

# An Advanced Catalytic Solvent for Lower Cost Post-combustion CO<sub>2</sub> Capture in a Coal-fired Power Plant

*Award # DE-FE0012926*

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**Overall Objective:** Address technical hurdles to developing an integrated process focused on a catalyzed solvent utilizing homogeneous catalyst.

## Project Details

- **Benefit from multiple CAER technologies: solvent; catalyst, membrane, process**
- **Project cost:**
  - **DOE share:\$2.97M**
  - **Cost share:\$742K**
- **Period performance: 10/1/2013 – 9/30/2016**

## Project Objectives

- Build towards low-cost CO<sub>2</sub> capture system via Integration of multiple CAER technologies
- To verify an advanced catalytic solvent with integrated membrane dewatering for solvent enrichment in our 0.1MW pilot plant (Proof of concept)

	<h1>CMRG</h1>		 <b>WorleyParsons</b> resources & energy
<ul style="list-style-type: none"> <li>• Project management</li> <li>• Catalytic solvent testing</li> <li>• ASPEN modeling</li> </ul>	<ul style="list-style-type: none"> <li>• Technical support</li> </ul>	<ul style="list-style-type: none"> <li>• PPE recommendation</li> <li>• EH&amp;S analysis</li> </ul>	<ul style="list-style-type: none"> <li>• Front-end engineering support</li> <li>• TEA</li> </ul>

	Previous work	Current Project				Future Development	
Yr	2011-2013	2013	2014	2015	2016	2017-2020	>2020
BP	-	1	1/2	2/3	3	-	-

Fundamental Development; Proof-of-Concept

Slipstream ~2 MWth

~20 MWe

Verification Testing on 0.1 MWth Unit

Laboratory Validation and Scale-up

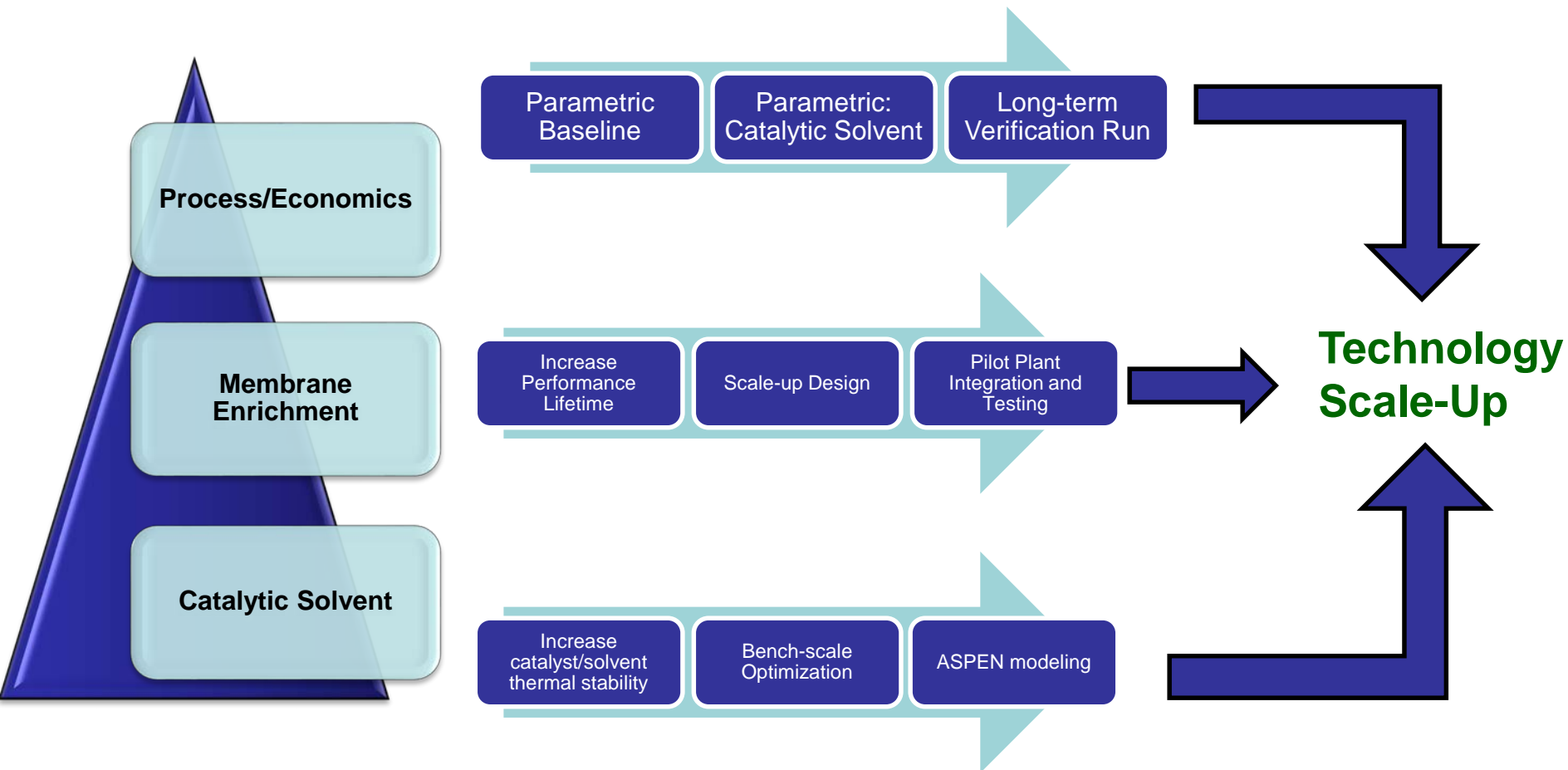
Parametric Testing on 0.1 MWth Unit

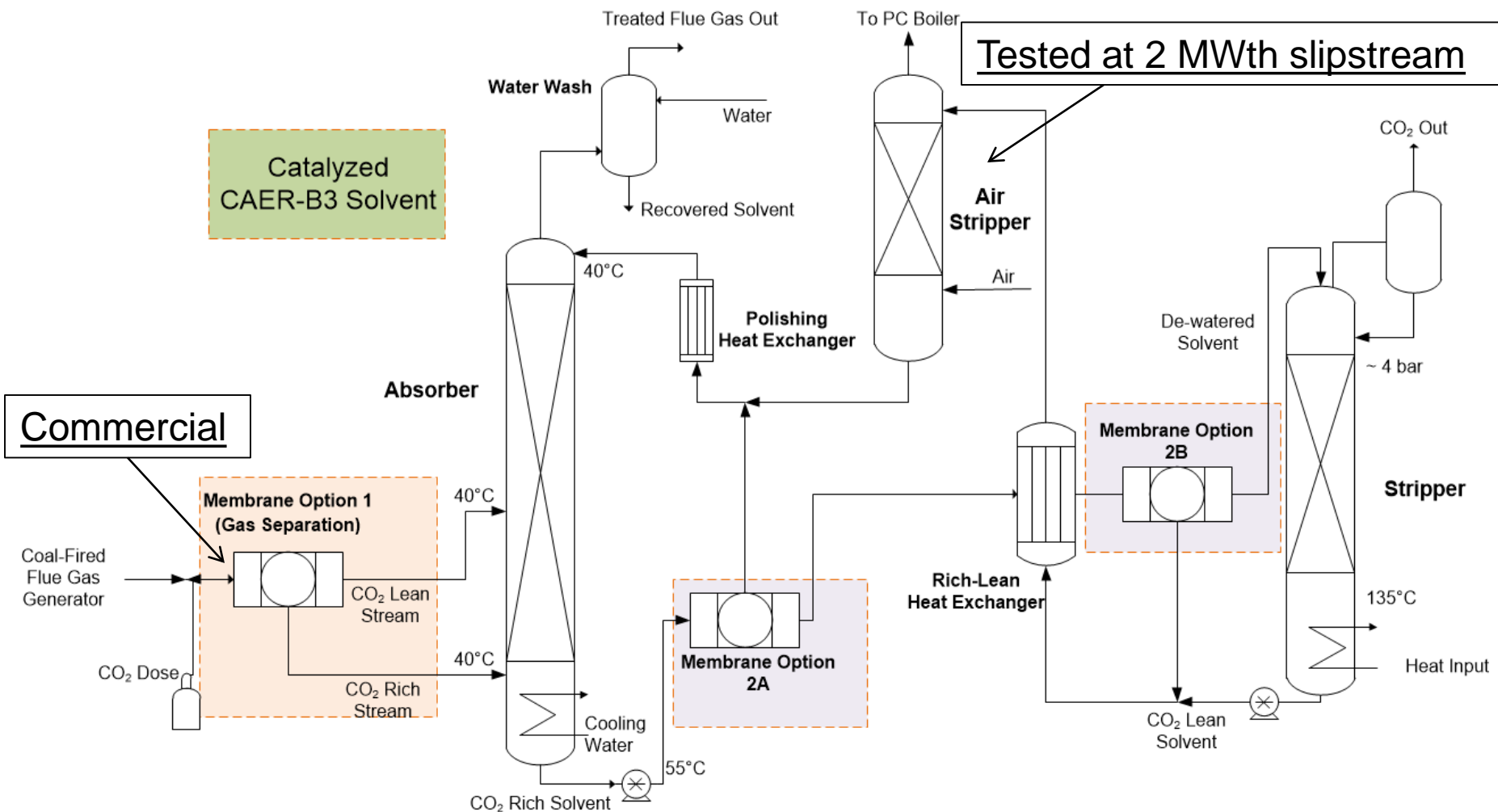
- **Solvent Optimization**
- **Milestone:** VLE and model regression
- **Membrane Enrichment**
- **Milestone:** 5% enrichment over 5hr
- **Catalyst Scale-up**
- **Milestone:** Develop method to produce 50g/batch
- **Milestone:** PPE recommendation & front-end engineering analysis

- **Catalyst Production**
- **Milestone:** 500g produced
- **Parametric Testing**
- **Milestone:** 100hr runs with and without catalyst completed
- **Membrane Enrichment**
- **Milestone:** 10% enrichment over 100hr and module design

- **Verification Run**
- **Milestone:** 500hr verification run
- **Membrane Enrichment**
- **Milestone:** Unit integrated and 20% dewatering observed
- **Techno-Economic Analysis**
- **Milestone:** TEA showing at least 20% reduction in energy cost compared to Case 12
- **EH&S**
- **Milestone:** Favorable EHS assessment

## 3-Prong Approach





Pre-absorber CO<sub>2</sub> enrichment and de-watered CAER-B3 used to lower the energy cost of CO<sub>2</sub> capture.



## Advantages

- Potential for reduced capital cost for post-combustion CO<sub>2</sub> capture
  - Increased scrubber kinetics (smaller absorber)
- Potential for reduced energy consumption compared to reference case (MEA)
  - High  $\alpha$ ; cyclic capacity
  - High stripper temperatures/pressure
  - Less solvent make-up rate

## Challenges

- Transition from lab- to bench-scale process under real flue gas conditions
- Solvent oxidation via catalyst addition
- Integration of multiple technologies

## ❖ $k_{obs}$ impacts PFO calculation

$$k_{obs} = k[amine] + k'[cat]$$

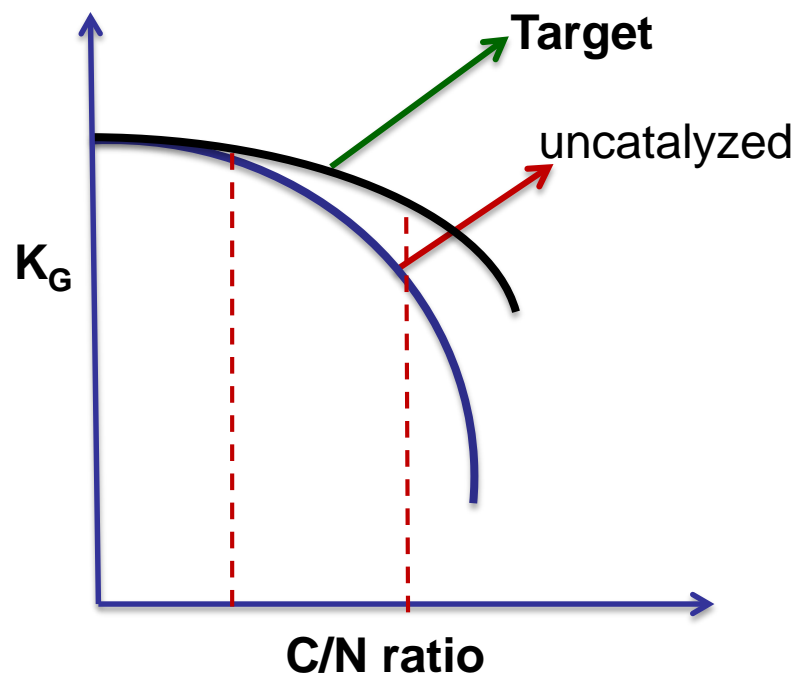
$$k_{g,PFO}' = \frac{\sqrt{D_{CO_2} \cdot k_{obs}}}{H_{CO_2}}$$

$$k_{g,PFO}' \propto \sqrt{k_{obs}}$$

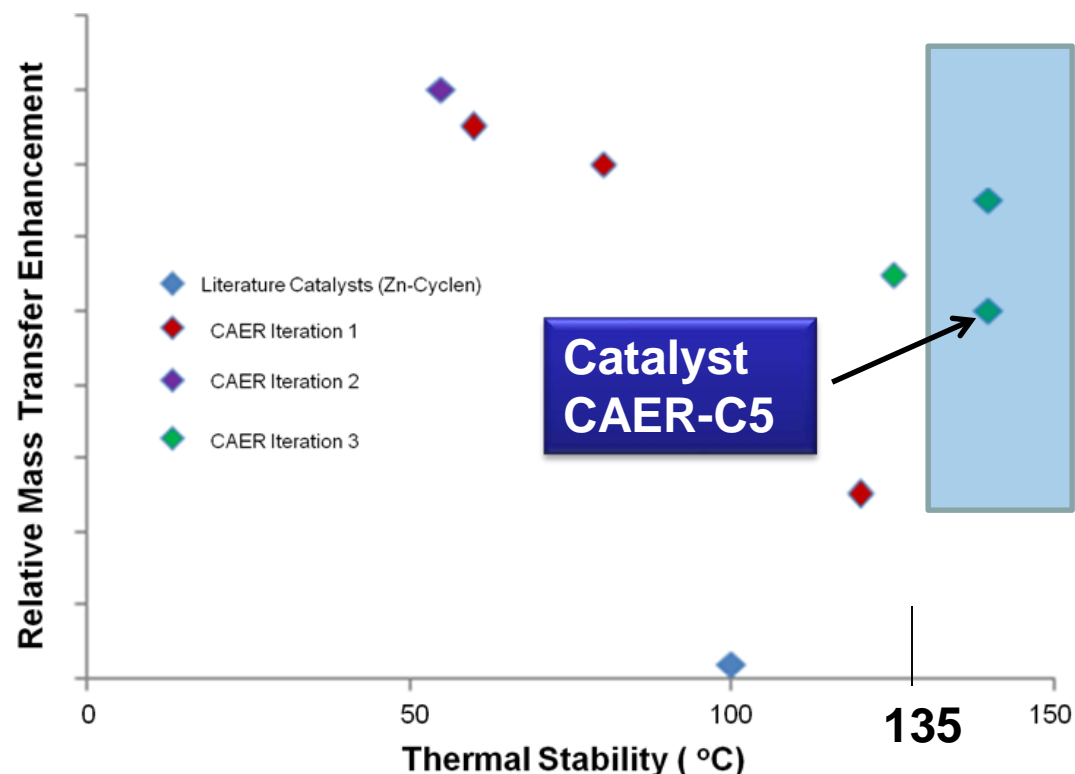
Higher the value of  $k_{obs}$  higher the mass transfer rate

Pseudo first order approximation

$$k_{g,PFO}' = \frac{\sqrt{D_{CO_2} \cdot k_2 \cdot [amine]}}{H_{CO_2}}$$



**Achieve rate enhancement at higher carbon loadings**



## Improved Catalyst

- 10% rate enhancement in CAER-B3
- Retains activity after heating at 145 °C for 150h
- Simple catalyst preparation; suitable for scale-up



Task Number	Title	Description
7	Updated Project Management Plan for budget period 2.	Review and update PMP/SOPO
8	CAER catalyst Production	Production of at least 500 g of CAER catalyst
9	Parametric CAER-B3 investigation	100 hour parametric study on conventional PCC using CAER solvent without catalyst at bench-scale completed
10	Parametric catalytic CAER-B3 investigation	100 hour parametric study with catalyst and gas pre-concentration membrane at bench-scale completed
11	Membrane dewatering	Membrane shown to dewater CAER-B3 solvent by at least 10% over 100 hours in lab-scale single element testing
11	Zeolite membrane bench-scale test module design	Completed design for dewatering membrane test module for integration in 0.1 MWth bench-scale test unit

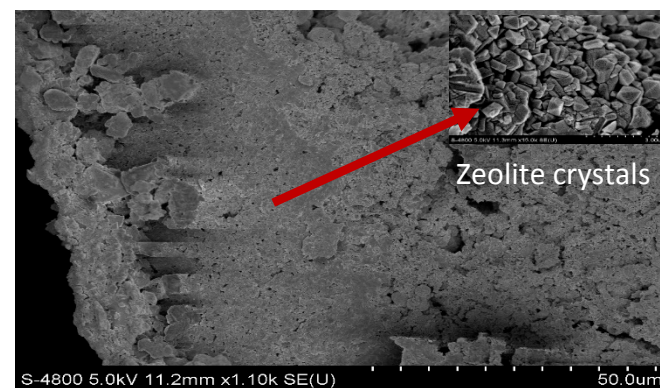
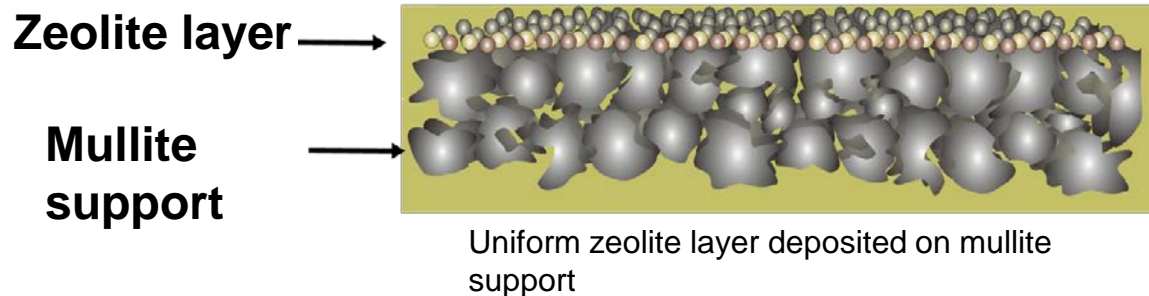
- BP2 has focused on testing in our 0.1 MWth unit
  - baseline testing, parametric catalytic solvent testing
  - short term degradation analysis
- Membrane improvement and module design for pilot integration

Task #	Milestone	Description	Deliverable Date	Completion Date
7	Updated PMP for BP 2	Review and update PMP/SOPO	10/30/14	12/18/14
8	CAER catalyst production	Production of at least 500 g of CAER catalyst	12/31/14	12/17/14
9	Parametric CAER-B3 investigation	100 hr parametric study on conventional PCC using CAER solvent without catalyst at bench-scale completed	2/28/15	1/29/15
10	Parametric catalytic CAER-B3 investigation	100 hr parametric study with catalyst and pre-concentration membrane at bench-scale completed	8/15/15	
11	Membrane dewatering	Membrane shown to dewater CAER-B3 by at least 10% over 100 hr in lab scale testing	5/31/15	2/24/15
11	Zeolite membrane bench-scale test module design	Completed designs for dewatering membrane test module for integration into 0.1 MWth bench-scale unit	6/30/15	6/8/15

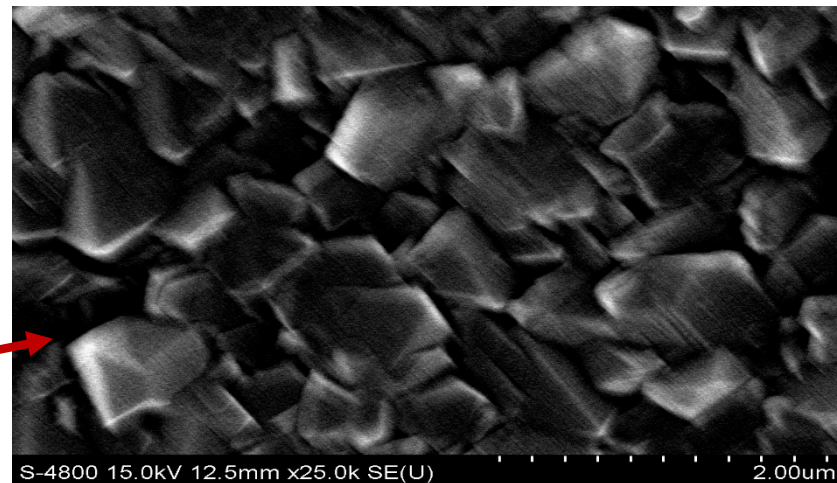
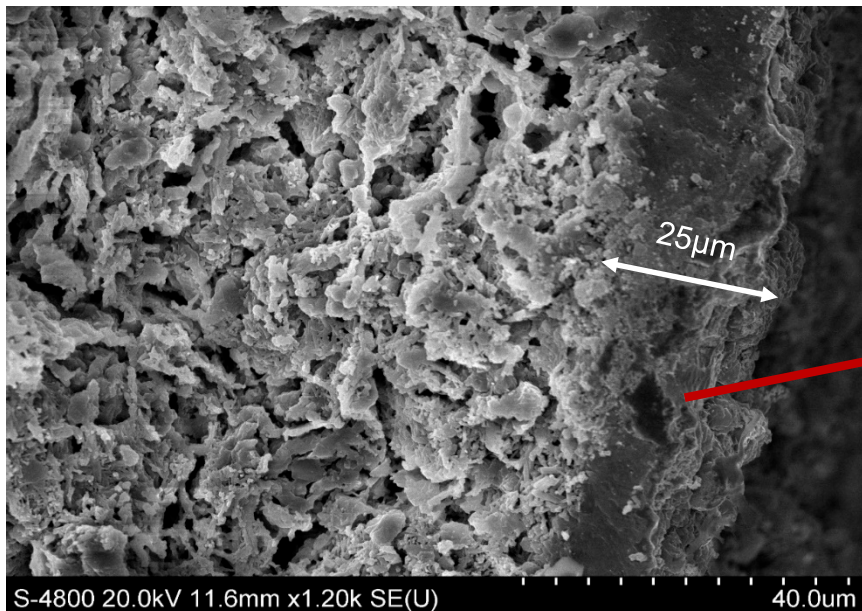
## Success Criteria

A minimum process energy operating condition established based on 100 hour parametric testing for the CAER catalyzed advanced amine solvent showing at least a 20% reduction in stripping energy compared to uncatalyzed CAER-B3 solvent.	80%
The validation to show the advantages of the CAER catalyzed amine to the uncatalyzed CAER advanced amine solvent verifying at least a 10% increase in overall mass transfer or 5% more rich solution at the CAER bench-scale evaluation.	80%
Zeolite membrane performance at the lab-scale maintains chemical and mechanical stability for 100 hour run verified.	100%

Scale (g)	Ligand Yield (%)	Ligand Purity (%)	Catalyst Yield (%)	Catalyst Purity (%)
1	> 90	> 90	50	> 90
5	> 90	> 90	86	> 90
20	70	> 90	77	> 90
50	> 90	> 90	81	> 90
100	> 90	> 90	92	> 90



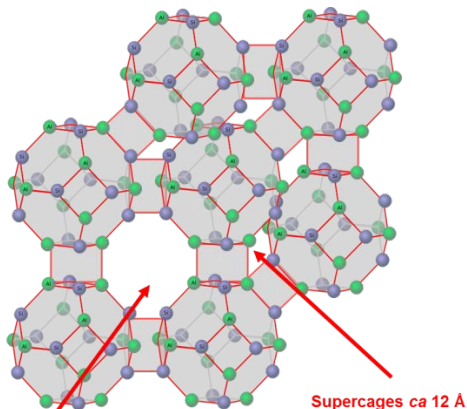
Zeolite layer *ca* 50 -60  $\mu\text{m}$



Y Zeolite crystals

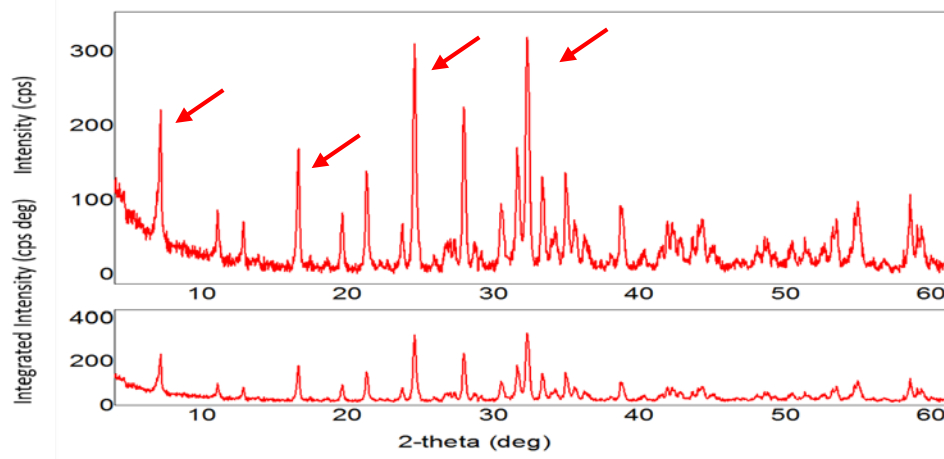
Successful creation of ~25 μm crystalline zeolite layer

Y Zeolite layer ca 25-30 μm



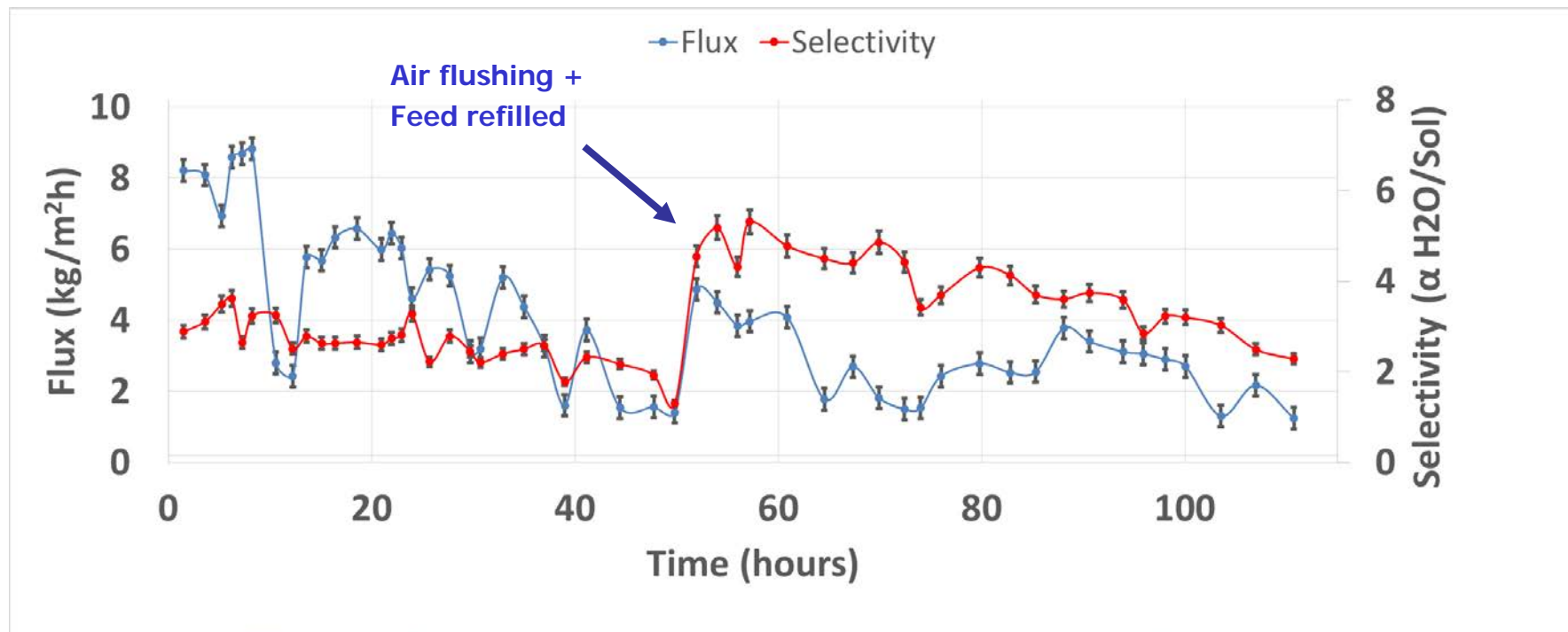
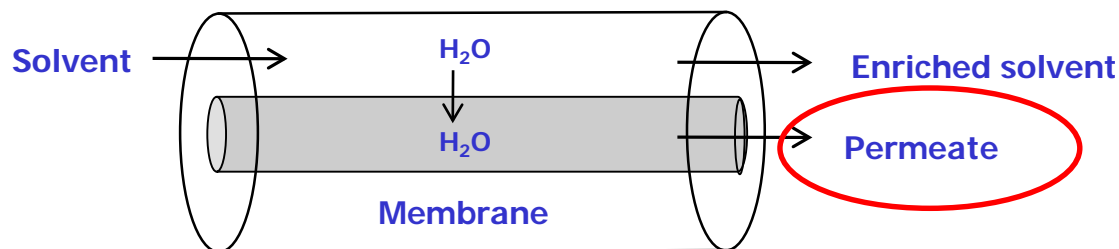
Pore size ca 7 Å

Characteristic Y zeolite peaks



# Long term dewatering operation

Membrane: TMAOH/ 1.0 % seed solution / 3 s deposition / 10h crystallization (100°C)  
 Conditions: CAER-B3 solvent  $\alpha=0.4$  , 100°C, 70 PSI, 20 mL/min



## High packing density of separation area

Values based on permeate flux of 10kg/m<sup>2</sup>h and flow rate of 40 mL/min



l = 10 cm

Active area of each membrane = 23.5 cm<sup>2</sup>

Target Permeate flow rate = 200ml/min

Assuming 20% rejection = 91 membrane tubes



l = 18cm

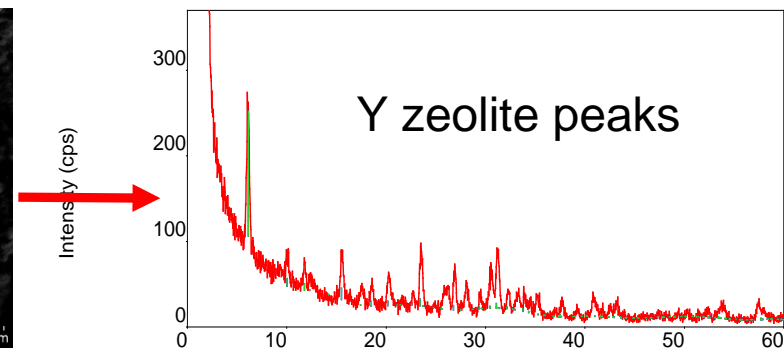
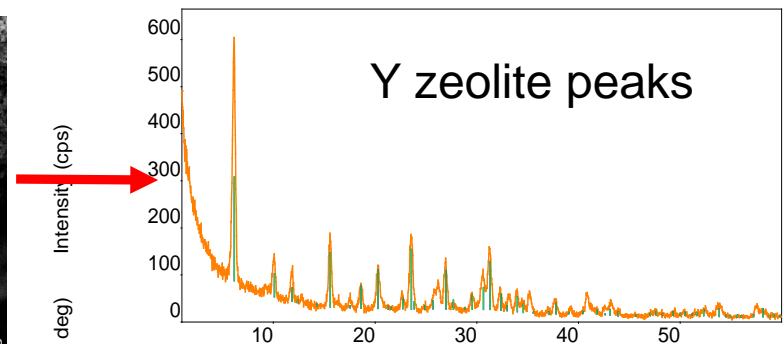
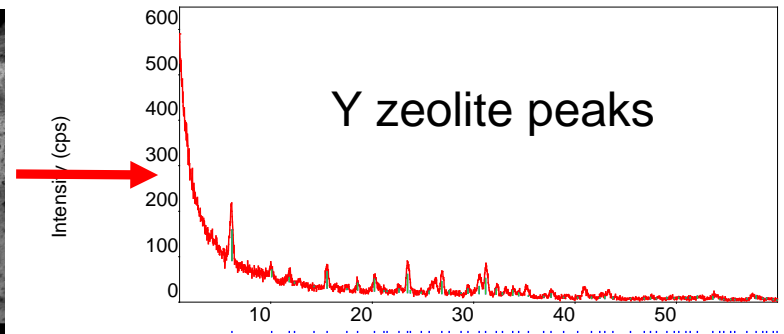
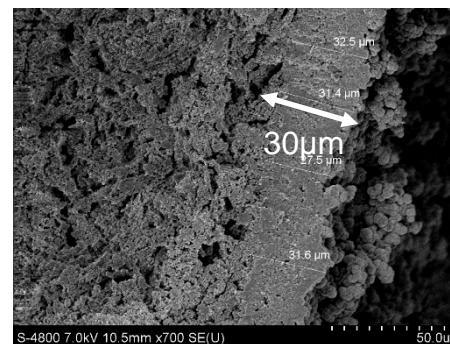
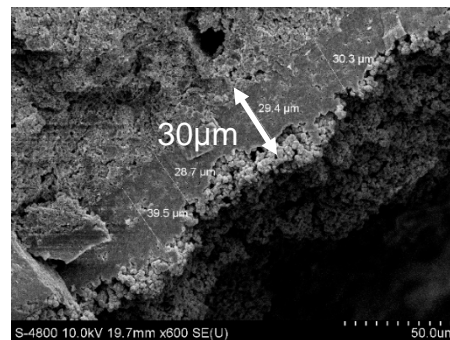
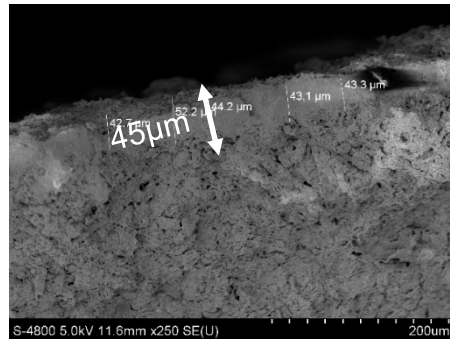
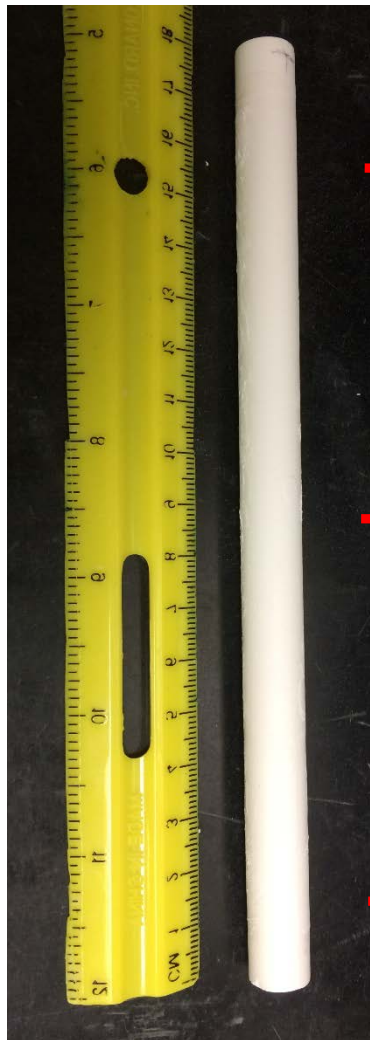
Active area of each membrane = 48.7 cm<sup>2</sup>

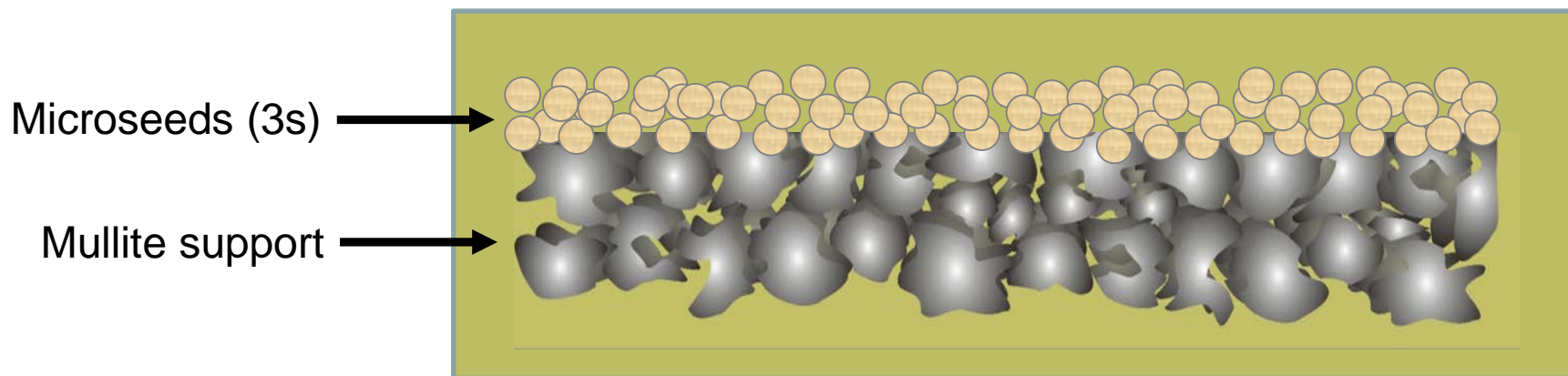
Target Permeate flow rate = 200ml/min

Assuming 20% rejection = 44 membrane tubes

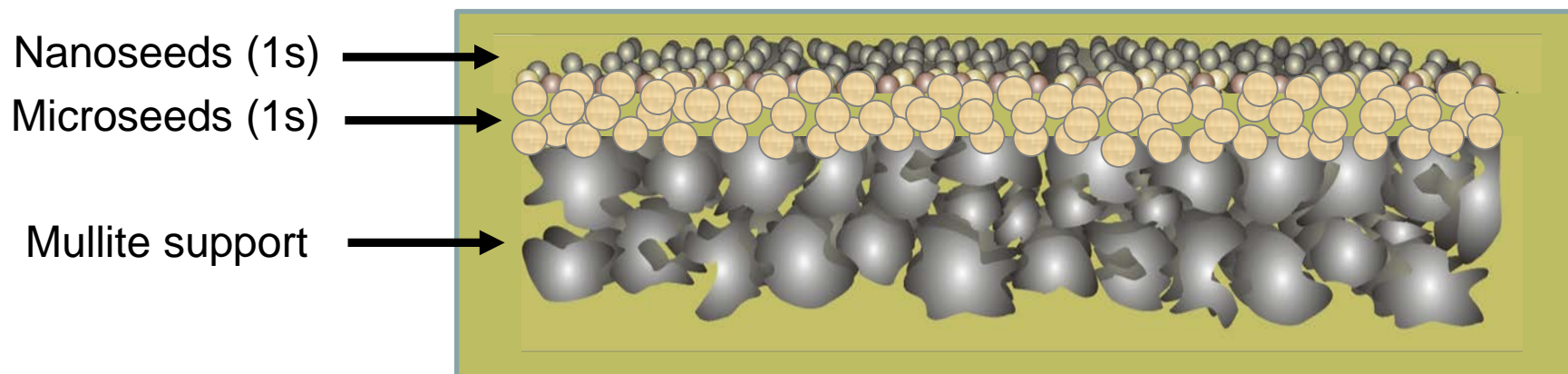


**6 Tubes/Module**



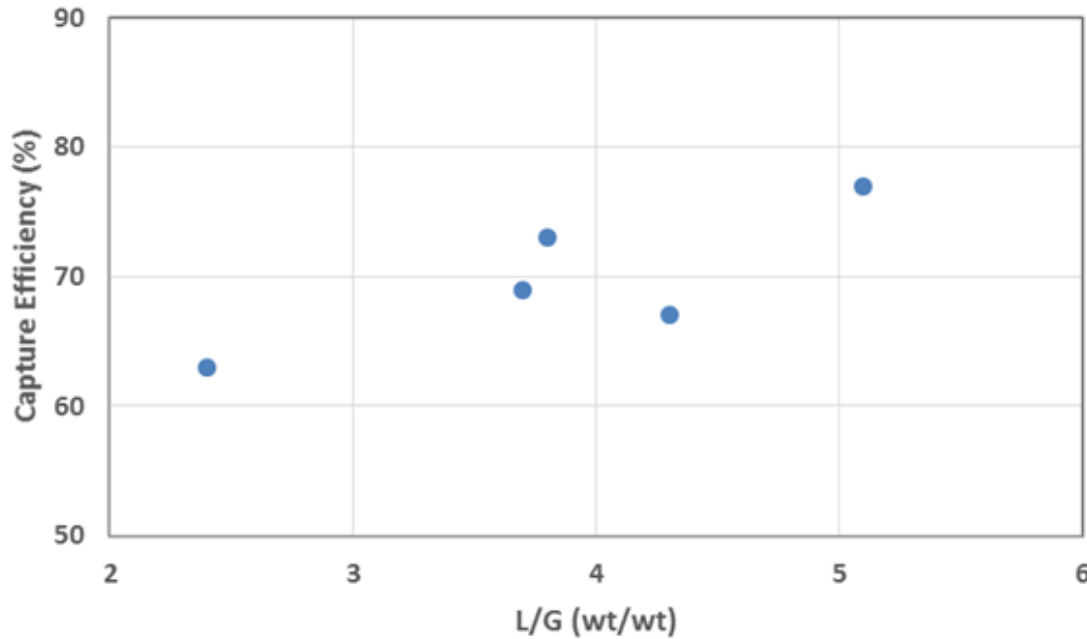


- Use of nanoseeds (OSU) towards the production of a defect-free membrane



- Ideal thickness 5-10  $\mu\text{m}$  to produce higher flux





- Different L/G ratio from variation in liquid circulation
- Absorber inlet temperature (30, 40°C)
- Stripper pressure – (45, 55 psia)



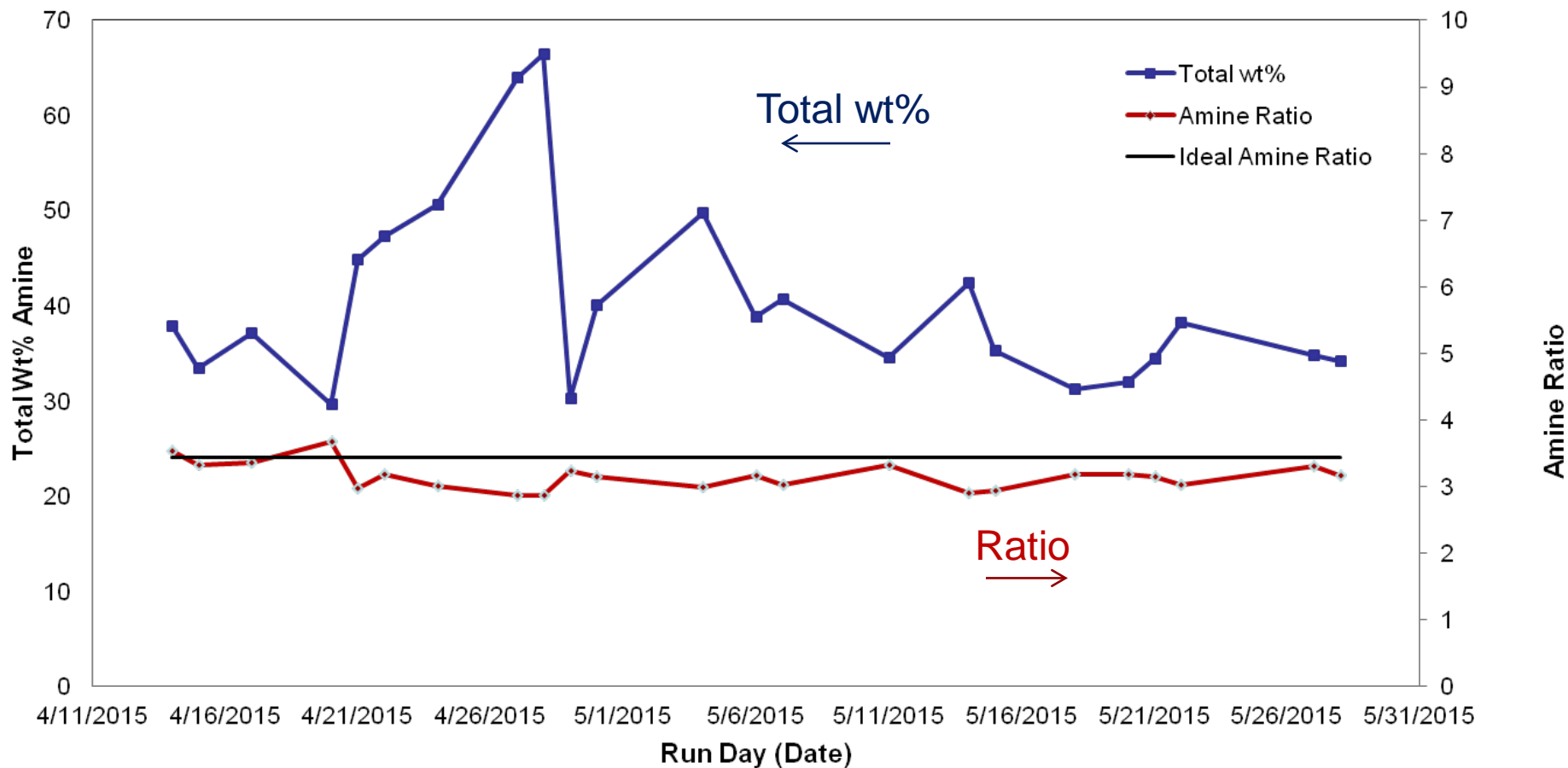
Experiments at liquid load of  $\sim 19 \text{ m}^3/(\text{m}^2 \cdot \text{h})$  and stripper pressure of 45 psia, absorber inlet temperature of 40 °C.

Baseline – No Catalyst Added

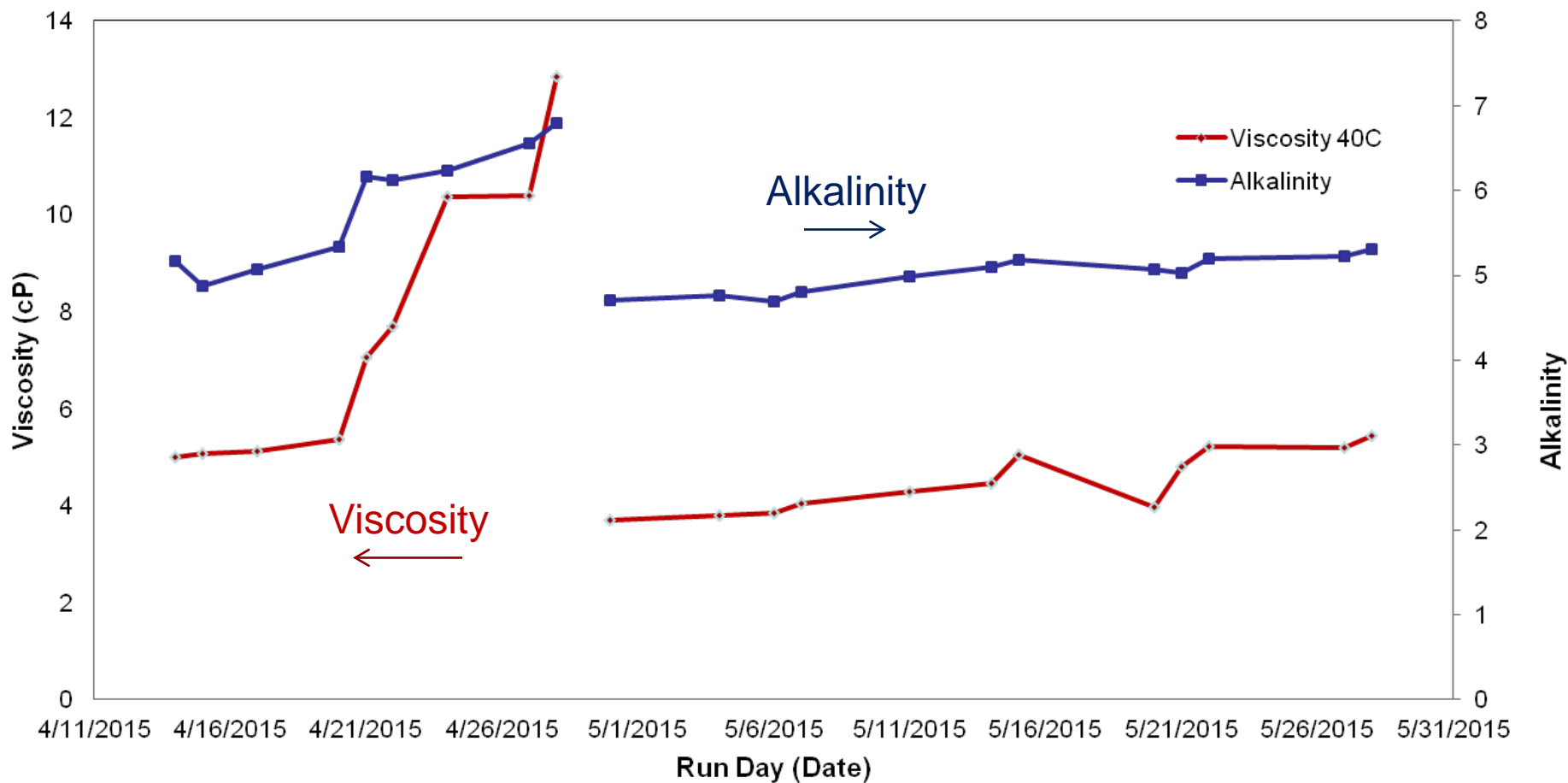
	Total amine wt%	Alkalinity (mol/kg)	Viscosity @ 40 °C (cP)	% Capture	Energy (Btu/lb CO <sub>2</sub> )
<b>April</b>	29.7	5.34	4.67	81	2006
	66.5	7.04	12.84	76	1980
	40.2	4.88	3.76	81	2114
<b>May</b>	38.8	4.87	3.85	81	2007
	40.66	4.98	4.04	75	2135
	36.61	5.05	4.28	79	2107
	42.45	5.10	4.46	79	2085
	35.3	5.18	5.06	77	2034
	36.41	5.03	3.98	78	2005

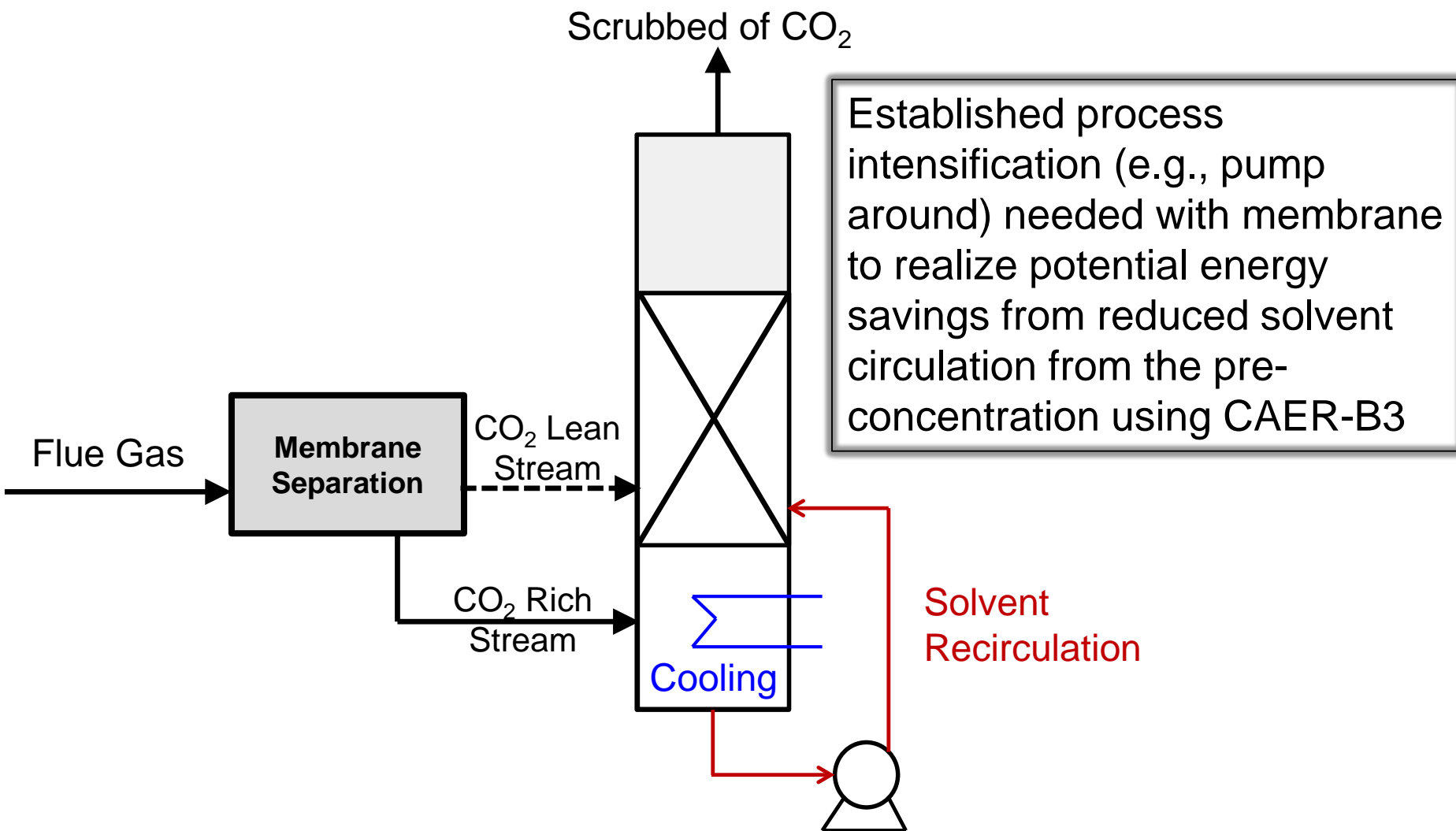
Maintained similar capture for a range of varying relative concentrations of amines in blend

- Slow decrease in [amine]
  - most likely from aerosol emissions



- Slow increase in alkalinity and viscosity





- Membrane successfully integrated with bench unit with accompanying vacuum pump and blower
  - Modified process
- Demonstrated CO<sub>2</sub> pre-concentration by a factor of ~2 obtainable in system







Run	L/G wt/wt	Liquid load m <sup>3</sup> /(m <sup>2</sup> .h)	Absorber LMTD (°C)	Stripper bottom temp. (°C)	Capture efficiency %	Energy (Btu/lb CO <sub>2</sub> )
Ref 1	5.1	19	50	134	77	2091
Ref 2	5.3	19	47	135	81	2006
<b>Membrane</b>						
1	4.9	17.8	43	137	82	1650
2	3.8	13.6	41	138	80	1473
3	3.7	13.4	42	140	82	1515
4	3.1	10.8	42	140	88	1412
5	2.9	10.6	43	140	86	1465

Energy savings of ~25% could be obtained from the reduced L/G runs with the membrane.

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- Carbon Management Research Group: Duke Energy, EPRI, LGE&KU, DEDI, AEP

		<h2>CMRG</h2>		
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