

# Predicting the Oxidation/Corrosion Performance of Structural Alloys in Supercritical CO<sub>2</sub>

**DE-FE0024120**

*S.C. Kung, J.P. Shingledecker, T. Lolla (EPRI)*

*I.G. Wright (WrightHT, Inc)*

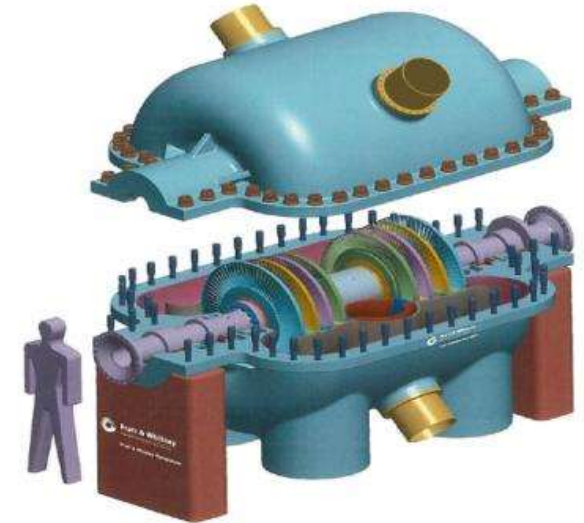
*A.S. Sabau (ORNL)*

2017 UTSR Project Review  
*Pittsburgh, PA*  
*November 2, 2017*



# Project objectives

- Overall Objectives
  - *predict the oxidation/corrosion performance of structural alloys in high-temperature high-pressure supercritical CO<sub>2</sub> (sCO<sub>2</sub>)*
  - *combine laboratory testing & computational modeling, incorporating unique attributes of sCO<sub>2</sub> heat exchangers and recuperators, to accomplish this goal*
- Identify materials to help enable U.S. DOE Program Goals for future sCO<sub>2</sub> Transformational Power Systems



sCO<sub>2</sub> Power Turbine  
(676 MW)

Graphics by Aerojet Rocketdyne,  
used with permission

*Some configurations of the sCO<sub>2</sub> Brayton power cycle might achieve **100% carbon dioxide capture and zero emissions of conventional pollutants** with little or no efficiency or capacity penalty.*

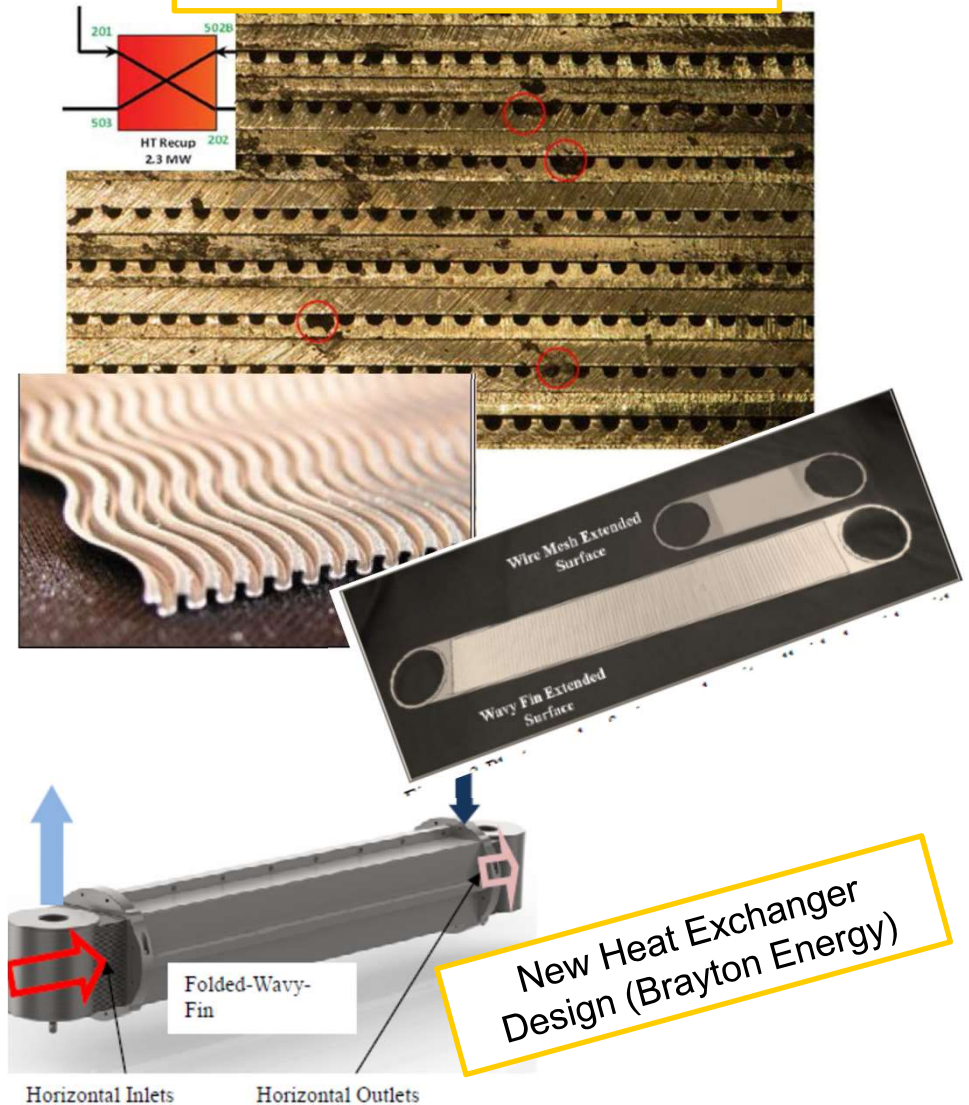
# Technology Challenges for sCO<sub>2</sub>

- HX Expensive: about 40% of the plant cost
- Unique designs
  - *small channels*
  - *large surface areas*
- Materials considerations
  - *thermal fatigue, creep (thin sections)*
  - *brazing/diffusion bonding*
  - *corrosion/oxidation/carburization*

## Corrosion/Oxidation

- Closed cycle = build-up of impurities
- **Open cycle = combustion products**
- Long-term performance, pluggage, blockage, etc.

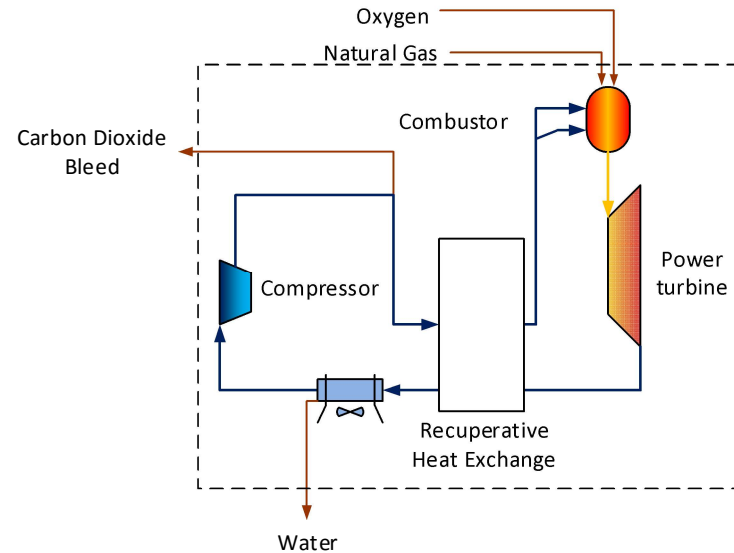
Compact Heat Exchanger Fouling (Sandia National Lab)



New Heat Exchanger Design (Brayton Energy)

# Realistic sCO<sub>2</sub> conditions for (open) Allam cycles

- **Survey of industry and current studies**
  - 700°C likely maximum temperature in heat-exchangers
- **Evaluation of impurities for near-term ‘open/direct-fired cycle’ – Allam Cycle**
  - H<sub>2</sub>O, O<sub>2</sub>, N<sub>2</sub>, Ar, NO<sub>x</sub>, SO<sub>x</sub>, HCl
  - mass-balance calculations for methane and cooled, raw syngas (checked against thermodynamic calculations)



Species	Composition (mol%)		
	Methane	Cooled raw coal syngas	Oxygen
CH <sub>4</sub>	100	1.0	
CO		39.0	
H <sub>2</sub>		28.3	
CO <sub>2</sub>		8.0	
H <sub>2</sub> O		20.0	
N <sub>2</sub> +Ar		2.0	0.5
H <sub>2</sub> S	2 ppm	0.9	
HCl		0.02	
O <sub>2</sub>			99.5
LHV	912 BTU/scf	218 BTU/scf	



Component	Composition (mol%)			
	Methane		Cooled Raw Coal Syngas	
	Combustor Inlet	Turbine Inlet	Combustor inlet	Turbine Inlet
CO <sub>2</sub>	95	90	90	85
H <sub>2</sub> O	250 ppm	5.3	250 ppm	5
N <sub>2</sub> +Ar	1	1	9	9
O <sub>2</sub>	3.8	3.6	1	1
HCl				20 ppm
SO <sub>2</sub>				1,000 ppm



**O<sub>2</sub> = 3.6 vol%, H<sub>2</sub>O = 5.3 vol%**



# Scope of Laboratory sCO<sub>2</sub> Corrosion Tests

- Conditions

- 650-750°C, 200 bar
- sCO<sub>2</sub>
  - commercial purity
  - simulated semi-open cycle impurities (O<sub>2</sub> + H<sub>2</sub>O)

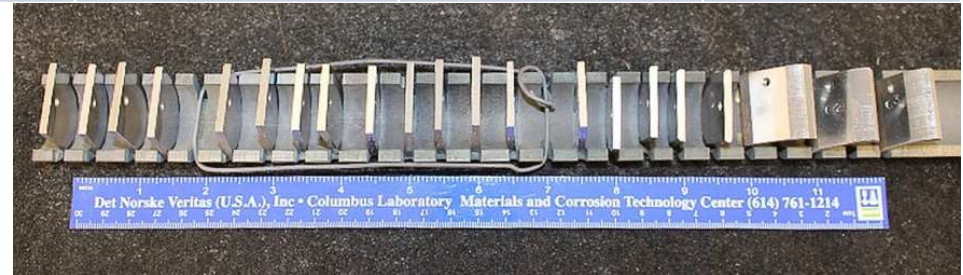
- Materials

- *commercially available*
- *Code approved/industry relevant*
- *focus on economics*

- Exposures

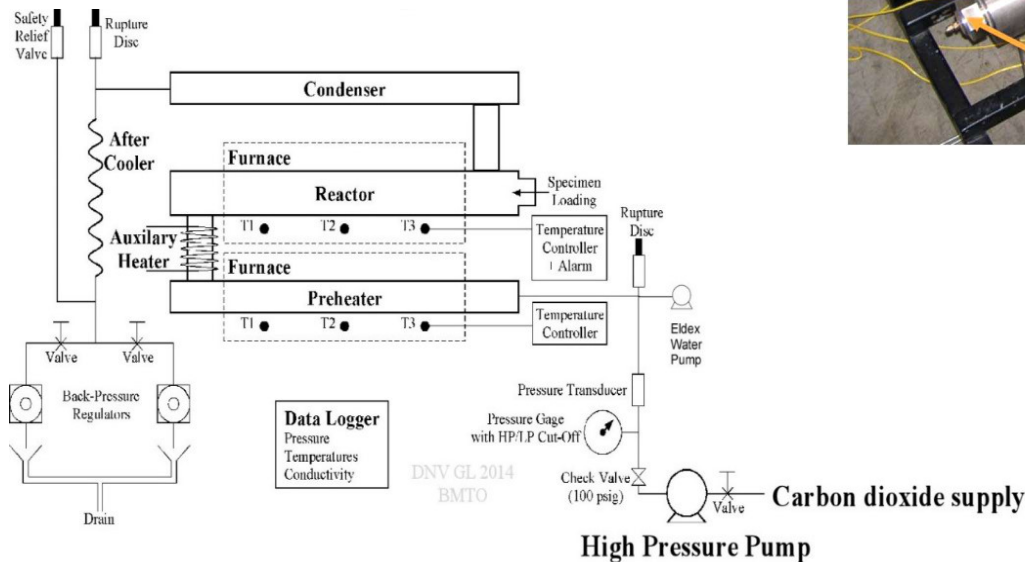
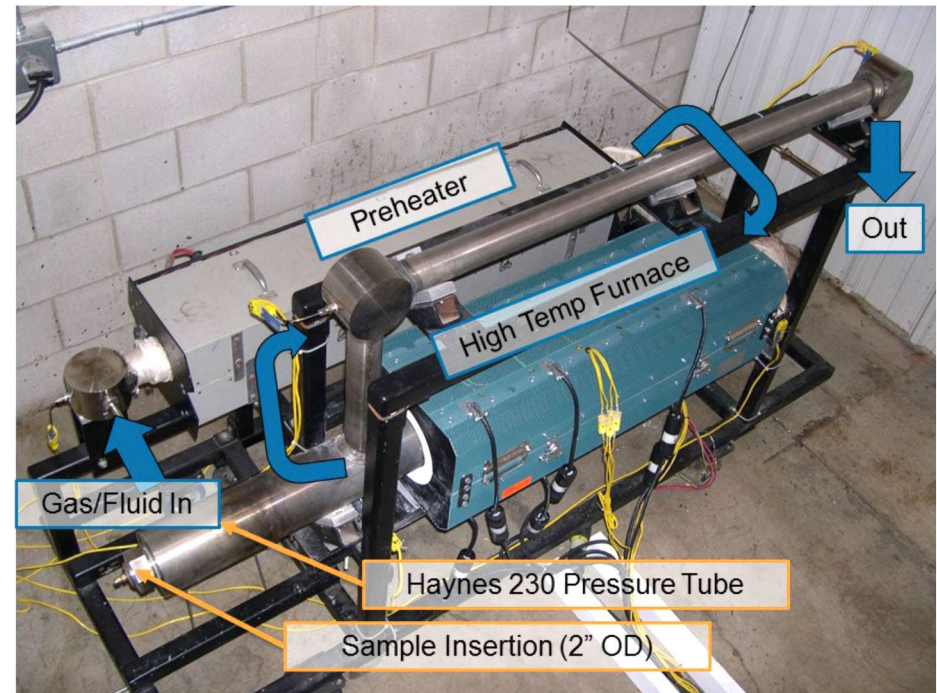
- *2 x 300-h shakedown tests in CO<sub>2</sub> ± impurities, 700°C, 200 bar (Gr91, TP304H, IN740H)*
- *3 x 1,000-h tests in CO<sub>2</sub> + impurities, 650, 700, 750°C, 200 bar (all 7 alloys)*
- *1 x 5,000-h test in CO<sub>2</sub> + impurities, 700°C, 200 bar (all 7 alloys), with scheduled sample retrievals*

Material Class	Alloys Selected		
Ferritic steels	<b>Gr 91</b> (8-9Cr)	<b>VM12</b> (11-12Cr)	<b>Crofer 22H</b> (20Cr)
Austenitic stainless	<b>TP304H</b> (18Cr)	<b>HR3C</b> (25Cr)	
Nickel-based	<b>IN617</b> (20Cr, solid soln. strengthened)	<b>IN740H</b> (25Cr, ppt. strengthened)	



# Laboratory testing facility

- Existing test facility modified for sCO<sub>2</sub> to ensure safety
- 650-750°C, 200 bar
- Introduction of impurities (3.6 O<sub>2</sub>, 5.3 H<sub>2</sub>O mol%)
- 5,000-h test in sCO<sub>2</sub> with impurities completed successfully



**Gas Volume: 4.41 liters**

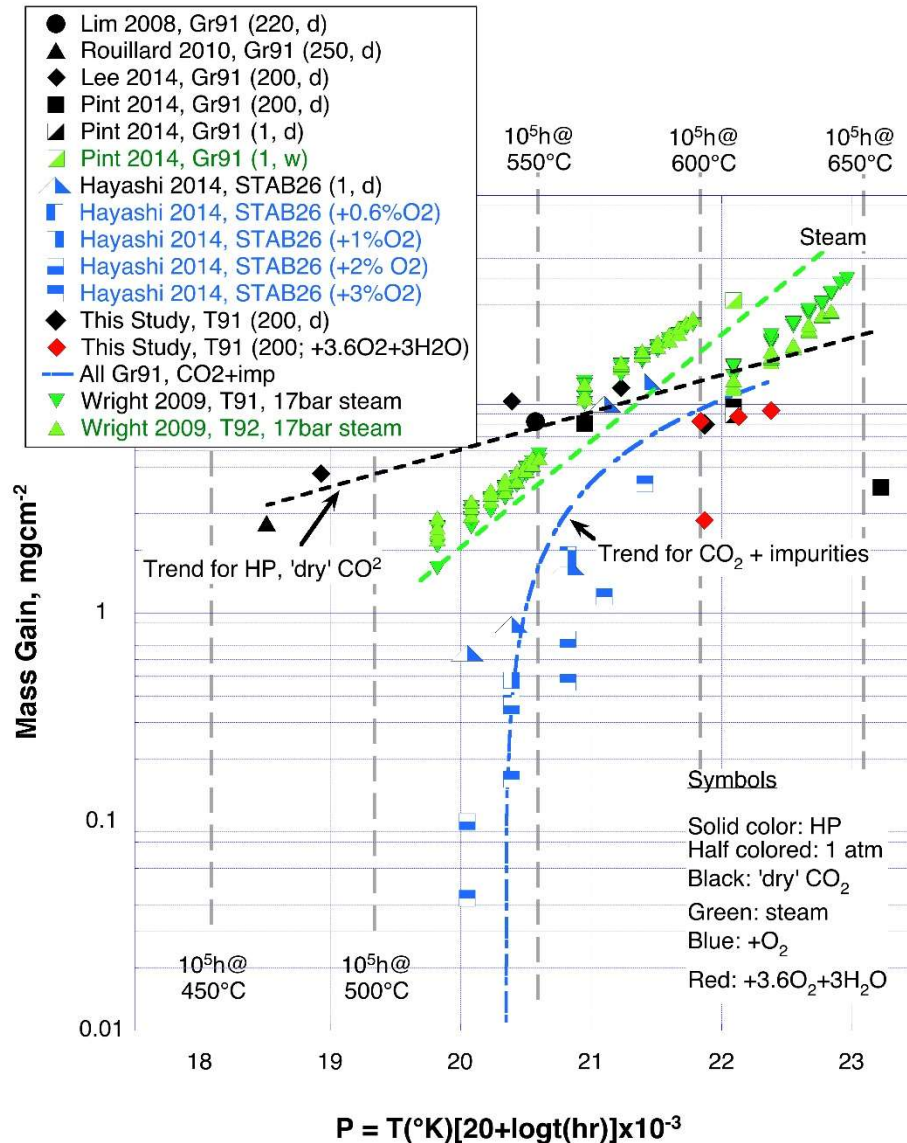
**Refresh Rate: static, with occasional replenishment**

**Temperatures: preheat 454°C. exit 149°C**

## Information needed for development of sCO<sub>2</sub> Exfoliation Model

- Oxidation rates as a function of oxide thickness vs.  $t$  and  $T$ 
  - *for oxide thickness:  $d^2 = 2.k_p.t$ , where:  $k_p = Ae^{-Q/RT}$ , or  $\ln k_p = A-Q/RT$*
  - *$Q$  from slope of an Arrhenius plot*
  - *mass-based oxidation data are of limited value*
    - both oxidation and carburization lead to to weight gain
    - weight gain cannot be easily converted to thickness
    - but, bulk of literature data rely on mass gain
    - nevertheless, mass gain data are useful for comparisons of available data, for examining trends, etc.
  
- Morphological data
  - *needed to infer modes of scale failure*
  - *current EPRI Exfoliation Model is based on morphologies formed in steam*
    - hence, importance of understanding similarities/differences in sCO<sub>2</sub>
  - *for HT alloys, adequate characterization of very thin scales is problematic*

# Ferritic steels: Available mass gain data in sCO<sub>2</sub> and steam



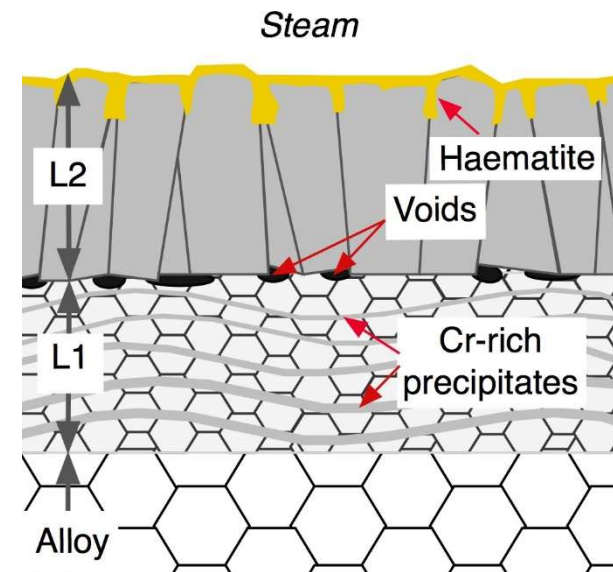
- Pure CO<sub>2</sub>
  - mass gain ±higher than in steam at lower values of Larson-Miller parameter (P)
- Impure CO<sub>2</sub>
  - significantly lower mass gains than in pure CO<sub>2</sub> at lower values of P
- Expect similar morphologies in sCO<sub>2</sub> & steam?
  - only after higher T, longer t?



# Potential differences in scale morphologies on ferritic steels from sCO<sub>2</sub> compared to HP steam?

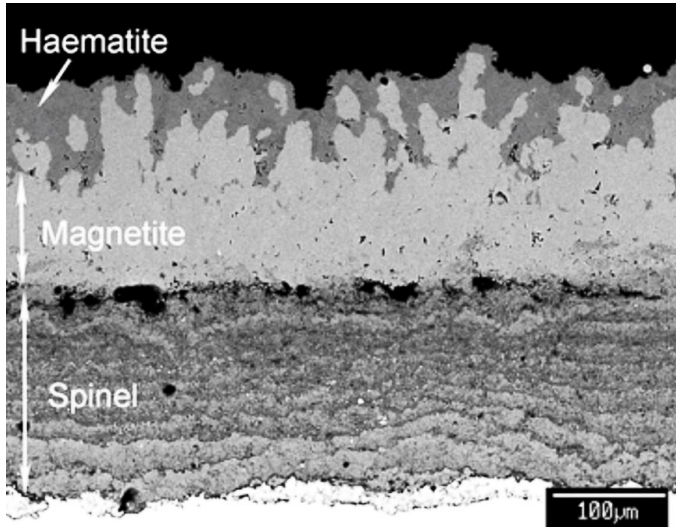
- Steam oxidation: different scale morphologies than in *air*
  - $H_2O = O^{-2} + 2H^+$ : mobility of all 3 in scales
  - for ferritic (& austenitic steels), typically 2-layered scales: Fe-Cr spinel (inner, L1), magnetite (outer, L2)

Schematic representation:  
cross section of fully-developed scale  
on T91 in HP steam (EPRI Atlas) →

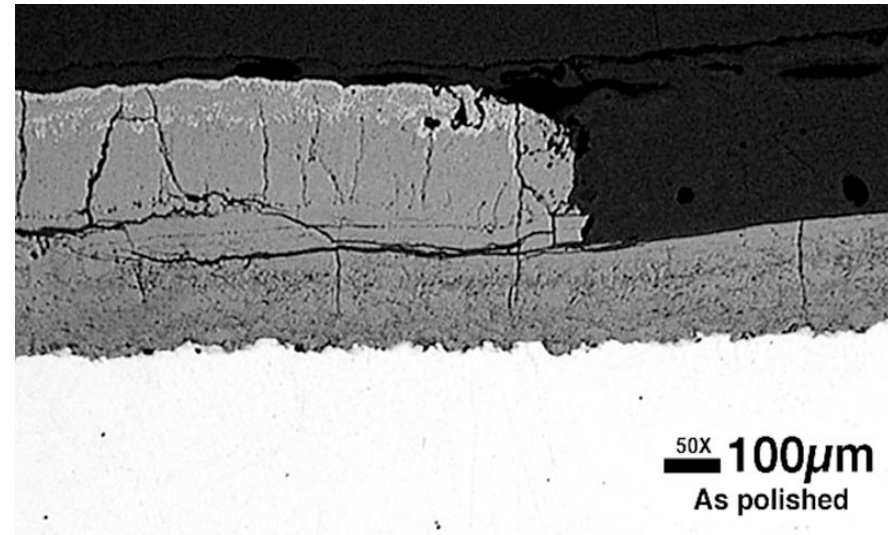


- CO<sub>2</sub> oxidation involves:  $CO_2 = O^{-2} + CO^{+2}$ 
  - similar oxidation process to steam (*i.e.* similar scale morphologies)
  - but, carburization could occur

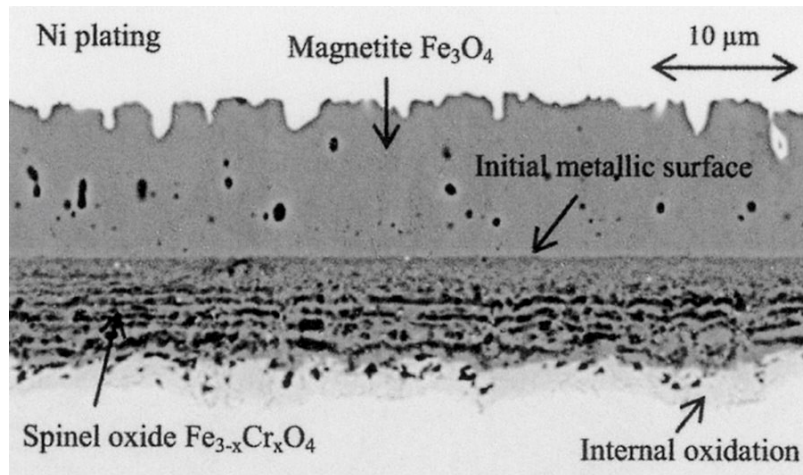
# Typical scale morphologies associated with ferritic steels



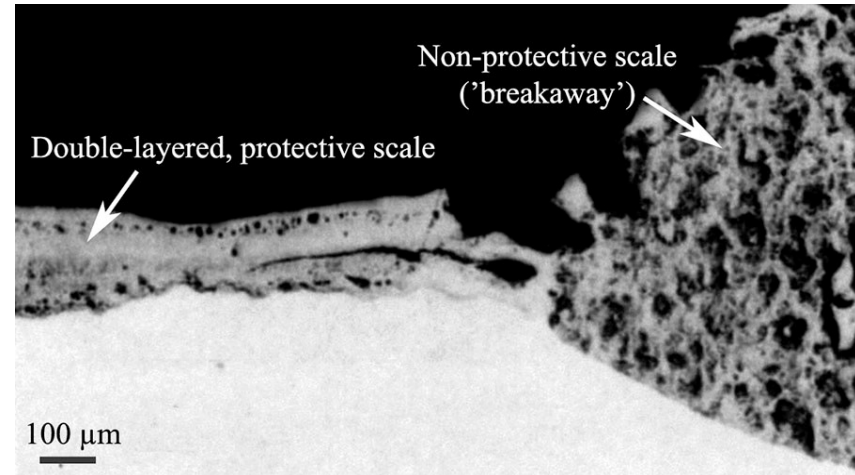
T91, 64kh at 566°C & 138 bar steam (EPRI Atlas)



T91, 155kh at 538°C & 17 bar steam (EPRI Atlas)



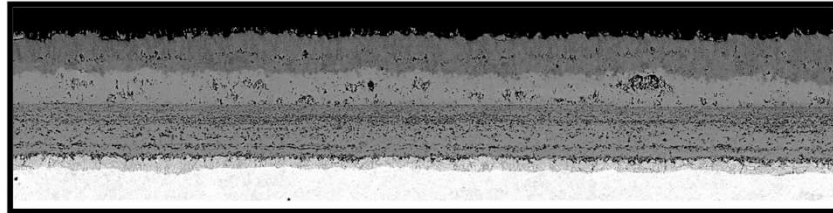
T91, ≈300h at 550°C & 250 bar CO<sub>2</sub> (Rouillard, 2010)



Fe-9Cr, ≤9kh at 550°C & 40 bar (imp) CO<sub>2</sub> (Harrison, 1974)

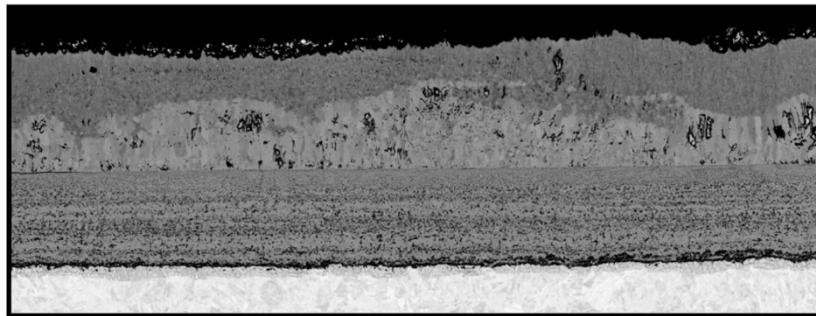
# New results: Gr 91, 700°C in sCO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 200 bar

1,000h



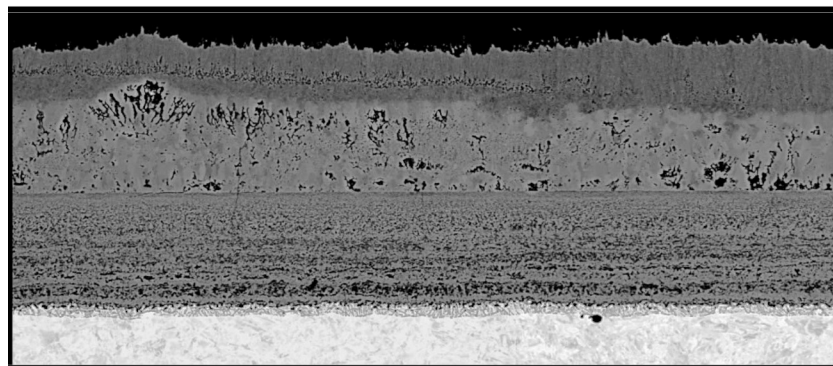
50 μm

3,500h



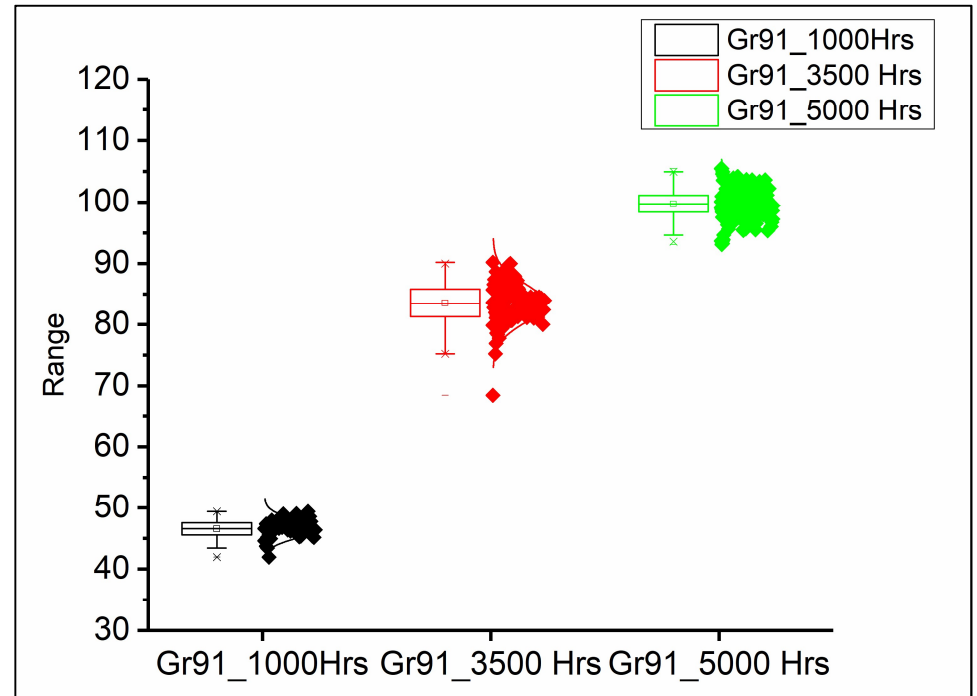
50 μm

5,000h

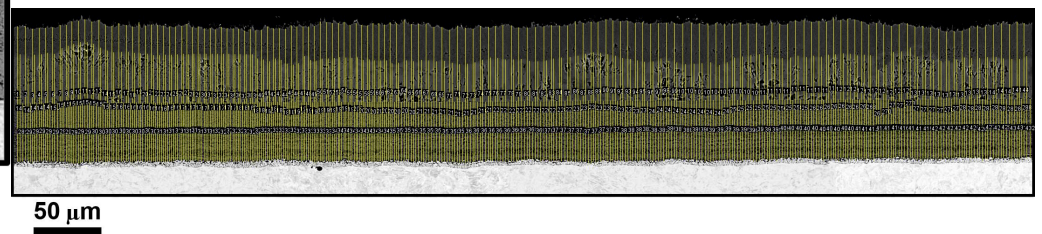


50 μm

Total Scale Thicknesses at 700°C



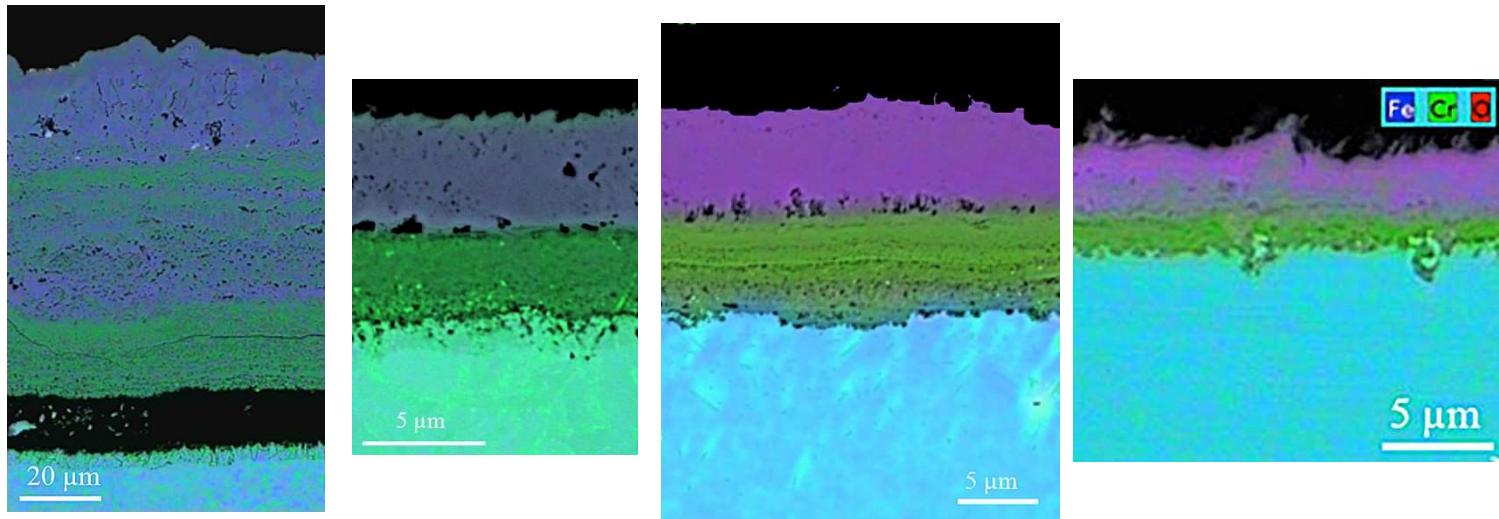
- Total scale thicknesses shown for simplicity
- Individual oxide layers were identified, and their thicknesses also measured



50 μm



## Scale morphologies after 1,000h in CO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 700°C and 200 bar – similar to those formed in steam



**Gr91**

**VM12**

**Crofer 22H**

**TP304H**

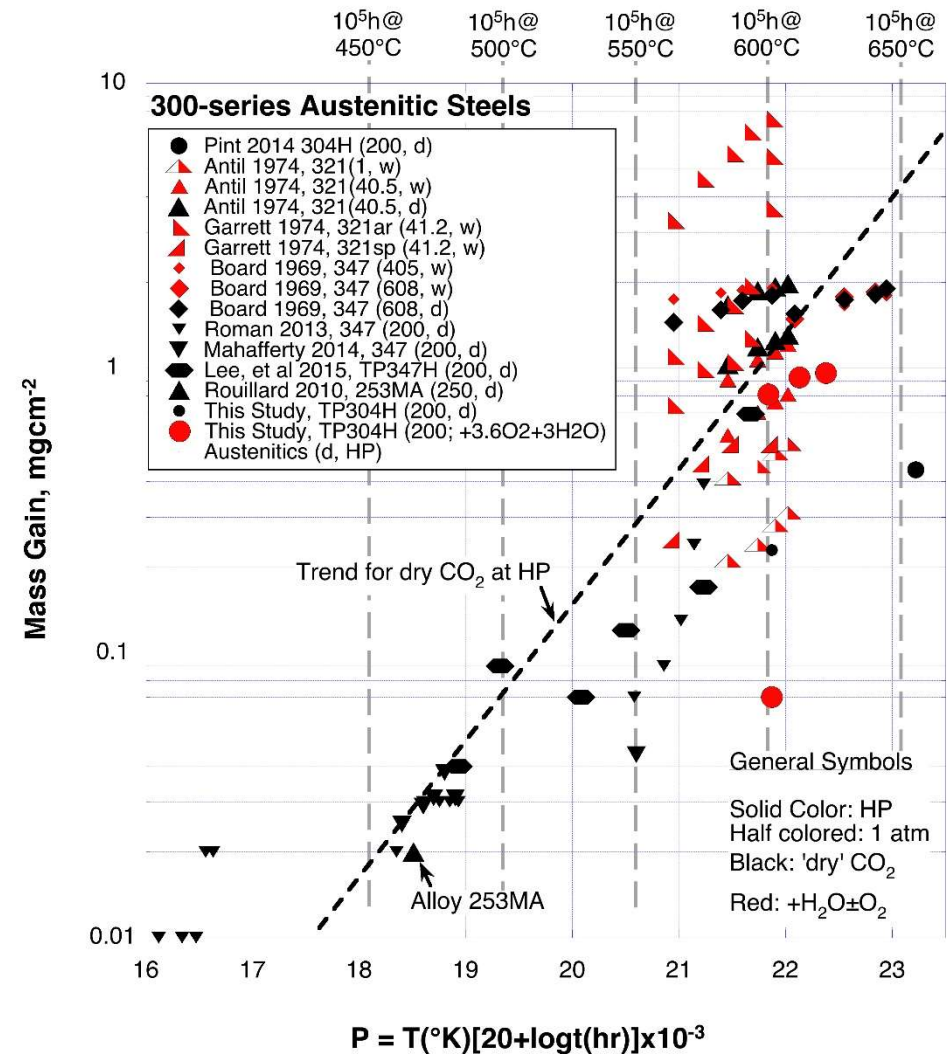
*EDS Maps for Fe, Ni, Cr & O Overlaid on SEM Images*

- Overall, similar structures to those in steam
  - voids associated with exfoliation are present in usual locations
- Some differences in apparent structure of inner layer (L1)
  - Gr91 shows layer(s) intermediate between L1 and L2 with Cr & Fe striations
  - some differences in extent and structure of 'IOZ' at alloy-L1 interface

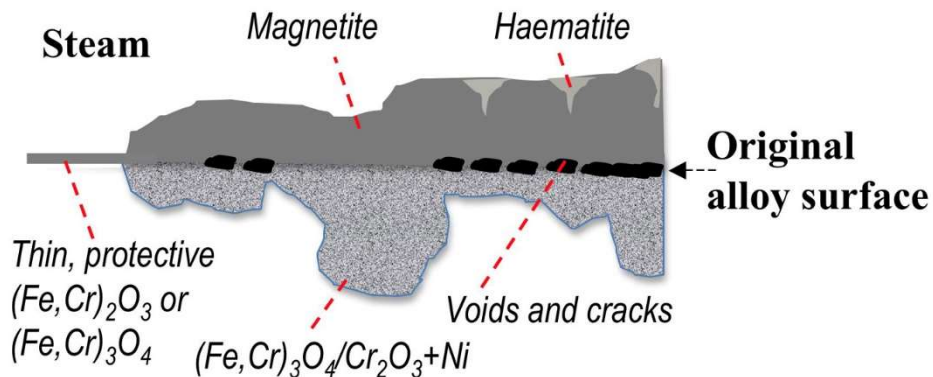


# Effects of impurities in sCO<sub>2</sub> on austenitic stainless steels?

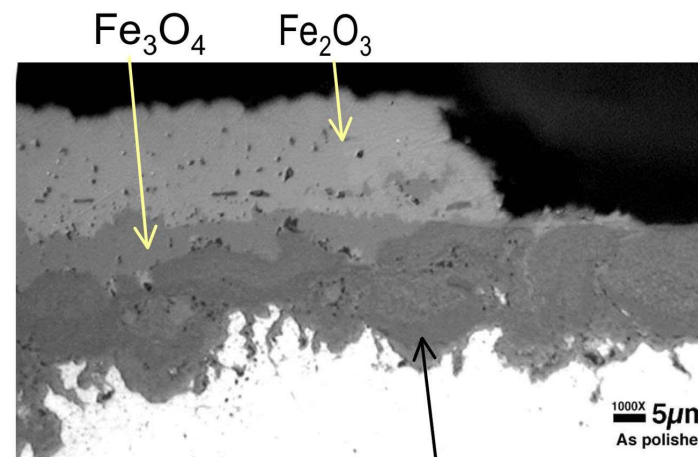
- No real trend for pure vs. impure CO<sub>2</sub>
  - or for HP vs. 1 atm
  - or vs. HP steam (not shown)
- Expect similar morphologies in CO<sub>2</sub>, sCO<sub>2</sub>, ±most impurities, and HP steam



# Scale morphologies observed on austenitic stainless steels



Main features of scale in HP steam (Wright & Dooley, 2011)

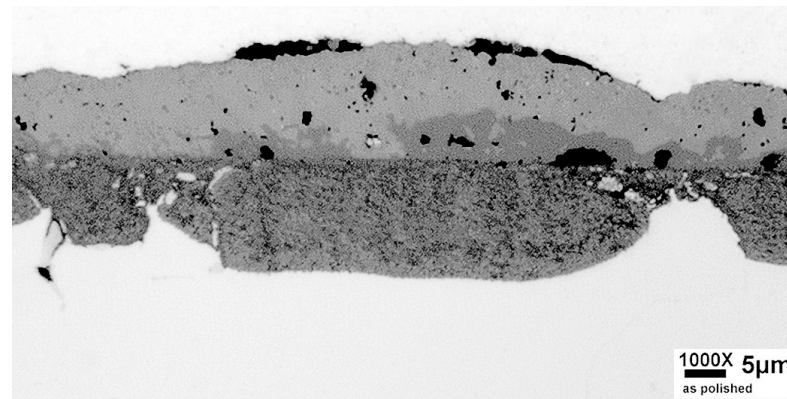


Fe-Cr spinel ( $\approx 28$  at% Cr)

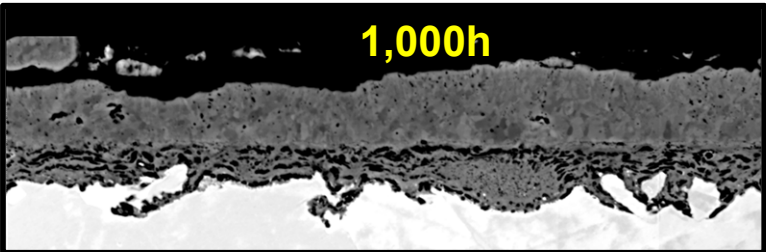
TP347HFG, 11kh at 670°C & 251 bar steam (EPRI Atlas)



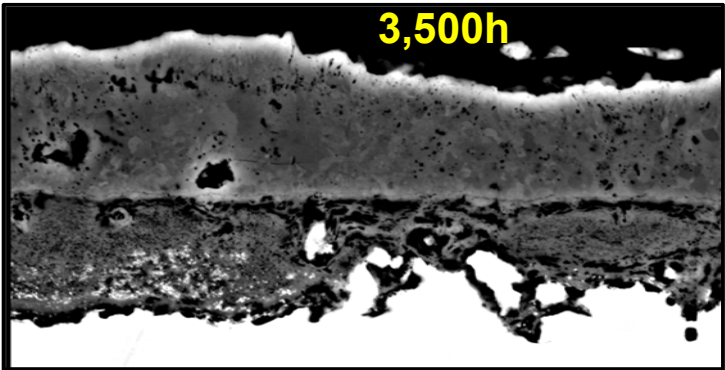
TP347HFG, 500h at 700°C & 200 bar CO<sub>2</sub> (Pint & Keiser, 2014)



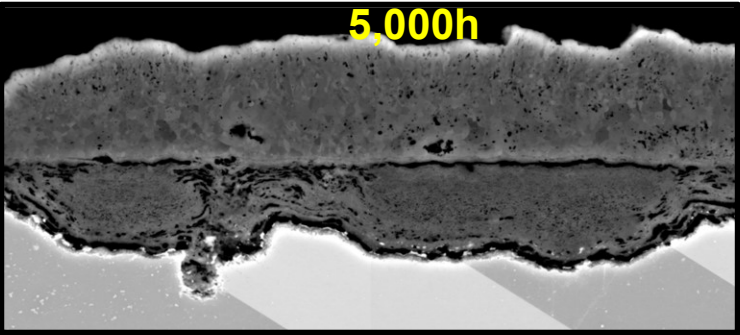
# New results: TP304H, 700°C in CO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 200 bar



10 μm

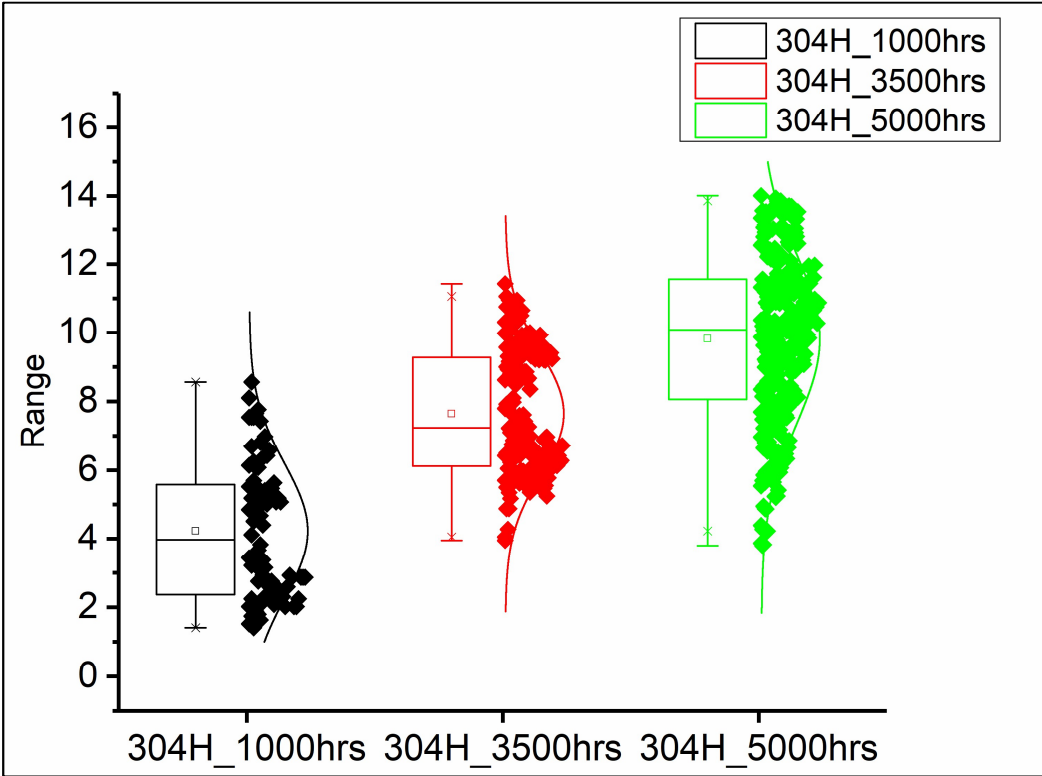


10 μm



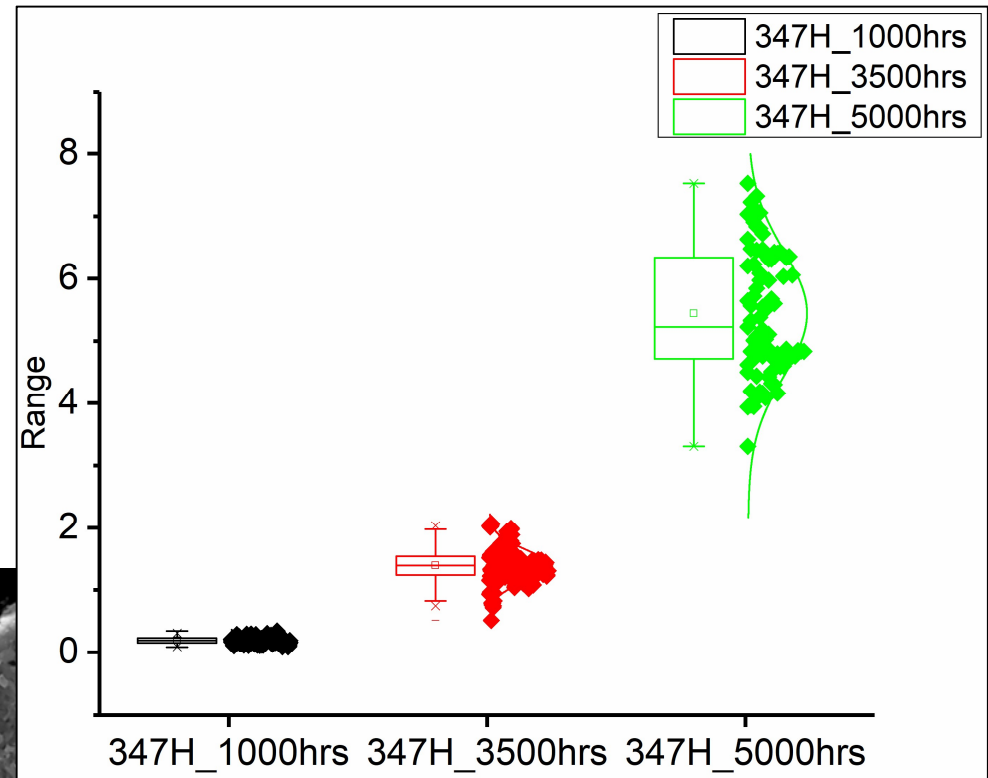
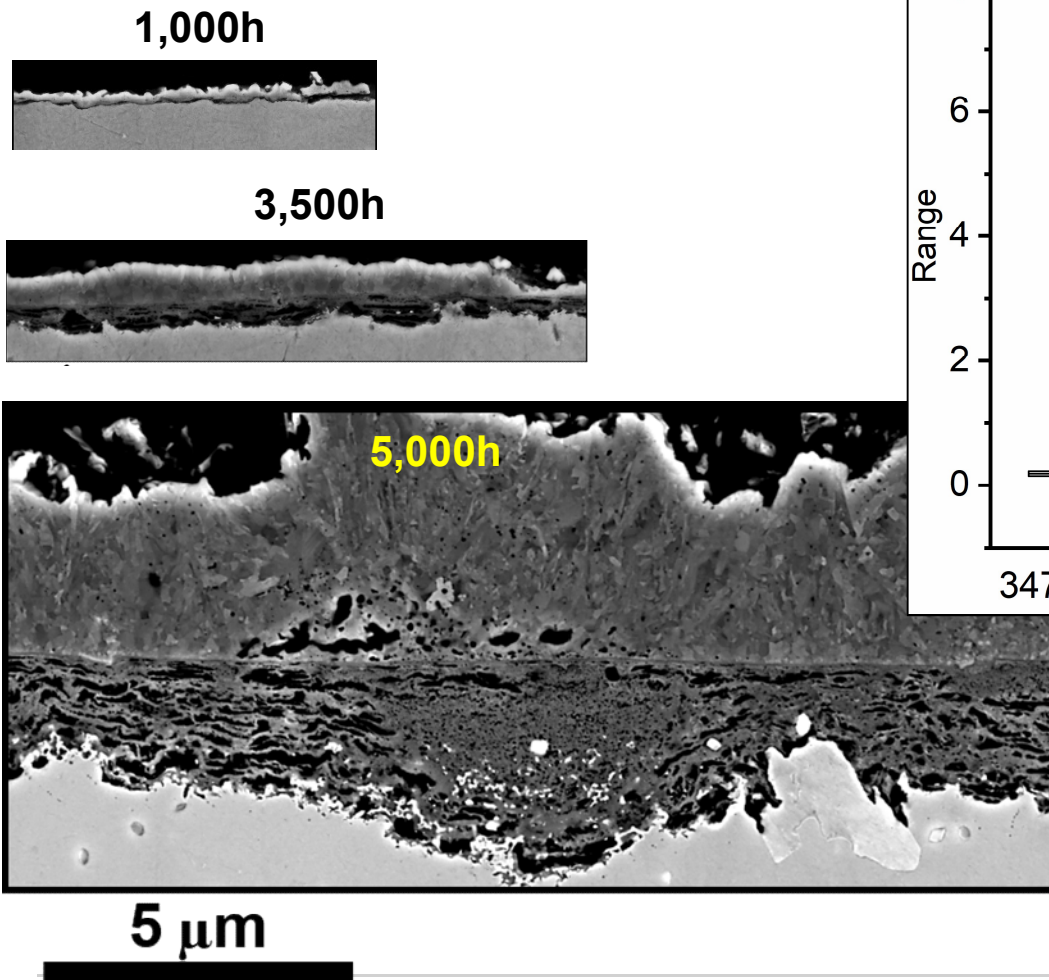
10 μm

### Total Scale Thicknesses at 700°C



# New results: TP347H – initially protective scale

## Total Scale Thicknesses at 700°C

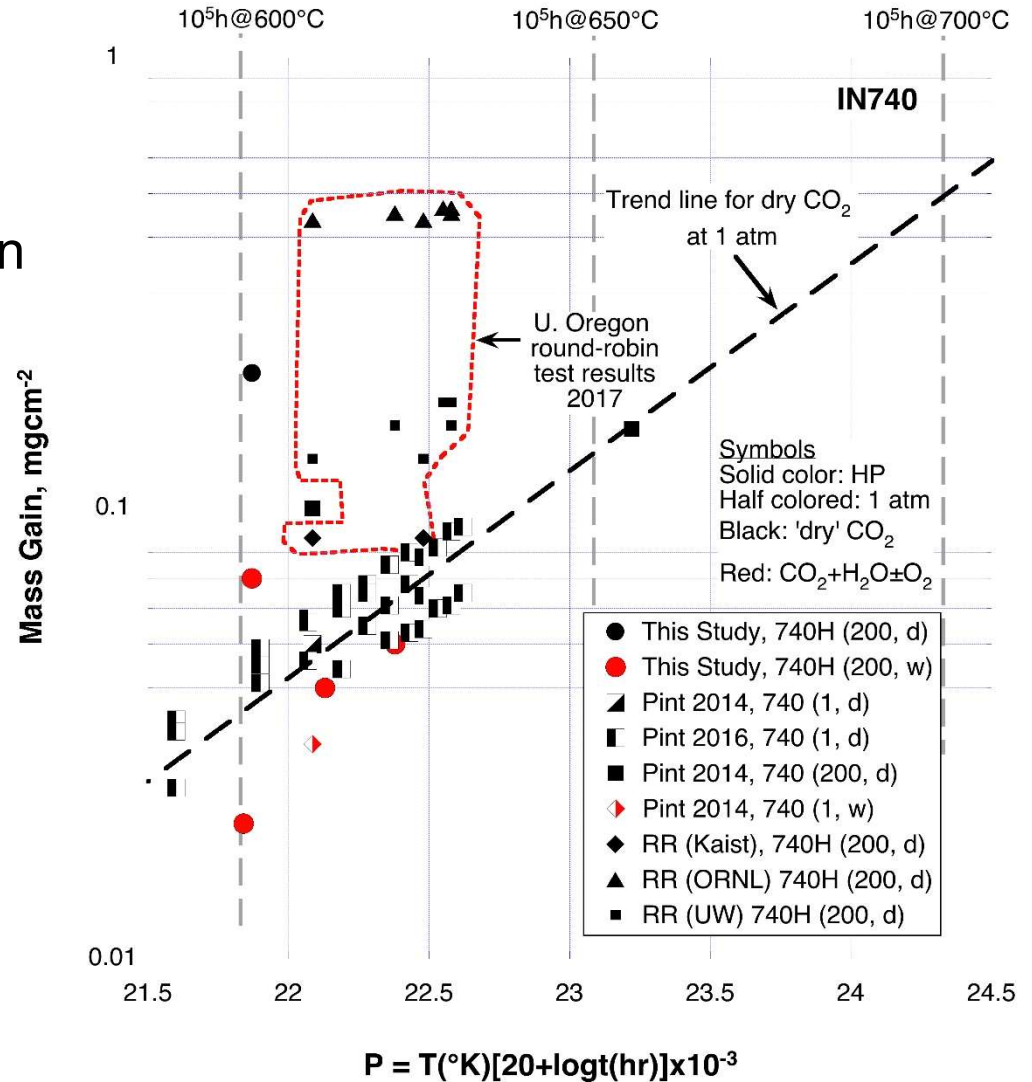




# Effects of Impurities in sCO<sub>2</sub> on IN740H?

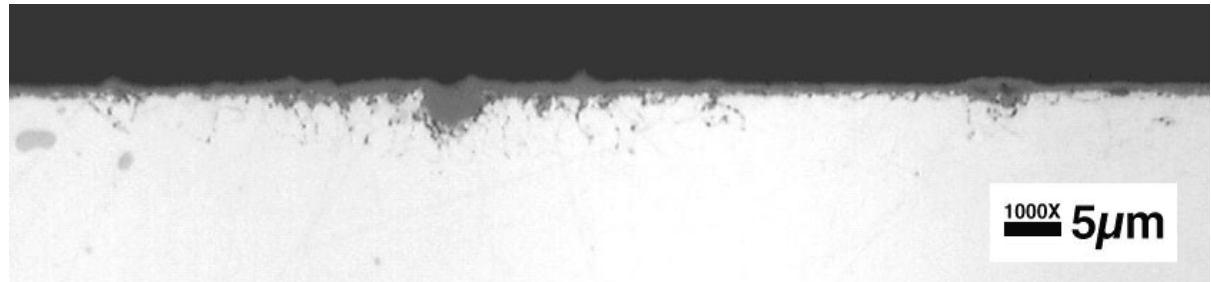
- So far, no obvious effects of impurities in CO<sub>2</sub> on weight gain
- More data are needed

EPRI supplied samples to all sCO<sub>2</sub> Round Robin testing laboratories and provides technical input



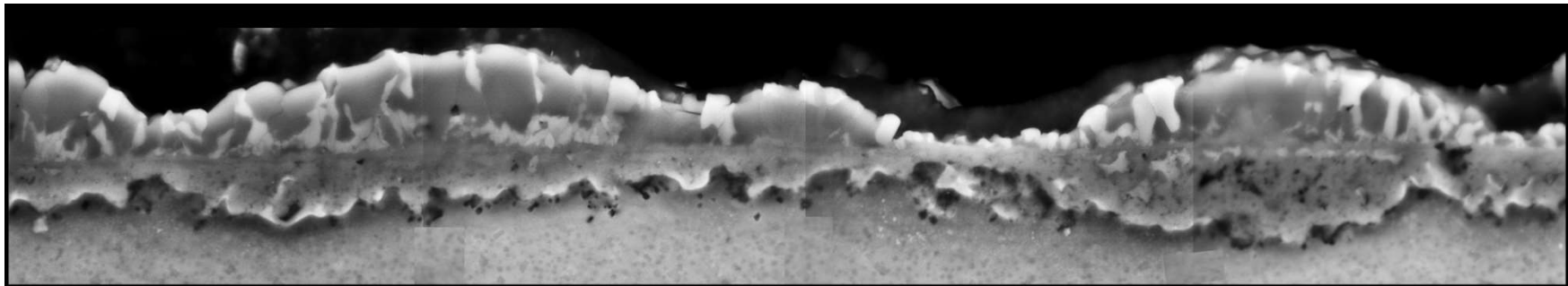
## Thin scales on HT alloys complicate thickness measurements

- Max power of optical microscopy produces marginal resolution



IN740 after 4kh at 700°C in steam at 17 bar (Wright, 2009)

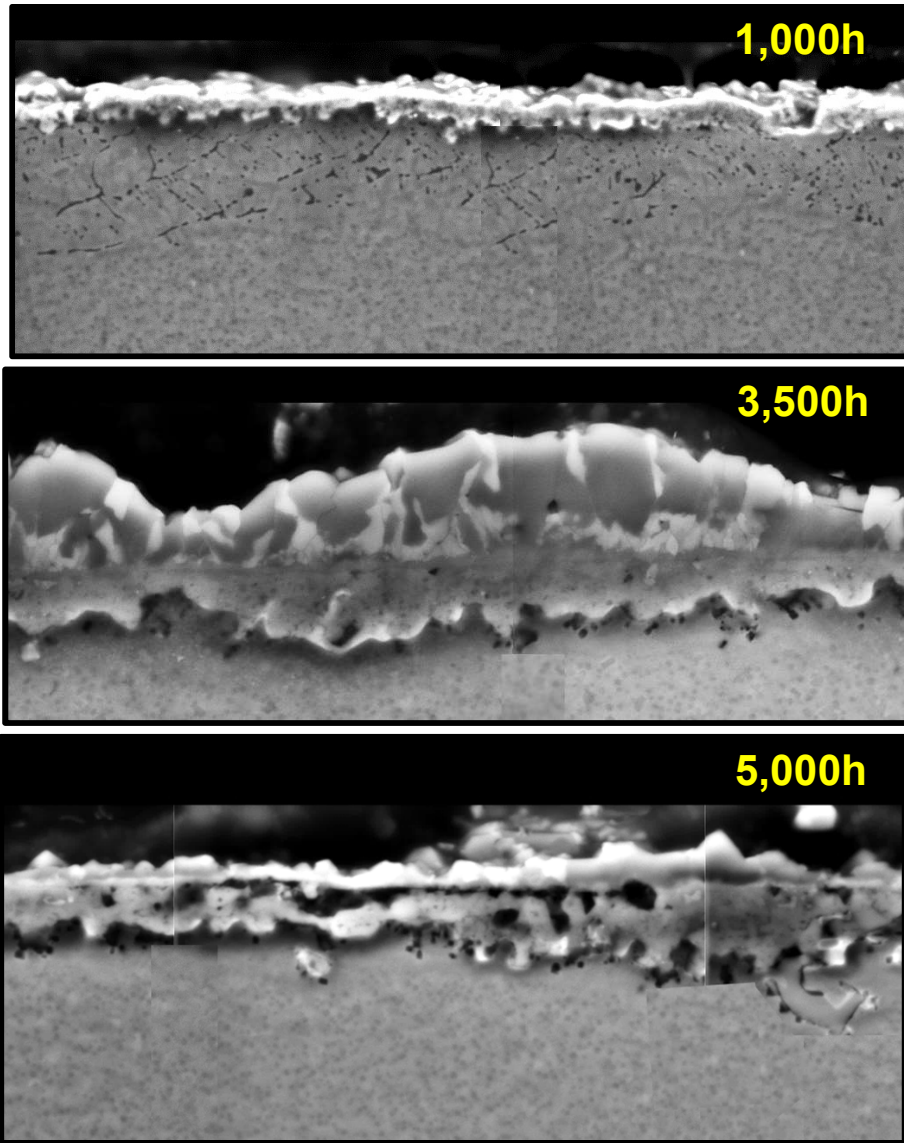
- New SEM/BSE/EDS/EBSD techniques explored by EPRI



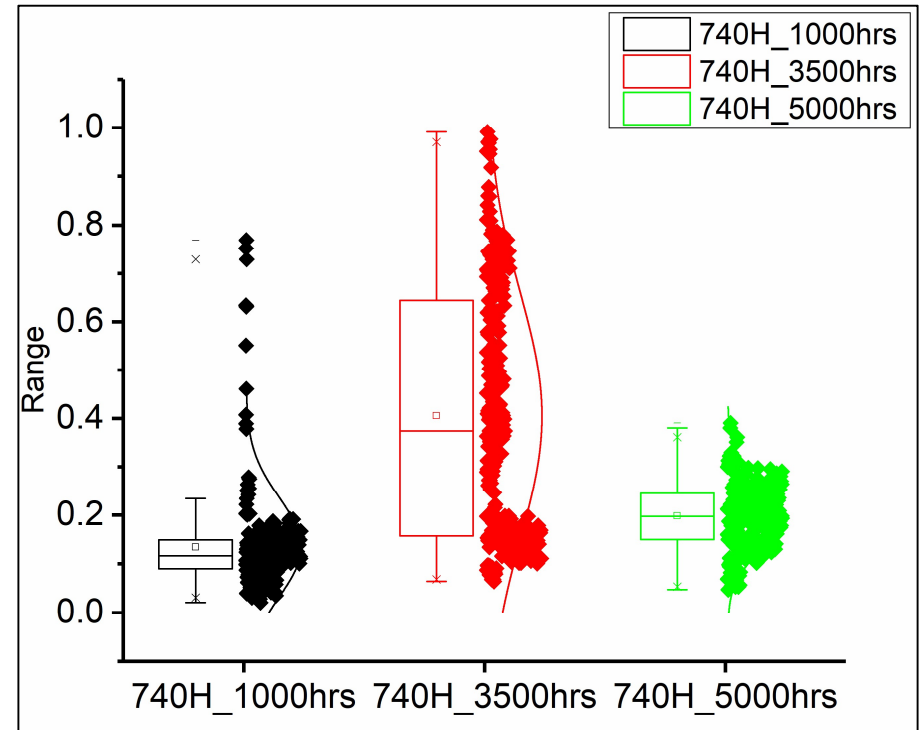
IN740 after 3.5kh at 700°C in 'impure' CO<sub>2</sub> at 200 bar (this study, 2017)

FIB-STEM offers better resolution, but it is time consuming and limited in area

# New results: IN740H, 700°C, CO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 200 bar

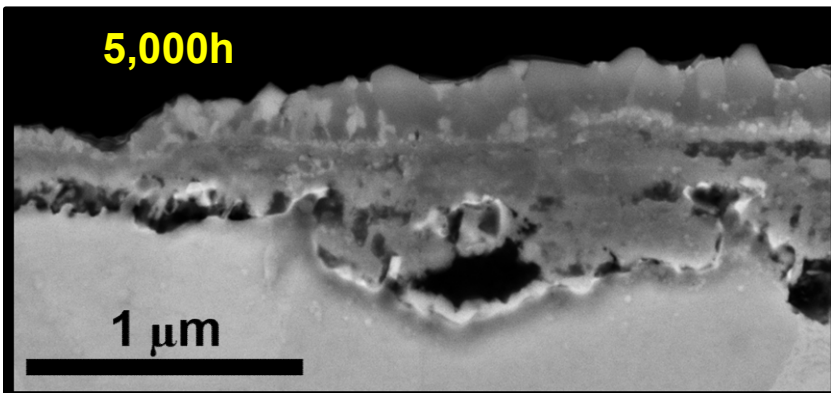
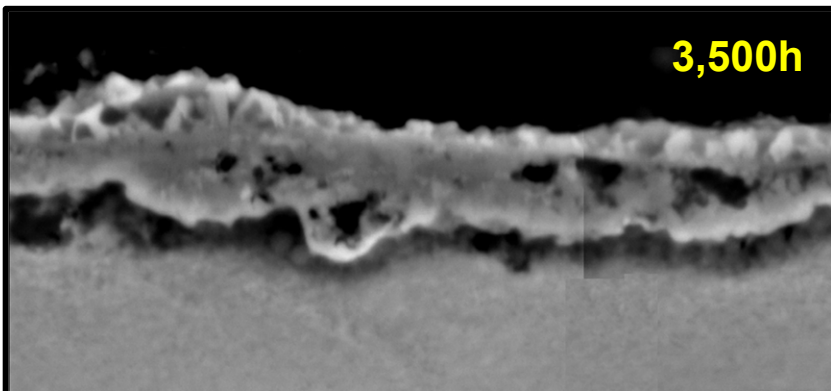
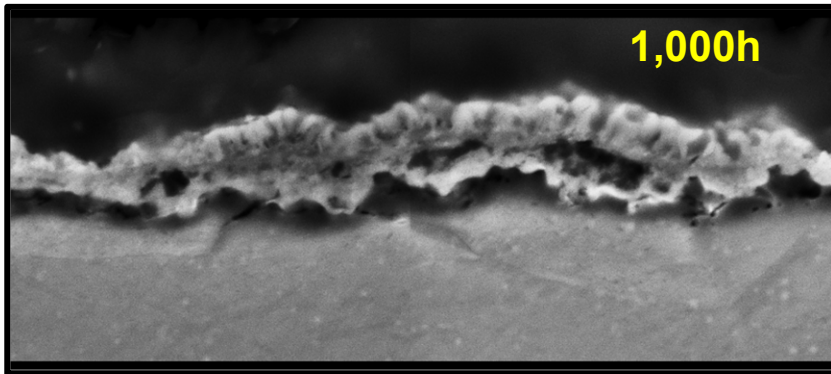


### Total Scale Thicknesses at 700°C

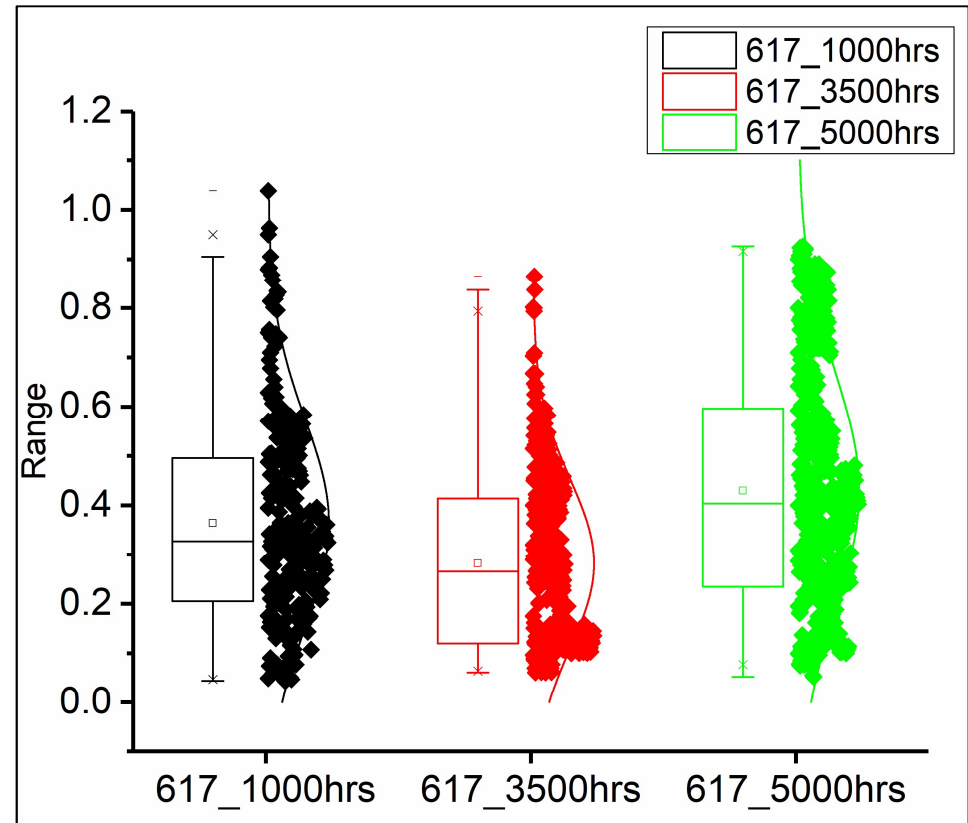


Thin, non-uniform scale clearly evident on 740H surface

# New results: IN617, 700°C in CO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 200 bar



Total Scale Thicknesses at 700°C





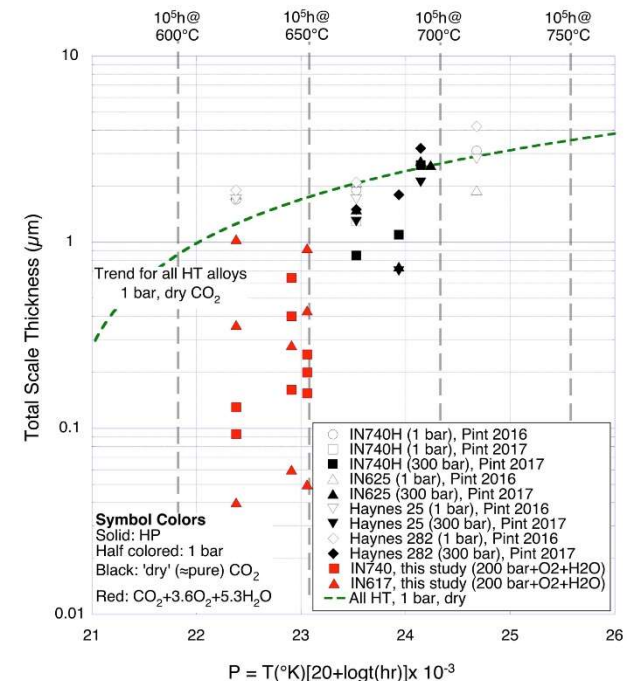
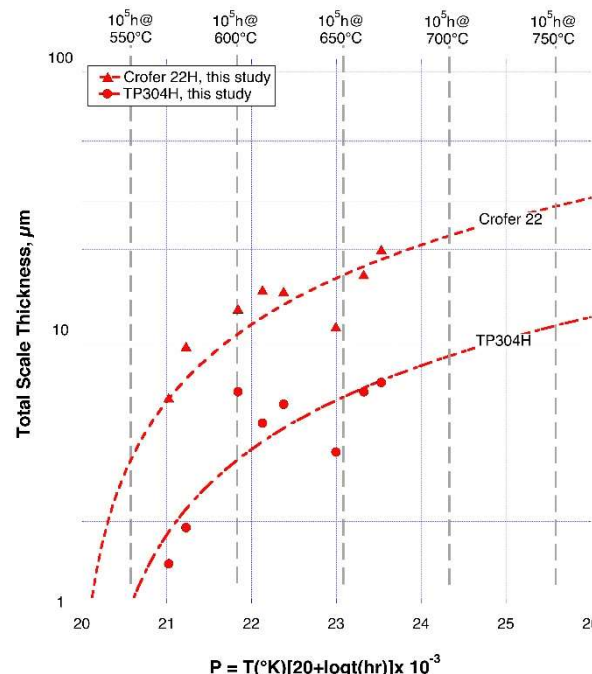
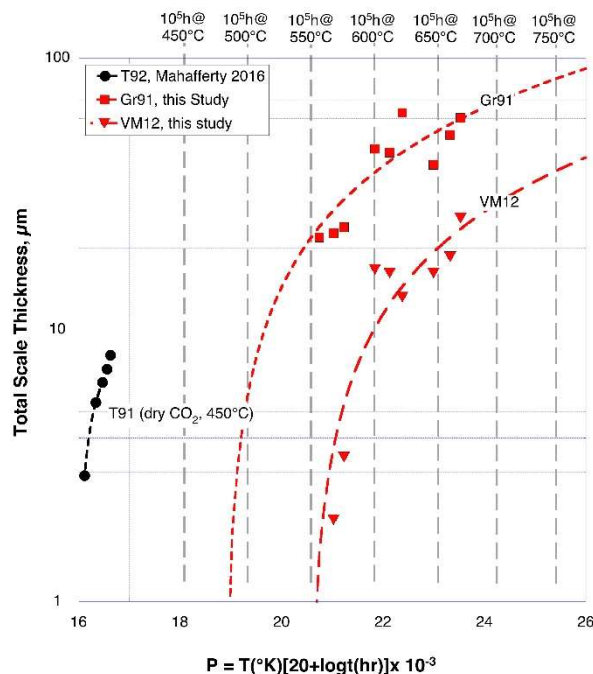
# Resulting oxidation kinetics available for use to modify EPRI Exfoliation Model (?)

## ■ Considerations:

- *caution in using 'literature' data due to apparent influence of oxidizing impurities*
- *new data appear mostly consistent, but are limited to results at 3 temperatures*

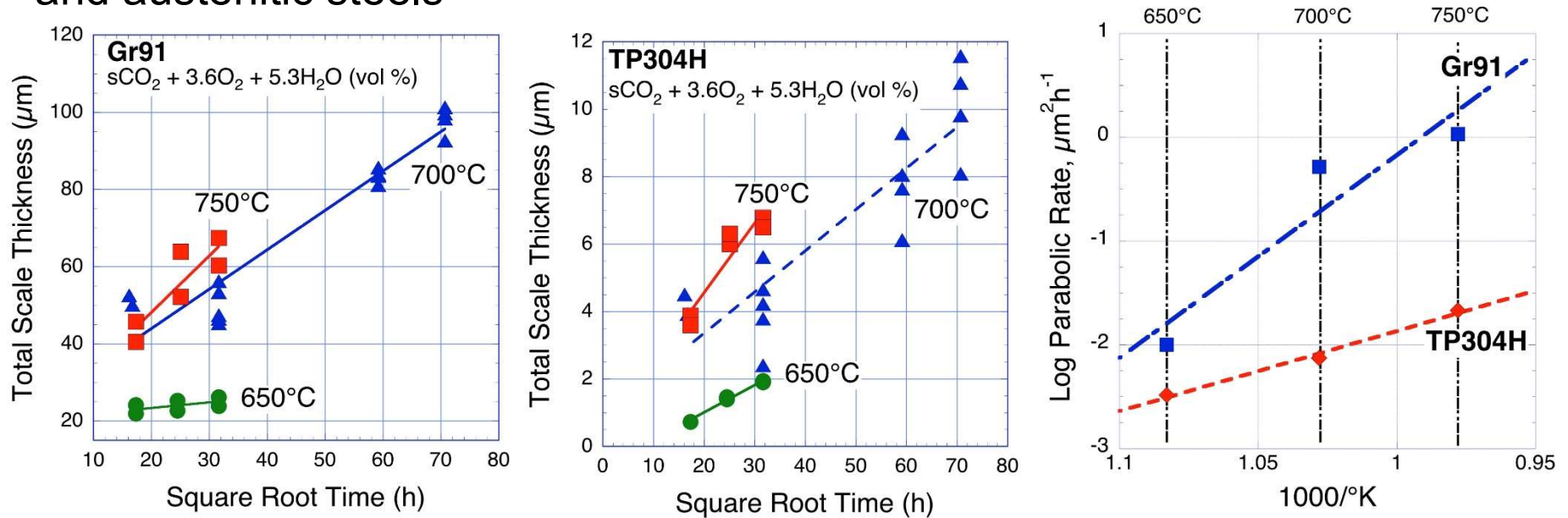
## ■ However, data for HT alloys show considerable scattering

- *augment measured thicknesses with converted mass-gain results?*



## Examples of resulting oxidation kinetics data (this study)

- Parabolic rate constants readily calculated from thickness data for ferritic and austenitic steels

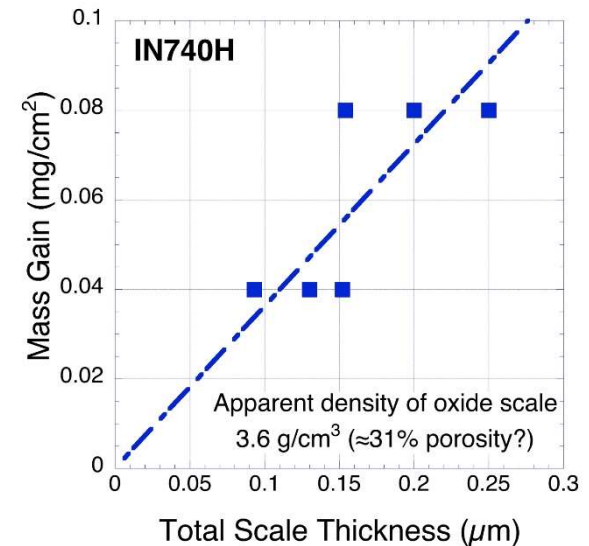
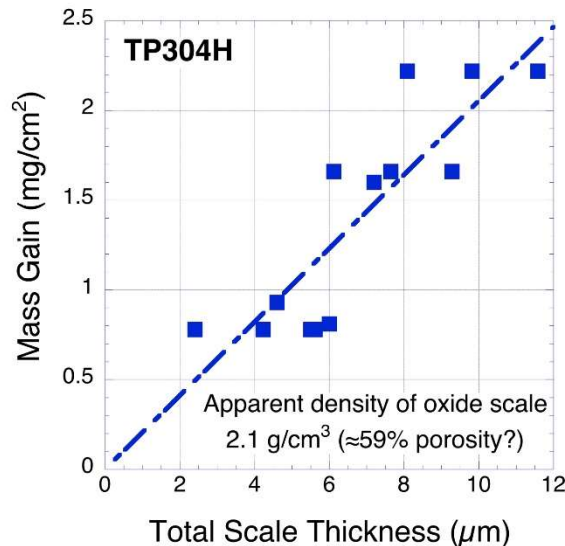
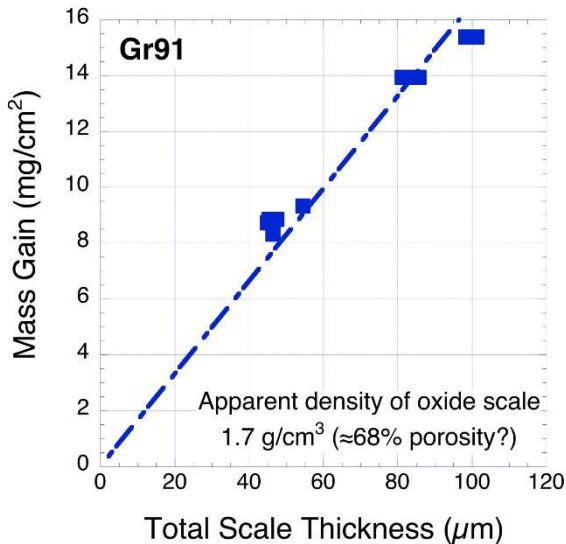


- Derived algorithms appear reasonable

Alloy	T Range ( $^\circ\text{C}$ )	A ( $\mu\text{m}^2/\text{h}$ )	Q (kJ/mole)
Gr91	650-750	$2.3 \times 10^{19}$	-374
TP304H	650-750	$7.6 \times 10^5$	-148

# Work in progress: Extraction of consistent kinetics for HT alloys

- Augmentation of thickness measurements with thicknesses derived from mass gain?



**Data for 700°C up to 5,000h. Compared with 30% porosity in T92 at 650°C in steam (Knodler & Ennis, 2001)**

- suggested densities of scales appear to be low  
e.g., porosity of scale on Gr91 ≈ 68% (vs. 30% for T92 in steam)
  - few matching scale thickness-mass gain measurements for IN740H, IN617
- Or, possible incorporation of oxidation kinetics for reliable chromia-forming alloys measured in HT steam (such as PMCr)?

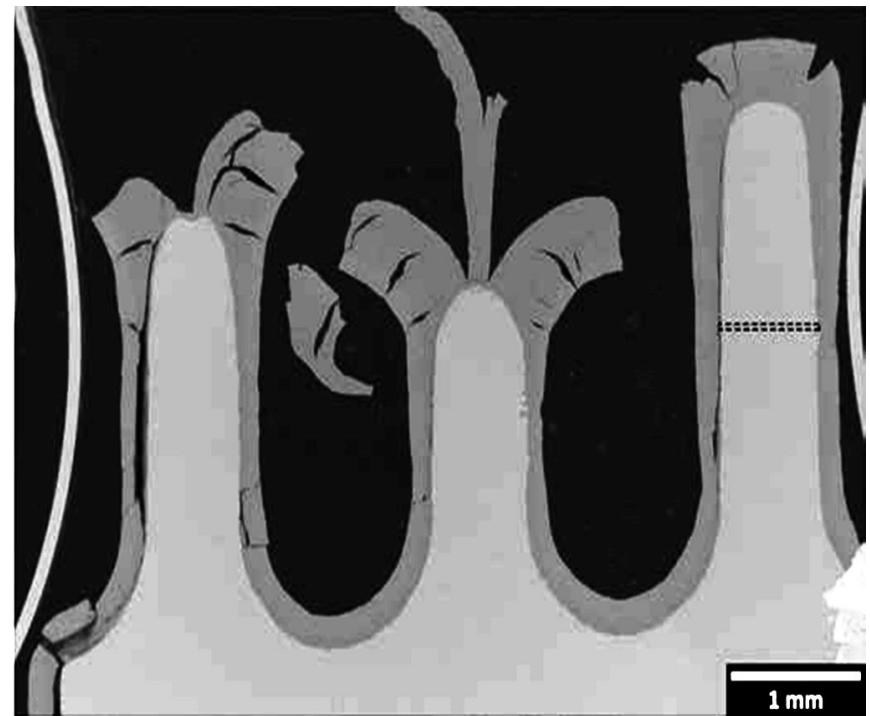
## Expected signs of C participation?

- Ferritic steels
  - *UK research (1970s) identified breakaway oxidation phenomenon under AGR conditions*
- Austenitic steels
  - *C pick-up observed only in initial stages, or when duplex scales present (or after exfoliation)*
- Chromia scales are known to be excellent barriers to C ingress
  - *confirmed by recent observations at 550-750°C, to 200 bar (Pint, ORNL)*
- High-Cr Ni-based (HT) alloys likely to be resistant to C pickup
  - *equally applicable to solid solution and precipitation-strengthened Ni-based candidate alloys?*



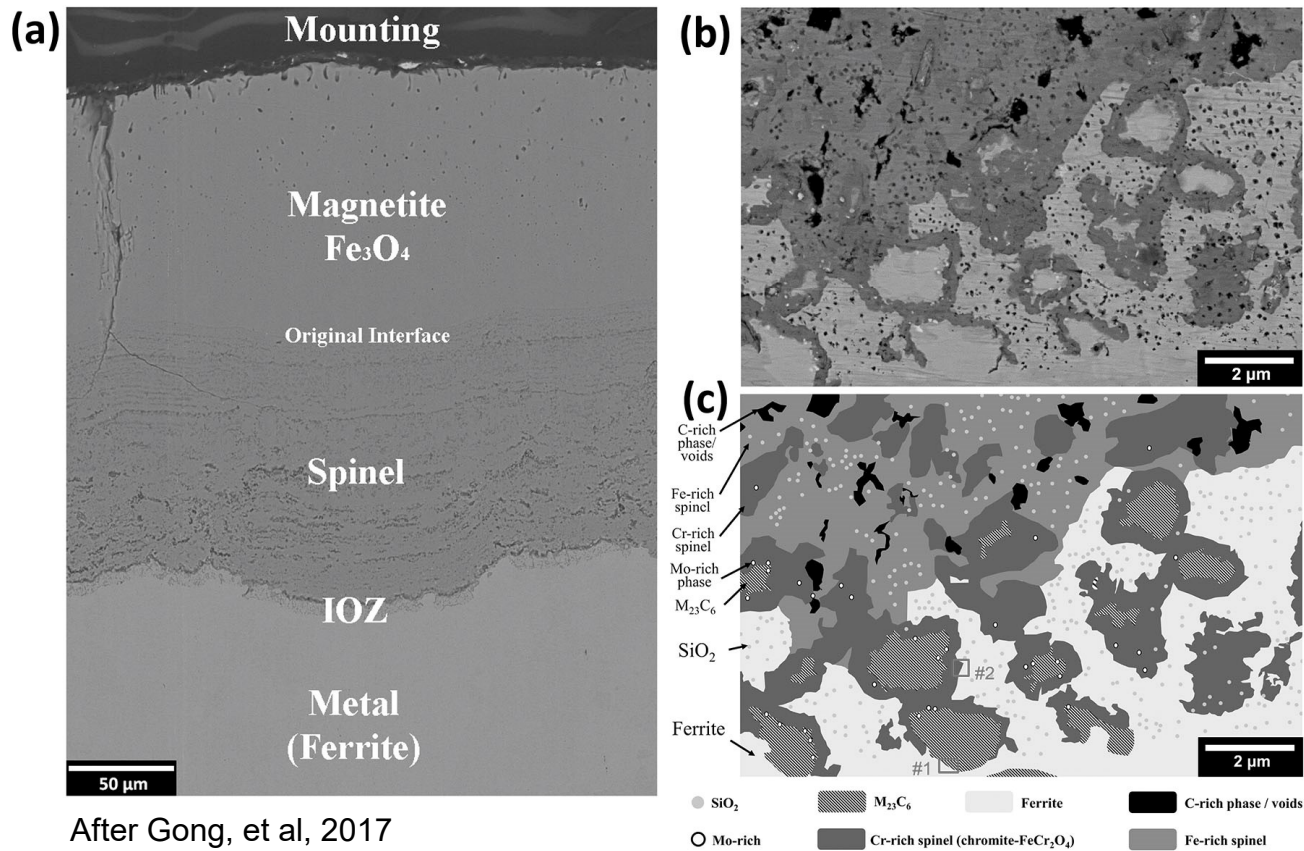
## Observed effects of C on oxidation of 9Cr ferritic steel

- AGR testing showed accelerated oxidation of 9Cr steels in (impure) CO<sub>2</sub>:
  - CO<sub>2</sub> containing 1 vol% CO-300 ppmv H<sub>2</sub>O-300 ppmv CH<sub>4</sub>-100 ppmv H<sub>2</sub> at 40 bar
  - breakaway oxidation > 580°C, ≥20 kh
  - C saturation model for breakaway criteria
- However, not much information on scale morphologies
  - measured alloy C volume fraction vs. *t*, *T*



9Cr -1Mo finned SH tube, 20kh at 600°C in 42 bar CO<sub>2</sub>  
(Gong et al, 2017)

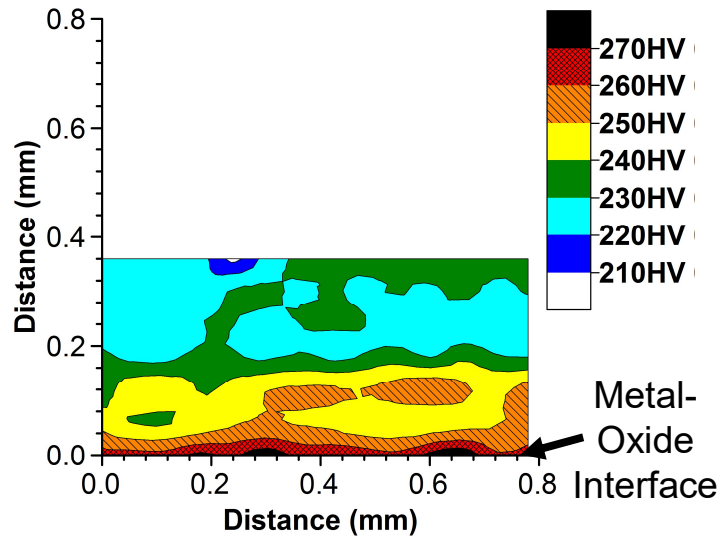
## Fe-9Cr-1Mo at transition to breakaway (RH fin in previous slide)



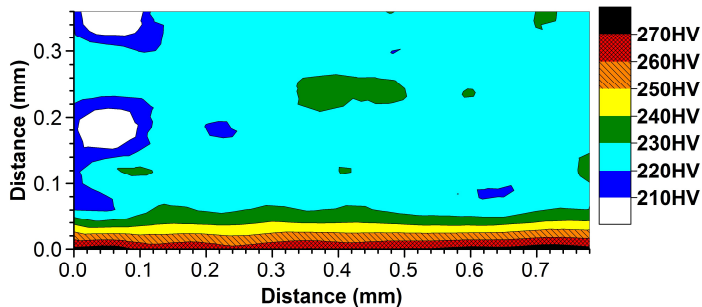
After Gong, et al, 2017

- Morphology of main scale indistinguishable from those in steam
- IOZ shows fine, distributed carbides
- Alloy substrate was saturated with C: large, blocky  $\text{M}_{23}\text{C}_6$ , and  $\text{M}_2(\text{C},\text{N})$  needles

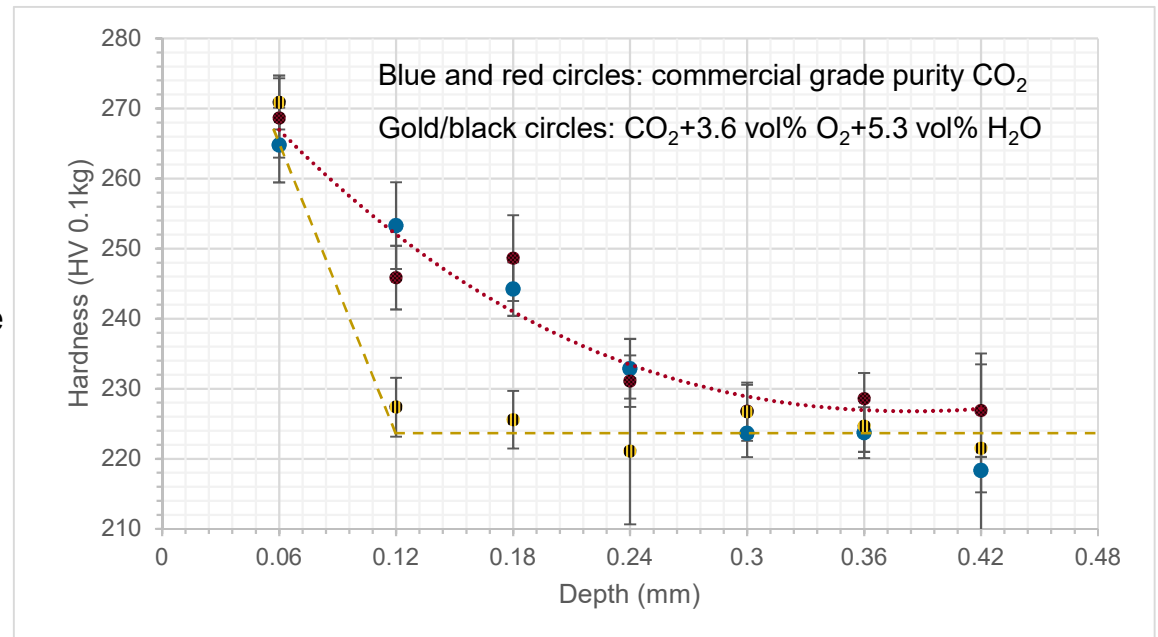
# This study: Gr91 had greater depth of hardening in CG purity than $\text{CO}_2 + 3.6\% \text{O}_2, 5.3\% \text{H}_2\text{O}$ at 200 bar after 300h at $700^\circ\text{C}$



Commercial-grade purity  $\text{CO}_2$



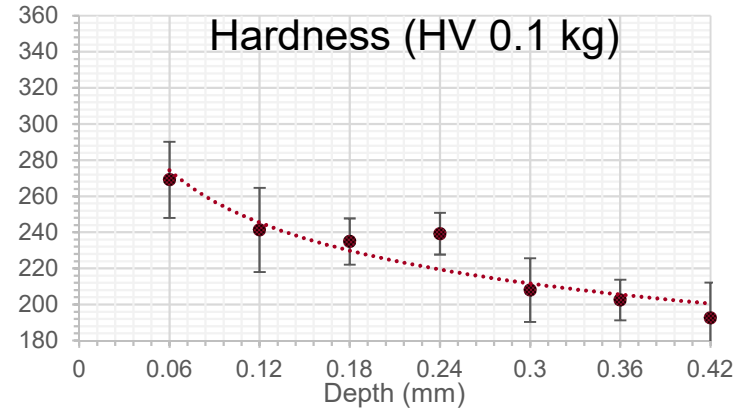
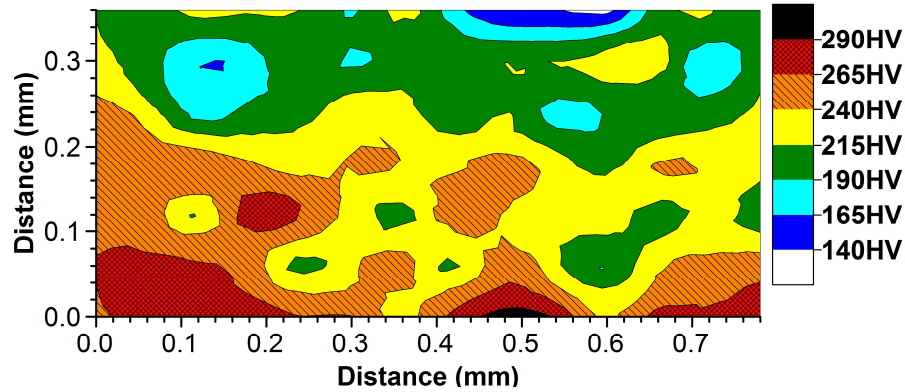
$\text{CO}_2 + 3.6 \text{ vol}\% \text{O}_2 + 5.3 \text{ vol}\% \text{H}_2\text{O}$



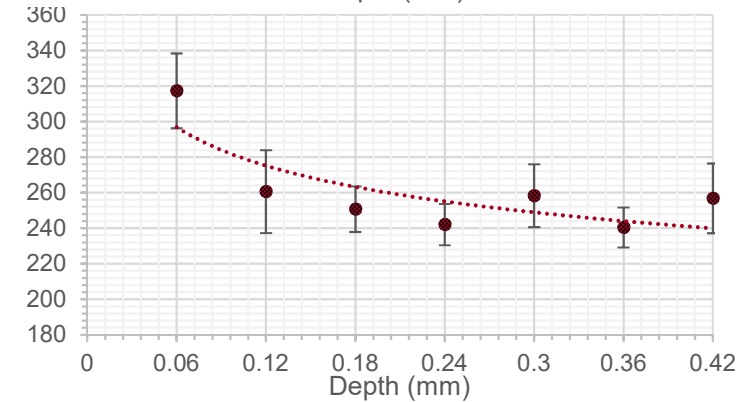
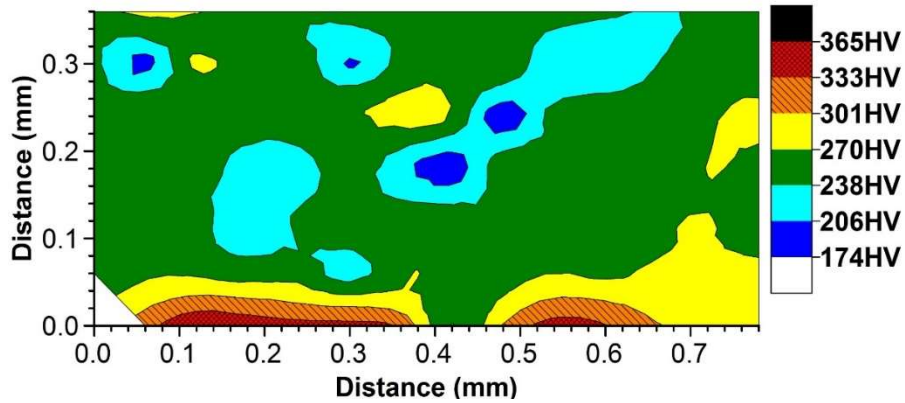
- Hardness maps from arrays of 7 x 14 indentations
- Alloy-oxide interface at  $y = 0.0$  (bottom of map)
- Oxidizing impurities appear to retard hardening

# Ferritic Steels: 1,000h at 700°C in CO<sub>2</sub> +3.6% O<sub>2</sub>, 5.3% H<sub>2</sub>O at 200 bar

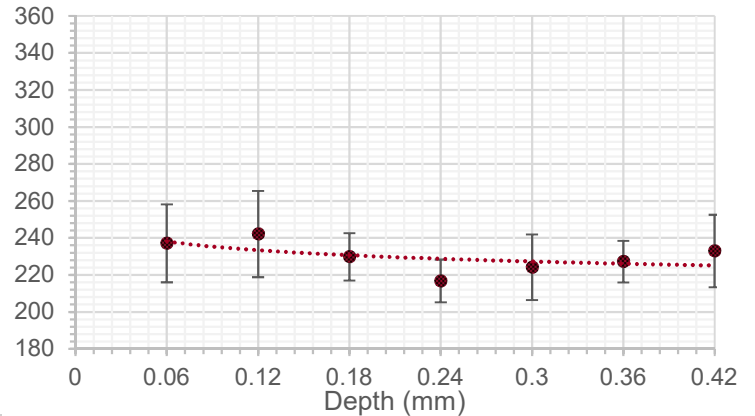
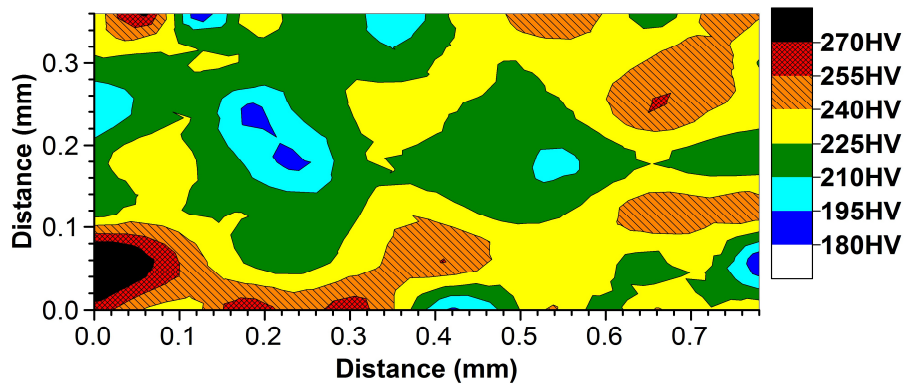
Gr91



VM12



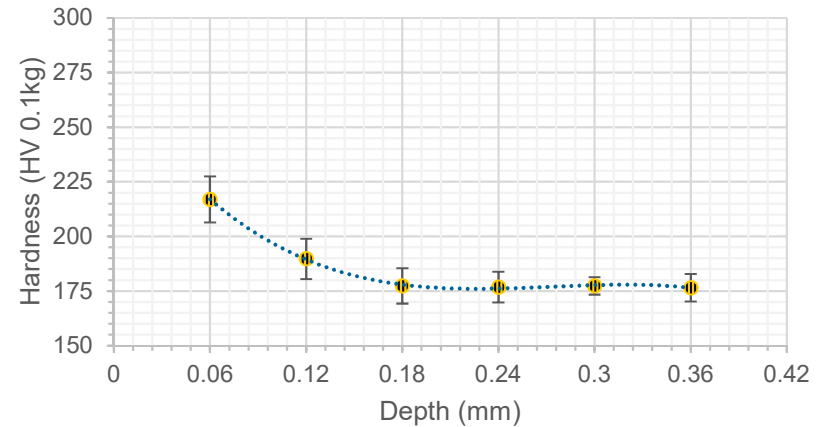
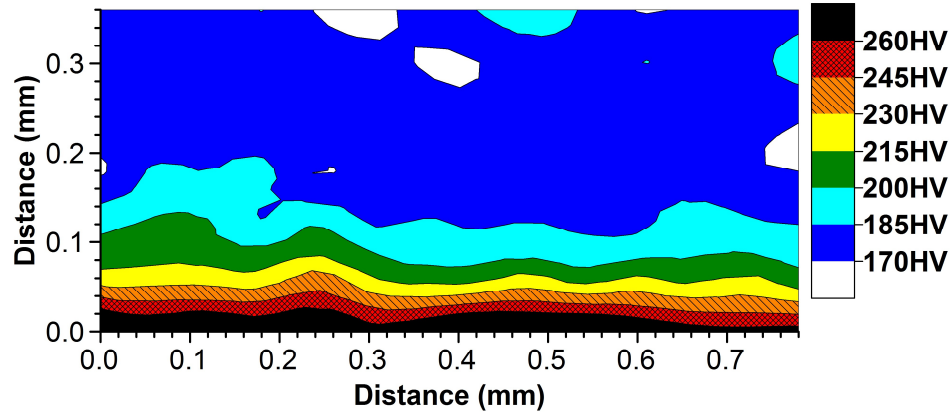
Crofer 22H



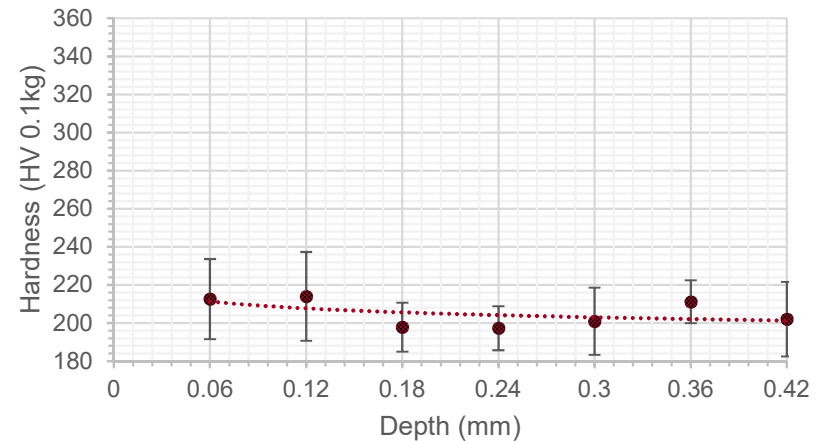
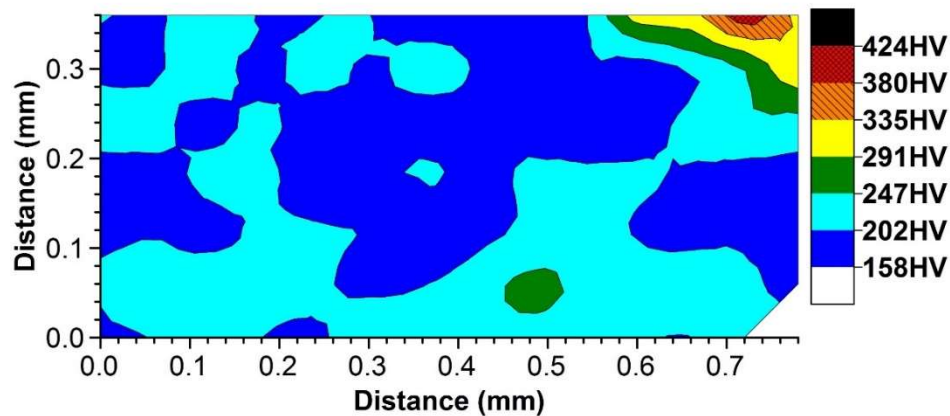


# Anomalous Hardness Changes in TP304H after exposure at 700°C to CO<sub>2</sub> containing 3.6 vol% O<sub>2</sub>, 5.3 vol% H<sub>2</sub>O at 200 bar?

- 300h: significant but highly-localized near-surface hardening

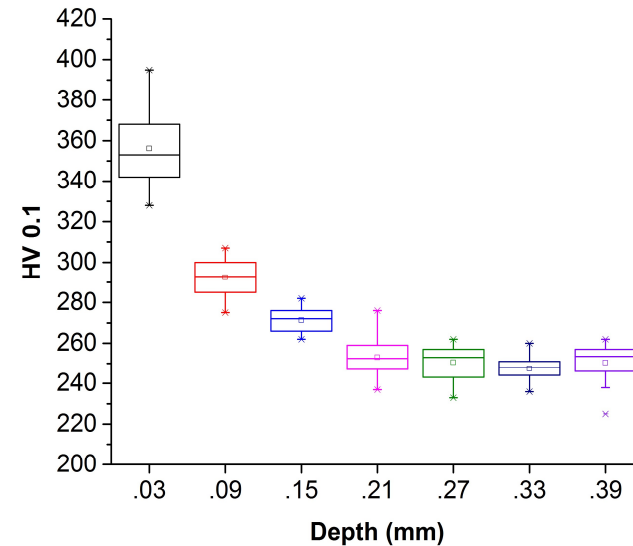
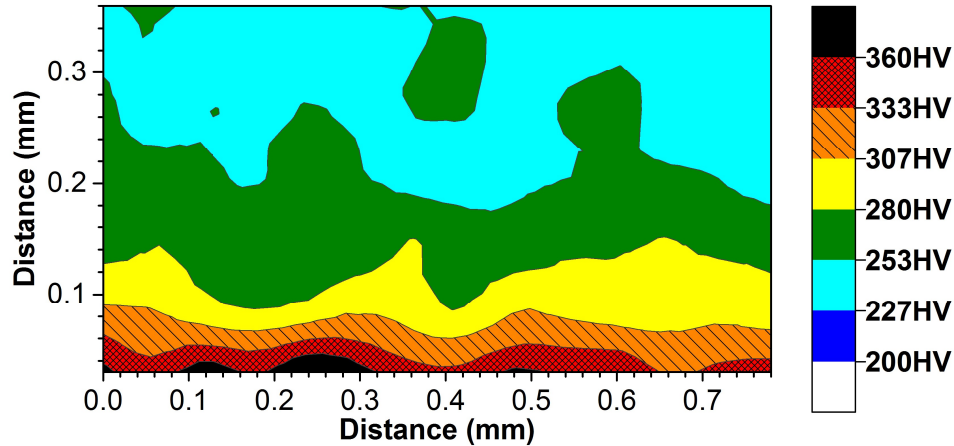


- 1,000h: no apparent increase in hardness



# Hardness changes in HT alloys after exposure for 1,000h at 700°C to CO<sub>2</sub> containing 3.6 vol% O<sub>2</sub>, 5.3 vol% H<sub>2</sub>O at 200 bar

## HR3C



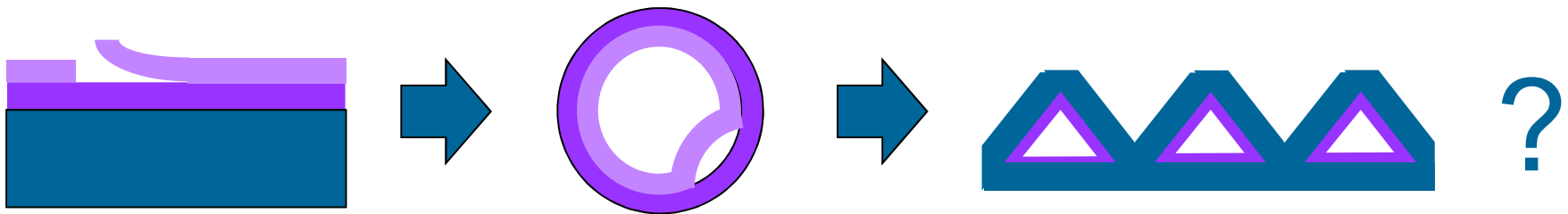
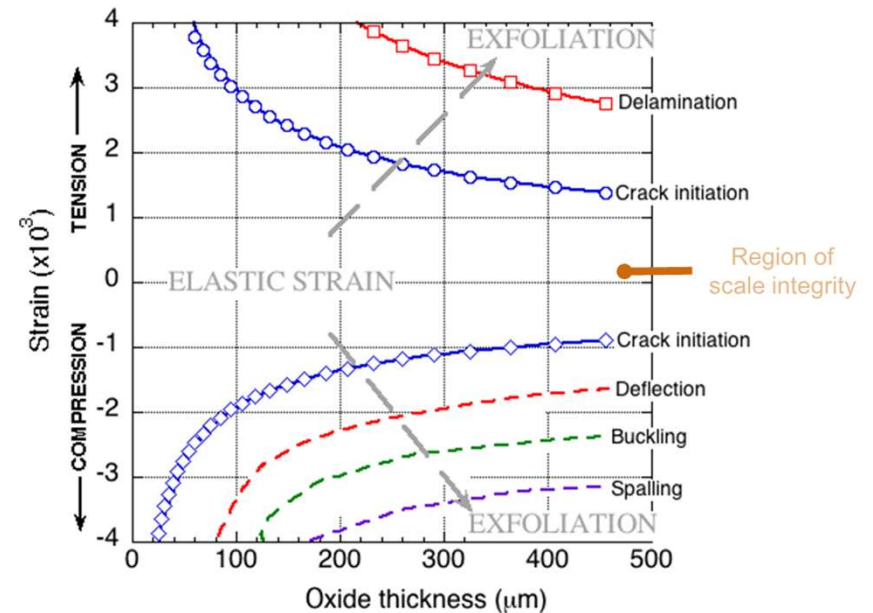
- Significant hardening to a relatively uniform depth of 150-200  $\mu\text{m}$
- Surprising, since HR3C is expected to form a protective chromia scale
- Could be associated with cold work from surface preparation ?

## IN617 and IN740H

- Micro-hardness measurements did not reveal any evidence of carburization.

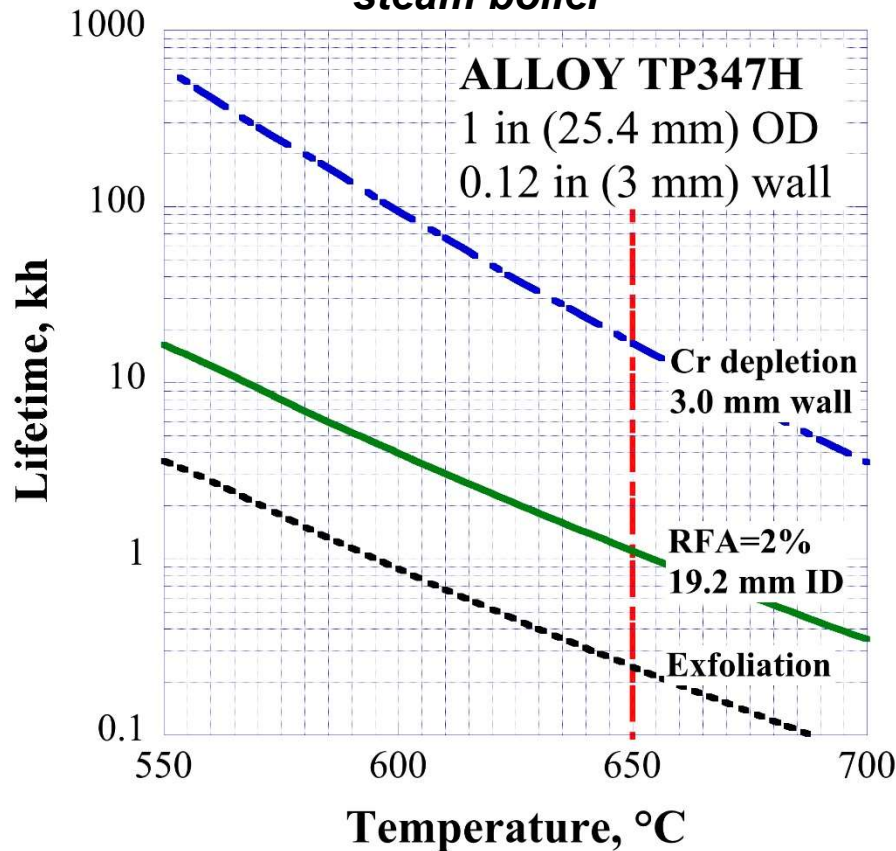
# Exfoliation Model to address unique influences of oxide growth on service life of small-channel heat-exchangers

- Lab studies: isothermal oxide growth on small flat coupons
- Real world: heat-flux, stress from complex geometries
- Modeling:
  - EPRI-developed strain trajectory approach for steam tubes
  - Properties of sCO<sub>2</sub> and alloys collected
  - **Discussion with vendors on convex vs. concave surfaces – need to develop a generic modeling approach**



# Proposed Overall Summary of Results

**Example for SH tube in conventional steam boiler**



- Actual examples that illustrate lifetime issues related to oxide growth
- Use critical dimensions from sCO<sub>2</sub> HX designs; all seven alloys
- Lifetime criteria used:
  - 1)  $\Delta P$  due to oxide thickening
  - 2)  $t$  to breakaway due to Cr consumption
  - 3)  $t$  to reach critical scale thickness for exfoliation
- Possible further criterion:
  - 4)  $t$  for wall thickness to be thinned to unsafe level (strength criteria?)



# Summary

- First project to address oxidation in semi-open Allam cycle sCO<sub>2</sub> conditions
  - *impurity concentrations determined via mass balance and thermodynamic calculations*
  - *a new test rig was assembled and laboratory tests to 5,000h with and without impurities completed*
- Oxidation rates in sCO<sub>2</sub>
  - *appear consistent, with similarities to those in HP steam*
  - *possibly slower when oxidizing impurities are present in sCO<sub>2</sub>*
  - *no systematic effect of total pressure*
- Scale morphologies
  - *nominally followed expectations from steam, with some potential influences of C*
- Surface hardening
  - *identified in Gr 91 and VM12; more severe in 'pure' sCO<sub>2</sub>*
  - *also in TP304H & HR3C, although behavior of TP304H appeared anomalous*
  - *none found in Ni-base alloys*
- Evaluation of impact of scale growth in sCO<sub>2</sub> on service lifetimes for compact heat exchanger designs in progress - built on existing EPRI Oxide Exfoliation Model

## Acknowledgements / Team

- This work is sponsored by the DOE Office of Fossil Energy Cross-Cutting Materials Research Program: ***DE-FE0024120***
- DOE/NETL
  - Vito Cedro, Project Manager
- EPRI
  - John Shingledecker (Principal Investigator)
  - David Thimsen
  - Steve Kung
- DNV-GL
  - Brett Tossey
- Oak Ridge National Laboratory
  - Adrian Sabau
- WrightHT, Inc.
  - Ian Wright



# Together...Shaping the Future of Electricity