

Advanced Alloy Design Concepts for High-Temperature Fossil Energy Applications

PI: Yukinori Yamamoto

Oak Ridge National Laboratory, Oak Ridge, TN 37831

E-mail: yamamotoy@ornl.gov

Co-investigators:

Bruce A. Pint

Oak Ridge National Laboratory, Oak Ridge, TN

Sudarsanam Suresh Babu, Benjamin Shassere*, Chih-Hsiang (Sean) Kuo

University of Tennessee, Knoxville, TN

(*Currently at Oak Ridge National Laboratory)

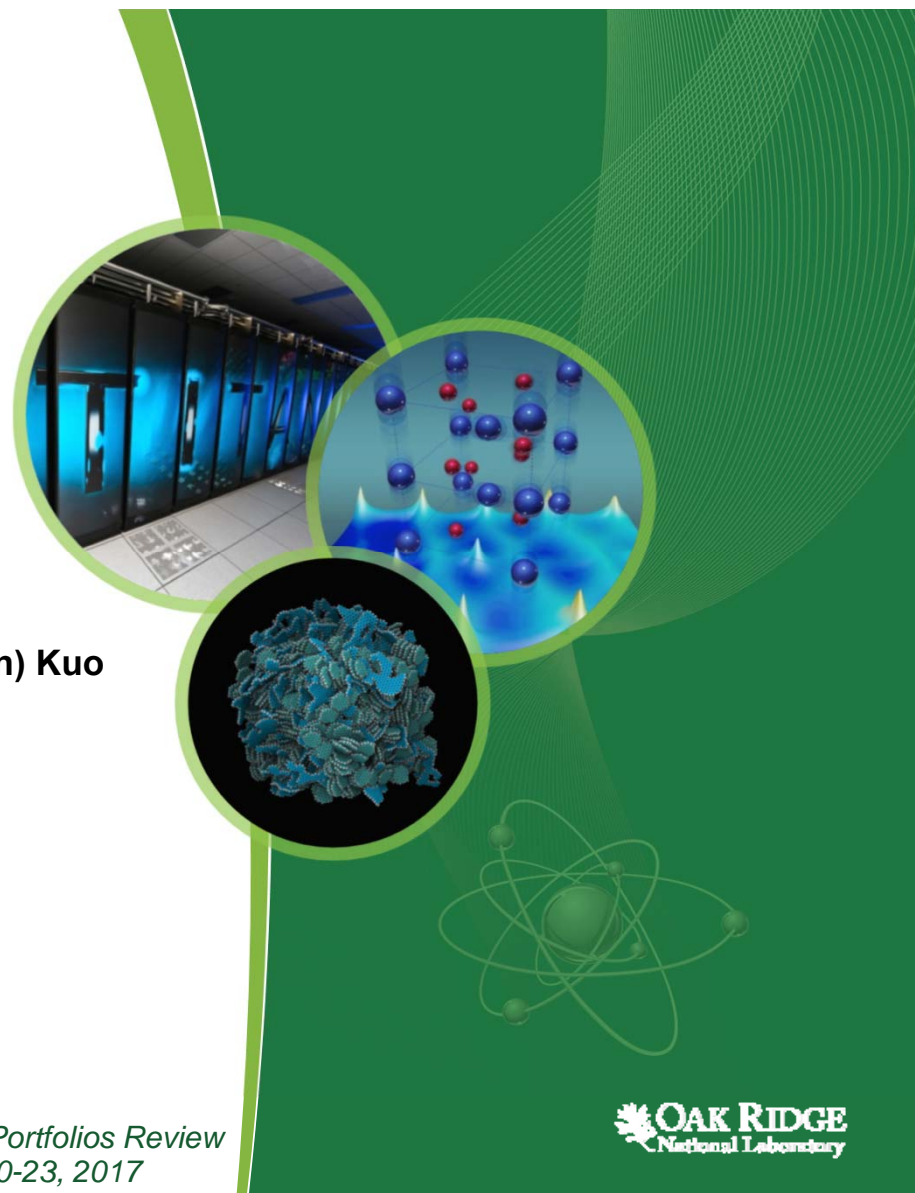
DOE Award Number: FEAA114

Period of Performance: Oct. 2013 - Sep.2018

Presentation: March 22nd, 2017 (in *Computational Materials*)

ORNL is managed by UT-Battelle
for the US Department of Energy

2017 Crosscutting Research Portfolios Review
Pittsburgh, PA, March 20-23, 2017



Acknowledgements

Program management:

Vito Cedro (NETL), Pete Tortorelli, Hiram Rogers (ORNL)

Scientific advice and support:

Mike Brady, Muralidharan Govindarajan, Roger Miller, Zhiqian Sun, Donovan Leonard (ORNL), Bernd Kuhn (Forschungszentrum Jülich GmbH, Germany), Kazuhiro Kimura (National Institute for Materials Science, Japan)

Technical support:

Cecil Carmichael, Dave Harper, Greg Cox, Dustin Heidel, Kevin Hanson, Daniel Moore, Tom Geer, Victoria Cox, Eric Manneschmidt, Jeremy Moser, Mike Stephens, George Garner, Doug Kyle, Brian Sparks (ORNL)

**Research sponsored by the Crosscutting Research Program, Office of Fossil Energy,
U.S. Department of Energy**



U.S. DEPARTMENT OF
ENERGY



Project Goals and Objectives

Goals: To identify and apply breakthrough alloy design concepts and strategies for incorporating improved creep strength, environmental compatibility/resistance, and weldability into three classes of alloys (ferritic, austenitic, and Ni-base) intended for use as heat exchanger tubes in fossil-fueled power generation systems at higher temperatures than possible with currently available alloys

Objectives: To develop and propose new creep-resistant, “alumina-forming”, cost-effective structural materials with guidance of computational thermodynamic tools

- Milestones (FY2017):

1. *Complete intermediate-term oxidation and creep testing of scale-up heats of “model alloy” (Met)*
2. *Submit a journal paper summarizing the new FeCrAl alloy design study (In Progress)*
3. *Initiate computational screening of next generation advanced alumina-forming austenitic and Ni-base alloys for improved strength and corrosion resistance (In Progress)*

Presentation Outline

- Backgrounds/Motivation:
 - *Concepts of “Advanced Alloy Design”*
 - *High-Cr containing FeCrAl “ferritic” alloy development*
- Update on FY16/17
 - *Effect of alloying on Laves-phase solvus and stability*
 - *Creep property evaluation*
 - *Oxidation/ash-corrosion tests*
 - *Scale-up efforts*
- Summary
- Future Activities

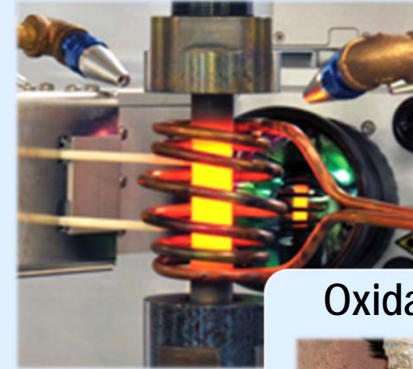
Backgrounds: Advanced Alloy Design

Strong demands on new high-temperature materials with improved properties for upper limit temperatures higher than that of the materials currently in service;

- *High-temperature strength (tensile, creep)*
- *Environmental compatibilities (oxidation, corrosion)*
- *Minimized weld-related issue for final components*
- *RT toughness (Charpy)*
- *Processibility (including microstructure control)*
- *Inexpensive materials cost (raw elemental costs, process routes, etc.)*

➔ **High-Cr containing FeCrAl ferritic alloy
(Fe-30Cr-3Al base, wt/%)**

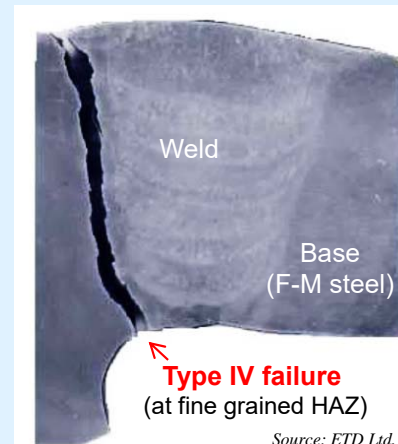
High-temp. Strength



Oxidation/Corrosion



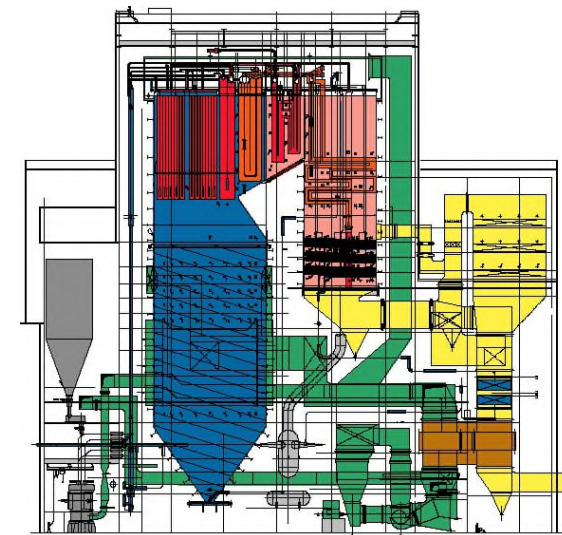
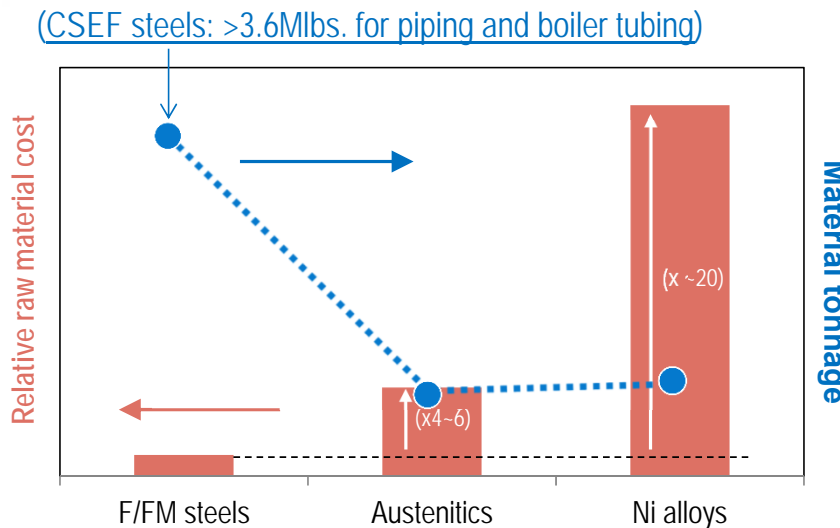
Weldments



Target Materials/ Applications

Three different grades of structural materials that are currently available for use by the US electric utility industry:

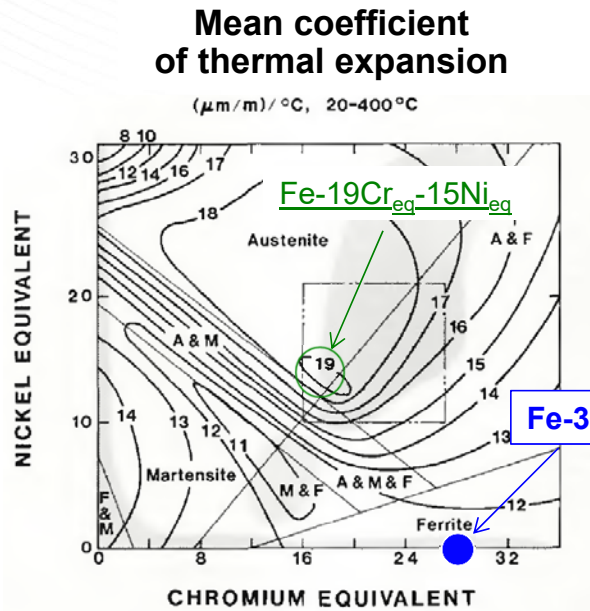
- 1) **Ferritic steels** for temperatures up to 600 °C, with **ferritic-martensitic** versions (F-M steels) having increased strength up to 600-620 °C;
- 2) **Austenitic stainless steels** with strength and environmental resistance up to 650 °C; and
- 3) **Ni-base alloys** for temperatures > 700 °C.



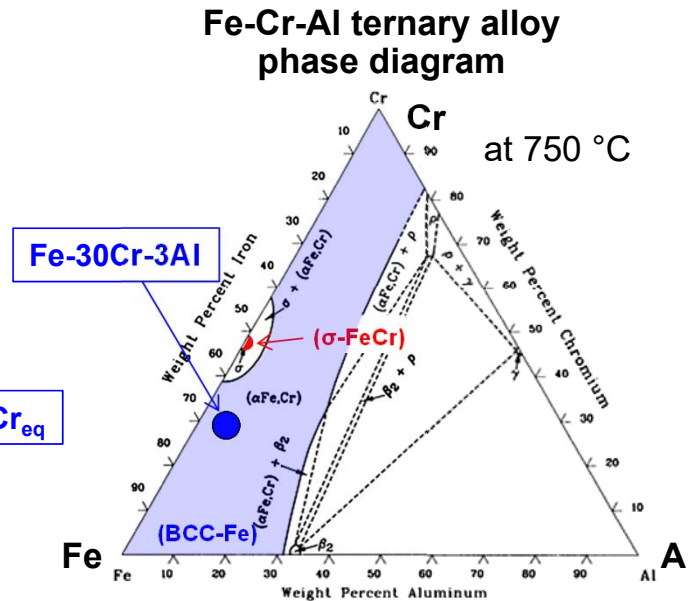
Alstom USC and AUSC Power Plants – J. Marion - NTPC/USAID Int. Conf. SC Plants - New Delhi, India, 22 Nov. 2013 – P 8

Why High Cr Containing FeCrAl Ferritic Alloy?

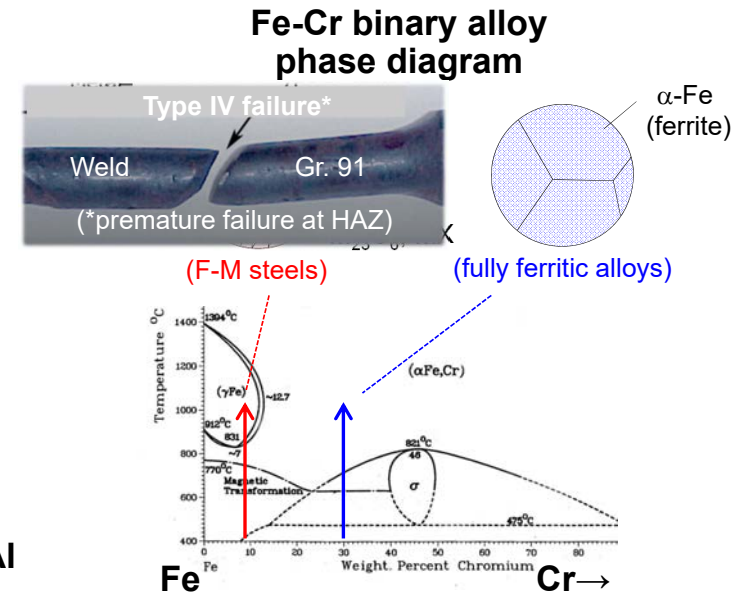
- High surface protection in both steam and ash-corrosion environments
- Less thermal expansion and high thermal conductivity
- Suppress brittle σ -FeCr formation by Al additions
- No Type IV failure because of no BCC-FCC transformation during welding



Ref. Welding Research Supplement (1982)



Ref. ASM international, Ternary alloy phase diagrams



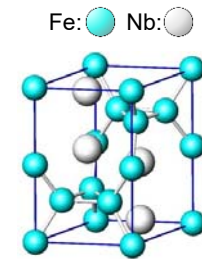
Ref. ASM international, Binary alloy phase diagrams

OAK RIDGE

What is the Challenge in High Cr Containing FeCrAl Ferritic Alloy?

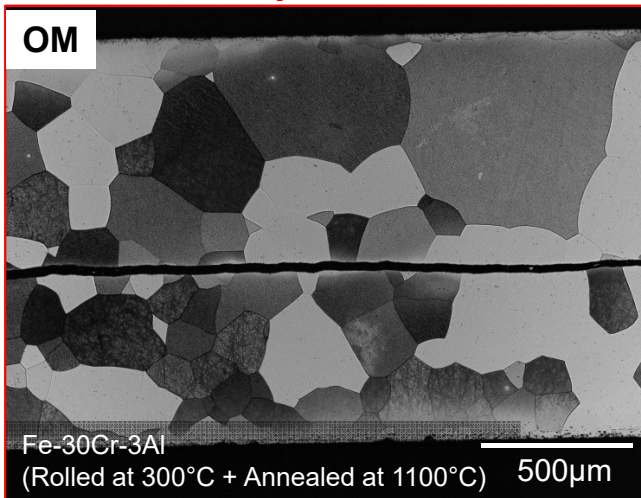
- Poor high-temperature strength due to BCC matrix
 - Rapid grain coarsening causes poor ductility at low temperature
- **Use alloying additions for GS stability and strengthening (e.g. Nb for Laves-phase formation)**

C14: Laves-Fe₂Nb

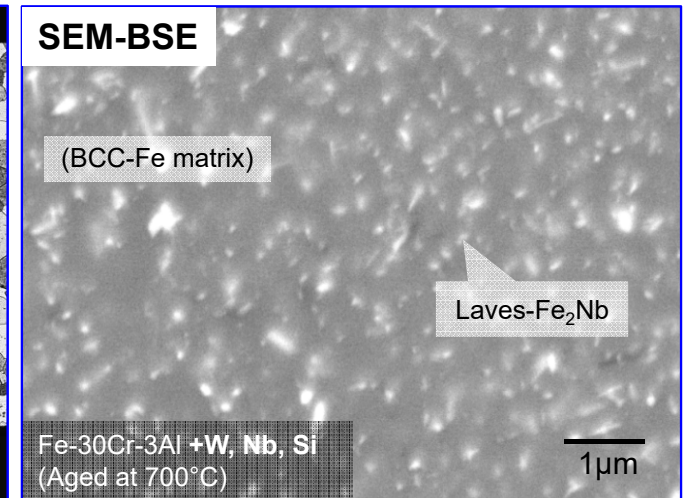
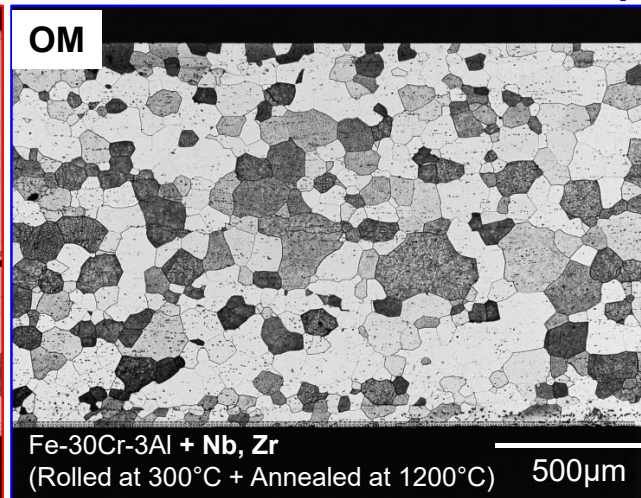


http://www.geocities.jp/ohba_lab_ob_page/structure5.html

No optimization



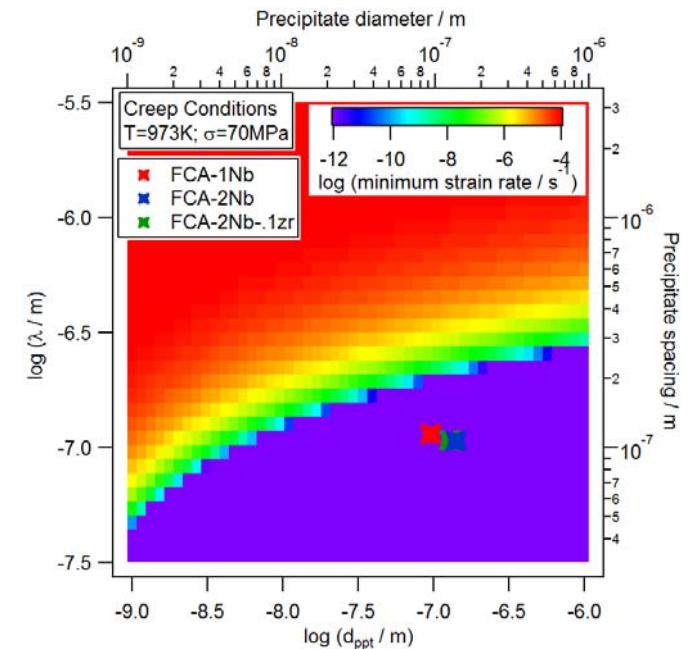
After optimization



Alloy Development in Progress

- **1st series: Fe-30Cr-3Al with (0~2)Nb-(0~0.3)Zr (in wt.%)**
 - Optimization of thermo-mechanical treatment
 - Tensile test (RT~800°C)
 - Oxidation test (800°C, in air + 10%H₂O)
 - Ash corrosion test (700°C, in ash + corrosive gas)
 - Creep-rupture test (650~700°C, 70~100MPa)
- **2nd series: Fe-30Cr-3Al with (0.5~2)Nb + W, Mo, Ti, and Y**
 - BCC solvus temperature control
 - Size stability of Lave-phase particle at 700°C
 - Oxidation/Ash-corrosion resistance
 - Property evaluation
 - Scale-up effort

Calc. minimum creep-rate map



*Used Bird-Mukherjee-Dorn (BMD) model

$$\frac{\dot{\epsilon}_m kT}{DEb} = A_{Dis} \left(\frac{\sigma_a - \sigma_{th}}{E} \right)^n$$

$$\sigma_{th} = \frac{Eb}{2\pi\lambda} \ln \frac{d_{ppt}}{b}$$

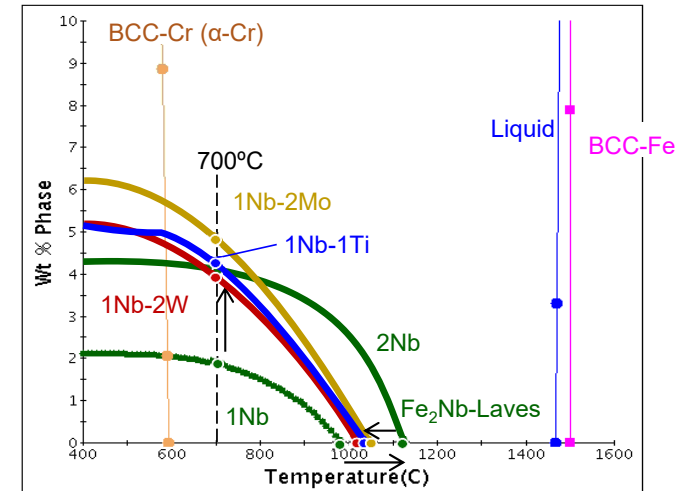
Major Concerns in FY17

- Effect of third element addition on microstructural stability, mechanical properties, and oxidation/ash-corrosion resistance:
 - *Nb, W, Mo, and Ti* (Laves-phase forming elements)
 - *Mn, Si, C* (simulate engineering alloys with the level of industrial impurities)
- Scale-up efforts of high-Cr FeCrAl alloy

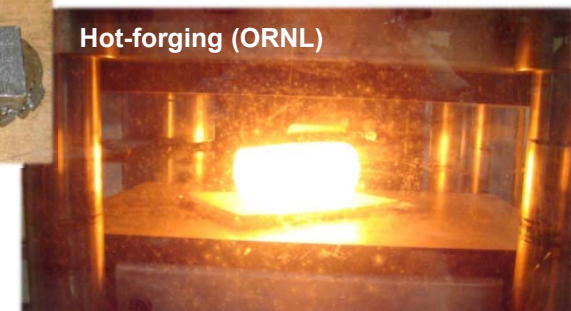
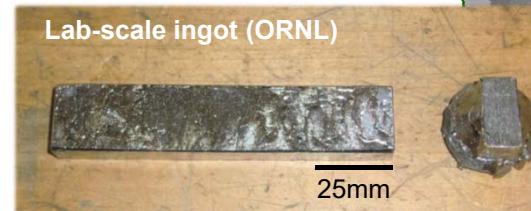
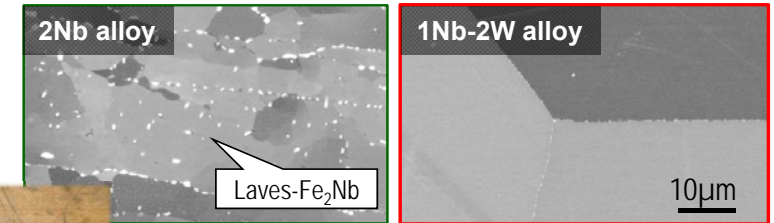
Experimental Procedure

- **Alloys to be evaluated:**
 - *Fe-30Cr-3Al base with (or without) Nb, Si, W, Mo, Ti, Y, Mn, and C, wt.%*
 - *Used thermodynamic computational tool (JMatPro, v.8-9) for downselection*
- **Lab-scale heat preparation:**
 - *Arc-melt and drop-cast the bar ingots (13 x 25 x 125 mm)*
- **Thermo-mechanical treatment:**
 - *Homogenization, forging, rolling, annealing*
 - *Water-quenching*
- **Microstructure analysis:**
 - *OM, SEM*
- **Property evaluations:**
 - *Creep, oxidation, ash-corrosion*

Fe-30Cr-3Al-0.2Si base alloys



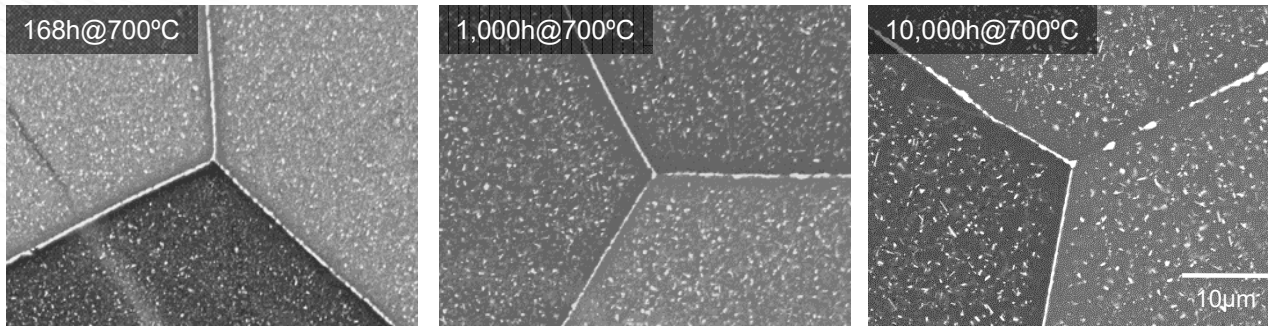
Annealed at 1200°C (SEM-BSE)



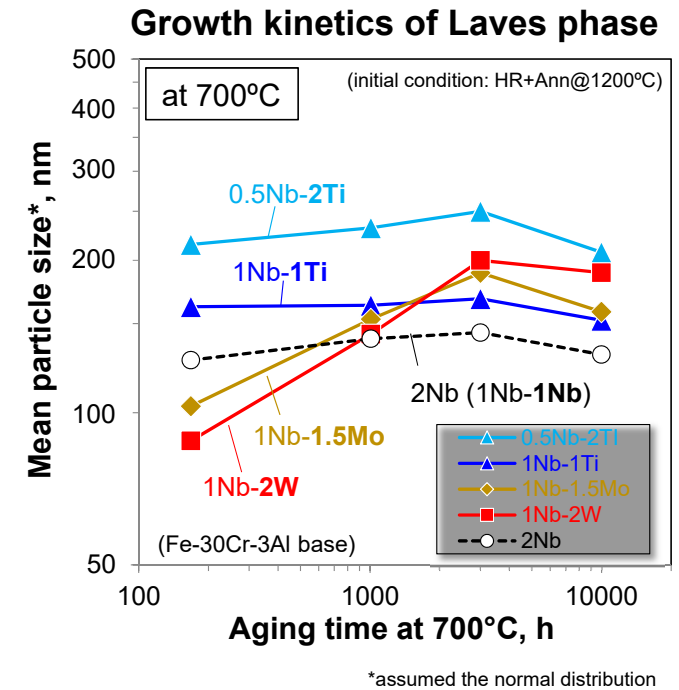
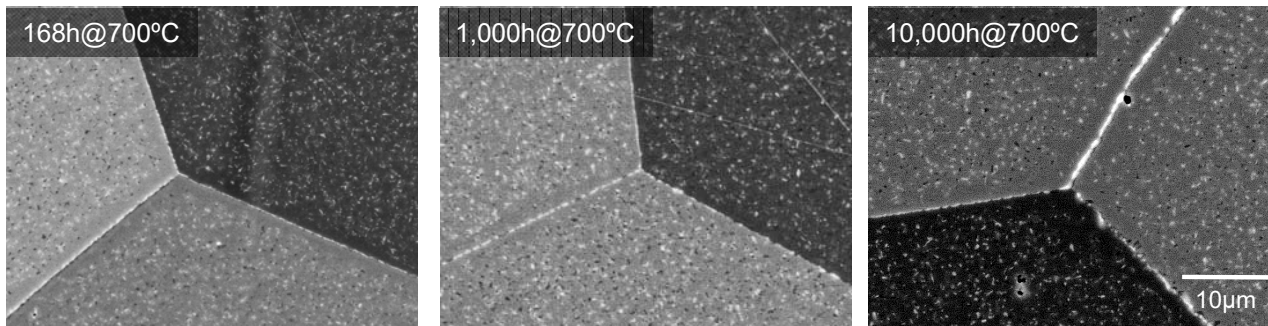
Impact of “Third Additional Elements” on Microstructure

- Fine particle size maintained even after 10,000h at 700°C

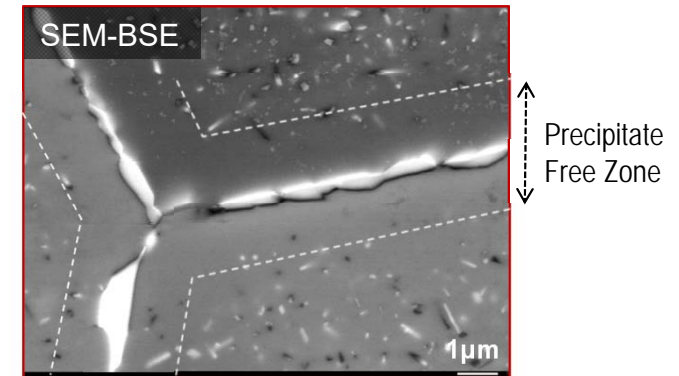
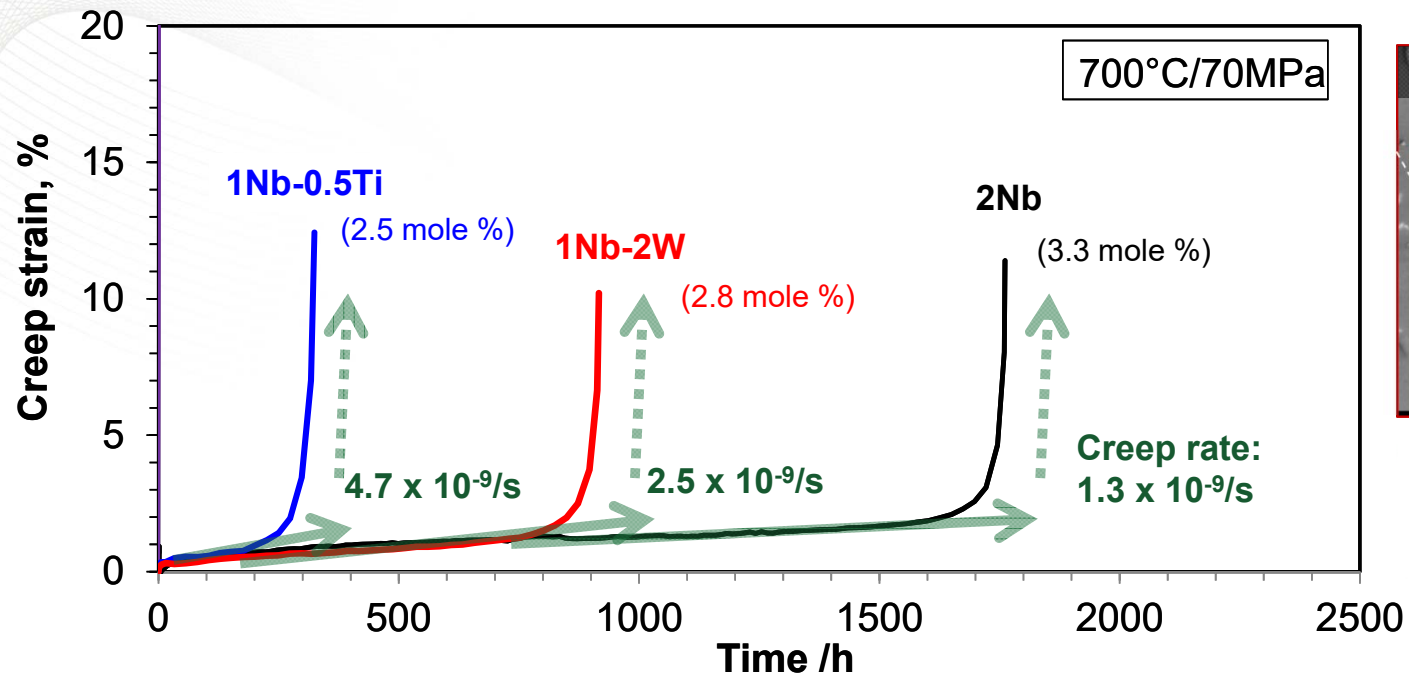
Fe-30Cr-3Al-1Nb-2W



Fe-30Cr-3Al-1Nb-1Ti



Third Element Addition Was Effective As Expected

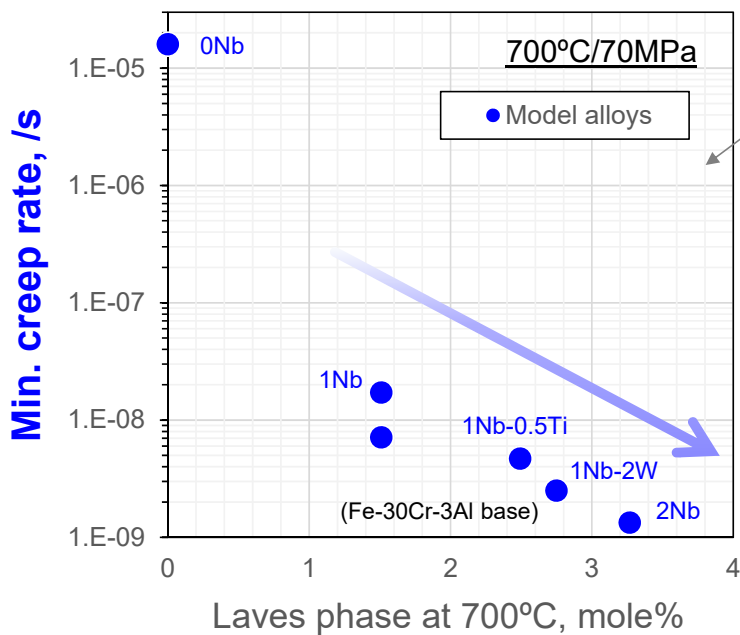


- Grain Boundary Zone Strength Factor:
$$\frac{\text{Grain boundary precipitate coverage}}{\text{Width of precipitation free zone}}$$

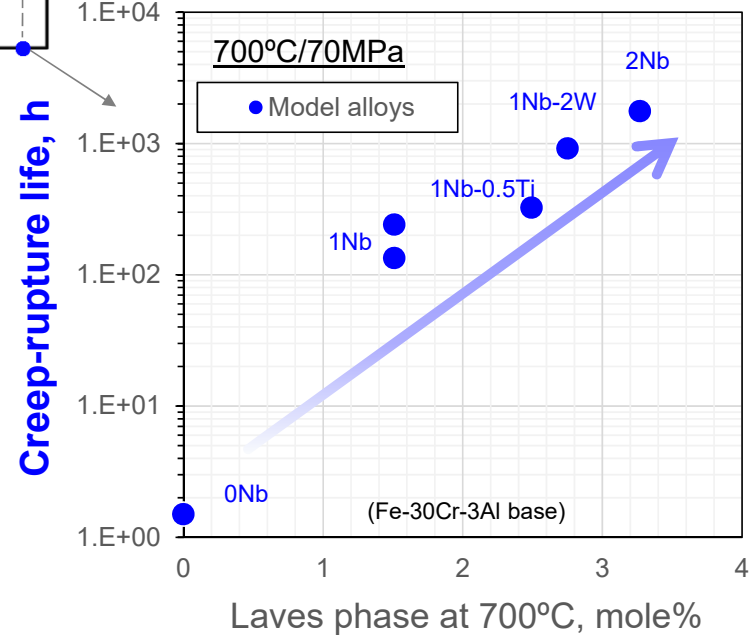
B. Shassere et al. (to be submitted)

Min. Creep Rate / Creep-rupture Life Depend On Fraction of Laves Phase Precipitates

Min. creep rate vs. Laves phase



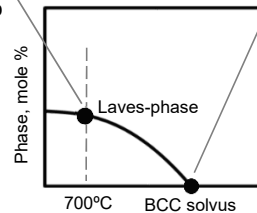
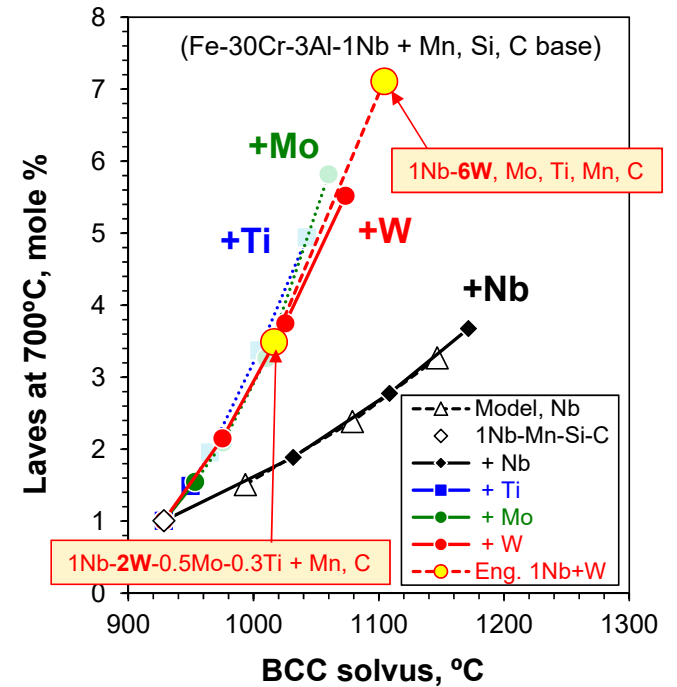
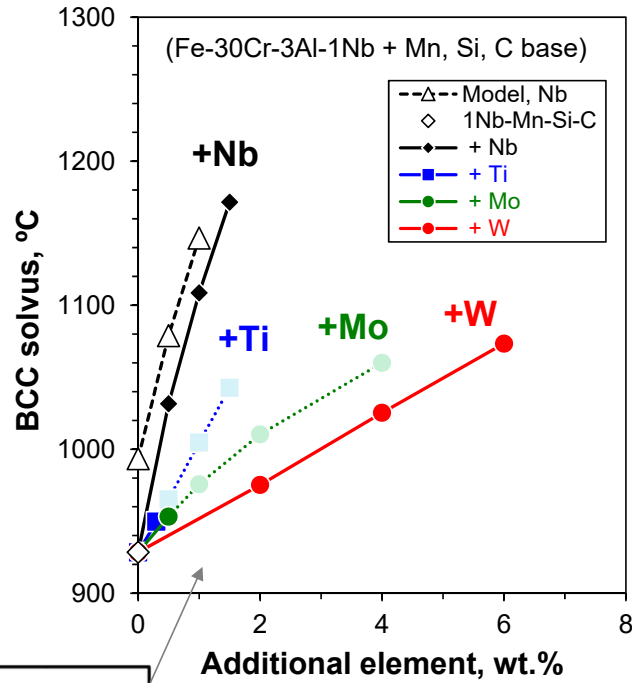
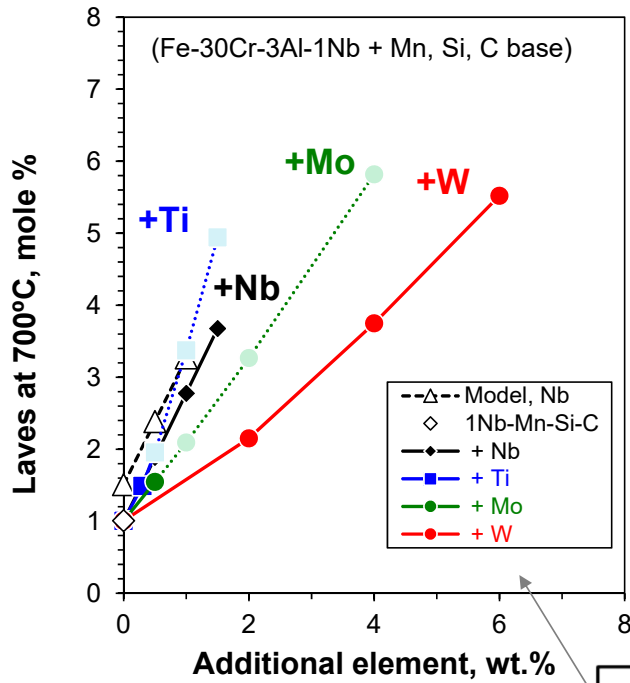
Creep-rupture life vs. Min. creep-rate



Moving toward Engineering Alloys

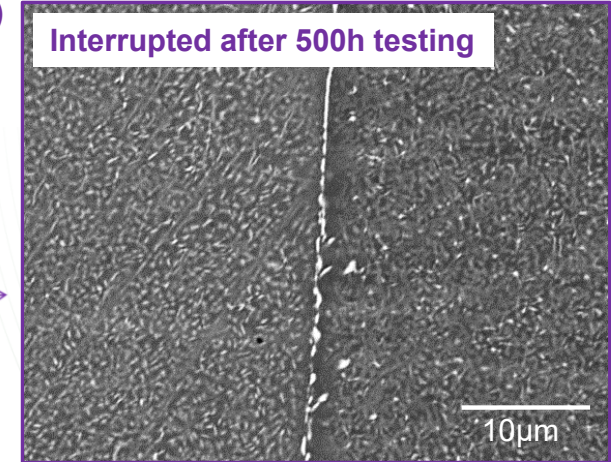
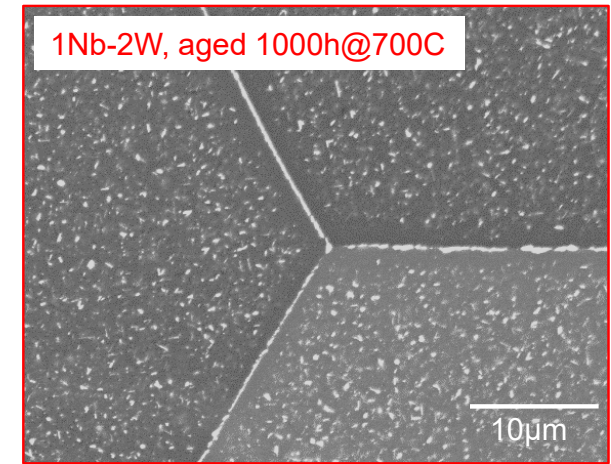
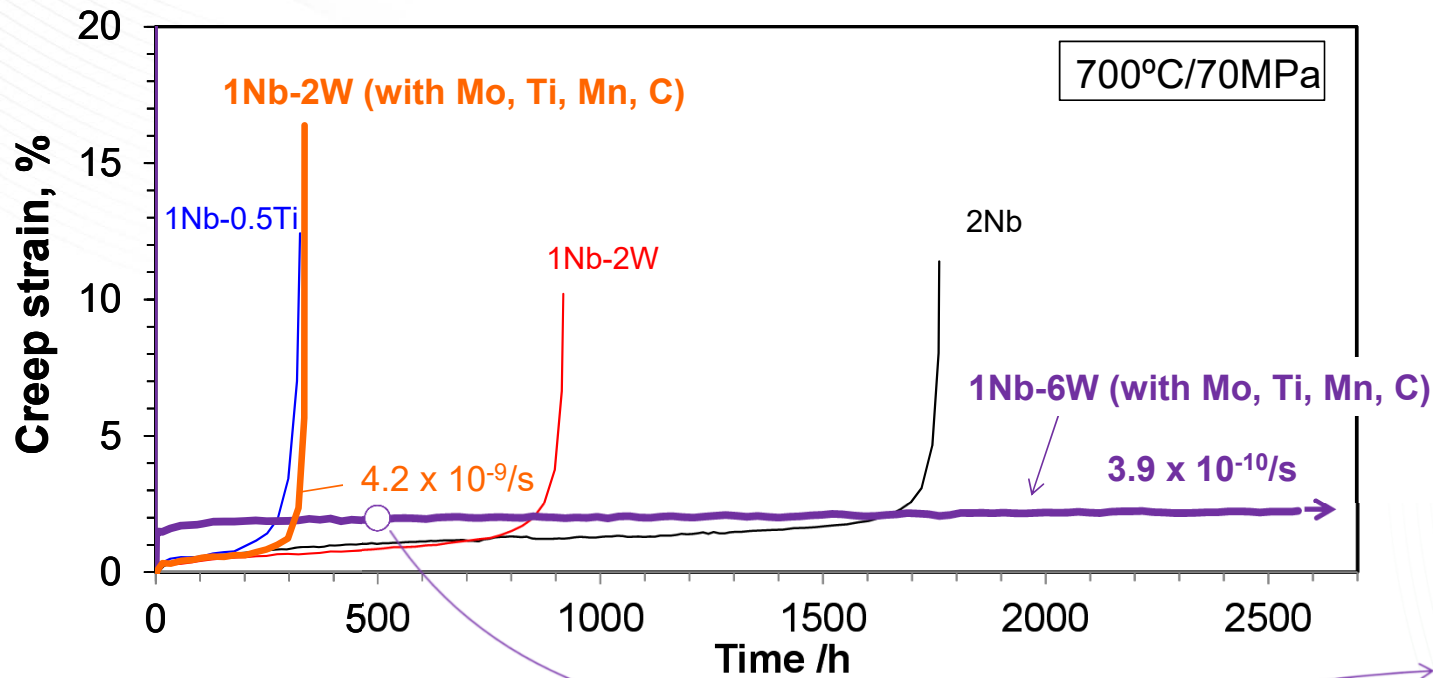
- Optimized the combination of Nb, W, Mo, and Ti:
 - Maximize Laves phase formation combined with reasonably low BCC solvus temperature (for Laves phase)
- Considerations of industrial quality chemistry:
 - Mn, Si, and C were intentionally added to simulate commercial steels
 - P, S, and N still need to be minimized for better weldability and oxidation resistance

BCC Solvus Temperature Control



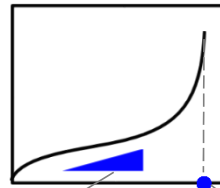
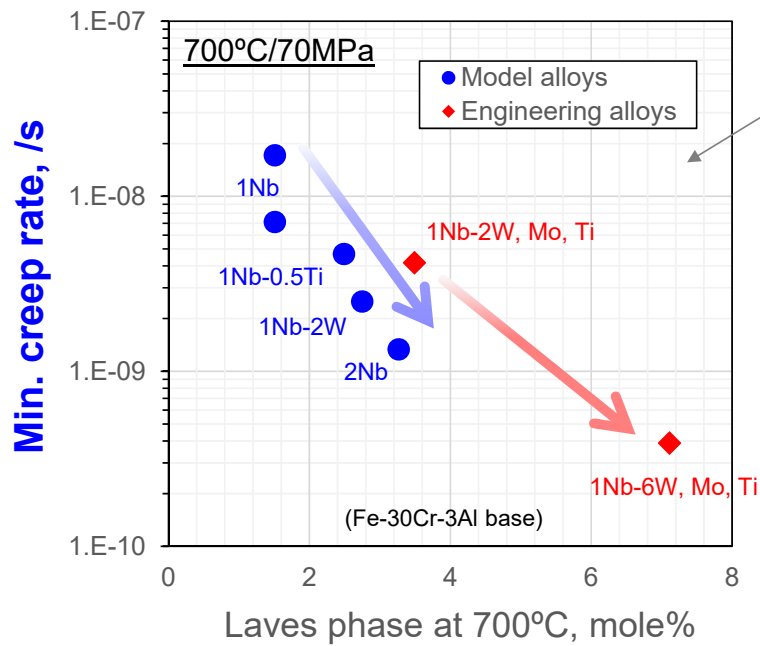
- Ti: ≤ 0.3 wt.% for better oxidation resistance
- Mo: ≤ 0.5 wt.% to avoid σ -FeCr or χ -FeCrMo formation

Improved Creep Resistance by Increased Volume Fraction of Laves-phase Precipitates

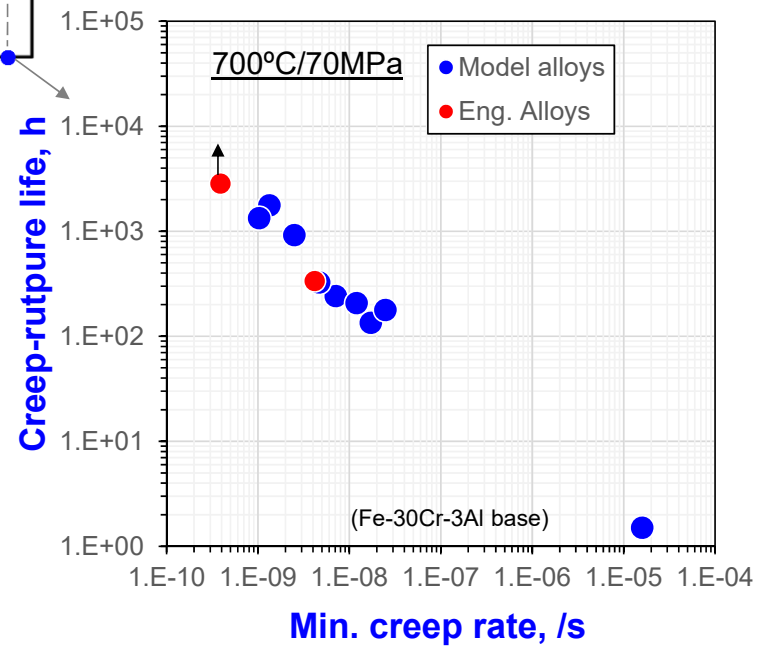


Min. Creep Rate / Creep-rupture Life Depend On Fraction of Laves Phase Precipitates

Min. creep-rate vs. Laves phase

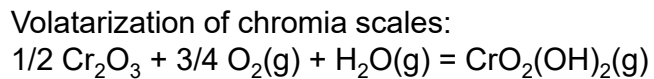


Creep-rupture life vs. Min. creep-rate

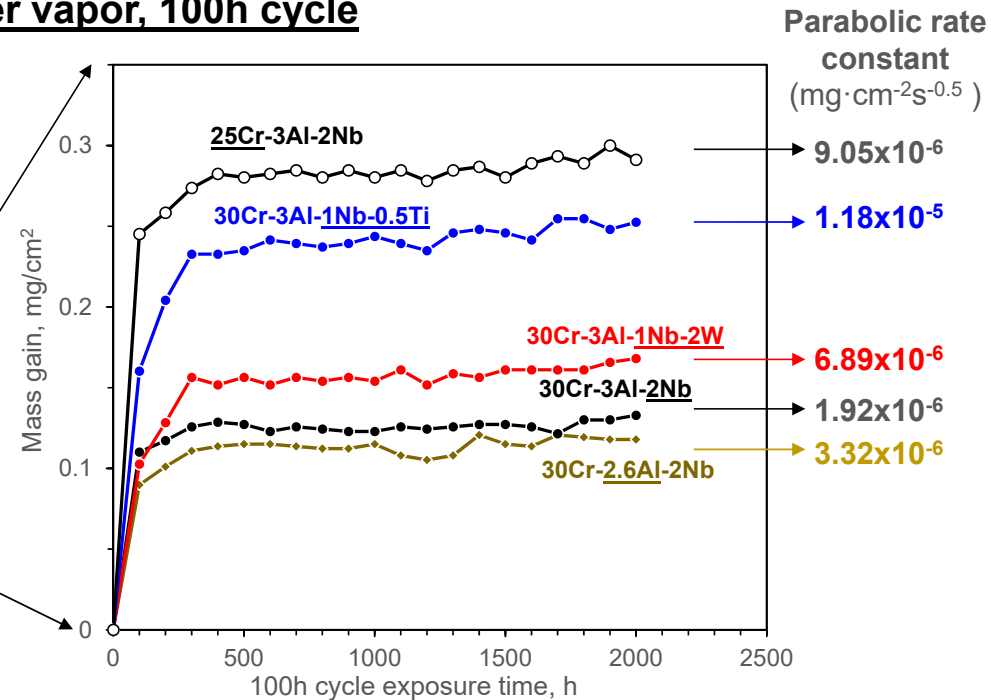
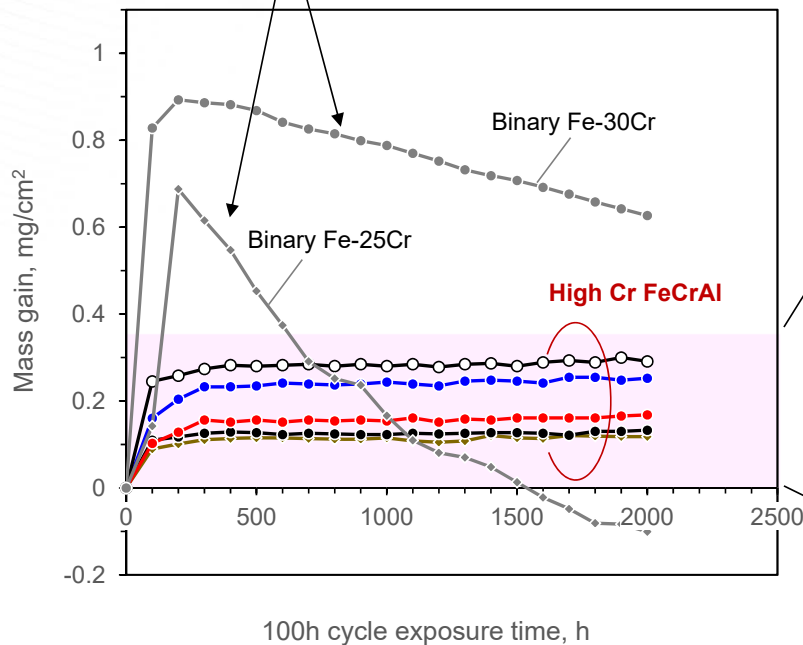


Important Role of Al Addition on Oxidation Resistance

- Slow oxidation kinetics indicates protective alumina scale formation



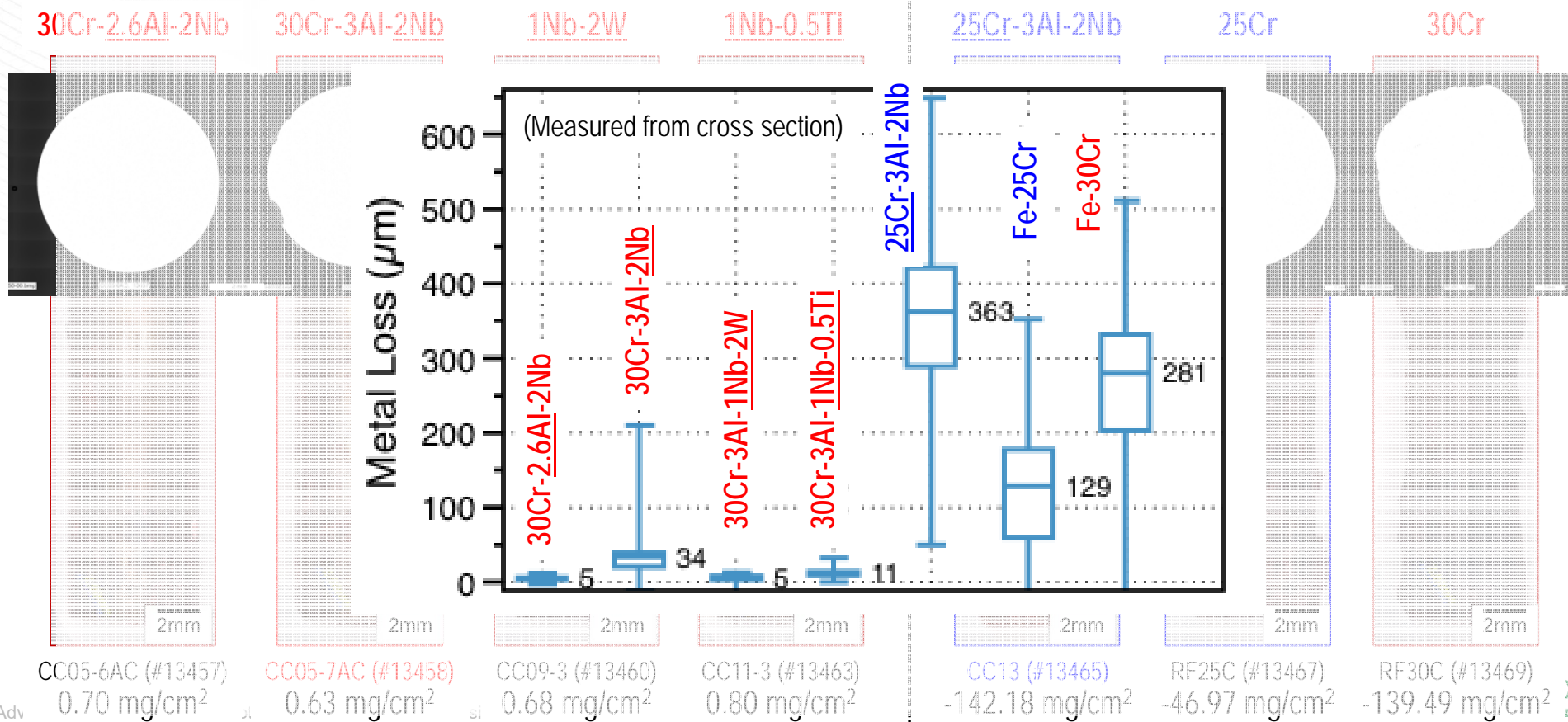
800°C, 10% water vapor, 100h cycle



Ash-corrosion Tested at 700°C for 500h

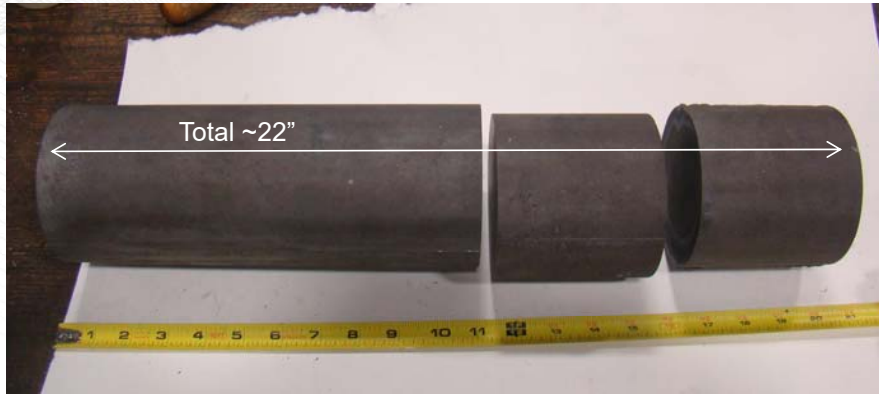
Ash: Al₂O₃ 16.9%, SiO₂ 22.6%, CaO 0.9%, Fe₂O₃ 7.8%, KOH 1%, TiO₂ 0.6%, MgO 0.2%, Fe₂(SO₄)₃ 19.8%, MgSO₄ 10.1%, K₂SO₄ 4.8%, Na₂SO₄ 15.1%

Gas: Synthetic gas simulating combustion environment in a steam plant (suggested by B&W/EPRI)

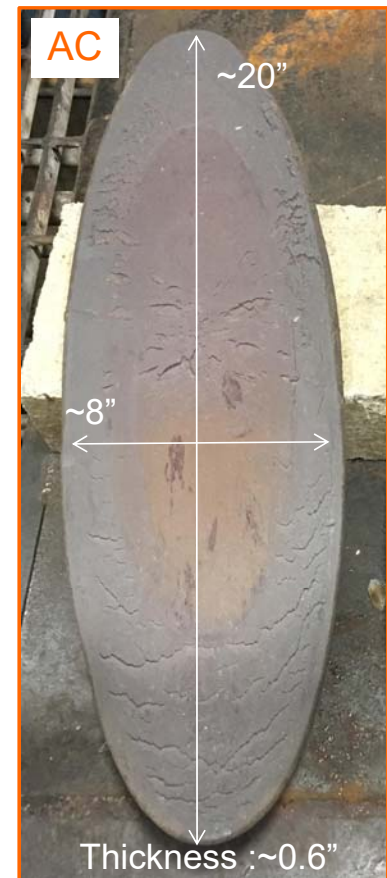
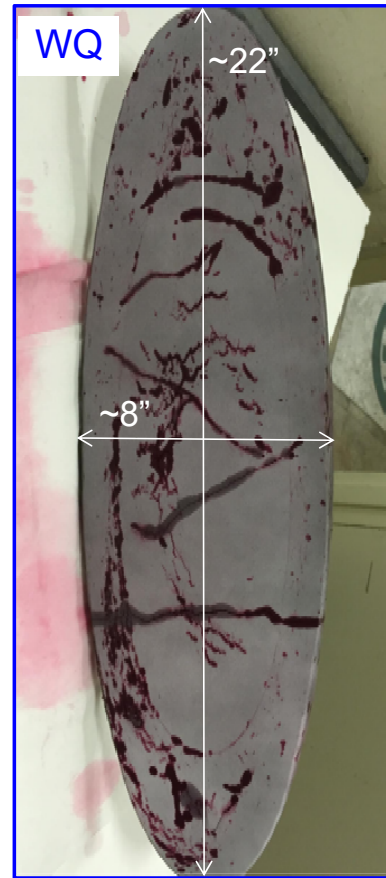


Scale-up Efforts (Fe-30Cr-3Al-2Nb-0.2Si-0.12Y)

VIM ingot (4" dia.) after HIPing



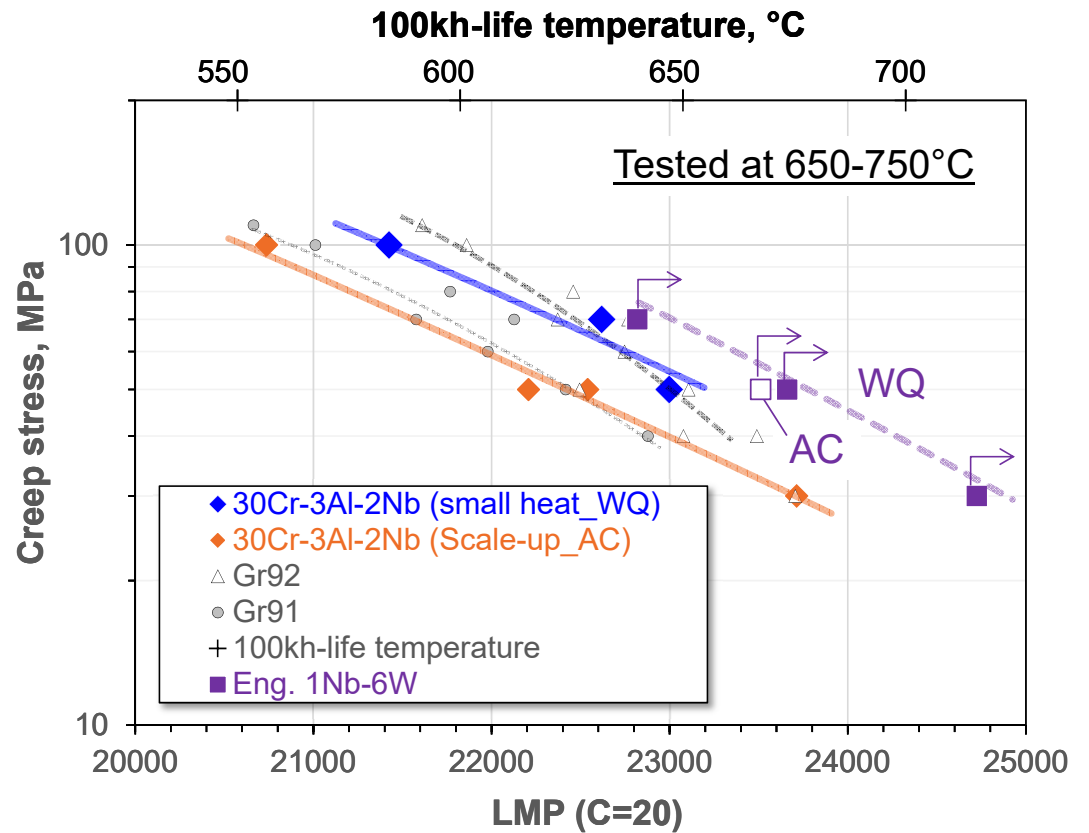
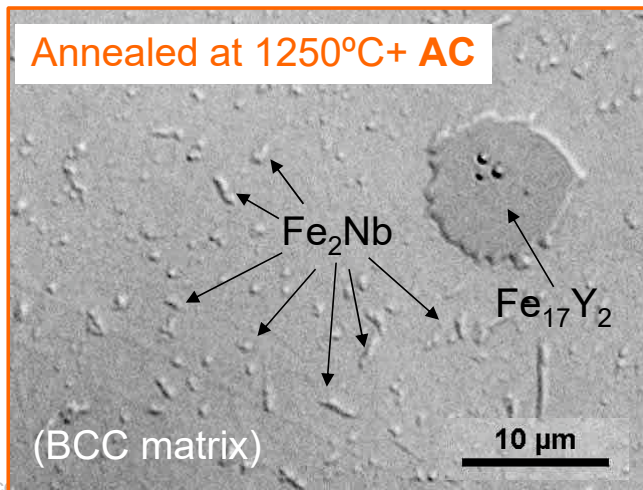
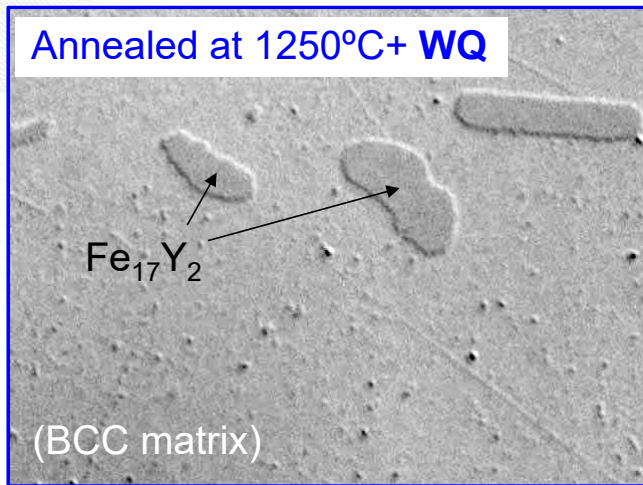
Rolled (and annealed) at 1250°C



Forged at 1250°C



Effect of Cooling Rate after Solution Annealing



Summary

- Effect of third element additions in Fe-30Cr-3Al base alloy on “microstructural stability” and “various high-temperature properties”:
 - W, Mo, and Ti additions are;
 - *Effective to lower the BCC solvus temperature*
 - *Detrimental to microstructure stability during creep deformation*
 - The larger volume fraction of Laves phase precipitate, the better for the creep resistance
 - Oxidation and ash-corrosion resistance are more sensitive to the amounts Cr and Al (and Nb) than the others
- Scale-up efforts in progress
 - Move to engineering alloys (based on 1Nb-6W +Mo, Ti, Mn, and C)
 - Require improvement of better processibility (against thermal shock, etc.)

Future Activities

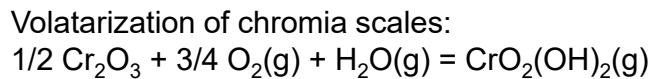
- **Property evaluation the second scale-up heat:**
 - *Screening basic properties (tensile, hardness, microstructure)*
 - *Intermediate/long-term creep-rupture test at or above 700°C*
 - *Ash-corrosion resistance evaluation at 700°C*
 - *GTAW screening*
- **New efforts on alumina-forming austenitic steels and Ni-base alloys**
 - *Target the applications in various extreme conditions (coking, metal dusting, sCO₂, etc.)*
 - *Communications with industrial partners initiated*



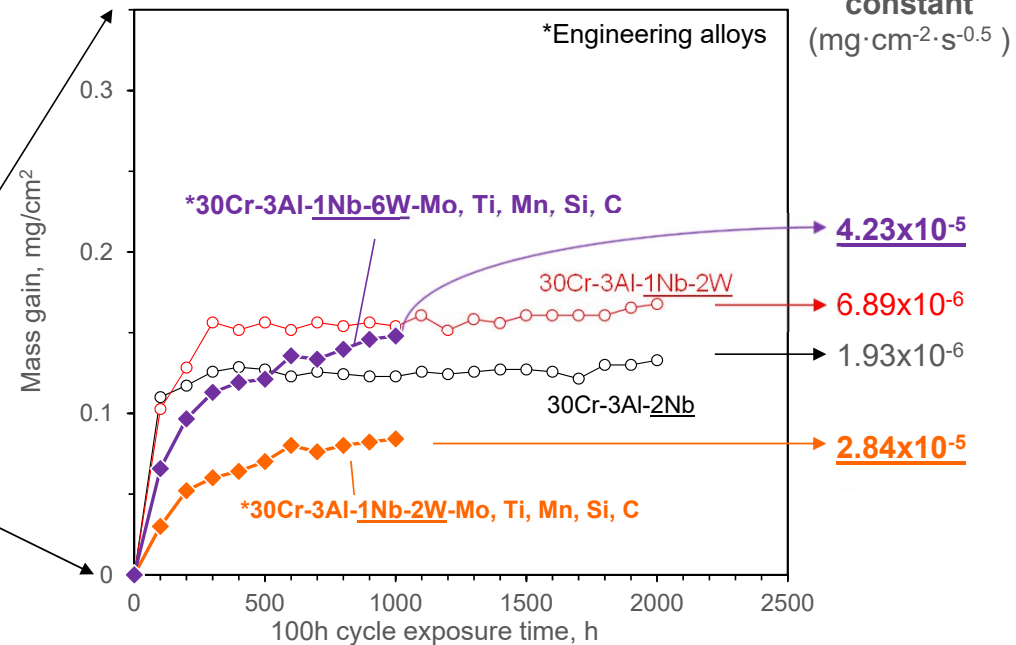
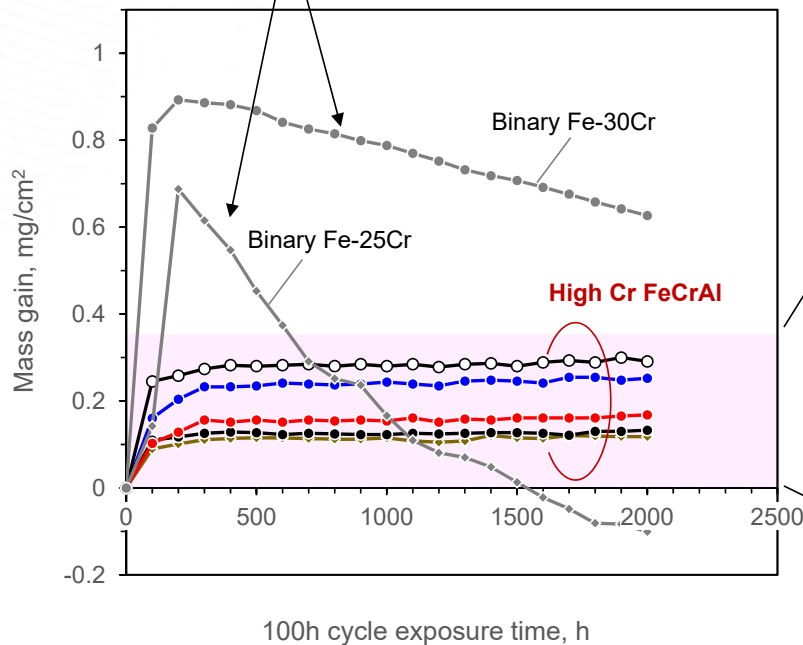
Thanks

Important Role of Al Addition on Oxidation Resistance

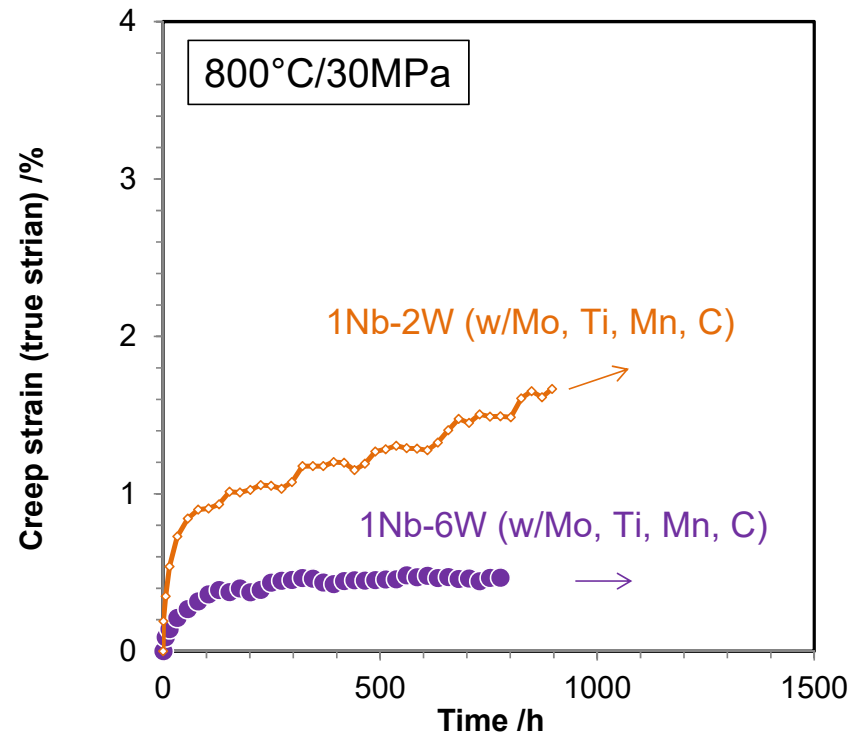
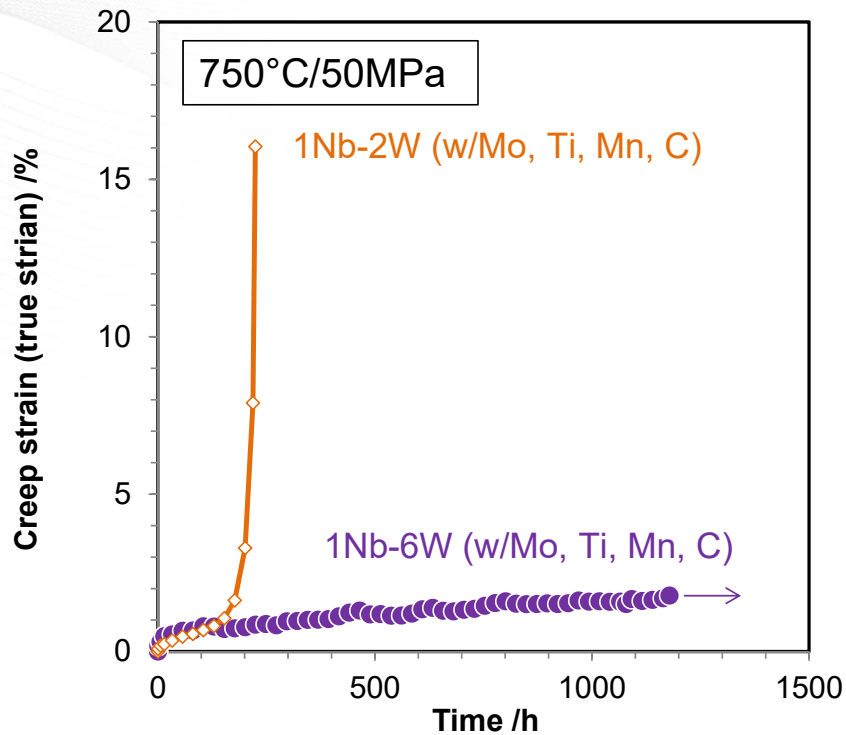
- Slow oxidation kinetics indicates protective alumina scale formation



800°C, 10% water vapor, 100h cycle

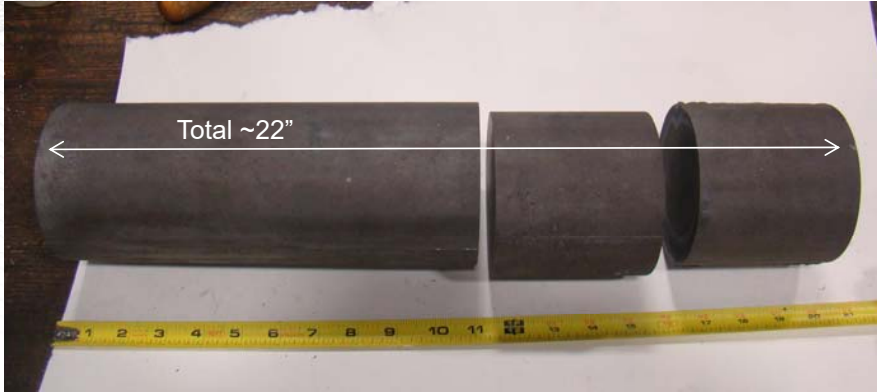


Higher Temperature Tests In Progress (750 & 800 °C)

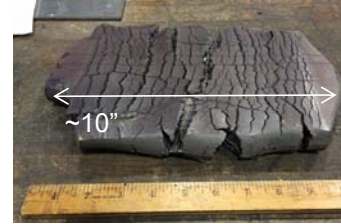


Scale-up Efforts (Fe-30Cr-3Al-2Nb-0.2Si-0.12Y)

VIM ingot (4" dia.) after HIPing



Rolled
at 800°C



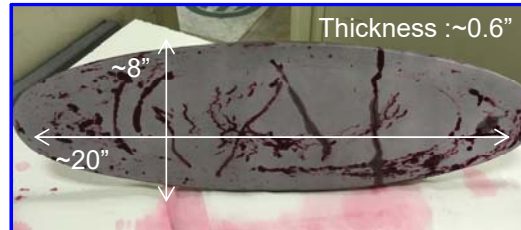
at 1000°C



Forged at 1250°C



at 1250°C
+ WQ



at 1250°C
+ AC

