Hydrate Formation and Dissociation in Simulated and Field Samples

Methane Hydrates Research at Oak Ridge National Laboratory

C.J. Rawn, PI T.J. Phelps, Co-PI M. Elwood-Madden, collaborator





Current Personnel

Tommy Phelps Co-PI, microbiology, large vessel experiments, and safety engineering

> Miguel Rodriguez, Jr R&D Associate and Technical Resource

Jonathan Alford Post-graduate, starting September 08

Claudia Rawn Co-PI, material science, X-ray/neutror diffraction and crystallography



Ji-Won Moon Postdoctoral Fellow



Megan Elwood-Madden University of Oklahoma

Hydrate background of PI's

T. J. Phelps, Distinguished staff scientist, 16 yrs at ORNL Published ~15 papers on gas hydrates, total of ~ 150 Trained 25 post-graduates

Related service includes 1st microbiologist and 1st DOE rep on IODP/ODP USSAC and committees 1999-2002 Since 2006 has served DOT in pipeline related issues





C. J. Rawn

Senior staff scientist with 11 yrs at ORNL Joint Faculty at the University of Tennessee

Published ~ 12 papers on gas hydrates, total of ~ 80 Interacts daily with DOE facility users, undergraduates and graduate students

Collaborativepublications with SNS, NIST, USGS, INL, Ga Tech, Texas A&M, Univ. of Oklahoma, Oregon State Univ.

Training since 2006 Hydrate Review



Megan Elwood Madden ORNL Wigner Fellow – 2005 - 2007 Assistant Prof. at the University of Oklahoma

ConocoPhillips SCHOOL OF GEOLOGY AND GEOPHYSICS

UNIVERSITY OF OKLAHOMA

Scott McCallum – post masters student geosciences - 2003 - 2006 Petro-geologist, PA

Phillip Szymcek – post masters student geosciences – 2005 – 2007 Petro-geologist, AR

Patricia Taboada-Serrano – 2005 – 2008 (Former Fulbright Fellow)

Shannon Ulrich – post bachelors student 2006 - 2008 Colorado School of Mines Environmental Science and Engineering



Facilities and Equipment

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- In-Situ Diffraction
 - Neutron Time, Temp, Pressure
 - XRD
- Hydrate synthesis capabilities
 - Deuterated samples for neutron diffraction
- Seafloor Processing Simulator
 - housed in a temperature controlled explosion-proof cold room
 - capable of simulating deep seafloor pressures and temperatures



Luna Distributed Sensing System (DSS) for observation of hydrate formation/dissociation



In-Situ neutron diffraction provides capabilities to do combined time, temperature, and pressure decomposition studies (spallation at the SNS provides the time component).

Two neutron powder diffraction patterns collected on samples provided by the USGS. Red: hydrogenous (H_2O), Black: deuterated (D_2O) showing how deuteration decreases the background and increases the Bragg scattering detailing atomic positions, lattice parameters, etc.

Spallation Neutrons and Pressure (SNAP)

Collaborators: Chris Tulk and Bryan Chakoumakos

SNAP Diffractometer allows studies of a variety of samples under extreme pressure and temperature conditions

Applies up to 100 GPa on an ~1mm³ sample



This enables us to collect temperature/pressure/time dependent data: Thermal expansion Bulk modulus (compressibility) Dissociation kinetics (formation?)

This new beamline has Increased neutron flux Large-volume pressure cells Next-generation detectors

In-Situ X-ray Powder Diffraction

- PANalytical (Phillips)
 X'Pert Pro Diffractometer
 - Anton Paar TTK 450 low temperature chamber
 - X'Celerator detector for fast data collection



Time/Temperature dependent dissociation studies On natural samples





Low Temperature X-ray Diffraction Studies of GOM Hydrates

Collaborator: R. Sassen (Texas A&M) GOM samples with ~30 – 50% sll hydrate

Temperature dependent data showing dissociation of sll hydrate to water ice





Future studies to determine rate constants and reaction parameters

Collect time dependent data at multiple constant temperatures until 100% decomposition Gas Chromatography/Mass Spectometery to analyze gases for gas mass balance



Need for Distributed Sensing System

- 2000 2006 hydrate tracked by bulk measurements or visually
- Purchased fiber optic Distributed Sensing System (DSS)
- DSS uses fiber optics to observe temp changes corresponding to hydrate formation (exothermic) or dissociation (endothermic)





Time-Resolved 3-D temperature monitoring



Distributed Sensing System (DSS)



The Luna® Distributed Sensing System (DSS) laser pulses down an optical fiber determine changes due to temperature and strain

Temperature and strain cause the fiber to expand or contract which are noted in the reflection time

Time for the reflection to return is translated into a hybrid temperature strain value (TSV).





Sensors are at 1-cm intervals



Integrated analysis of hydrate formation in heterogeneous systems



In situ optical, pressure, and temperature observations of hydrate in free gas systems



Elwood Madden et al., Marine and Petroleum Geology (in press)

Model of hydrate growth from free gas phase

•Hydrate films observed along the surface of methane gas bubbles

•Films crystallized to form hydrate nodules in sediment

•Massive hydrate deposits form initially in areas of gas bubble accumulation



Elwood Madden et al., Marine and Petroleum Geology (in press)

Suggests in systems containing free gas the stratigraphy, tectonic, and sedimentary structures likely control the location of massive gas hydrate deposits

Corroborates Indian Ocean hydrates observed in sands and fractures (Kastner, Goldschmidt 2008)

Data Collection with the DSS

- Fibers were coiled in a spiral
 - Attached to a circle of plastic mesh
 - Beginning of fiber on the outside of the spiral
 - End of fiber at the center of the spiral





Synthetic sediment experiments Ottawa Sand/Silt

Saturated to 30% with H_2O

Homogeneous experiments

Heterogeneous experiments With a 3" layer of silt

Ex-Situ Fiber test

• Four fibers and thermocouple exposed to same conditions





Nitrogen Control Experiments

- Four homogeneous sediments experiments
- Two where temperature increases
- Two where pressure decreases (one shown at right)

Fibers show similar overall trends Some fibers could be more sensitive to changes than others

Note: Fiber 115 always shows reproducible variability despite location in SPS, possibility more sensitive to changes



Methane Experiments

Dissociation by

- Four homogeneous sediment experiments
- CH₄ gas at T/P •
- < 50 g of hydrate expected to form
- ~1 L of CH_4 added at 0.6 mL/min • via HPLC pump
 - Initial vessel cooling
 - System reaches equilibrium
 - **Depressurization**
 - Warming

Possible hydrate formation indicated by upward arrows and dissociation by downward arrows





FY09 – FY11 Directions for SPS - DSS Experiments

- Homogeneous and Heterogeneous sediment experiments pressuring with CH₄
- Large data set analysis
 - Temp/strain monitored every 60 sec
 - Plotting the distance as polar coordinates
 - Igor Pro, Origin, MatLab (subcontract with University of Oklahoma)
- Massive amounts of hydrate to compare to earlier CH₄ experiments
- Examining relationships between overheating and depressurization for gas production





FY09 – FY11 Diffraction Directions

- More in-situ XRD work on natural (hydrogeneous) samples from Texas A&M
 - Time dependent studies (kinetics)
 - Lattice parameters as a function of temperature (thermal expansion)
- In house synthesis of deuterated samples for neutron powder diffraction
 - ORNL's High Flux Isotope Reactor (HFIR) powder diffractometer upgrades to be completed in FY09
 - SNAP open for general users in FY09
 - Data collection as a function of pressure
 - Dissociation pressures
 - Bulk modulus from analyzing lattice parameters as a function of pressure





Productivity – Publications 2007 and 2008

Elwood Madden, M.E., S.M. Ulrich, T.C. Onstott, and T.J. Phelps, 2007, Salinity-induced hydrate dissociation: a mechanism for recent CH₄ release on Mars. *Geophysical Research Letters* 34, Issue 11, CiteID L11202.

McCallum, S., D. Riestenberg, O. Zatsepina and T. Phelps, 2007, Effect of pressure vessel size on the formation of gas hydrates, *Journal of Petroleum Science and Engineering – Natural Gas Hydrate/Clathrate* (Elsevier), edited by D. Mahajan, C. Taylor and G. Ali Mansoori, Volume 56, Issues 1-3, p. 54-64, March.

Colwell, F., S. Boyd, M. Delwiche, D. Reed, T. Phelps, and D. Newby, 2008, Estimates of biogenic methane production rates in deep marine sediments at Hydrate Ridge, Cascadia Margin, *Applied and Environmental Microbiology*, 74: 3444-3452.

Elwood Madden, M.E., P. Szymcek, S.M. Ulrich, S. McCallum, and T.J. Phelps, 2008, (in press) Experimental formation of massive hydrate deposits from accumulation of CH4 gas bubbles within synthetic and natural sediments. *Marine and Petroleum Geology*

Taboada-Serrano, P. S. Ulrich, P. Syzmcek, S. McCallum, T. J. Phelps, A. Palumbo, and C. Tsouris. 2008. A multi-phase, micro dispersion reactor for the continuous production of methane gas hydrate. (Submitted).

Proceedings:

Rawn, C.J., R. Sassen, S.M. Ulrich, T.J. Phelps, B.C. Chakoumakos, and E.A. Payzant, "Low Temperature Xray Diffraction Studies of Natural Gas Hydrate Samples from the Gulf of Mexico", *Proceedings of the 6th International Conference on Gas Hydrate (ICGH 2008)*, CD (2008) (In preparation for a journal)

Ulrich, S.M., M.E. Elwood-Madden, C.J. Rawn, P. Szymcek, and T.J. Phelps, "Application of Fiber Optic Temperature and Strain Sensing Technology to Gas Hydrates", *Proceedings of the 6th Internationation on Gas Hydrate (ICGH 2008)*, CD (2008) (Topic is in preparation for a journal)



Productivity – Presentations

- 6th ICGH Vancouver July 2008 (Two Presentations)
- AGU Fall meetings both in 2006 and 2007
- GSA Annual Meeting 2007
- Science and Technology Issues in Methane Hydrate R&D workshop, Hawaii, 2006
- Inter-Laboratory Hydrates Workshop, Sept 2006, CSM
- Hosted Tim Grant at ORNL December 2006
- ORNL Gas Hydrates Workshop February 2007 to inform the ORNL community of special capabilities and ongoing hydrate research conducted at the laboratory, and provided a platform for future collaborations



Other Publications, Milestones, and Quarterly Reports

- Publication planned for the Review of Scientific Instruments on the data collection and processing with the DSS with Luna Technologies
- Publication planned on the time/temperature dependent x-ray diffraction data once more experiments at different temperatures are completed
- All milestones from FY08 FWP completed:
 - 12/07 Collected data with the DSS with heterogeneous sediment column
 - 3/08 submitted report on homogeneous sediment column experiment (precursor to ICGH paper)
 - Switched 6/08 (conduct neutron diffraction experiment) with 3/09 (publication on time dependent xrd studies) milestones – ICGH paper on time dependent xrd studies
- Quarterly reports sent to Robert Vagnetti (NETL)



Budget Considerations

- In FY08 reduced budget by 15% and delayed capital equipment request
- FY09 Capital Equipment to design and build a coldloading system for Paris-Edinburgh pressure cells used on the SNAP beamline
 - ORNL has several PE cells, and the cooling capability, but lack a method/system to load cold samples
 - This cold loading system would give us the ability to pre-cool and load ices and gas hydrates at liquid nitrogen temperature, into the anvil package



Collaborations: Ga Tech/ORNL

With Costas Tsouris, Joint ORNL/GA Tech Faculty Department of Civil and Environmental Engineering

Methane Recovery from Hydrate-Bearing Sediments

With J. Carlos Santamarina GA Tech Faculty

Coalbed Methane Produced Water Treatment by Hydrate Formation and Dissociation





Collaborations: University of Oklahoma Megan Elwood-Madden



Geologic indicators of gas hydrate formation and dissociation in the lab and field- examining ~65 million year old carbonate seeps for geochemical and sedimentological traces of gas hydrates

Funded through Oak Ridge Associated Universities

Geochemistry of shales associated with methane seep mounds

Analyzing: Oxidation state C, O, H, and S isotopes Carbonates Redoxgradients (below)



High Pressure in the Paris-Edinburgh Cell

-room *T*, 2 - 20 GPa (encapsulated gasket ensures isostatic to 10GPa)
 -with micro-furnace, *P*_{max} 7 GPa at 1200K
 -with WC anvils 2.0 to 20 GPa
 -withsintered diamond anvils to 30 GPa
 100 mm³ sample volume





10 GPa Paris-Edinburgh Pressure cell in a dedicated liquid N₂/CCR cooler

26 K

