

## Lab and Bench-Scale Testing of CO<sub>2</sub> Capture Using Physical Sorbents

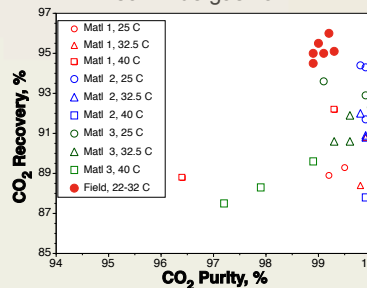
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### Executive Summary

- Physical sorption to produce dry CO<sub>2</sub> at high purity (>98%) and high recovery (>90%) from the flue gas
- An impurities removal stage followed by a CO<sub>2</sub> adsorption stage
  - Both the stages have been extensively tested in the lab and in the field (up to 100 scfm flue gas flow) for over 6 years
- Potential for more than 40% reduction in the capital and more than 40% reduction in parasitic power for CO<sub>2</sub> capture compared to MEA
- The estimated total energy required, excluding compression, is 450-500 kcal/kg of CO<sub>2</sub>
  - Potential to provide CO<sub>2</sub> at a cost (<\$40/ton) and quality (<1 ppm H<sub>2</sub>O, <1 ppm SO<sub>x</sub>, <10 ppm O<sub>2</sub>) suitable for EOR applications
- The DOE projects addressed various process risks and scale up issues through lab and field testing, process simulation, and techno-economic evaluation

### Lab and Field Testing Summary

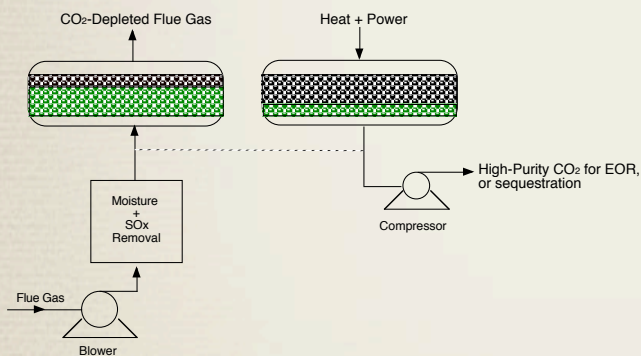
- Lab testing for initial adsorbent screening
  - >90% CO<sub>2</sub> recovery, >99% CO<sub>2</sub> purity, 7-9 wt% net CO<sub>2</sub> capacity (~15% CO<sub>2</sub> in the feed)
  - Same or higher CO<sub>2</sub> purity and recovery as reactive sorbents
- The bench unit testing at NRG's Indian River, DE plant
  - The feed to bench unit saturated at 60°C, about 50 ppm SO<sub>2</sub>, and 10-12% CO<sub>2</sub>
  - Eight weeks of testing with 22-32°C feed temperature, 80-100 scfm flue gas flow



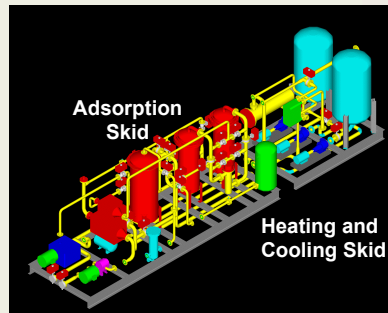
Field performance better than the lab performance

- 8-10.5 wt% net CO<sub>2</sub> capacity in the field
- >94% CO<sub>2</sub> recovery, 98.5-99.5% CO<sub>2</sub> purity

### Process Overview

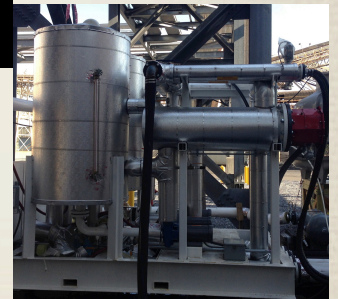


- Flue gas pretreatment to remove moisture and SO<sub>x</sub> to < 1 ppm each, adsorption at 25-40°C and regeneration at ~100°C
- High purity CO<sub>2</sub> (>98% purity, <10 ppm O<sub>2</sub>) at >90% recovery
- The key innovation is **the combination** of process and materials (physical sorbents) that provides performance similar to or better than reactive systems with a much lower energy penalty



Skid Layout

Heating and Cooling Skid



Adsorption Skid

## Techno-Economic Analysis (550 MMW SCPC Plant, 2012 Basis)

Estimated Capital Cost	\$388 MM
Power consumption including compression	99 MW
Steam cost with capture (per 1,000 lbs)*	\$8.60 <b>(+47%)</b>
Electricity cost with capture*	\$0.095/kWh <b>(+48%)</b>
CO <sub>2</sub> production rate, million tons/yr	3.5
<b>CO<sub>2</sub> Recovery Cost**</b>	<b>\$46/ton</b>

\*Based on the DE-FOA-0000403 guidelines

\*\*85% plant utilization factor. Includes capital charge, power, maintenance, and transportation cost

### Comparison with MEA (2011 Cost Basis, Case 12)

- 44% lower capital compared to MEA excluding the CO<sub>2</sub> compressor
- 38% lower capital including the CO<sub>2</sub> compressor
- 18% parasitic power load compared to 28% parasitic power load for MEA

## Potential Approaches for Further Cost Reduction

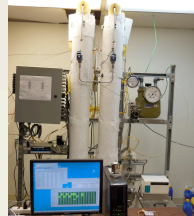
- The DOE bench scale project used particulate adsorbents which can be
  - Subject to fluidization, attrition and higher pressure drop
- The DOE SBIR project is looking at structured sorbents which are
  - Not subject to fluidization, attrition and have a significantly lower pressure drop
- The key goals for the project are
  - Adsorption isotherms, process simulation and engineering design for CO<sub>2</sub> capture based on monoliths
  - Quantify benefits over MEA as well as with InnoSeptra's process using particulate adsorbents with a detailed techno-economic analysis
- The key benefits include
  - Lower pressure drop, faster cycling and lower skid and shipping cost can overcome the higher cost of structured adsorbents
  - Potential for up to 10% reduction in CO<sub>2</sub> capture cost compared to particulate adsorbents

## Key Tasks for the Structured Adsorbent Project

- Vendor fabrication of sorbent modules
- Equipment modification and moisture removal testing
- Sorption isotherms and CO<sub>2</sub> capture testing
- Process simulation and engineering design
- Techno-economic analysis

## Accomplishments So Far

- Structured sorbents for flue gas purification and CO<sub>2</sub> capture fabricated
- Breakthrough capacities determined in bed sizes up to 3" diameter



**Lab Unit for  
Adsorbent Testing**

- Capacities comparable to particulate adsorbents, much lower pressure drop (<1/5<sup>th</sup>)
- Process modeling and preliminary TEA indicating potential for CO<sub>2</sub> capture cost reduction

## Overall Accomplishments

- The InnoSeptra CO<sub>2</sub> capture process combines several innovative features to reduce the capital cost and parasitic power for CO<sub>2</sub> capture
- It is possible to obtain very high recovery (>90%), and high purity (>99%) CO<sub>2</sub> with physical sorbents while meeting the EOR/sequestration oxygen specification
  - $\Delta H_{ads} < 200$  Kcal/kg, parasitic power <500 Kcal/kg
  - High net CO<sub>2</sub> capacity (>8 wt%)
  - <1 ppm each of SO<sub>x</sub> and H<sub>2</sub>O, 10-30 ppm O<sub>2</sub> in product CO<sub>2</sub>
- The capital cost and parasitic power estimates based on a detailed component level analysis indicate that we are close to DOE's LCOE target (<35% increase) and the CO<sub>2</sub> cost target (<\$40/ton)
- Possible to obtain a CO<sub>2</sub> product suitable for EOR and sequestration
- Potential for further cost reduction through the use of structured sorbents