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April 27-30, 2015, Pittsburgh, PA

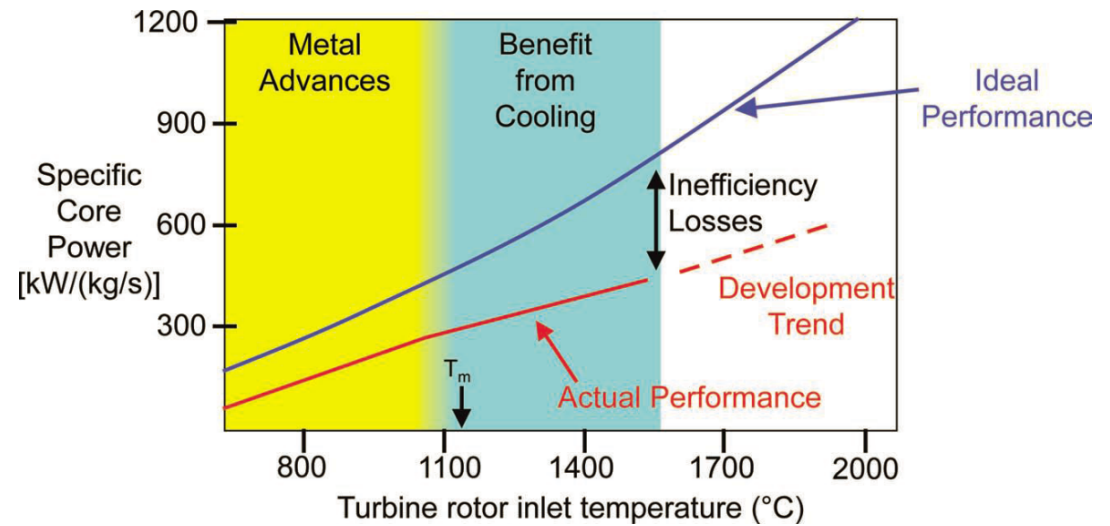


# **Mechanically Activated Combustion Synthesis of MoSi<sub>2</sub>-Based Composites**

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<b>Program Manager:</b>	<b>Richard Dunst</b>

# The Goal

- ❑ To increase the thermal efficiency of power plants, we need materials that can operate **beyond 1300°C** without cooling.
- ❑ **Molybdenum silicides and borosilicides** are potential replacements for Ni-based superalloys.
- ❑ This project investigates a **combustion-based** route for fabricating these materials.



**Specific core power vs turbine inlet temperature for gas turbine power plants**

Dimiduk and Perepezko, *MRS Bull.* 28 (2003) 639  
 Perepezko, *Science* 326 (2009) 1068  
 Lemberg and Ritchie, *Adv. Mater.* 24 (2012) 3445



# Material Design Considerations



- ❑ **MoSi<sub>2</sub> advantages and drawbacks**
  - Melting point: 2030°C
  - Excellent high-temperature oxidation resistance
  - Low fracture toughness at room temperature
  - Low strength at elevated temperatures
  
- ❑ **Mo-rich phases such as Mo<sub>3</sub>Si and Mo<sub>5</sub>Si<sub>3</sub> (T<sub>1</sub> phase)**
  - Better mechanical properties
  - Worse oxidation resistance
  
- ❑ **Boron-containing phases such as Mo<sub>5</sub>SiB<sub>2</sub> (T<sub>2</sub> phase)**
  - Improved oxidation resistance due to a borosilicate glass layer



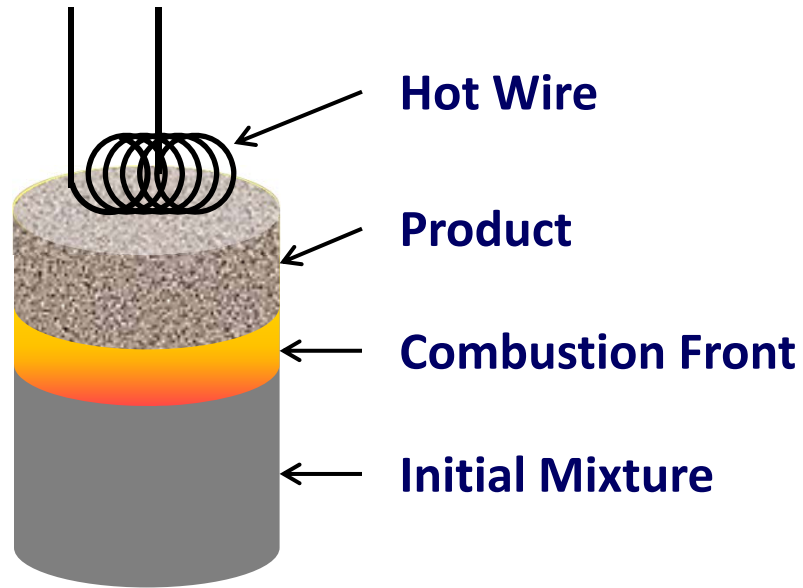
# Fabricating Mo-Si-B Alloys



- ❑ **Arc-melting and optical floating zone**
  - Extremely high temperatures limit scalability.
  - High energy consumption
  
- ❑ **Mechanical alloying**
  - Long milling times
  - Contamination by grinding media
  
- ❑ **Self-propagating high-temperature synthesis (SHS)**
  - Short time
  - High purity
  - Low energy consumption



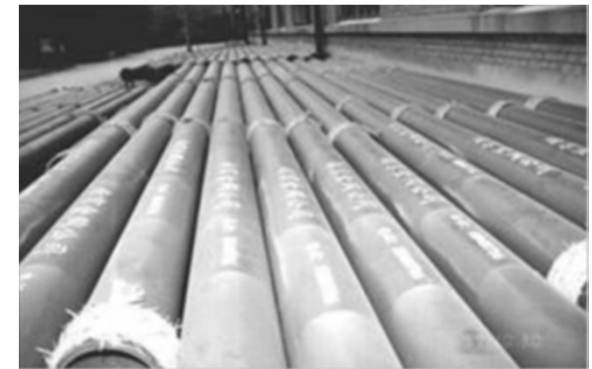
# Self-propagating High-temperature Synthesis (SHS)



Schematic of SHS process



Image: [www.ism.ac.ru/handbook/shsf.htm](http://www.ism.ac.ru/handbook/shsf.htm)



Large-scale ceramic-lined steel pipes produced using SHS  
A.G. Merzhanov, *J. Mater. Chem.* 14 (2004) 1779



# SHS Techniques Used in the Project

- Since increasing Mo content decreases exothermicity, methods for facilitating ignition were used:
  - Mechanical activation of the mixture (MASHS)
  - Additional exothermic reactions *inside* the mixture
  - Additional exothermic reactions *outside* the mixture (“chemical oven”)
  
- Since SHS often produces porous products, SHS compaction (pressing immediately after combustion) was used.



# Materials Fabricated in the Project

- $\text{MoSi}_2$ – $\text{Mo}_5\text{Si}_3$  composites
  
- Materials based on  $\text{Mo}_5\text{SiB}_2$  phase
  - $\text{Mo}_5\text{SiB}_2$  – MoB
  - $\text{Mo}_5\text{SiB}_2$  –  $\text{TiB}_2$
  - $\alpha$ -Mo/ $\text{Mo}_5\text{SiB}_2$ / $\text{Mo}_3\text{Si}$  (Mo–12Si–8.5B)



# Experimental Procedure



# Pellet Preparation

Mixing



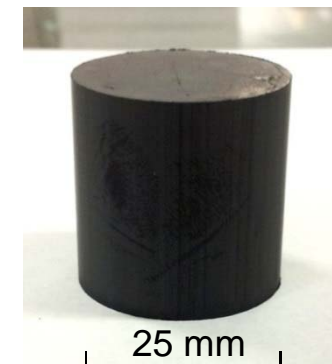
**3-D inversion  
kinematics mixer  
(Inversina 2L)**

Mechanical activation

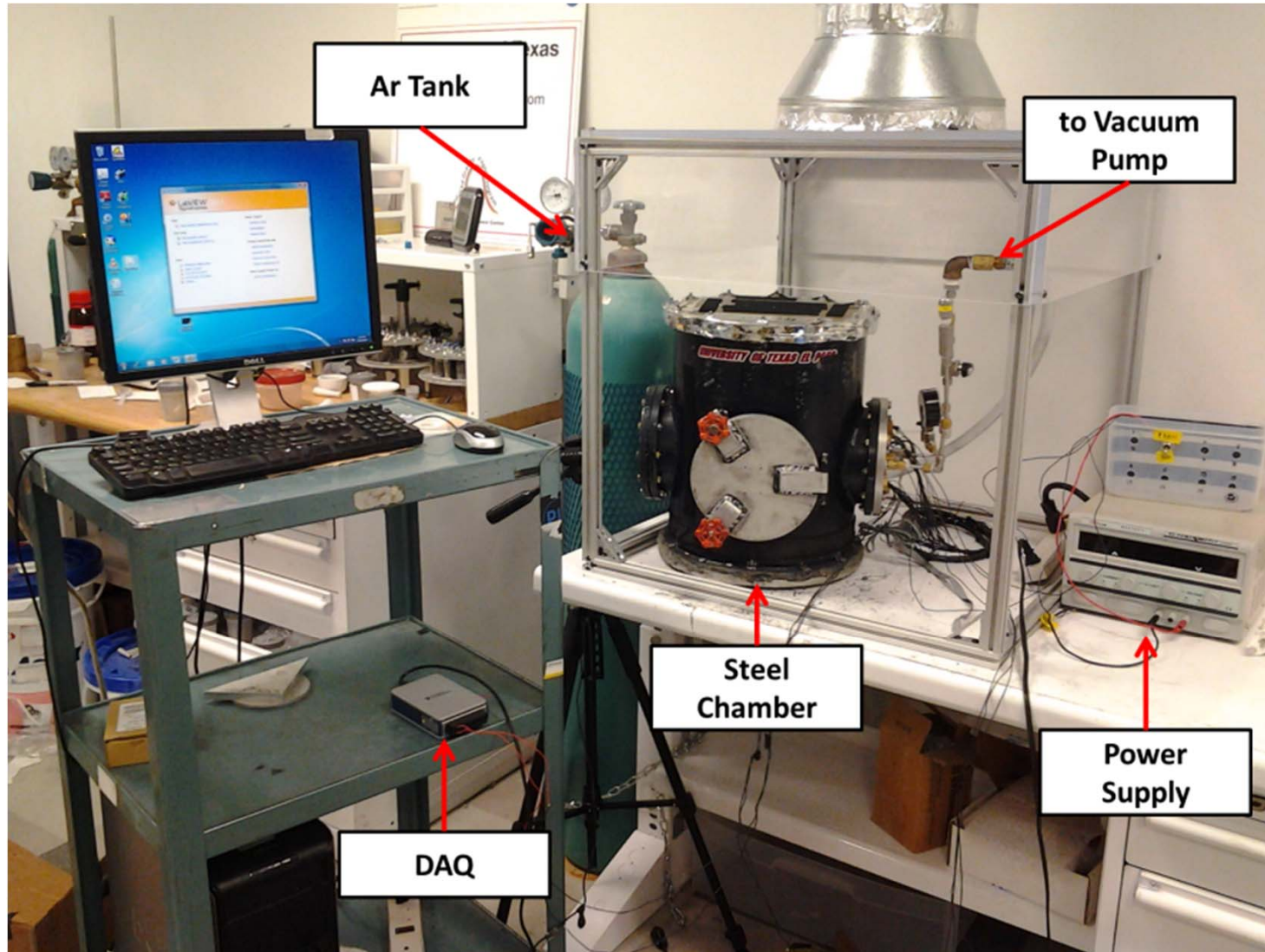


**Planetary ball mill  
(Fritsch  
Pulverisette 7)**

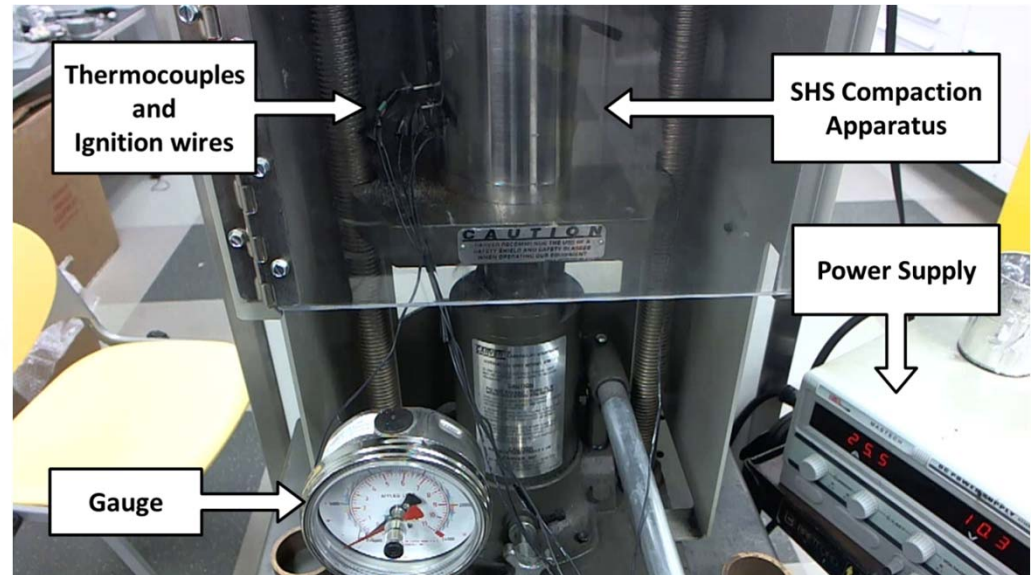
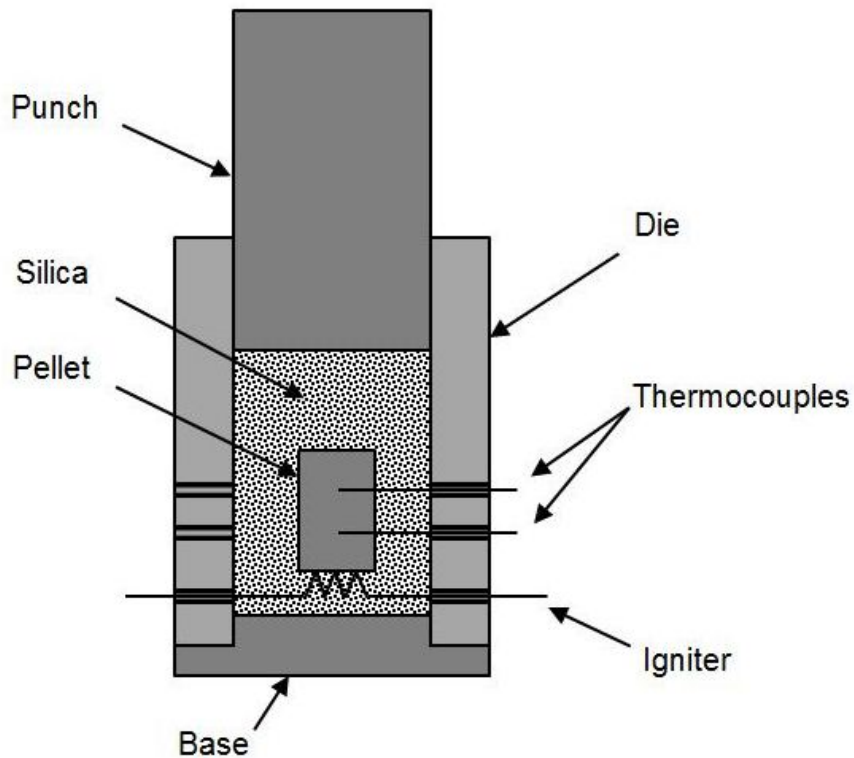
Pressing



# Setup for SHS in Argon



# SHS Compaction



- SHS products often have high porosity and low density.
- Stronger materials can be obtained using SHS compaction: quasi-isostatic pressure is applied through a pressure-transmitting medium immediately after combustion.

- Particle size distribution
- Thermogravimetric analysis
- Compression test
- XRD
- SEM



Particle size analyzer  
(Microtrac Bluewave)



Thermogravimetric analyzer  
(Netzsch TGA 209 F1 Iris)



Fatigue test machine  
(Instron 8801)



# **Synthesis of $\text{MoSi}_2$ - $\text{Mo}_5\text{Si}_3$ Composites**

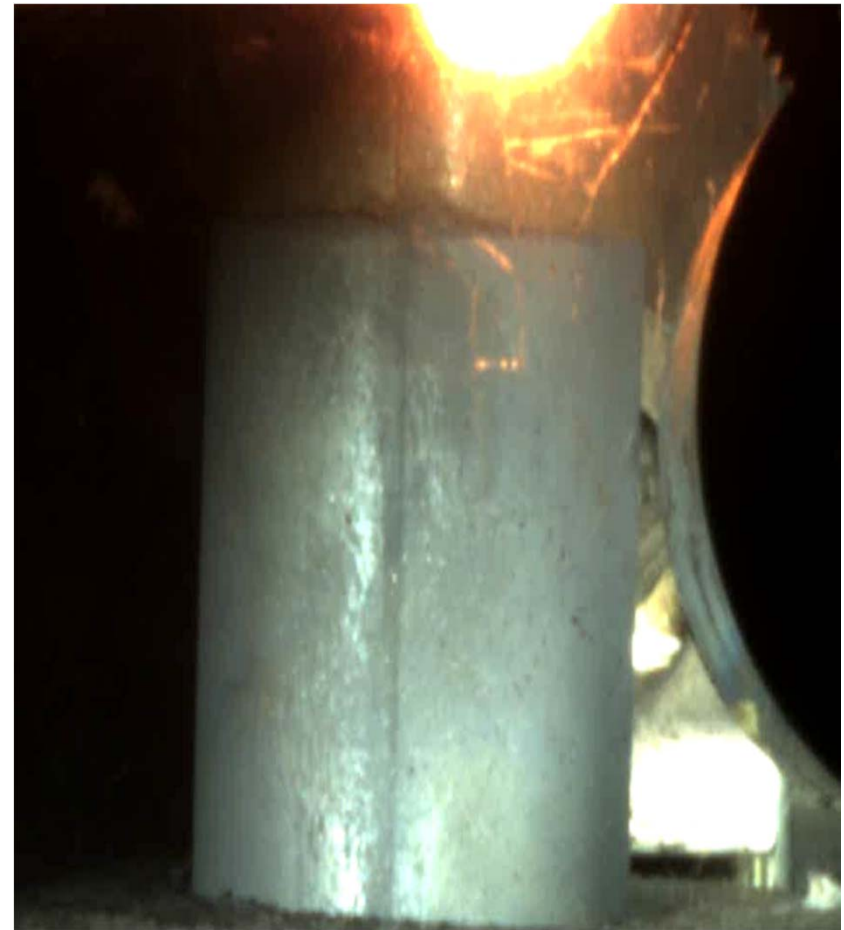


# MoSi<sub>2</sub>-Mo<sub>5</sub>Si<sub>3</sub> Composites



- Goal: to improve properties of MoSi<sub>2</sub>.
  - MoSi<sub>2</sub>: Good oxidation properties, poor mechanical properties
  - Mo<sub>5</sub>Si<sub>3</sub>: Good mechanical properties, poor oxidation properties
  
- Mixtures were prepared based on the desired product composition:
  - 10 – 50 vol% Mo<sub>5</sub>Si<sub>3</sub>
  - The balance MoSi<sub>2</sub>

- Mechanical activation enables combustion of mixtures with 10 – 40 vol%  $\text{Mo}_5\text{Si}_3$ .

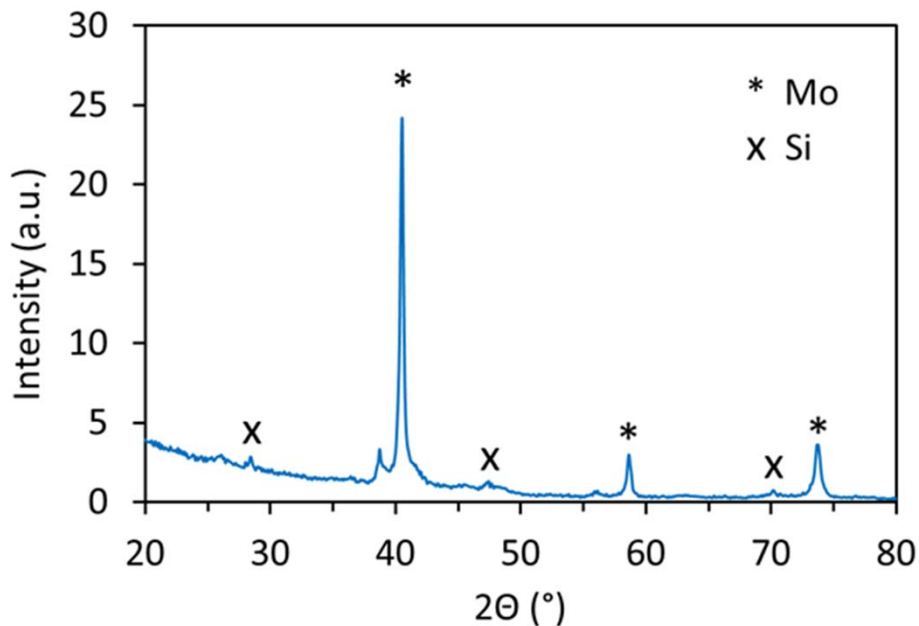


20 vol%  $\text{Mo}_5\text{Si}_3$

# XRD of $\text{MoSi}_2\text{-Mo}_5\text{Si}_3$ Product

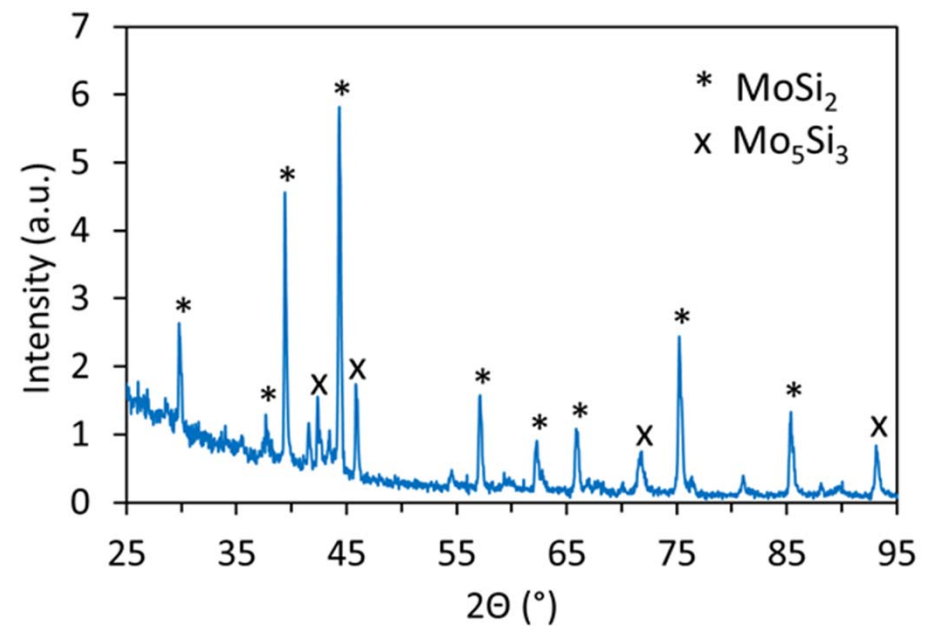
20 vol%  $\text{Mo}_5\text{Si}_3$

As-milled powder



- Initial components
- No  $\text{MoSi}_2$  or  $\text{Mo}_5\text{Si}_3$

Combustion products

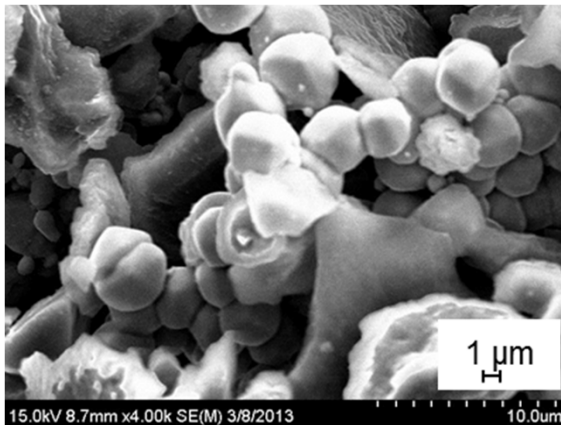


- No initial components
- $\text{MoSi}_2$  and  $\text{Mo}_5\text{Si}_3$

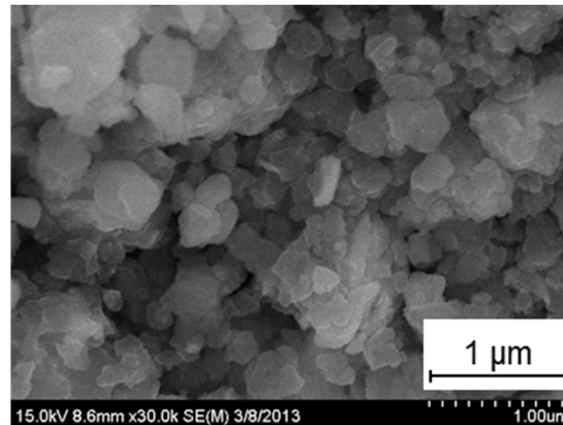


20 vol%  $\text{Mo}_5\text{Si}_3$

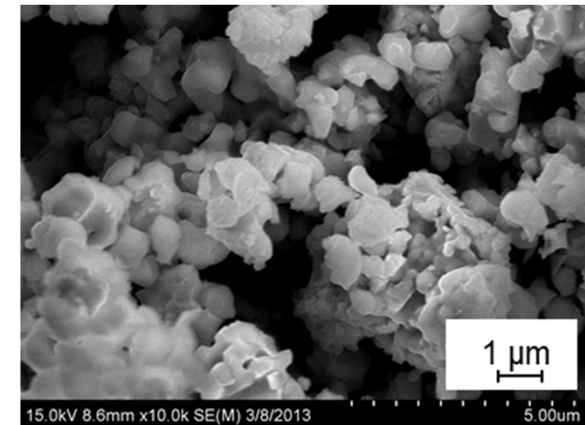
Before milling



After milling



After combustion



- Milling reduces particle size to submicron range.
- In the combustion products, most particles: 0.5 – 1  $\mu\text{m}$ , agglomerated, 3-D network structure.

# Effect of SHS Compaction

20 vol% Mo<sub>5</sub>Si<sub>3</sub>

After combustion in Ar

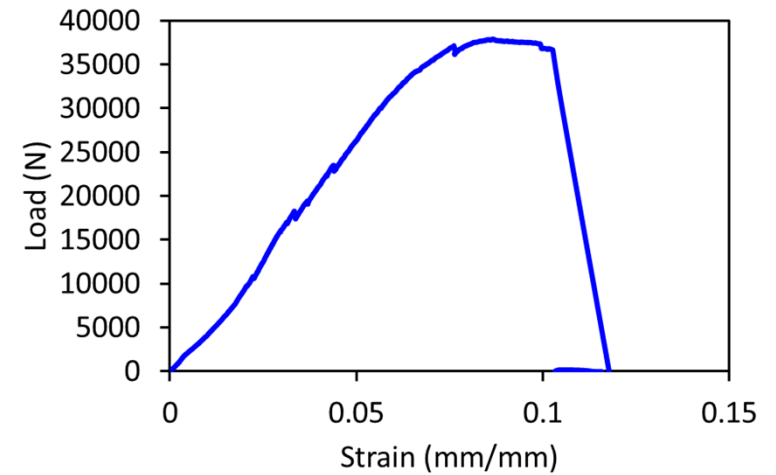
After SHS compaction



Relative density: 39%



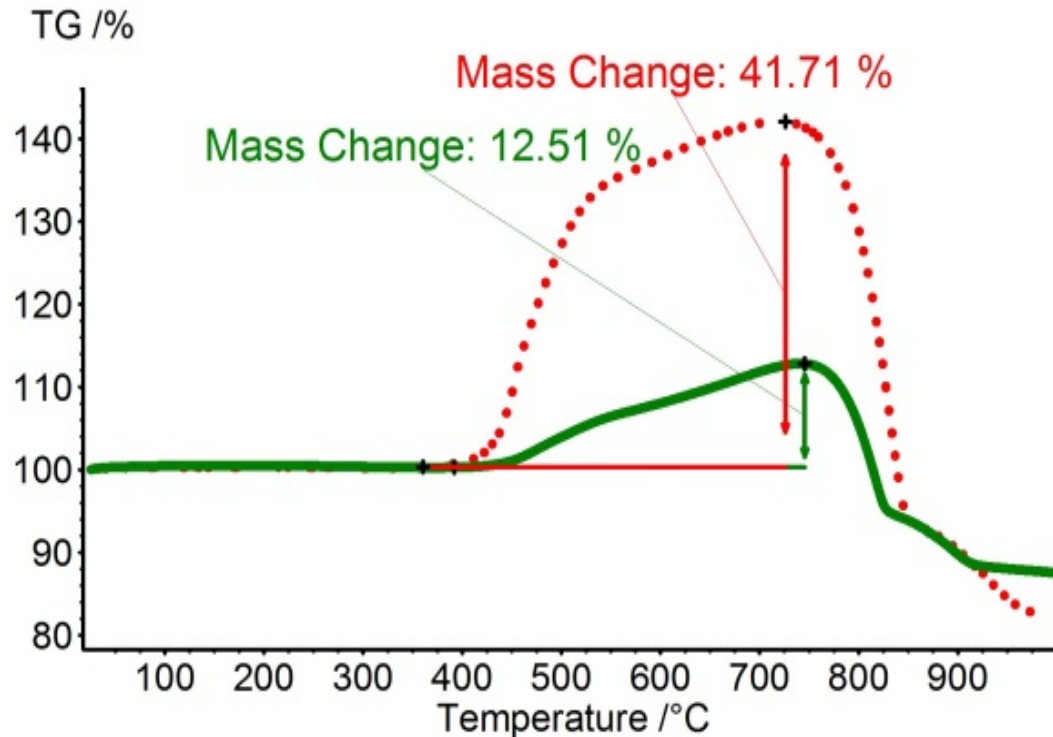
Relative density: 60%



Compressive load-strain curve of the SHS compaction product

- SHS compaction increased the relative density by 52%.
- Maximum compressive strength: 80 MPa

# Oxidation of $\text{MoSi}_2\text{-Mo}_5\text{Si}_3$ Product



- 20 vol%  $\text{Mo}_5\text{Si}_3$
- 20%  $\text{O}_2$ , 80% Ar
- 10°C/min
  
- SHS compaction decreases oxidation.

Mass change during oxidation of  $\text{MoSi}_2\text{-Mo}_5\text{Si}_3$  materials obtained by combustion in Ar (·····) and by SHS compaction (—).



# **Synthesis of $\text{Mo}_5\text{SiB}_2$ - MoB Materials**



# Mo<sub>5</sub>SiB<sub>2</sub> – MoB Materials



- Mo<sub>5</sub>SiB<sub>2</sub> has good mechanical and oxidation properties
  - A borosilicate glass layer protects from oxidation
- The addition of Mo and B ensures higher exothermicity through the reaction  $\text{Mo} + \text{B} \rightarrow \text{MoB}$
- Mixtures were prepared based on the desired product composition:
  - 10 – 67 vol% MoB
  - The balance Mo<sub>5</sub>SiB<sub>2</sub>

61.8% Mo  
11.8% Si  
26.4% B



A single hot spot, then  
two counter-  
propagating hot spots

61.1% Mo  
11.1% Si  
27.8% B



Three hot spots

59% Mo  
9% Si  
32% B

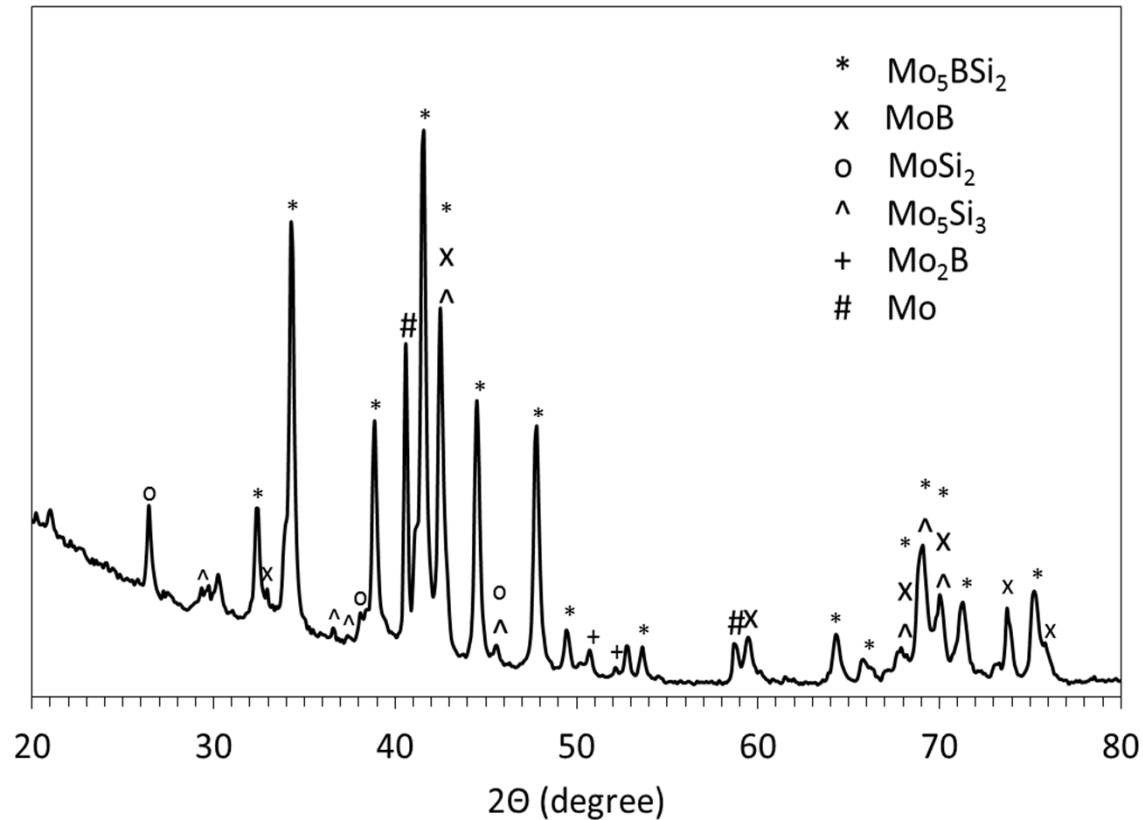


Multiple hot spots

- The results confirm the theory of *spin combustion*.

# XRD of Mo-Si-B Product

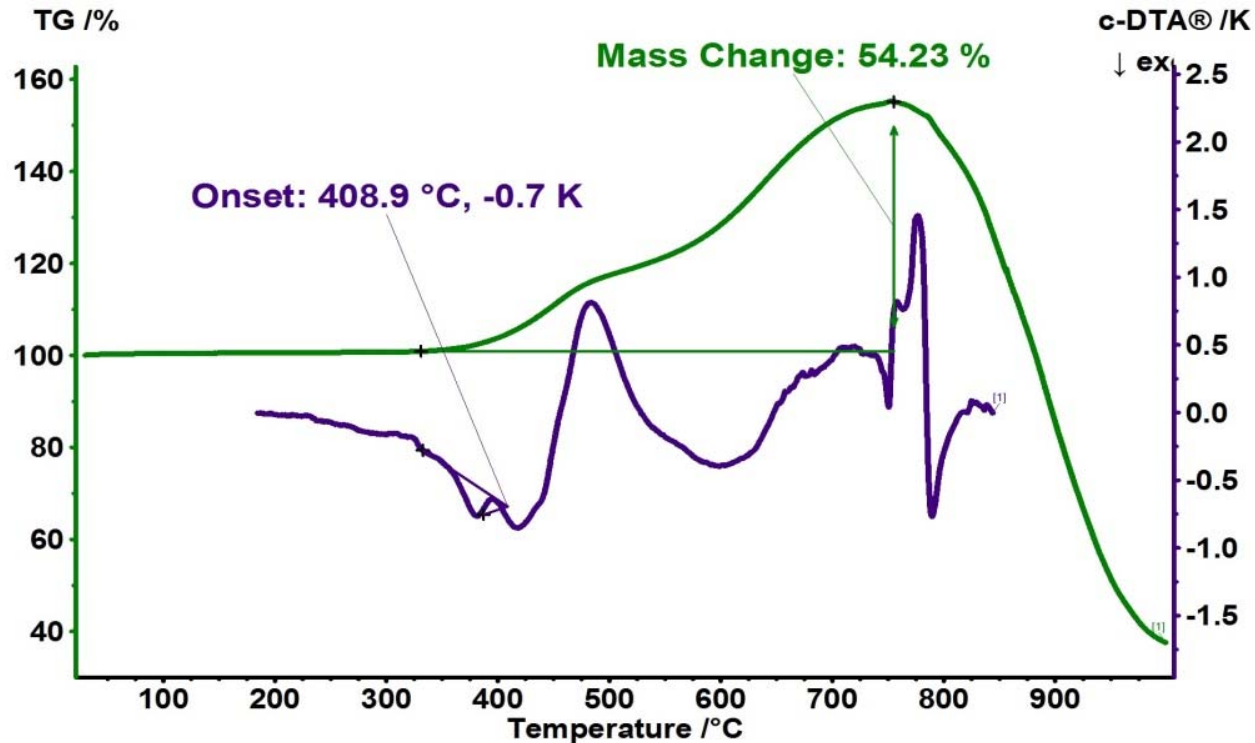
58.3% Mo, 8.3% Si, 33.4% B



- $\text{Mo}_5\text{SiB}_2$  and MoB were obtained, but Mo and some other phases were also present.

# Oxidation of Mo-Si-B Materials

58.3% Mo, 8.3% Si, 33.4% B



- Oxidation resistance of the obtained Mo-Si-B materials is poor.
- SHS compaction or other densification methods may improve the oxidation resistance.





# **Synthesis of $\text{Mo}_5\text{SiB}_2$ – $\text{TiB}_2$ Materials**



# Mo<sub>5</sub>SiB<sub>2</sub> – TiB<sub>2</sub> Materials

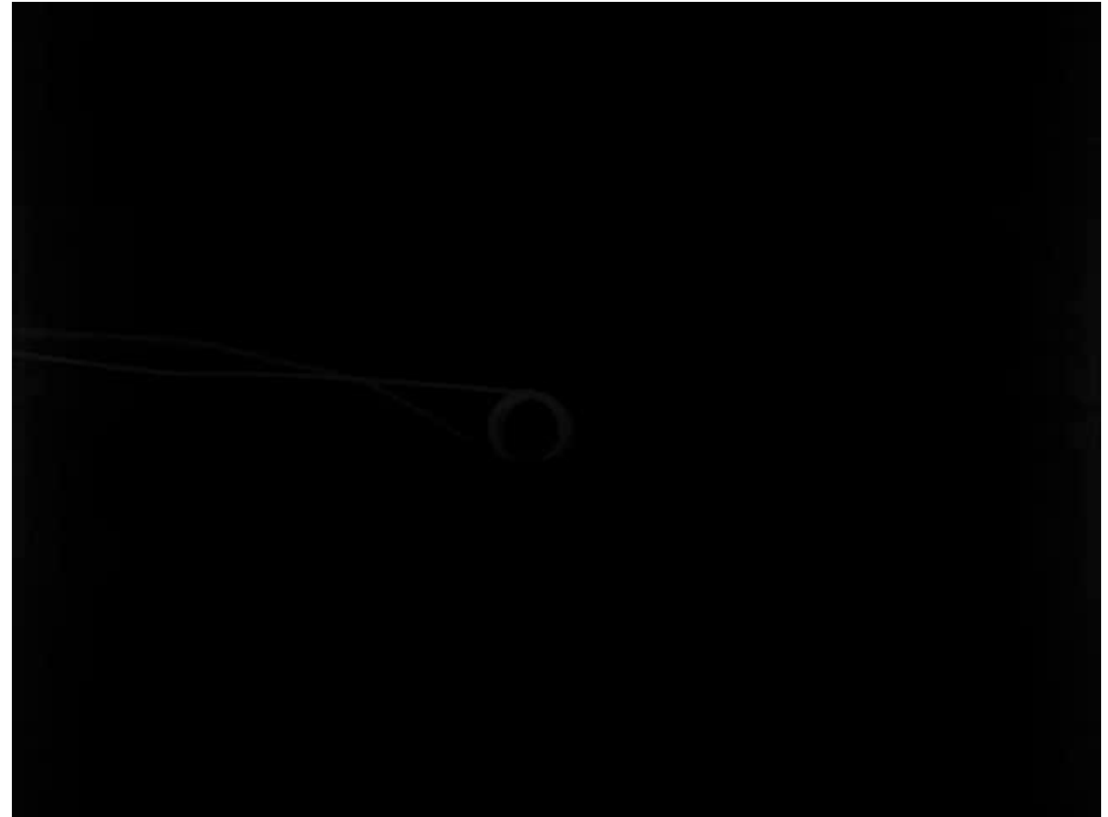


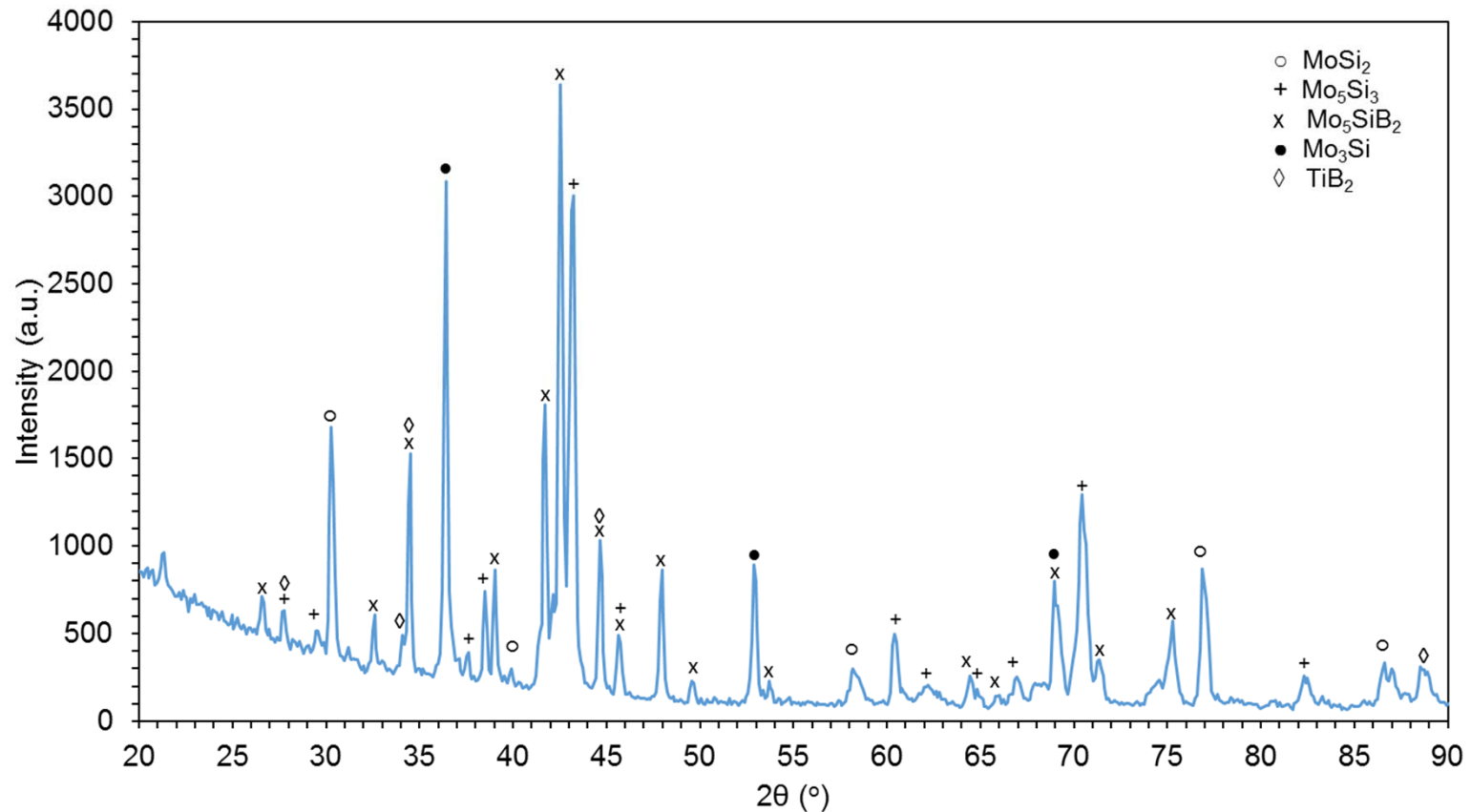
- Recently, TiC-added Mo-Si-B alloys were obtained in Japan using arc-melting in Ar.
  - TiC: low density (4.9 g/cm<sup>3</sup>) and excellent mechanical properties
  
- TiB<sub>2</sub> has similar properties.
  - Lower density: 4.5 g/cm<sup>3</sup>
  
- Highly exothermic reaction  $\text{Ti} + 2\text{B} \rightarrow \text{TiB}_2$  may enable self-sustained combustion of the mixture designed for fabricating Mo<sub>5</sub>SiB<sub>2</sub> – TiB<sub>2</sub> materials.

S. Miyamoto, K. Yoshimi, S.-H. Ha, T. Kaneko, J. Nakamura, T. Sato, K. Maruyama, R. Tu, and T. Goto, *Metall. Mater. Trans. A* 45A, 1112 (2014)

# Combustion of Mo-Si-B-Ti Mixture

- ❑ Mixtures were prepared based on the desired product composition:
  - 10 – 50 wt%  $\text{TiB}_2$
  - The balance  $\text{Mo}_5\text{SiB}_2$
- ❑ Video shows combustion of the mixture designed for 15%  $\text{TiB}_2$ .





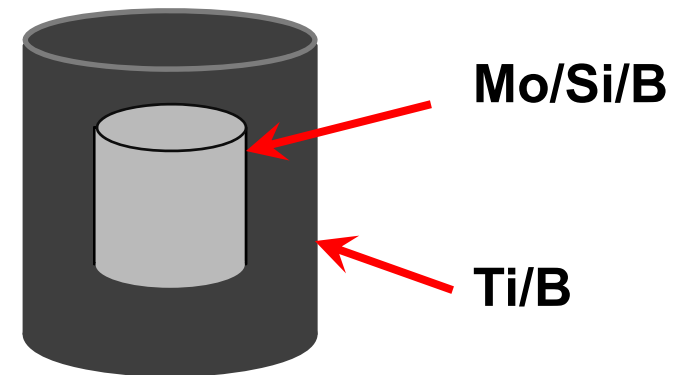
- Along with desired Mo<sub>5</sub>SiB<sub>2</sub> and TiB<sub>2</sub> phases, there are also Mo<sub>5</sub>Si<sub>3</sub>, Mo<sub>3</sub>Si, and MoSi<sub>2</sub>.



# **Synthesis of $\alpha$ -Mo/Mo<sub>5</sub>SiB<sub>2</sub>/Mo<sub>3</sub>Si (Mo-12Si-8.5B) Alloys**

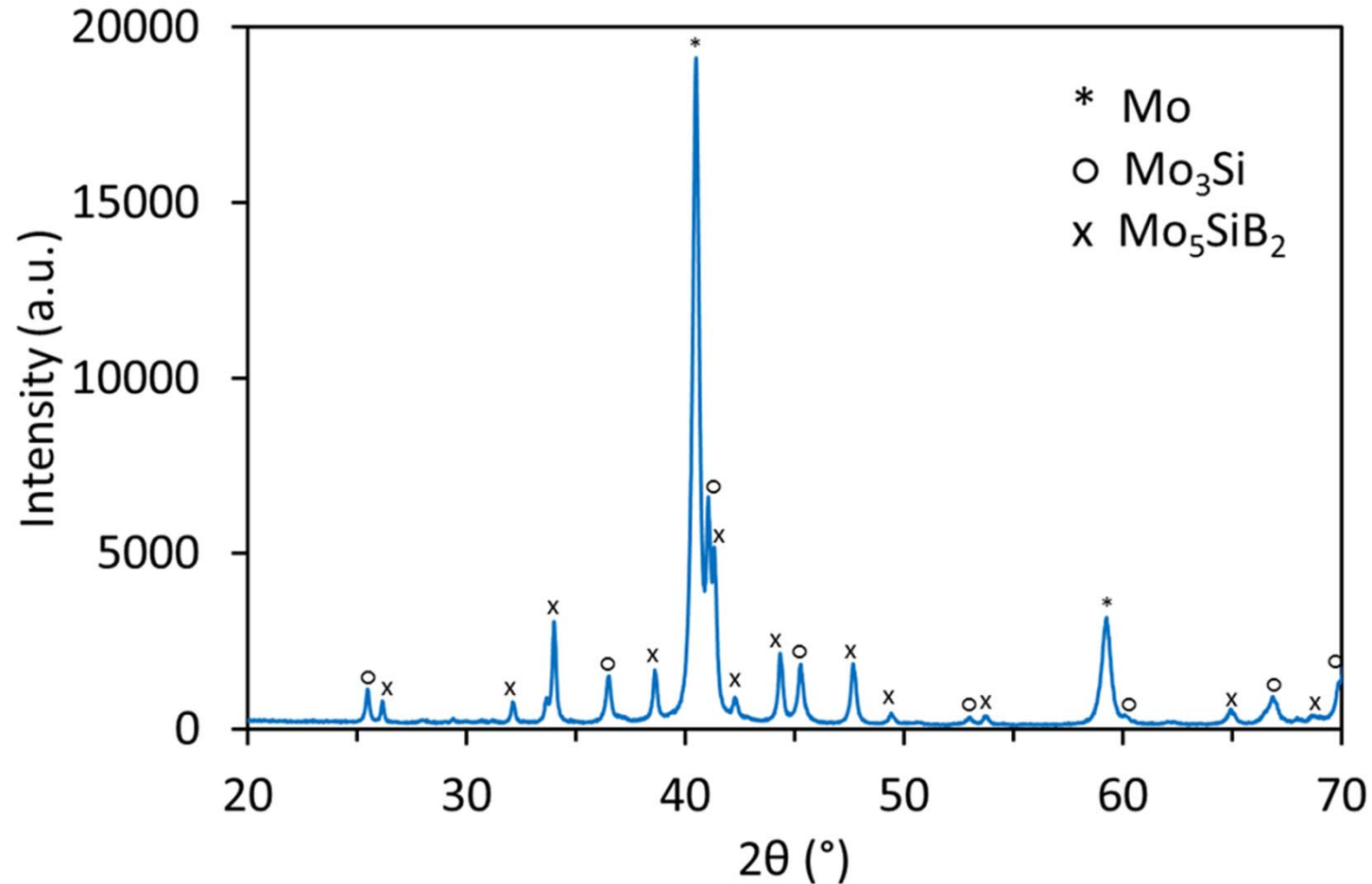
# $\alpha$ -Mo/Mo<sub>5</sub>SiB<sub>2</sub>/Mo<sub>3</sub>Si (Mo-12Si-8.5B) Alloys

- ❑  $\alpha$ -Mo/Mo<sub>5</sub>SiB<sub>2</sub>/Mo<sub>3</sub>Si alloys offer good mechanical properties and oxidation resistance.
- ❑ Exothermicity of Mo-12Si-8.5B mixtures for their fabrication is too low for a self-sustained combustion.
- ❑ The chemical oven technique enables combustion of these mixtures.



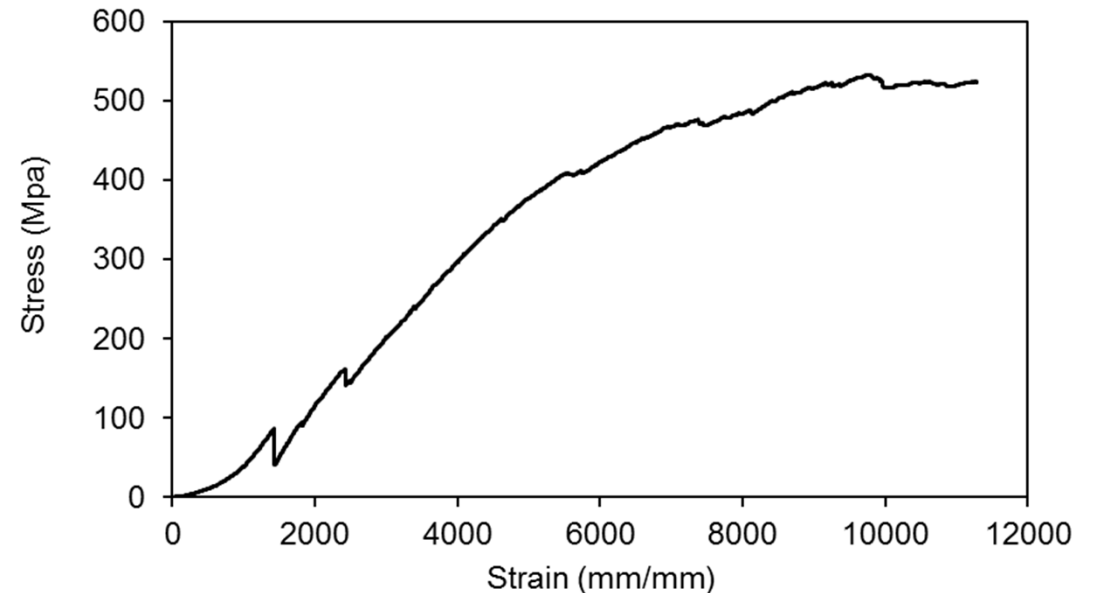
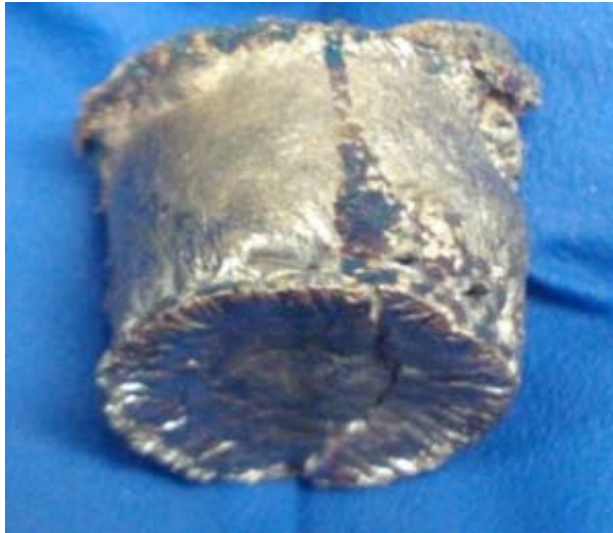
Chemical oven technique

# Mo-12Si-8.5B Product



- Three-phase  $\alpha$ -Mo/ $\text{Mo}_5\text{SiB}_2$ / $\text{Mo}_3\text{Si}$  alloy has been obtained.

# Density and Strength

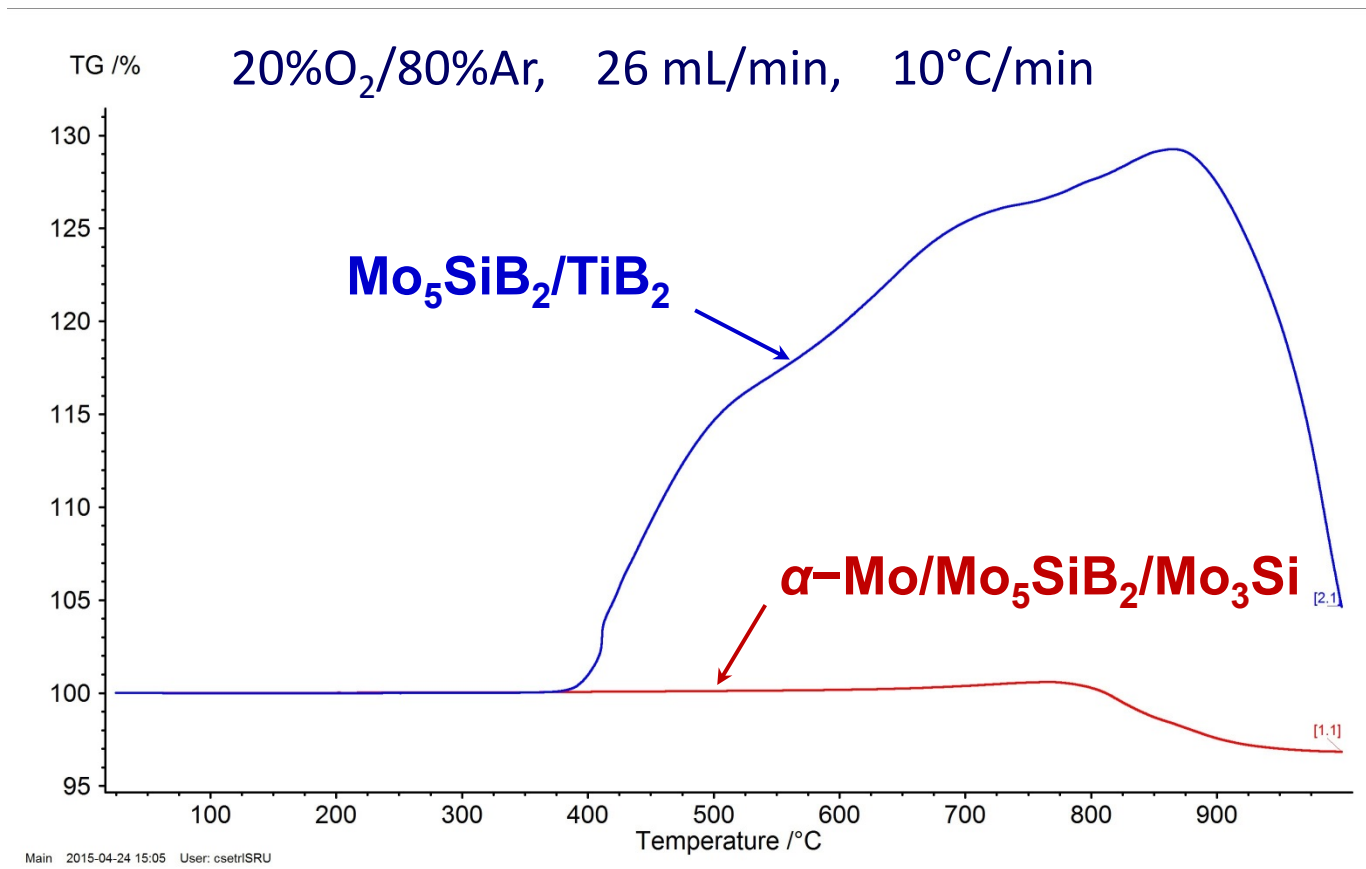


**Compression test**

- Density: 7.0 – 7.6 g/cm<sup>3</sup> (Mo: 10.2 g/cm<sup>3</sup>)
- Compressive strength: over 500 MPa (Mo: 400 Mpa)



# Oxidation of $\text{Mo}_5\text{SiB}_2/\text{TiB}_2$ and $\alpha\text{-Mo}/\text{Mo}_5\text{SiB}_2/\text{Mo}_3\text{Si}$ Materials



- $\alpha\text{-Mo}/\text{Mo}_5\text{SiB}_2/\text{Mo}_3\text{Si}$  obtained by the chemical oven technique exhibits good oxidation resistance at temperatures up to 1000°C.





# Conclusions

- Mechanically activated combustion synthesis has been used for fabricating the following materials:
  - $\text{MoSi}_2\text{-Mo}_5\text{Si}_3$
  - $\text{Mo}_5\text{SiB}_2\text{-MoB}$
  - $\text{Mo}_5\text{SiB}_2\text{-TiB}_2$
  - $\alpha\text{-Mo/Mo}_5\text{SiB}_2\text{/Mo}_3\text{Si (Mo-12Si-8.5B)}$
  
- Compaction after combustion significantly improves oxidation resistance of the obtained  $\text{MoSi}_2\text{-Mo}_5\text{Si}_3$  composites.
  
- The chemical oven technique has enabled fabrication of  $\alpha\text{-Mo/Mo}_5\text{SiB}_2\text{/Mo}_3\text{Si (Mo-12Si-8.5B)}$  alloys, which offer favorable combinations of high-temperature mechanical properties and oxidation resistance.



# Current and Future Work



- ❑ Current efforts are focused on improving the quality of  $\text{Mo}_5\text{SiB}_2\text{-TiB}_2$  materials using the chemical oven technique
  
- ❑ Mechanically activated combustion synthesis could be used for fabricating various other materials that are of interest for DOE, for example:
  - Electrode materials for MHD direct power extraction
  - Thermoelectric materials for energy harvesting



# Publications and Presentations



Esparza, A.A., Alam, M.S., and Shafirovich, E., 9<sup>th</sup> U.S. National Combustion Meeting, May 17-20, 2015, Cincinnati, OH, 2B06.

Alam, M.S., Esparza, A.A., and Shafirovich, E., *MRS Proceedings*, 2015, 1760, mrsf14-1760-yy03-02, DOI:10.1557/opl.2014.964 (presented at the 2014 Fall Meeting of the Materials Research Society).

Alam, M.S., and Shafirovich, E., *Proceedings of the Combustion Institute*, Vol. 35, 2015, No. 2, pp. 2275–2281, DOI:10.1016/j.proci.2014.05.019 (presented at the 35<sup>th</sup> International Symposium on Combustion).

Alam, M.S., and Shafirovich, E., Spring Technical Meeting of the Central States Section of the Combustion Institute, March 16-18, 2014, Tulsa, OK, B-601 (*First Place for Technical Merit Award in the Combustion Art Competition*).

Alam, M.S., and Shafirovich, E., 12<sup>th</sup> International Symposium on Self-Propagating High Temperature Synthesis, 21 - 24 October 2013, South Padre Island, TX, p. 92.

Alam, M.S., and Shafirovich, E., 8<sup>th</sup> U.S. National Combustion Meeting, May 19-22, 2013, Park City, UT, 070HE-0301.



# Acknowledgments



## □ Support

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## □ Students involved

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**Thank you for your attention!**