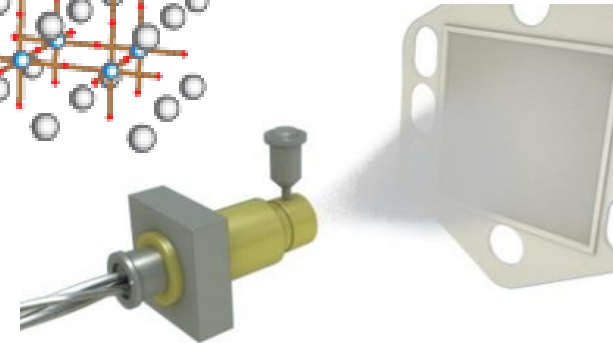
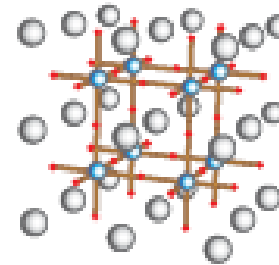




\*



# Development of a Thermal Spray, Redox Stable, Ceramic Anode for Metal Supported SOFC

Richard Hart  
GE Global Research  
Pitt Review July 20, 2016

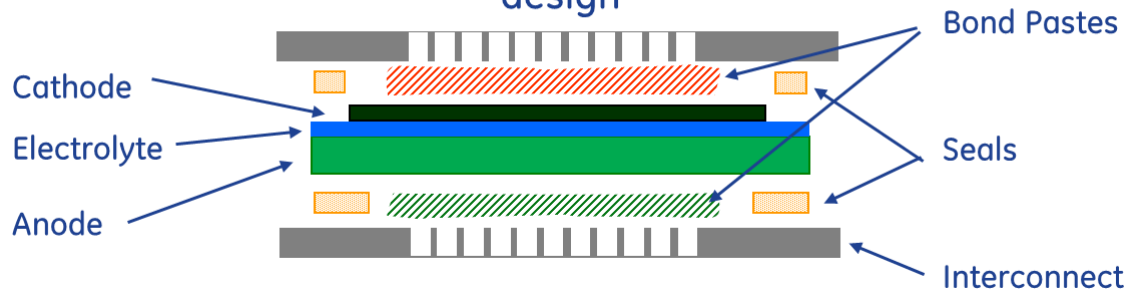
**Imagination at work.**

**SOFC Innovative Concepts and Core Technology Research**  
DE-FOA-0001229 Award FE0026169

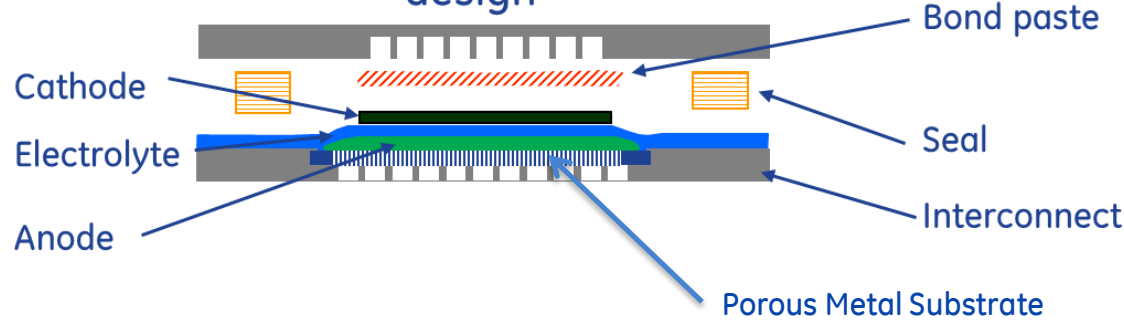
\*  
Trademark of General Electric Company

# Metal supported SOFC cells

Anode supported cell design



Metal supported cell design

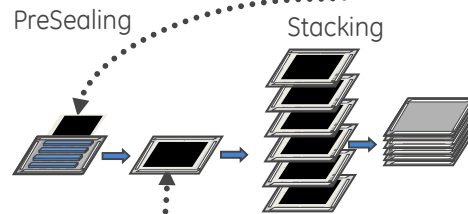
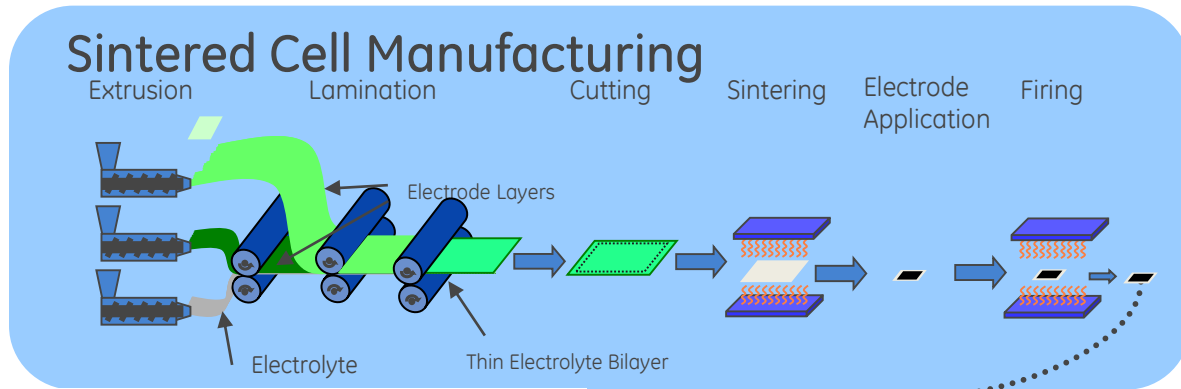


- Advantages:
- Integrated anode seal
  - Electrolyte in compression
  - Improved anode electrical contact
  - Increased active area
  - Lower anode polarization

- Challenges:
- Dense / hermetic electrolyte
  - Porous metal substrate degradation



# Low-cost manufacturing



## Advantages

- Larger area / Scalable
- Simplified sealing
- Low Capex / Modular
- Lean Manufacturing

## Thermal Spray



Leverage GE thermal spray expertise



# Fuel Cell Pilot Facility – Malta NY



# Traditional NiO(Ni)/YSZ anodes

- Advantages:

- High initial electrochemical activity
- Good electronic conductivity
- Low cost
- Well understood, wealth of data

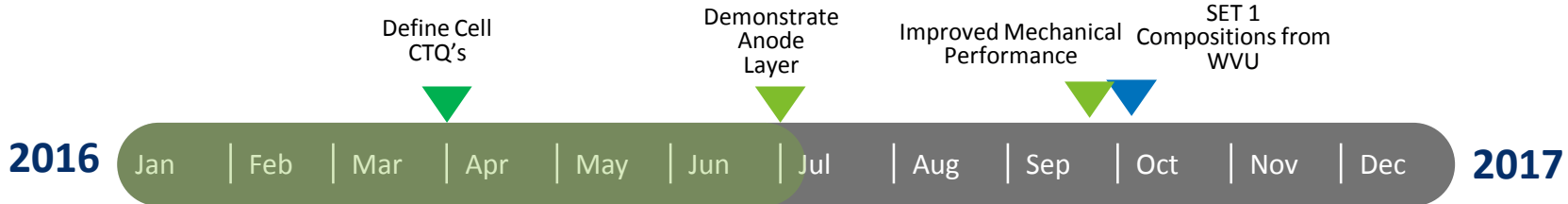
- Disadvantages:

- High redox Vol change (fuel↔air)
- Ni particle ripening/poisoning
- EHS concerns (NiO)
- Sourcing concerns (REACH in Eu)



# Project Plan & Deliverables (~\$3.5M, 3 year, 25% cost share)

Task	Owner	Timing	Objectives
1	GE Global Research	Months 1-36	<ul style="list-style-type: none"> <li>Defined by DOE; risk management, coordination, reporting</li> </ul>
2	GE Global Research GE-Fuel Cells	Months 1-12	<ul style="list-style-type: none"> <li>Derive anode layer requirements from existing systems models</li> <li>Tailor Global Research thermal spray process using single baseline composition</li> <li>Streamline (cost and lead time) powder engineering methods</li> <li>Establish redox cycle cell test procedures</li> </ul>
3	West Virginia University	Months 1-24	<ul style="list-style-type: none"> <li>Develop key materials properties measurements</li> <li>Hand off to GRC SET1 and SET2 Anode Compositions</li> </ul>
4	GE Global Research	Months 13-27	<ul style="list-style-type: none"> <li>Optimize thermal spray process for improved formulations</li> <li>Go/No – Does single scaled cell (100-400cm<sup>2</sup>) meet CTQs?</li> </ul>
5	GE Global Research GE-Fuel Cells	Months 28-36	<ul style="list-style-type: none"> <li>Powder scale up, cell fabrication scale up.</li> <li>Build and test, 5 kW SOFC stack for 1000 hr, Nat Gas/Sim Nat Gas fuel.</li> </ul>



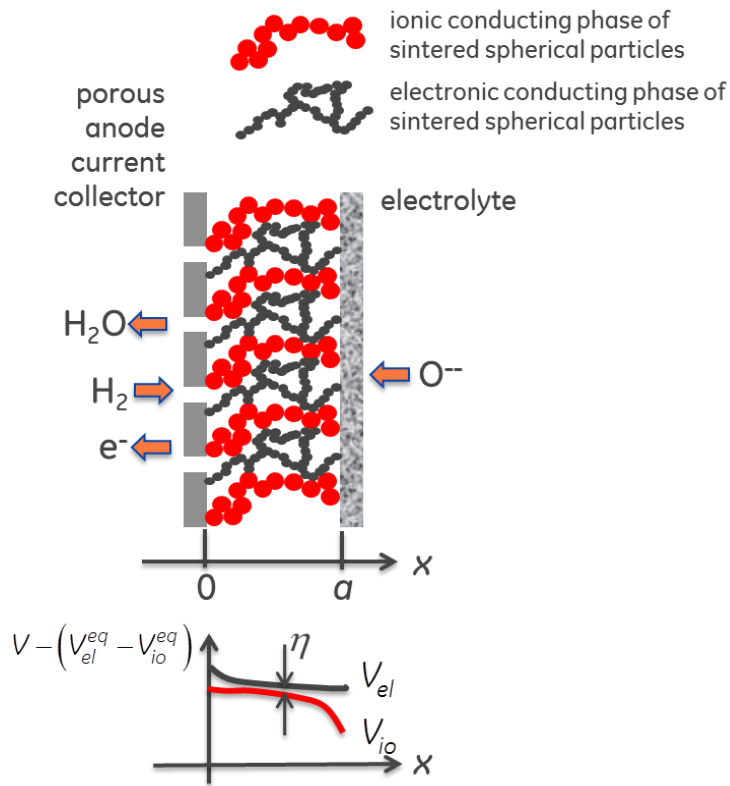
- Define cell specifications for Go/No Go decision point (27months) [March 31]
- Demonstrate a working all ceramic anode layer (OCV on a cell) [June 31]
- Demonstrate improvement mechanical perf. (no failure @ 1 cycle) [Sept 30]
- Deliver SET 1 compositions (WVU-> GE) [Oct 1]



# Electrochemical Model



# Electrochemical Model



For our system:

Red =  $Gd_{0.2}Ce_{0.8}O_{1.9}$

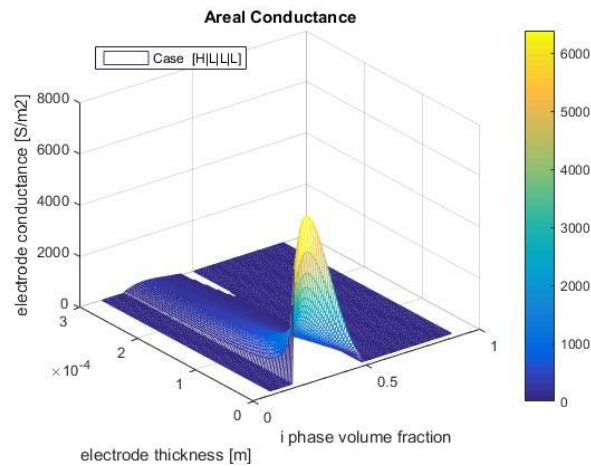
Black =  $La_{0.35}Sr_{0.65}TiO_3$

- Adapted simple Literature Model (Costamagna)
- Initial programming complete (Matlab)
- Completed 6 factor DOE exploring:
  - Electrode thickness
  - Particle size & ratio of particle sizes
  - Volume fraction of phases
  - Effective electron conductivity
  - Effective ion conductivity
- WVU performing screen printed electrode study for model validation/calibration (kinetics)

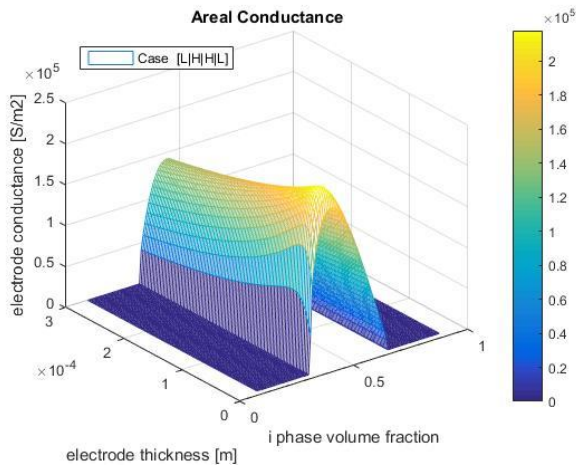




# Electrochemical Model - Example DOE Results

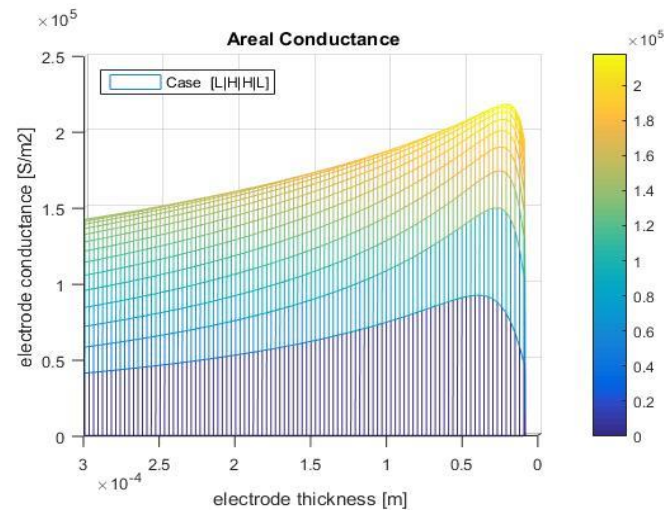


Quadrant 3:  $P=2$  and  $\sigma_{0el} = 20$  S/m



Quadrant 2:  $P=1$  and  $\sigma_{0el} = 2000$  S/m

- Wide range of electrode area conductance (1/ASR)
- Model results match qualitative expectations
- Identified regions of performance near GE goals



-Exchange current density for reactions largely unknown for these systems. Goal of WVU study.

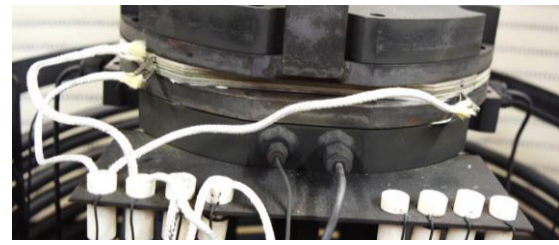
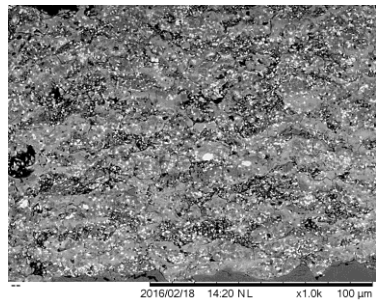
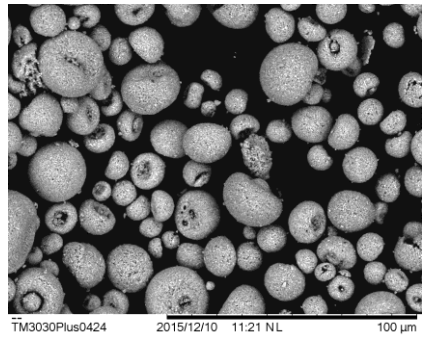
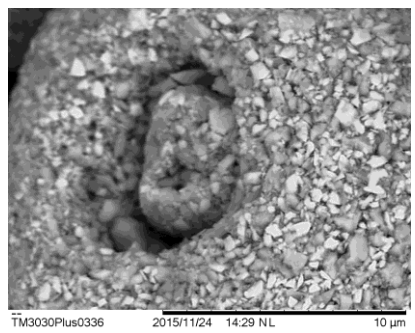
-Used model to aid material spec. definition



# Cell Testing Results



# Demonstrating Ceramic Anode Metal Supported Cells:



Sourced  
Engineered Powders

LST ( $\text{La}_{0.35}\text{Sr}_{0.65}\text{TiO}_3$ )  
GDC ( $\text{Gd}_{0.2}\text{Ce}_{0.8}\text{O}_{\sim 1.9}$ )

Coupon Screening  
Experiments

XRD, SEM, Permeability,  
DE, Roughness, etc...

100cm<sup>2</sup> Cells

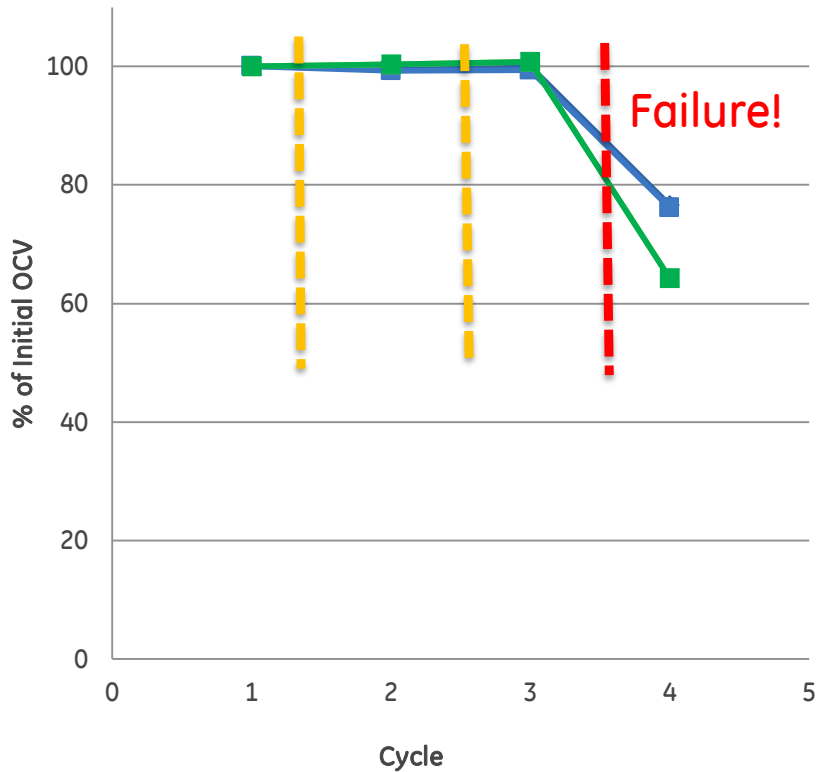
(2 cell stacks)

OCV, W/cm<sup>2</sup>  
Redox Stability

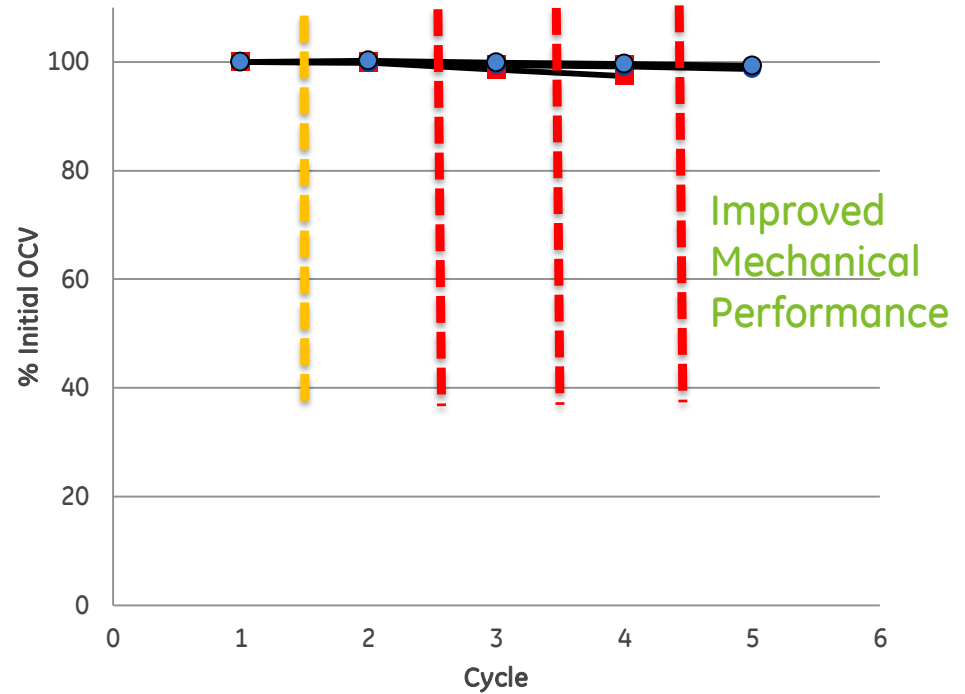


# Stack Redox Cycling – Ni/YSZ vs. Alt Anode Stacks

## Ni/YSZ



## LST/GDC



- Orange = Standard Thermal Cycle w/ H2 Flow (we did two of these, to check cell health)
- Red = Redox Thermal Cycle (no protective flow)

Ni/YSZ cells fail after a single redox cycle

Ceramic anode cells survive up to 5 cycles

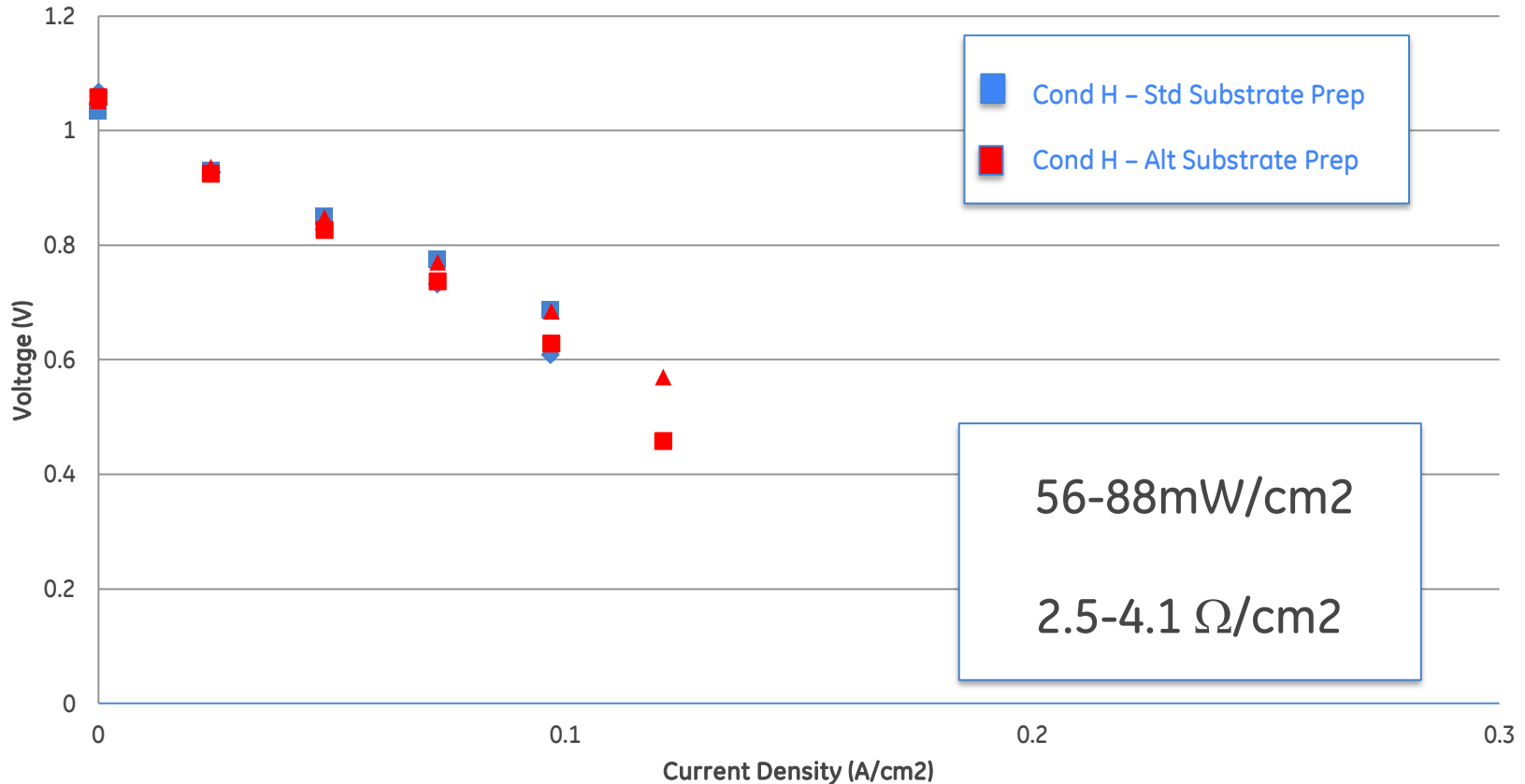


Confirmation of damage mechanism! (similar to mechanism previously reported for sintered Cells)

Hart, Renko, Northey

# LST/GDC Anodes- Power curves (low Uf)

## Agglomerated, Uncalcined, LST/GDC - Cond H Results



Demonstrated working ceramic anode cells (June 31 + Sept 30 Milestones)

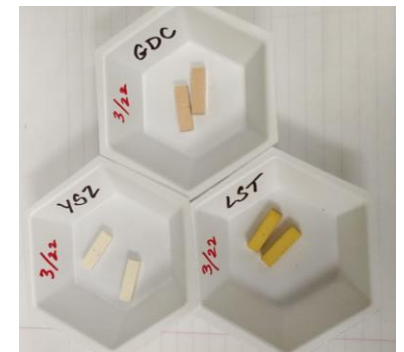
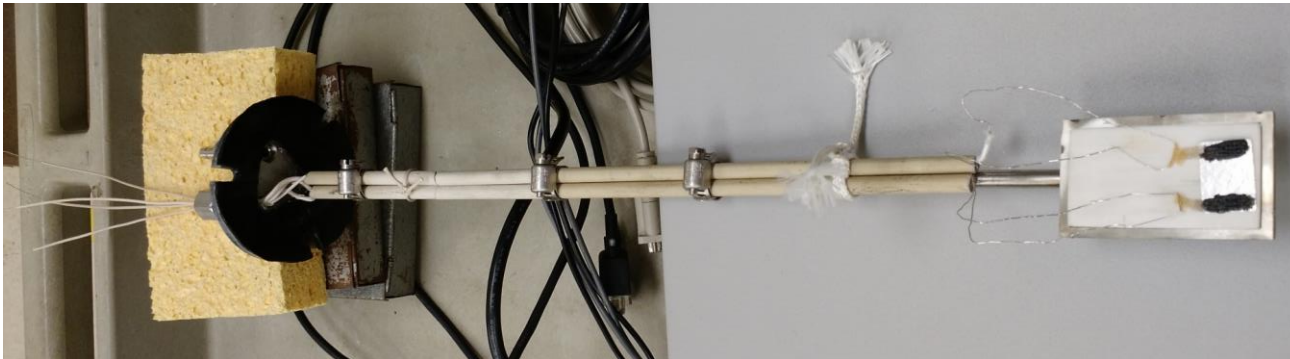
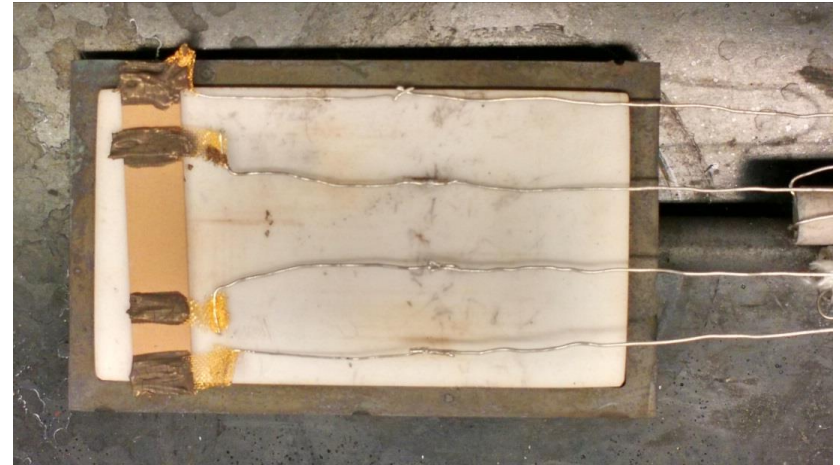
Next step: improve upon extreme low power density! (improve microstructure & test new formulations)



# Material Conductivity Testing Results

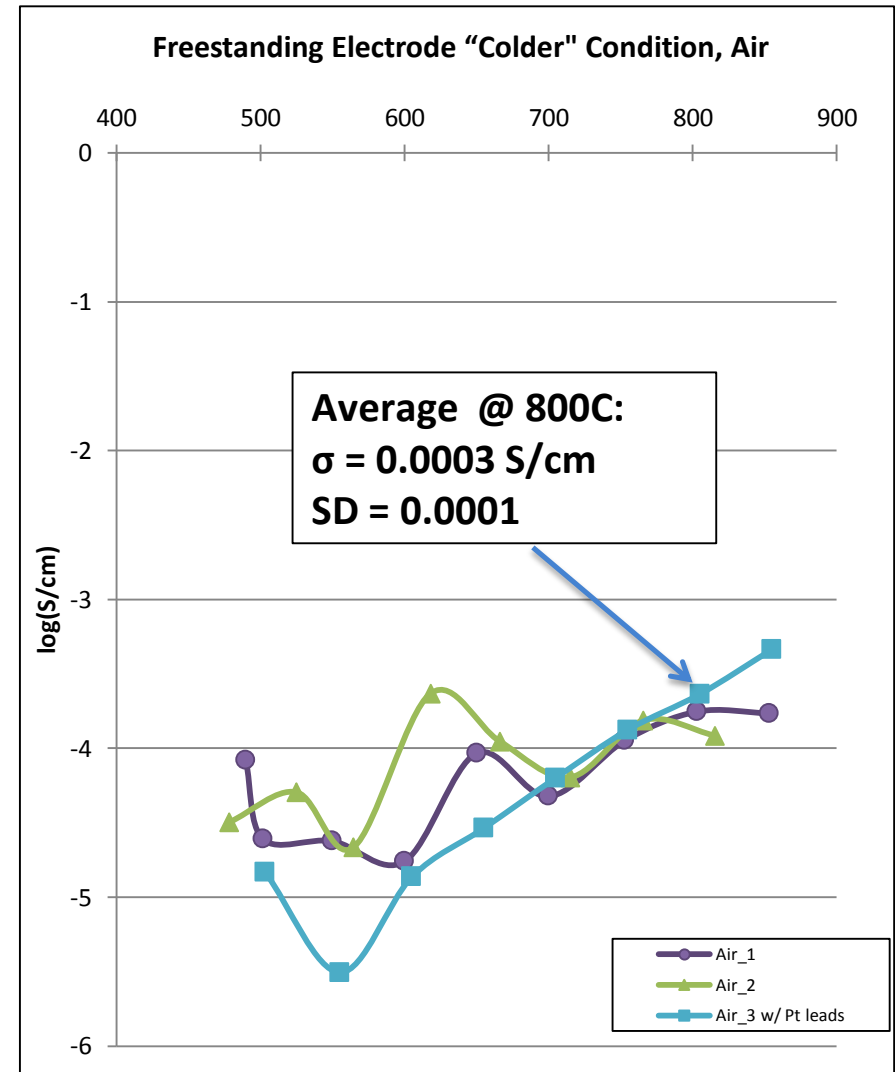
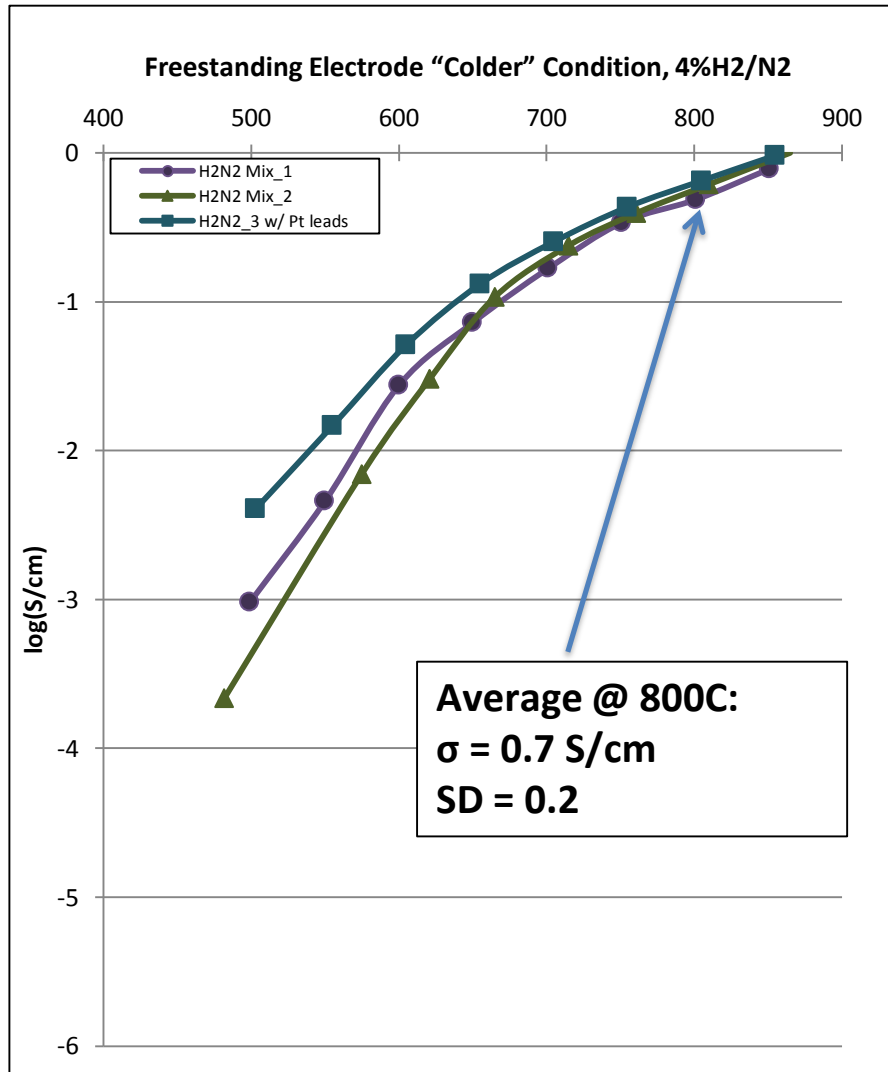


# Conductivity Test Setup (GE-GRC)



# Conductivity Results – Replicate Measurements

## Free Standing LST/GDC thermal spray films

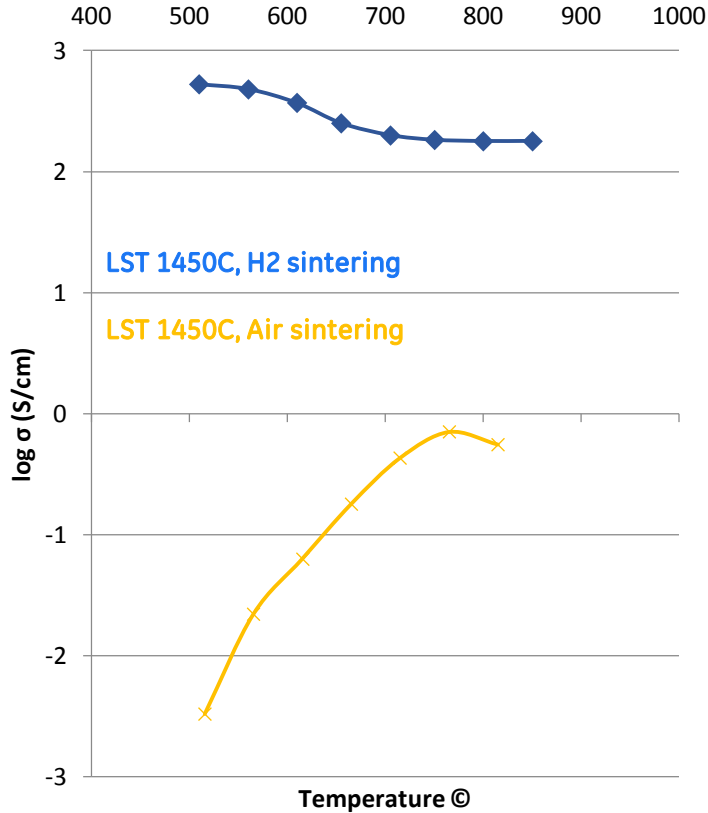


Solatron 1287/1260, 4pt, AC impedance, ~1kHz



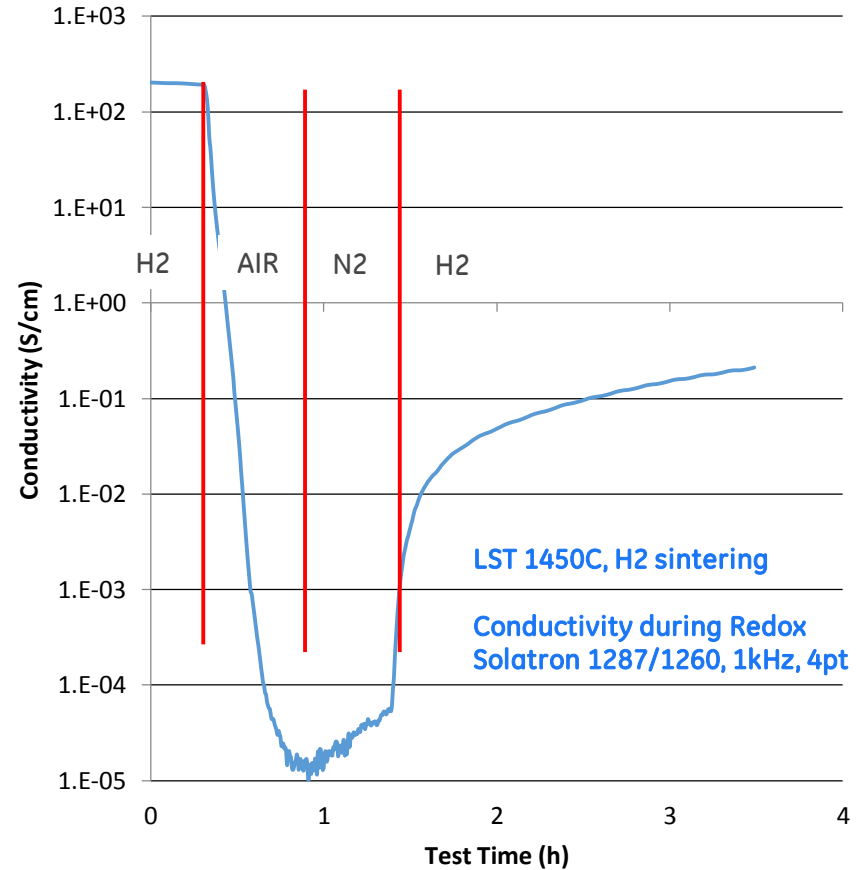
# LST Conductivity – Effect of Sintering Atm, and Redox:

### LST – 1450C sintered, effect of atm:



Solatron 1287/1260, 4pt, AC impedance, ~1kHz

### LST Pellet Conductivity – Redox Cycling



E-chem Model -> need to identify materials w/ >10-20S/cm after redox



# WVU & GE Anode Material Development



# Material Development Testing Plan

## Conductivity Testing

- Screen w/ pressed pellets or free-standing films
- Electron Conductivity > 20S/cm (~30x improvement)
- Ion Conductivity >  $1 \times 10^{-2}$  S/cm (~100x improvement)

## Mechanical Stability During Redox Cycling (800C)

- Redox Vol. Change target still in progress (Mech E) < GDC soft target
- Measuring vol change w/ redox dilatometry (**good baseline**)

## SOFC Cell Testing

- WVU using 1" button testing
- GRC using 100cm<sup>2</sup> metal supported cells (2-6 cell stacks)



# Formulation Development Plan:

## GE Global Research:

- Pivot: added on ceramic synthesis efforts
  - GE Targeting lower risk/reward candidates
- Pivot: Testing GE lab scale spray dry (schedule risk abatement)

## WVU:

- Starting from WVU's previous Anode Composition work
- Developed Redox Dilatometry methods
- Using 1" SOFC test bed: model validation & comp screening

Goal: thermal spray 1<sup>st</sup> new ceramic formulation by Oct 1



# *Lit Overview Alternative Ceramic Anodes*

Structure	Performance Attributes	Examples	Research Level Required (RLR)
<b>Perovskite</b>	Good/excellent e- or hole-conductor, low catalysis of HCs	LST, YST, LSCM	Low
<b>Layered-Perovskite</b>	Good e- conductor (very slight ionic), some catalysis of HCs	$\text{Sr}_2\text{MgMoO}_{6-x}$	Low
<b>Fluorite</b>	Ionic conductor (very slight electronic), low/high catalysis	doped- $\text{CeO}_2$	High
<b>Pyrochlore</b>	Low e- conductor, some catalysis of HCs, high redox stability	doped- $\text{Gd}_2\text{Ti}_2\text{O}_7$	High
<b>Ruddlesden-Popper</b>	Low e- conductor, high redox stability (for Ti or Nb-oxides)	$(\text{Sr},\text{La})_3(\text{Mg},\text{Nb})_2\text{O}_7$	High
<b>Tungsten-Bronzes</b>	Good/low e- conductor, redox stable, chemical stability issues	$\text{Sr}_{0.2}\text{Ba}_{0.4}\text{Ti}_{0.2}\text{Nb}_{0.8}\text{O}_3$	High

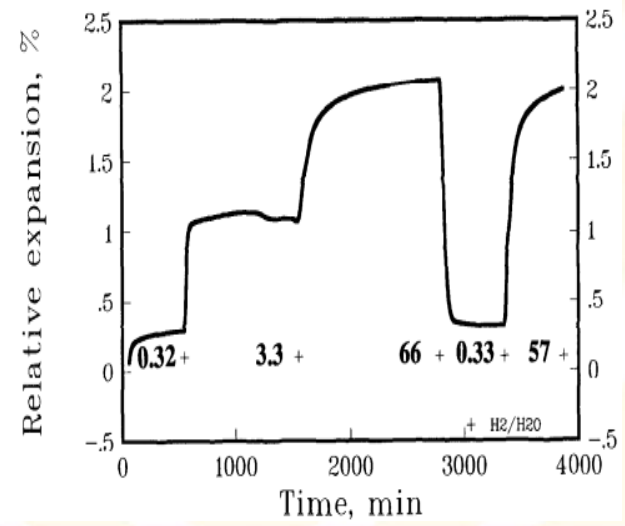
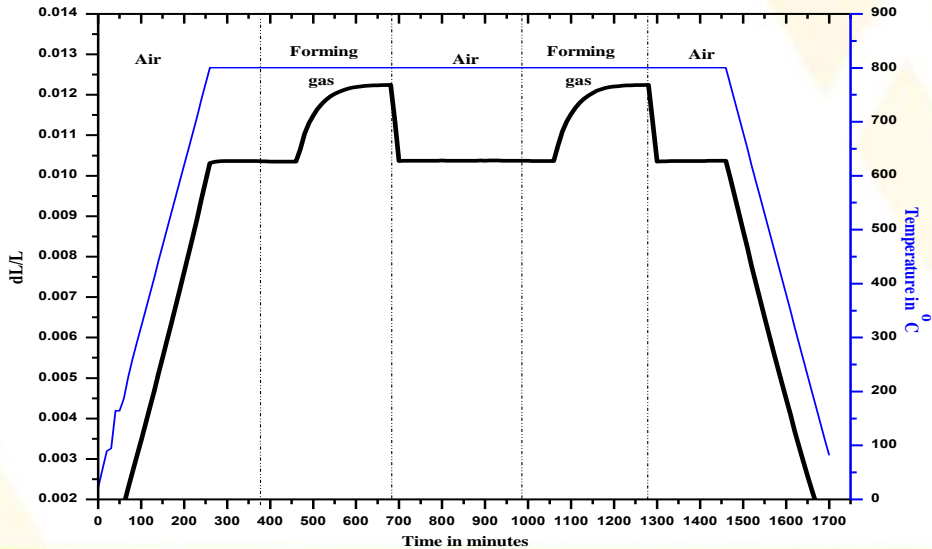


# Redox Dilatometry



**NETZSCH dilatometry setup in WVU for thermomechanical analysis**

- Change in protocol was necessary (longer dwell times)
- Redox behavior for GDC now matches lit data shapes:

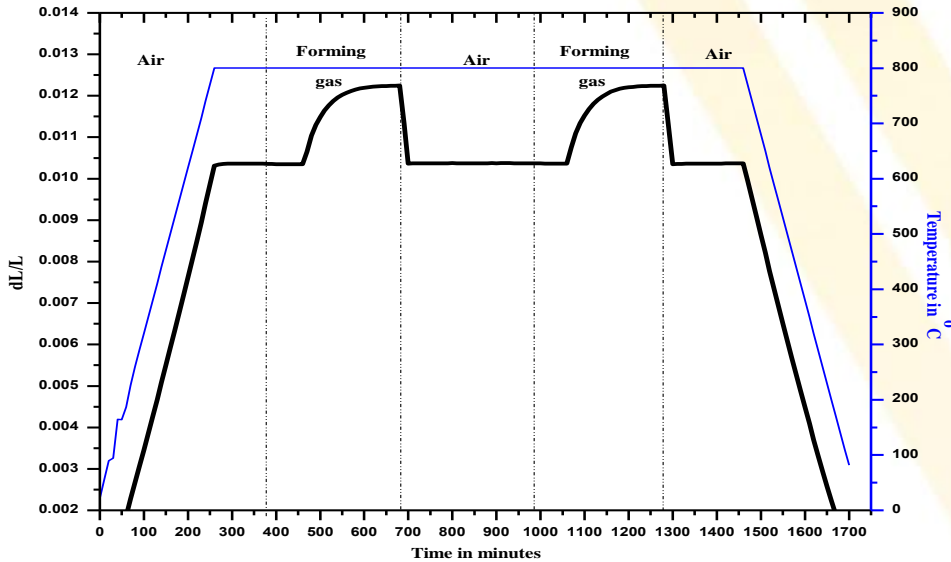


CTE in Air between 25-800°C is  $\sim 13.23 \times 10^{-6}$   
 $\sim 0.2\%$  volume changes due to redox

G Mogensen, M. Mogensen/Thermochim Acta 214 (1993) 47-50

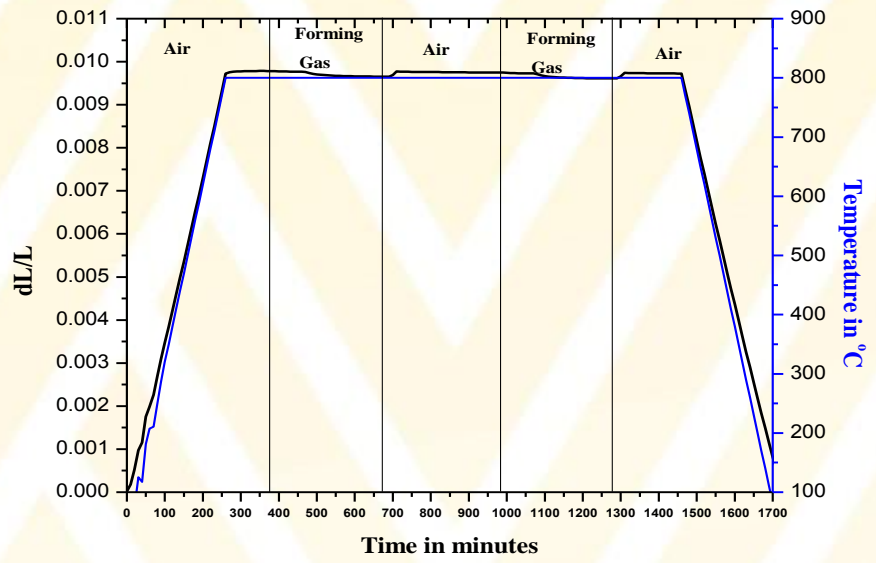


# Redox Dilatometry (LST and GDC)



GDC (GRC Supplied)

CTE in Air between 25-800°C is  $\sim 13.23 \times 10^{-6}$   
 $\sim 0.2\%$  volume changes due to redox



LST (GRC Supplied)

CTE in Air between 25-800°C is  $12.51 \times 10^{-6}$   
 $\sim 0.02\%$  volume changes due to redox?



# Summary

- Demonstrated working ceramic anode cells
  - Improved mechanical performance vs. NiO/YSZ anodes
  - Very low power density (formulation + microstructure)
- Redox conductivity tests identified insufficient materials properties for baseline composition (LST)
  - Short term microstructure optimization delayed temporarily
- Formulation development in progress and additional resources at GE added to help accelerate





# GE Team:

Rich Hart

PI, testing & direction

Larry Rosenzweig, Bastiann Korevaar,  
Paul Thomas

Thermal Spray GRC

Stephen Bancheri, Susan Corah

Powder development

Erik Jezek, Becky Northey

Materials testing, microstructure &  
degradation

Dayna Kinsey, Luc Leblanc, Matt Alinger  
Todd Striker, Andy Shapiro

GE Fuel Cells, scale up Thermal Spray  
Systems Support

Mike Vallance

Echem Model

Jae Hyuk Her, Erik Telfeyan, Matt Ravalli

Analytical Support

Johanna Wellington, Steve Duclos,  
Katharine Dovidenko, Vanita Mani

GE Management Support



## **Principle Investigators:**

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## **Research Assistants:**

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West Virginia University



# Acknowledgements

- GE Fuel Cells SOFC Team
  - GE Global Research Team
  - WVU (Dr. Sabolsky, Dr. Liu, Dr. Zondlo, & team)
  - Steven Markovich @ DOE/NETL
- 
- Funding provided by the US Department of Energy through cooperative agreement FE0026169

This material is based upon work supported by the Department of Energy under Award Number FE0026169. However, any opinions, findings, conclusions, or recommendations expressed herein are those of the authors and do not necessarily reflect the views of the DOE.

