

# Surface Modified LSCF Powders for SOFC Cathodes

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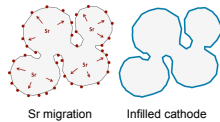
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## Introduction

Improvements are needed to mitigate effect of cathode degradation on long-term SOFC performance

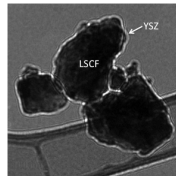
- Sr migration to surface of perovskite cathode materials appears to be linked to decreased cell performance at elevated temperature
- Post-sintering infill approaches have shown promise for improved short and long-term performance
- Surface modified powder is an alternate, potentially cost-effective route to improved long-term performance



Idealized process route for surface modified cathode powder



Pre-demonstration of ALD coating of LSCF: YSZ via Atomic Layer Deposition (ALD) TEM micrograph



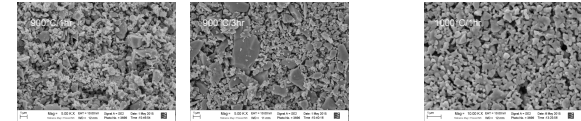
## Choice of Shell Material

Desired attributes of shell material:

- Conductive
- Strong blocking potential for Sr migration to surface
- Non-interfering with sintering
- Non-reactive with cathode
- Long term stability

Infill material	Attributes
Pd, Ag	catalyst
La, Sr manganate (LSM)	Triple phase boundaries
Sm-doped ceria (SDC)	Expensive compared to Gd
La, Ca ceria (LCC)	Three component
Sm, Sr cobaltate (SSC)	Barrier properties unknown
Gd-doped ceria (GDC)	Existing barrier in SOFC
Y-doped ceria (YDC)	Lower conductivity than GDC
Pr-doped ceria (PDC)	Expensive compared to GDC
ZrO <sub>2</sub>	Non-conductive

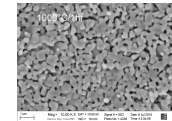
## Sintering Behavior



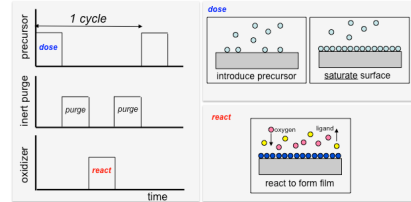
- La<sub>0.8</sub>Sr<sub>0.2</sub>Co<sub>0.8</sub>Fe<sub>0.2</sub>O<sub>3-δ</sub> (LSCF) sintered in air
- 1000°C/1hr shows good sintered structure
- Non-uniform starting particle size distribution

- LSCF ball milled for 12hr
- Ball milled structure shows more uniform particle size

- LSCF with 60 cycles GDC (240°C)
- Minimal impact of coating on sintering



## Core - Shell Approach



ALD's highly self-regulating process enables precise control of surface modification

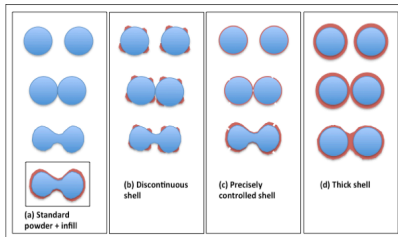
- Thickness of coating
- Morphology of coating

Examples of surface modification of particles via ALD

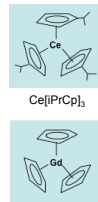
Particle	Coating
ZrO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub> <sup>1</sup>
BN	Al <sub>2</sub> O <sub>3</sub> <sup>2</sup>
Li(Li <sub>0.2</sub> Mn <sub>0.3</sub> Ni <sub>0.13</sub> Co <sub>0.36</sub> )O <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub> <sup>3</sup>
Polymers	W <sup>4</sup>

<sup>1</sup>J.A. McCormick, B.L. Clouder, W. Weimer, and S.M. George, "Rotary reactor for atomic layer deposition on large quantities of nanoparticles," *J. Vac. Sci. Technol. A* 25 (2007) 67.  
<sup>2</sup>J.R. Wank, S.M. George, and A.W. Weimer, "Nanocoating individual cohesive boron nitride particles in a fluidized bed by ALD," *Powder Tech.* 142 (2004) 99.  
<sup>3</sup>Y.S. Jung, A.S. Cavanagh, Y. Yan, S.M. George, and A. Manthiram, *J. Electrochem. Soc.* 158 (2011) A208.  
<sup>4</sup>C.A. Wilson, J.A. McCormick, A.S. Cavanagh, D.N. Goldstein, A.W. Weimer, and S.M. George, "Targeted atomic layer deposition on polymers," *Thin Solid Films* 516 (2008) 6175.

Potential sintering routes of surface modified particles (post-sintered infill process at left)



## GDC ALD Process



Cyclopentadienyl based chemistry:

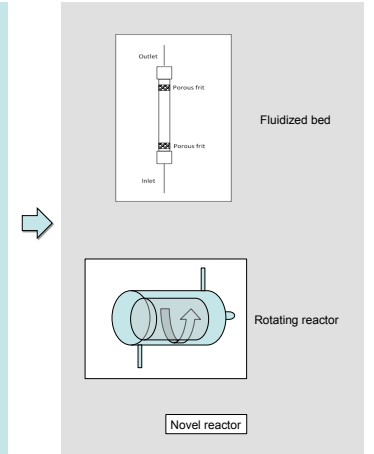
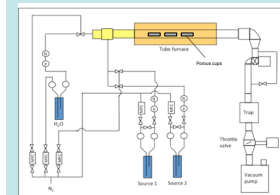
- Volatile
- Thermally stable
- Highly reactive with water as oxidizing agent

GDC films targeting 10mol% Gd have been deposited on LSCF at 230-240°C

- Loosely packed bed in porous cup
- 20 cycles
- 60 cycles
- 196 cycles



## Manufacturing Scale Up



## Acknowledgements

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