

Novel Inorganic/Polymer Composite Membranes for CO₂ Capture DE-FE0007632

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The Ohio State University**

Steve Schmit, Director of R&D

Gradient Technology

**FE0007632 Continuation Application Status Meeting
NETL, Pittsburgh, PA, August 7, 2014**

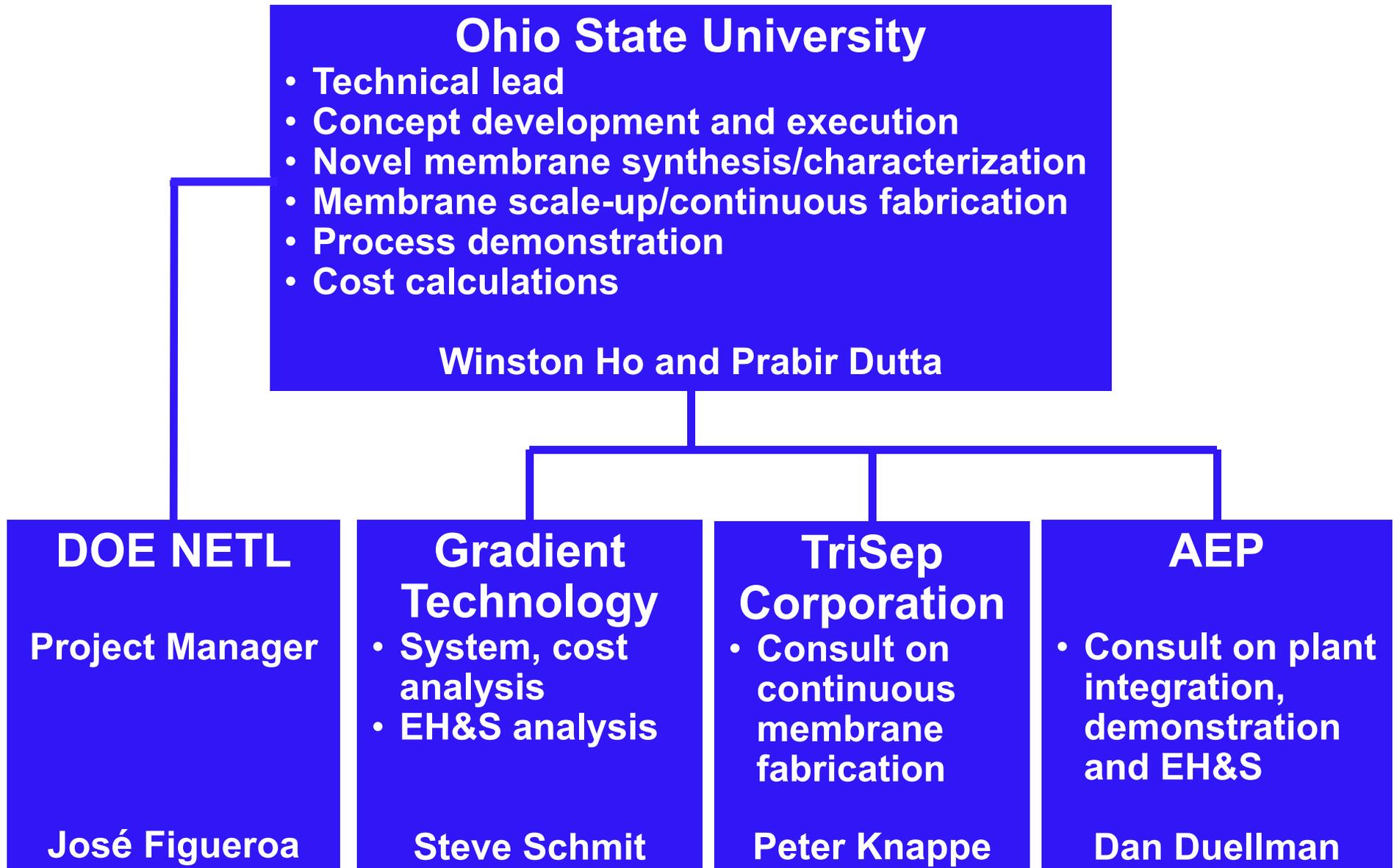
Outline

- **Project Objective and Scope**
- **Membrane Approaches**
- **BP1 Accomplishments**
- **BP2 Accomplishments**
- **Membrane and Process Performance**
- **Techno-Economic Analysis Results**
- **Proposed BP3 SOPO**
- **Closing Statement**

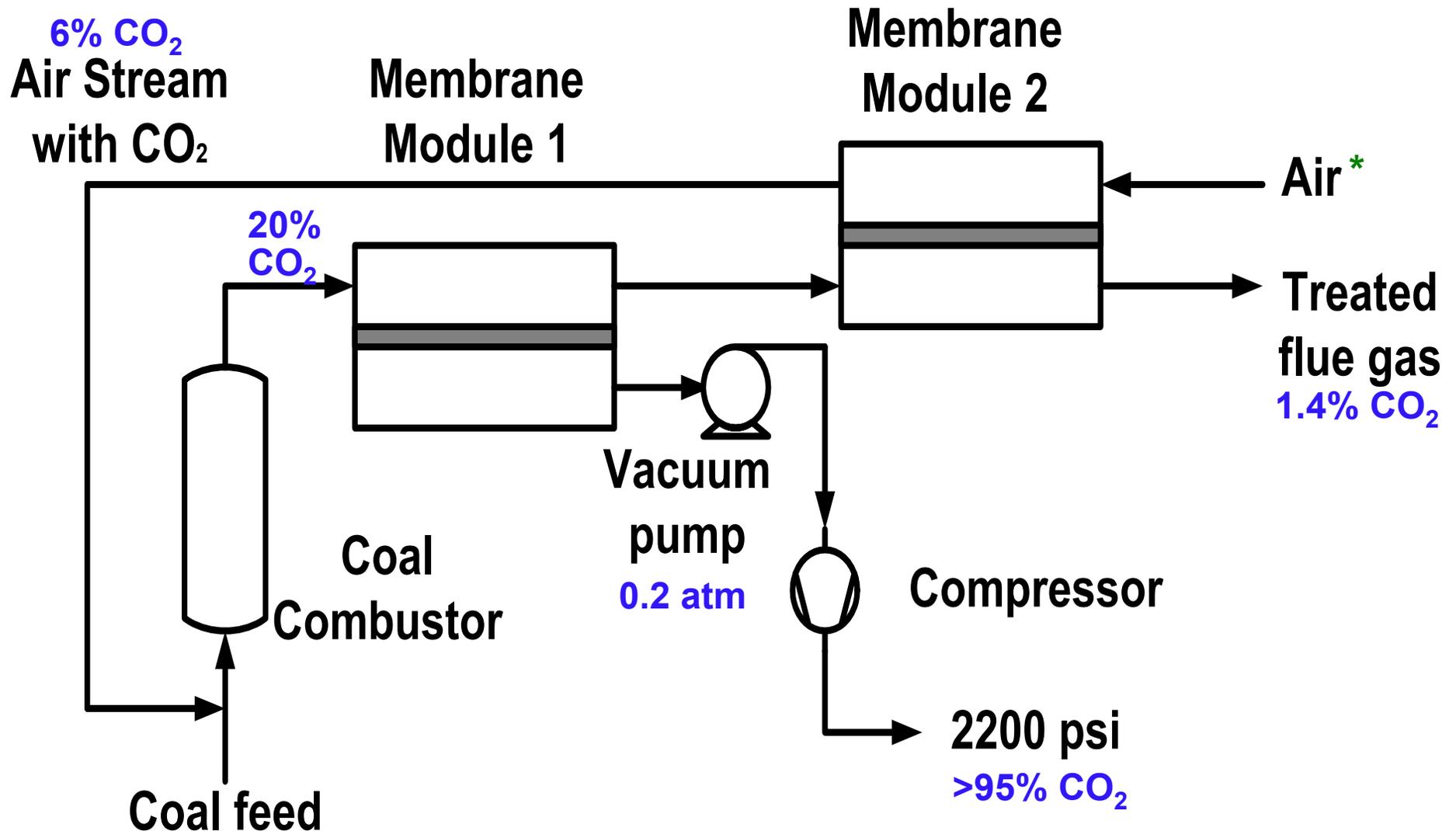
Project Objective

- **Develop a cost-effective design and manufacturing process for new membrane modules that capture CO₂ from flue gas**
 - **BP1**
 - Bench scale membrane synthesis, characterization, downselection, and gas separation performance
 - Preliminary techno-economic analysis
 - **BP2**
 - Bench scale membrane synthesis, characterization and gas separation performance to continue
 - Continuous membrane fabrication
 - Membrane module testing in lab (CO₂, N₂, MOISTURE)
 - Update techno-economic analysis
 - **BP3**
 - 3 prototype modules for testing with simulated flue gas
 - Update techno-economic analysis
 - EH&S evaluation report will be developed
- **Comprehensive program with fundamental studies, applied research, synthesis, characterization and transport studies, scale-up, techno-economic analysis, and EH&S**

Project Organization and Roles



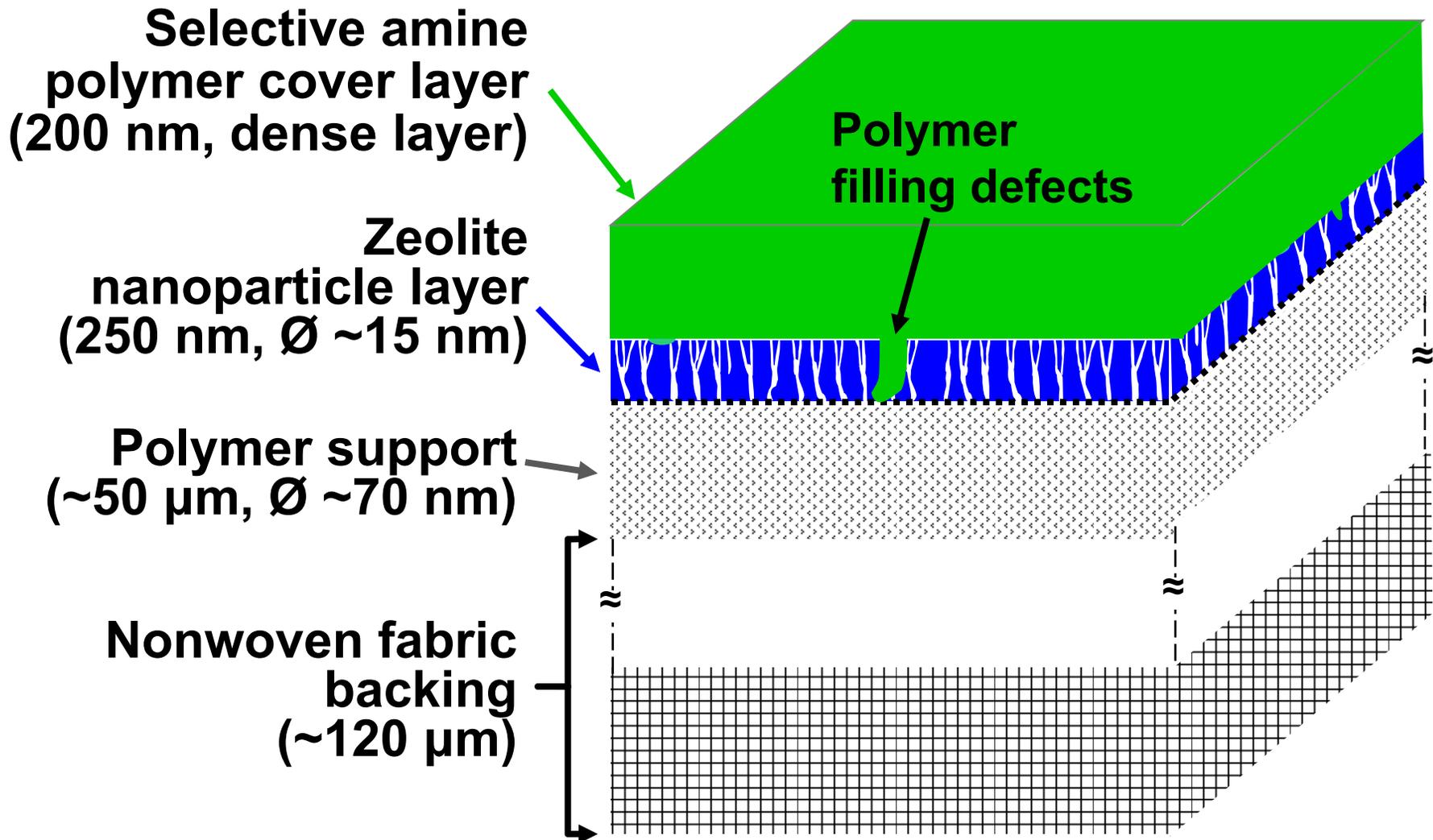
Process Proposed for CO₂ Capture from Flue Gas in Coal-Fired Power Plants



*Air Sweep first used by MTR

Approach 1: Selective Amine Polymer Layer / Zeolite Nanoparticle Layer / Polymer Support

High Inorganic Performance and Low-Cost Polymer Processing Benefits



Approach 1: Selective Amine Polymer Layer / Zeolite Nanoparticle Layer / Polymer Support

- **Selective Amine Polymer Layer**

- Facilitated transport of CO₂ via reaction with amine



- High CO₂ permeance and CO₂/N₂ selectivity

- **Zeolite Nanoparticle Layer**

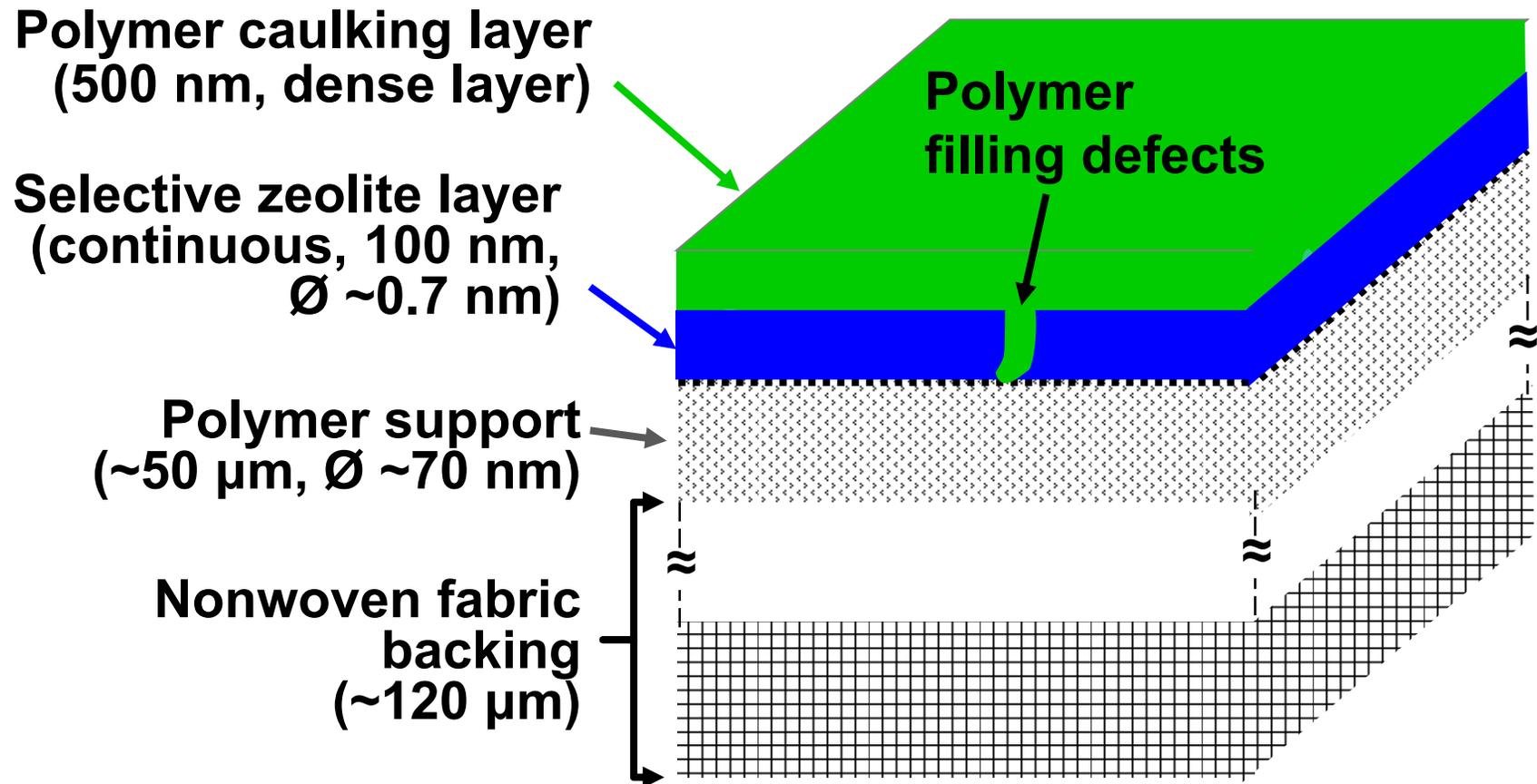
- Increased porosity

- Reduced pore size → Thinner selective amine layer

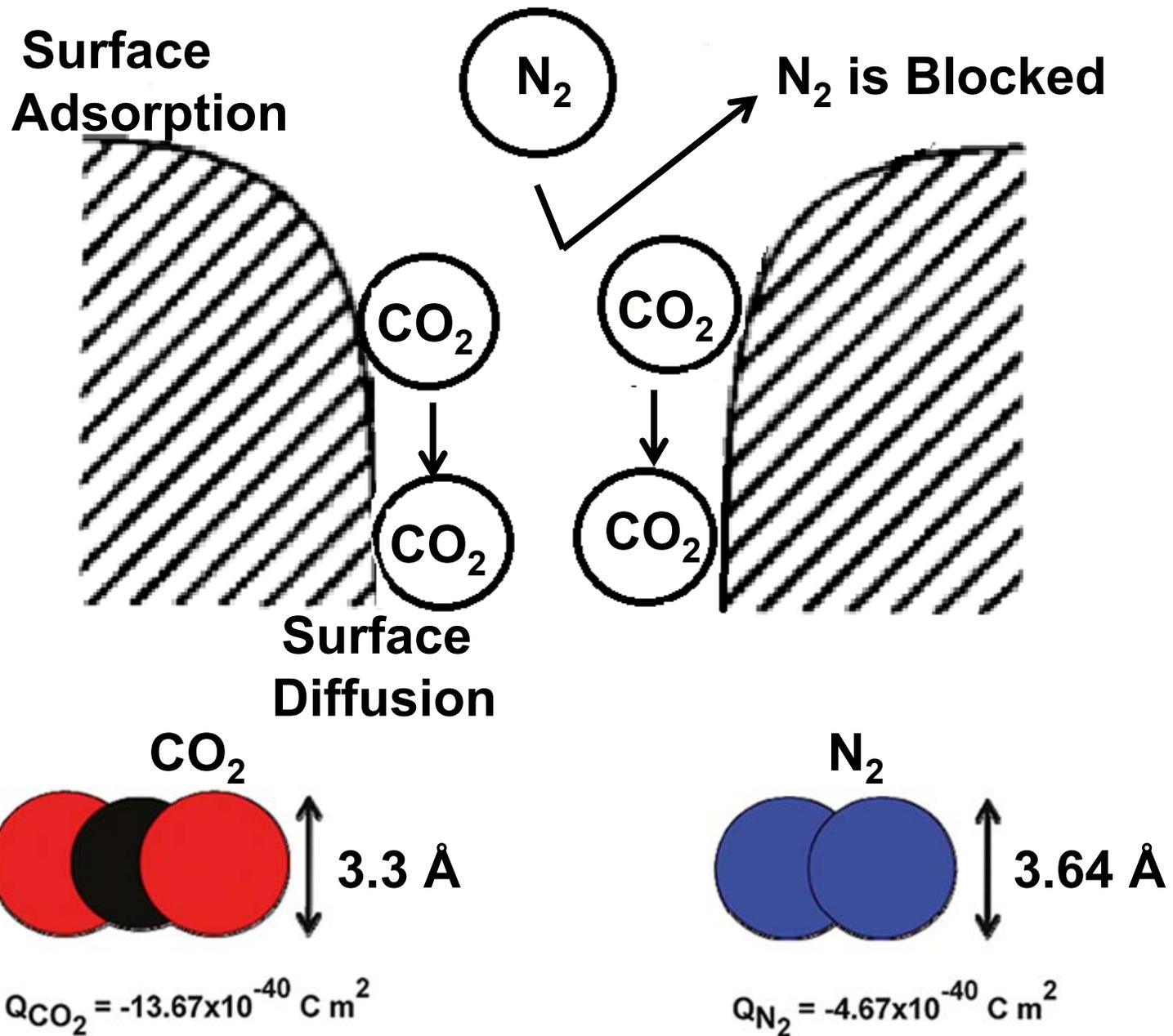
- Higher CO₂ permeance

Approach 2: Polymer Caulking Layer / Selective Zeolite Membrane / Polymer Support

High Inorganic Performance and
Low-Cost Polymer Processing Benefits



Approach 2: Transport Mechanism through Zeolite



BP1 Accomplishments

- **Approach 1: Zeolite/Amine Polymer Composite Membranes Synthesized and Showed:**
 - 1100 GPU with ~800 CO₂/N₂ selectivity at 102°C
 - 690 GPU with 123 CO₂/N₂ selectivity at 57°C
 - Zeolite/polymer element hand rolled successfully (6" x 6" membrane leaf)
- **Approach 2: Significant Membrane Synthesis Improvements**
 - Discovery of rapid zeolite particle synthesis (< 1 hr vs. 8 hrs)
- **Preliminary Techno-economic Calculations**
 - Techno-economic model developed
 - 690 GPU with ~123 selectivity at 57°C (based on 2007\$)
 - ~\$43/tonne CO₂

BP2 Accomplishments

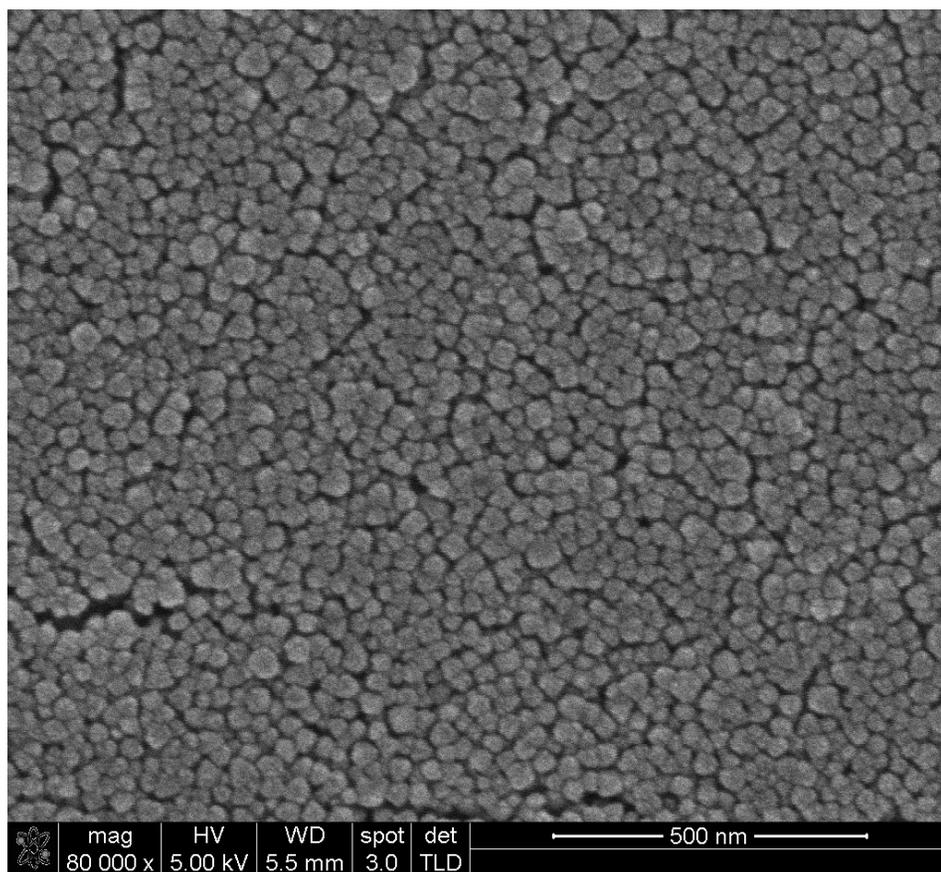
- **Approach 1: Zeolite/Amine Polymer Composite Membranes Prepared in Lab Showed:**
 - 1100 GPU with ~140 CO₂/N₂ selectivity at 57°C
 - 1460 GPU with >1000 CO₂/N₂ selectivity at 102°C
 - Patent application filed
- **Approach 1: Composite Membrane Scaled up to Prototype Size**
 - Membrane scaled up to 14” wide using continuous membrane rolling machine
 - 870 GPU with 218 CO₂/N₂ selectivity obtained at 57°C
 - 1800 GPU with 160 CO₂/N₂ selectivity obtained at 102°C
 - Developed affordable nanoporous polymer support (PES)
 - 1.8” (1.5” OD central tube) by 14” long spiral-wound membrane elements fabricated using rolling machine

BP2 Accomplishments (continued)

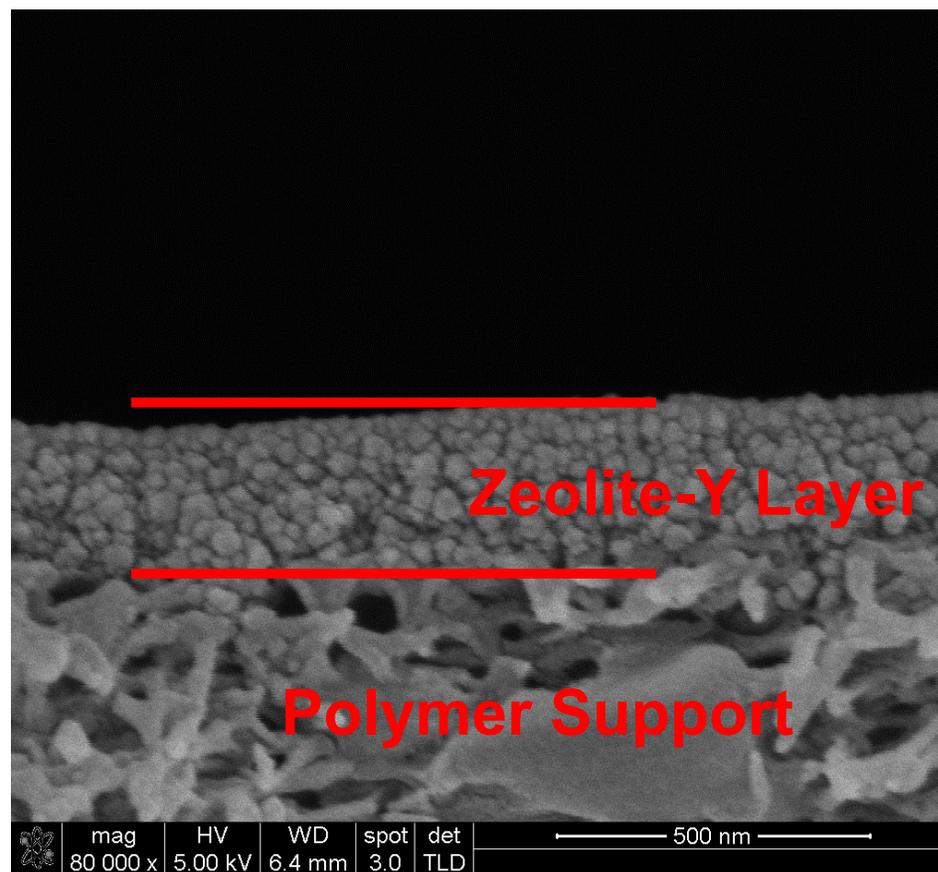
- **Preliminary Techno-economic Calculations showed**
 - **1100 GPU with ~140 selectivity at 57°C (based on 2007\$)**
 - \$37.5/tonne CO₂ – Exceed DOE target of \$40/tonne CO₂
 - 52.2% COE increase
- **Approach 2: Rapid Zeolite Membrane Growth (1 hour)**
 - Patent application filed
 - Published in *Langmuir*, 2014, 30, 6929-6937
- **Effects of SO₂ and CO₂/SO₂ Mixture on Amine Carriers being Studied by in-situ FTIR**
 - SO₂ permeated with CO₂
 - Amine regenerated by air sweep at 57°C – Confirmed by in-situ FTIR

Approach 1: Zeolite Nanoparticles Deposited on Polymer Support Successfully

Top View

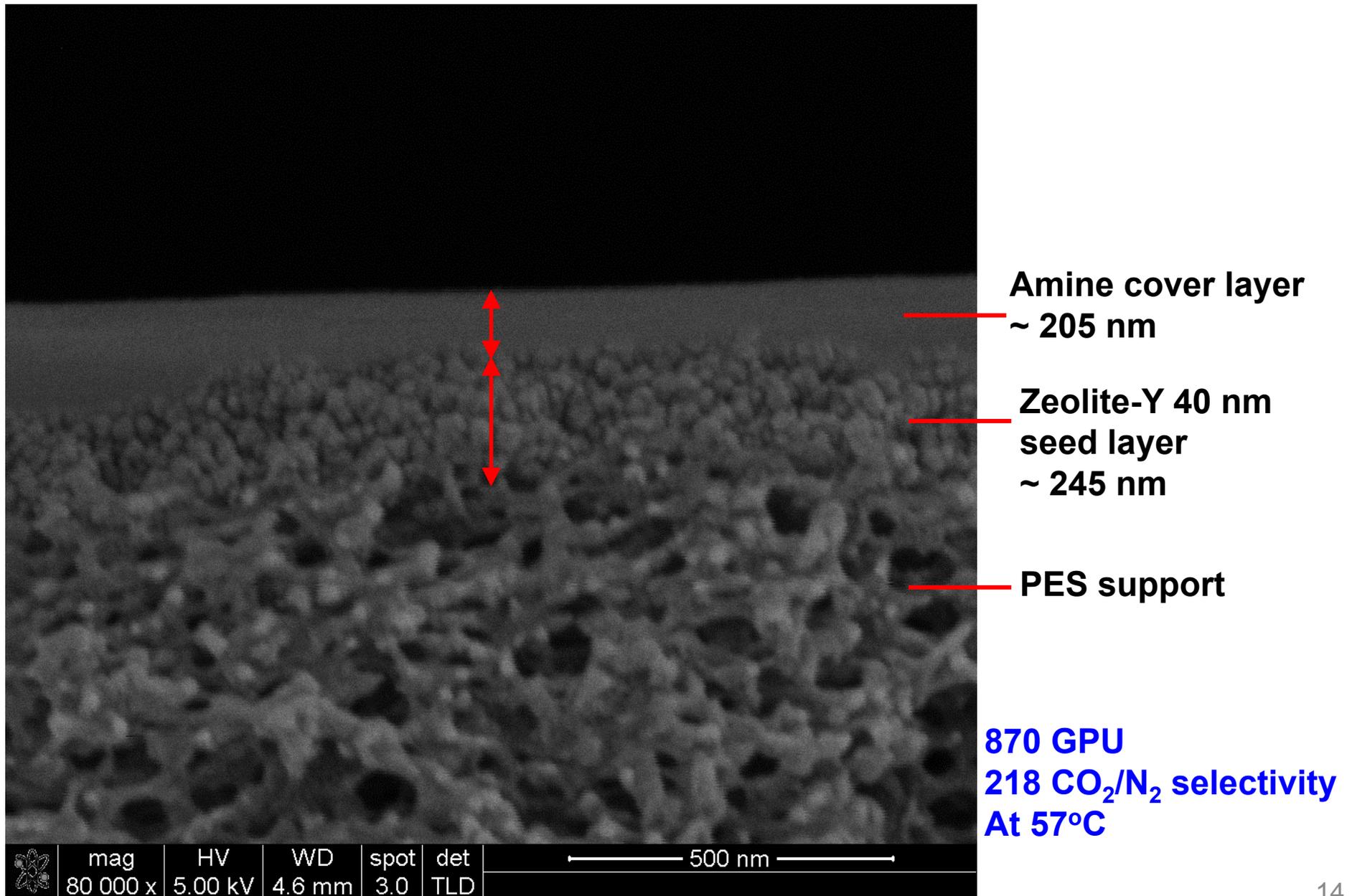


Cross-section

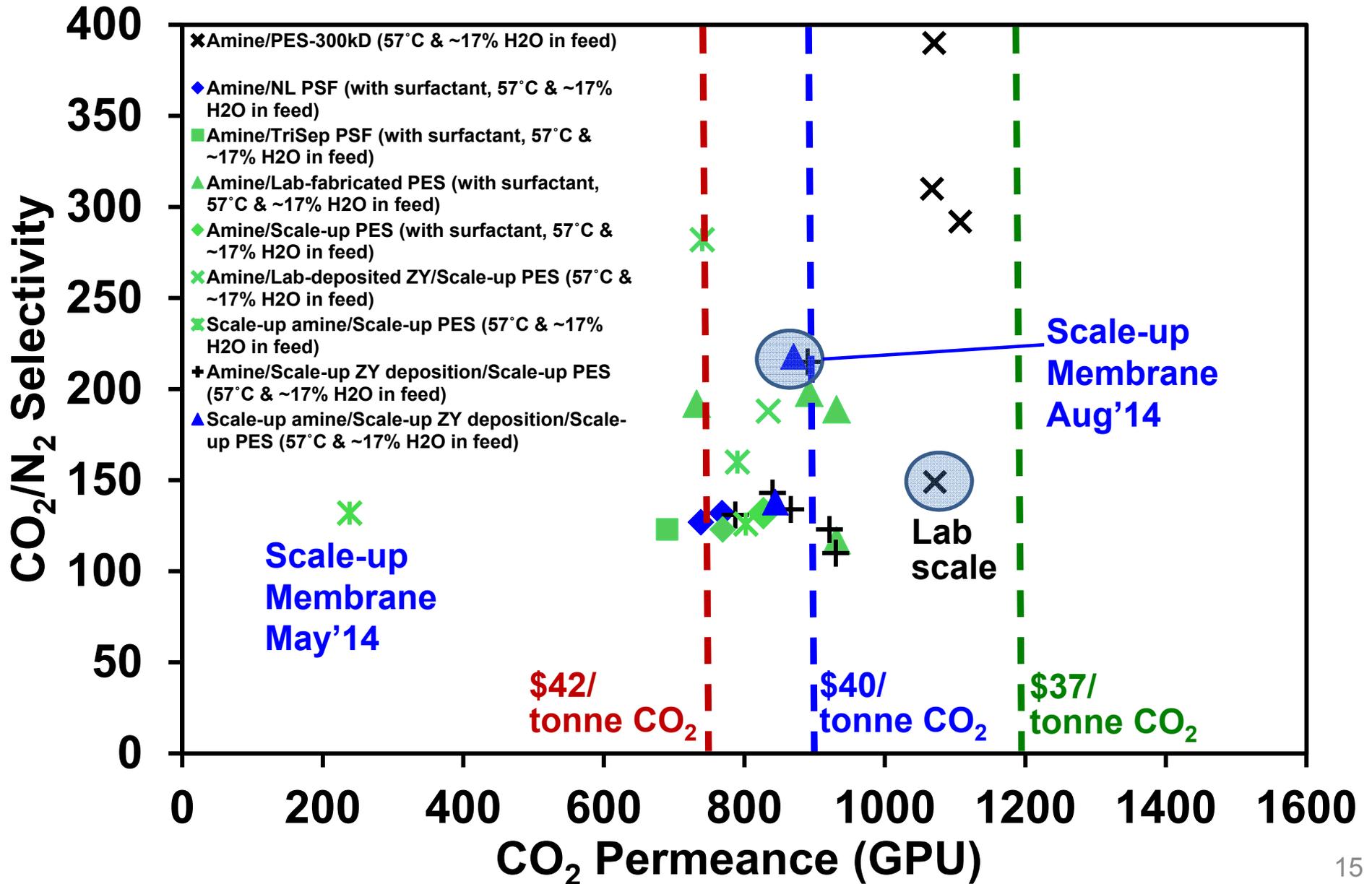


- High quality deposition with good repeatability

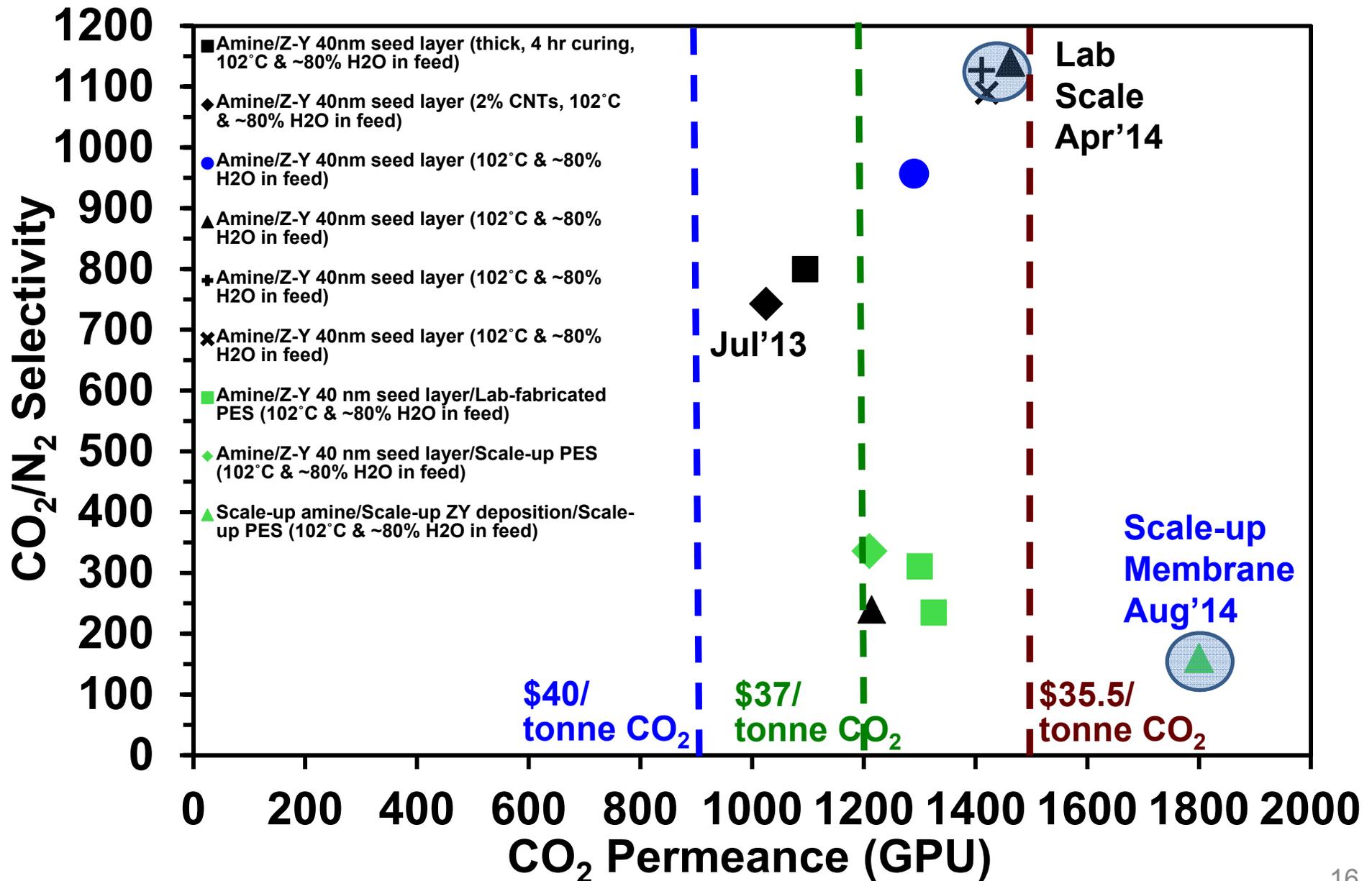
Amine/Zeolite Seed Layer/Polymer Support



Approach 1: Zeolite/Polymer Composite Membranes Containing Amine Cover Layer at 57°C



Approach 1: Zeolite/Polymer Composite Membranes Containing Amine Cover Layer at 102°C



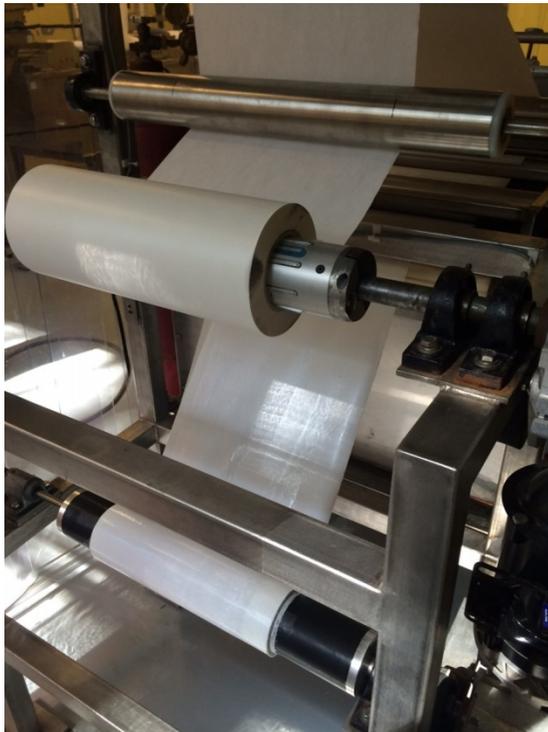
Membrane Scale-up: Usable for Approaches 1 and 2

Continuous Membrane Fabrication Machine at OSU



Successful Continuous Fabrication of Affordable PES Support (applicable to Approaches 1 and 2)

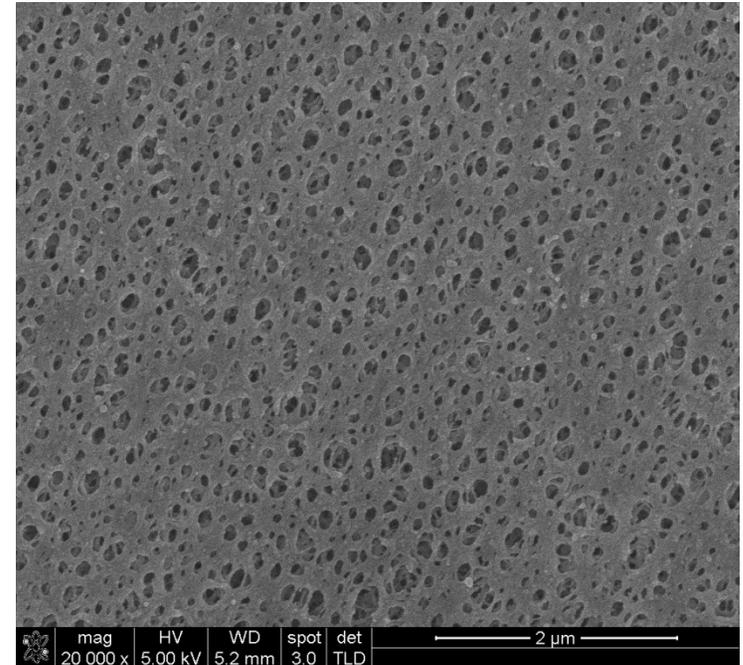
Casting Machine



14-inch PES Support



SEM – Top View



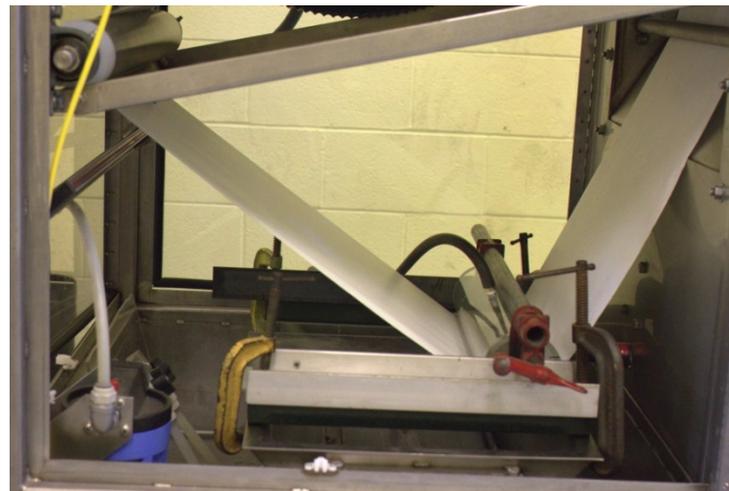
- **Manufacturer could not supply PES needed for scale-up**
- **PES synthesized/developed at OSU to resolve supply issue**
- **Technology transfer to TriSep**

Approach 1: Scale-up Zeolite-Y Deposition and Amine Coating

14-inch PES Support



14" ZY Deposition on PES Support



14" Amine Coating on ZY Layer on PES

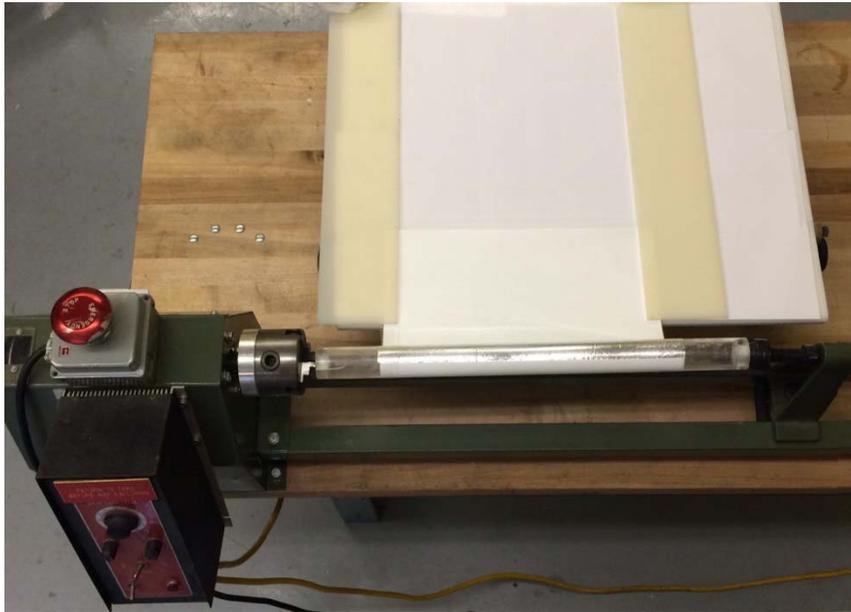


Approach 1: Membrane Element Fabrication

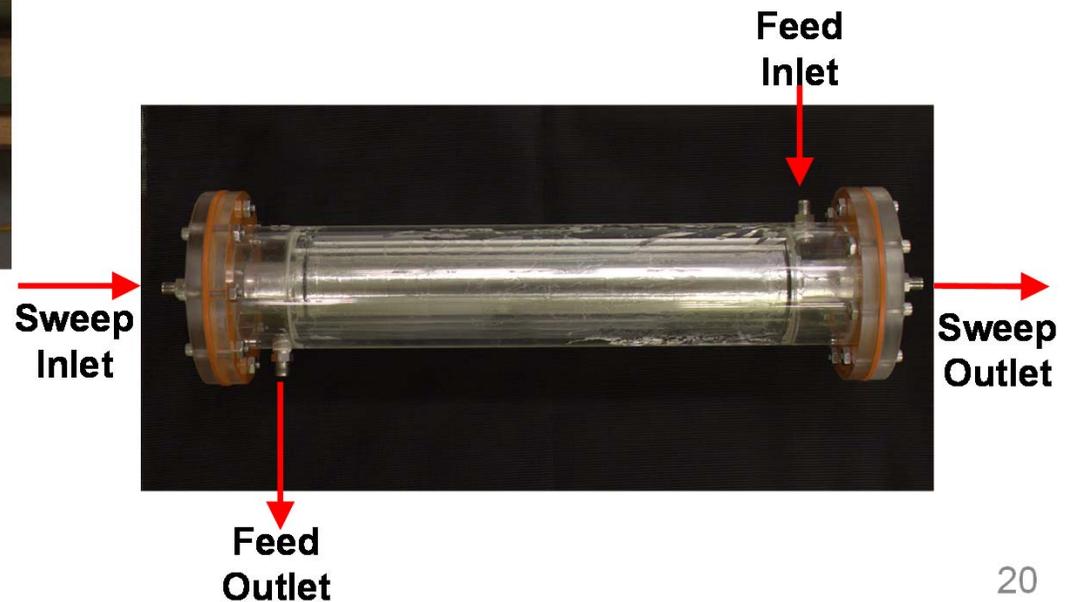
Spiral-Wound Membrane Element



Element Rolling Machine



Membrane Module

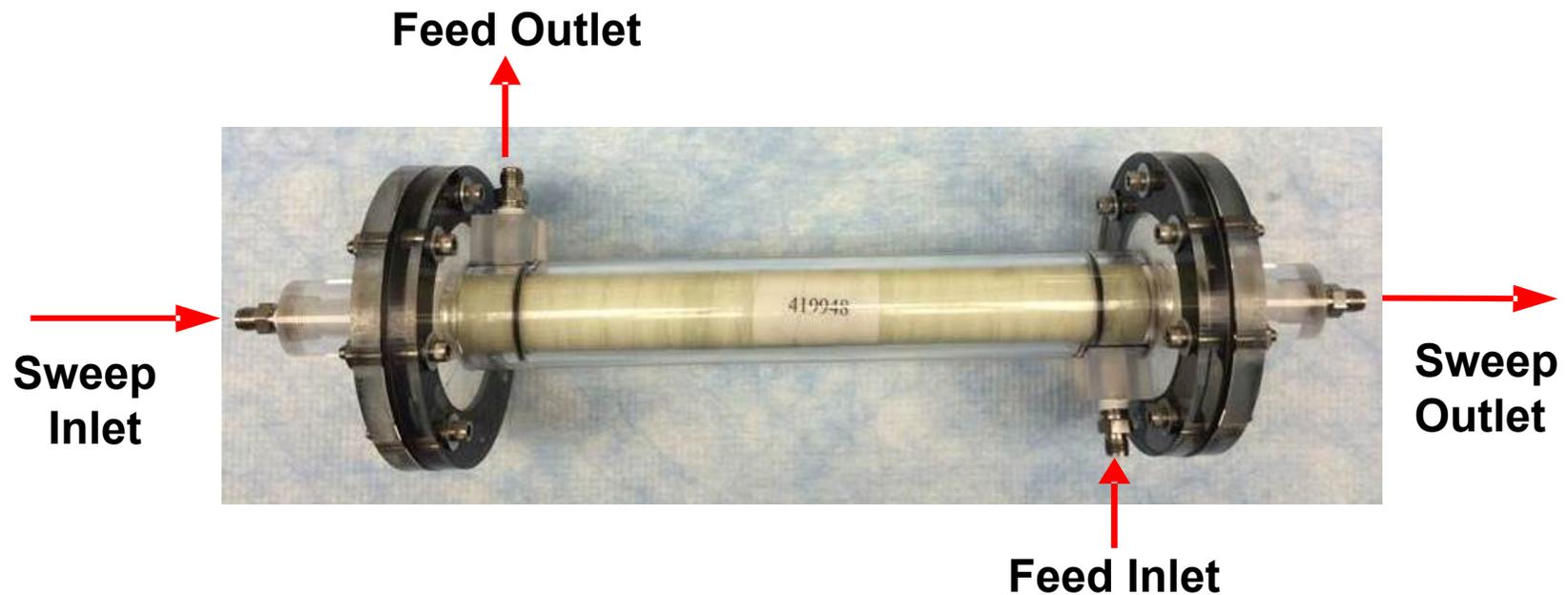


Approach 1: TriSep also Made Elements for us

Spiral-Wound Membrane Element Made by TriSep



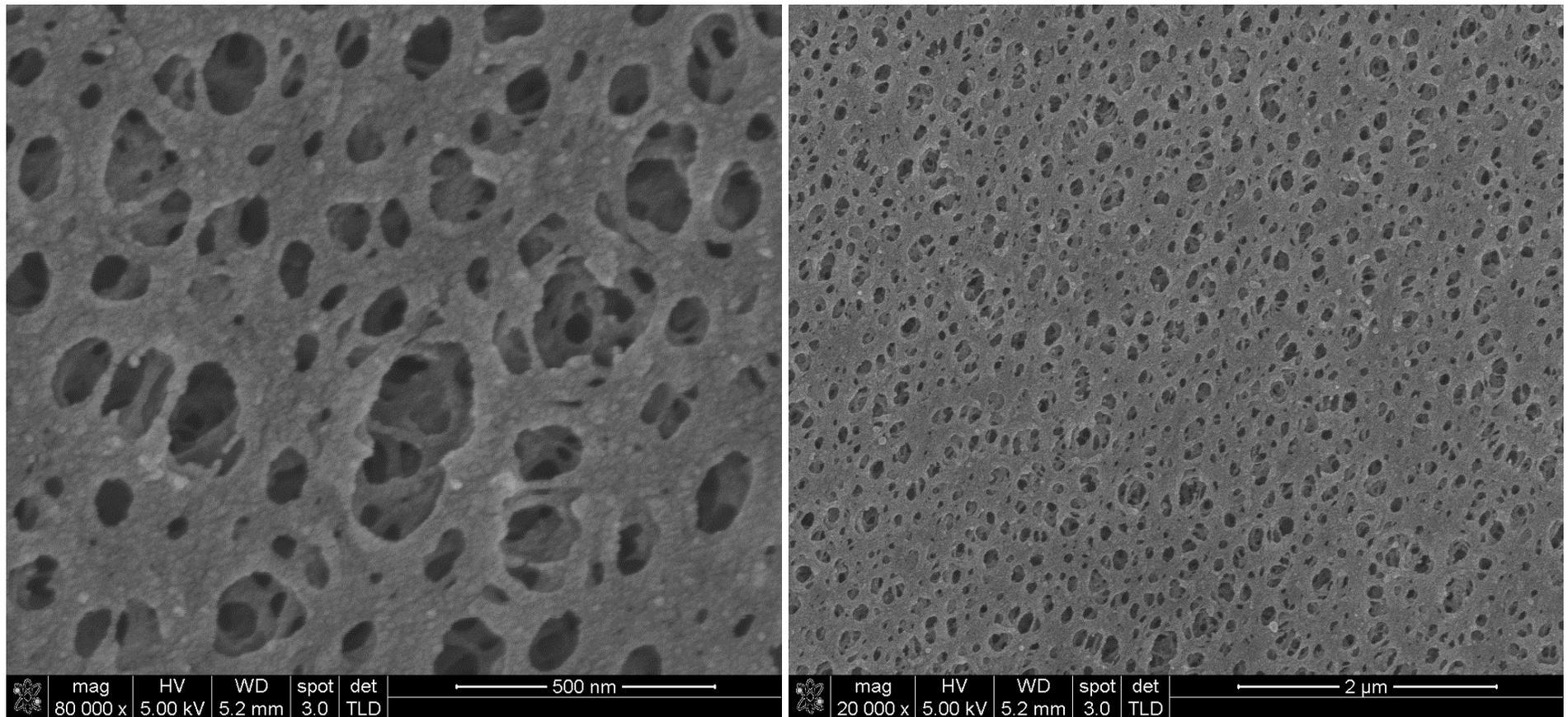
Membrane Module: Element Made by TriSep in our Housing



Technical Details – BP2

Successful Continuous Fabrication of Affordable PES Support (applicable to Approaches 1 and 2)

SEM Analysis of 14-inch PES Support



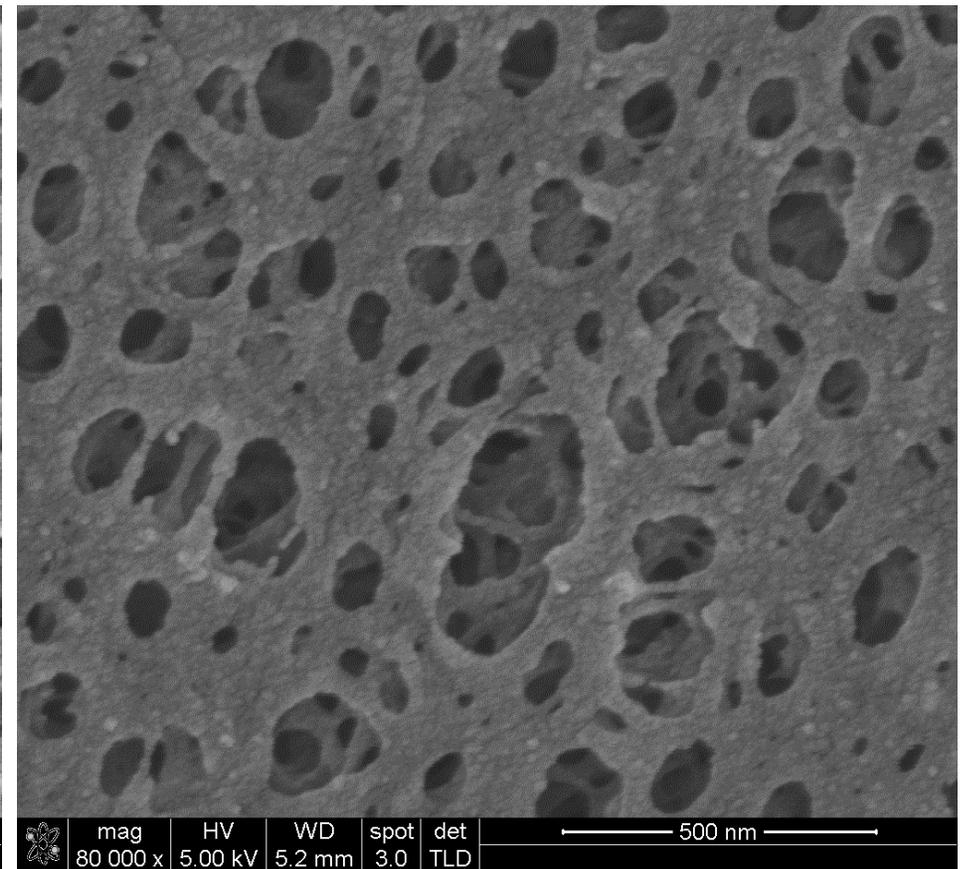
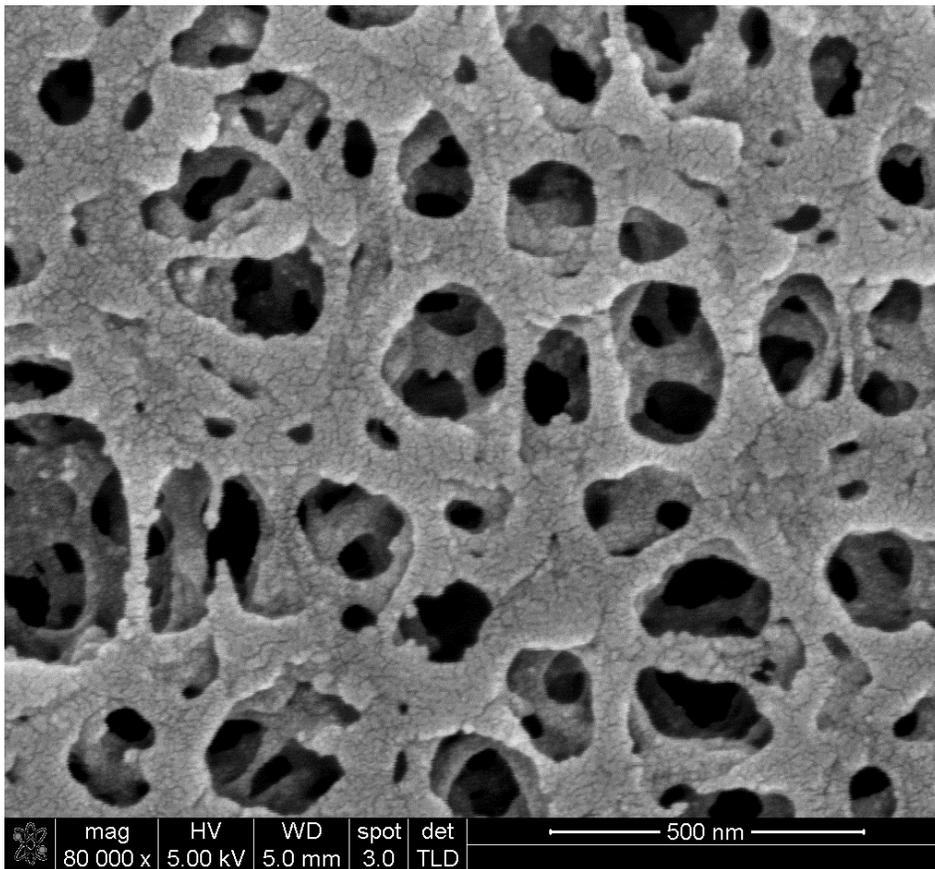
Ave. pore size = 69.5 nm, Porosity = 16.9%

Comparison between Commercial PES-300kD and 14-inch Scale-up PES

Large magnification (80,000 x)

PES-300kD

14-inch PES

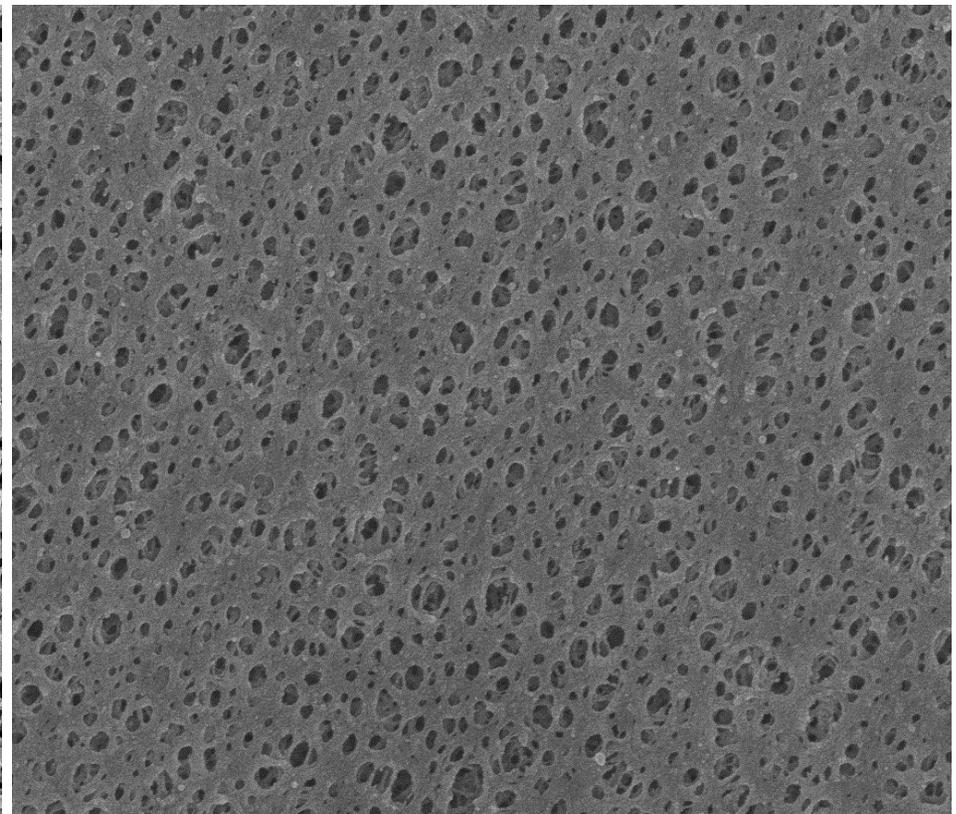
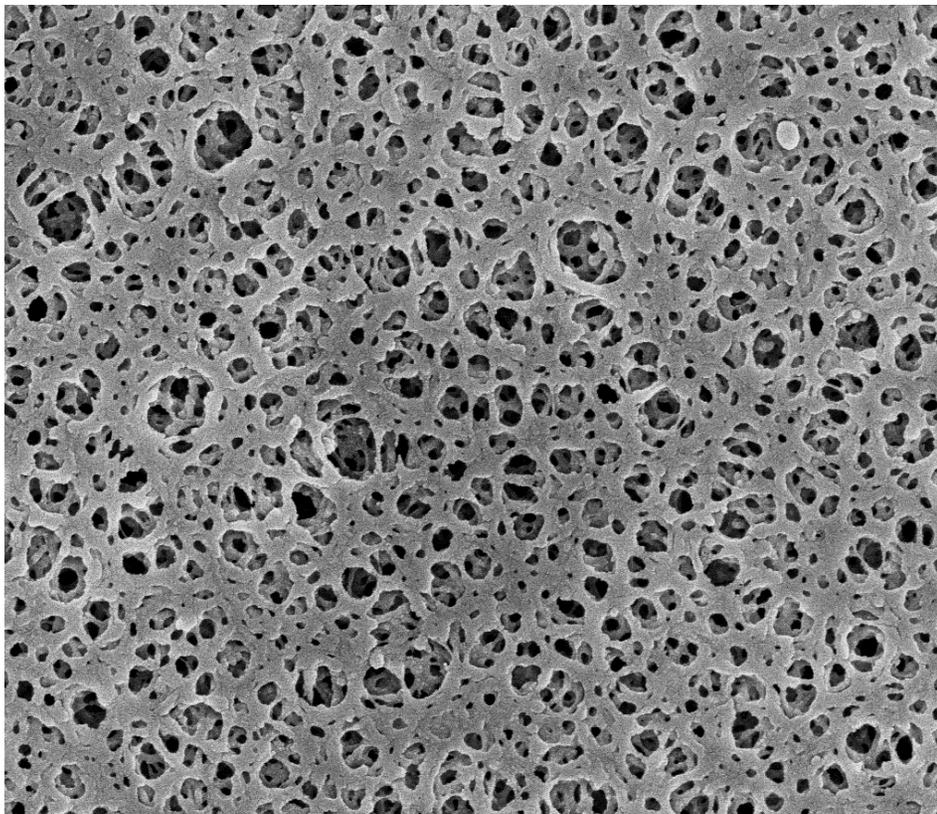


Comparison between Commercial PES-300kD and 14-inch Scale-up PES

Small magnification (20,000 x)

PES-300kD

14-inch PES



mag	HV	WD	spot	det	
20 000 x	5.00 kV	5.0 mm	3.0	TLD	2 μm

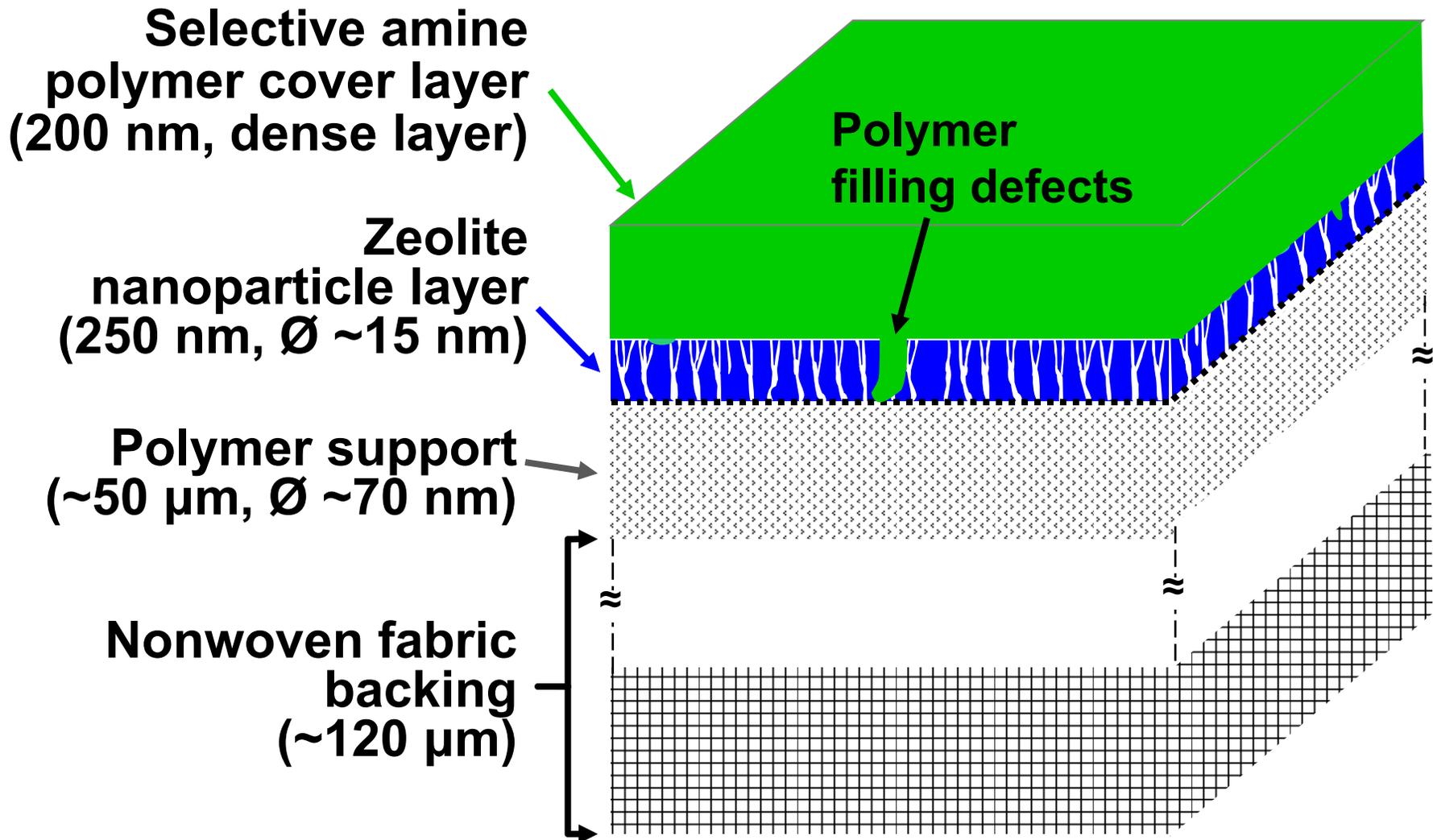
mag	HV	WD	spot	det	
20 000 x	5.00 kV	5.2 mm	3.0	TLD	2 μm

Ave. pore size (nm)	72.3	69.5
Porosity (%)	15.8	16.9
Ave. roughness (nm)	17.0	14.0

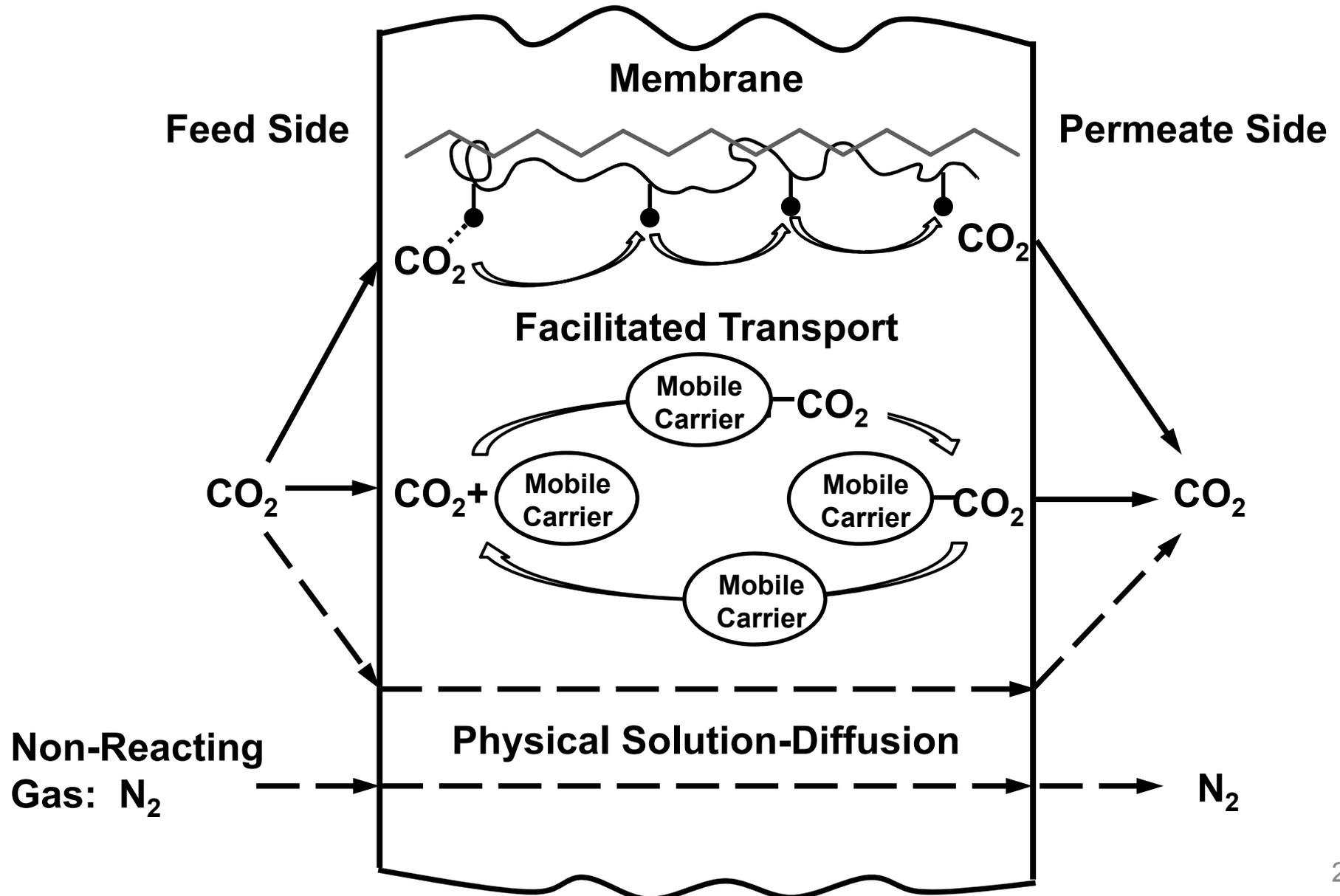
- Support surface morphology close to that of PES-300kD

Approach 1: Selective Amine Polymer Layer / Zeolite Nanoparticle Layer / Polymer Support

High Inorganic Performance and Low-Cost Polymer Processing Benefits



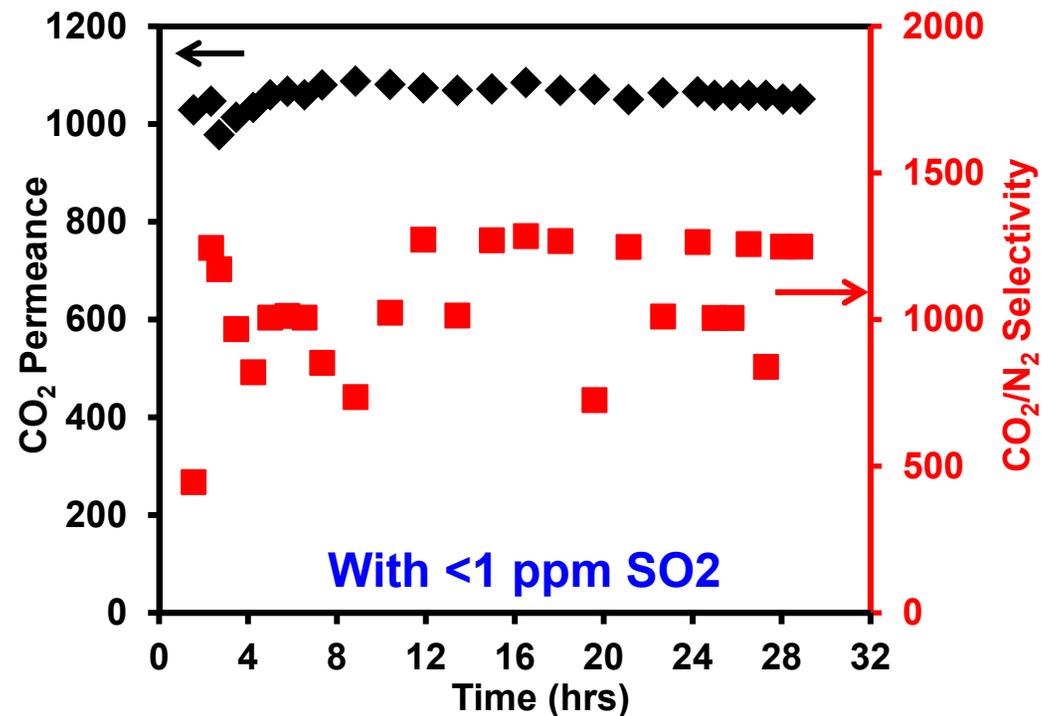
Approach 1: Amine Cover Layer Contains Mobile and Fixed Carriers



Approach 1: SO₂ Effects on Amine-containing Membranes

- **SO₂ Effects**

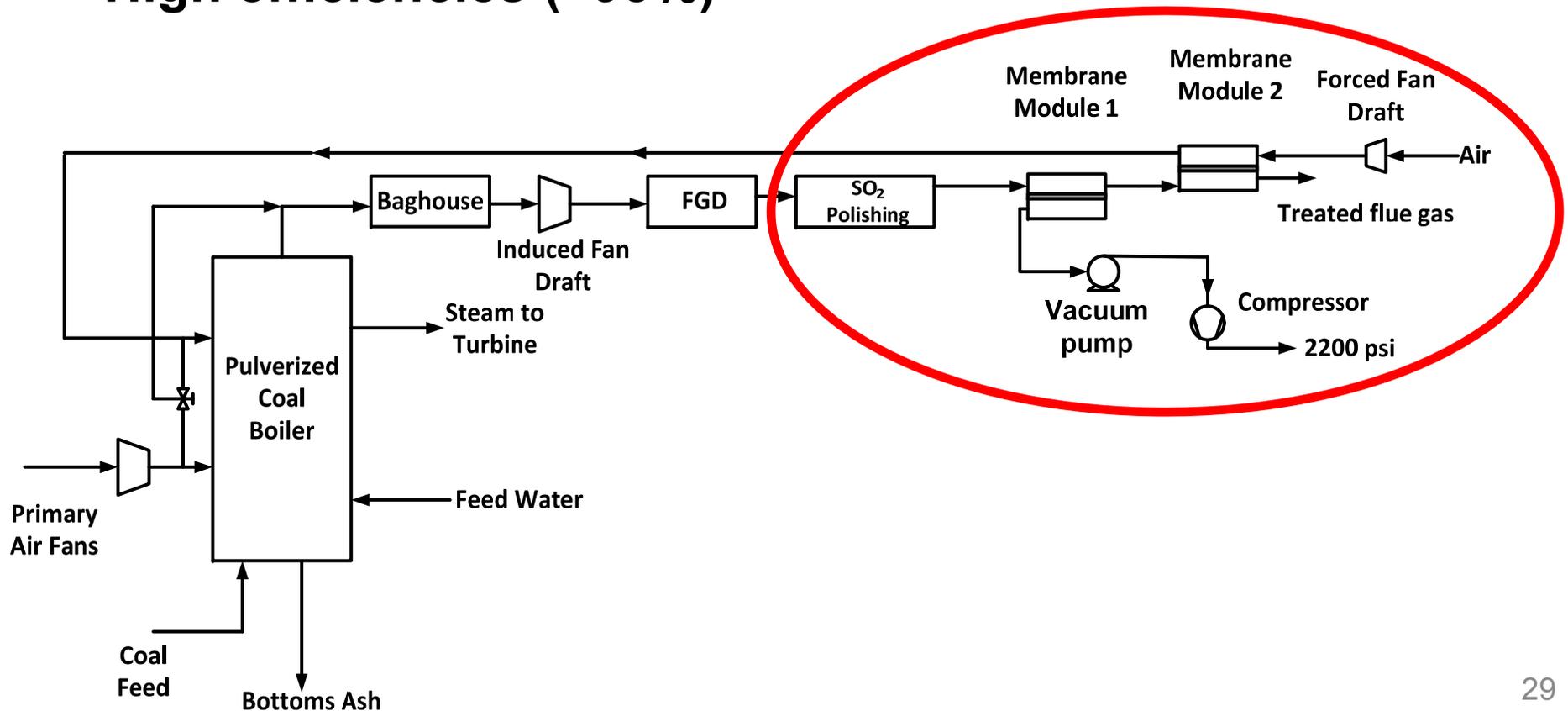
- SO₂ at ≤1 ppm appeared not to affect stability of membrane with amine cover layer
- More study underway between 1 – 43 ppm SO₂



- **Propose SO₂ Polishing Step before membrane**

Approach 1: SO₂ Membrane Mitigation

- **Absorption into 20 wt% NaOH Solution**
 - Polishing step based on NETL baseline document
 - Estimated to be about \$4.3/tonne CO₂ (6.5% COE increase)
 - Non-plugging, low-differential-pressure, spray baffle scrubber
 - High efficiencies (>95%)



Breakthroughs to Make Approach 1 Feasible

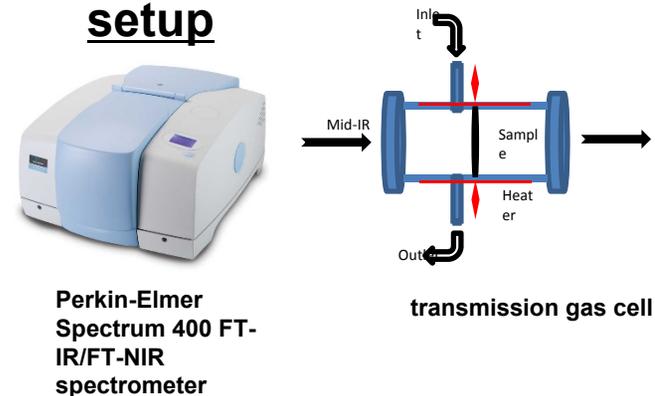
- **Speeding up Synthesis of Nanozeolites**
- **Atomic Force Microscopy for Optimization of PES Support**
- **Influence of SO₂ on Membrane Components**

Approach 1: Interaction of SO₂ with Membrane

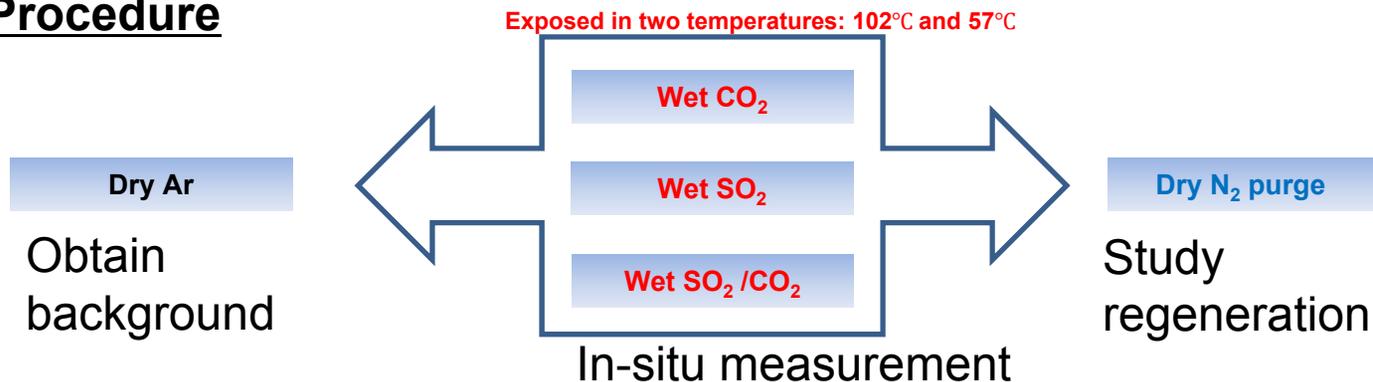
Objectives

- Investigate each component separately and then their combinations
- Understand interaction of CO₂/SO₂ with amine membranes
- correlate the membrane performance/degradation in the presence of SO₂ and suggest ways to circumvent any negative effects

Experimental setup



Procedure



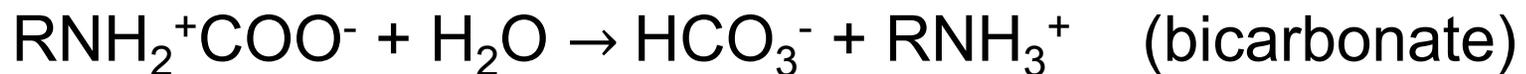
Approach 1: Chemical Reactions of CO₂/SO₂ with Membrane Components

- Reactions with CO₂

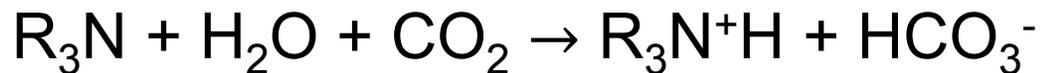
For primary and secondary amine



If sterically hindered,



For tertiary amine

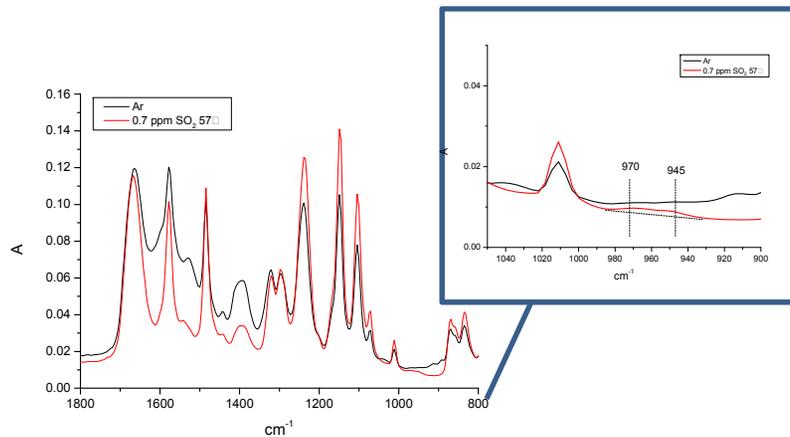


- Reactions with SO₂



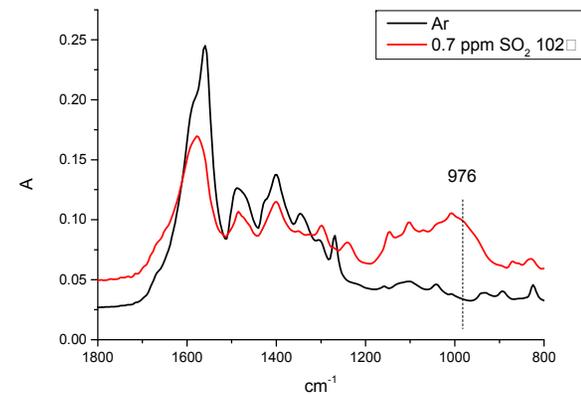
Approach 1: SO₂ Exposed Membranes Characterized by Infrared Spectroscopy

Two sets of membrane samples



- Membrane 1 exposed to 0.7 ppm SO₂ 57°C

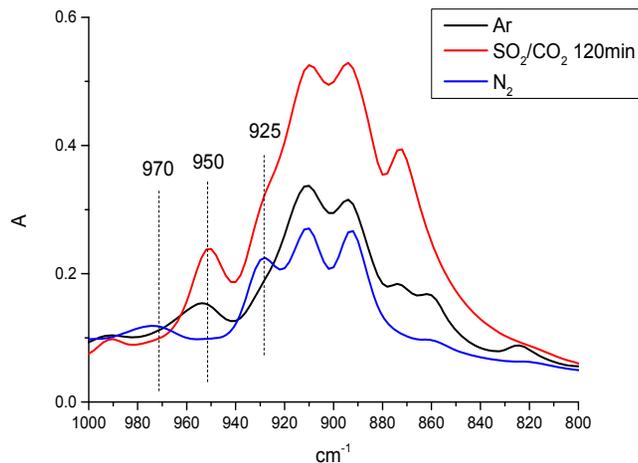
➤ Weak bands for sulfite species



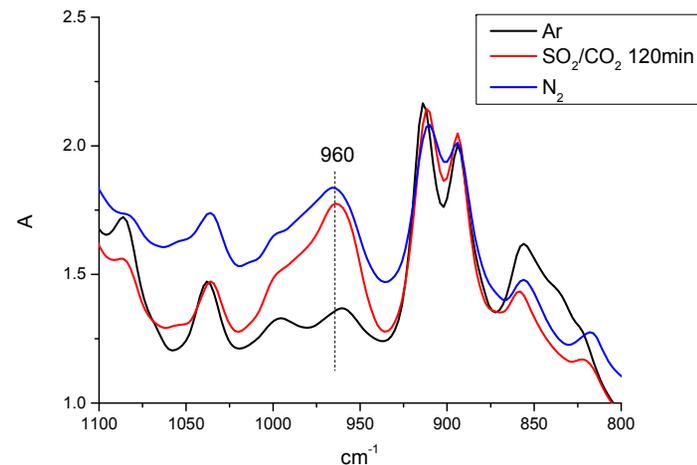
- Membrane 2 exposed to 0.7 ppm SO₂ at 102°C

➤ Strong band of adsorbed sulfite species

Approach 1: IR Spectrum of Amine at 57°C and 102°C (45 ppm SO₂/10% CO₂)

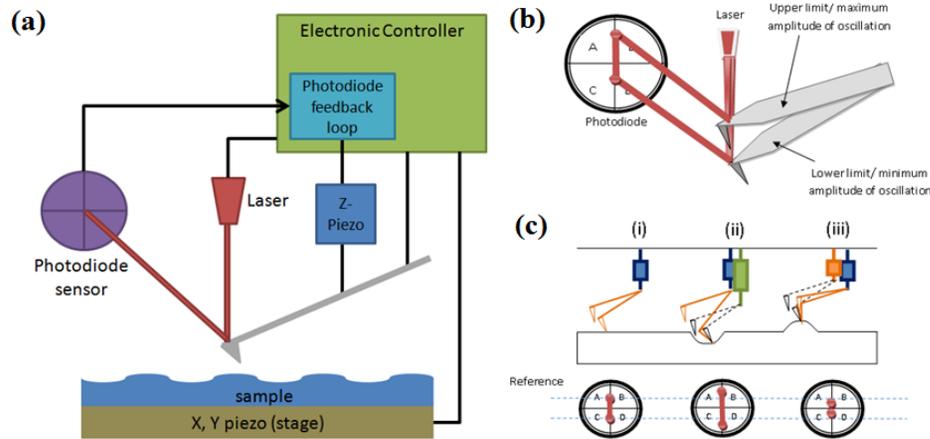


57°C : SO₂ primarily
in adsorbed form,
removed with N₂ or
air purge >1 hour



102°C : SO₂ primarily in
reacted form, not
removed with N₂ or air
purge > 1 hour

Approach 1: Measuring Surface Roughness of PES Synthesized at Ohio State



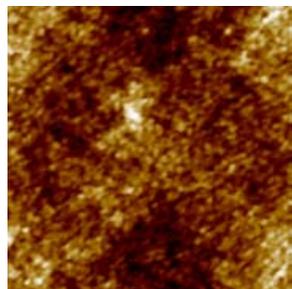
- The AFM generates surface topography (e.g roughness) of membrane
- Roughness measurement
=> membrane quality control

Polymer Characterization with the Atomic Force Microscope

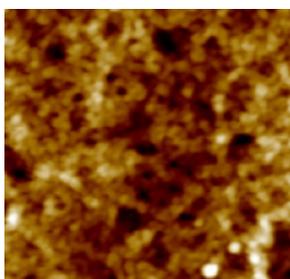
By U. Maver, T. Maver, Z. Peršin, M. Mozetič, A. Vesel, M. Gaberšček and K. Stana-Kleinschek, Polymer Sc, Chapter 4

Approach 1: Support Roughness Characterized by AFM

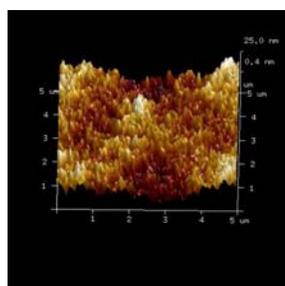
PES support



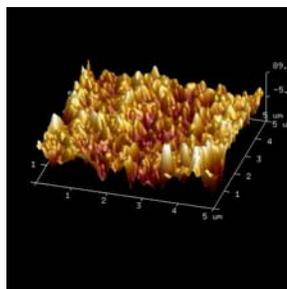
Sample A (top)



Sample B (top)



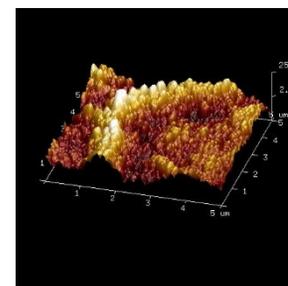
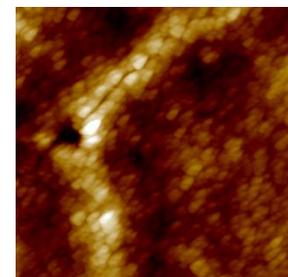
Sample A (3D)



Sample B (3D)

Sample name	Avg roughness
Sample A	8 nm
Sample B	28 nm

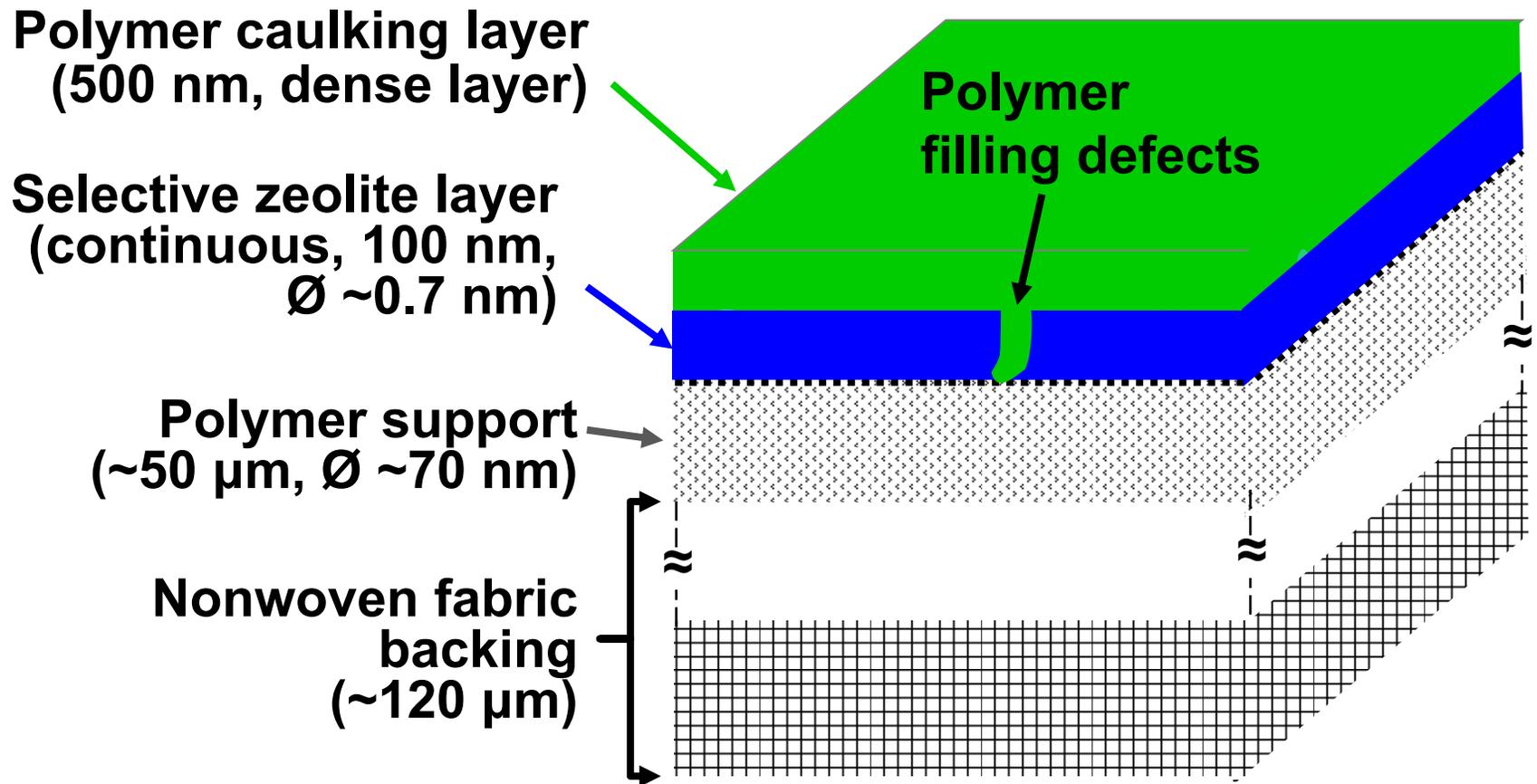
Nanozeolite coated PES membrane



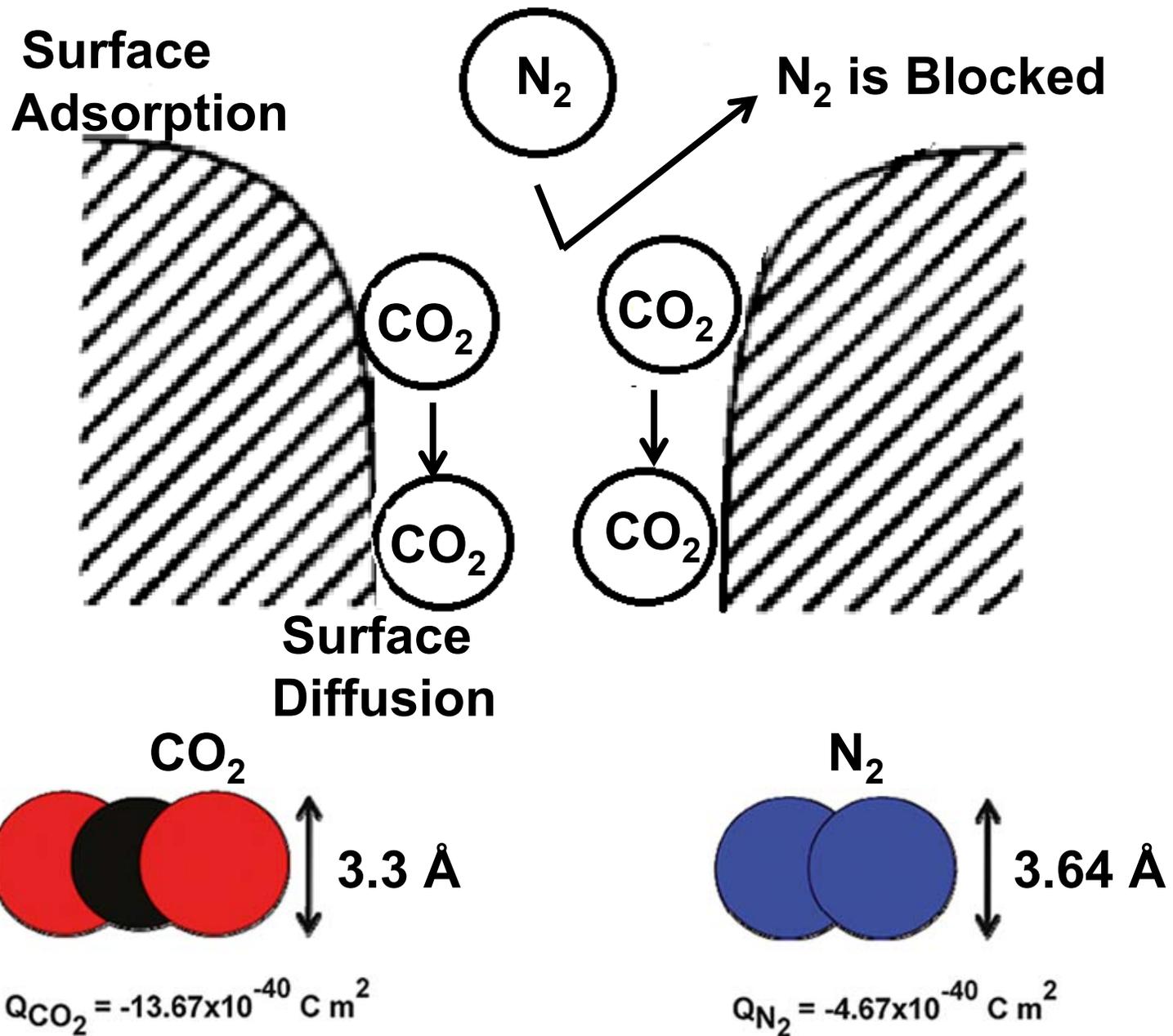
Sample name	Avg roughness
Nano y coated	6 nm

Approach 2: Polymer Caulking Layer / Selective Zeolite Membrane / Polymer Support

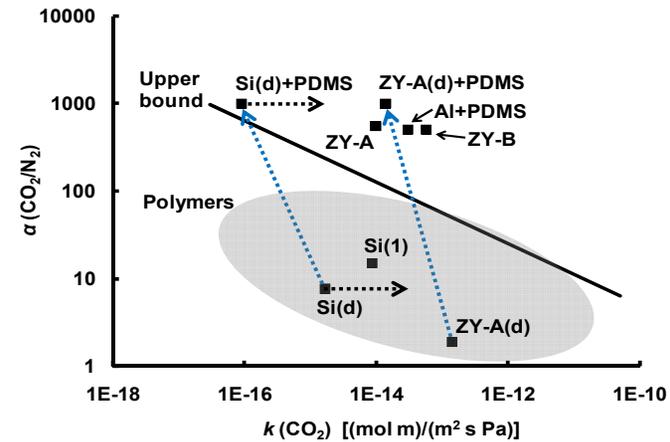
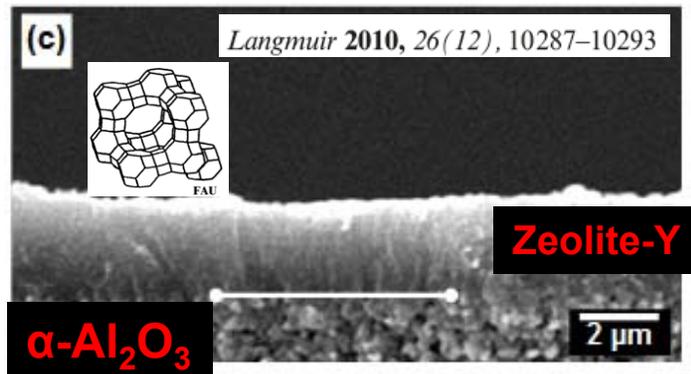
High Inorganic Performance and
Low-Cost Polymer Processing Benefits



Approach 2: Transport Mechanism through Zeolite



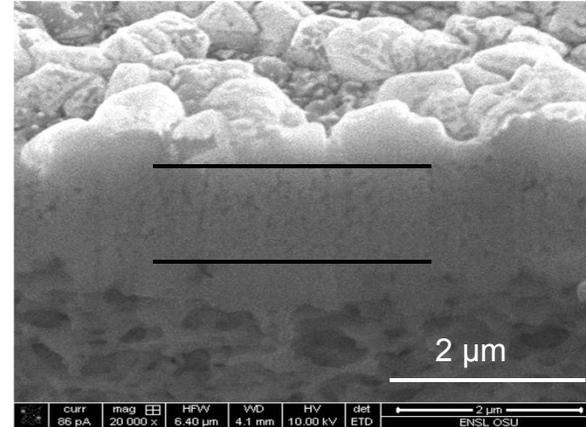
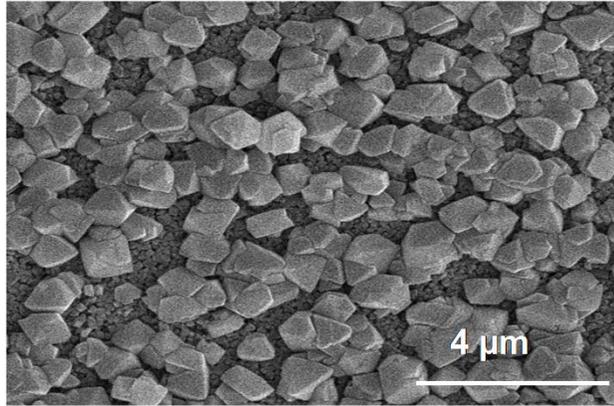
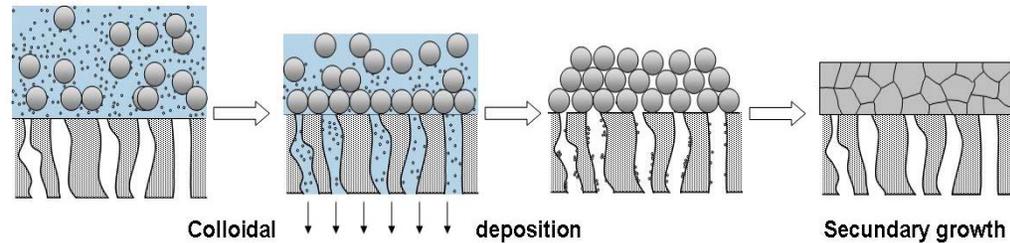
Precedence to Approach 2



Journal of Membrane Science 421–422 (2012) 299–310

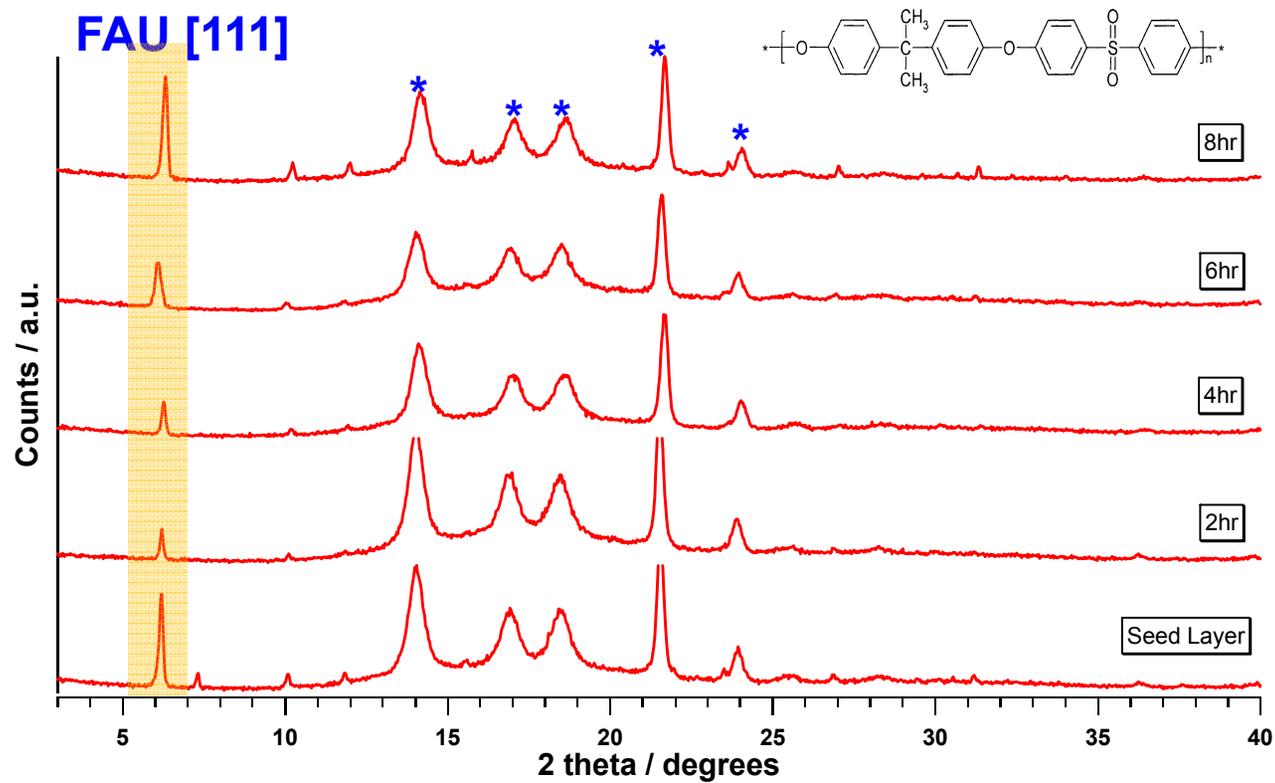
- Zeolite Y on alumina support
- 4-day synthesis
- GPU 328, Permenace 29 (CO_2/N_2)
- Synthetic protocols limit practical application
- Change to Polymer support and speed up zeolite synthesis

Approach 2: First Step – Polymer Support



Growth Process takes 8 hours : Impractical

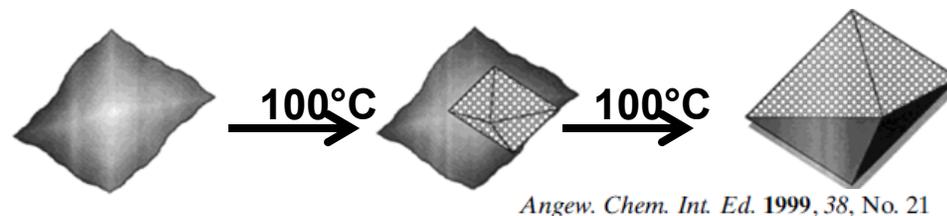
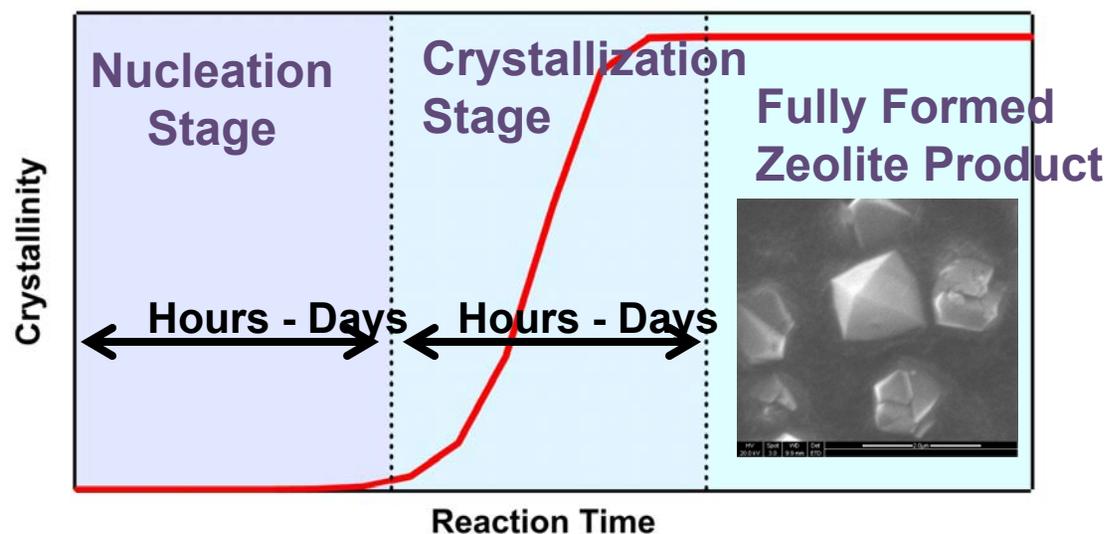
Approach 2: Conventional Zeolite Membrane Growth vs. Time



Poor physical adhesion and partial dissolution limit the membrane quality.

Approach 2: Decrease Zeolite Synthesis Time

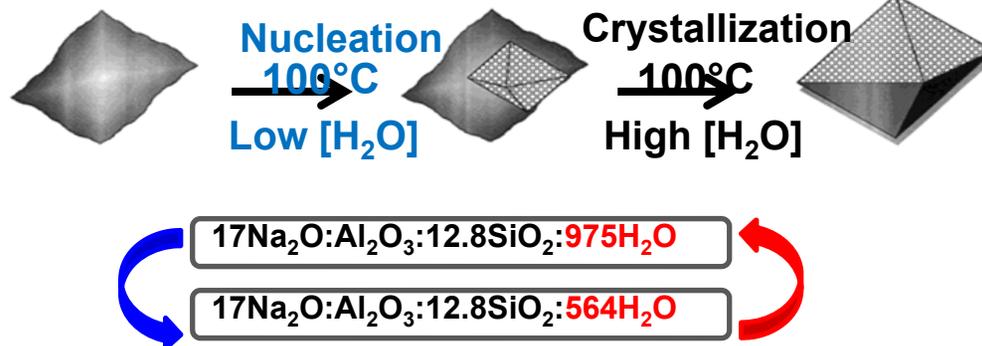
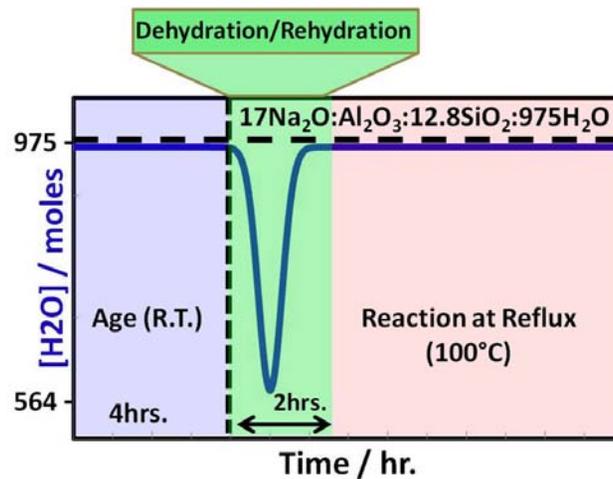
Traditional zeolite synthesis requires long reaction times at high temperatures.



Approach 2: Novel Zeolite Synthesis Approach

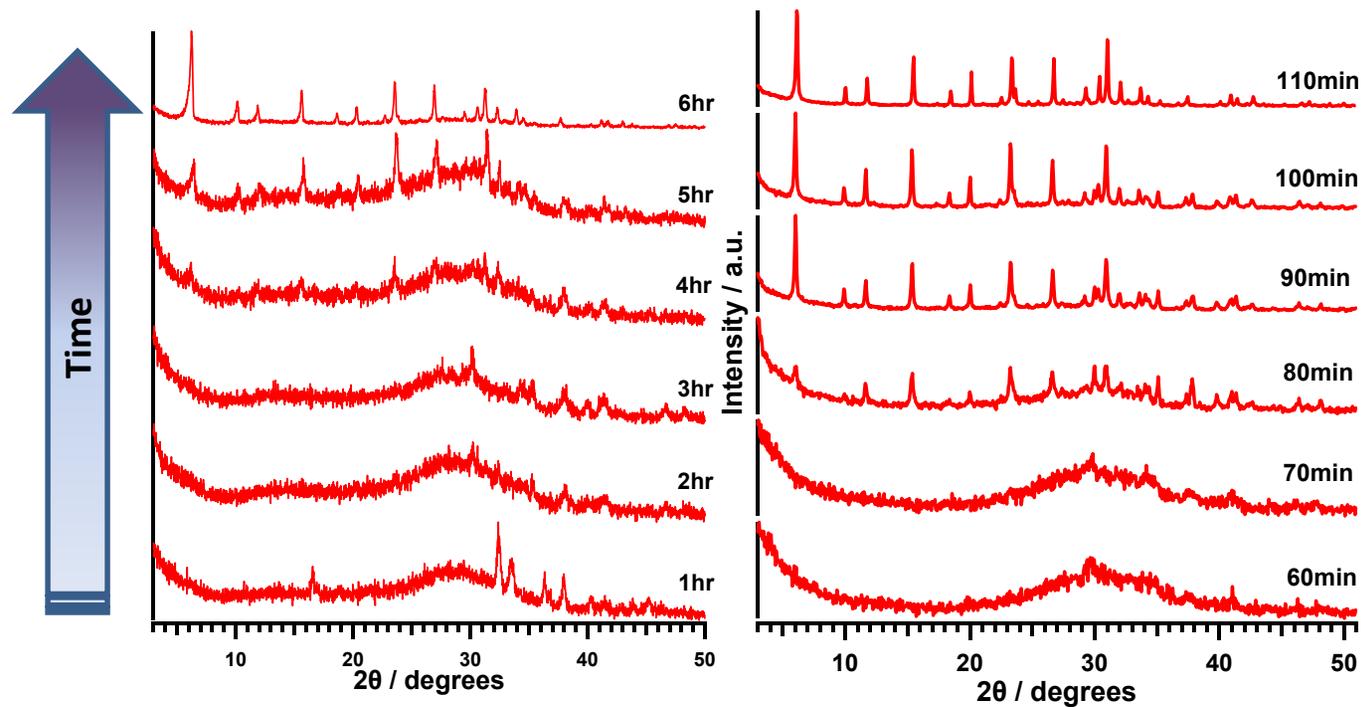
Rapid Synthesis

- Open system
- Control water concentration
- Remove H₂O (nucleation)
- Re-add H₂O (crystallization)



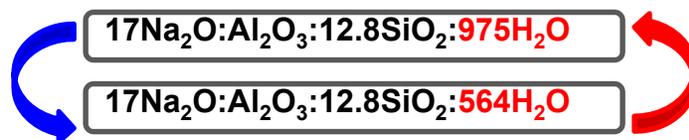
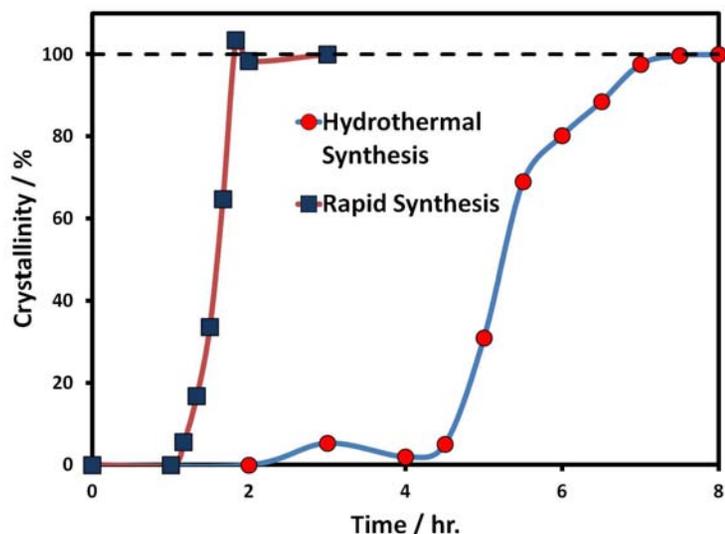
Approach 2: Rapid Synthesis of Zeolite Powders

Hydrothermal Synthesis $17\text{Na}_2\text{O}:1\text{Al}_2\text{O}_3:12.8\text{SiO}_2:975\text{H}_2\text{O}$ Rapid Synthesis

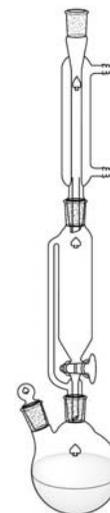


Rapid synthetic method yields pure crystalline zeolite powder in 110min

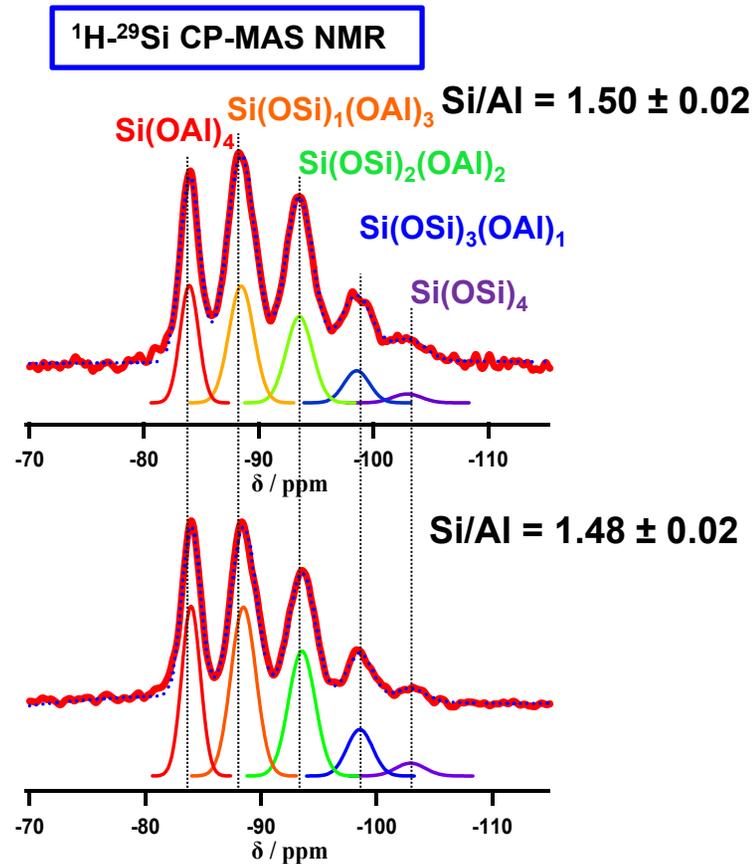
Approach 2: Zeolite Powder Formation Kinetics



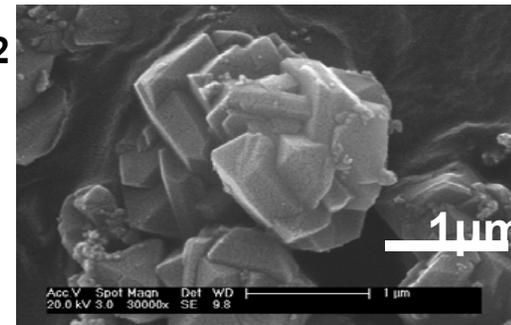
Crystallization in 2hrs. vs. 8hrs.



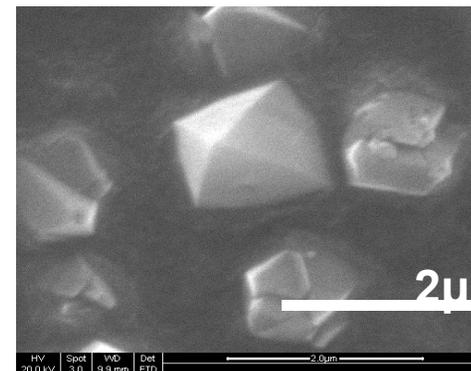
Approach 2: Comparing Rapid and Conventional Processes



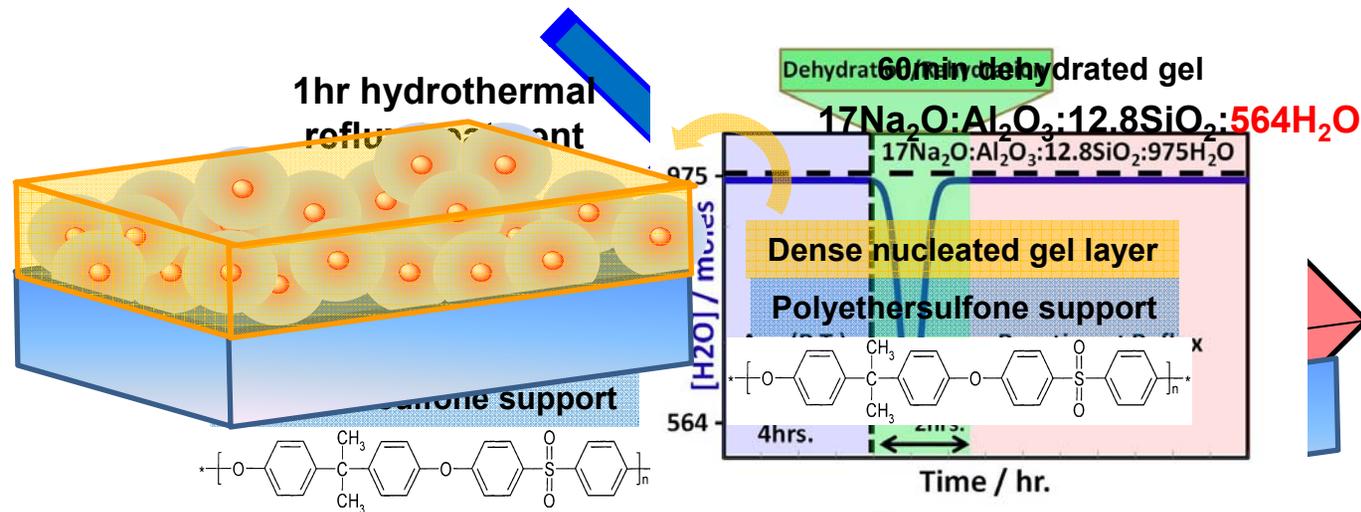
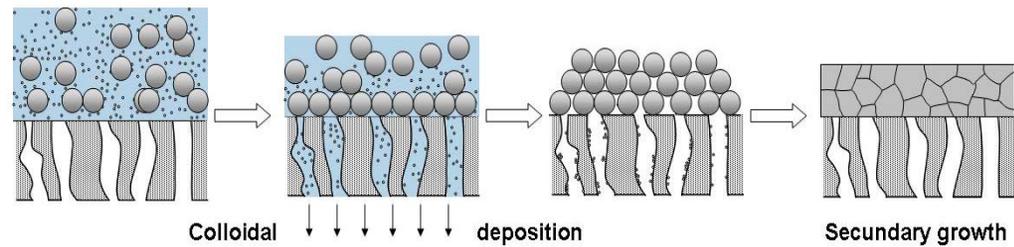
Rapidly grown product



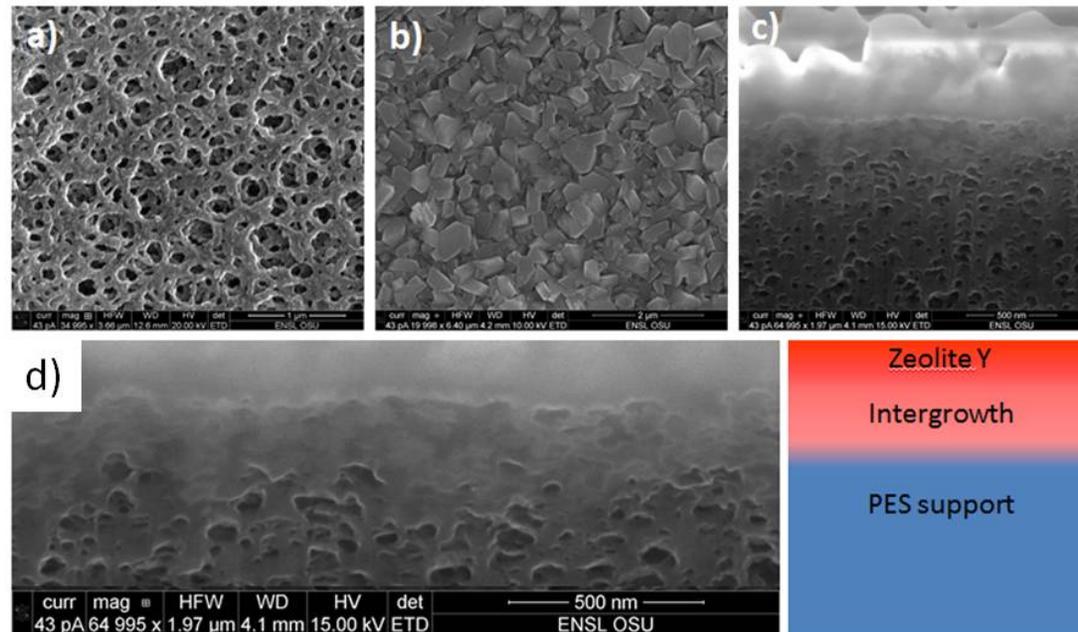
Hydrothermal product



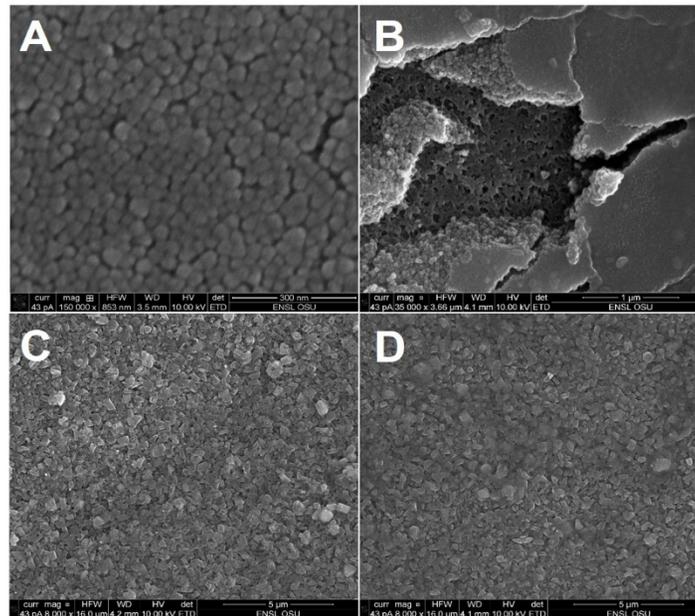
Approach 2: Application of Nucleated Gel to Membrane Growth



Approach 2: Characteristics of Membrane Grown by Rapid Process – 1 hour



Approach 2: Characteristics of Membrane Grown by Rapid Process – Tape Test

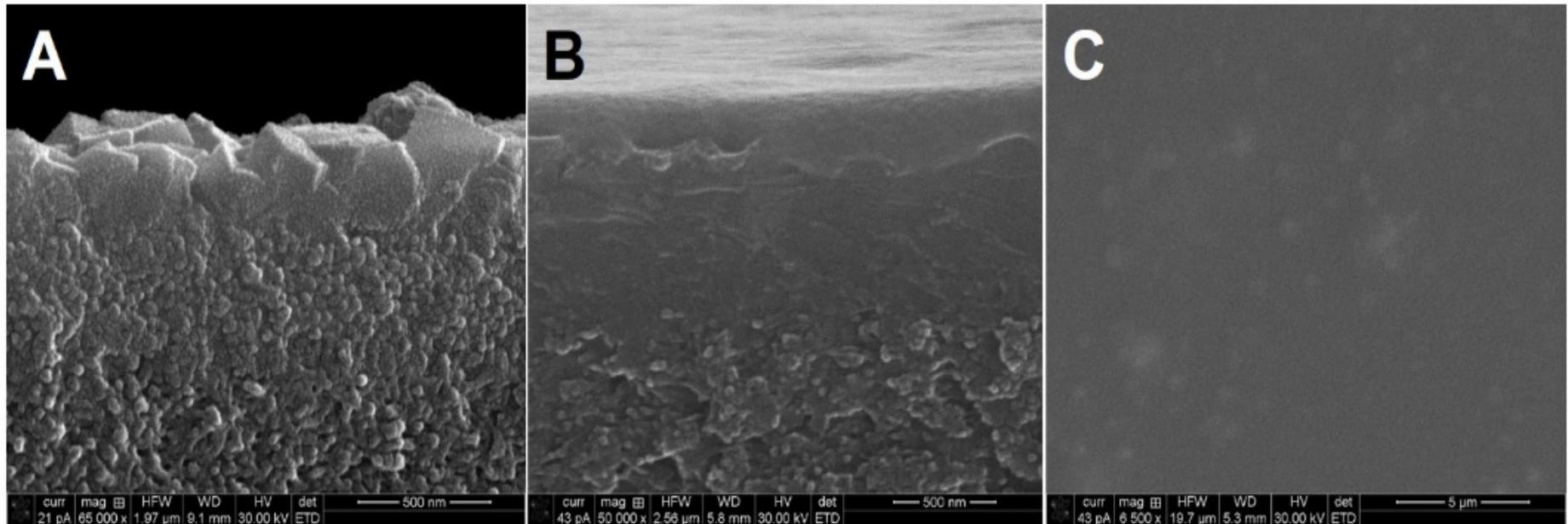


**Seeded zeolite
on PES
support**

**Zeolite
membrane on
PES support**

Before Tape test After Tape test

Approach 2: PDMS/Zeolite/PES Membrane Synthesized



A. Membrane

B. PDMS
caulking
layer

C: Top view

Techno-Economic Calculations (applicable to Approaches 1 and 2)

Performed by Gradient Technology (based on 2007\$)

- **Scale-up Prototype Membrane Results**
 - **870 GPU and 218 Selectivity at 57°C**
 - **\$40.4/tonne CO₂** – Nearly meet DOE target of \$40/tonne CO₂
 - **57.2% Increase in cost of electricity (COE)**
 - **1800 GPU and 160 Selectivity at 102°C**
 - **\$34.4/tonne CO₂** – Exceed DOE target of \$40/tonne CO₂
 - **47.1% Increase in cost of electricity (COE)**
 - **Lab-Size Membrane Results: 1100 GPU & 140 Selectivity at 57°C**
 - **\$37.5/tonne CO₂** – Exceed DOE target of \$40/tonne CO₂
 - **52.2% Increase in COE**
-
- **If:**
 - **CO₂ Permeance = 3000 GPU and 140 CO₂/N₂ Selectivity**
 - **\$30.9/tonne CO₂** – Exceed DOE target of \$40/tonne CO₂
 - **44% COE increase**
-
- **Significantly Lower Cost than Amine Scrubbing**

BP1 Milestone Status Report

Milestone Title Description	Planned Completion Date	Actual Completion Date	Verification Method	Comments
Task 1	09/30/12			
No-cost extension requested	09/30/12	08/23/12	Project Manager	Additional time for meeting BP1 target
PI change	12/31/12	12/05/12	Project Manager	Change PI/Co-PI
Task 2	01/31/13			
Polymer support source selected	09/30/12	09/30/12	Performance data	Identified PES and PSF supports
Cover layer composition	01/31/13	--	Performance data	BP1 target met, continue BP2 and BP3
Protocol ceramic support >3000 GPU	04/01/12	09/30/12	Performance data	Completed; protocol delivered in Q3
Meso-porous scaffold	09/30/12	09/30/12	Performance data	Support blocking solved (see the above milestone)
Task 3	03/31/13			
Membrane benchmark for polymer supports	03/15/13	03/15/13	Transport measurement	BP1 target met
Membrane benchmark for ceramic supports	09/30/12	09/30/12	Transport measurement	Ceramic supports are no longer pursued.
Task 4	03/01/13			
Model incorporated in ASPEN Plus program	03/01/13	03/01/13	Model calculations	Model in ASPEN Plus program completed
Task 5	05/01/13			
Quarterly reports	Q1 – Q6 + 30 days	Q1: 01/31/2012 Q2: 05/01/2012 Q3: 07/31/2012 Q4: 10/26/2012 Q5: 01/31/2013 Q6: 04/30/2013 Q7: 08/14/2013	Project Manager	Mostly on schedule
Task 6	07/01/13			
Budget Period 1 report	07/01/13	06/14/13	Project Manager	On schedule

BP2 Milestone Status Report

Milestone Title Description	Planned Completion Date	Actual Completion Date	Verification Method	Comments
Task 7	08/31/14			
3-Month no-cost extension requested	08/31/14	05/31/14	Project Manager	Additional time for completing BP2 tasks
Task 8	08/31/14			
Improved Membrane Synthesis	08/31/14	08/01/14	Performance data	
Continuous Deposition Procedure	08/31/14	08/01/14	Performance data	
Task 9	08/31/14			
Membrane Characterization	08/31/14	08/01/14	Transport measurement	
Task 10	08/31/14			
Continuous Membrane Machine Operational	12/31/13	12/31/13	Demonstration	The machine is operational at OSU.
7.5' 14" prototype membrane	08/31/14	08/01/14	Performance data	
Task 11	08/31/14			
Prototype Membrane Characterization	08/31/14	08/01/14	Transport measurement	
Task 12	08/31/14			
Spiral-Wound Membrane Element Rolling	08/31/14	08/01/14	Demonstration	
Task 13	08/31/14			
Pressure Drop Measurements of Membrane Elements	08/31/14		Pressure measurement	
Task 14	08/31/14			
Use and Refining of the System and Cost Analysis	08/31/14		Model calculations	
Task 15	08/31/14			
Quarterly reports	Q8 – Q11 + 30 days	Q8: 10/28/2013 Q9: 01/27/2014 Q10: 04/27/2014 Q10: 07/18/2014	Project Manager	On schedule
Task 16	10/01/14			
Budget Period 2 report	10/01/14		Project Manager	

Changes of BP2 Milestones

- **Zeolite Membrane Growth Time Revision to 1 h**
 - Over the past two years, we have shortened this time from 8 h to 1 h
 - Due to significant amount of time spent in fixing irreproducibility
 - Not enough time to decrease zeolite membrane growth time
 - Original growth time set for 16 minutes based on minimum web speed of 1'/min for continuous machine provided by manufacturer
 - We have found that minimum web speed is actually 0.15'/min
 - This translates to 107 min available in this machine
 - Investigate strategies for shortening growth to <60 min in BP3
- **Element/Module Size Revision to 1.8" dia. x 14"**
 - We have had to develop PES support needed for scale-up
 - Due to issues encountered from PES support manufacturer
 - This has taken available time and resources
 - Our PES support costs much less – membrane cost much reduced
 - Due to zeolite growth time issue, limited resources to make zeolite nanoparticles – limited supply of nanoparticles
 - New microwave approach will alleviate this obstacle
 - Modified Milestone: Fabricate 7.5', in lieu of 50' in BP2
 - 15" x 14" membrane leaf used for element rolling
 - Will fabricate >50' membrane in BP3
 - Make three 2" dia. x 14" elements/modules in BP3

Partners Involved in BP3 Tasks

- **Optimized Continuous Membrane Fabrication**
 - Continuous Membrane Fabrication Machine Available at OSU
 - **TriSep Membrane Company is Involved**
- **Real Flue Gas Testing**
 - Quick Testing at NCCC
 - + Membrane module testing with real flue gas
 - + For comparison with simulated flue gas results
 - **NCCC (Tony Wu, Frank Morton) for Testing**
 - **AEP is Involved in EH&S and Power Plant integration**
 - **Gradient is Involved in Modeling and Cost Calculations**
- **Gradient Task**
 - **Task 24.0 – Use and Refining of System and Cost Analysis**

BP3 Budget

Budget Period	Federal Share	Non-Federal Share	Project Total
3	\$1,143,566	\$420,293	\$1,563,859

- **Budget amounts do not change**
 - Not asking for more money for BP3
- **Reallocating funds for proposed technology development acceleration**
- **ODOD cash match of \$500,000 cost share**
 - Add 2.5 Ph.D. students to this project
 - + Zeolites with different pore openings as compared to Zeolite-Y
 - + Subsequent zeolite-polymer composite membranes
 - This should complement and contribute to this project significantly

BP3 Tasks – No Major Changes

Task 17.0 – Project Management and Planning

Task 18.0 – Further Improved Membrane Synthesis

- Subtask 18.1 – Optimiz. of inorg. sel. layer: rapid zeolite membrane growth
- Subtask 18.2 – Optimization of inorganic / polymer membrane

Task 19.0 – Membrane Characterization – *SO₂ Mitigation*

Task 20.0 – Optimized Prototype Membranes Fabrication

- Subtask 20.1 – Fabrication of optimized inorganic selective layer
- Subtask 20.2 – Fabrication of optimized inorganic/polymer membrane
 - + At least 50 ft of 14-inch width

Task 21.0 – Optimal Prototype Membrane Characterization

Task 22.0 – Prototype Module Fabrication – 3 Modules (each 2 m²)

- Subtask 22.1: Spiral-wound module fabrication
- Subtask 22.2: Optimized fabrication w.r.t. pressure drop & packing density

Task 23.0 – Membrane Module Testing

- Subtask 23.1: Bench scale testing using simulated flue gas
- Subtask 23.2: Pressure drop measurements with simulated flue gas

Task 24.0 – Use and Refining of System and Cost Analysis

Task 25.0 – Quarterly Progress Reports

Task 26.0 – Final Technical Report

BP3 Task Schedule

Task Name	Total Cost of Task (\$)	Start	Finish	1st Quarter			2nd Quarter			3rd Quarter			4th Quarter								
				Sep	Oct	Nov	Dec	Jan	Feb	Mar	Apr	May	Jun	Jul	Aug	Sep	Oct	Nov			
Budget Period 3	1,563,859	9/1/2014	8/31/2015																		
Task 17: Project Management and Planning	10,753	9/1/2014	8/31/2015																		
Task 18: Further Improved Membrane Synthesis	262,545	9/1/2014	8/31/2015																		
Subtask 18.1: Optimization of inorganic selective layer	131,272	9/1/2014	8/31/2015																		
Subtask 18.2: Optimization of inorganic/polymer membrane	131,273	10/1/2014	8/31/2015																		
Task 19: Membrane Characterization	149,745	10/1/2014	8/31/2015																		
Task 20: Optimized Prototype Membrane Fabrication	300,545	10/1/2014	8/31/2015																		
Subtask 20.1: Fabrication of optimized inorganic selective layer	150,272	10/1/2014	8/31/2015																		
Subtask 20.2: Fabrication of optimized inorganic/polymer membrane	150,273	11/1/2014	8/31/2015																		
<i>Milestone 6: Successful fabrication of at least 50-ft prototype membrane</i>																					
Task 21: Optimal Prototype Membrane Characterization	211,065	11/1/2014	8/31/2015																		
<i>Milestone 7: Prototype membrane with \$40/tonne CO₂</i>																					
<i>Milestone 8: Prototype membrane with \$39/tonne CO₂</i>																					
Task 22: Prototype Module Fabrication	221,894	11/1/2014	5/31/2015																		
Subtask 22.1: Spiral-wound module fabrication	110,947	11/1/2014	5/31/2015																		
Subtask 22.2: Optimized fabrication w.r.t. pressure drop/packing density	110,947	12/1/2014	5/31/2015																		
<i>Milestone 9: Successful fabrication of 3 modules of 2-inch dia. by 14 inches</i>																					
Task 23: Membrane Module Testing	221,086	9/1/2014	8/31/2015																		
Subtask 23.1: Bech scale testing with simulated flue gas	138,179	9/1/2014	2/28/2014																		
Subtask 23.2: Presure drop measurements with simulated flue gas	82,907	3/1/2015	8/31/2015																		
<i>Milestone 10: Achieving acceptable pressure drop (0.05 bar)</i>																					
Task 24: Use and Refining of the System and Cost Analysis	174,475	9/1/2014	8/31/2015																		
Task 25: Quarterly Progress Reports	5,876	11/1/2014	10/1/2015																		
Task 26: Final Technical Report	5,875	6/1/2015	12/1/2015																		

Field Testing for BP3

- **Prototype Membrane Module Field Testing**
 - NCCC with real flue gas
 - Discussion with NCCC (Tony Wu and Frank Morton) in progress
- **CO₂ and SO₂ will be Measured**
- **After Test, comparative analysis with baseline data**
 - Module will be measured in lab for permeance / selectivity
 - For comparison with the module without exposure
- **Post Mortem Analysis**
- **Any Residues/Contaminants Collected will be Analyzed**
 - Testing on treating flue gas with pretreatment
 - + Filtration using 5- μ m filter before membrane
 - + Based on field test data, revisit pretreatment requirements
 - Designing Effective Pretreatment
 - Determining O&M Requirements

Technology Development Roadmap

July 2013	8" x 8" Membrane leaf hand-rolled into element
May 2014	Continuous PES support fabrication
June 2014	Continuous zeolite nanoparticle deposition
July 2014	Continuous amine cover layer on zeolite nanoparticle layer Spiral-wound element fabricated using rolling machine "Model-T" membrane module fabricated
December 2014	Membrane module testing at NCCC
2016 – 2017	1 TPD at NCCC
2019 – 2020	20 TPD (1 MW) at NCCC

Technical and Economic Analysis

- Aspen Plus modeling software utilized to model the capture process
- PC plant has not been modeled
- Model has been integrated into a Supercritical PC plant (Case 11) that produces 550 MW of net power
- The capture equipment has been sized to capture 90% of the CO₂ at a purity of 95 vol% and compress it to 2,215 psia at 124°F
- Economic model is based on the factored cost estimating technique (distributive percentages)
 - Recognized by American Association of Cost Engineers (AACE)
- **Technical & economic analysis conducted similarly to the analysis completed in the NETL report**
 - *Cost and Performance Baseline for Fossil Energy Plants Volume 1: Bituminous Coal and Natural Gas to Electricity, Revision 2, Nov. 2010*

Technical & Economic Analysis (Cont'd)

- 27 different cases utilizing the Aspen Plus and economic model were conducted
- Report details the results and includes stream tables, process flow diagrams, equipment size and cost information, capital cost summaries, variable and fixed cost summaries, and all performance and cost metrics
- Primary parameters investigated:
 - ✓ Flue gas feed pressure to the membrane-based CO₂ capture unit
 - ✓ First-stage membrane permeate pressure
 - ✓ CO₂ permeance
 - ✓ CO₂ / N₂ selectivity
 - ✓ H₂O / CO₂ selectivity

Technical & Economic Analysis Case Parameters

	Case A	Case B	Case C	Case D	Case E	Case F	Case G	Case H	Case I	Case J
Flue Gas Pressure (atm)	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5
Membrane 1 Permeate Pressure (torr)	150	150	150	150	150	150	150	150	150	150
Membrane 2 Air Sweep (%)	80	80	80	80	80	80	80	80	80	80
Air Sweep Gas Pressure (atm)	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05
CO ₂ Permeance (GPU)	1100	1300	1500	1700	2000	3000	1100	1100	1100	1100
H ₂ O:CO ₂ Permeance	1	1	1	1	1	1	1	1	1	1
CO ₂ :N ₂ Permeance	140	140	140	140	140	140	160	180	200	300
H ₂ O Permeance (GPU)	1100	1300	1500	1700	2000	3000	1100	1100	1100	1100
N ₂ Permeance (GPU)	7.9	9.3	10.7	12.1	14.3	21.4	6.9	6.11	5.5	3.7
SO _x and NO _x Permeance (GPU)	0	0	0	0	0	0	0	0	0	0
MB-02 Sweep Side Pressure	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05
Turboexpander	Yes									
Cost of Electricity Increase (%)	54.36%	52.77%	51.60%	50.71%	49.70%	47.80%	53.85%	53.57%	53.41%	53.47%
Cost of CO ₂ Capture	38.60	37.47	36.64	36.00	35.29	33.94	38.29	38.12	38.05	38.17

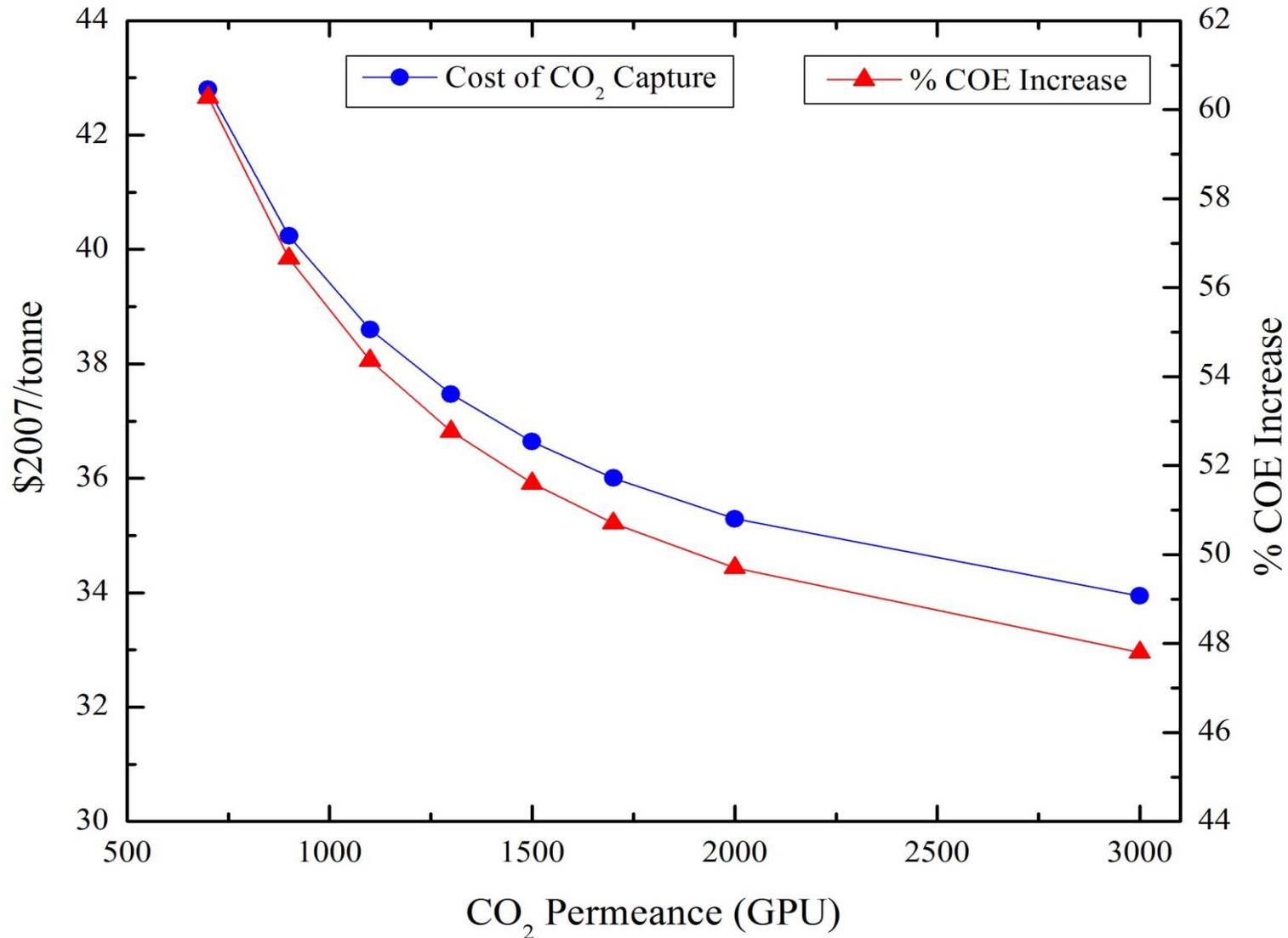
	Case K	Case L	Case M	Case N	Case O	Case P	Case Q	Case R	Case S	Case T
Flue Gas Pressure (atm)	1.6	1.4	1.3	1.2	1.1	1.07	1.054	1.5	1.5	1.5
Membrane 1 Permeate Pressure (torr)	150	150	150	150	150	150	150	170	150	150
Membrane 2 Air Sweep (%)	80	80	80	80	80	80	80	80	80	80
Air Sweep Gas Pressure (atm)	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05
CO ₂ Permeance (GPU)	1100	1100	1100	1100	1100	1100	1100	1100	900	700
H ₂ O:CO ₂ Permeance	1	1	1	1	1	1	1	1	1	1
CO ₂ :N ₂ Permeance	140	140	140	140	140	140	140	140	140	140
H ₂ O Permeance (GPU)	1100	1100	1100	1100	1100	1100	1100	1100	900	700
N ₂ Permeance (GPU)	7.9	7.9	7.9	7.9	7.9	7.9	7.9	7.9	6.4	5.0
SO _x and NO _x Permeance (GPU)	0	0	0	0	0	0	0	0	0	0
MB-02 Sweep Side Pressure	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05	1.05
Turboexpander	Yes	Yes	Yes	Yes	Yes	No	No	Yes	Yes	Yes
Cost of Electricity Increase (%)	54.26%	55.01%	56.46%	60.38%	74.43%	90.72%	157.98%	53.52%	56.66%	60.28%
Cost of CO ₂ Capture	38.31	39.30	40.61	43.75	54.29	66.04	115.13	38.13	40.24	42.80

Technical & Economic Analysis Case Parameters

	Case U	Case V	Case W	Case X	Case Y	Case Z	Case AA
Flue Gas Pressure (atm)	1.7	1.5	1.5	1.5	1.5	1.5	1.6
Membrane 1 Permeate Pressure (torr)	150	130	190	210	230	170	150
Membrane 2 Air Sweep (%)	80	80	80	80	80	80	80
Air Sweep Gas Pressure (atm)	1.05	1.05	1.05	1.05	1.05	1.05	1.05
CO ₂ Permeance (GPU)	1100	1100	1100	1100	1100	900	900
H ₂ O:CO ₂ Permeance	1	1	1	1	1	1	1
CO ₂ :N ₂ Permeance	140	140	140	140	140	140	140
H ₂ O Permeance (GPU)	1100	1100	1100	1100	1100	900	900
N ₂ Permeance (GPU)	7.9	7.9	7.9	7.9	7.9	6.4	6.4
SO _x and NO _x Permeance (GPU)	0	0	0	0	0	0	0
MB-02 Sweep Side Pressure	1.05	1.05	1.05	1.05	1.05	1.05	1.05
Turboexpander	Yes						
Cost of Electricity Increase (%)	54.46%	56%	53%	53%	52%	56%	56%
Cost of CO ₂ Capture	38.2	39.5	37.8	37.6	37.5	39.8	39.7

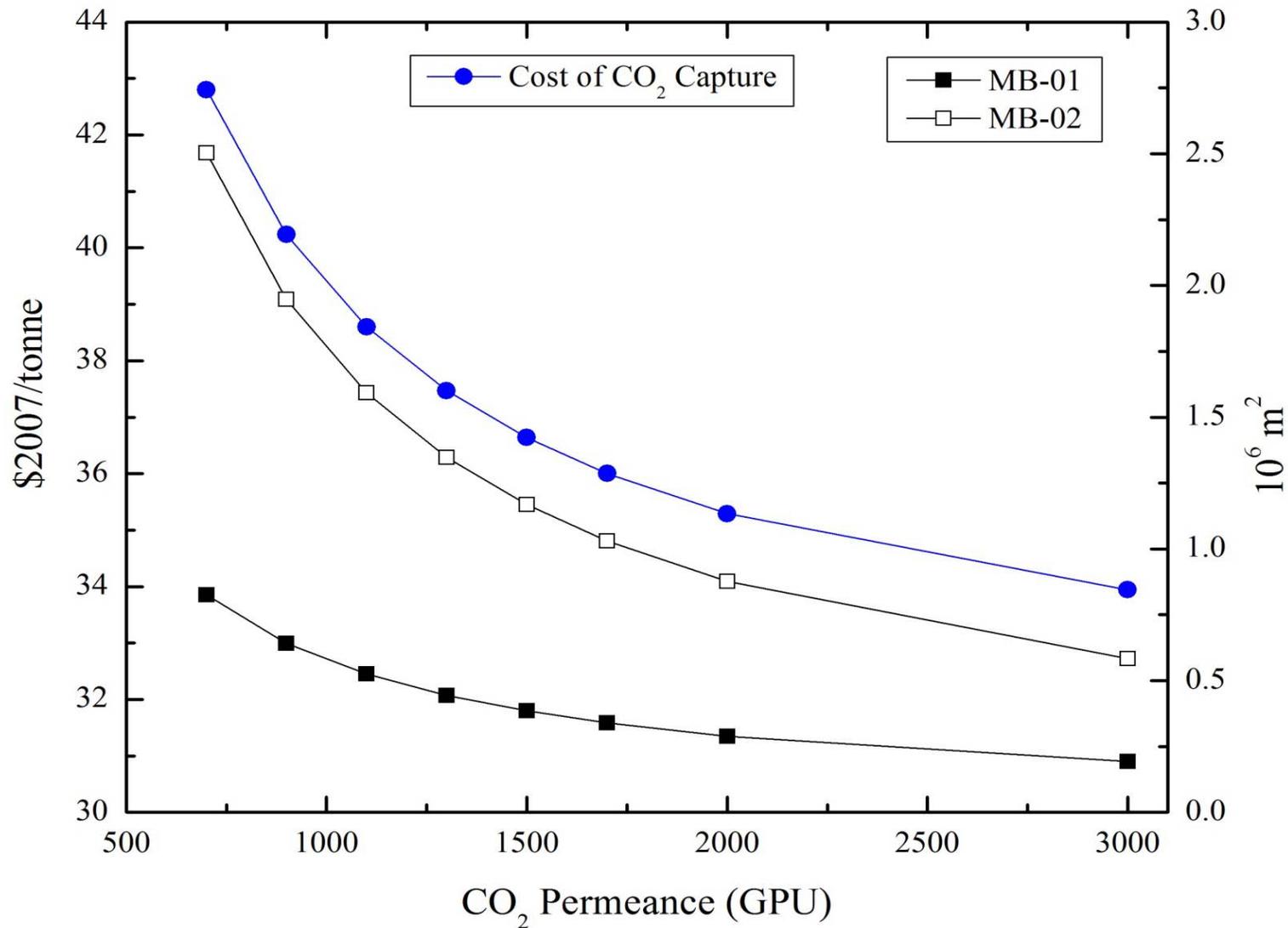
Technical & Economic Analysis Results

Effect of Permeance



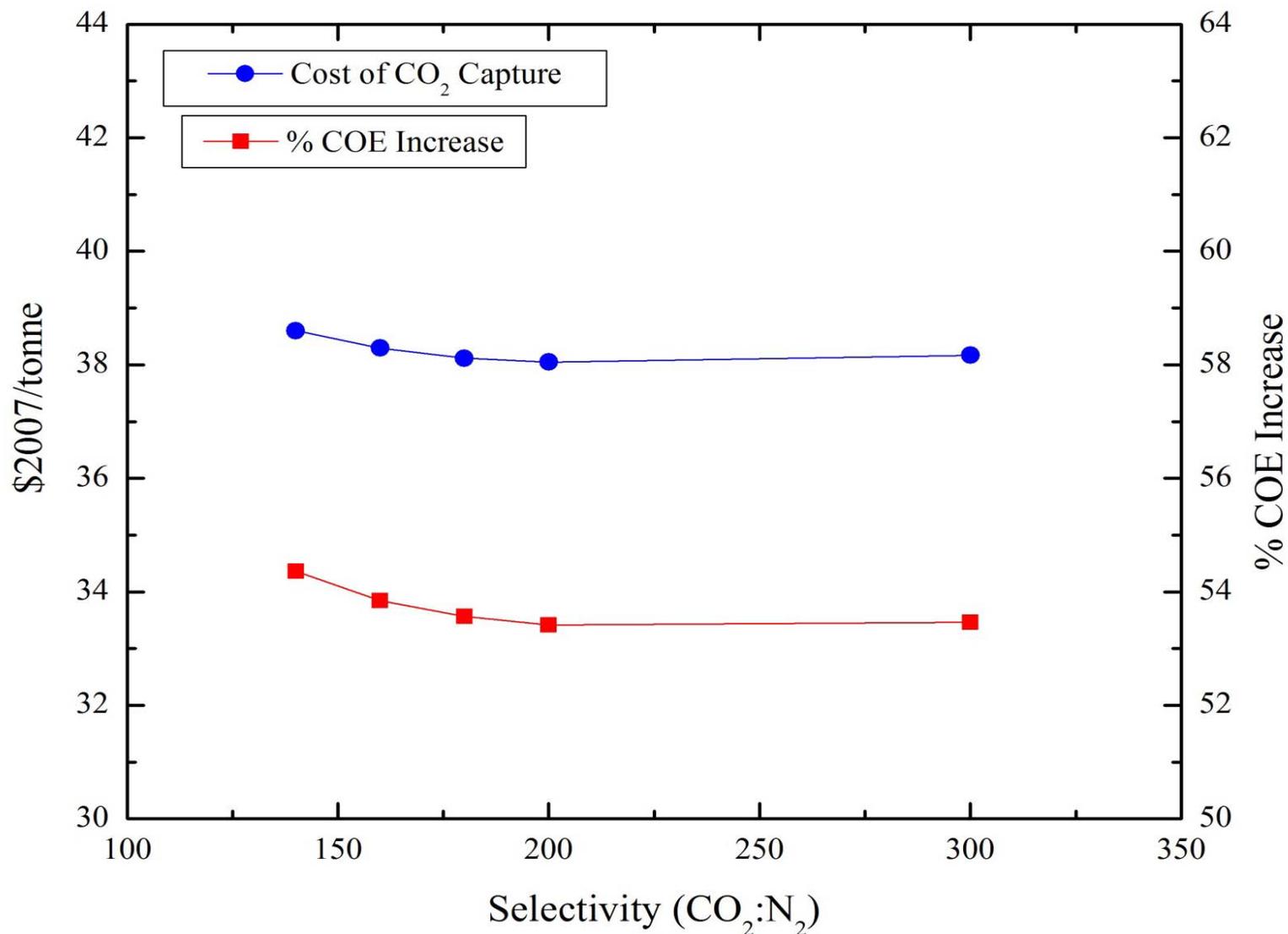
Technical & Economic Analysis Results

Effect of Permeance



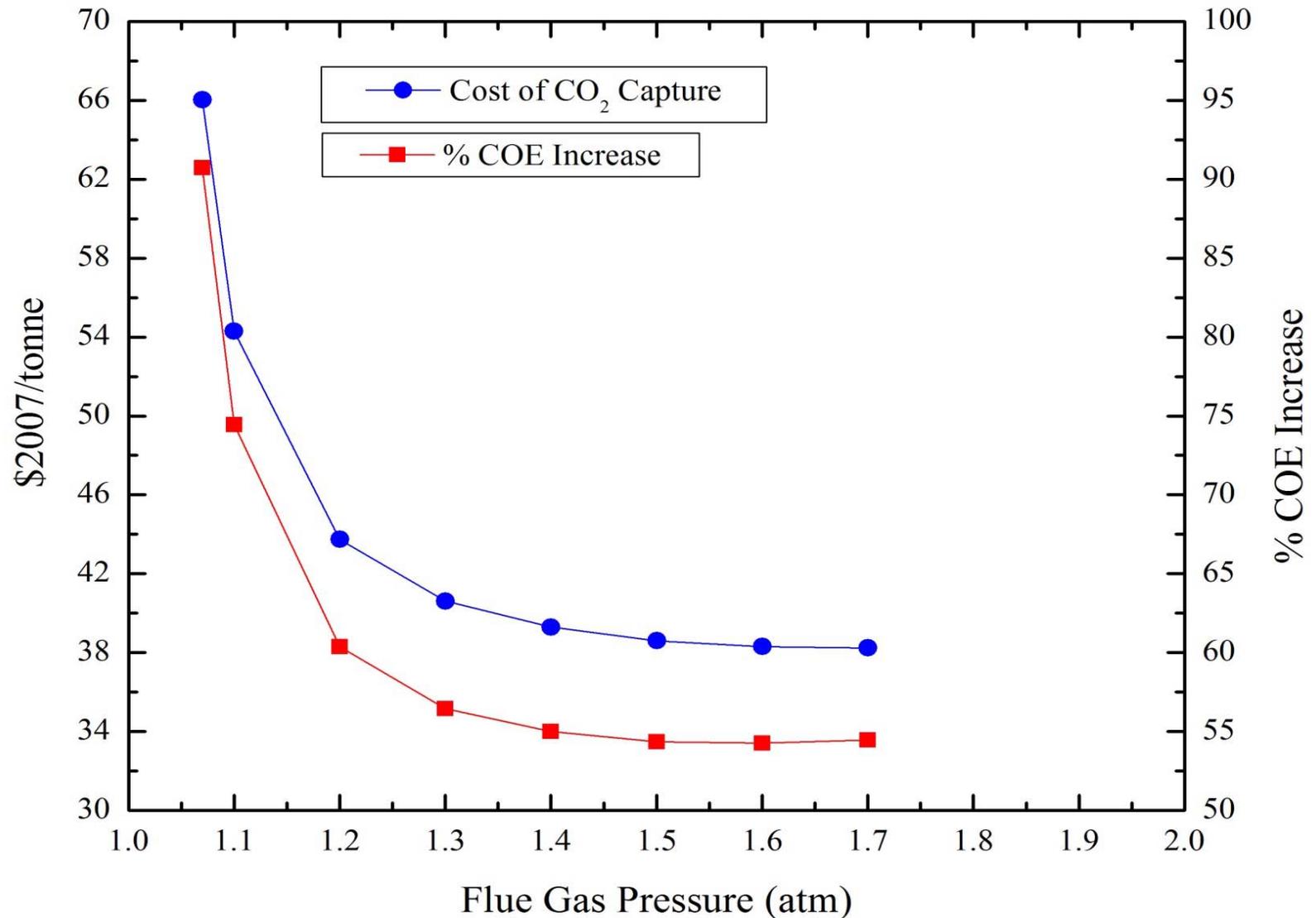
Technical & Economic Analysis Results

Effect of Selectivity



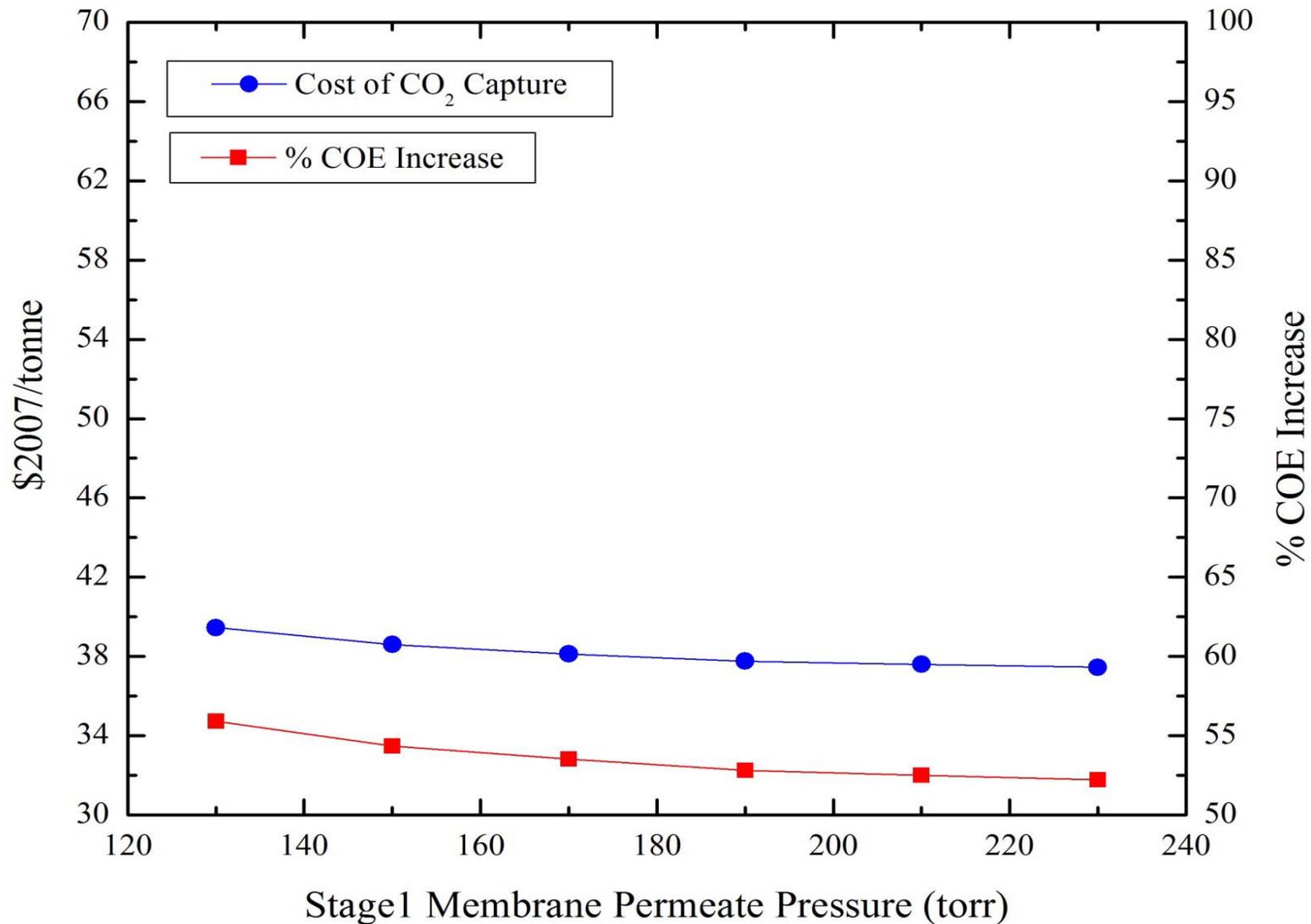
Technical & Economic Analysis Results

Effect of Flue Gas Pressure



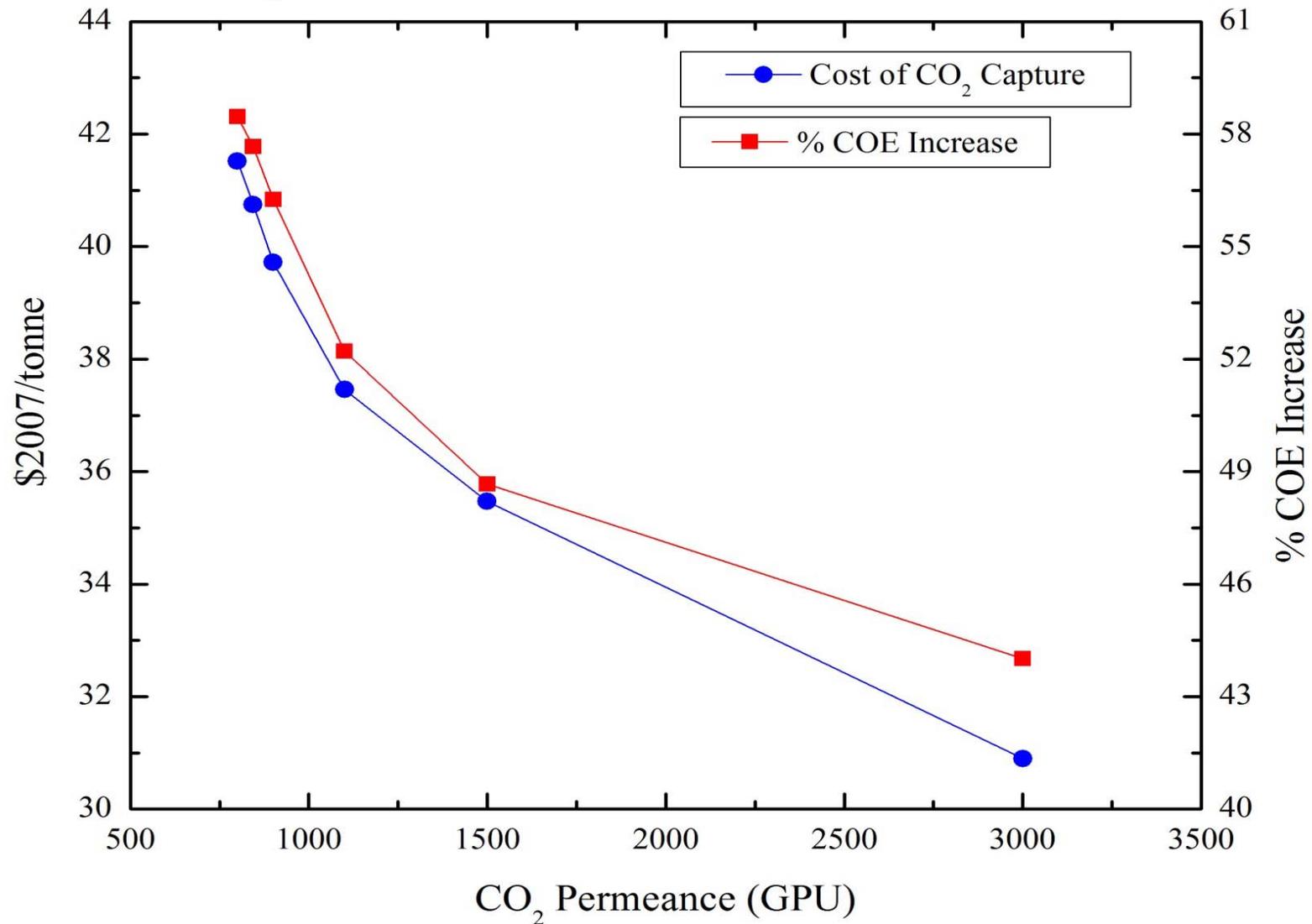
Technical & Economic Analysis Results

Effect of Permeate Pressure



Technical & Economic Analysis Results

Optimum as a Function of Permeance



Technical & Economic Analysis Results

- **Effect of CO₂ permeance**
 - Improved permeance reduces membrane areas and, hence, cost
- **Effect of CO₂ / N₂ selectivity**
 - Selectivity (>140) has negligible impact on economics
- **Effect of Flue Gas Pressure**
 - Flue gas pressure optimum approximately 1.5 – 1.6 atm and decreases with increasing permeance
- **Effect of Permeate Pressure (Stage 1)**
 - Increased permeate pressure slightly improves economics due to reduced vacuum demand

Technical & Economic Analysis Summary

- Technical & economic analysis has shown that the cost of carbon capture can be reduced to **less than \$40/tonne**
- SO₂ removal equipment has been included to reduce SO₂ to less than 10 ppm
- Improved economic analysis will require additional equipment design and quotations

Summary

- **Achieved milestones/success criteria for BP2**
 - **Scale-up prototype membrane results**
 - + **CO₂ permeance = 870 GPU, CO₂/N₂ selectivity = 218 at 57°C**
 - ++ **Capture cost = \$40.4/tonne CO₂ – Nearly meet DOE target of \$40/tonne**
 - + **CO₂ permeance = 1800 GPU, CO₂/N₂ selectivity = 160 at 102°C**
 - ++ **Capture cost = \$34.4/tonne CO₂ – Exceed DOE target of \$40/tonne**
 - **Lab-size membrane with CO₂ permeance = 1100 GPU, CO₂/N₂ selectivity = ~140 at 57°C**
 - + **Capture cost of \$37.5/tonne CO₂ – Exceed DOE target of \$40/tonne**
- **Updated techno-economic analysis**
- **Proposed BP3 tasks**
 - **Accelerating technology development**
- **Received \$500K Matching funds from Ohio Coal Office**
- **Asking NETL to authorize proposed BP3 research, budget, and schedule**