

A Bifunctional Ceramic Fuel Cell Energy System

An Update on Reversible Air-electrodes Development

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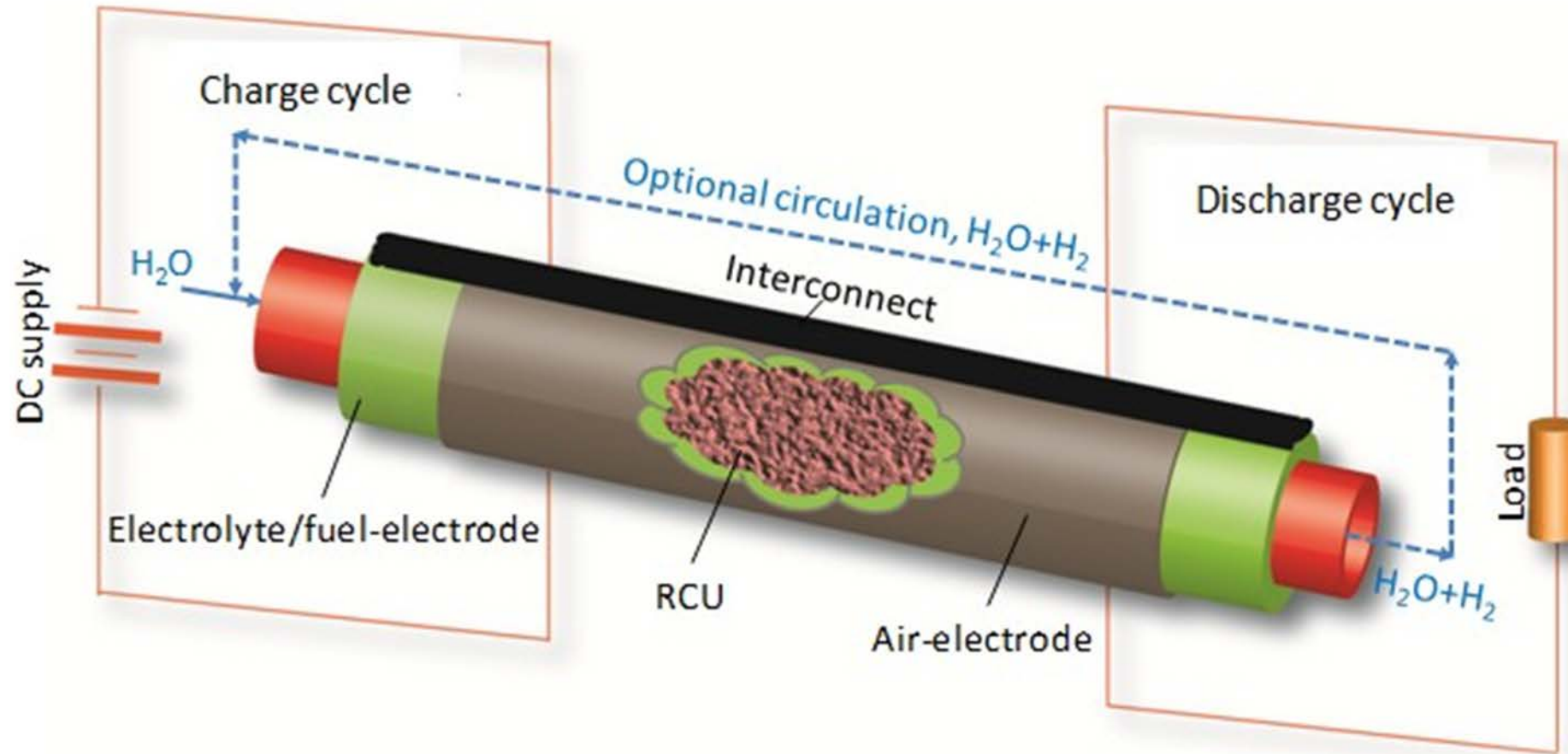


Award#: DE-AR-0000492

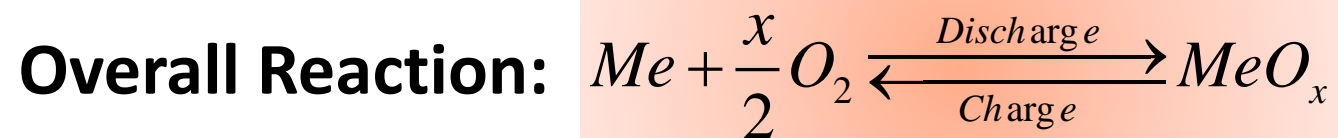
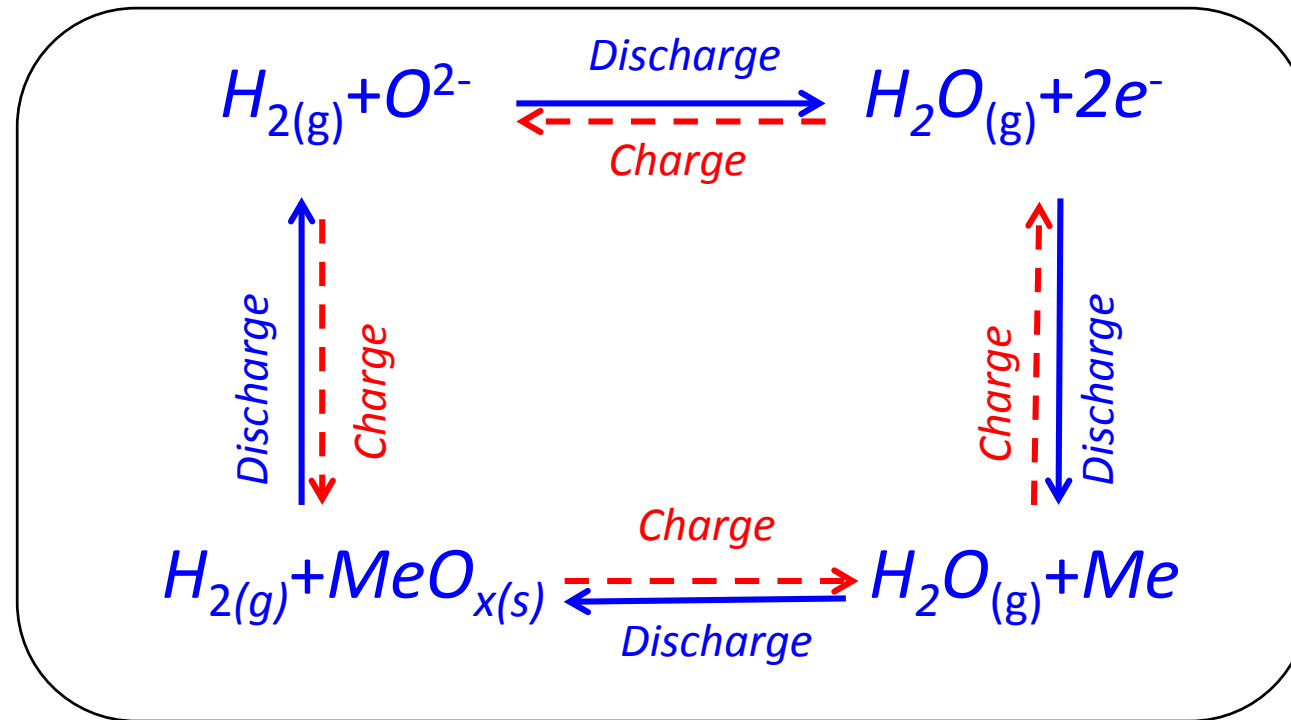
About this REBELS Project

- Demonstration of power generation and energy storage functionality of the new fuel-cell/battery hybrid system at bench-scale
- Development of reversible IT-air-electrodes
- Development of IT-electrolytes
- Development of Fe-based energy storage materials
- Development of computational models

The Battery Configuration

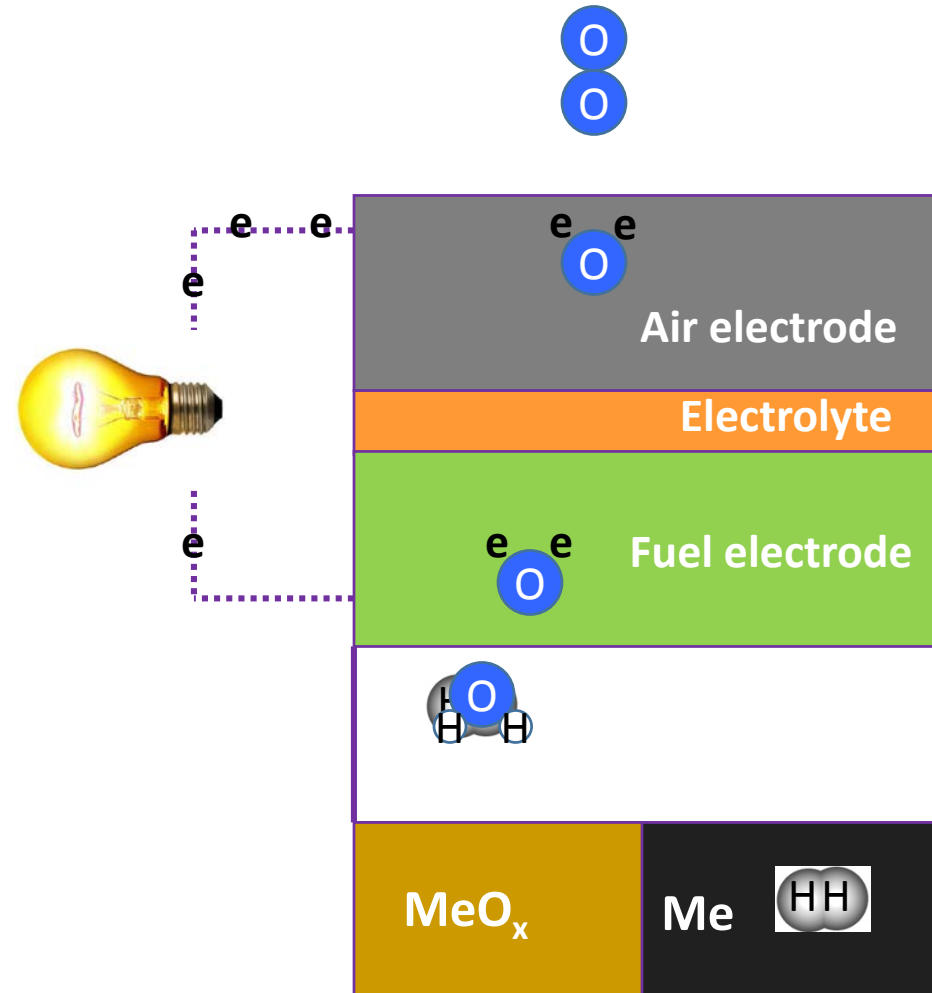


The New Battery Chemistry

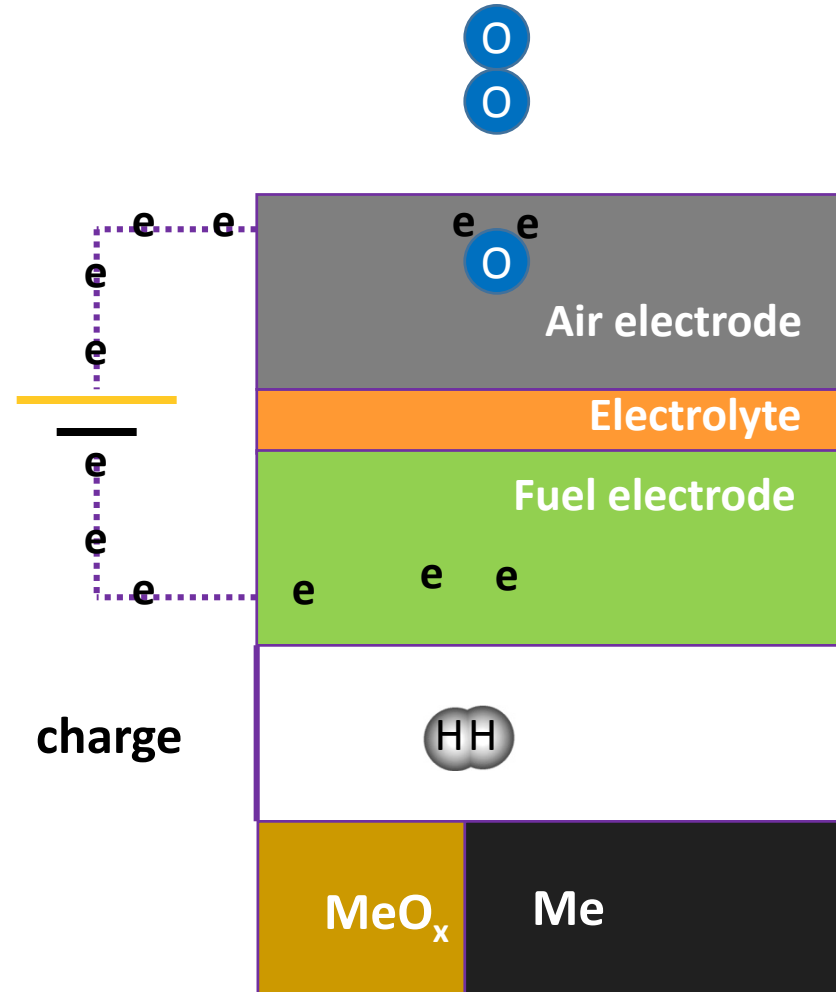


Solid Oxide Metal Air Redox Battery (SOMARB)

Working Principle: Discharging

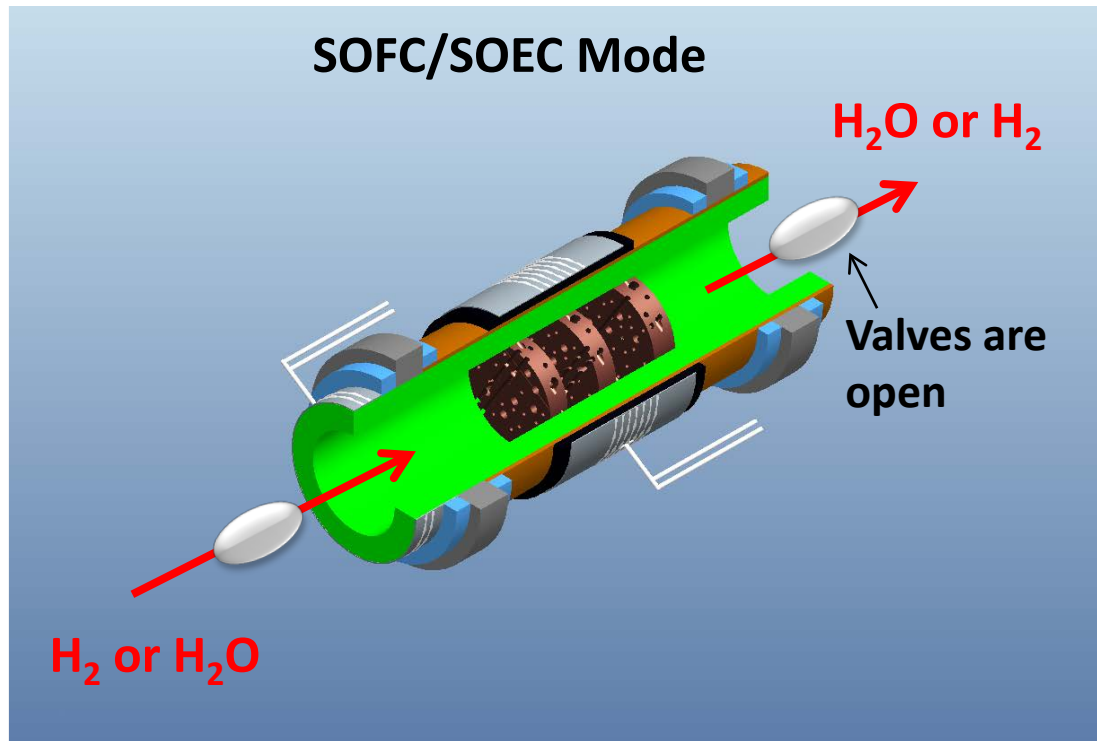


Working Principle: Charging

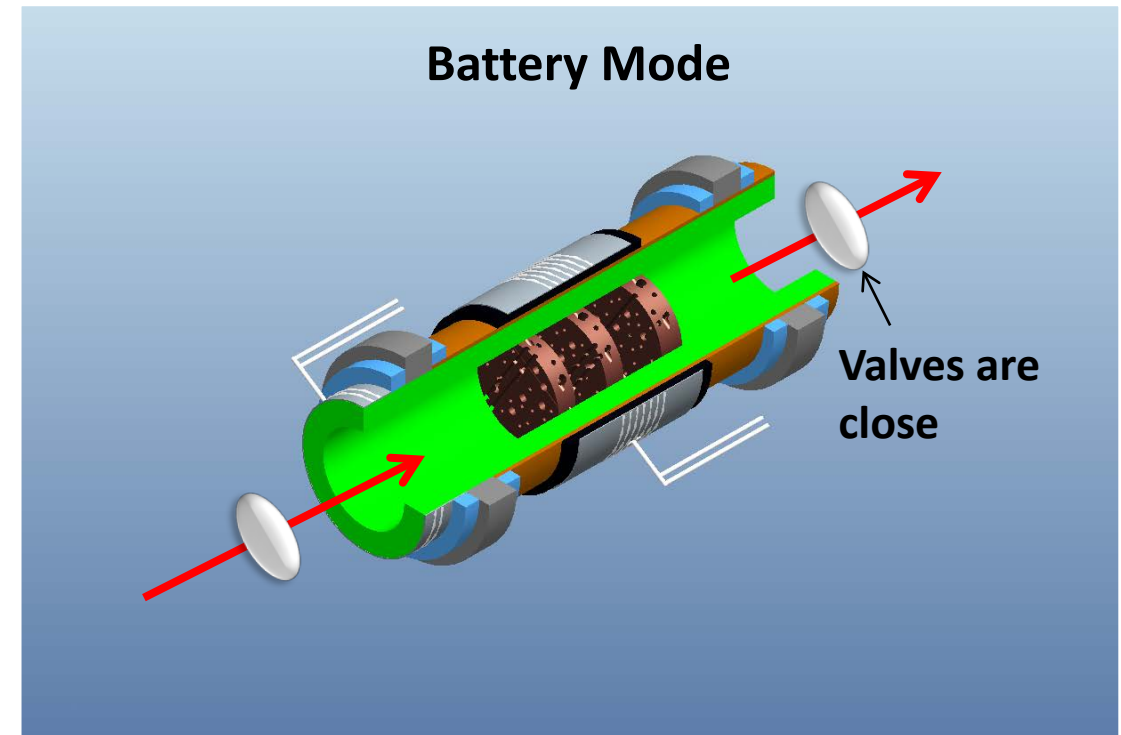


Easy Switching between Fuel Cell and Battery Mode

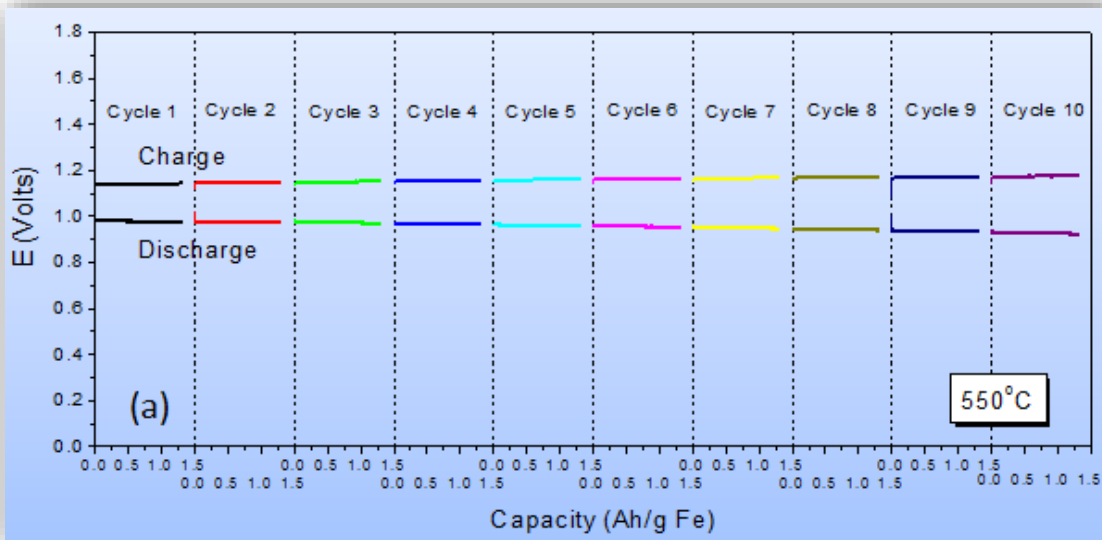
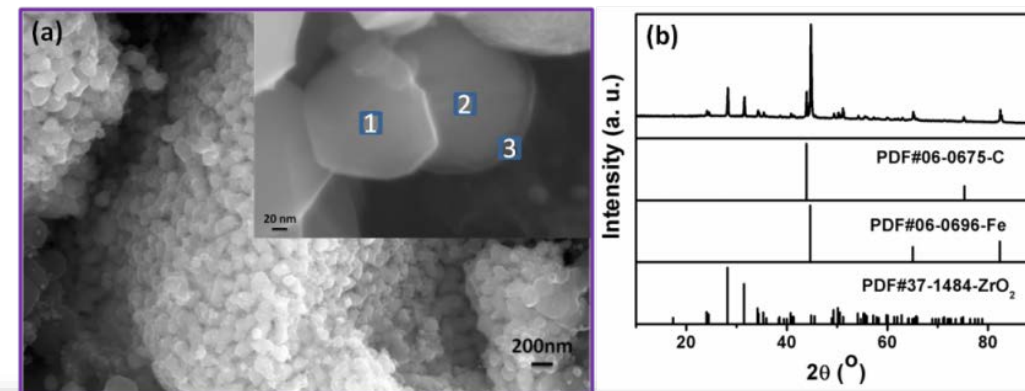
SOFC/SOEC Mode



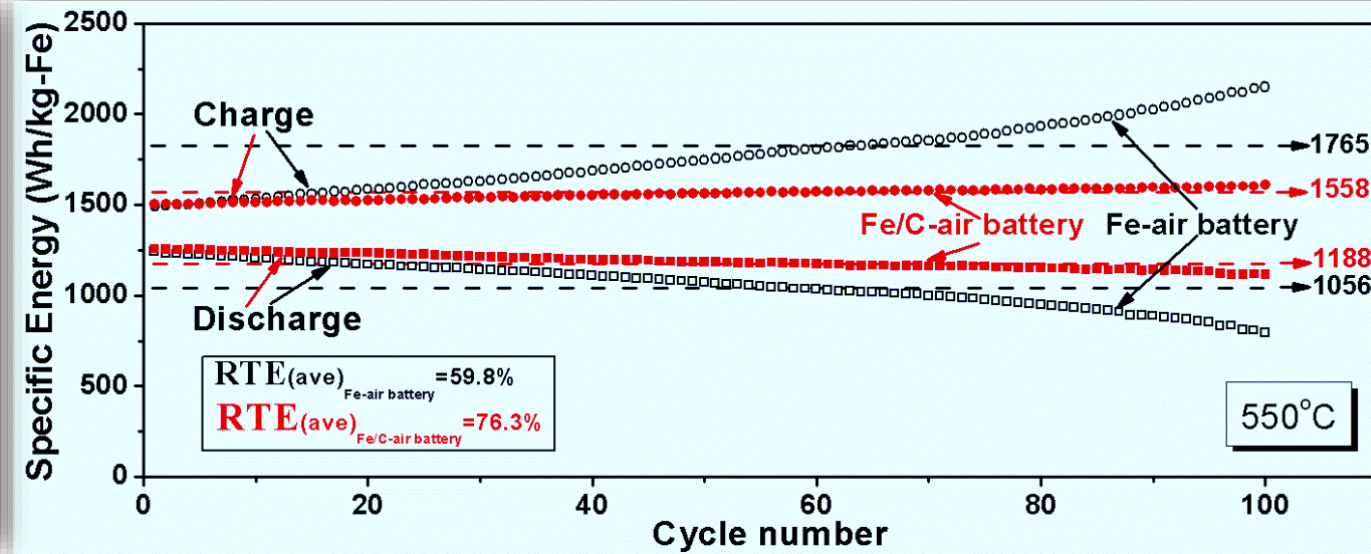
Battery Mode



A Typical Performance of the SOMARB Battery



Chem. Comm., 2014, 50, 623.



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Distinguished Advantages

- All ceramic components
- Transfer of two electrons by O^{2-}
- Only gaseous O_2 involved in ORR and OER
- Energy storage with high-capacity chemical redox bed
- Energy cycling at high rates
- Independent power and energy
- Adaptable to new metal-air chemistries
- Scalable, sustainable & safe

A Reversible Air-electrode of Choice: $\text{SrCoO}_{3-\delta}$

Important Features:

- Highly oxygen catalytically active at IT range
- Unstable at elevated temperatures: losing oxygen to decompose into $\text{Sr}_2\text{Co}_2\text{O}_5$:

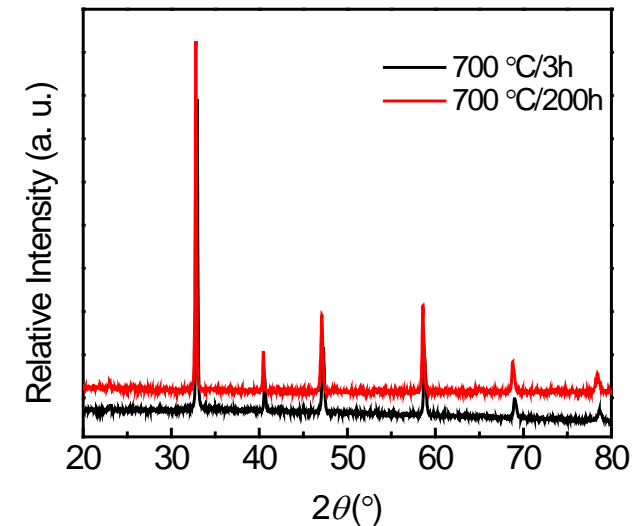
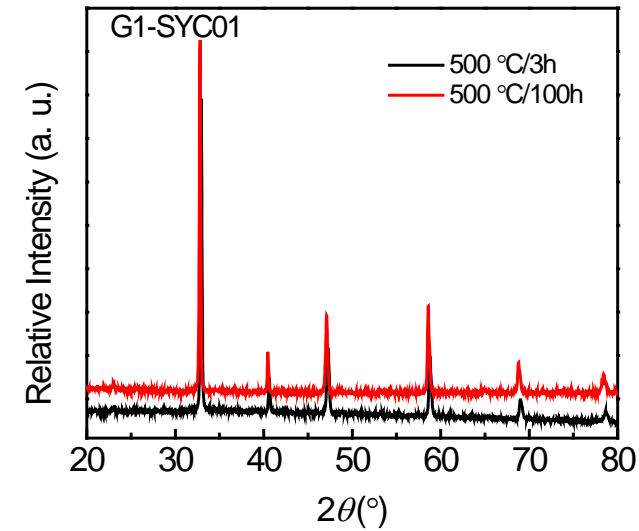
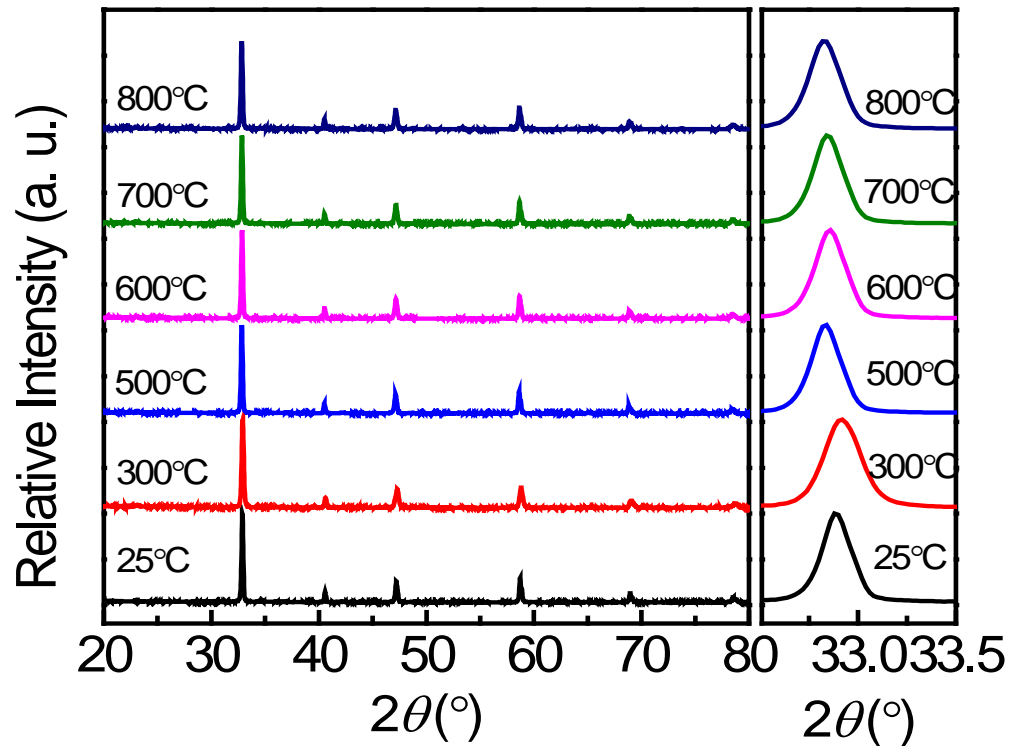


Making $\text{SrCoO}_{3-\delta}$ stable:

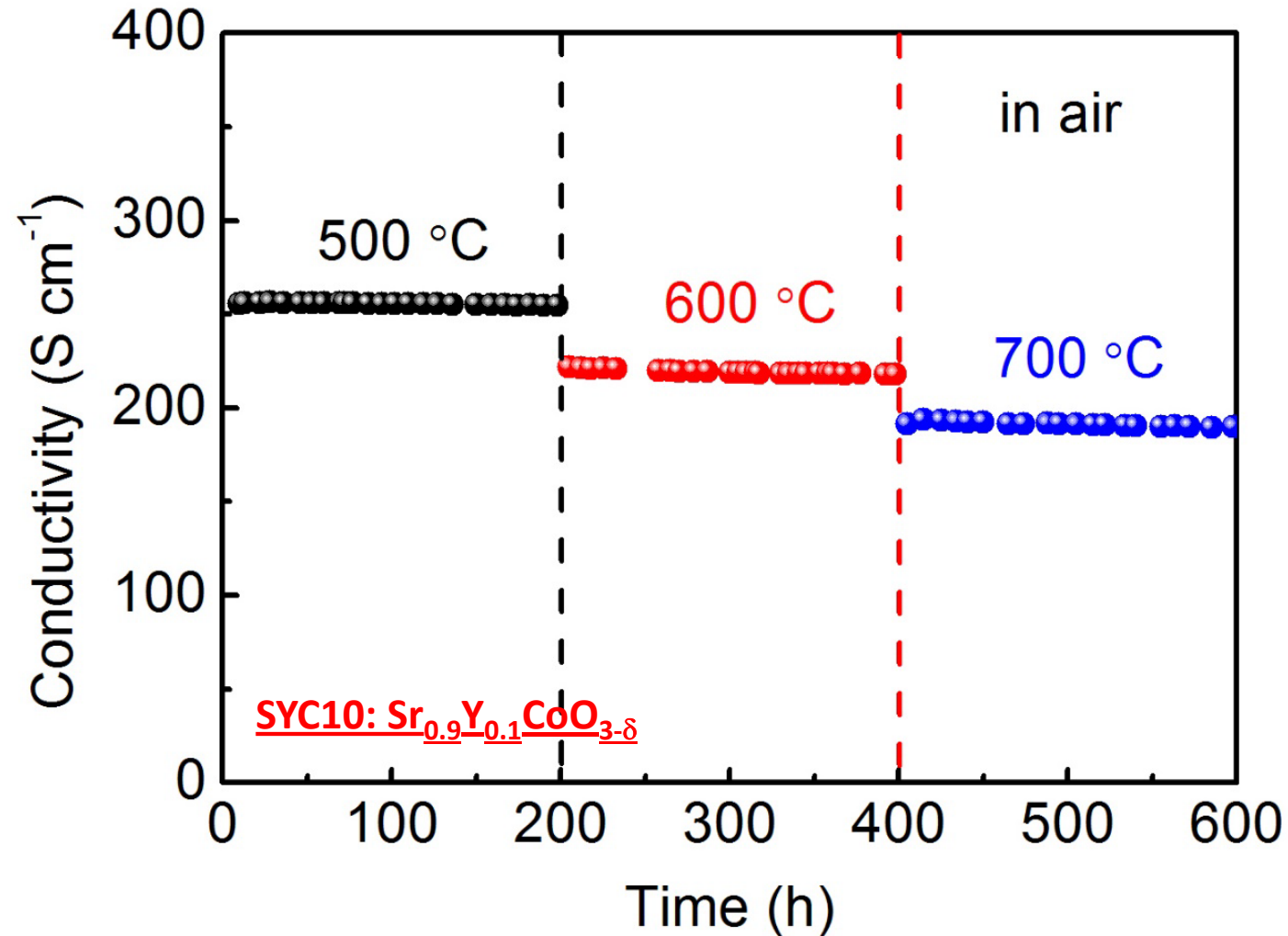
- Phase stabilization by A- and B-site donor doping
- Morphological stabilization of nanostructures by atomic layer deposition

Phase Thermal Stability – SYC Series

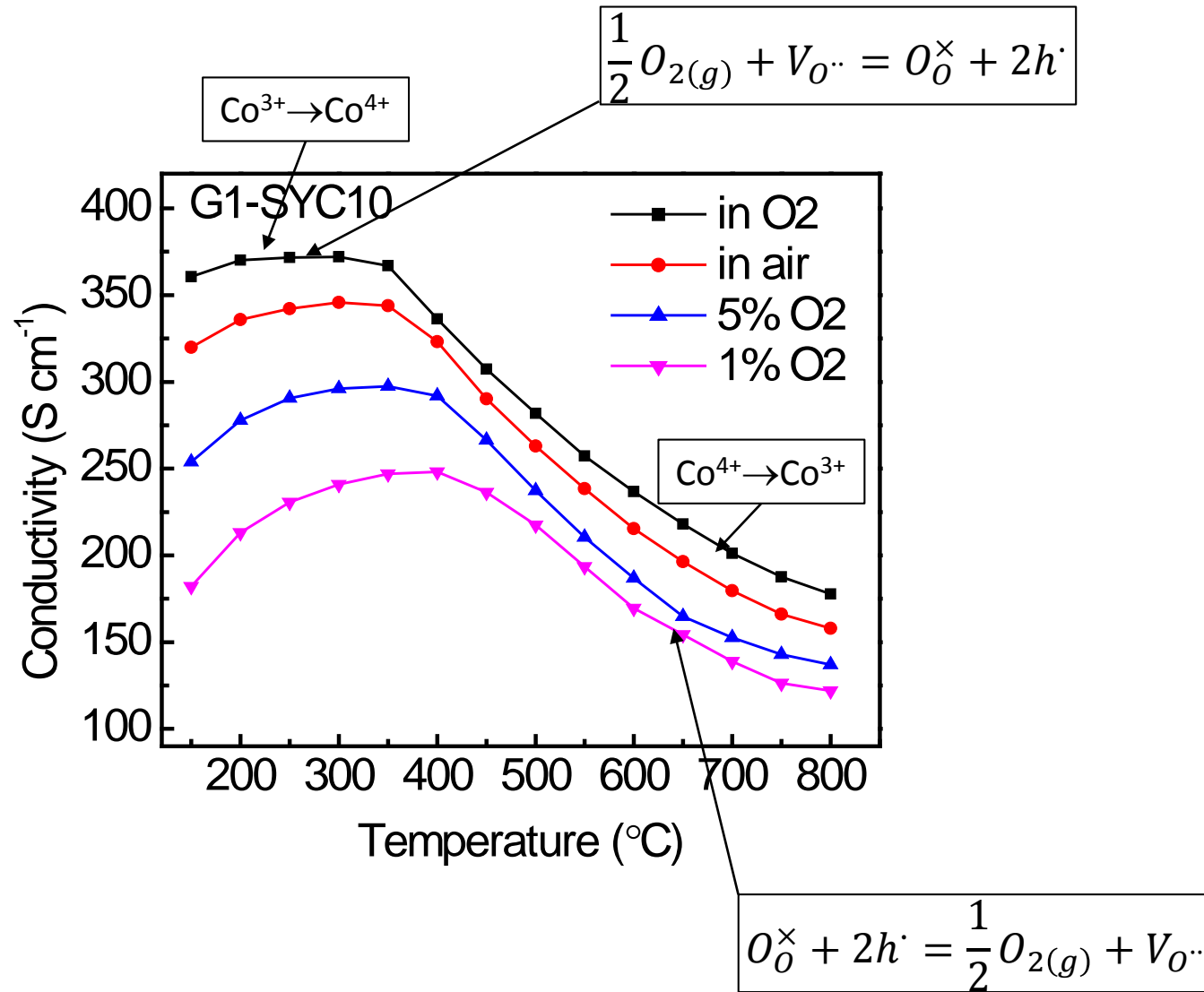
SYC10: $\text{Sr}_{0.9}\text{Y}_{0.1}\text{CoO}_{3-\delta}$



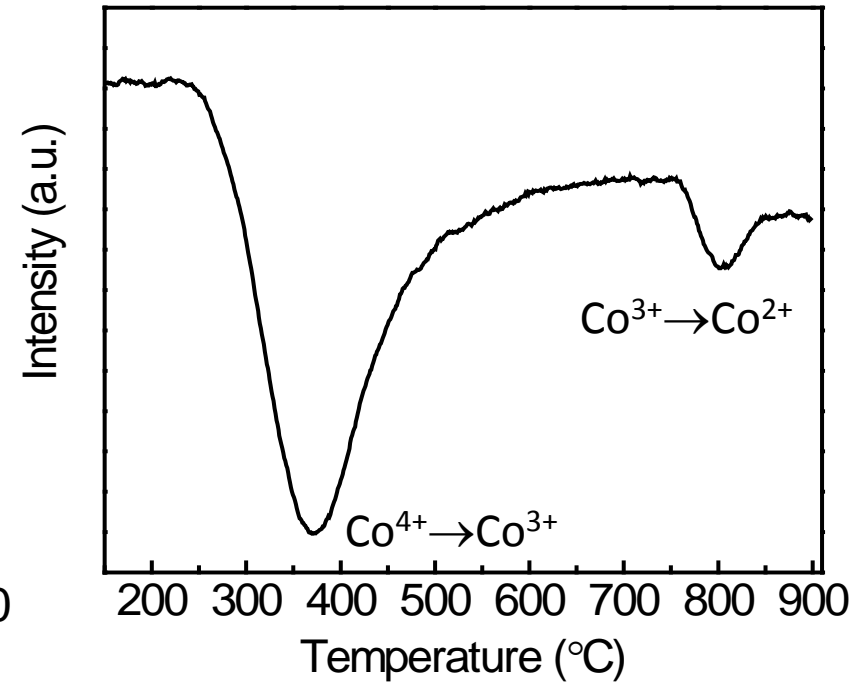
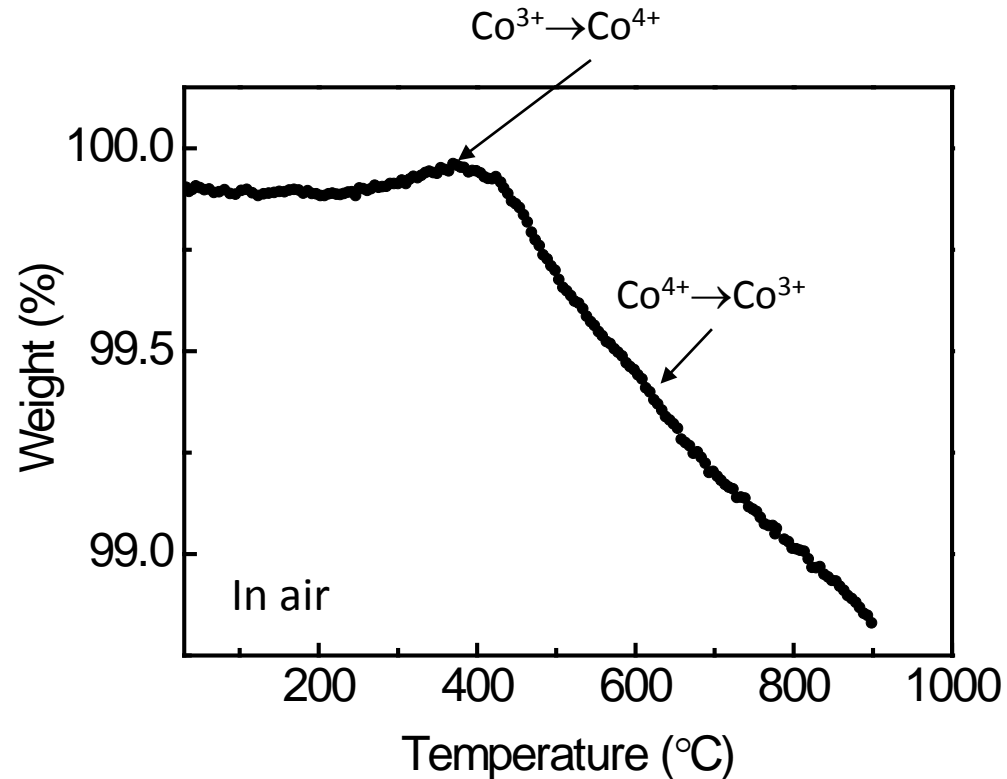
Conductivity Stability – SYC Series



Conductivity vs T – SYC Series

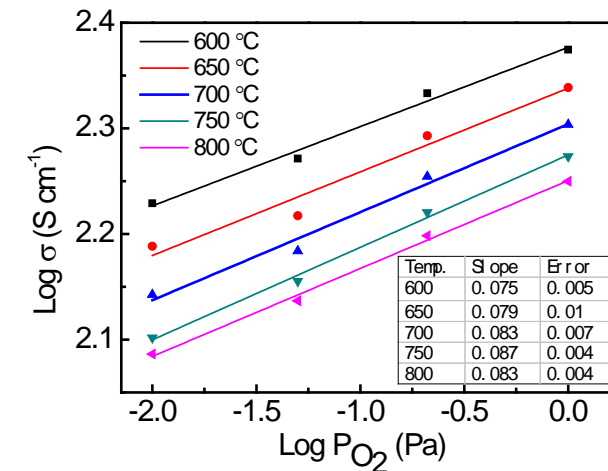
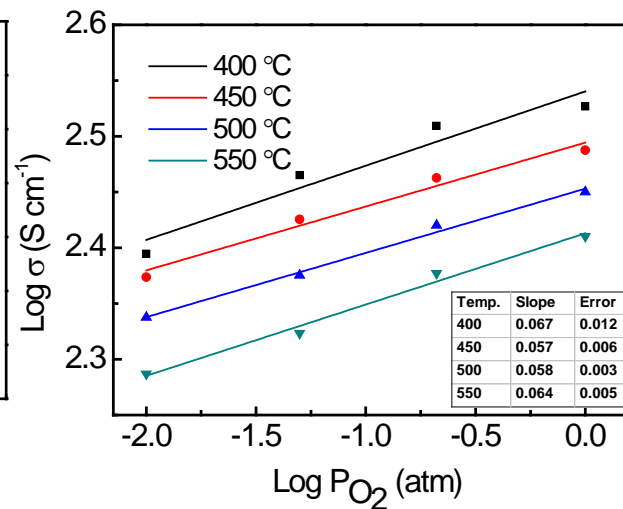
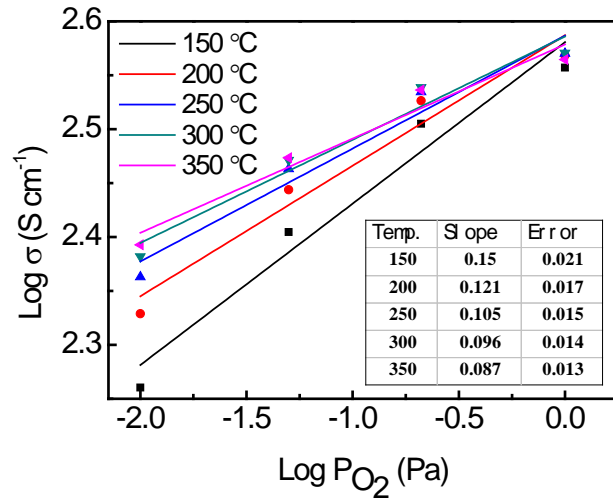


TGA and TPD: SYC Series

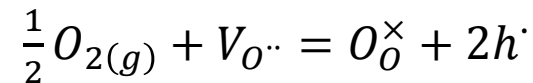


Oxygen intake and release are reversible

Conductivity vs P_{O_2} – SYC Series

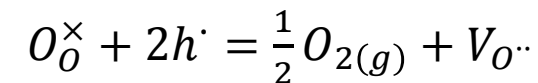


$T < 350^\circ\text{C}$,



Higher order of reaction m

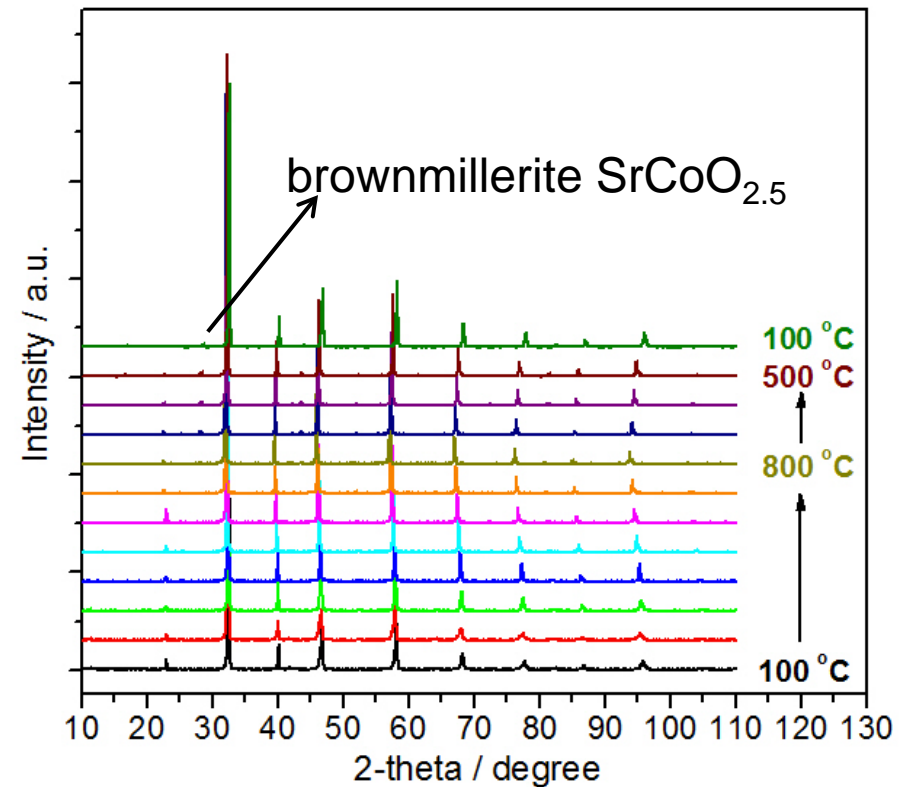
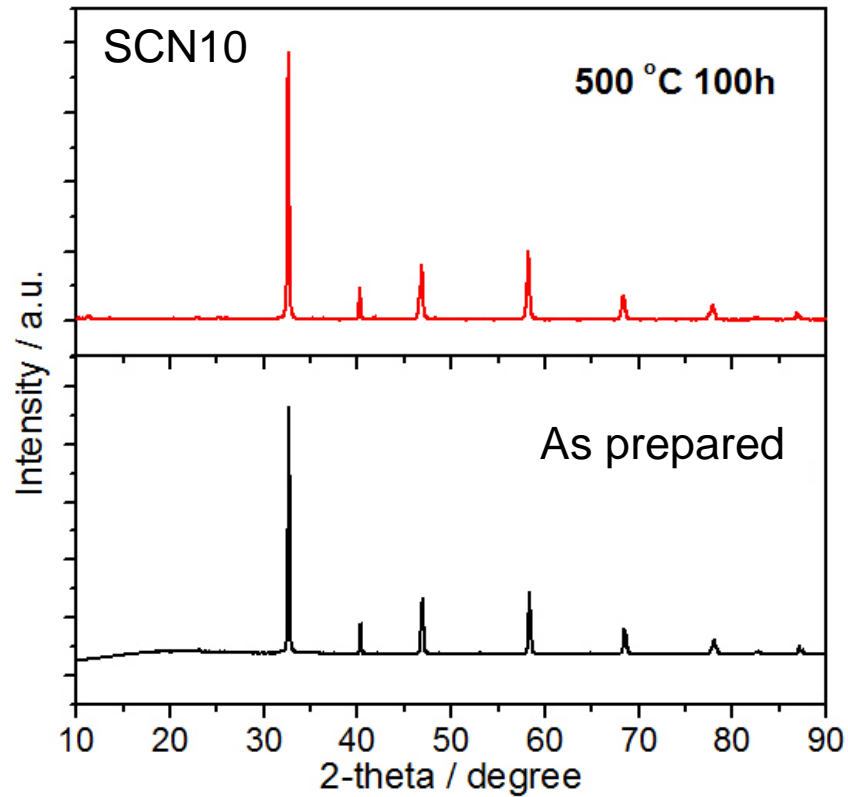
$T > 350^\circ\text{C}$,



Lower order of reaction m

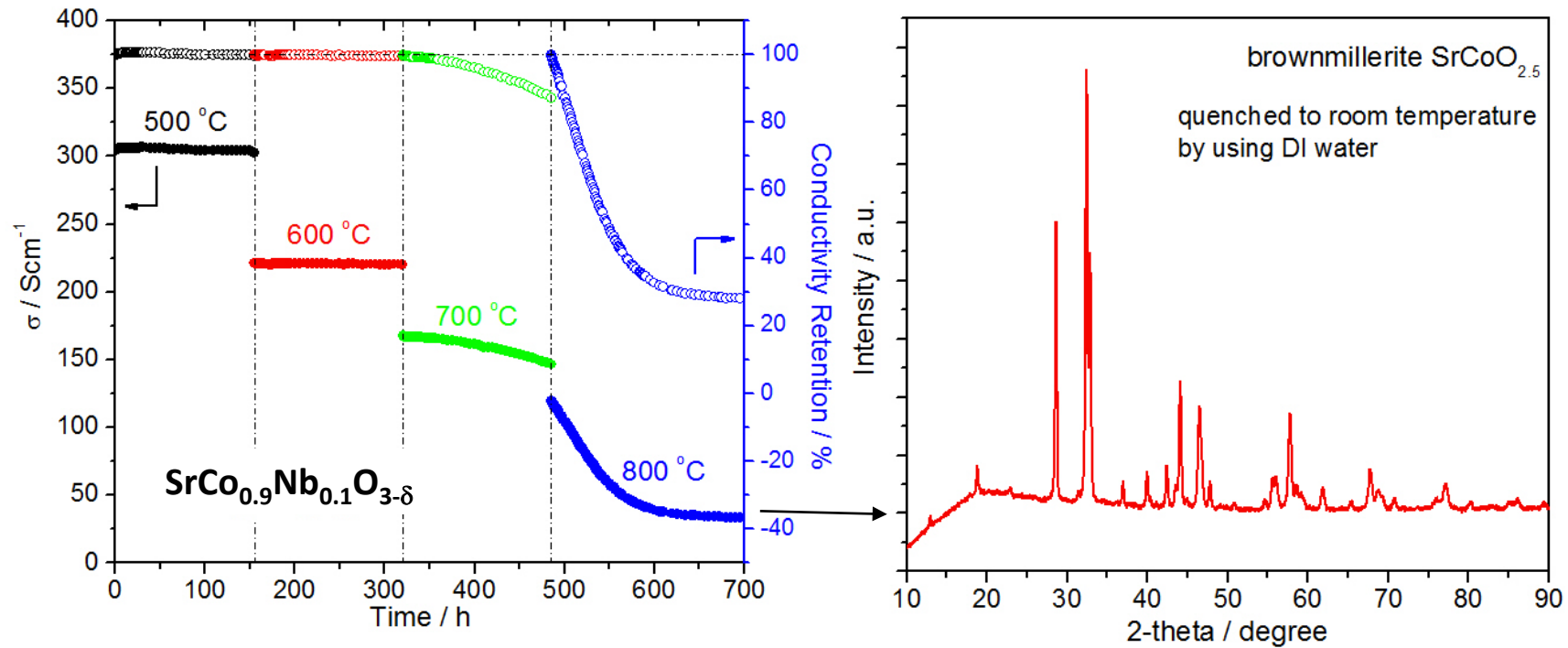
Charge neutrality: $x + 2\delta + p = n$

Phase Thermal Stability – SCN Series

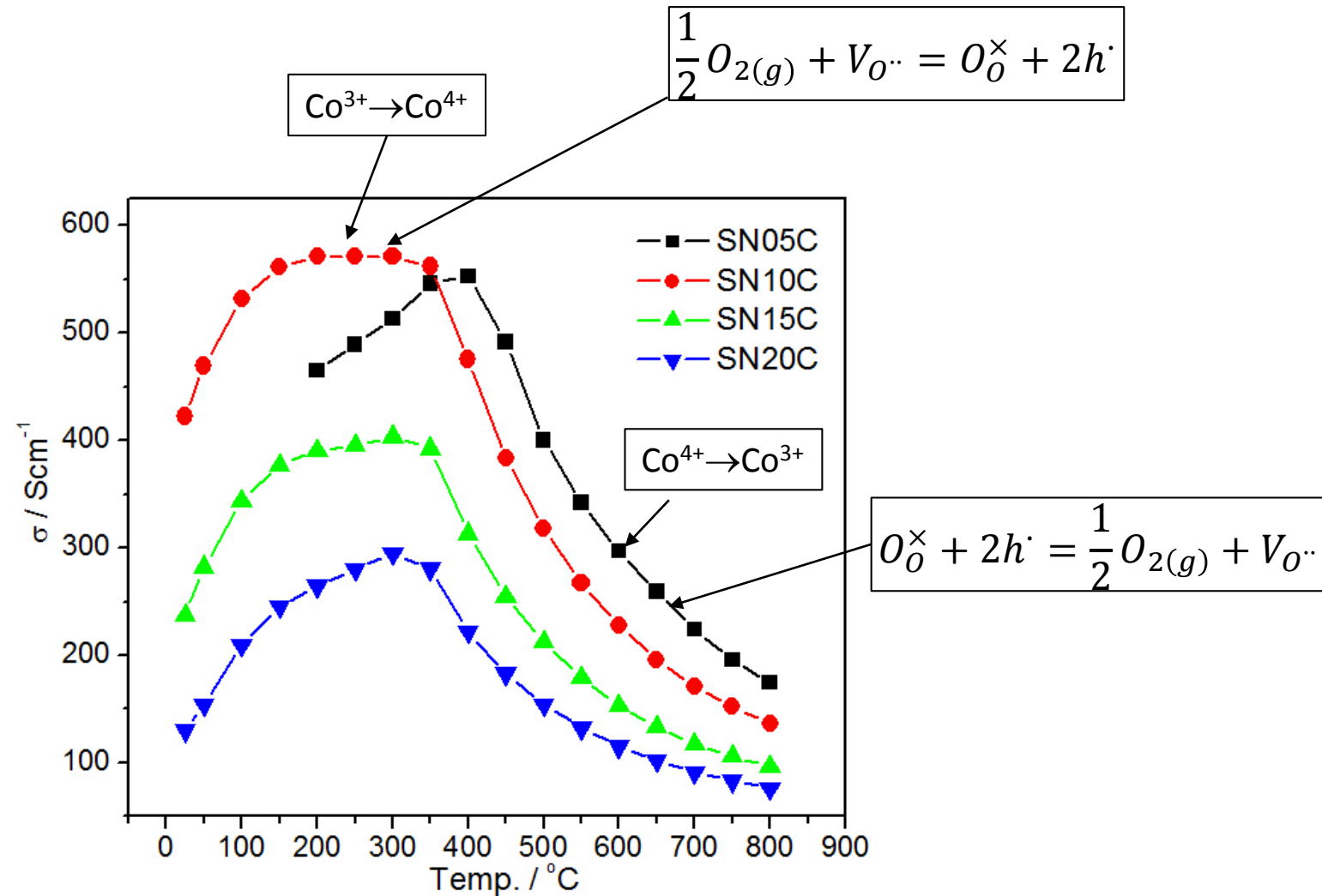


Conductivity Stability – SCN Series

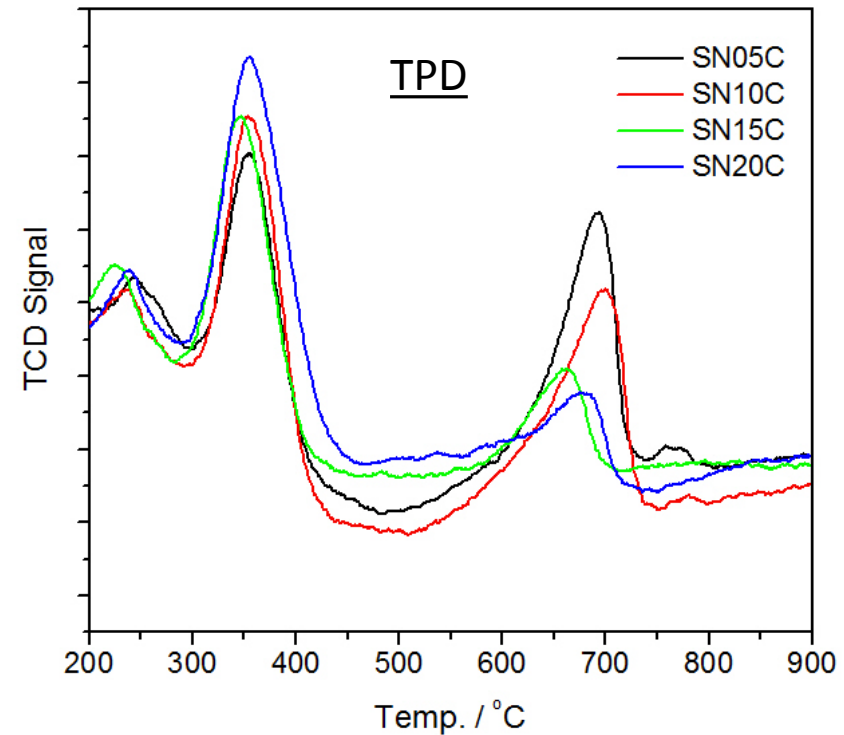
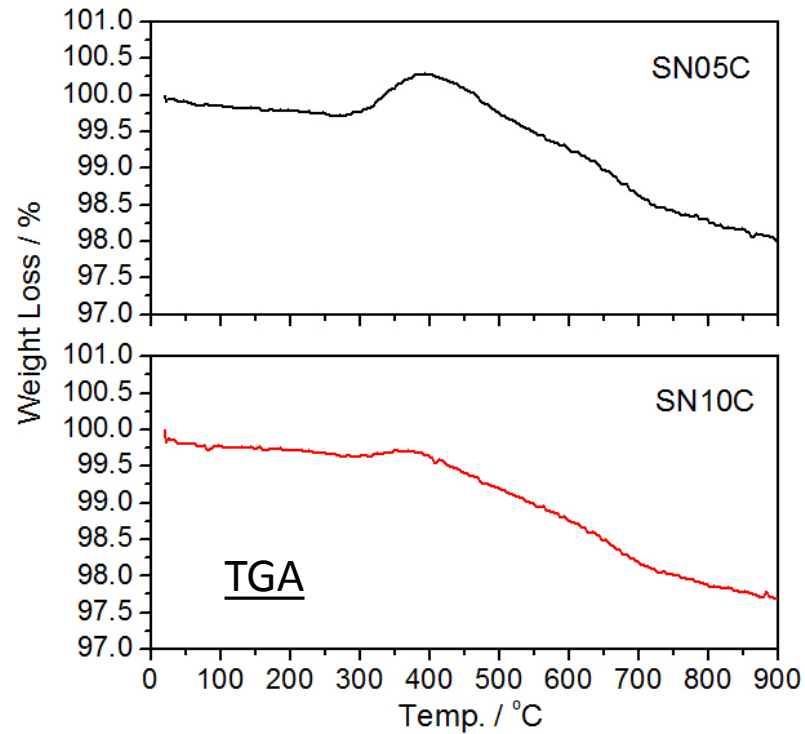
SCN10: $\text{SrCo}_{0.9}\text{Nb}_{0.1}\text{O}_{3-\delta}$



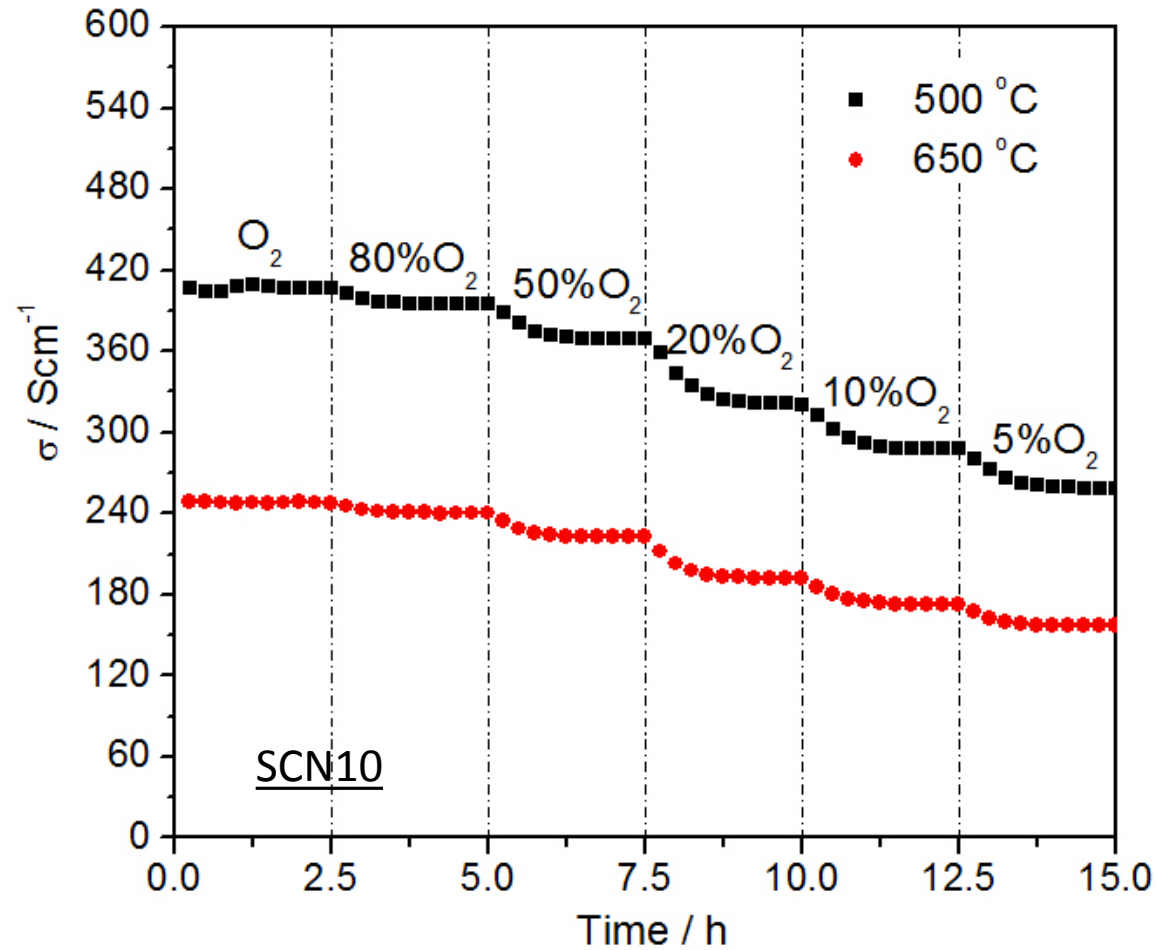
Conductivity vs T – SCN Series



TGA and TPD: SCN Series

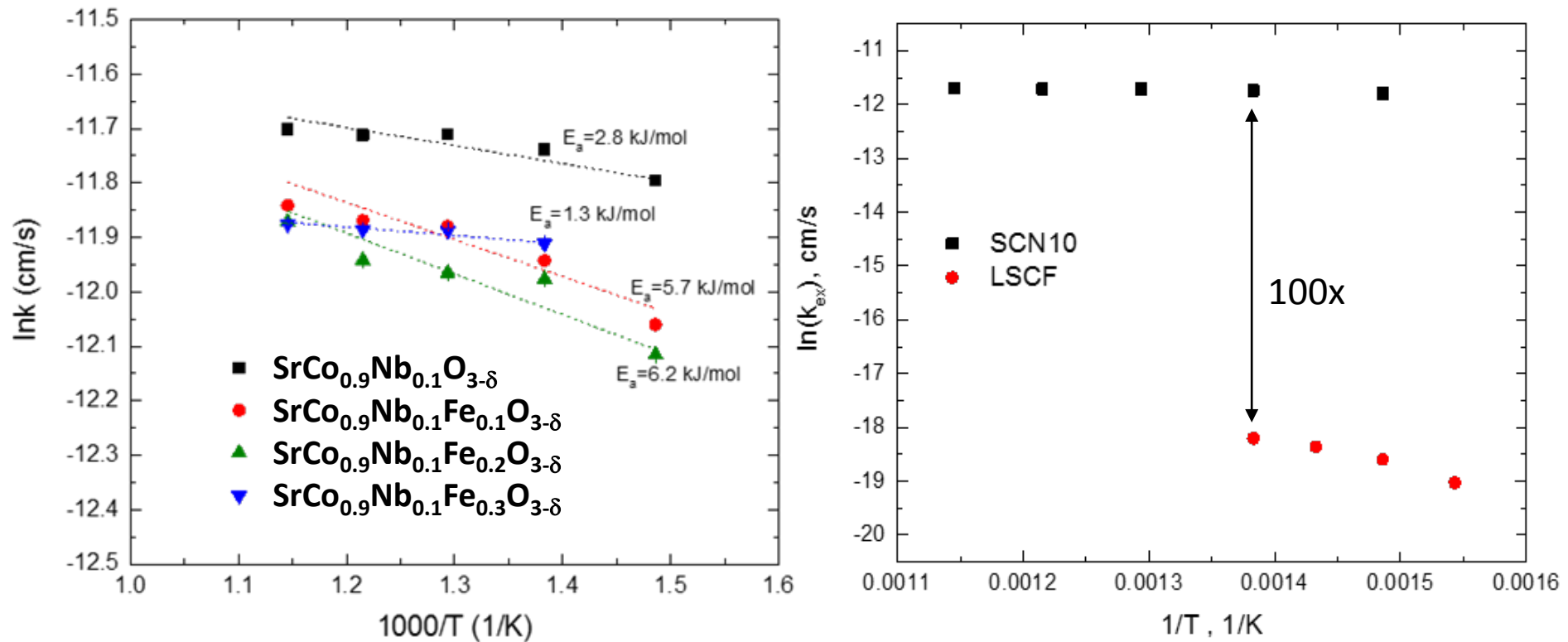


Conductivity vs P_{O_2} – SCN Series



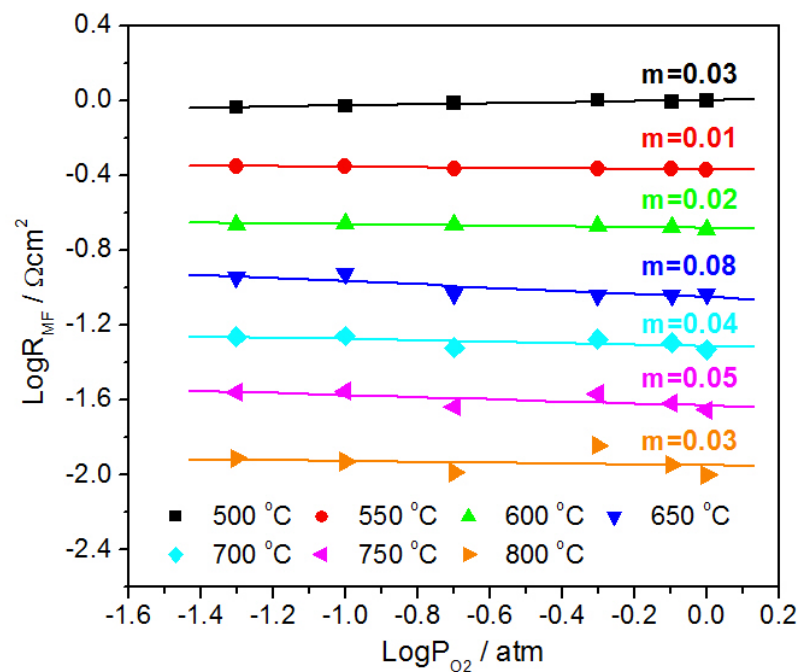
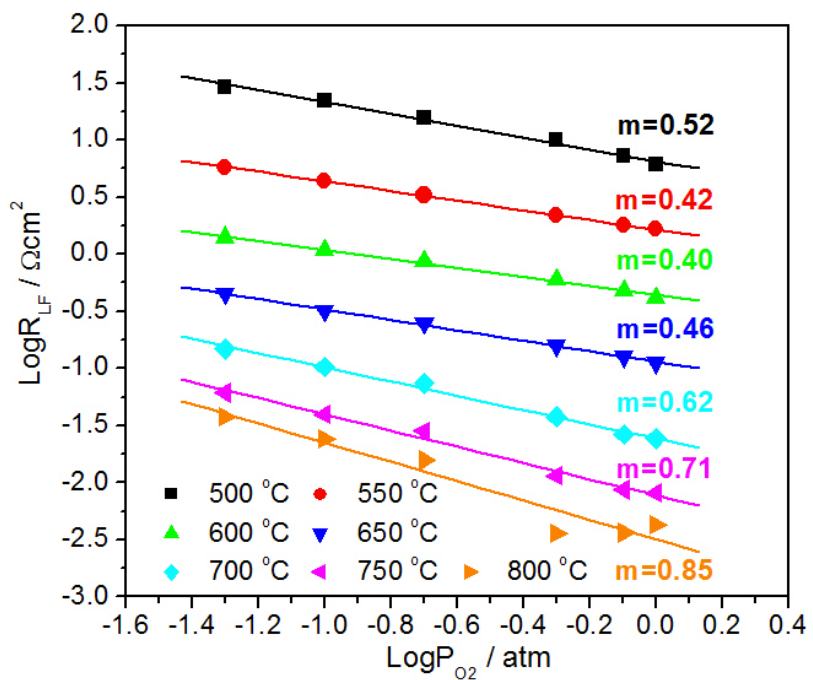
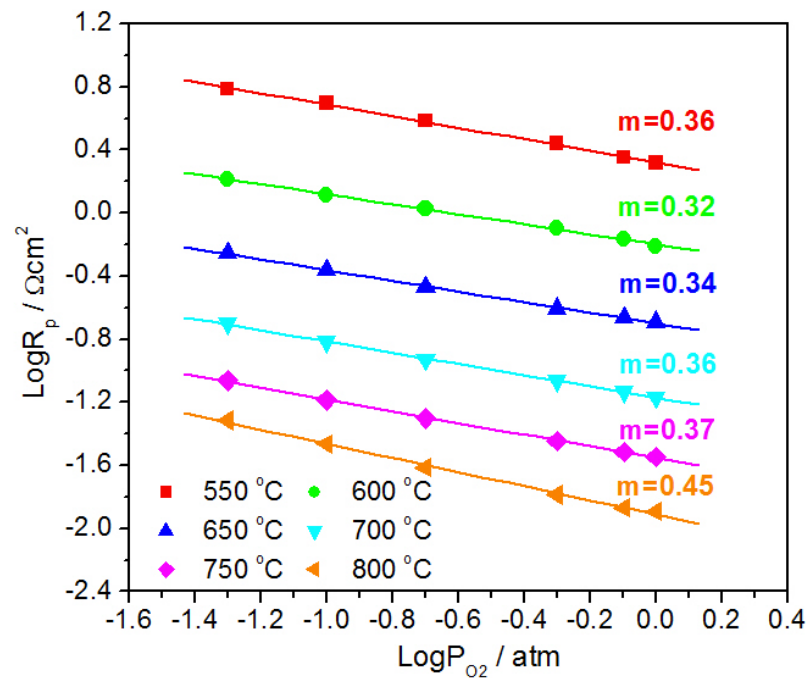
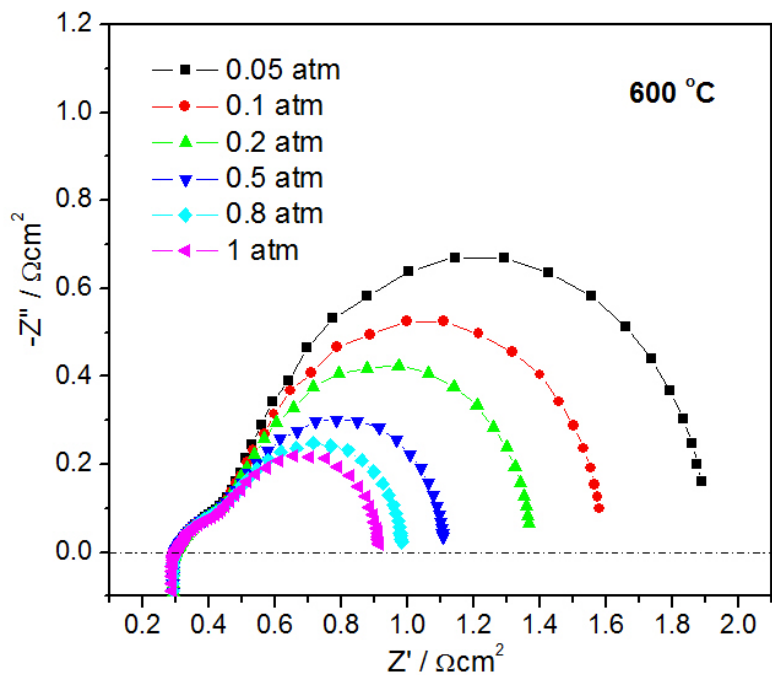
Surface O₂ Exchange Coefficient

Oxygen Isothermal Isotope Exchange

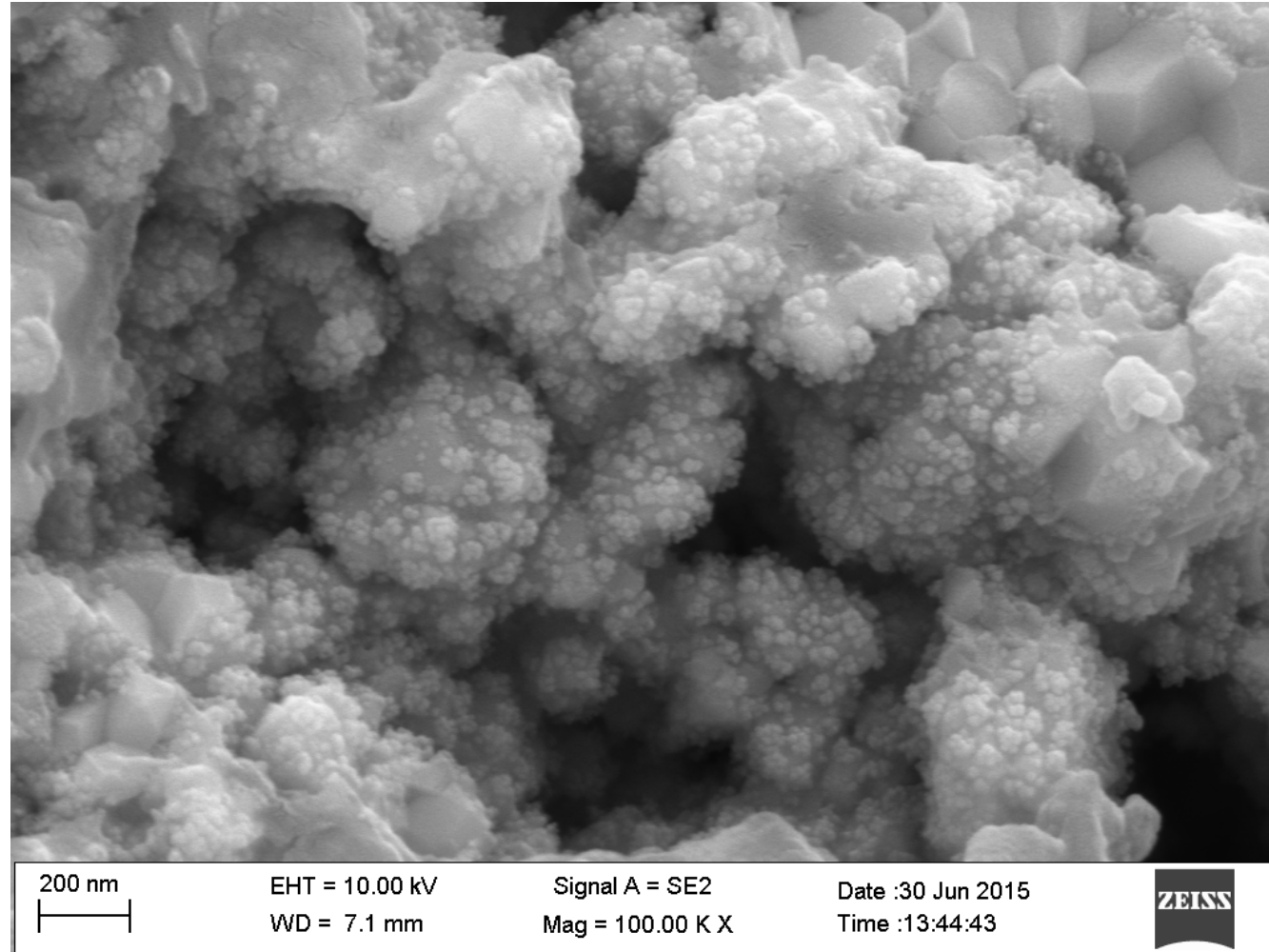


By E. Wachsman's group at University of Maryland

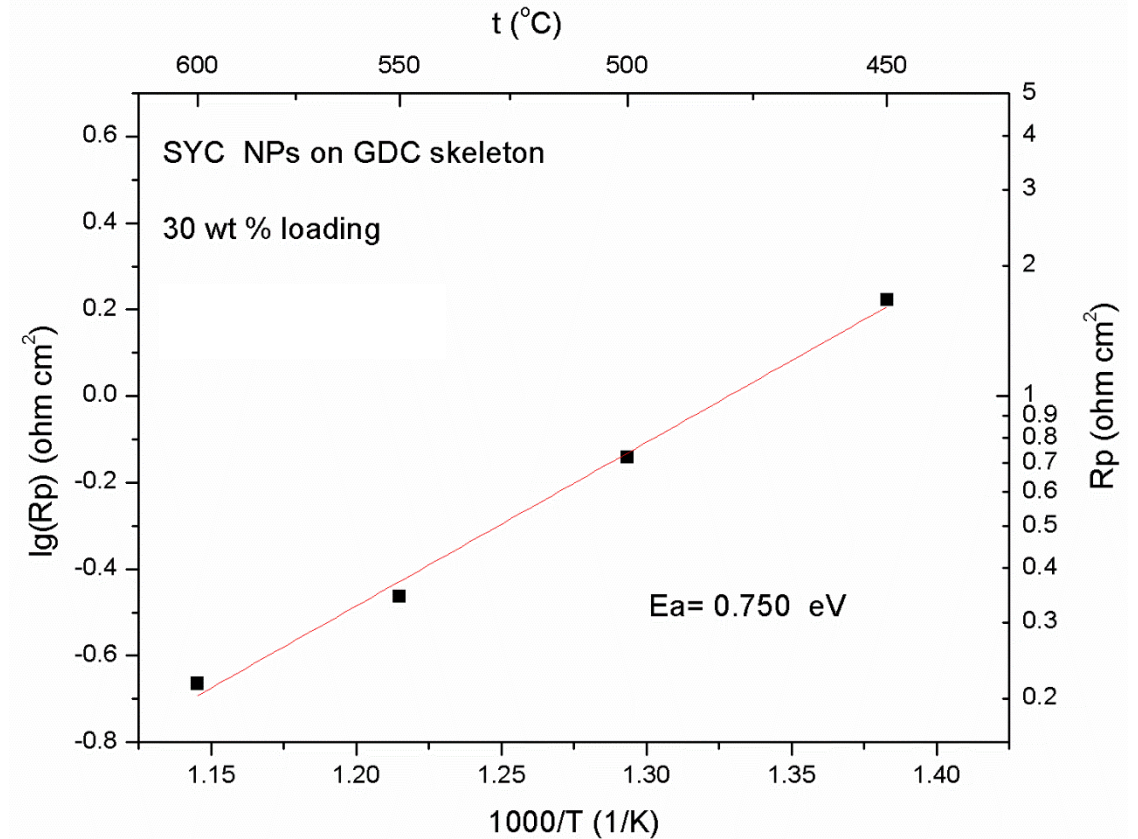
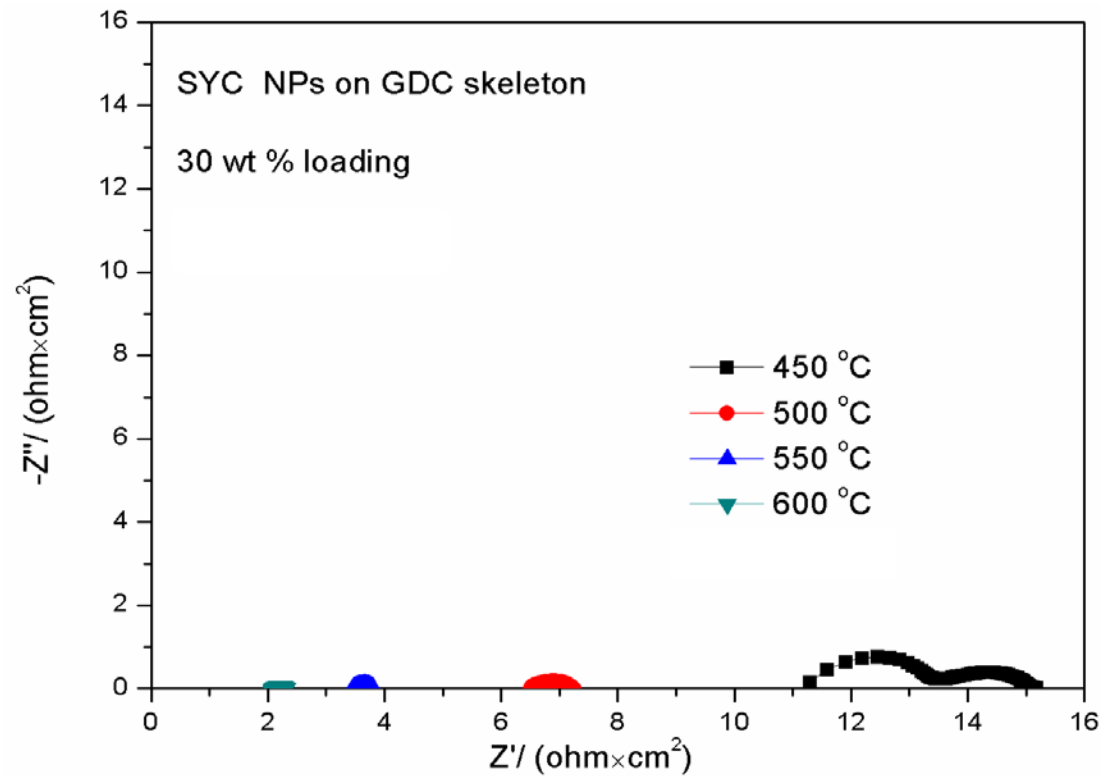
SCN Series Micron size



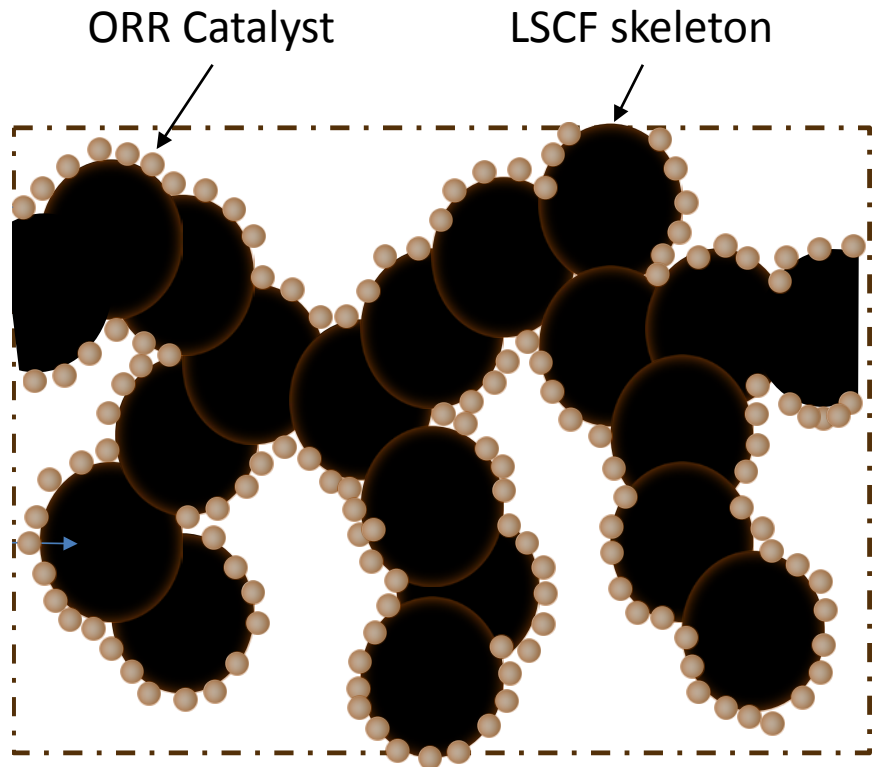
Microstructure of a Nanostructured SYC



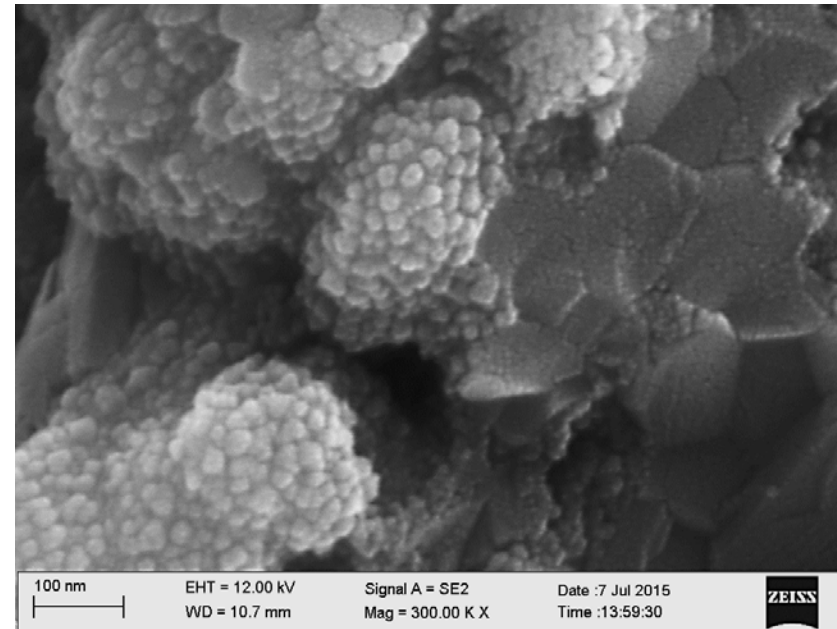
Performance of Nanostructured SYC Cathode



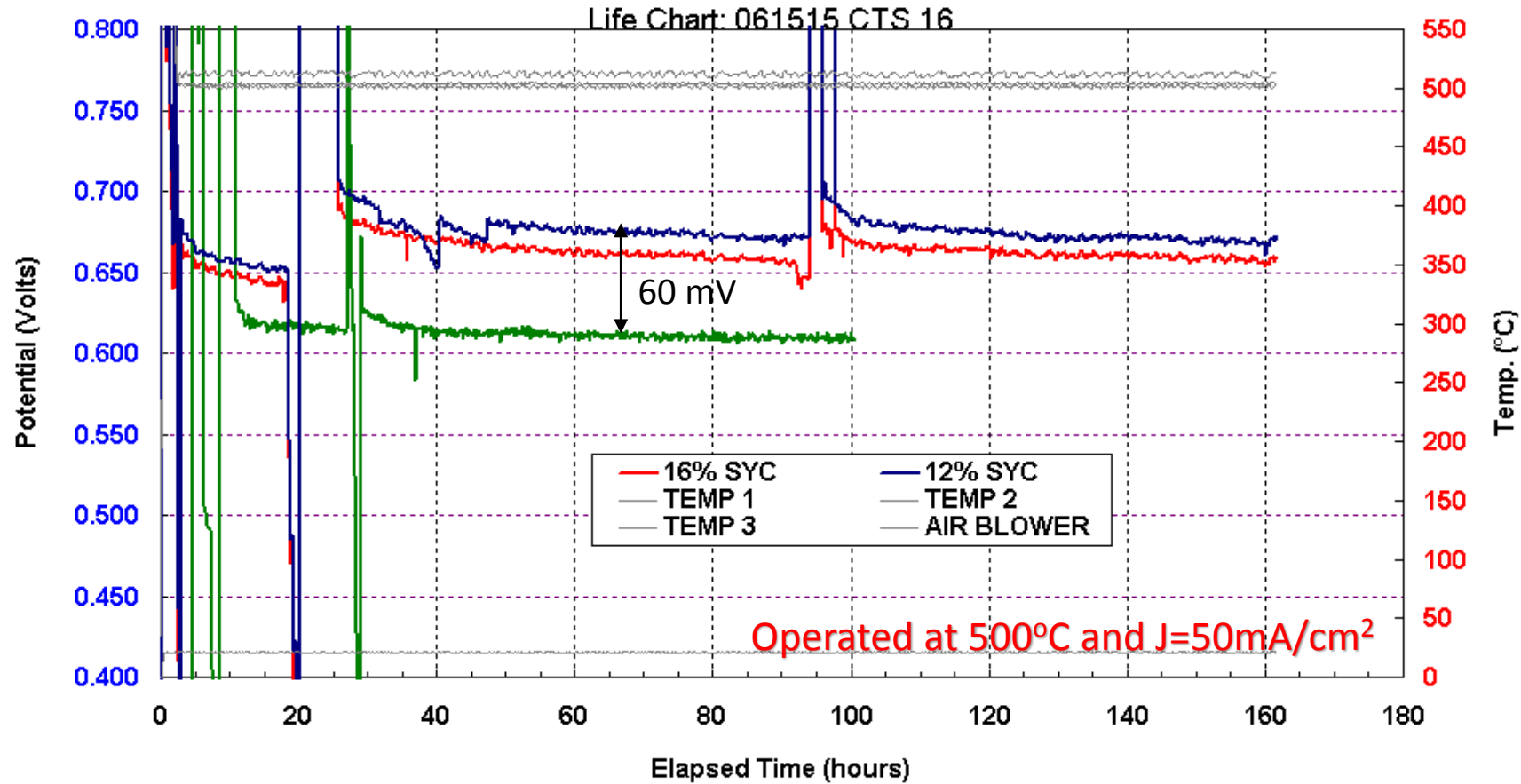
Nanostructured SYC as an ORR/OER Catalyst for Air-electrodes



SYC on LSCF



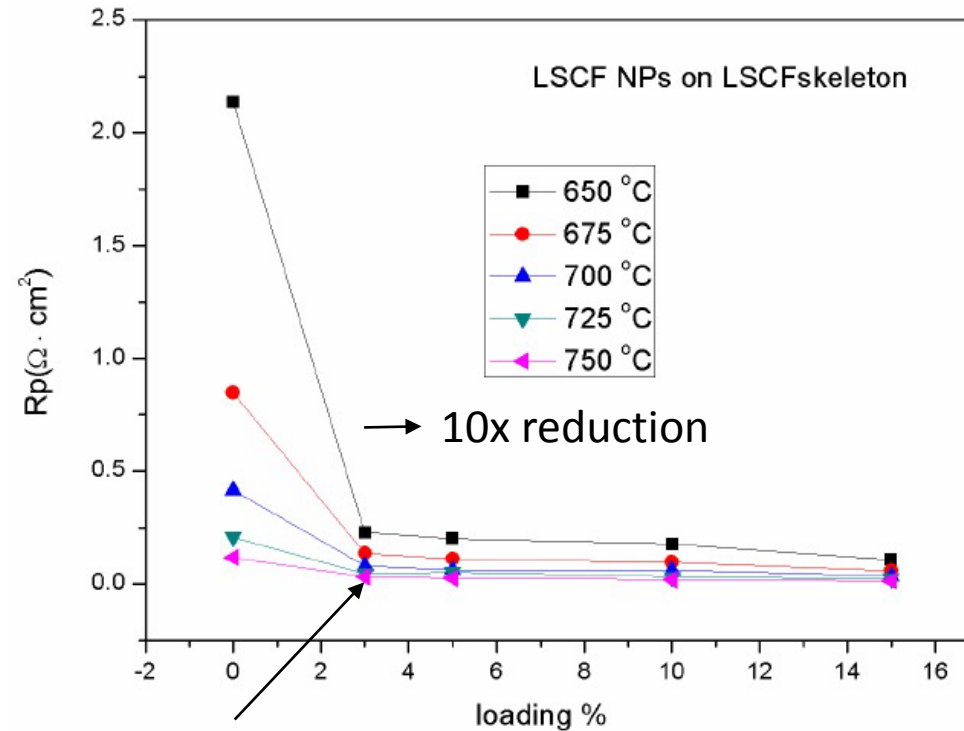
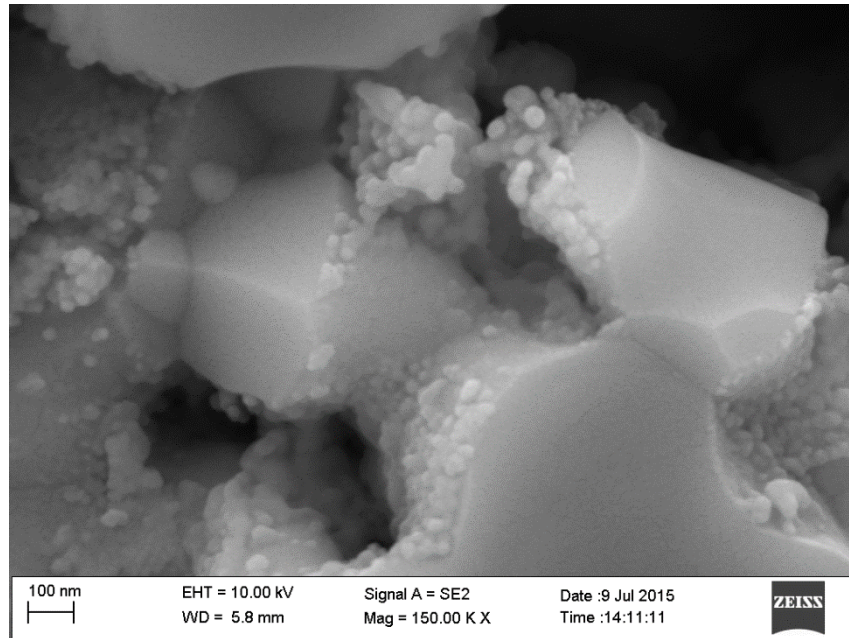
Performance Improvement by SYC Catalyst



Acumentrics 50-cm long anode-supported tubular YSZ cell
Provided by Dr. Doug Schmidt and Dr. Wensheng Wang

Decorating LSCF NPs on Commercial LSCF

5wt% loading



Single infiltration

Summary

- Integration of a redox chemical bed can enable energy storage functionality of conventional RSOFC
- Donor doped $\text{SrCoO}_{3-\delta}$ oxygen-deficient perovskites are a class of promising reversible ORR/OER catalysts for IT-RSOFCs
 - Phase stabilization by chemical doping
 - Nanostructuring by solution infiltration
 - Morphological stability by ALD
- Decorating nanoscaled SYC ORR catalysts on a commercial IT-SOFC cathode LSCF can be beneficial

Acknowledgements

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 - Acumentrics: Dr. Doug Schmidt and Dr. Wensheng Wang
 - USC: Dr. Fengzhan Si, Dr. Jie Wang, Dr. Long Jiang
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