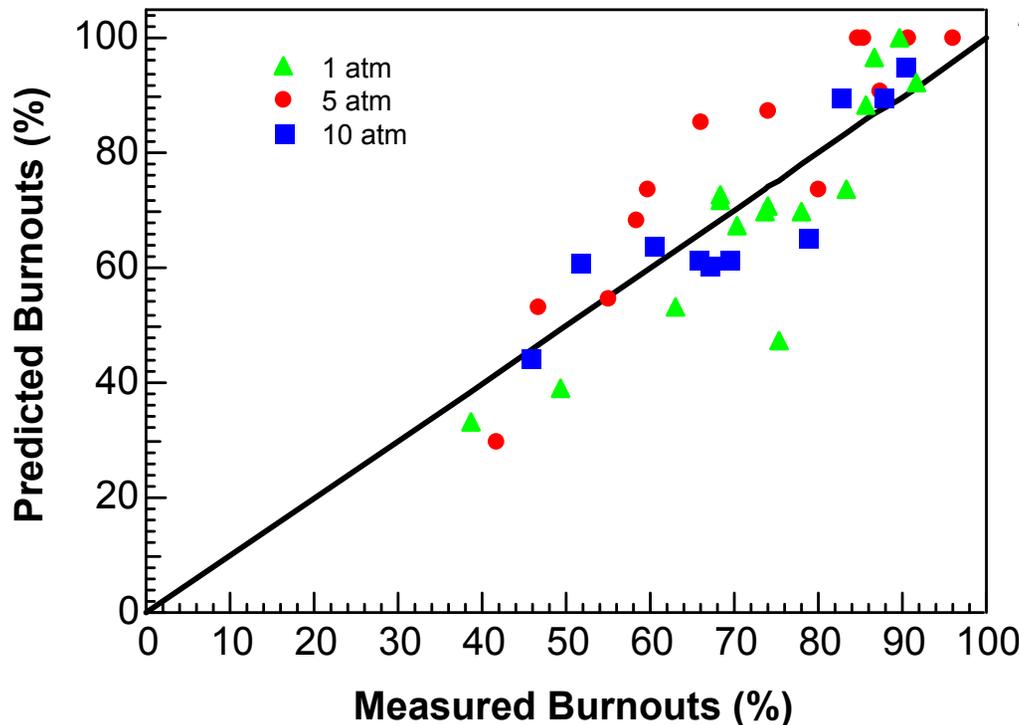
Hecker HP-TGA data

North Dakota lignite char in He/O₂ at 375°C



The HP-TGA data do not seem to support the findings of Monson

Hong modeling efforts



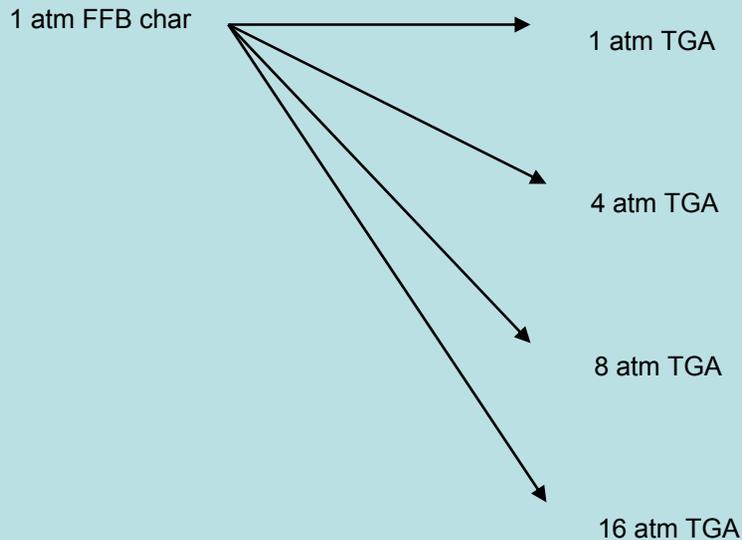
$$r_{obs}''' = \eta \frac{k_l C_s}{1 + KC_s}$$

$$M_T = L \sqrt{\frac{v_0 k_l}{2D_e} \frac{KC_s}{1 + KC_s} [KC_s - \ln(1 + KC_s)]}^{\frac{1}{2}}$$

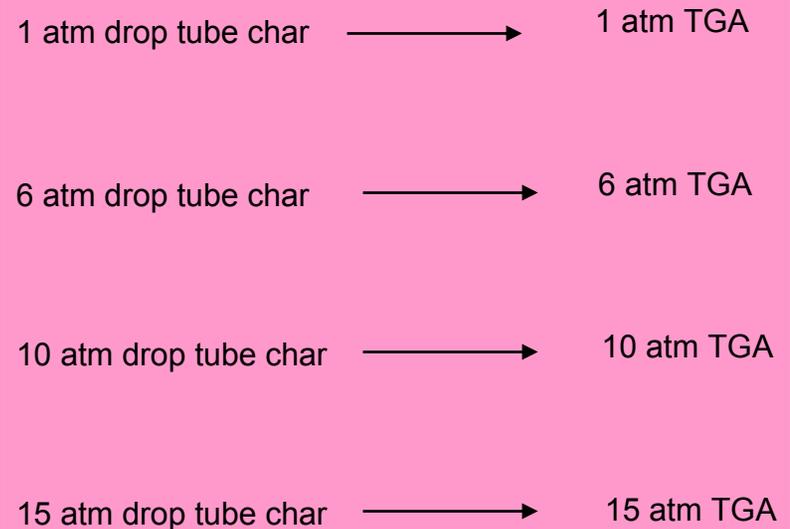
- Langmuir rate with Thiele-type modulus
- Fit five sets of data, including Monson's
- Could not explain Hecker's HP-TGA data (constant $n=0.7$)

High Temperature, High Pressure Chars

Previous Approach



New Approach

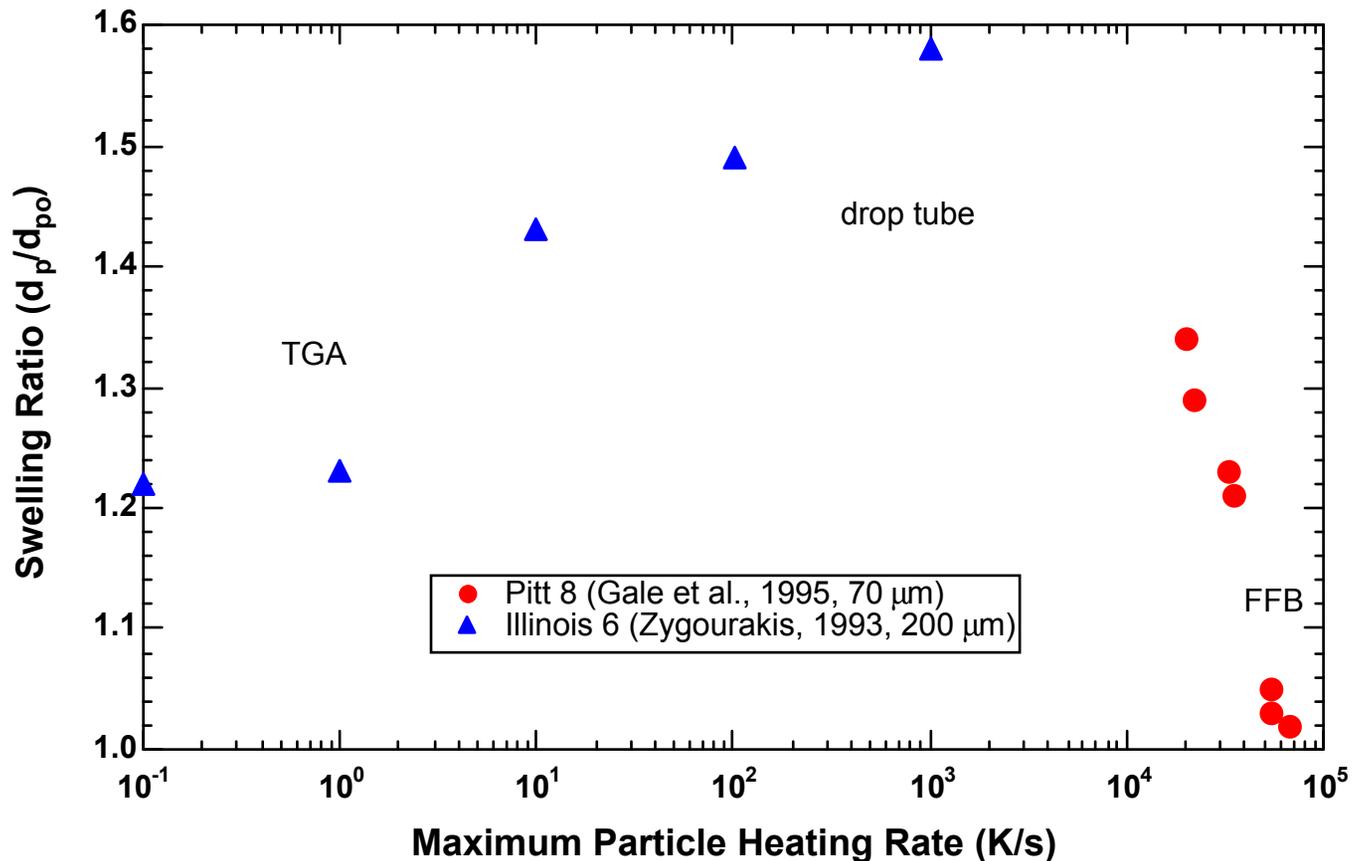


Why?

char density and diameter = $f(P_{\text{tot}}, \text{heating rate})$
intrinsic char reactivity = $f(T_{\text{final}}, \text{heating rate})??$

Swelling vs. Heating Rate at 1 atm

- Heating rate significantly affects swelling properties during pyrolysis of bituminous coals at atmospheric pressure (Gale et al., Comb. Flame, 1995)
- **High pressure experiments at moderate heating rates may show too much swelling**



Flat Flame Burner (FFB) in HP drop tube

- **Advantages:**
 - Char preparation at high temperature and heating rate (~ 1500 K, 10^5 K/s)
 - Adjust stoichiometry for %O₂ in post-flame zone
 - Fast heat-up and shut-down times for ease of use
- **Disadvantages:**
 - Fuel-rich operation at pressure produces soot from CH₄
 - Axial temperature profile not constant

Status

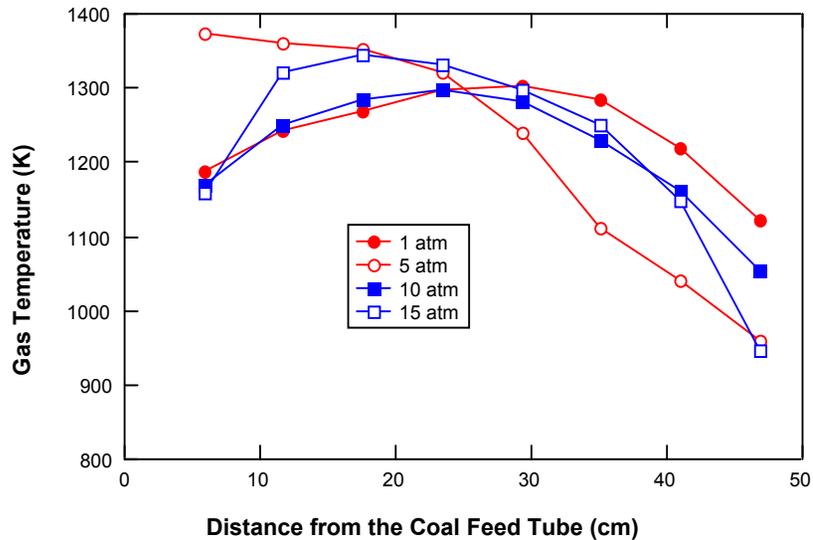
- Operational water-cooled FFB
- First set of chars produced
 - Pittsburgh #8 hva bituminous coal ($\sim 75 \mu\text{m}$)
 - Wyodak subbituminous coal (45-75 μm)
 - Knife River lignite ($< 45 \mu\text{m}$)
 - Koonfountain South African bituminous coal (63-75 μm)
- Beginning analysis of chars

Initial Tests

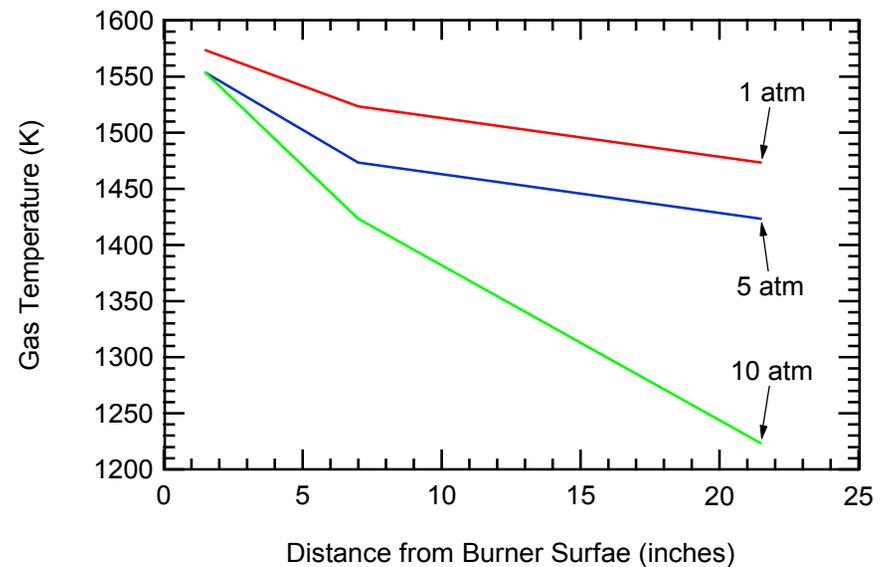
	Pressure			
	1 atm	6 atm	10 atm	15 atm
Pittsburgh #8 hva bit	x	x	x	
Knife River lignite	x	x	x	
Wyodak sub	x			
Koonfountain bit	x			

Temperature Profile Comparison of HP-FFB and HPDT

High-pressure Drop-tube

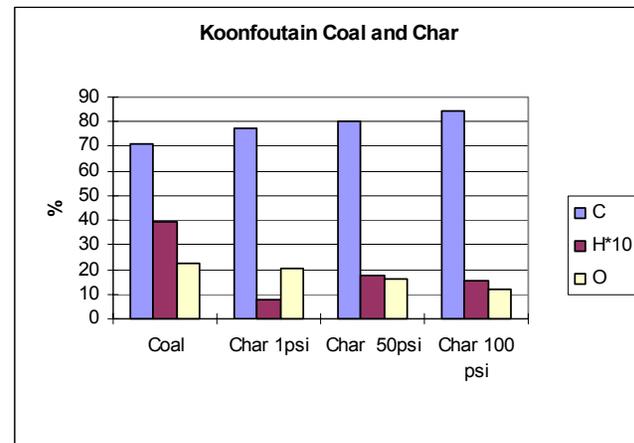
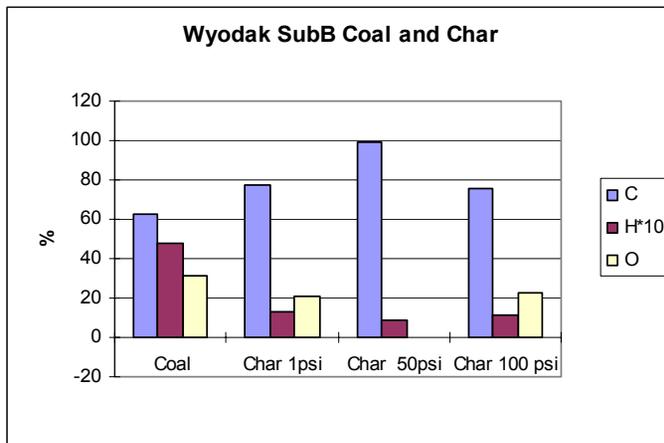
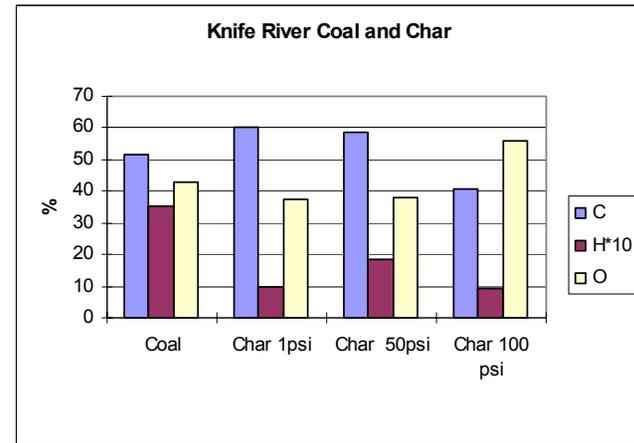
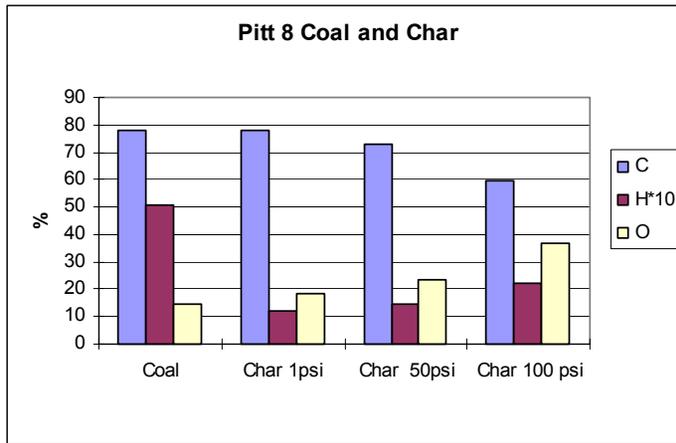


High-pressure FFB



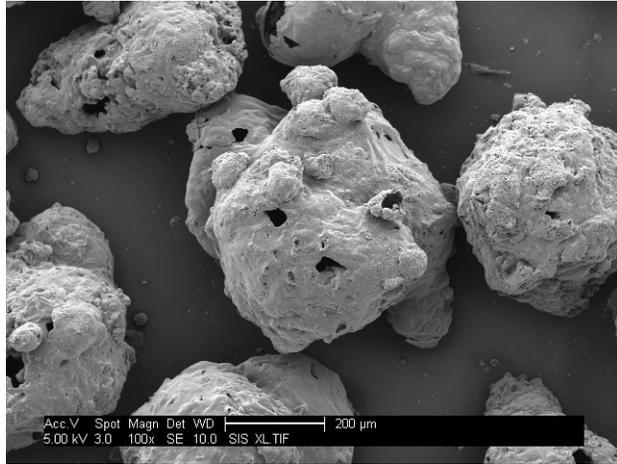
New HP-FFB has much higher particle heating rate than drop tube (10^5 K/s vs 10^4 K/s)

Elemental Compositions of Chars

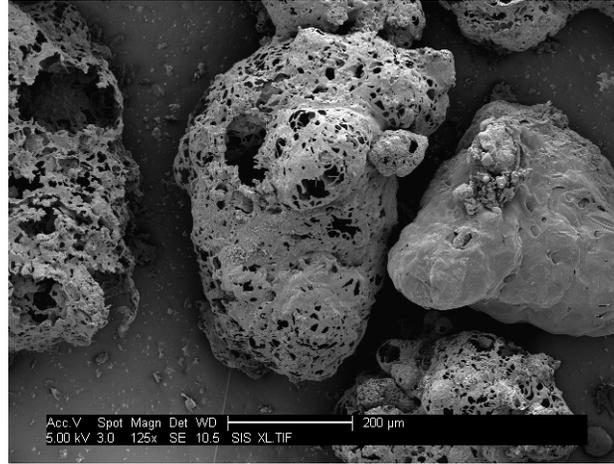


- Hydrogen content of chars is low enough to represent industrial conditions

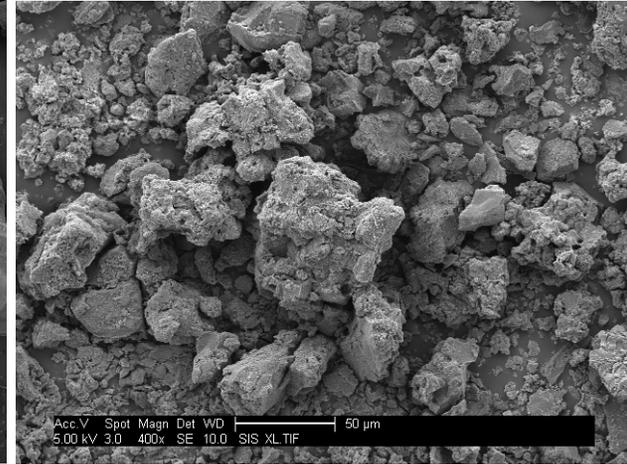
SEM Photos of Pitt #8 Chars



Pitt # 8 char (P=1 ATM)



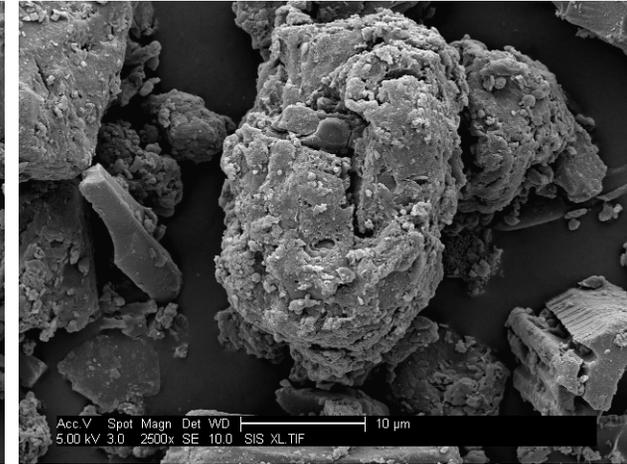
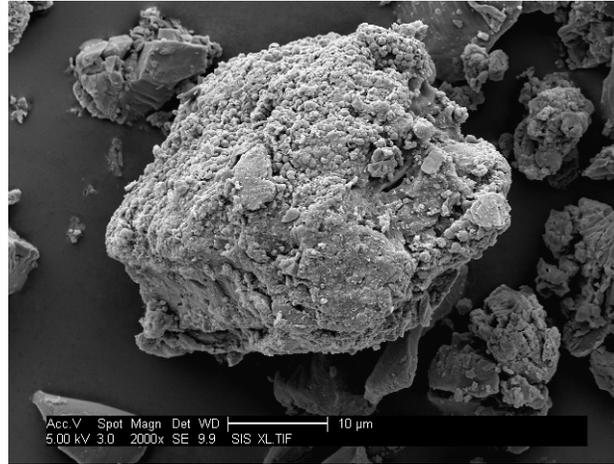
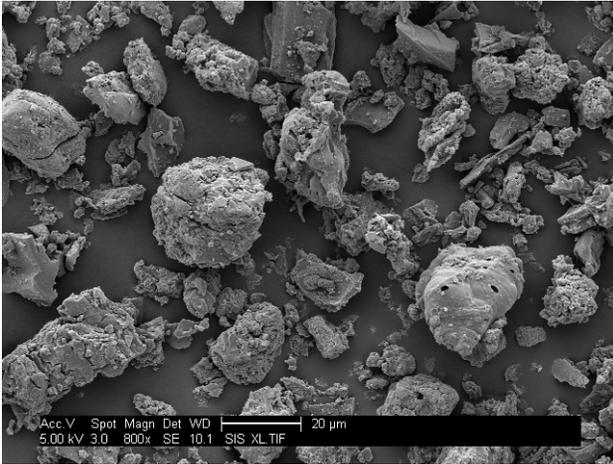
Pitt # 8 char (P=6 ATM)



Pitt # 8 char (P=10 ATM)

- Increased pyrolysis pressure increased
 - tar precursors left in char due to vapor pressure effects
 - resistance for the volatiles to transport from the interior to the exterior
- With more volatiles retained in coal particles, char made from higher pressure showed higher fluidity
- SEM photos of char made from HP-FFB, medium pressure (6 ATM) showed popped bubbles
- On 10 ATM char, pressure hold the volatile, no distinct large holes.

SEM Photos of Knife River Lignite Chars



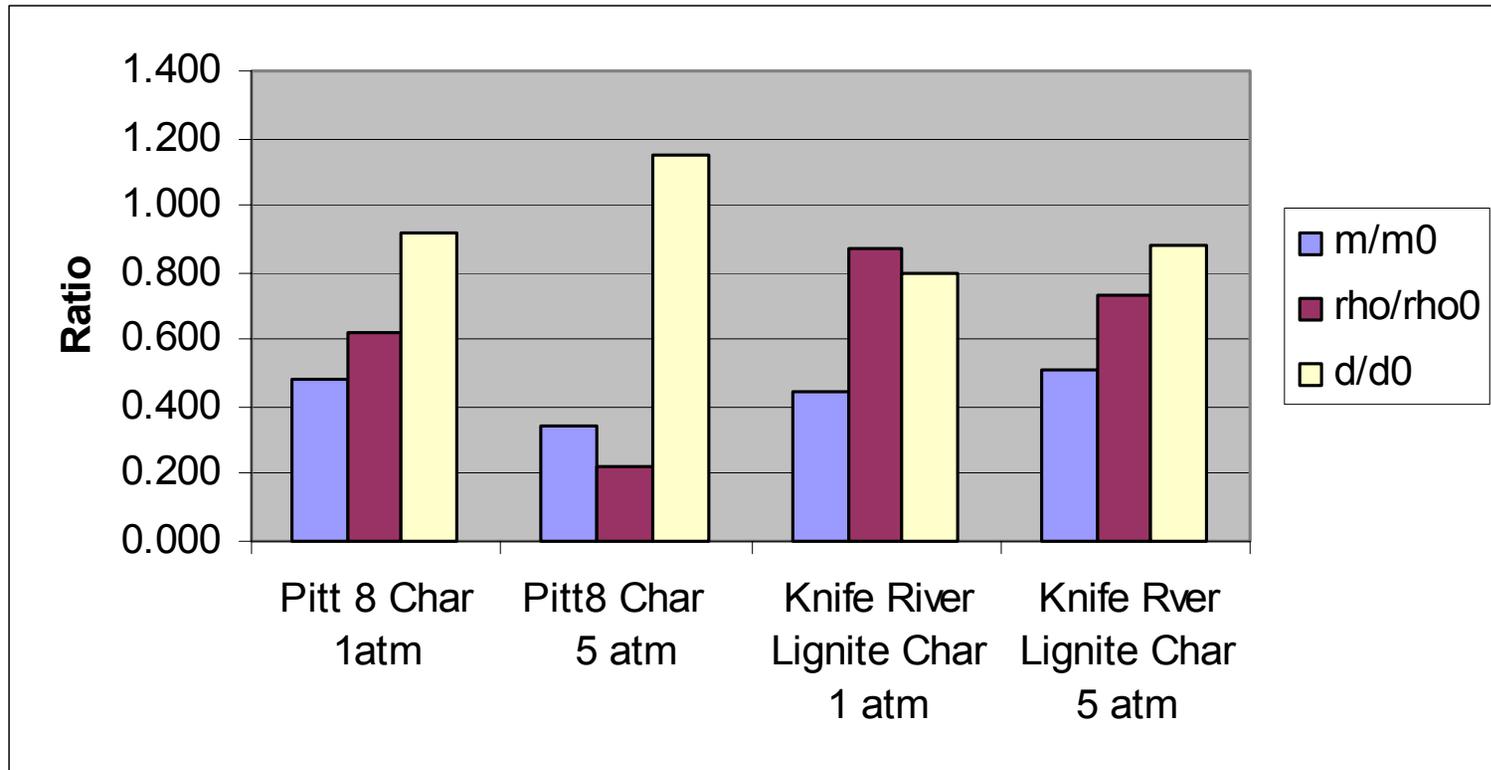
KRL char (P=1 ATM)

KRL char (P=6 ATM)

KRL char (P=10 ATM)

- No clear effect of pressure for this coal
- Effects of char formation pressure on morphology greatest for bituminous coals

Preliminary Swelling Data

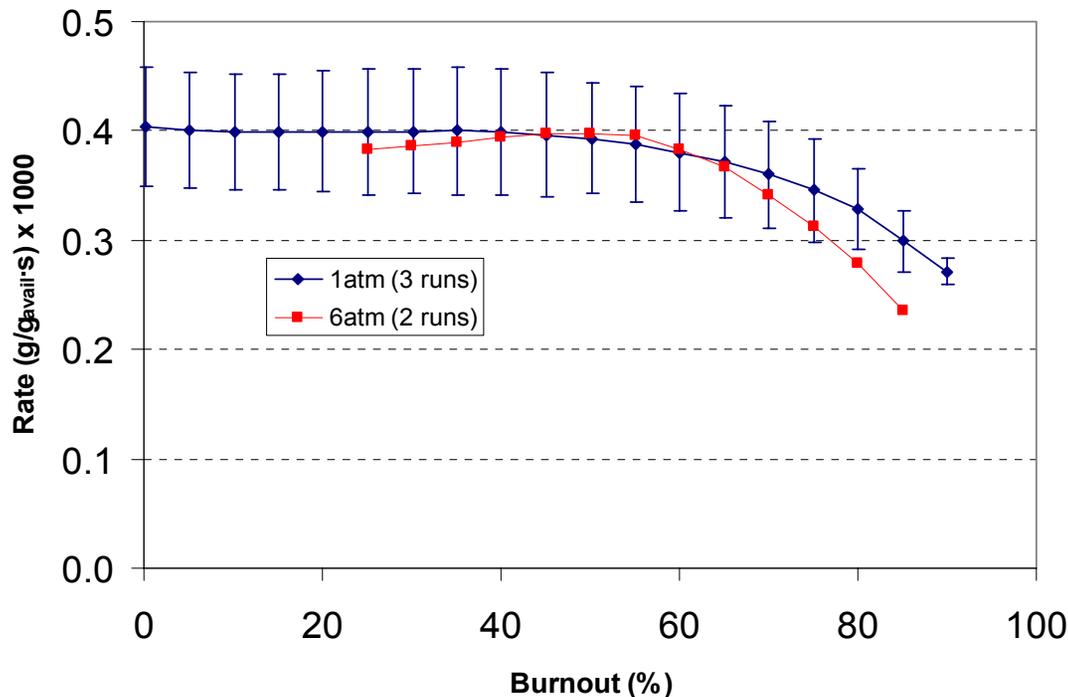


- Mass release from ash tracer (soon to use Al and Ti tracers)
- Tap densities measured, ratio eliminates packing factor
- Shrinkage indicated for 3 of 4 samples
- Increased swelling with pressure for Pitt #8 char, but not as much as in lower heating rate experiments

Pitt 8 TGA Reactivity Data

(3-5 mg samples, P_{tot} = char formation pressure)

Pittsburgh #8 Char
 $P_{\text{O}_2} = 0.32 \text{ atm}; T = 715 \text{ K}$

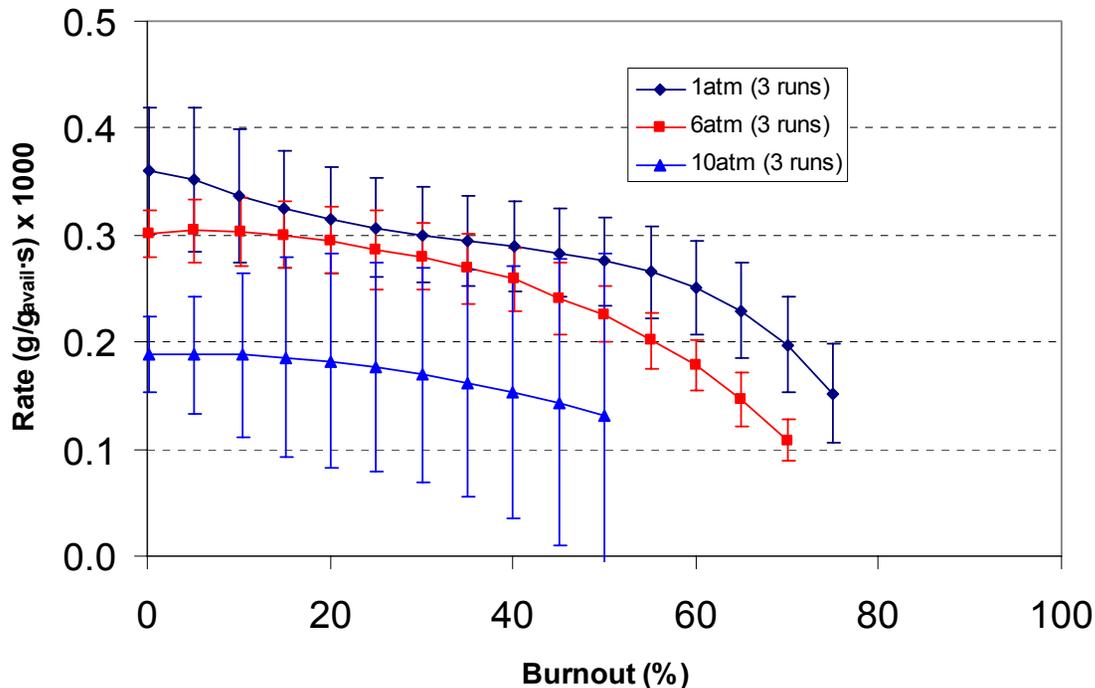


- TGA (intrinsic) reactivity relatively constant until 60% burnout
- Only late burnout reactivity changes for high pressure char

Lignite TGA Reactivity Data

(3-5 mg samples P_{tot} = char formation pressure)

Knife River Lignite Char
 $P_{\text{O}_2} = 0.28 \text{ atm}$; $T = 615 \text{ K}$



- TGA (intrinsic) reactivity not constant like the Pitt 8 char
- High pressure char has 15% lower reactivity at these conditions
- Lignite reactivity much higher than Pitt 8, so TGA temperature lowered to get intrinsic rates

The Advanced Combustion Engineering Research Center (ACERC)



- **Founded by the NSF in 1986**
(Founding Director: L. Douglas Smoot)
- **Brigham Young University & University of Utah**
- **Initial Mission:** Clean and Efficient Use of Fossil Fuels
Initial research was aimed at improving the comprehensive 3-D entrained flow coal combustion model (PCGC-3)
- **NSF base support** funding (~\$2 M/yr) for first 11 years
- **Results:**
 - Industry using CFD now
 - Many students placed in industry
 - Numerous publications and computer software products

Current ACERC Mission

***To develop advanced combustion technology
through fundamental engineering research
and educational programs
aimed at the solution of critical national problems***





ACERC Leadership

Thomas H. Fletcher
Director
Chemical Engineering
BYU



JoAnn Lighty
Associate Director
Chemical & Fuels Engineering
U of Utah



Larry Baxter
Associate Director
Chemical Engineering
BYU





Current Research Interests

Industries

- Electric power
- Chemical processes
- Forest products
- Transportation
- Cement
- Glass
- Military (safety)

Fuels

- Coal
- Biomass
- Black liquor
- Petroleum coke
- Live fuels
- Natural gas
- Jet & diesel fuel
- Metallurgical coke
- Soot
- Foams
- Model hydrocarbons

Subjects

- Organic reactivity
- Ash
- Soot formation
- NO_x, SO_x
- Heat transfer
(convective, radiative)
- CFD
- Parallel computing



22 Professors Involved

- | | | | |
|----------------------|------|-------------------|------|
| • Calvin Bartholomew | ChE | • Warren Lucas | CE |
| ♠ Larry Baxter | ChE | • Mardson McQuay | ME |
| • Merrill Beckstead | ChE | • Justin Peatross | Phys |
| • Jeffrey Bons | ME | • Dave Pershing | VP |
| • Mark Clement | CS | • Ron Pugmire | ChFE |
| ♠ Eric Eddings | ChFE | • Adel Sarofim | ChFE |
| ♠ Tom Fletcher | ChE | • Doug Smoot | ChE |
| • Fernando Fonseca | CE | • Quinn Snell | CS |
| • Hugh Hales | ChE | ♠ Dale Tree | ME |
| • Bill Hecker | ChE | • Brent Webb | ME |
| ♠ JoAnn Lighty | ChFE | | |
| • Matt Linford | Chem | | |

♠ = Executive Committee



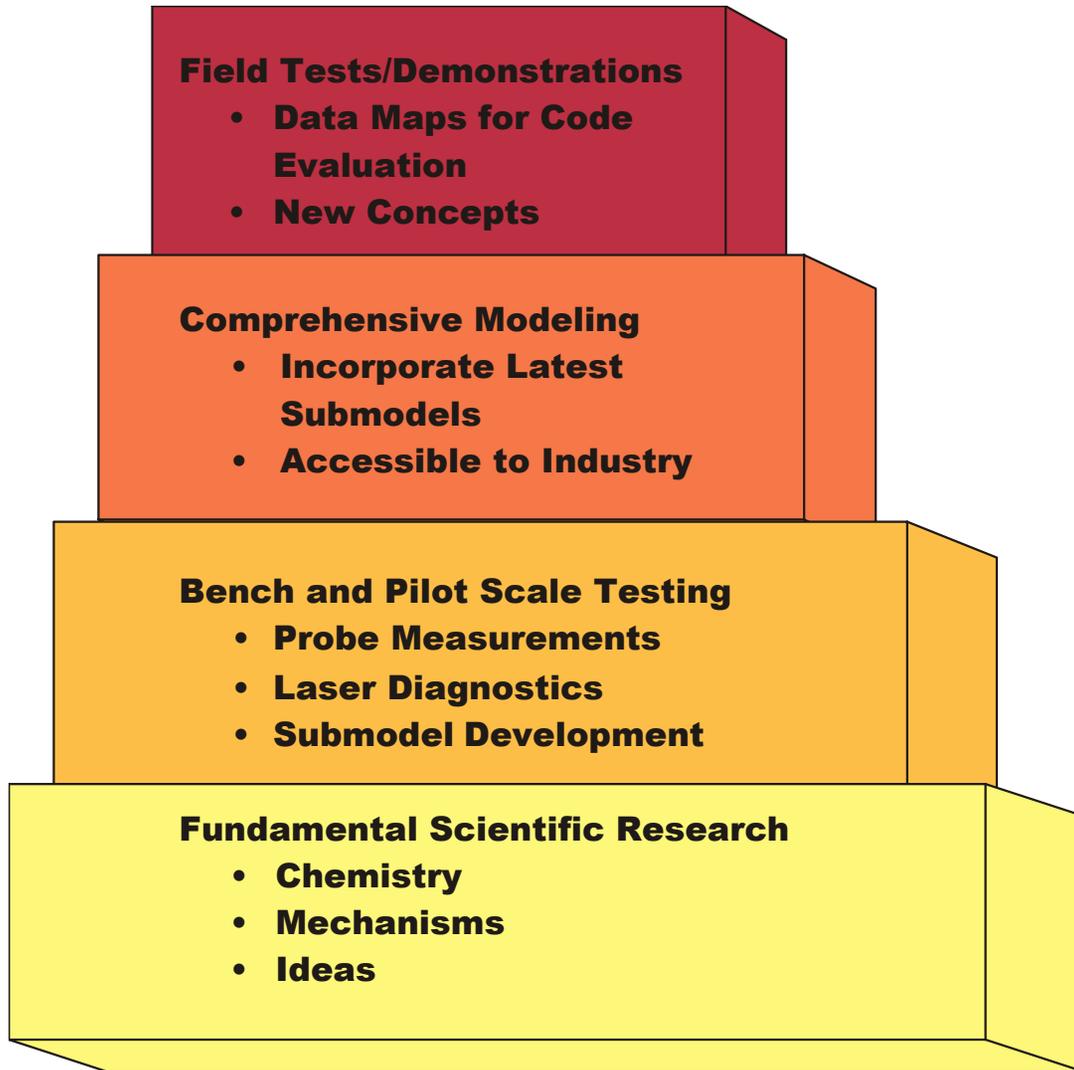


Who Really Does The Work

- **~ 30 Graduate Students**
- **~ 45 Undergraduate Students**
- **~ 5 Research Faculty (at U of Utah)**
- **Visiting Scientists**

ACERC secretaries – Jessica Higginbotham, Trang Tran

Range of Research Interests in ACERC



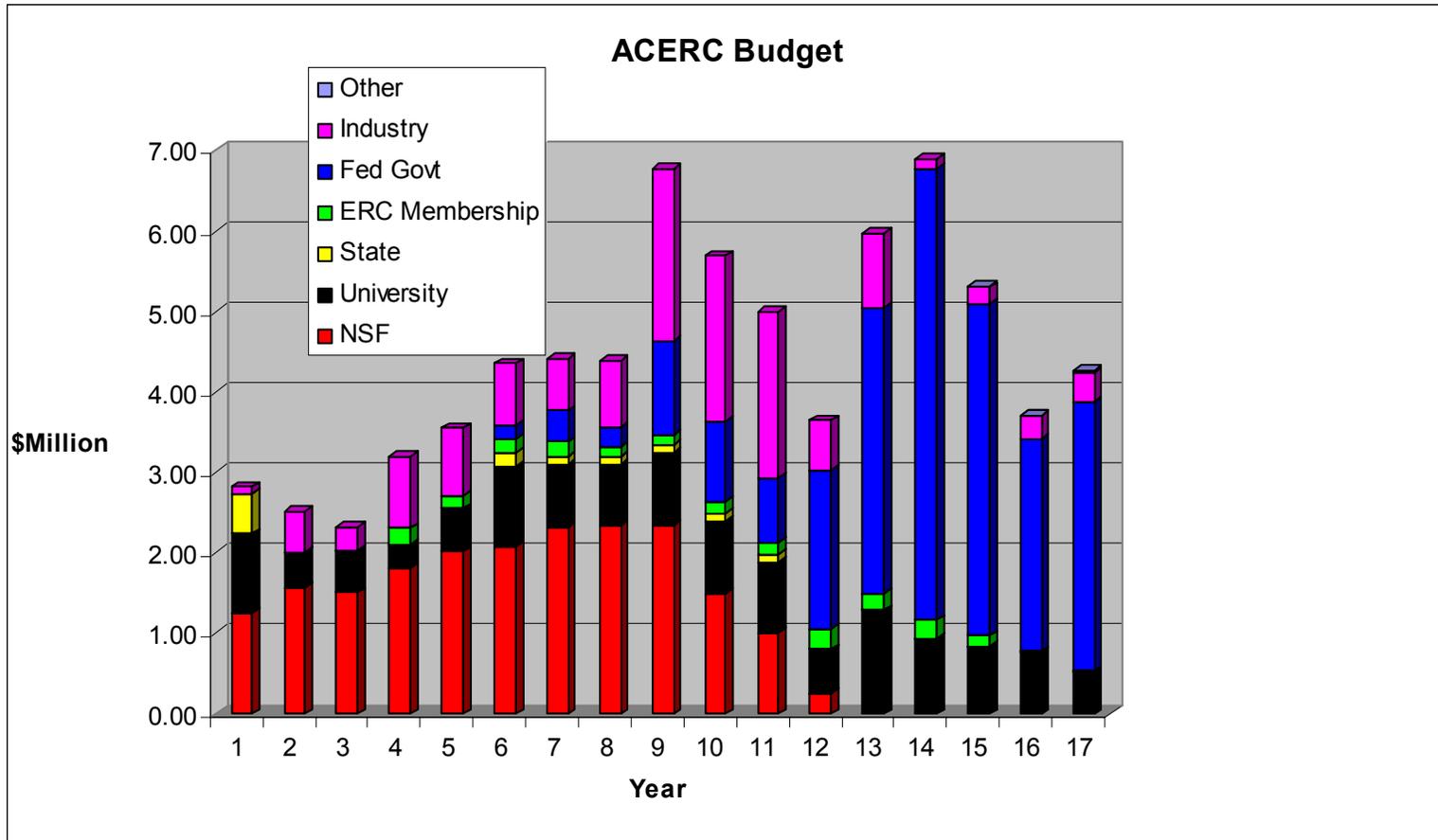
Types of Reactors

- Entrained Flow
- Fluidized Bed
- Fixed Bed
- Rotary Kiln
- Suspended drops & particles
- Levitated particles

Scale of Reactors

- Single Particle
- 1 g/hr
- 5 lbs/hr
- 40 lbs/hr
- 1 MW
- 150 MW
(small industrial)

ACERC Research Funds 1987-2002



Over \$75 million in research over 17 years!



**Next Annual ACERC Conference: Feb. 12-13, 2004
in Provo, UT (see www-acerc.byu.edu)**



