

# Characterization of Fine Particulate Matter at Elementary Schools in Central and Southeastern Ohio

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*Ohio Environmental Protection Agency*  
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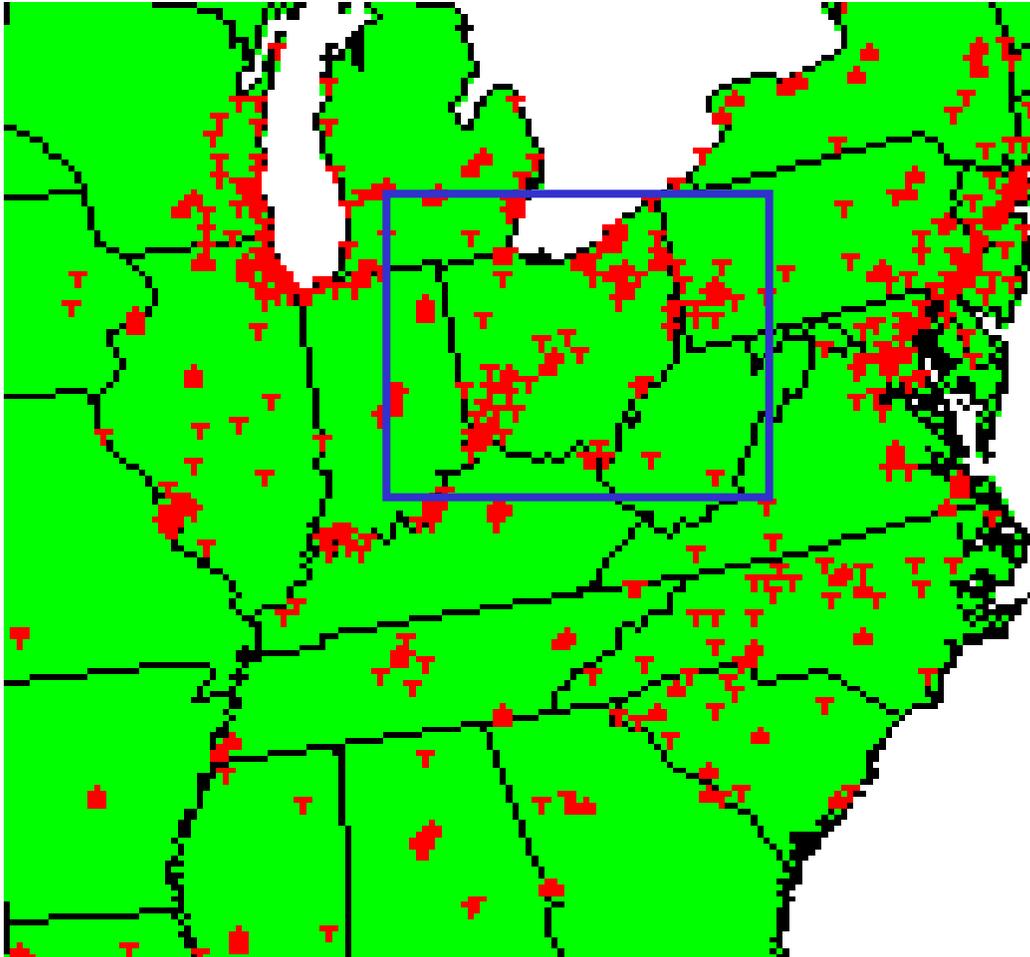
# OBJECTIVES

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- **Evaluate Air Quality**
  - **Regional, Seasonal, Daily**
  - **Meteorological Influence**
- **Correlation: Outdoor, Indoor, and Personal Exposures**
- **Correlation Between Health and Air quality**
  - **Prospective (PM<sub>2.5</sub>, O<sub>3</sub>, Health Data)**
  - **Retrospective (Criteria Pollutants, Health Data)**

# RETROSPECTIVE

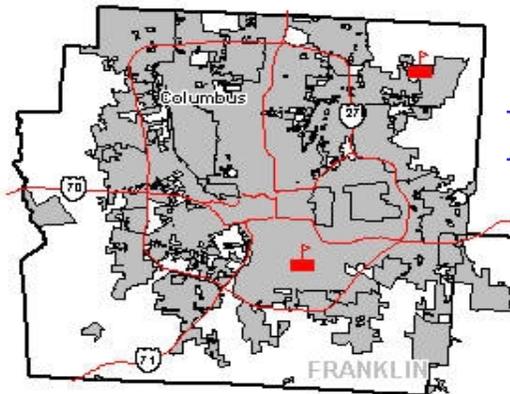


## Statistical Correlation 1992-1998

- Hospital Admissions  
over 300,000/yr
- Criteria Pollutants  
( $O_3$ ,  $SO_2$ ,  $NO_2$ ...)

▽ Ozone Monitors

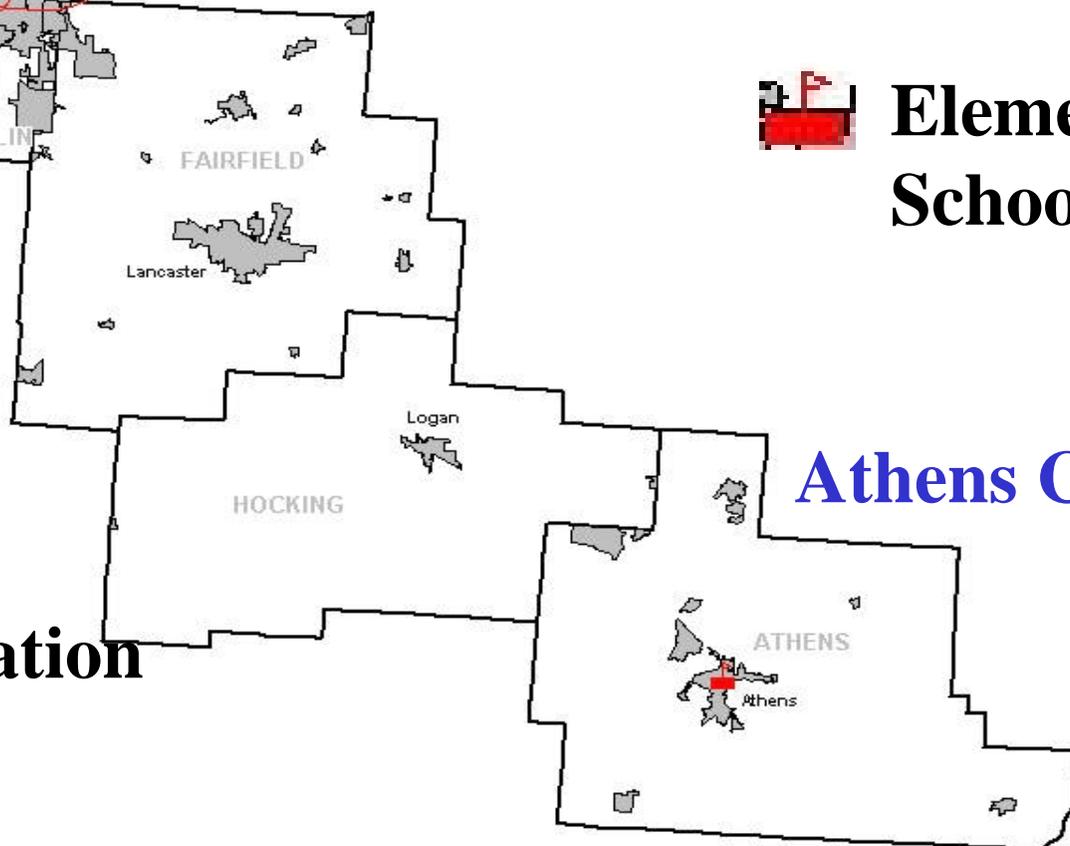
# PROSPECTIVE STUDY



Franklin County



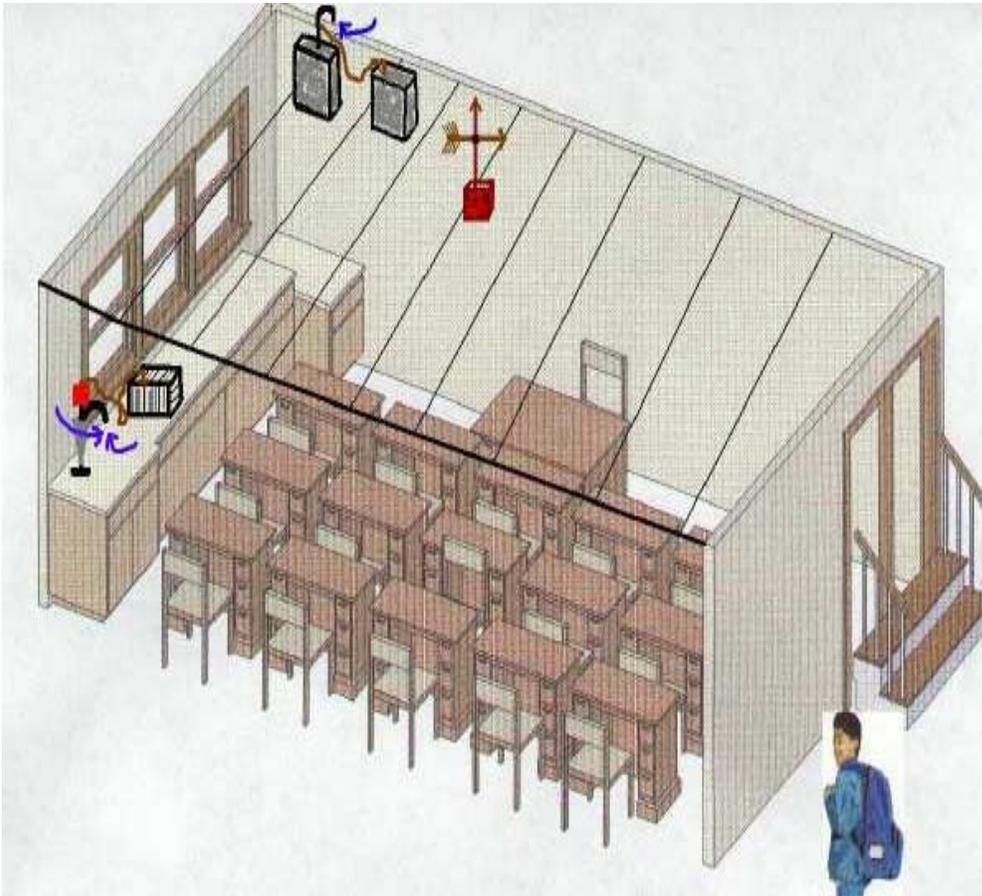
Elementary School



Athens County

- Susceptible Population
- Neighborhood

# MONITORING



- **PM<sub>2.5</sub>**
  - **Ambient**
  - **Indoor**
  - **Personal**
- **Meteorological**
- **Ozone**

# Installation

## Urban Location

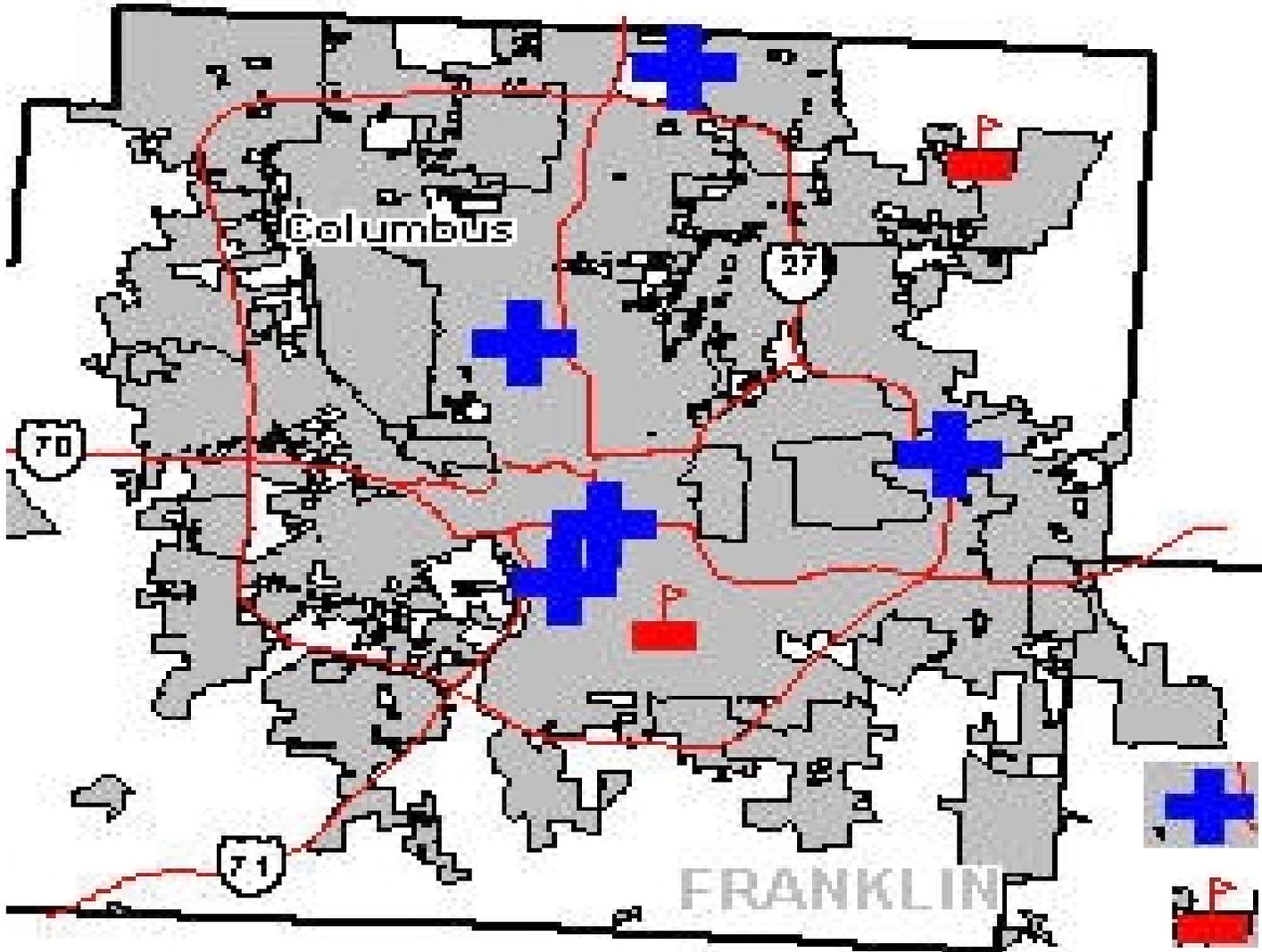


# Health Monitoring



- **Peak Flow Meters**

# Health Monitoring Sentinel Sites



# Ambient Monitoring



- **PM<sub>2.5</sub> Continuous**
- **PM<sub>2.5</sub> -Filter**
- **Weather**
- **Ozone**

# Personal Monitoring

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- **Filter Samples**
  - **7 hr day –5 days per week**

**OVMs –Air Toxics**  
**Personal,**  
**Outdoor, Indoor**

# Analysis of Trace Elements

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## Energy Dispersive X-ray Fluorescence

Elements : Si, P, S, Cl, K, Ca,  
Ti, Co, Ni, V, Cr, Mn, Fe, Cu,  
Zn, As and Cd

## Ion chromatography

Anions: F<sup>-</sup>, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>-2</sup>,  
PO<sub>4</sub><sup>-3</sup>

Cations: Li<sup>+</sup>, Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>,  
Mg<sup>+2</sup>, Ca<sup>+2</sup>

# Analysis

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- Water Soluble Ions
- Metals

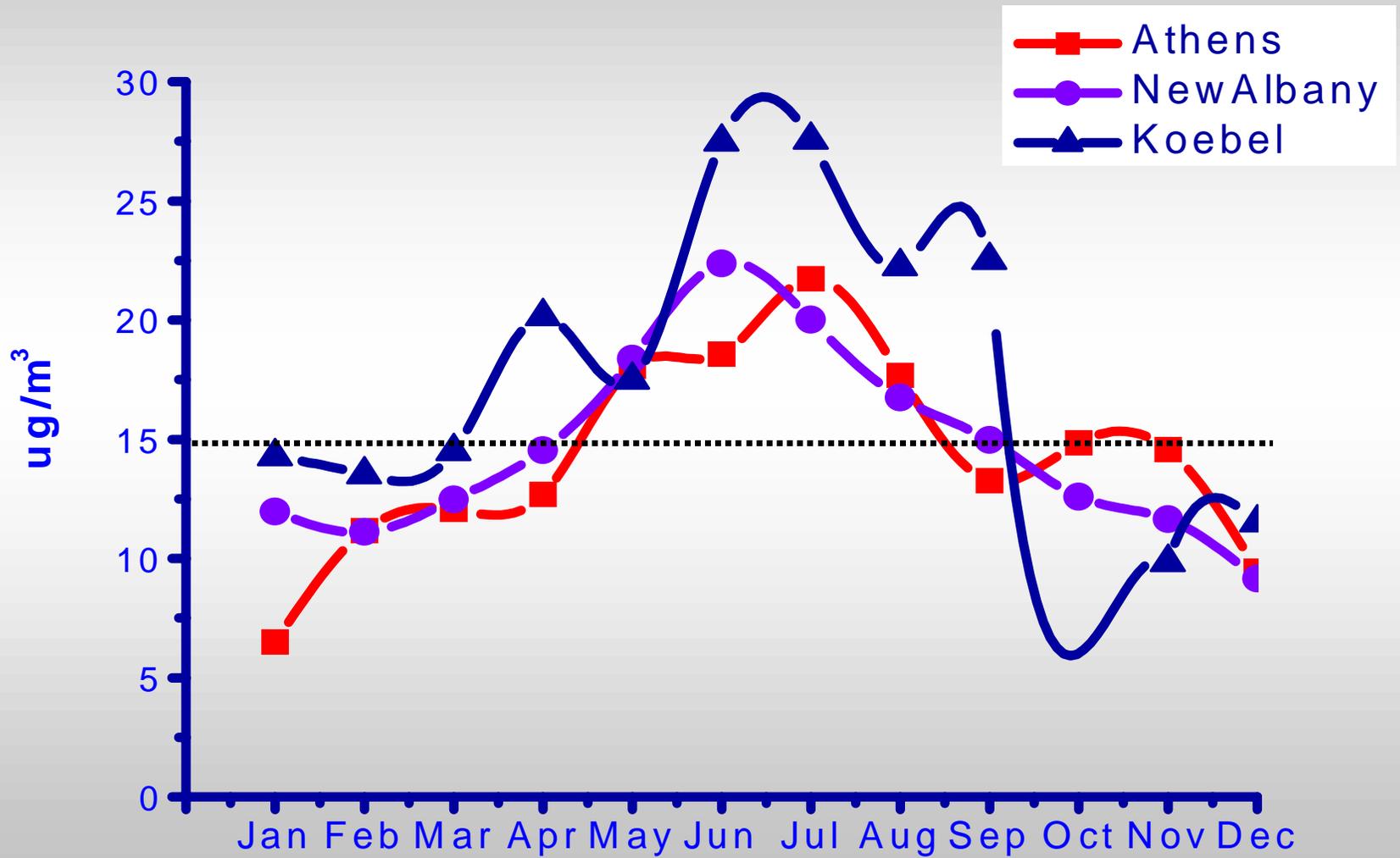


## *Seasonal Trend of Ambient PM<sub>2.5</sub>*

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- PM<sub>2.5</sub> concentrations show clear seasonal changes at all three sites, indicating the trend of *higher PM<sub>2.5</sub> concentrations during the summer* months.
- *Higher ambient PM<sub>2.5</sub>* concentrations were identified *at the urban site*---- Koebel, which was influenced by local sources.
- Data from 1999 and 2000 suggest that all three sites will **exceed** the annual standard of 15µg/m<sup>3</sup>.

# Monthly TEOM PM<sub>2.5</sub> Concentrations, 1999



# Results and discussion

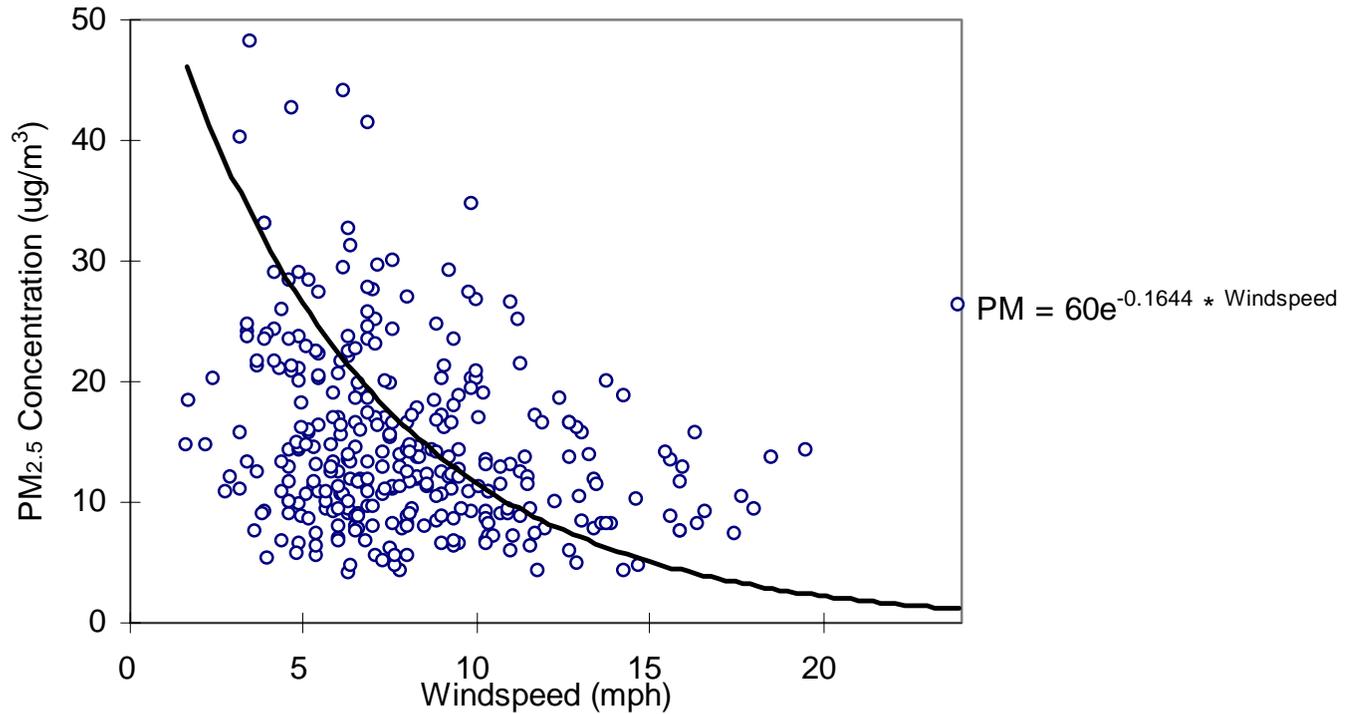
## *PM<sub>2.5</sub> and Meteorological Parameters*

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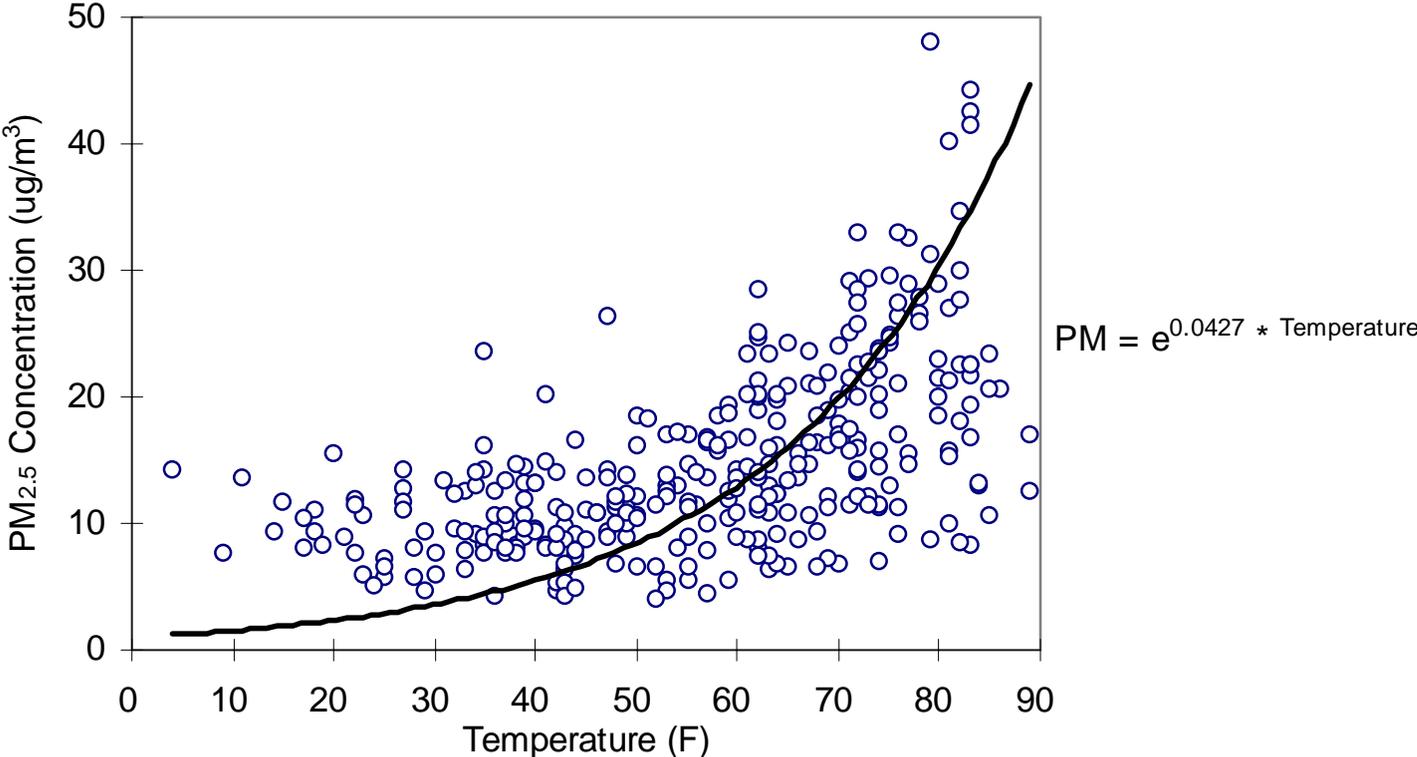
### Key Results for Ambient Monitoring

- PM<sub>2.5</sub> concentrations tended to increase with rising temperatures
- PM<sub>2.5</sub> concentrations decreased with increasing wind speed
- PM<sub>2.5</sub> concentrations in general were highest when the winds were blowing from the south & southeast directions despite the low frequency of occurrence of this particular wind direction.

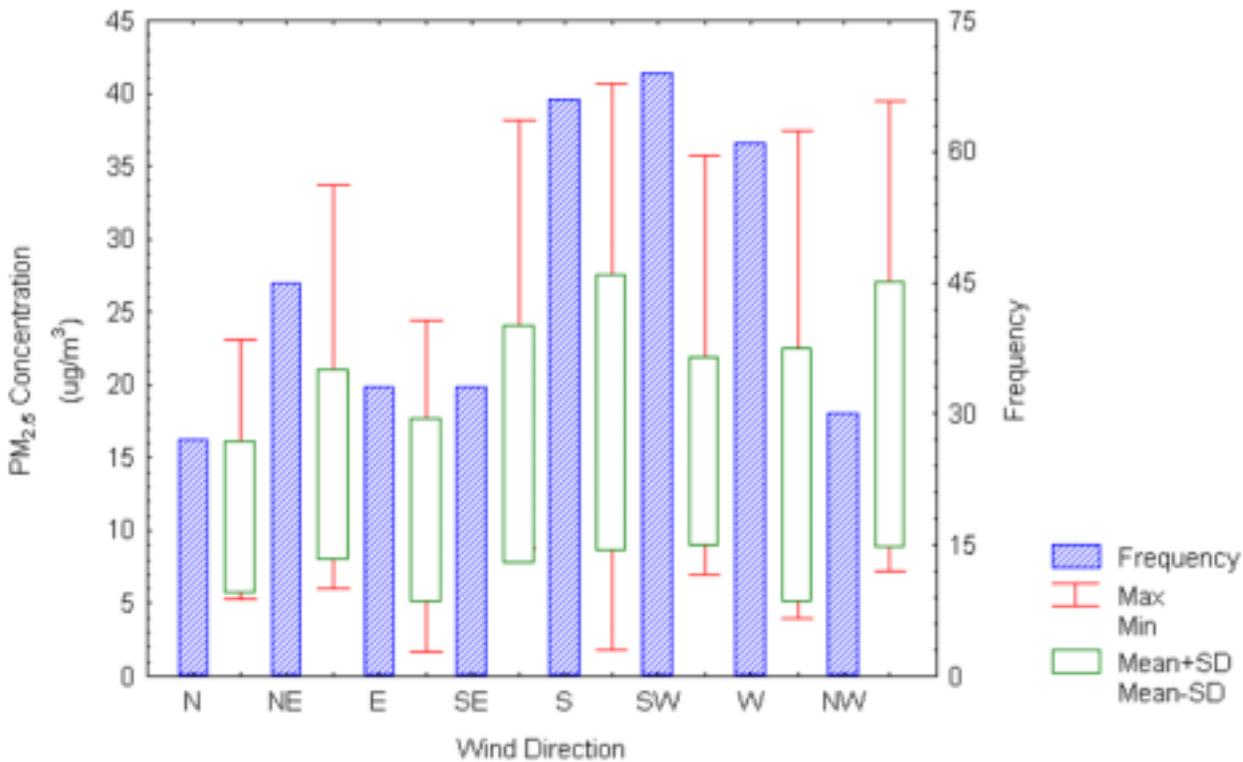
# Relation Between PM<sub>2.5</sub> & Wind Speed at New Albany



# Relation Between PM<sub>2.5</sub> & Temperature at New Albany



# Relation between PM<sub>2.5</sub> & wind direction at Athens (rural)



# Results and discussion

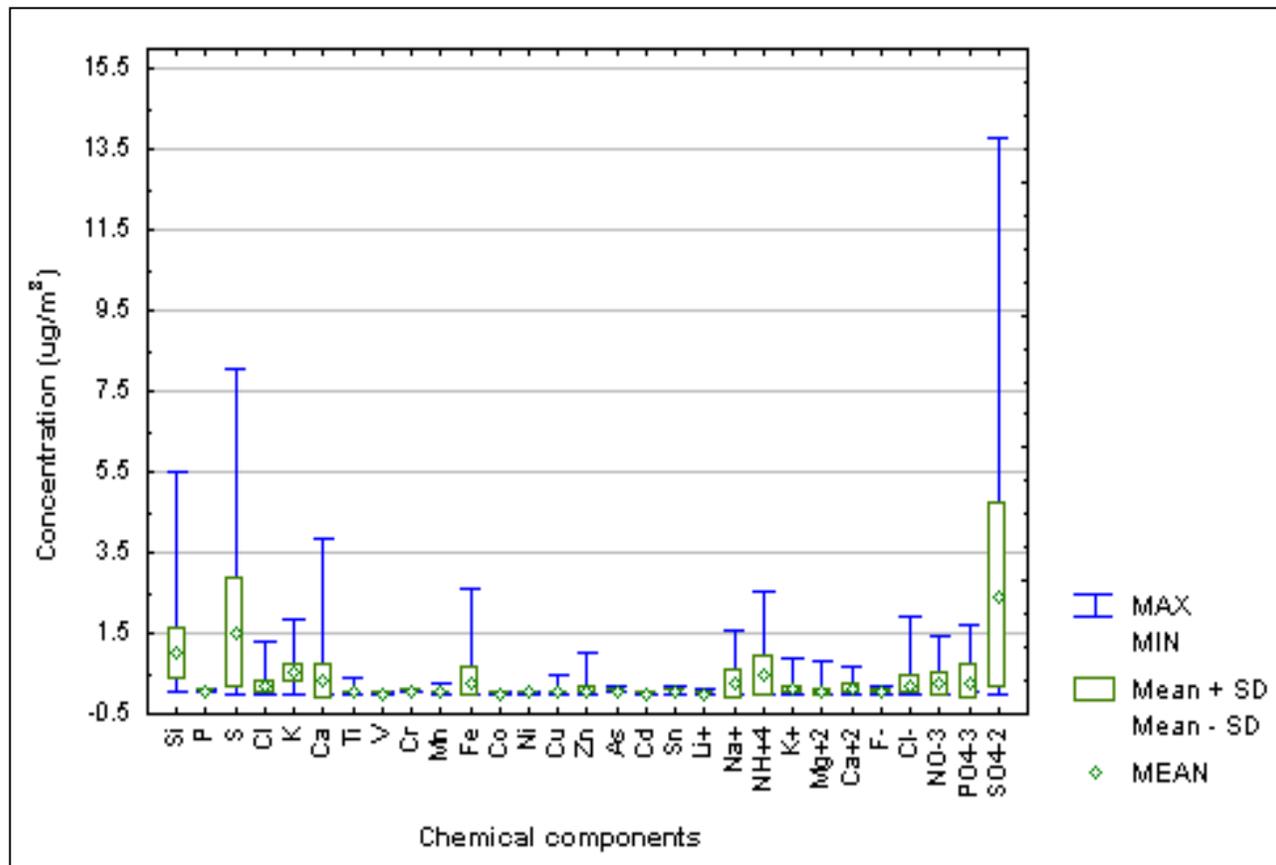
## *PM<sub>2.5</sub> Chemical Species Concentrations*

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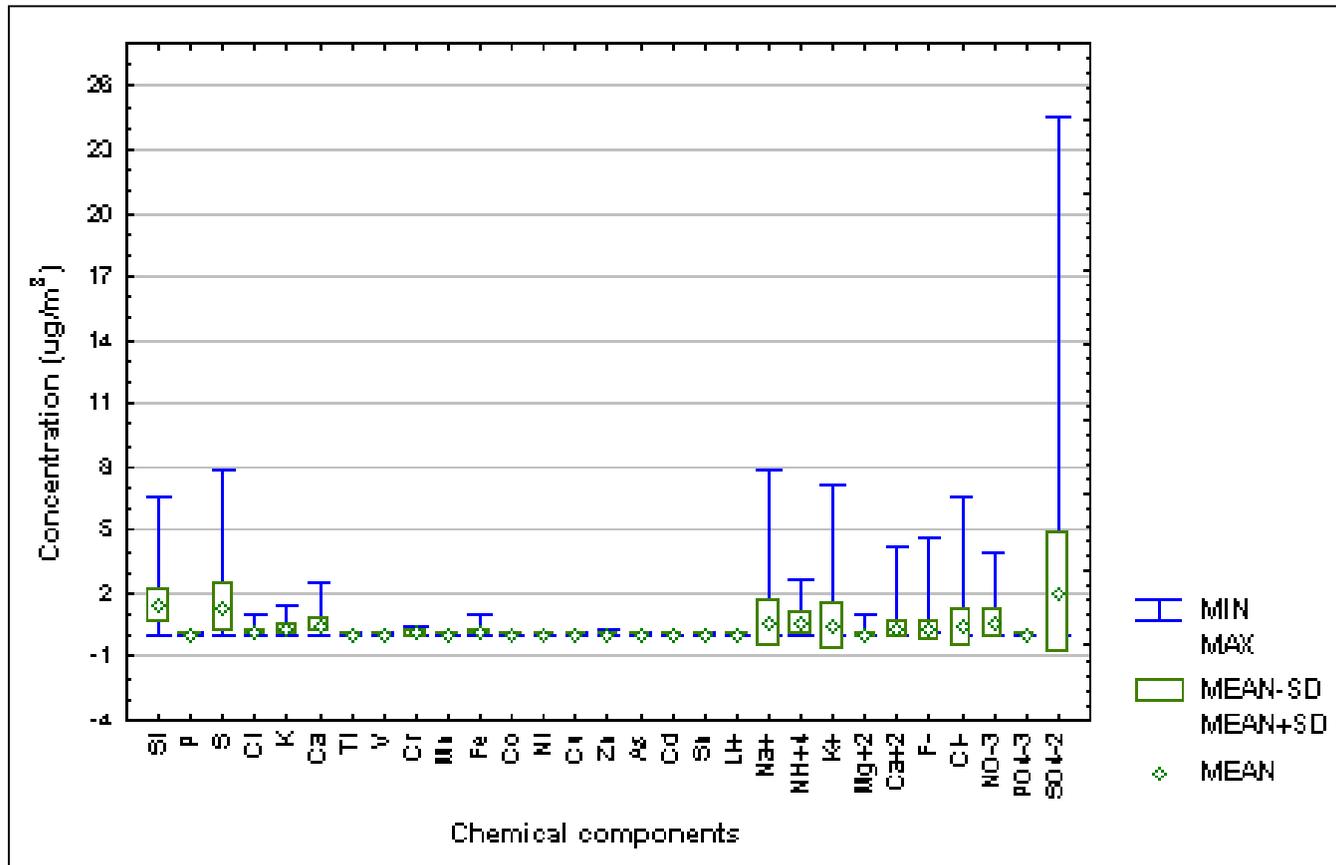
### Key Results at Outdoor & Indoor Sites

- Components Identified: Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>-2</sup>, PO<sub>4</sub><sup>-3</sup>, Li<sup>+</sup>, Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>+2</sup>, Ca<sup>+2</sup>, Si, P, S, Cl, K, Ca, Ti, V, Cr, Co, Ni, Mn, Fe, Cu, Zn, As, Cd and Sn
- Sulfate major component present in all PM<sub>2.5</sub> samples.
- Other abundant components included nitrate and ammonium ions and silicon
- Anion and cation average concentrations followed the pattern SO<sub>4</sub><sup>-2</sup> > NO<sub>3</sub> > PO<sub>4</sub><sup>-3</sup> > Cl<sup>-</sup> and Na<sup>+</sup> > NH<sub>4</sub><sup>+</sup> > Ca<sup>+2</sup> > K<sup>+</sup> > Mg<sup>+2</sup>
- Heavy metals such as Ti, V, Mn, Fe, Cu and Zn found in all the samples, and iron was the most abundant species

# Concentrations of the chemical components present in the samples at Koebel outdoor site (February 1999-August 2000)



# Concentrations of the chemical components present in the samples at Koebel indoor site (February 1999-August 2000)



# Results and Discussion

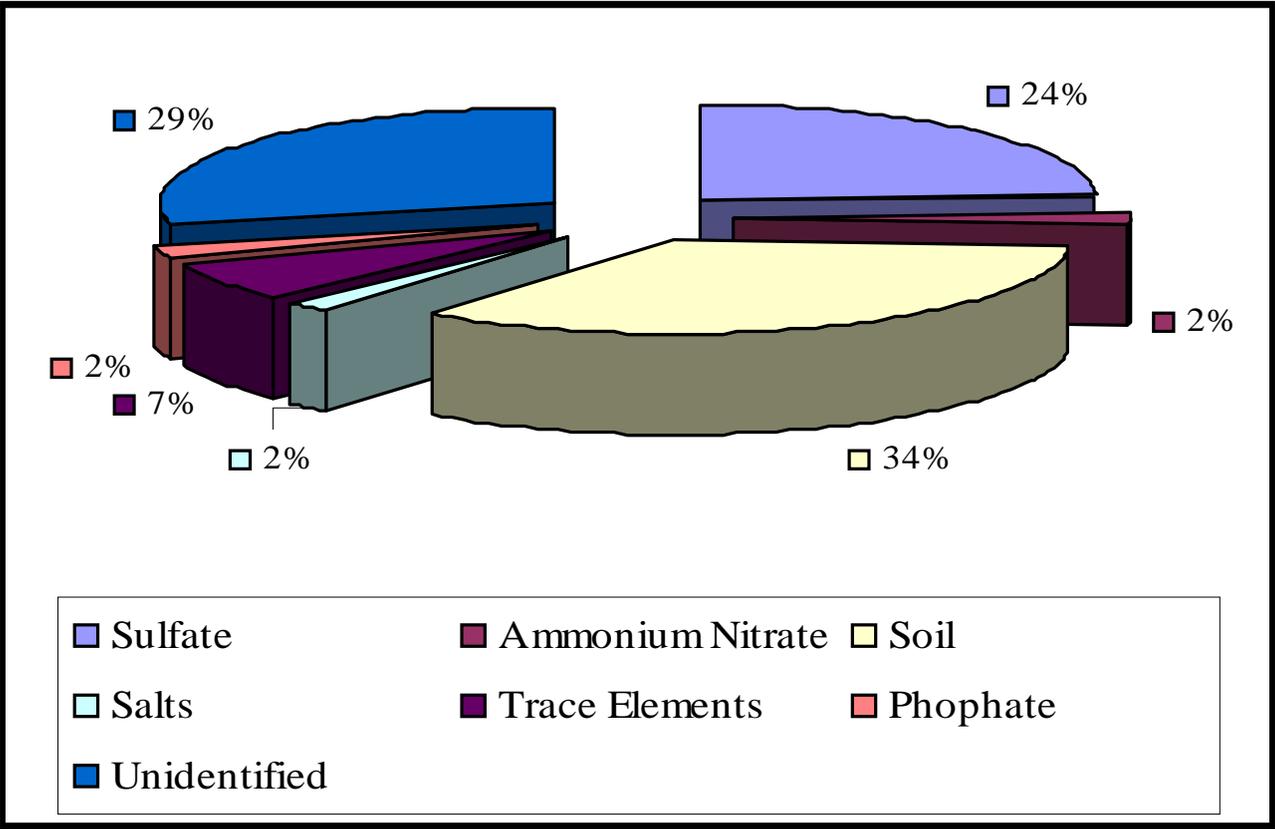
## *Major Chemical Components of PM<sub>2.5</sub>*

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### Key Results for Outdoor Sites

- On an average, 65% of the total particulate (PM<sub>2.5</sub>) mass successfully analysed at the outdoor sites
- Sulfate percentage ranged from 22% to 25% of the total particulate matter mass. Sulfate was highest at the East site
- The percentage range of soil varied from 27% to 34% with Koebel showing the highest percentage
- East rural site showed significantly higher percentage (10%) of phosphates compared to the other two sites

# PM<sub>2.5</sub> components for outdoor ambient air at Koebel (February 1999–August 2000)



## Results and Discussion

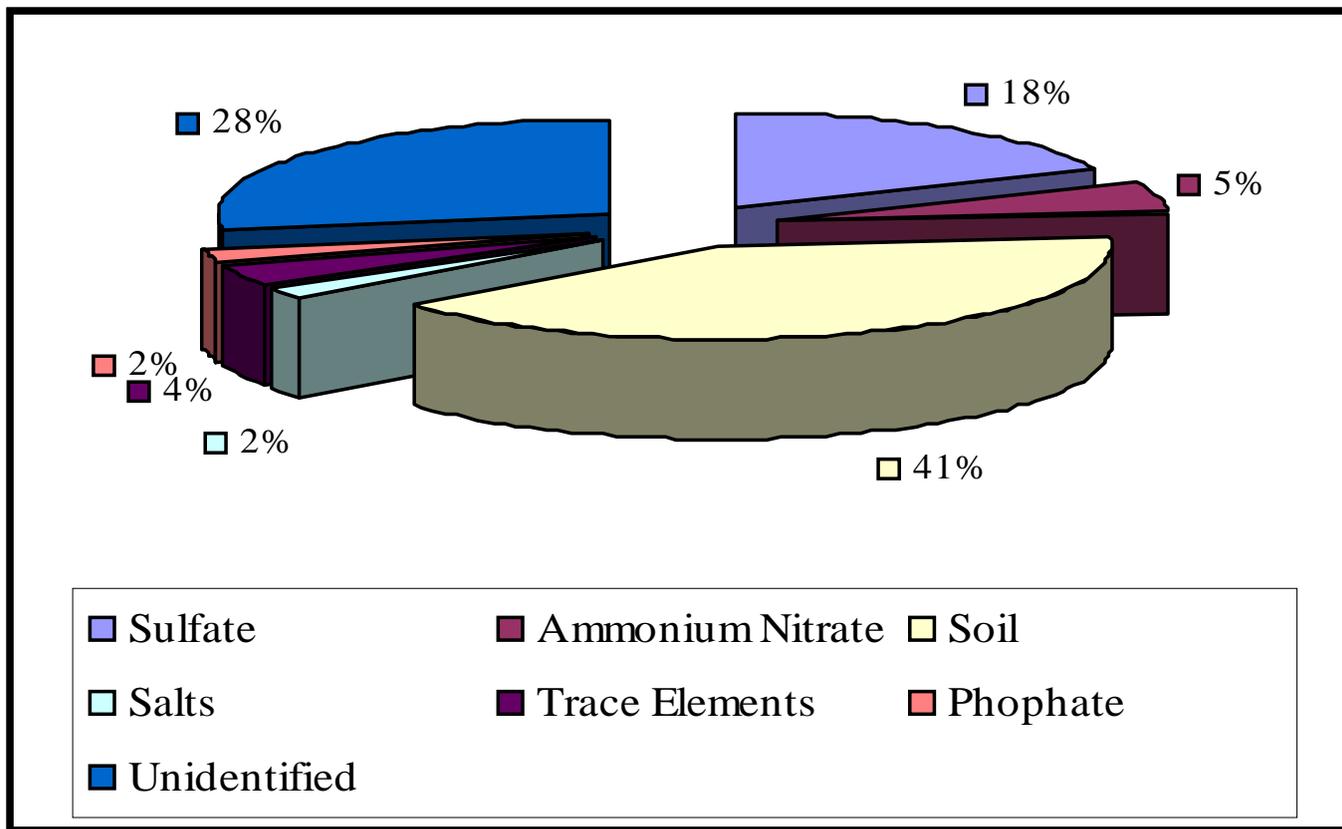
### *Major Chemical Components of PM<sub>2.5</sub>*

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#### Key Results at Indoor Sites

- Soil concentration at all indoor sites were higher compared to the outdoor site soil concentrations
- Soil percentage ranged from 36% to 56% at the New Albany and the East indoor site, respectively
- Sulfate concentration was highest at the Koebel site and ranged from 15% to 18 % of the total particulate matter mass
- Little or no phosphate found at the indoor sites

# PM<sub>2.5</sub> Chemical Composition for Koebel Indoor Site (February 1999–August 2000)



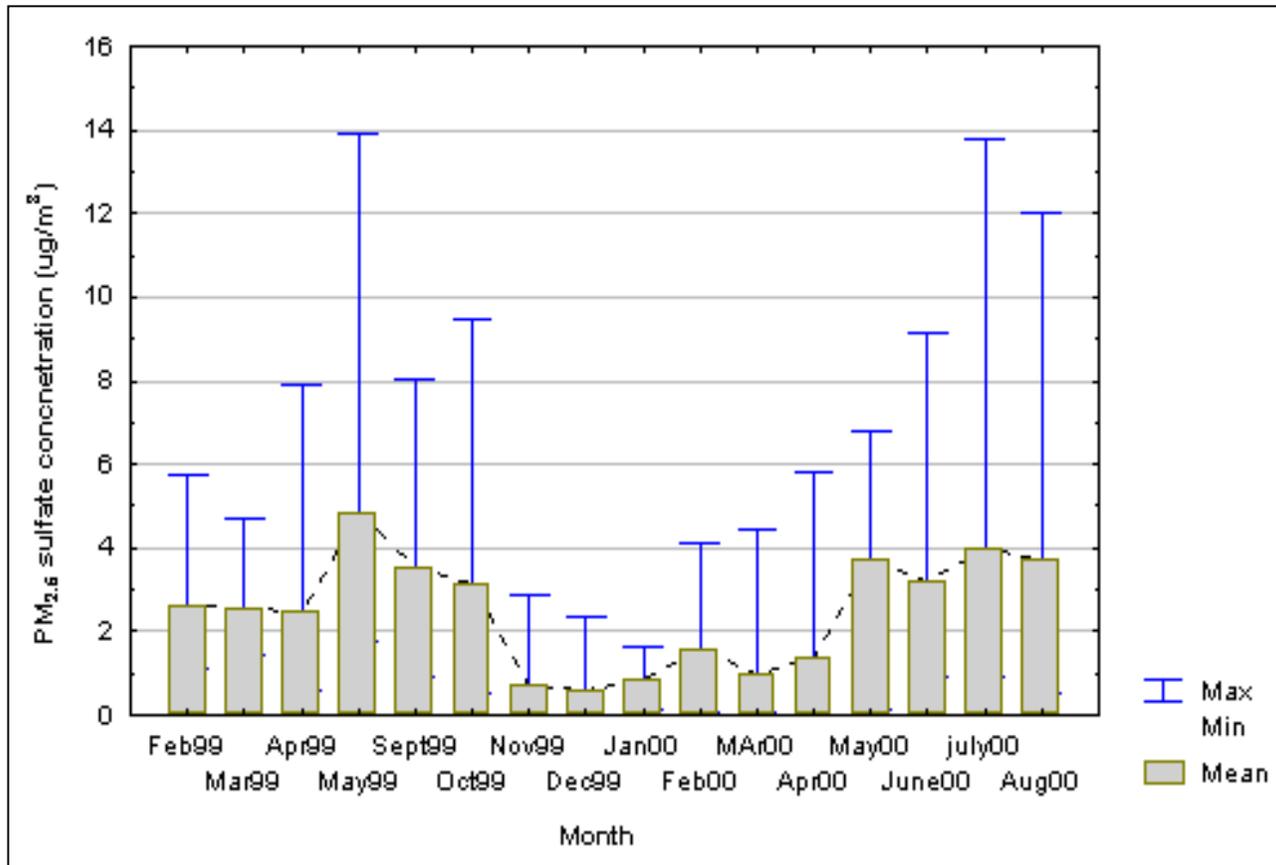
# Results and Discussion *Monthly Variations in Sulfate Concentration*

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## Key Results

- Sulfate ion concentration showed a strong seasonal variations with maximum concentrations observed during the summer months
- Sulfate concentrations increased from winter to summer
- Average PM<sub>2.5</sub> sulfate concentrations for East, New Albany, and Koebel sites were 2.47, 2.09, and 2.46  $\mu\text{g}/\text{m}^3$ , respectively

# Monthly variations in Sulfate Concentrations for New Albany Outdoor Samples (February 1999-August 2000)

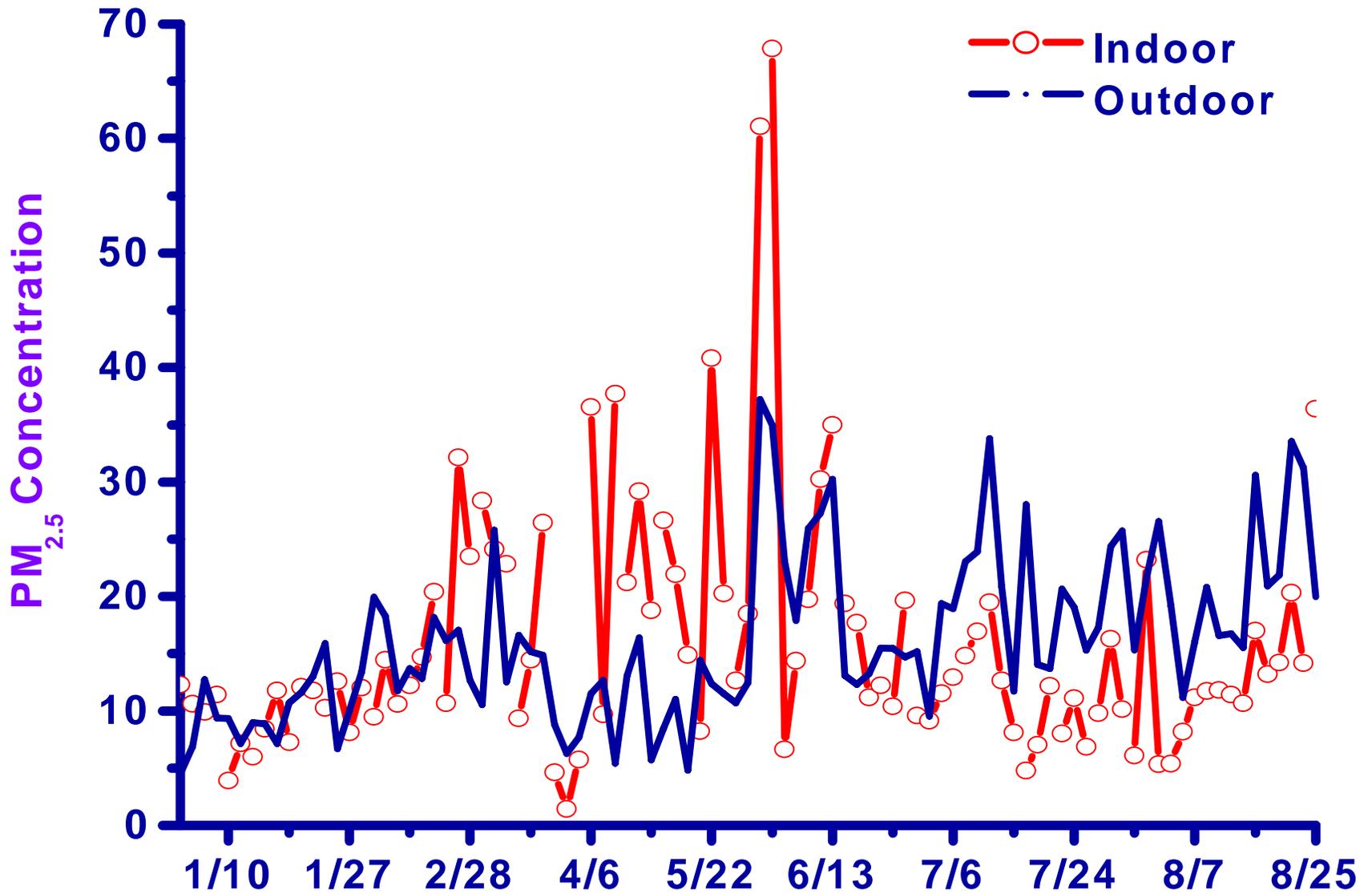


## Results and Discussion *Comparison of Indoor and Outdoor Analysis*

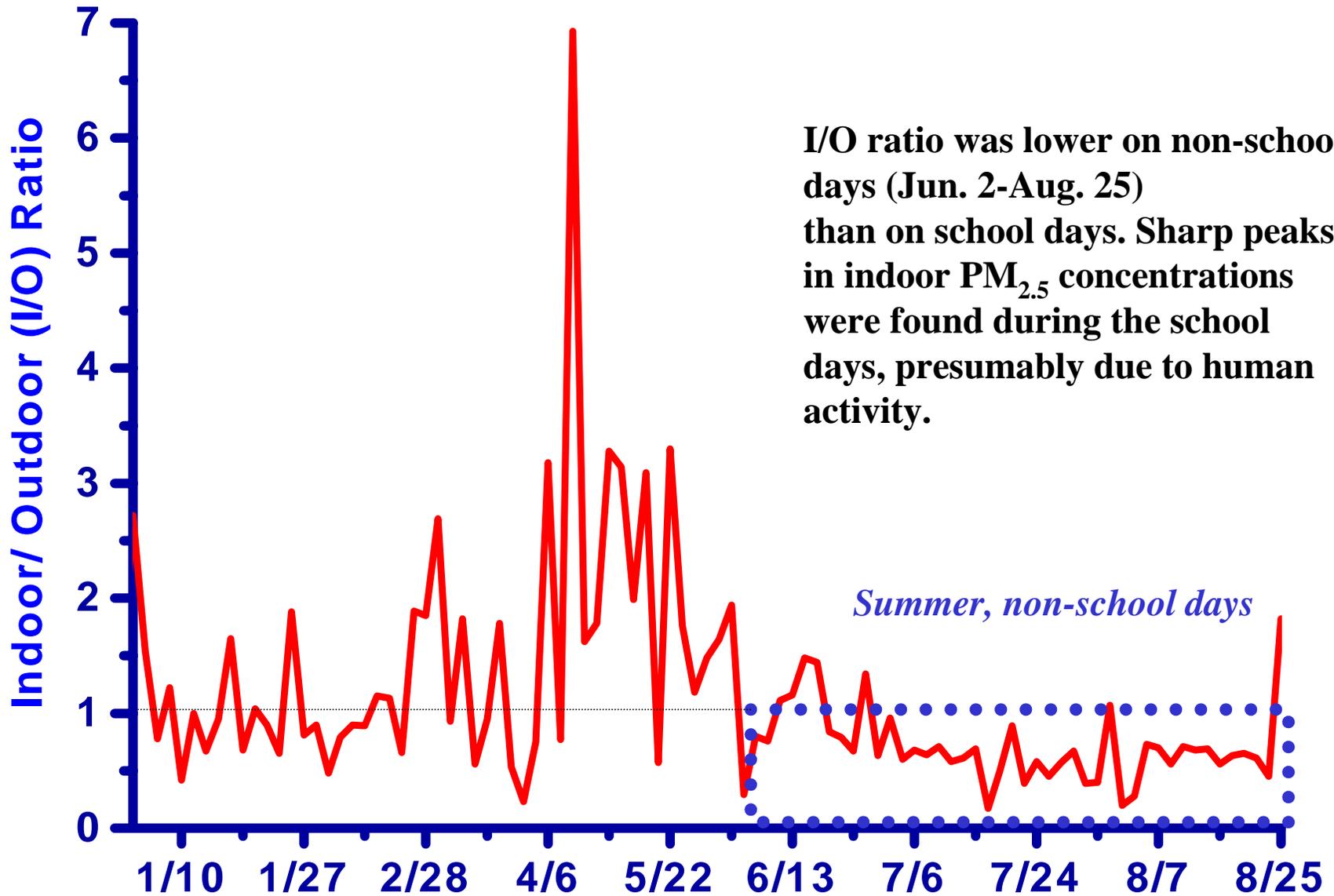
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- In general, indoor PM<sub>2.5</sub> concentrations were higher than outdoor levels.
- Human activity in the indoor settings impacts the PM<sub>2.5</sub> concentration.

# Indoor and Outdoor PM<sub>2.5</sub> concentrations at Athens, 2000



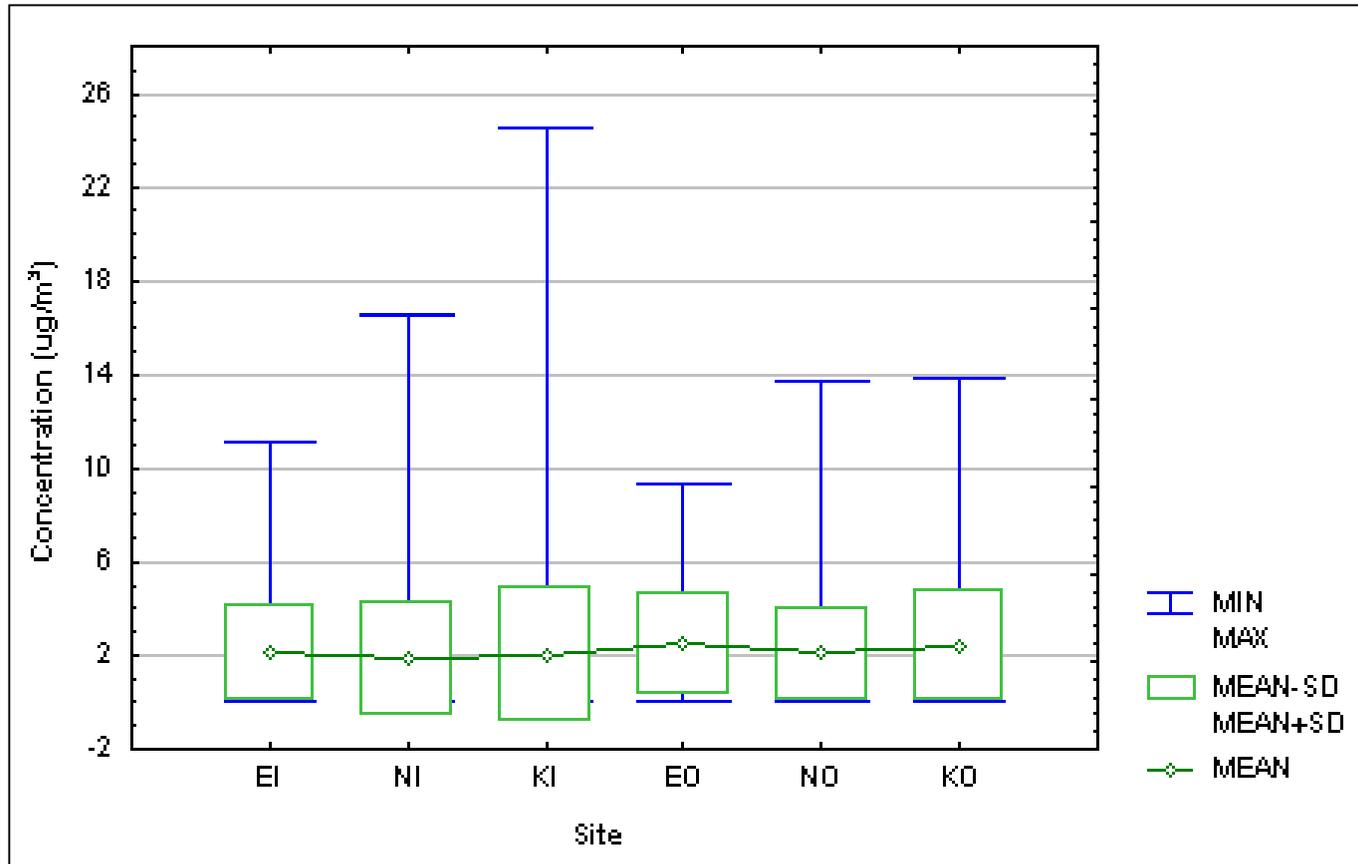
## I/O ratio at Athens, 2000



**I/O ratio was lower on non-school days (Jun. 2-Aug. 25) than on school days. Sharp peaks in indoor  $PM_{2.5}$  concentrations were found during the school days, presumably due to human activity.**

*Summer, non-school days*

# Sulfate ion distribution in Ohio (February 1999 – August 2000)



# Summary & Conclusions

- On an average, 65% of the total particulate (PM<sub>2.5</sub>) mass was successfully analysed at the indoor and outdoor sites.
- The components determined by the chemical analysis included: F<sup>-</sup>, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>-2</sup>, PO<sub>4</sub><sup>-3</sup>, Li<sup>+</sup>, Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>+2</sup>, Ca<sup>+2</sup>, Si, P, S, Cl, K, Ca, Ti, Co, Ni, V, Cr, Mn, Fe, Cu, Zn, As and Cd.
- Sulfate comprised the largest fraction.
- Other abundant components included phosphate, nitrate, ammonium, chloride, sodium, calcium, silicon, and iron.
- Soil concentration (36% - 56%) at the indoor sites were found to be significantly higher compared to the outdoor site soil concentrations

# Summary & Conclusions cont.

- Sulfate ion concentration showed strong seasonal variations with maximum concentrations observed during the summer months
- $PM_{2.5}$  concentrations tended to increase with rising temperatures, and decreased with increasing wind speeds.
- $PM_{2.5}$  concentration was highest when the winds were blowing from the south and the southeast direction at the three sites.

# Summary & Conclusions cont.

- In general, indoor PM<sub>2.5</sub> concentrations were higher than outdoor levels
- Human activities in the indoor settings affected the monitored levels.

# *Acknowledgements*

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**Ohio EPA**

**Ohio Air Quality Development Authority**