

Bench-Scale Development and Testing of Rapid PSA for CO₂ Capture

James A. Ritter & The Team



U N I V E R S I T Y O F
SOUTH CAROLINA



catacelTM

Driving Reaction Technology



Johnson Matthey
Process Technologies

Battelle

The Business of Innovation

GRACE

Talent | Technology | TrustTM

**2015 NETL CO₂ Capture Technology Meeting
Pittsburgh, PA, June 25, 2015**

Overall Project Objectives

- ❖ design, develop and demonstrate a bench-scale process for the efficient and cost effective separation of CO₂ from flue gas using Pressure Swing Adsorption (PSA)
- ❖ goal to reduce energy consumption, capital costs, and environmental burdens with novel PSA cycle/flow sheet designs
- ❖ applicable to both large (500-1000 MW) and small (5-50 MW) capacity power plants, and industries with 10 to 100 times less CO₂ production

Process simulations and experiments; structured adsorbent material development, CFDs and experiments; and complete flow sheet analyses being used for demonstrating and validating the concepts.

The Team

thin film
materials
development and
characterization

investigation

Grace
(Ehrlich)

USC
(Ritter &
Ebner)

Catacel
(Cirjak)

materials
characterization,
and process
modeling and
experimentation

specification

Battelle
(Saunders &
Swickrath)

technology
development and
process integration

validation

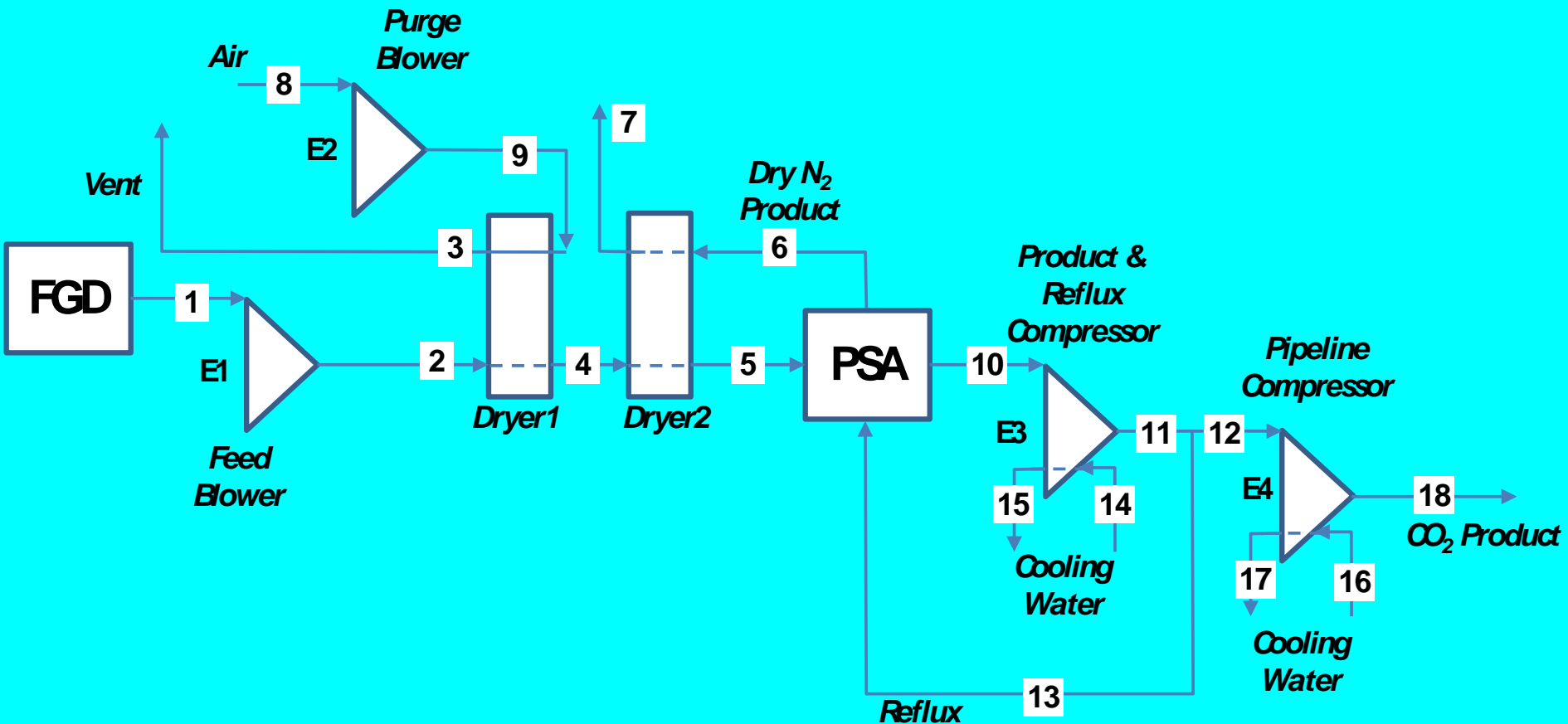
Key PSA Technology Challenge

....to develop a structured adsorbent around an efficient PSA cycle that exhibits a high enough packing density to allow the fastest possible cycling rate (→ smallest possible beds), while improving pressure drop and mass transfer and eliminating attrition issues....



Where have we been?

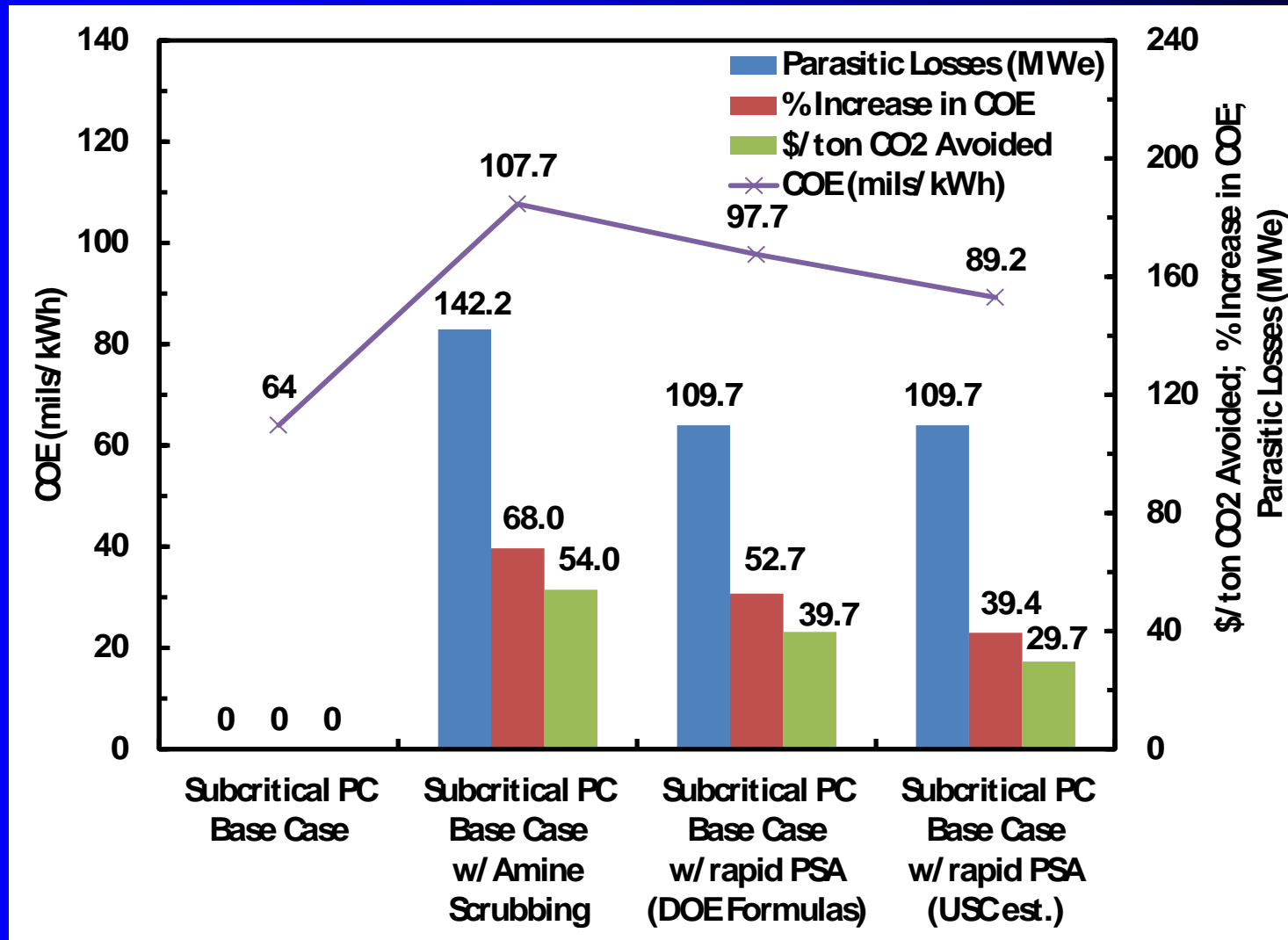
USC Rapid PSA Process Flow Sheet*



* USC Provisional Patent Filed

Preliminary Technical and Economic Feasibility Study

Overall Outcome





**Where are we now in Budget
Period (BP) 2?**

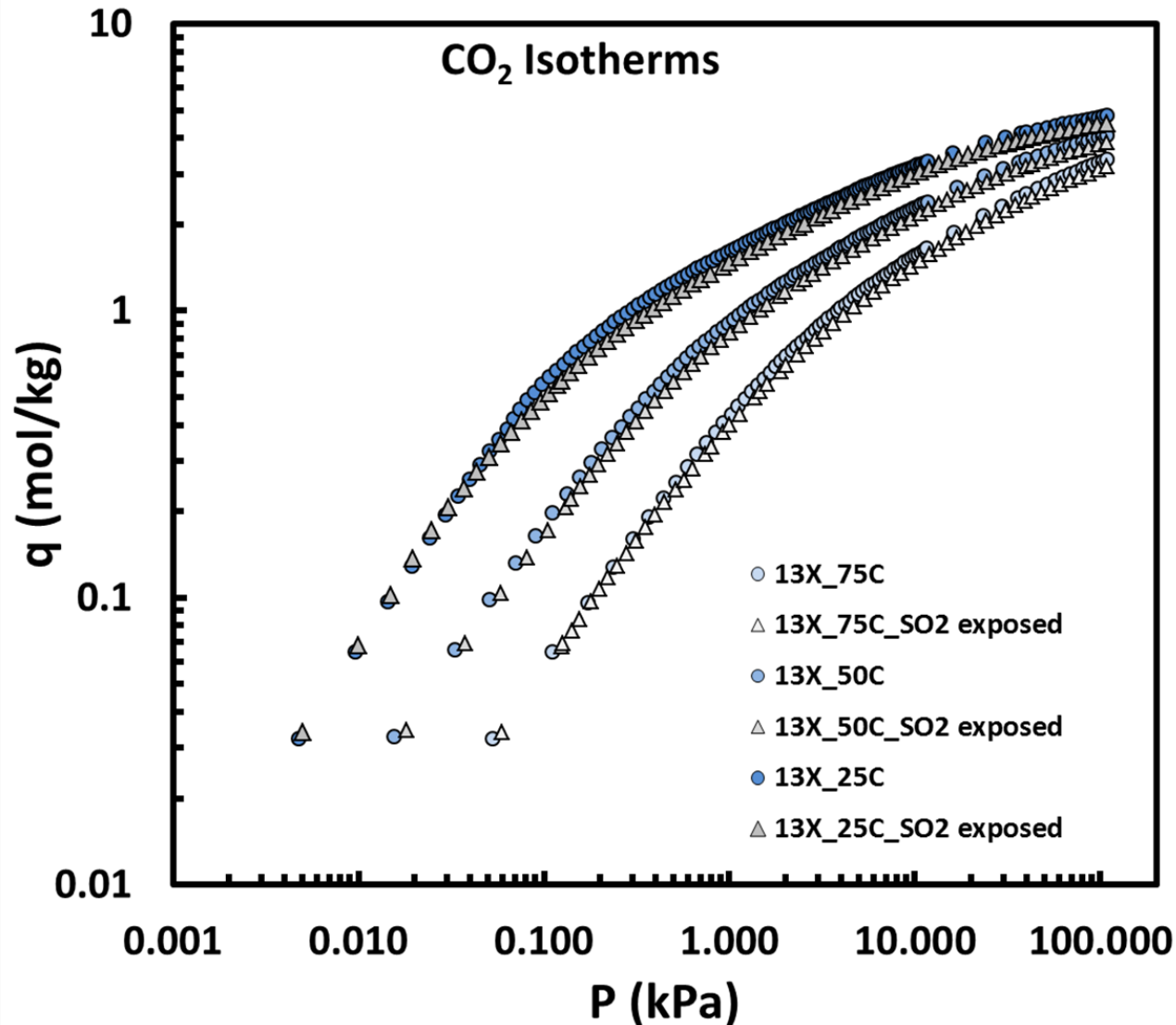
Significant Outcomes so far in BP 2

- developed test procedures for measuring effects of trace levels of both SO_x and NO_x on zeolites and silica gels; *some preliminary results have been obtained*
- completed start-up and trouble-shooting of 1-bed bench scale PSA apparatus; *system now capable of mimicking all cycle steps of multi-bed PSA process*
- completed *preliminary* CO_2 - N_2 cycling with 3 mm zeolite beads in larger 1-bed bench scale PSA apparatus; *results showing very good separation/recovery at reasonable throughput*
- fabricated three, 6 inch Catacel cores coated with 50 μm thick layers of zeolite crystals, achieving 240 kg/m^3 bed density; *awaiting testing in larger 1-bed bench scale PSA system*
- fabricated two, 6 inch Catacel cores, one uncoated and one coated with 50 μm thick layers of zeolite crystals; *awaiting testing in VFR apparatus to determine mass transfer rates*

Significant Outcomes so far in BP 2

- completed construction of multi-bed bench scale PSA apparatus; currently undergoing start-up, troubleshooting and testing with 3 mm zeolite beads
- developed parallel channel structured adsorbent pressure drop correlation (PD) for use in DAPS; currently being used with DAPS to simulate full scale PSA process
- completed CFD modeling showing when plug flow (packed bed) conditions prevail in parallel channel structured adsorbent (PCSA); simpler 1-D packed bed models can now be used to study PCSAs in DAPS with modified PD correlation
- completed first phase of CFD modeling, revealing use of slower 2-D and 3-D models to train much faster 1-D models; currently being used to determine optimum Catacel core structure

Equilibrium Adsorption Isotherms for CO_2 on 13X Zeolite



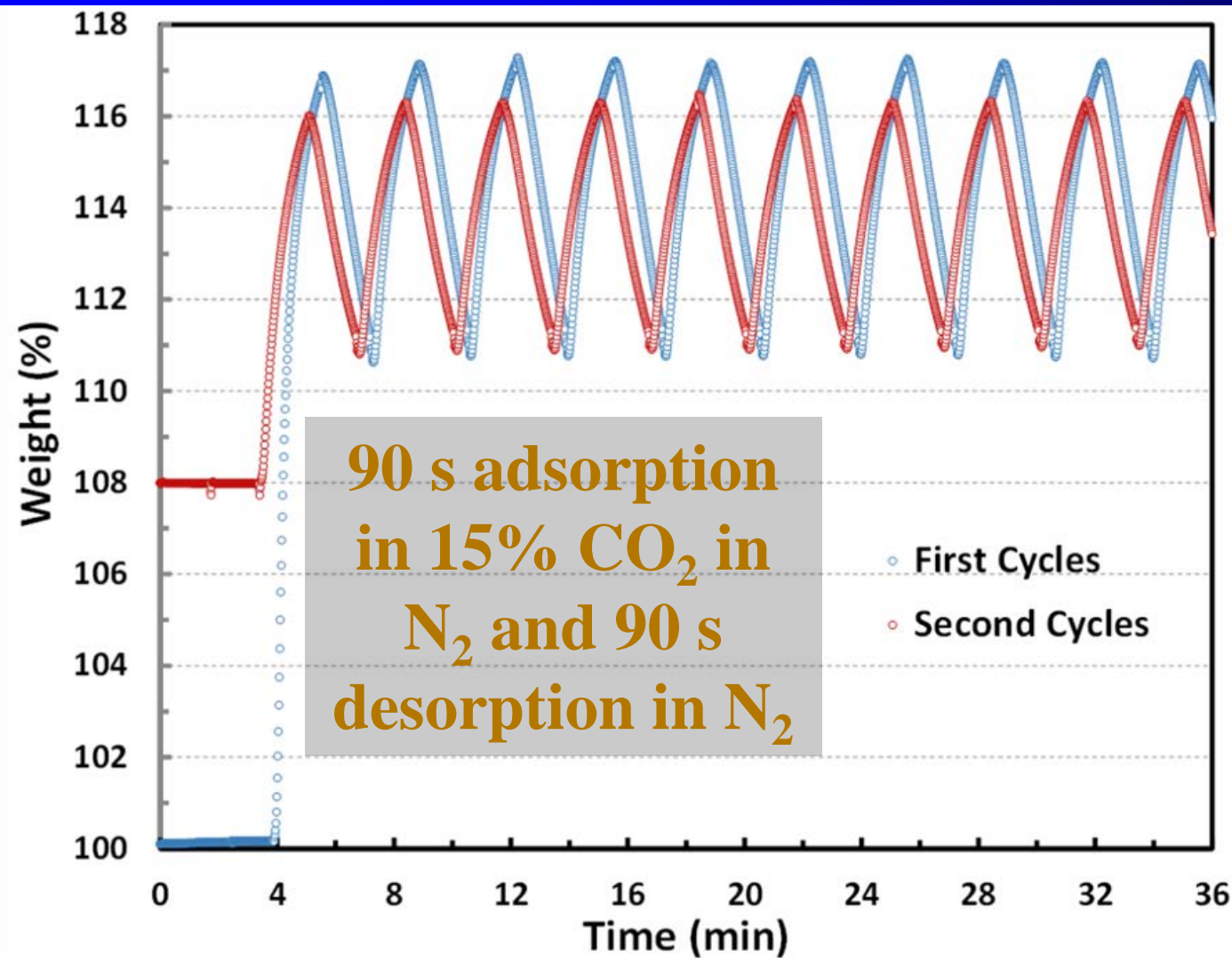
Effect of
Exposure to
500 ppm of
 SO_2 in N_2 with
Subsequent T
Regeneration

5% to 10%
Decrease in
Capacity Realized
Especially at High
Loadings

Working Capacity (wt%) of CO₂ on 13X Zeolite

Effect of SO₂ and NO_x Exposure

TGA CO₂ Cycling Tests at 70 °C and 1 atm



sample
exposed to
2.5 hr of 42
ppm of SO₂
in N₂
between sets
of cycling
tests – no
regeneration

Working Capacity (wt%) of CO₂ on 13X Zeolite

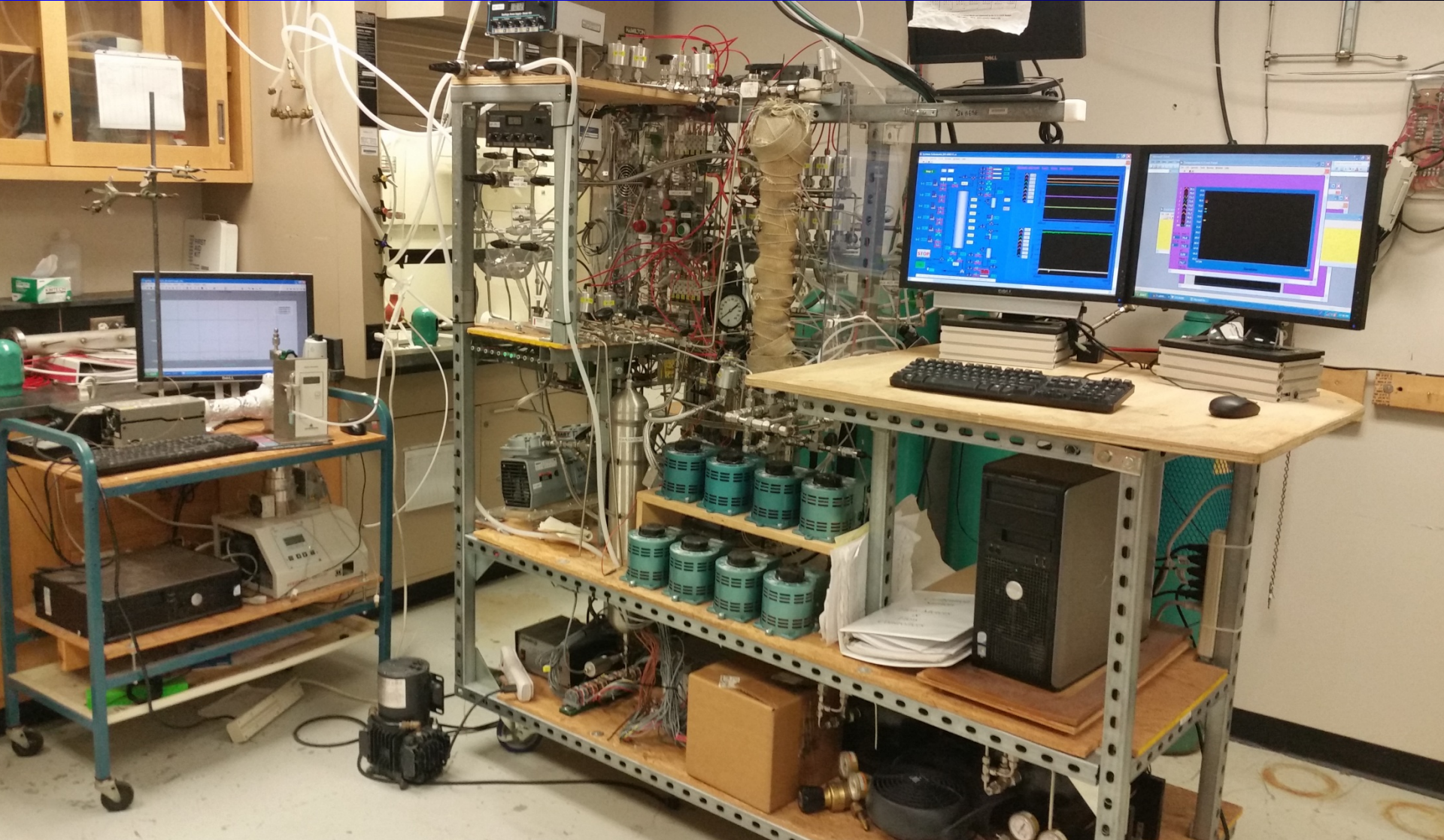
Effect of SO₂ and NO_x Exposure

Gas	Before Exposure	After Exposure	% Decrease
He*	7.05	5.87	16.7
SO ₂ (42 ppm)	6.56	5.18	21.0
SO ₂ (500 ppm)	6.88	5.66	17.7
NO ₂ (74 ppm)	6.44	5.43	15.7

* only helium run, so sample not exposed to any SO₂ or NO_x – implies decrease in all cases, since similar, most likely due to trace H₂O vapor leaking into TGA

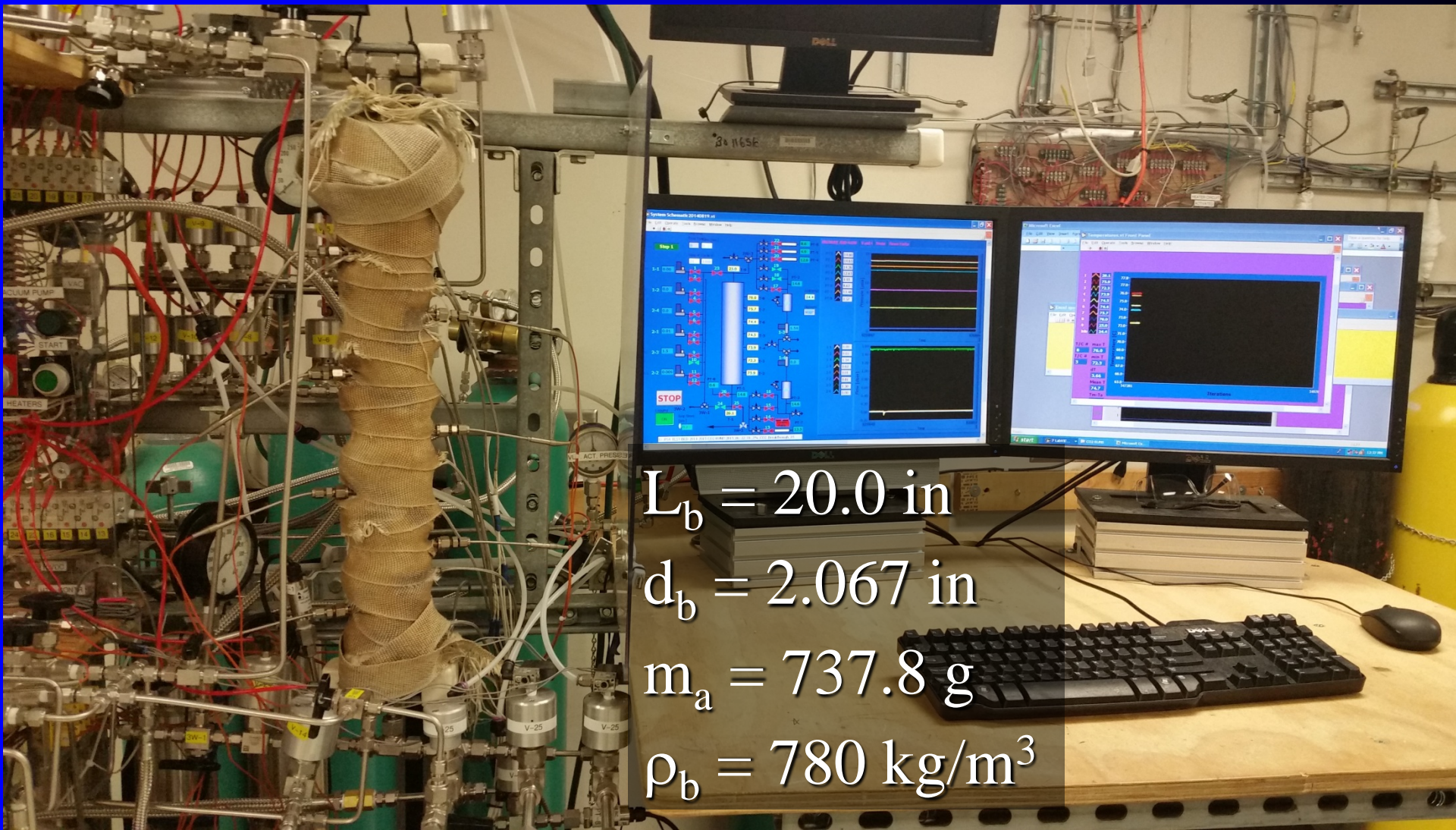
1-Bed PSA System

Rapid Complex PSA Cycle Schedule Analysis



1-Bed PSA System

Rapid Complex PSA Cycle Schedule Analysis



$$L_b = 20.0 \text{ in}$$

$$d_b = 2.067 \text{ in}$$

$$m_a = 737.8 \text{ g}$$

$$\rho_b = 780 \text{ kg/m}^3$$

1-Bed PSA System

Preliminary Results with Zeolite Beads

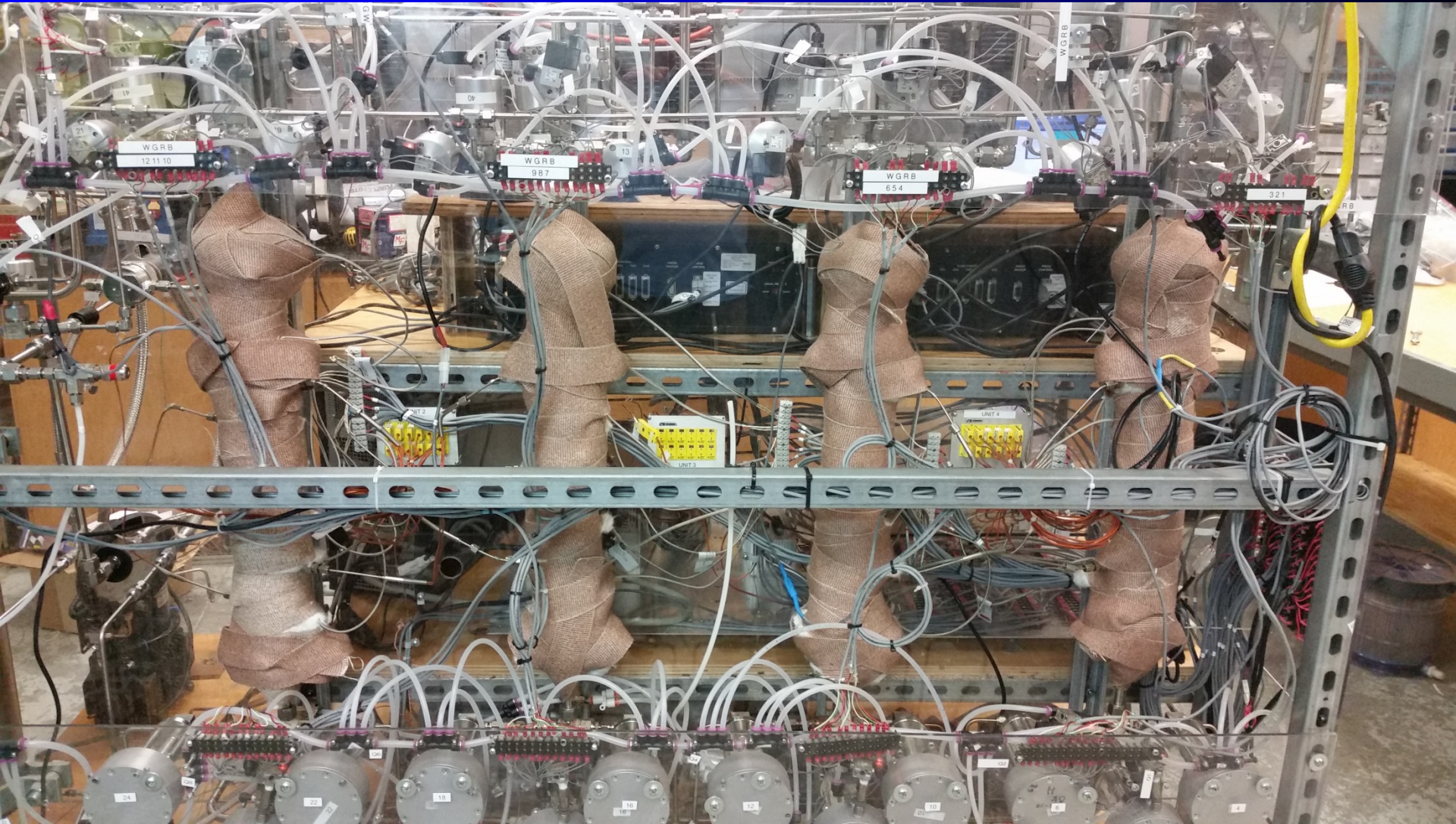
Exp No.	Feed Flow Rate [SLPM]	Cycle Time [sec]	Feed Throughput [L(STP)/hr/kg]
1	12.97	720	351.60
2	14.26	720	386.55
3	7.13	1440	193.32
4	7.85	1440	212.67

15 vol% CO₂ in N₂
Fed at 120 kPa with
Column at 70 °C
and Regeneration
by Vacuum with
 $P_{\text{low}} = 5 \text{ kPa}$

Exp No.	HP CO ₂ Pur (%)	HP CO ₂ Rec (%)	LP N ₂ Pur (%)	LP N ₂ Rec (%)
1	88.31	92.16	98.18	96.53
2	89.82	89.98	97.40	97.42
3	88.50	94.50	98.46	97.50
4	89.37	93.08	98.18	97.35

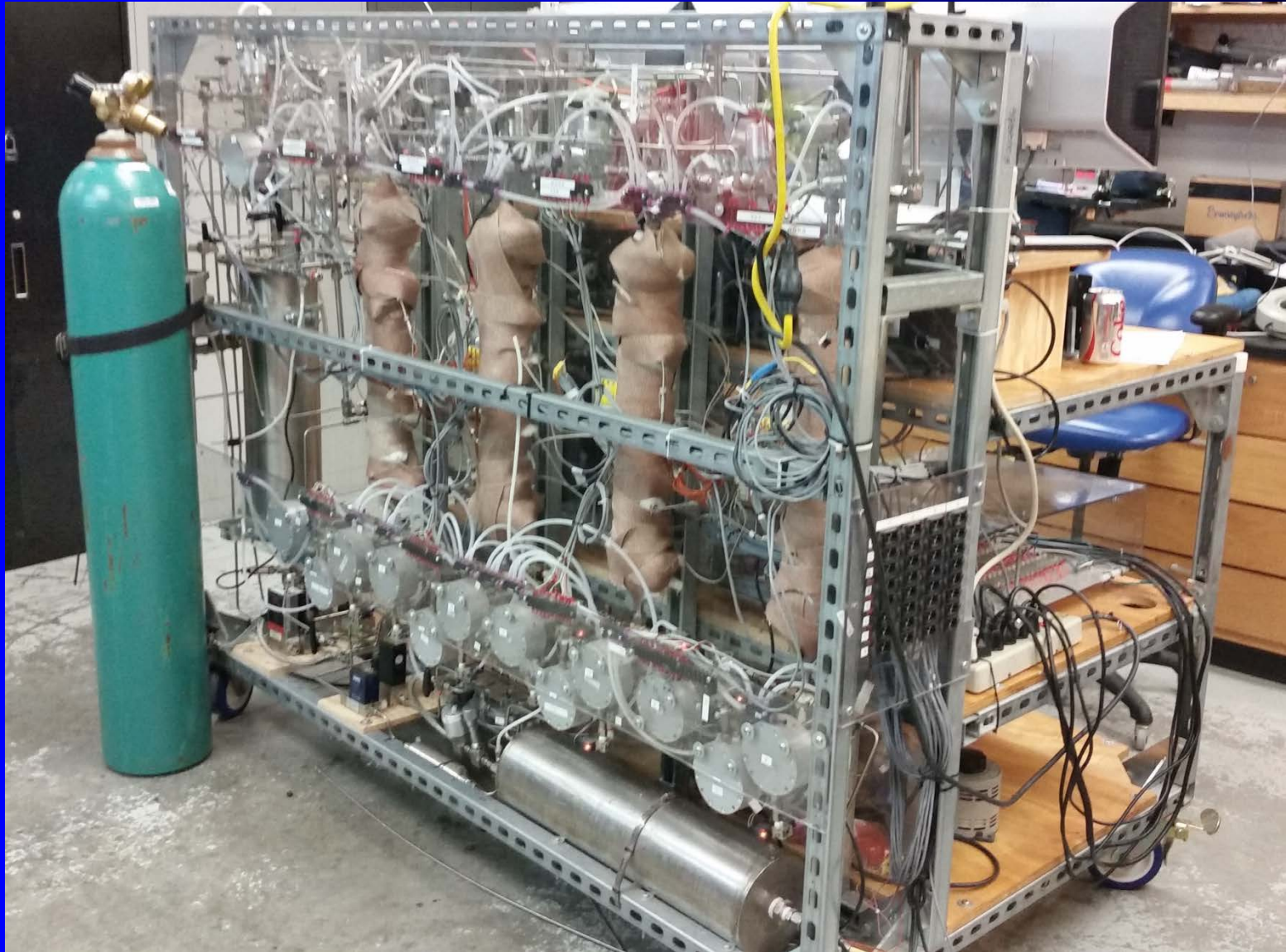
4-Bed PSA System

Suitable for Power Plant Demonstration



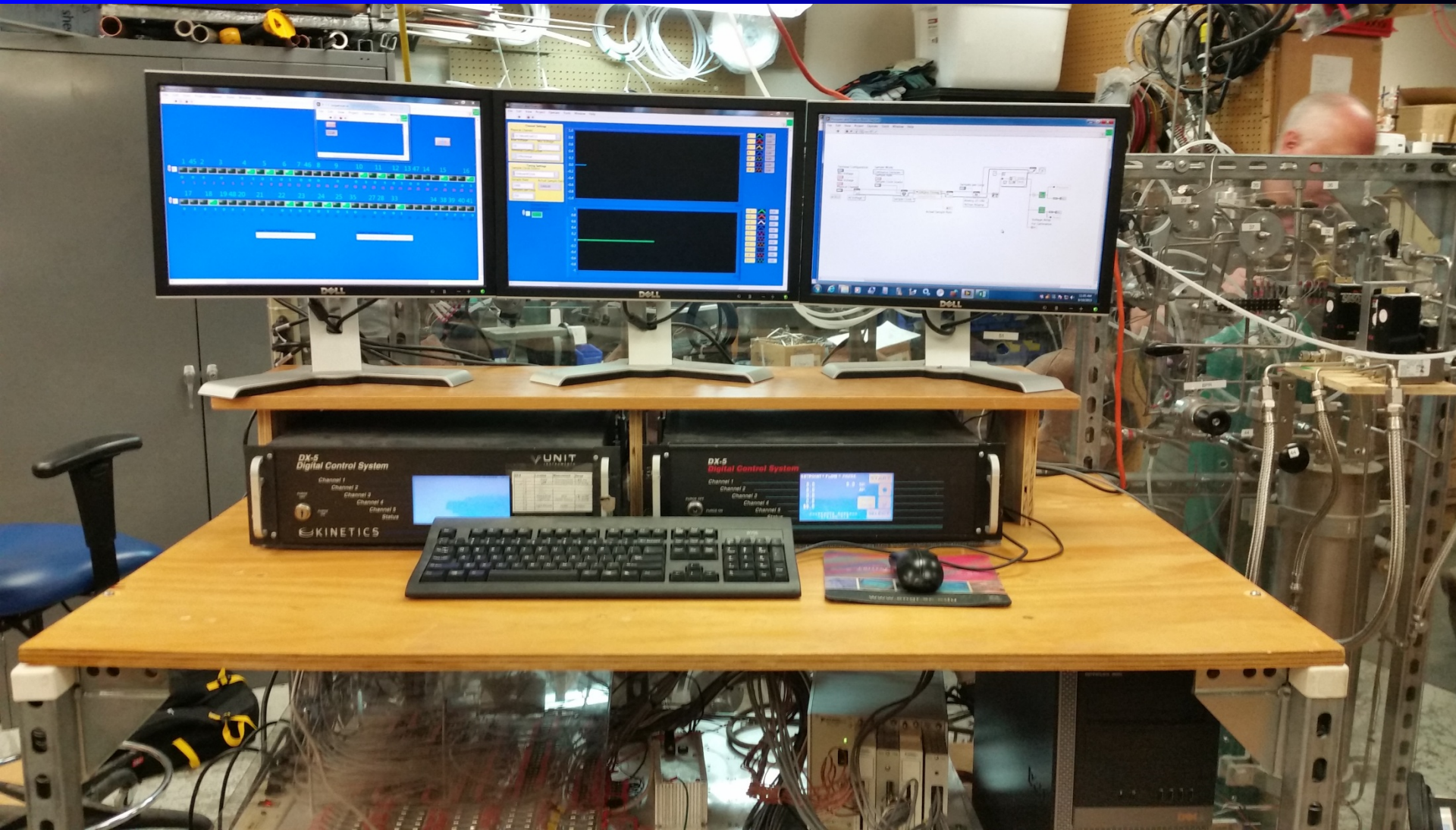
4-Bed PSA System

Suitable for Power Plant Demonstration

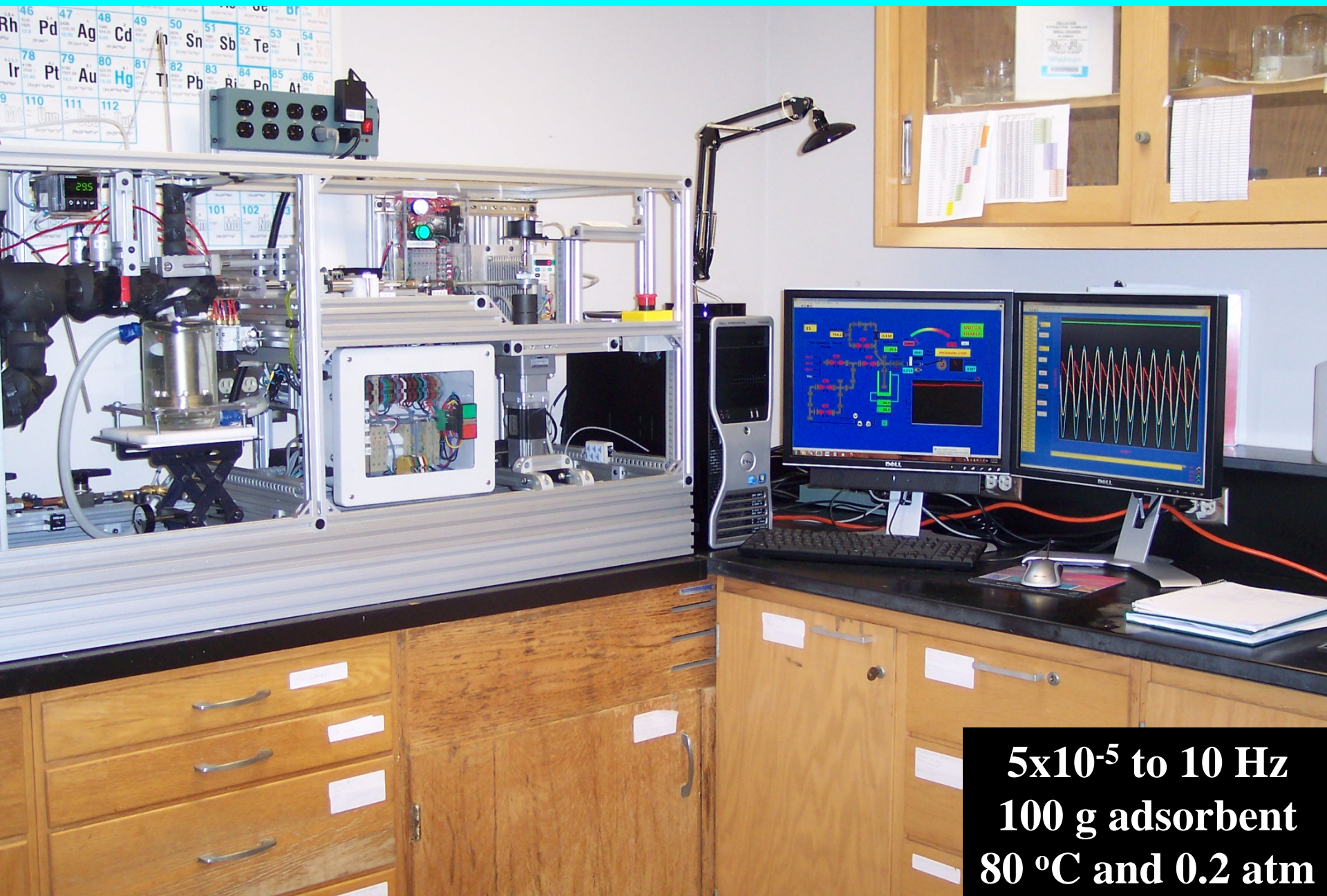


4-Bed PSA System

Suitable for Power Plant Demonstration



Volumetric Frequency Response Apparatus



5×10^{-5} to 10 Hz
100 g adsorbent
80 °C and 0.2 atm

Comparison of Mass Transfer Coefficients N₂ and CO₂ on 13X Zeolite Beads at 25 °C

	k s⁻¹	
	VFR	1-Bed RPSA
CO₂	3.3	7.5
N₂	5.1	4.6

- VFR: volumetric frequency response
- 1-Bed rapid PSA experiments

Higher values are expected in the structured adsorbent!

Uncoated and Zeolite Coated Catacel Cores

Specially Designed for Use in VFR Apparatus



$$\text{CPSI} = 741$$

$$\varepsilon_b = 0.64$$

$$w_{\text{foil}} = 52 \mu$$

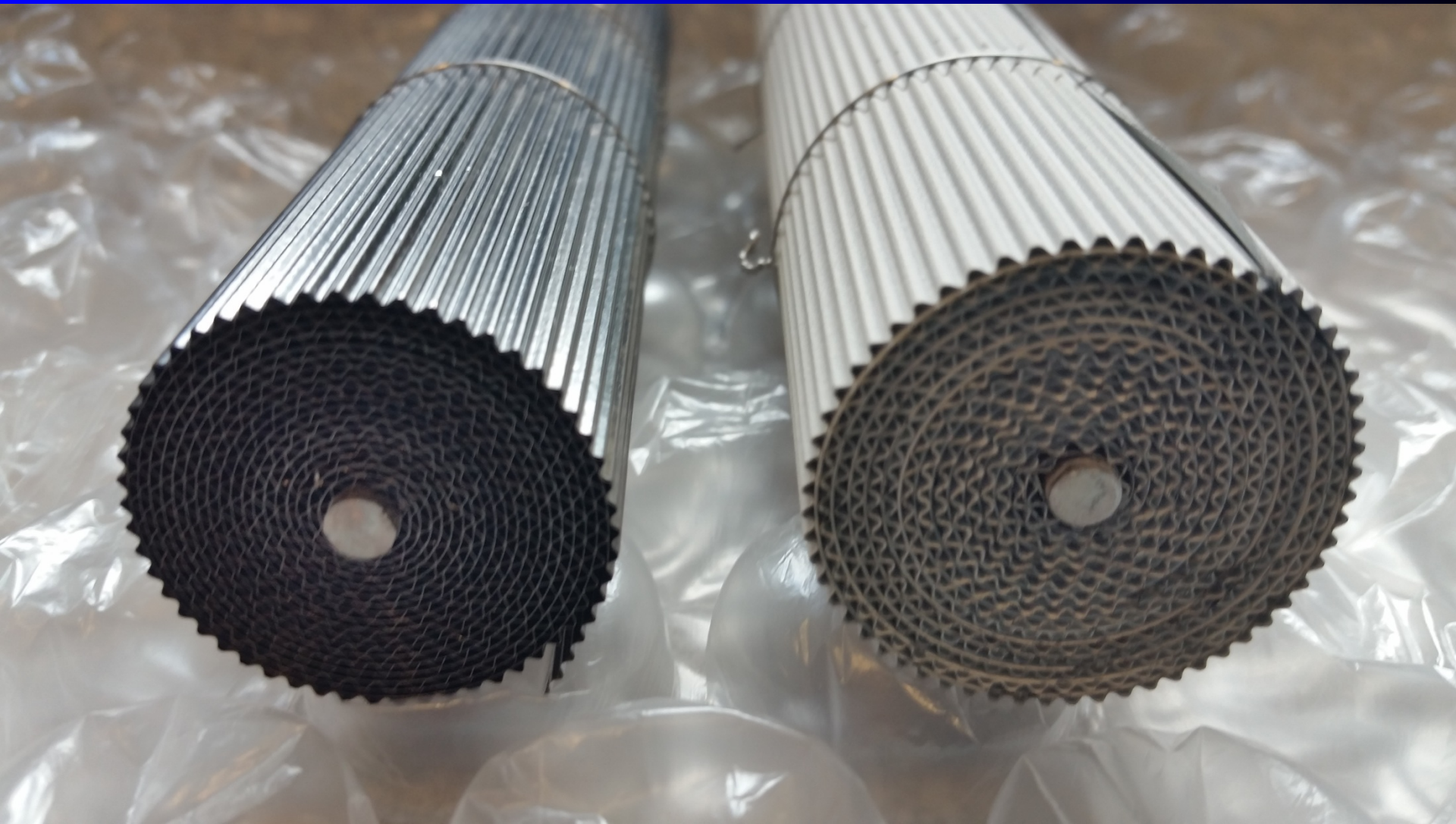
$$w_{\text{coating}} = 51 \mu$$

$$\rho_b = 241.93 \text{ kg/m}^3$$

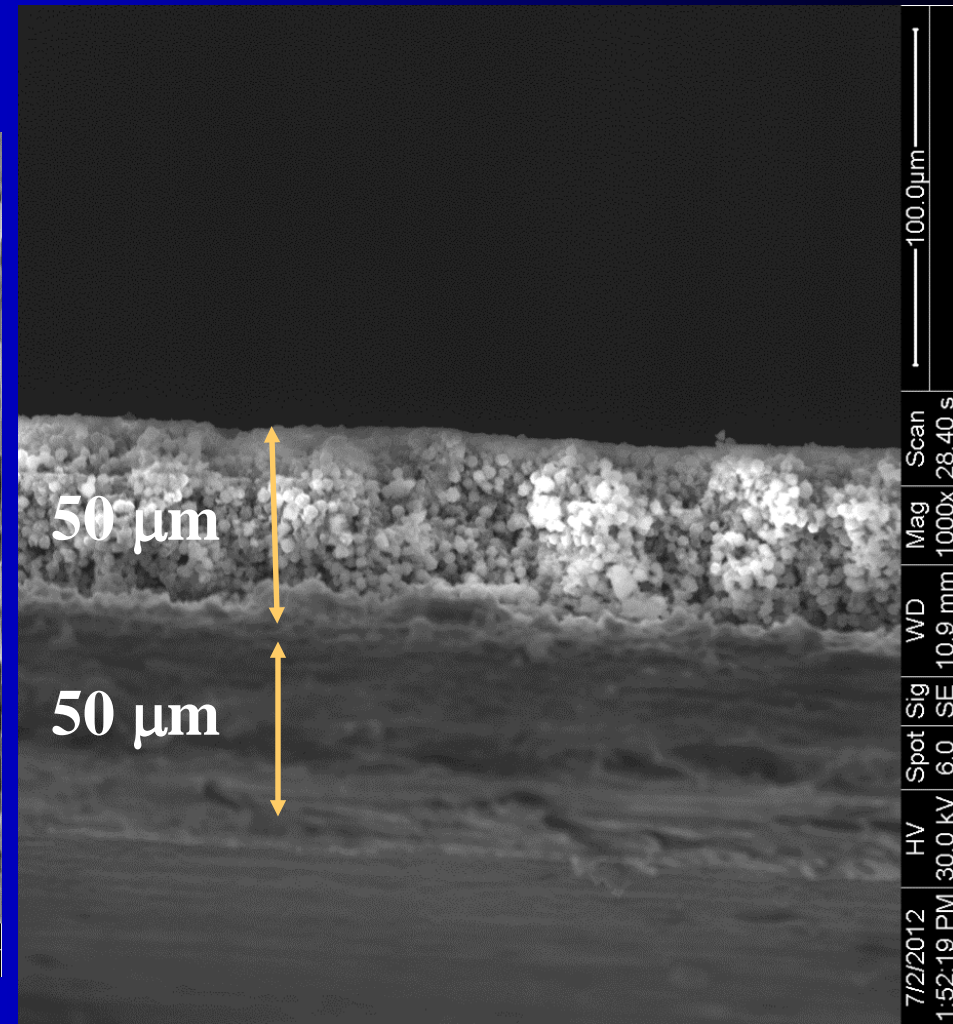
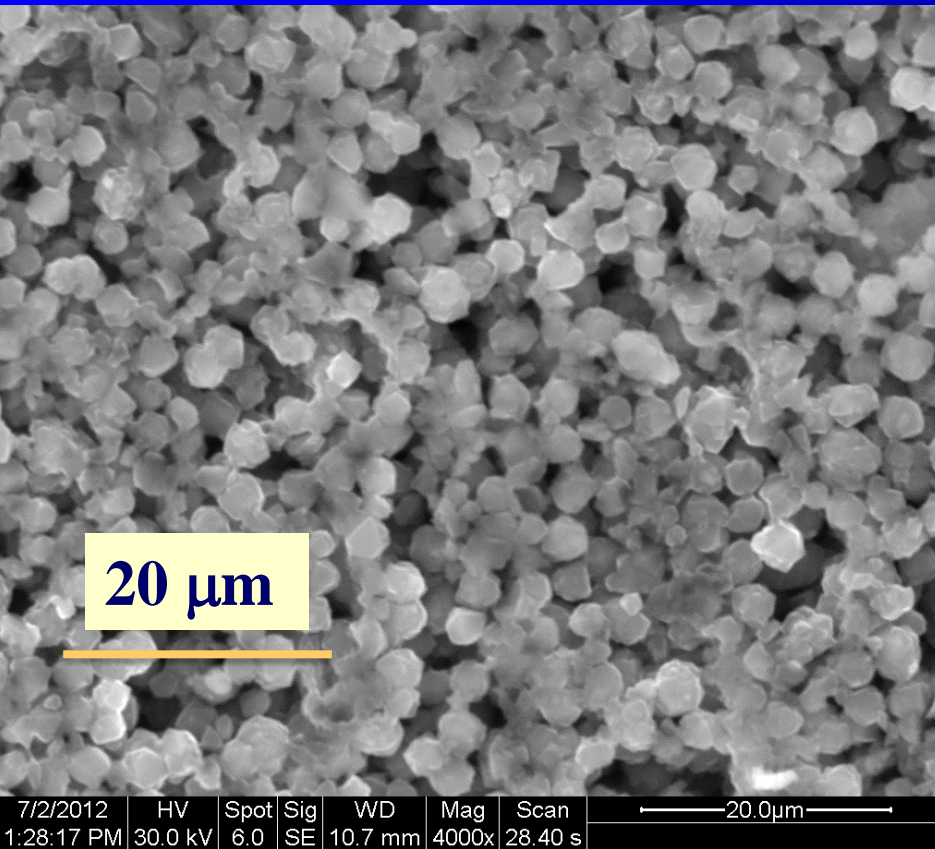
Uncoated and Zeolite Coated Catacel Cores

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Zeolite Coated Catacel Metal Foil



Parallel Channel Structured Adsorbent Column Containing Three 6" Zeolite Coated Catacel Cores



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Parallel Channel Structured Adsorbent Column Containing Three 6" Zeolite Coated Catacel Cores

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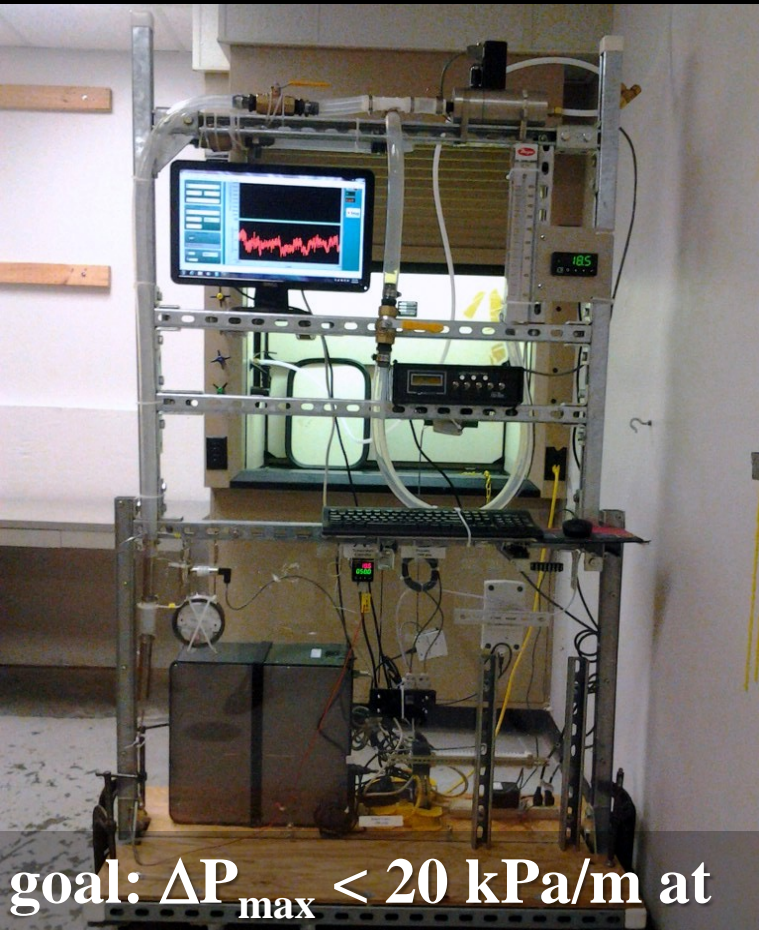
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Structured Adsorbent Pressure Drop

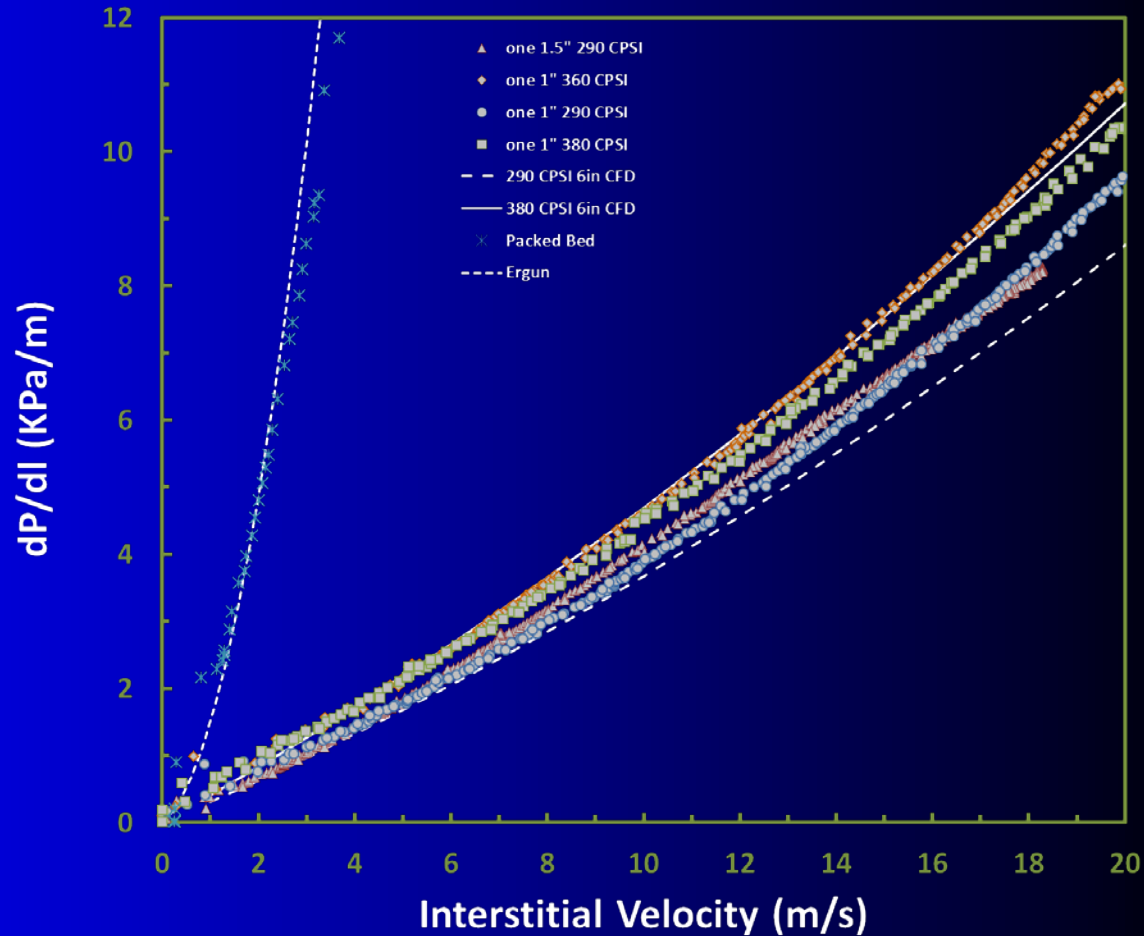
Open Cell Corrugated Structure and Beaded Media

Pressure Drop Apparatus

$Q_{\max} = 1000 \text{ SLPM}$
 $\Delta P_{\max} = 30, 70 \text{ or } 140 \text{ in H}_2\text{O}$



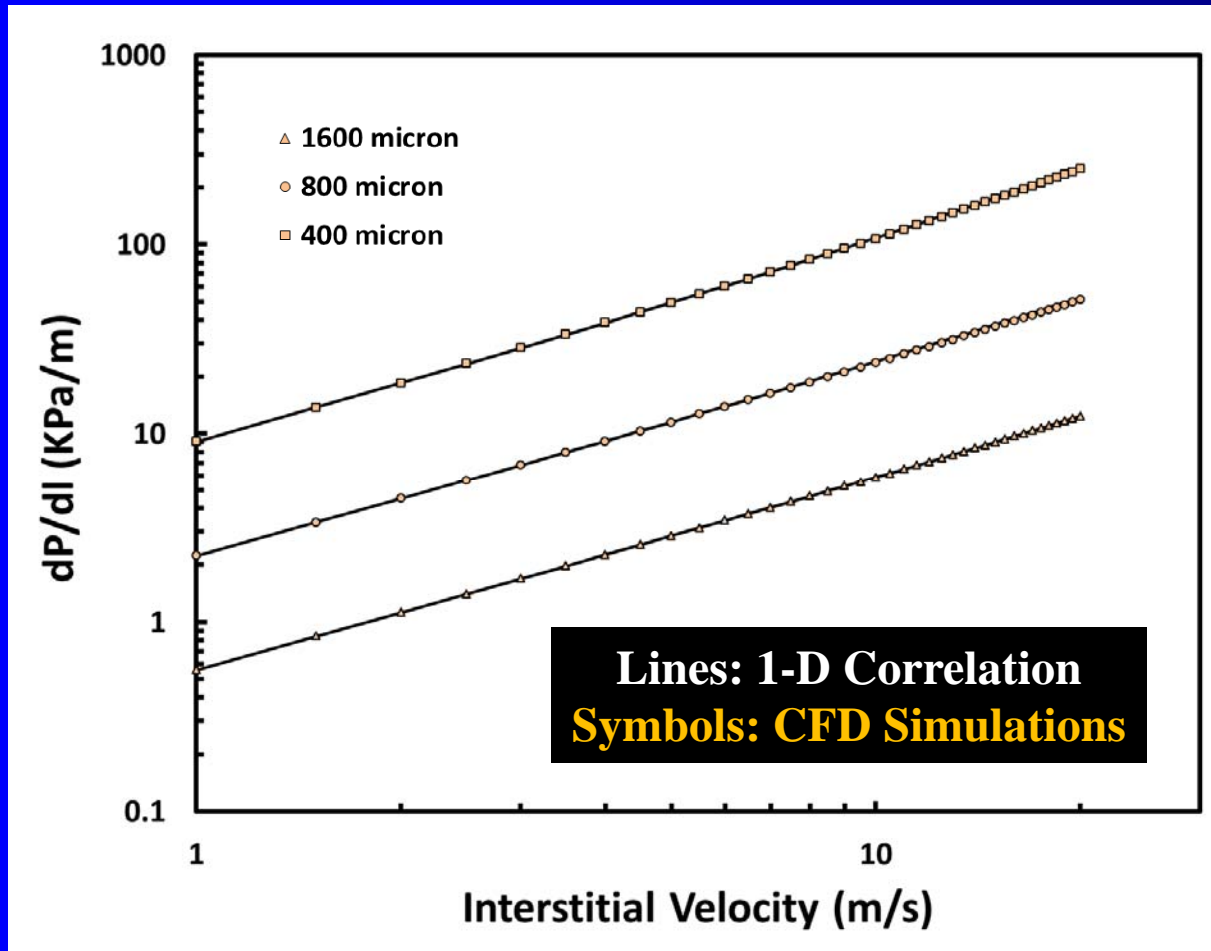
goal: $\Delta P_{\max} < 20 \text{ kPa/m}$ at design velocity of 20 m/s



3-D CFD

Compressible Navier-Stokes Equations

New 1-D Pressure Drop Correlation for Parallel Channel Structured Adsorbent and Use in DAPS

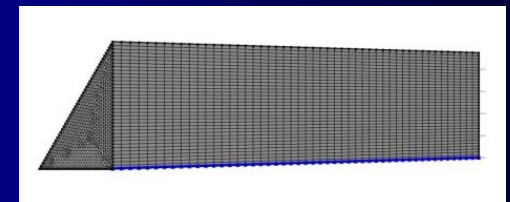
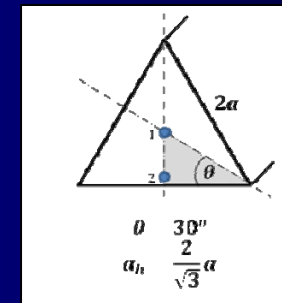


$$f(Re) = f_1 + \frac{f_2}{Re}$$

$$Re = \frac{2\rho V_z a_n}{\mu}$$

$$f_1 = 2.5 \times 10^{-3}$$

$$f_2 = 25.42$$

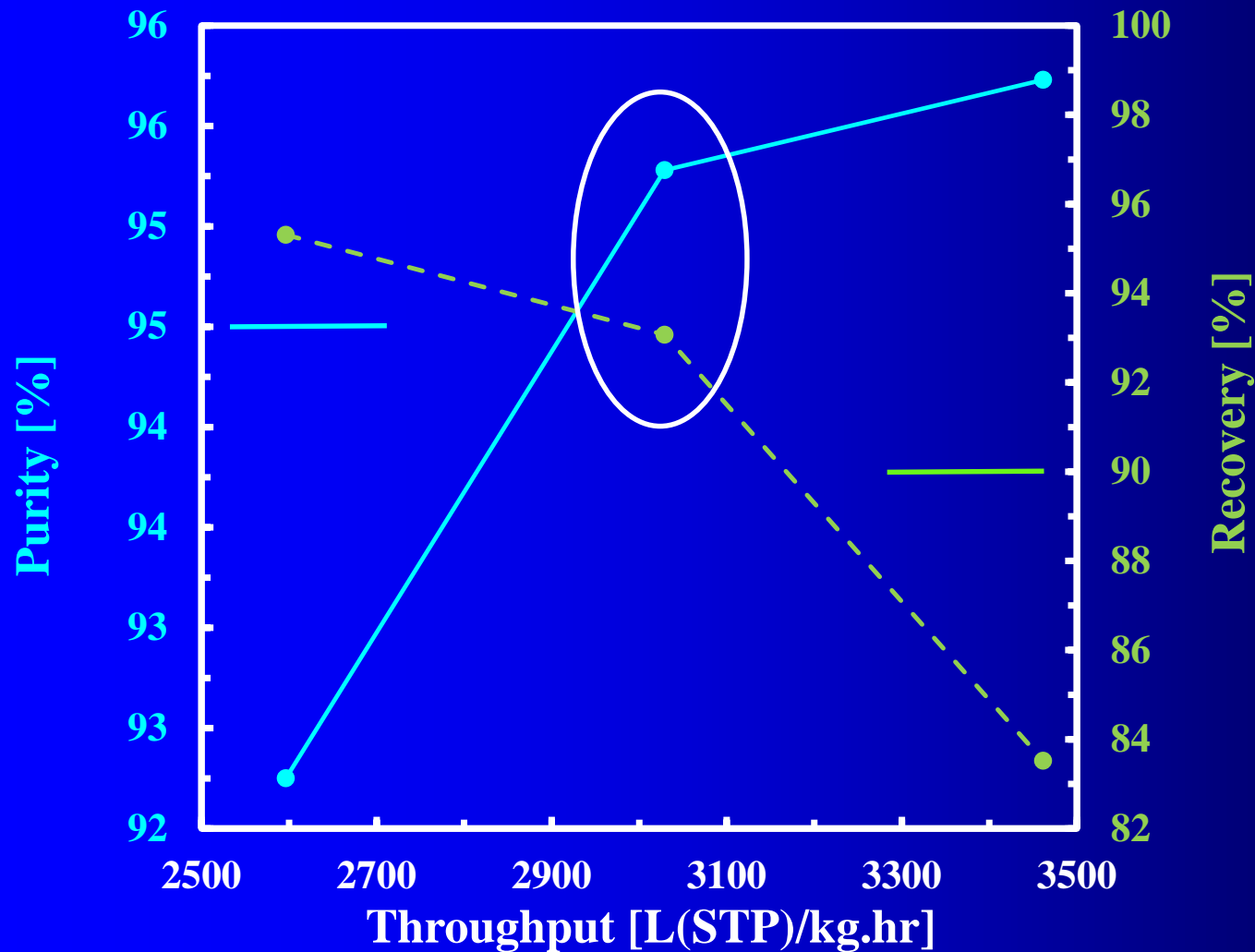


Developed from 3-D CFD Simulations

DAPS of Bench Scale PSA Processes

Zeolite Coated Catacel Structured Core

$y_F = 0.1592 \text{ CO}_2, 0.8029 \text{ N}_2 \text{ and } 0.0379 \text{ O}_2$



$P_H = 103.3 \text{ kPa}$

$P_L = 5 \text{ kPa}$

$t_c = 120 \text{ s}$

$T = 75 \text{ }^\circ\text{C}$

$L_b = 0.125 \text{ m}$

$d_b = 0.098 \text{ m}$

$k_{LDF} = 1 \text{ s}^{-1}$

$\text{CPSI} = 741$

$\varepsilon_b = 0.64$

$w_{\text{foil}} = 52 \text{ } \mu$

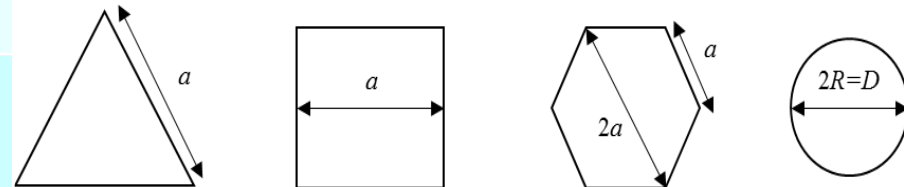
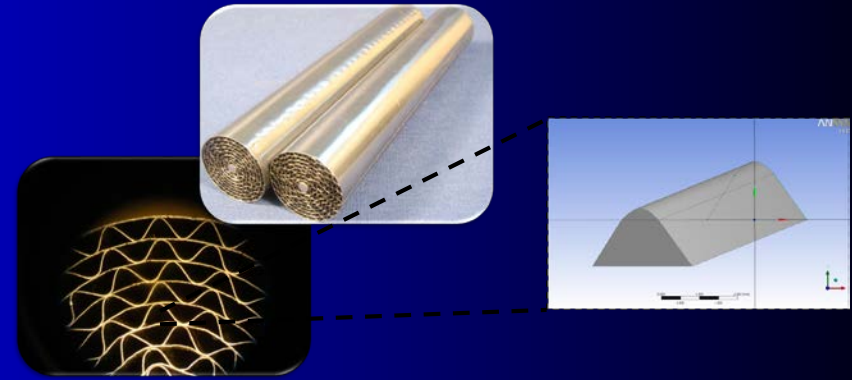
$w_{\text{coating}} = 51 \text{ } \mu$

$\rho_b = 241.93 \text{ kg/m}^3$

Determination of Optimal Channel Shape

- use 1-D models with friction factors and mass transfer parameters determined by 3-D CFD
- match performance to predictions from DAPS, then find minimum parasitic energy over key parts of cycle

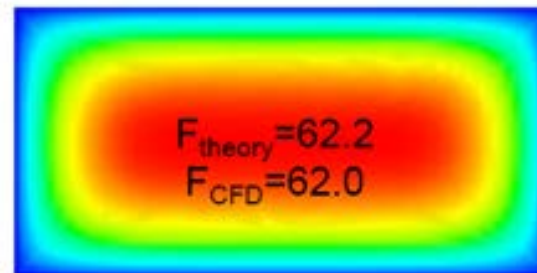
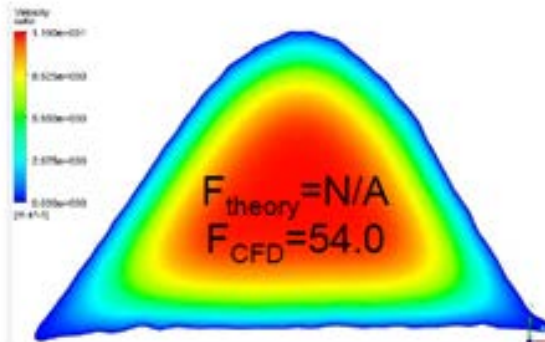
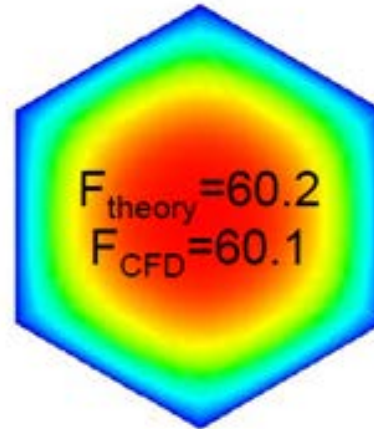
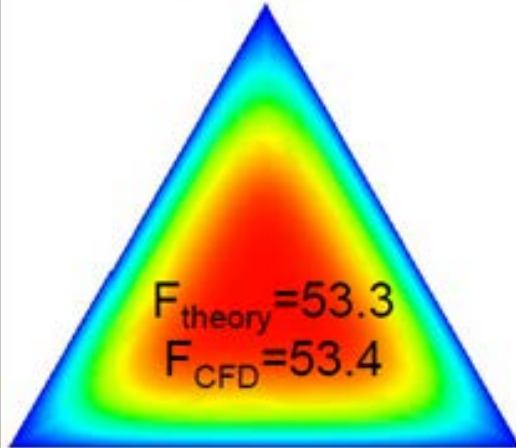
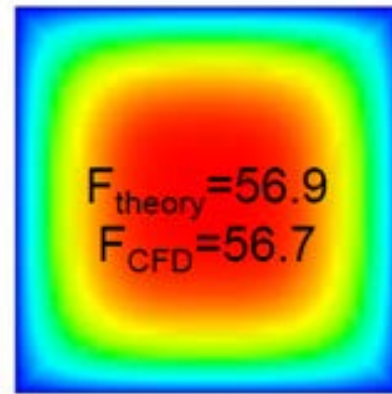
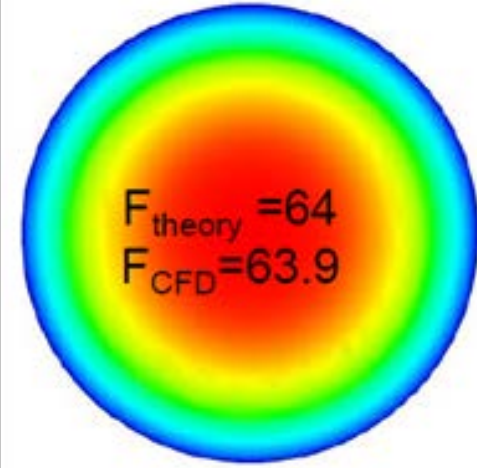
Shape (sides)	Hydraulic Diameter, D_h	Friction Factor, f
Triangle ($n=3$)	$D_h = \frac{\sqrt{3}}{3} a$	$f = \frac{53.3}{Re}$
Square ($n=4$)	$D_h = a$	$f = \frac{56.9}{Re}$
Hexagon ($n=6$)	$D_h = \sqrt{3}a$	$f = \frac{60.2}{Re}$
Circle ($n=\infty$)	$D_h = 2R$	$f = \frac{64}{Re}$



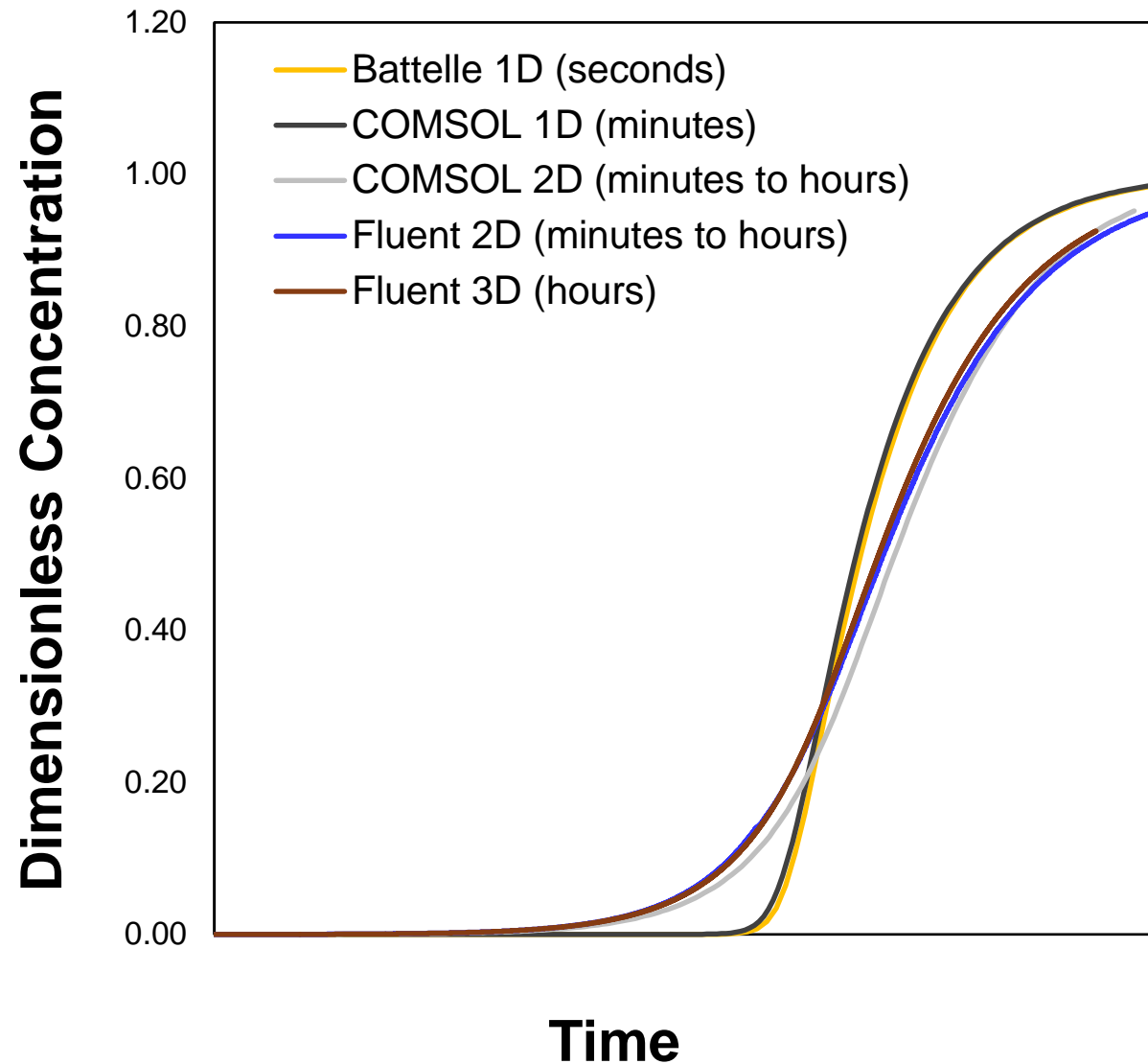
Best Shape Estimates from Rapid 1D Model

Based on
Parameters from
3-D CFD Models

Comparison of
Friction
Factors from
CFD and
Theory



Best Shape Estimates from Rapid 1-D Model Based on Parameters from 3-D CFD Models



**Comparison of
Breakthrough
Times from
Different Mass
Transfer
Models**



Where are we headed?

On-Going Tasks to Complete in BP 2

- test breakthrough and cycling behavior of zeolite coated Catacel cores with $\text{CO}_2\text{-N}_2$ in 1-bed PSA apparatus
- test cycling behavior of multi-bed PSA apparatus with $\text{CO}_2\text{-N}_2$ using 3 mm zeolite beads
- validate DAPS with results from bench scale PSA systems utilizing zeolite beads and zeolite coated Catacel cores
- measure pressure drop through zeolite coated Catacel cores
- characterize thermodynamic and mass transfer properties of zeolite coated Catacel cores and refine PSA cycle schedule via modeling with DAPS
- study adsorbent (zeolite crystals and silica gel) stability in presence of trace levels of NO_x and SO_2
- validate 1-D, 2-D and 3-D CFD models by comparison to DAPS
- investigate friction factor and mass transfer assumptions during dynamic adsorption and desorption to refine 1-D models
- use 1-D models to optimize Catacel channel shape

Budget

Project Team Member	Budget Period 1		Budget Period 2		Budget Period 3		Total
	Gov. Share	Cost Share	Gov. Share	Cost Share	Gov. Share	Cost Share	
	Grace	139441	34860	75084	18772	145089	
USC	670000	167500	490000	122500	490000	122500	2062500
Battelle	239115	59978	191791	47930	159744	39998	738556
Catacel	125592	31398	172187	43047	100662	25166	498052
TOTAL	1174148	293736	929062	232249	895495	223936	3748626

Breakdown in % of Total Budget

USC 55.0%

Battelle 19.7%

Catacel 13.3%

Grace 12.0%

Acknowledgements

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DOE/NETL and

SAGE is greatly appreciated!



Thank You!

