

Bespoke Materials Surfaces

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Fireside Corrosion, Particularly under Low NO_x Firing Conditions, Limits Boiler Performance

- Temperatures up to 600°C
- Aggressive species
 - Sulfur
 - High and low pO₂
 - Alkali
 - Ash
- Thermal fatigue cracking issues with some overlays
 - decreased heat flux



weld overlay

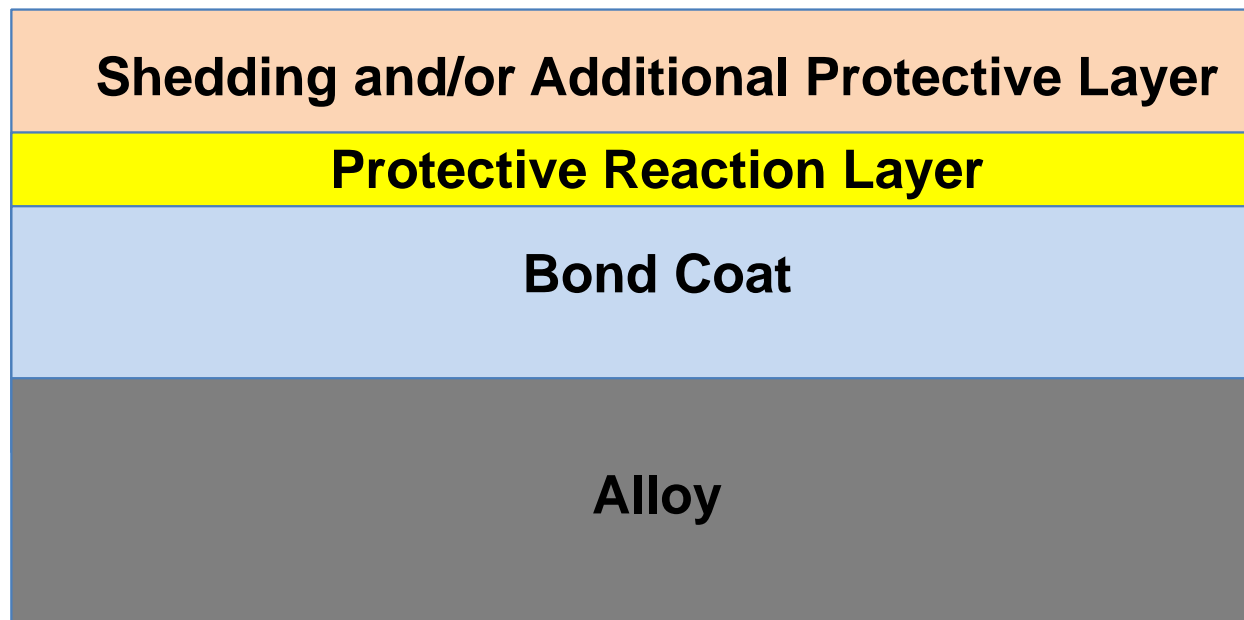
- Need better methodologies for tailoring and producing highly protective, thin, cost-effective surface treatments
- Project expected to use ORNL's experience in synthesis, high-temperature corrosion, and physical and mechanical properties to define material systems approaches to such

This Project Focuses on Fireside Surface Protection

- Objective: *Develop holistic system approaches to synthesis of **TAILORED** material surfaces that are corrosion resistant, have appropriate physical properties, and can be fabricated easily and economically*
- Based on understanding and control of key factors affecting
 - synthesis
 - performance
 - sulfidation/oxidation resistance
 - thermal expansion matches
 - thermal conductivity (thin oxide coating?)
 - ash/slag shedding (how to shed and remain durable?)

Overall Systems Approach to Achieve Properties Not Available in One Coating

Generic Example of a High-Temperature Corrosion Protection System



Approach

- **Modeling to guide material selection and help explain observations**
 - phase stabilities, reaction products
 - CALPHAD
 - Evaluate Gibbs energies of phases in the lower order systems
 - Extrapolate to higher order system to predict phase stabilities at any given conditions (T, P, wt. %, ...)
- **Experimental Verification – synthesis, characterization, performance evaluation**
 - Focus on building an affordable coating system: slurry coatings
 - Initial experimental coatings guided by what we know works
 - Characterization of metal surface properties (roughness, microstructure, composition, etc.) and coatings thereon
 - Evaluation and testing of new candidate materials/surfaces
- **Coordination with other Fossil Advanced Research Materials (ARM) projects**

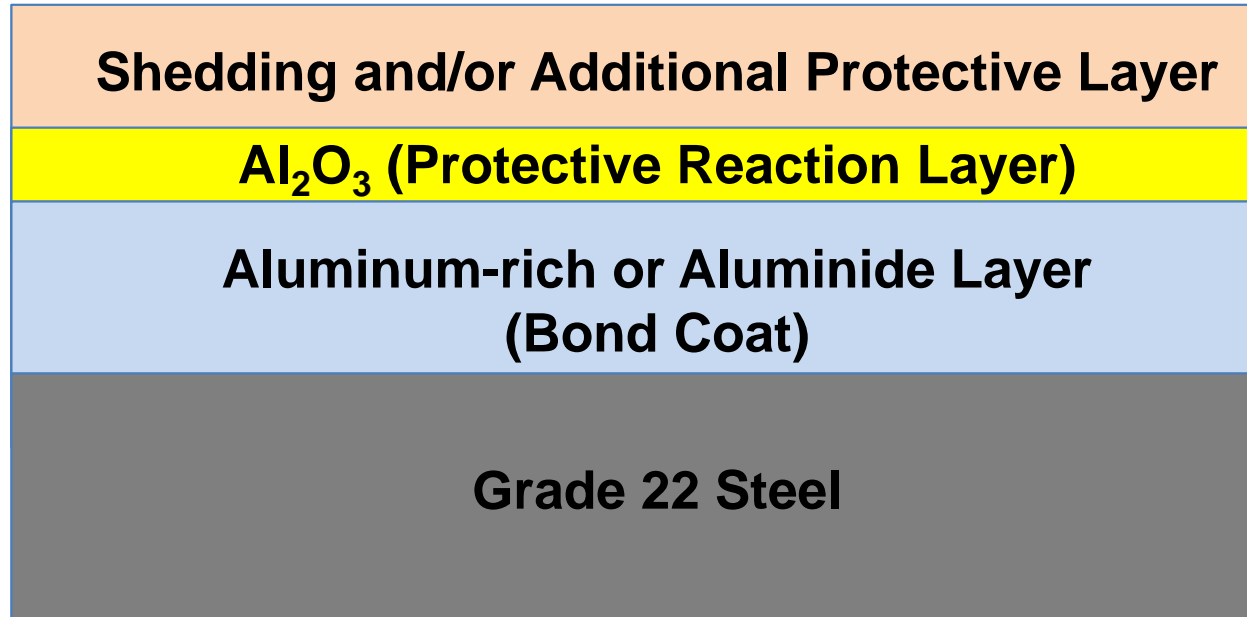
Project failed Go/No-Go milestone: Ending September 2014

- Complete experimental and modeling work on approach to multi-functional surface modification of steels and produce first set of specimens using such. (8/30/13)
- Complete a Go/No-Go analysis on effectiveness of approach to multi-functional surface modifications (9/30/13)
- Complete a review paper of the major research worldwide as well as the state of the art commercial coatings for boiler and high temperature heat exchanger tubes (7/30/13)
- Complete evaluation of the physical properties and coal ash corrosion resistance of a set of surface modified steels and assess effectiveness. (9/30/13)
- Assess effectiveness of coating approach in preferring predicted performance improvement and define next steps (9/30/13)

This Project Focuses on Fireside Surface Protection

- Objective: *Develop holistic system approaches to synthesis of **TAILORED** material surfaces that are corrosion resistant, have appropriate physical properties, and can be fabricated easily and economically*
- Based on understanding and control of key factors affecting
 - synthesis ← work was hung up here
 - performance ← never got to here
 - sulfidation/oxidation resistance
 - thermal expansion matches
 - thermal conductivity (thin oxide coating?)
 - ash/slag shedding (how to shed and remain durable?)

Starting Point was Aluminum-Containing Bond Coating



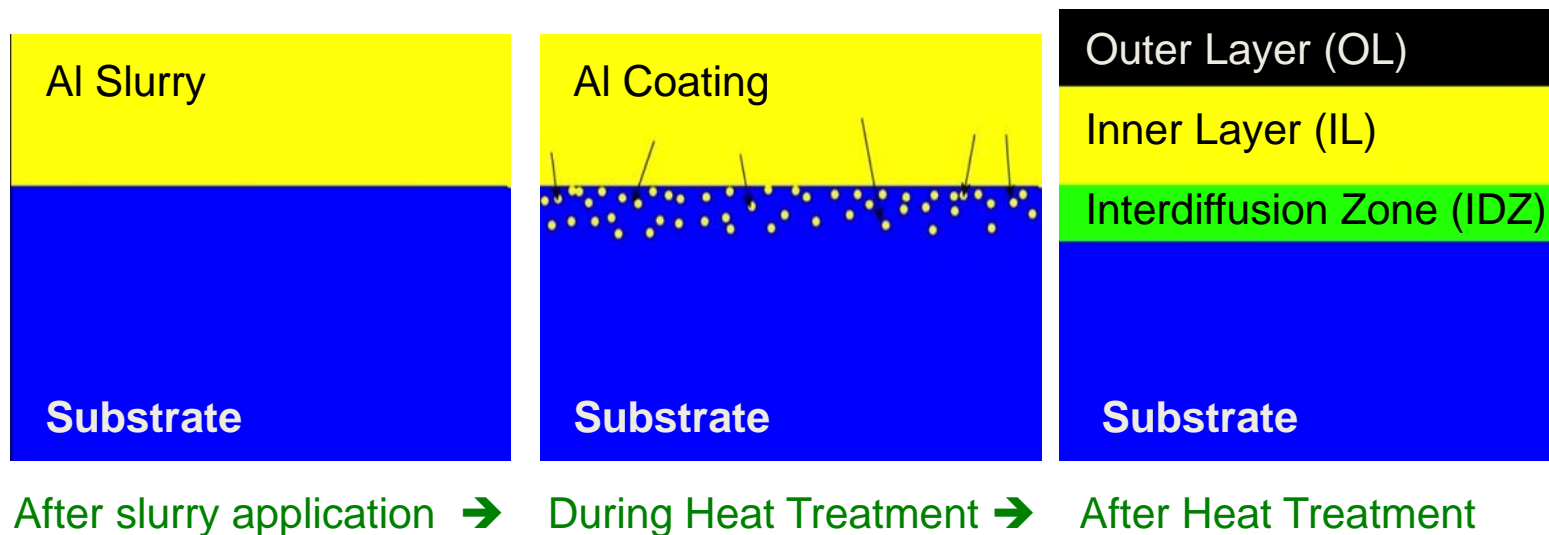
Slurry coating low Cr steel was easier said than done

First Layer: Al-rich Coatings on Gr22 Steel (2.25Cr-1Mo), Relevant to Waterwalls

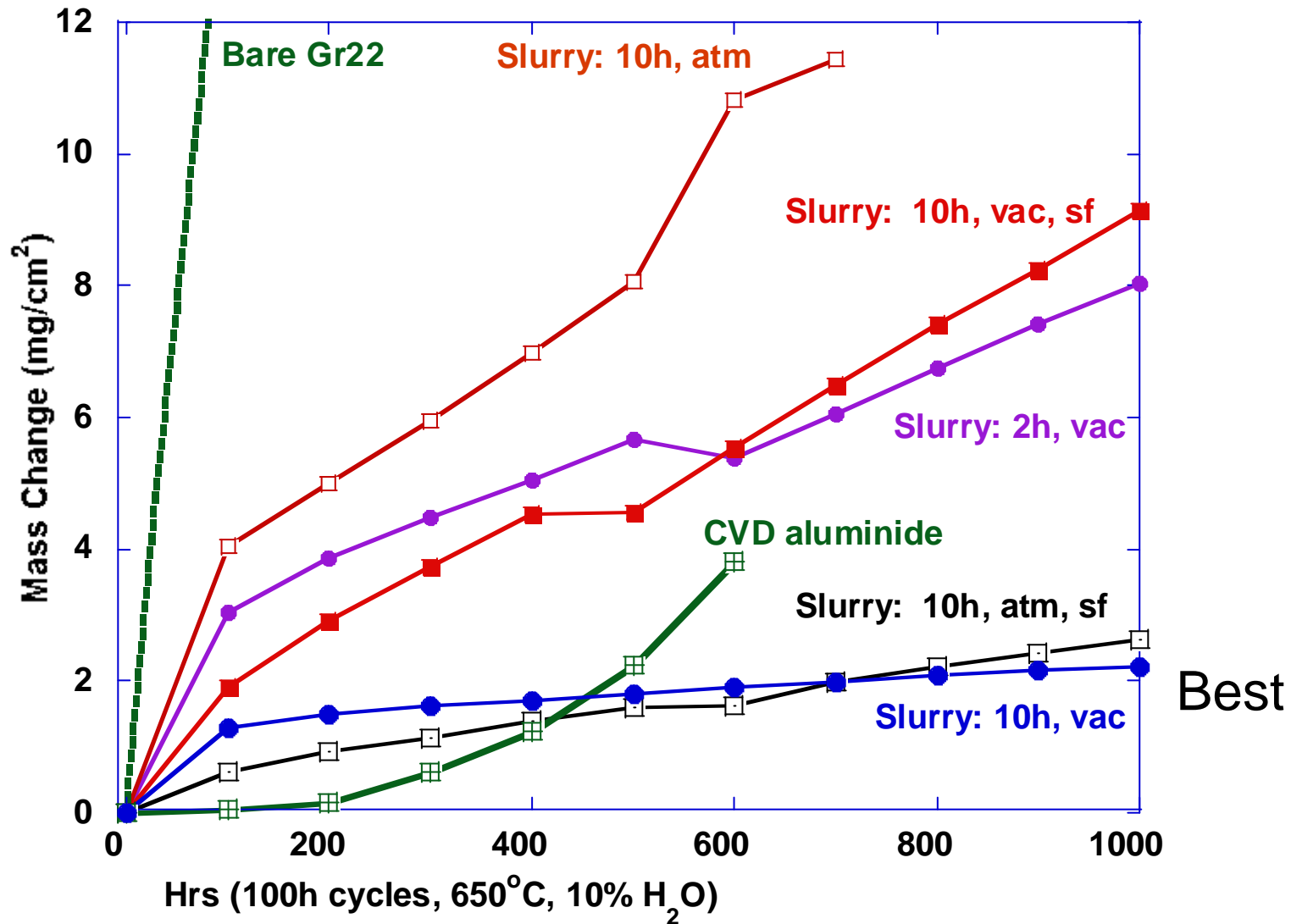
- Building on a foundation of previous FE-ARM aluminide work at ANL, ORNL, Lehigh, Foster-Wheeler, and EPRI using a variety of synthesis techniques
 - Good-to-excellent in highly reducing conditions, oxidation-sulfidation, water vapor
 - Problematic with ash
- ORNL's slurry aluminide coating process as an easy-to-apply route
- Examination of resulting microstructures
 - Formation of stable surface dependent upon processing conditions
 - Coating efficacy dependent upon alloy composition
- Testing
- Refine and tailor
- Examine other compositions and structures

Integrated Architecture is more complicated than earlier schematic

- Slurry is applied
- Heat treatment leads to diffusion of Al into the substrate (enriched surface and/or formation of another phase)
- Excess Al is removed after heat treatment
- In high-temperature environment, protective Al-rich scale forms

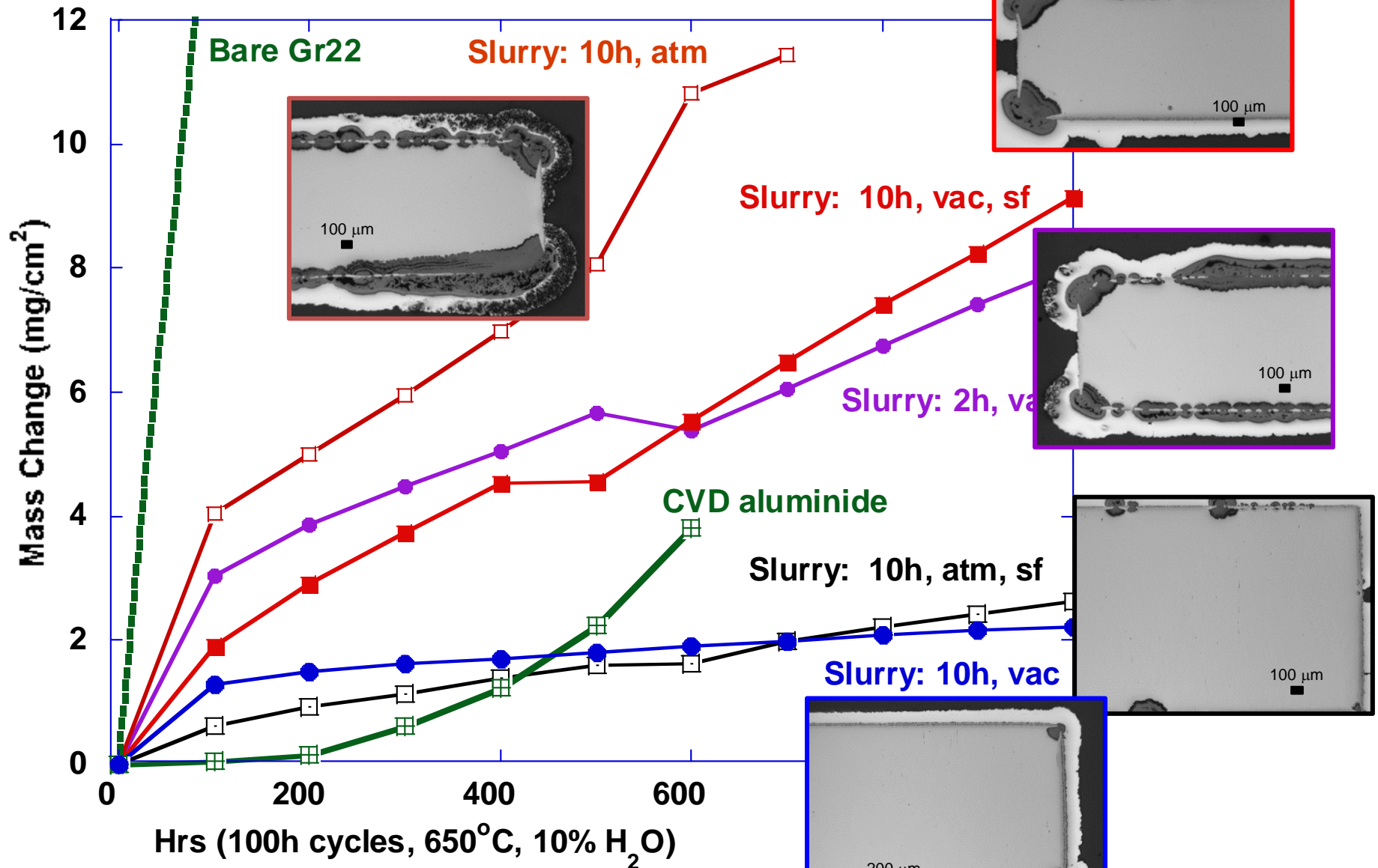


FY13: Parametric study of slurry coating process yielded no clear path forward



Vac = annealed at 610 Torr
Atm = annealed without vacuum
sf = surface finishing after annealing

FY13: Slurry Aluminides on Gr22



Vac = annealed at 610 Torr

Atm = annealed without vacuum

sf = surface finishing after annealing

FY14 Wrap-up Activities

Complete work by September 2014.

- 1. Prepare review article on state of the art for coatings for boilers and high temperature heat exchanger tubes (Milestone #1).
 - Completed January 2014 (submitted as ORNL TM report)
- 2. Additional fabrication, oxidation testing and characterization of slurry coatings to better define preferred coating strategies.
 - In progress
- 3. Complete a peer-reviewed publication highlighting the effect of the role of coating processing conditions on the efficacy of oxidation-resistant slurry diffusion coatings (Milestone #2).
 - September 2014
- 4. Proof of concept: Deposit slurry YSZ or mullite over an alumina-forming alloy for ash testing.
 - September 2014

Highlights from Literature Review

- Reviewed literature on slurry aluminides from 1960's to present.
- Slurry work in the 1960's and 70's focused on coatings for Ni- and Co-based alloys.
 - Maxwell and Gabriel (United Aircraft Corp.) patented a **FeCrAlY slurry coating** for Ni- and Co-based superalloys for the USAF (D. H. Maxwell and J. M. Gabriel, "Slurry Coating Superalloys with FeCrAlY Coatings," U.S. Patent # 3,741,791, 1973).
 - **Slurry Fe-Cr-Al could be an area of potential future interest.**
- Very little literature in this area in the 1980's.
- In the 1990's United Technologies Corp. and Sermatech patented and published modifications to Al-based slurry technologies.
- In 1996, the National Research Council formed a committee on Coatings for High-Temperature Structural Materials and published a report on the trends and opportunities. Coatings for both metals and ceramic systems were highlighted. However, overlay coatings were the focus and **slurry coating techniques were not discussed.**

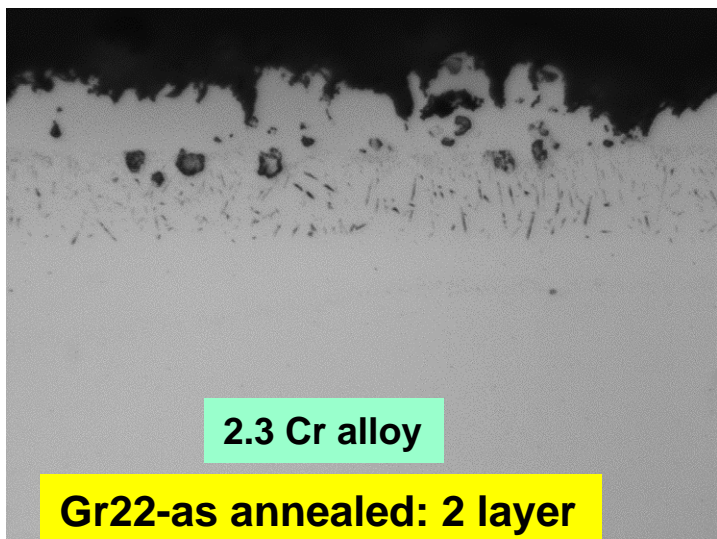
Highlights from Literature Review

- In the early 2000's, a move from superalloys to alternative, iron-based coatings for steel was evident - particularly for slurry-applied coatings on ferritic steels.
 - Work was driven primarily by the European COST Action 522: Ultra Low Emission Power Plant into the 21st Century.
 - The work of Agüero et al. at the Instituto Nacional de Tecnica Aeroespacial in Spain demonstrated a resurgence of interest in slurry-based coatings for metallic substrates.
- DOE funded work on ceramic and metallic slurry coatings.
 - Included slurry aluminizing of cast austenitic stainless steels at ORNL.
- Sermatech introduced vapor phase slurry coating process.
- Slurry aluminizing of Ti-based alloys has been reported in the past decade.
- Sixteen currently active US industrial sources of slurry coating technology were identified in the review.

FY14: 650°C Cyclic Oxidation Evaluation

- Coatings: Slurry aluminide dip coatings (aqueous, ORNL)
- Alloys coated
 - Gr. 22 (Fe-2.3wt.%Cr-0.9Mo)
 - Gr. 315 (Fe-2.9Cr-0.7Mo-1.7W)
 - Gr. 91 (Fe-8.3Cr-0.9Mo)
 - CF8C-Plus (cast aust. steel) (Fe-19.2Cr-12.8Ni-4Mn-0.25N)
- Slurry coating heat treatment (best conditions from FY13)
 - Annealing: 750°C
 - Time: 10 h
 - Environment: argon (Ar) at 610 Torr
- Oxidation Testing Conditions:
 - 650°C, 100 h cycles, 10% water vapor
 - Low cost initial evaluation, rapid attack of base alloy in wet air

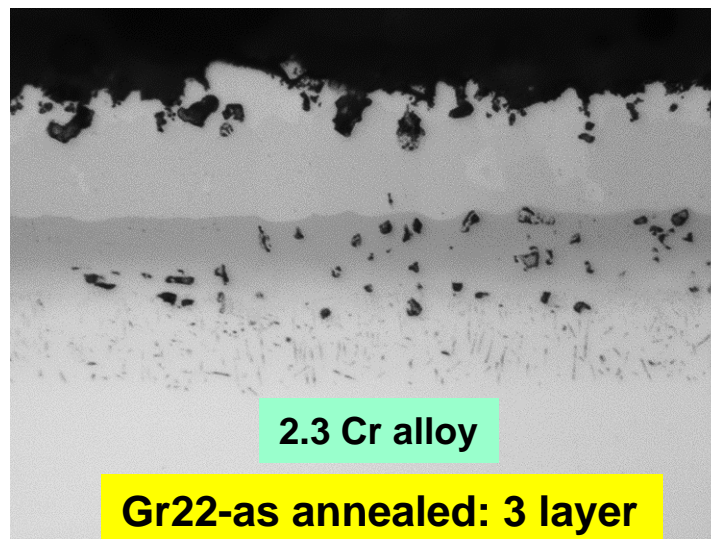
Microstructures of FY14 As-Deposited Slurry Aluminide Coatings (750°C, 10h, 610 Torr)



14-0598-07

T22-1
Al031417, 0hr

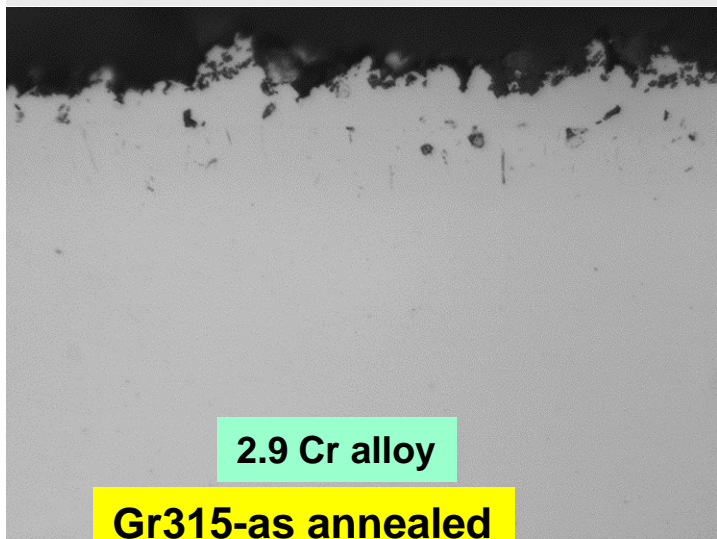
1000X 5µm
as polished



14-0598-08

T22-1
Al031417, 0hr

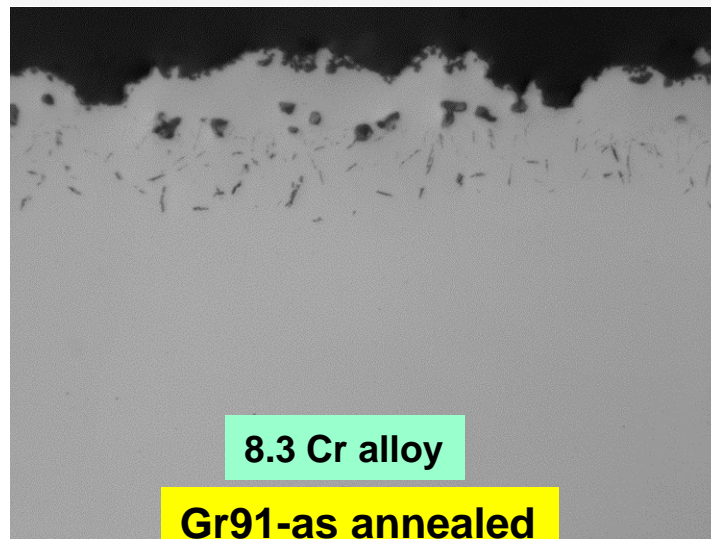
1000X 5µm
as polished



14-0600-07

T33-1
Al031417, 0hr

1000X 5µm
as polished



14-0602-07

T91-3
Al031417, 0hr

1000X 5µm
as polished

High-Al outer layer

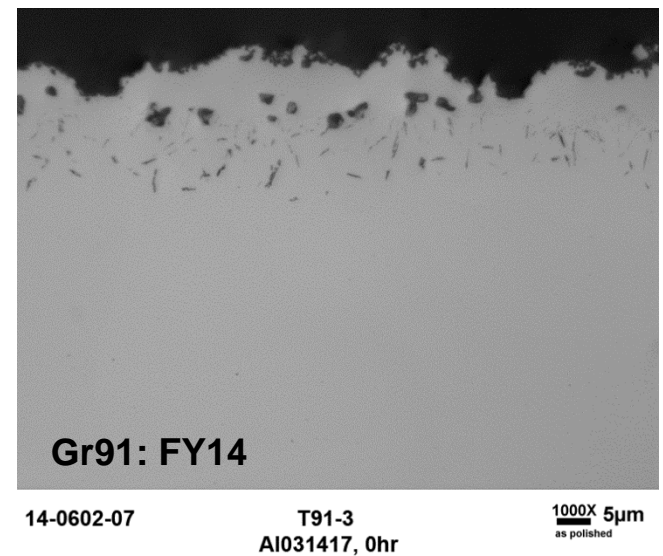
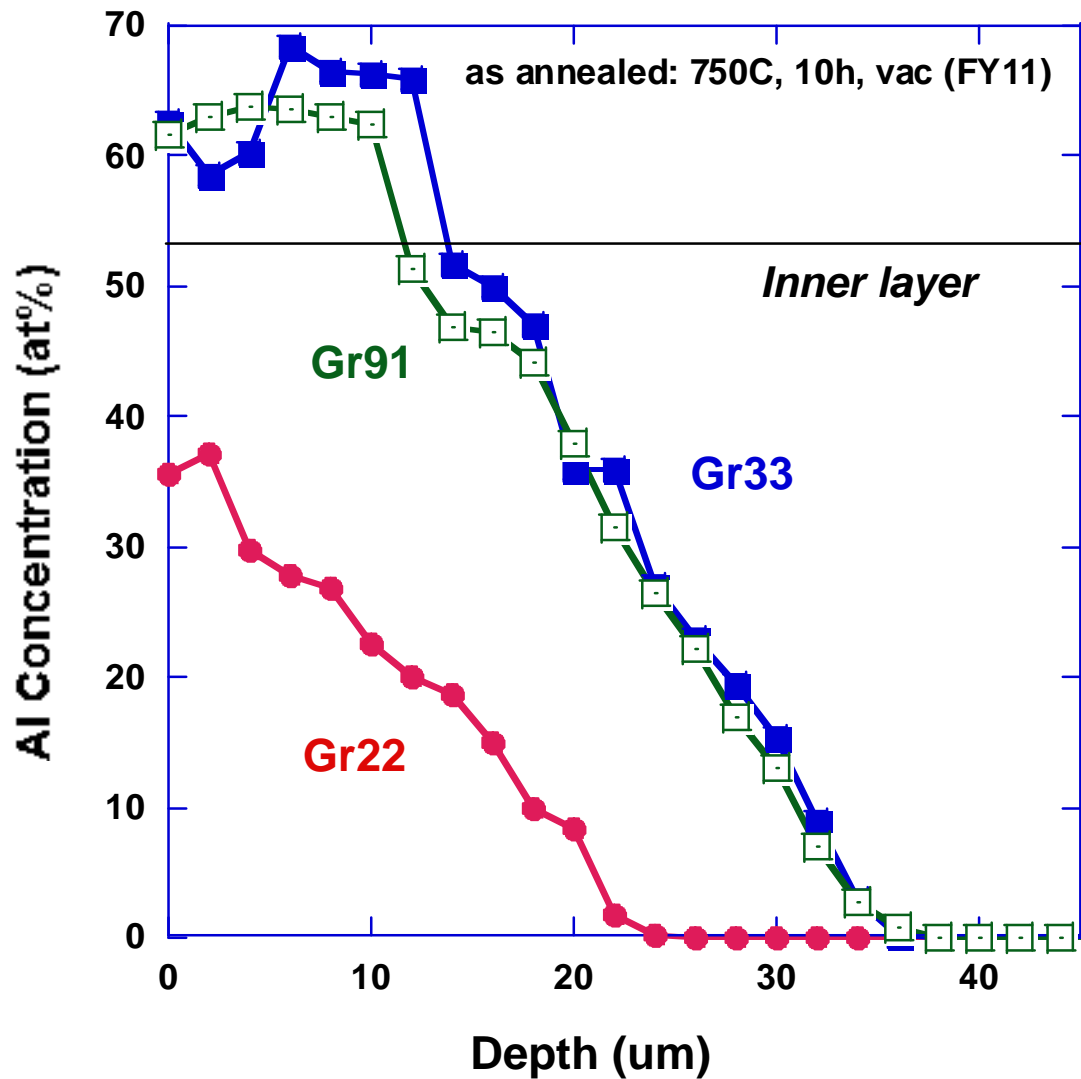
Porous inner layer

IDZ

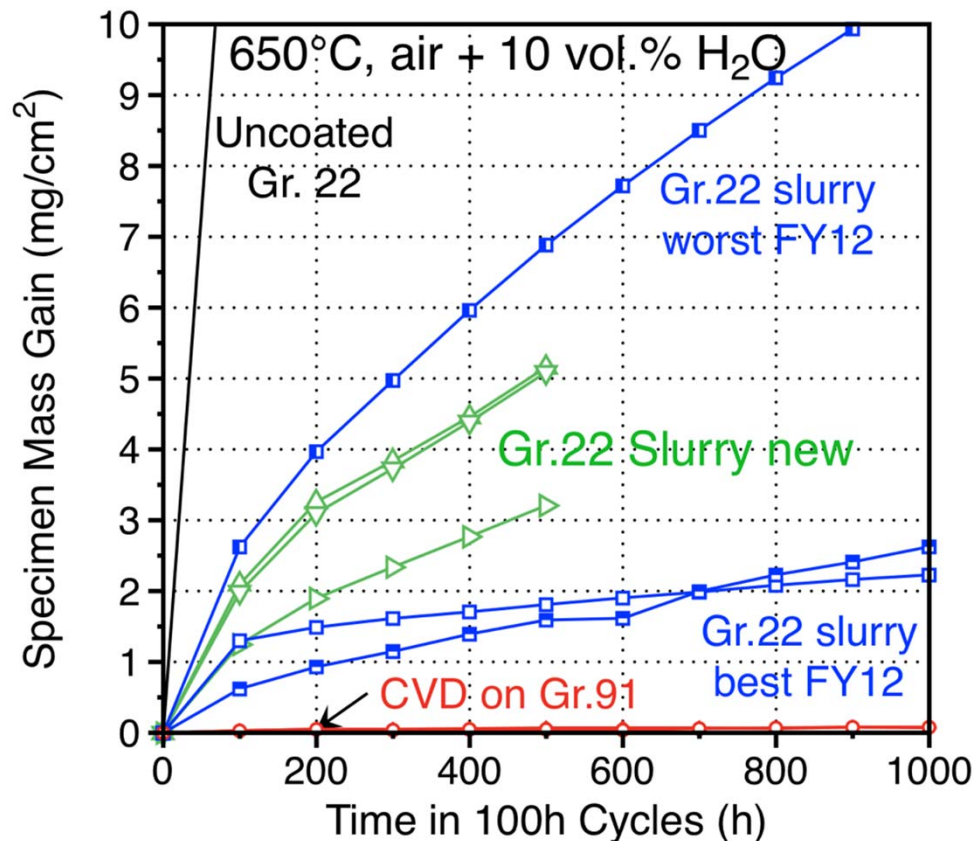
Porous inner layer

IDZ

Elevated Al contents of slurry coating inner layer on higher-Cr ferritic steels



650° C oxidation behavior of slurry coated Gr.22 was similar to FY12 results

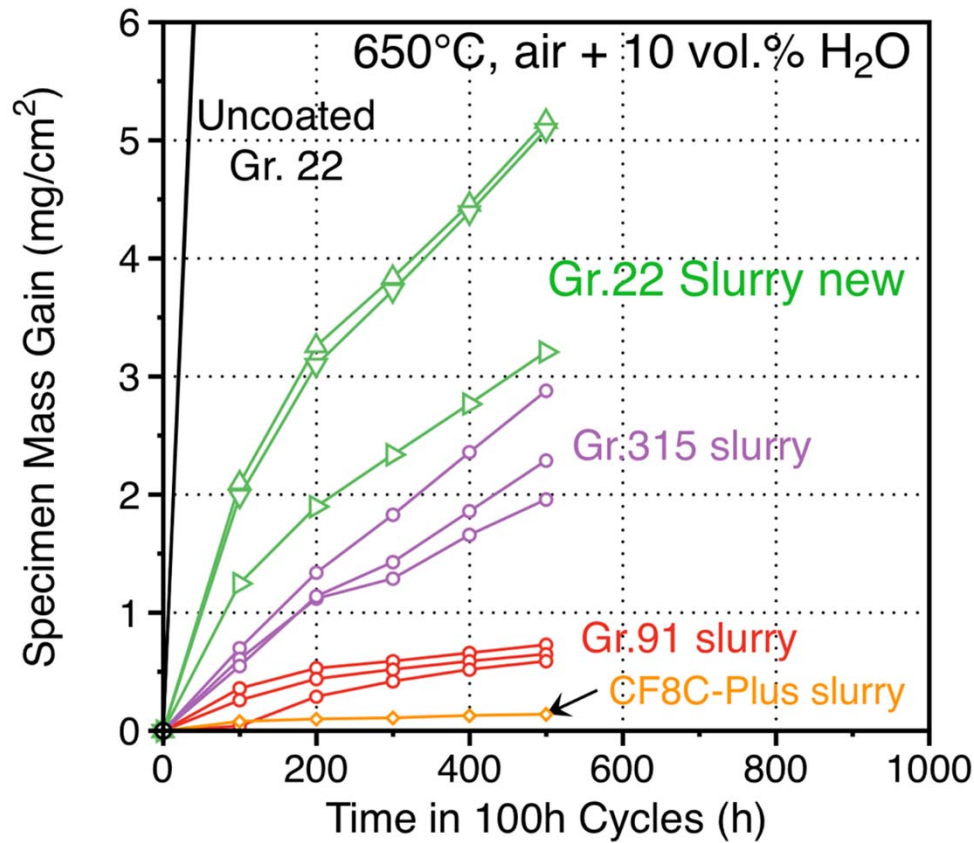


Air+10%H₂O:

- Rapid attack of uncoated Grade 22
- Quick indication of coating performance

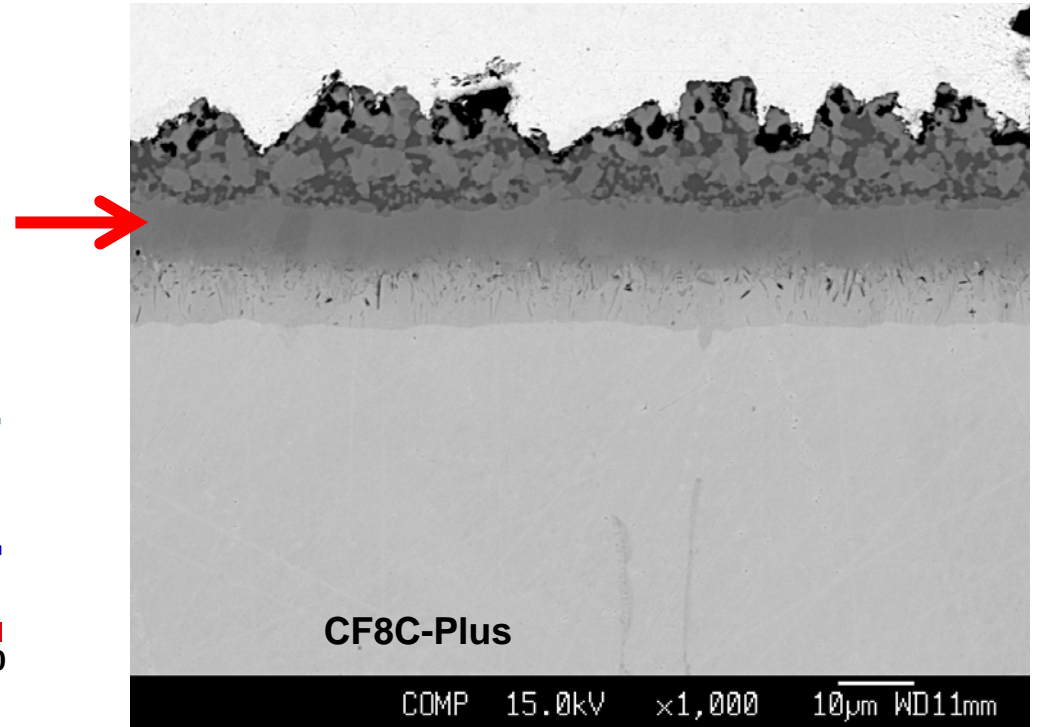
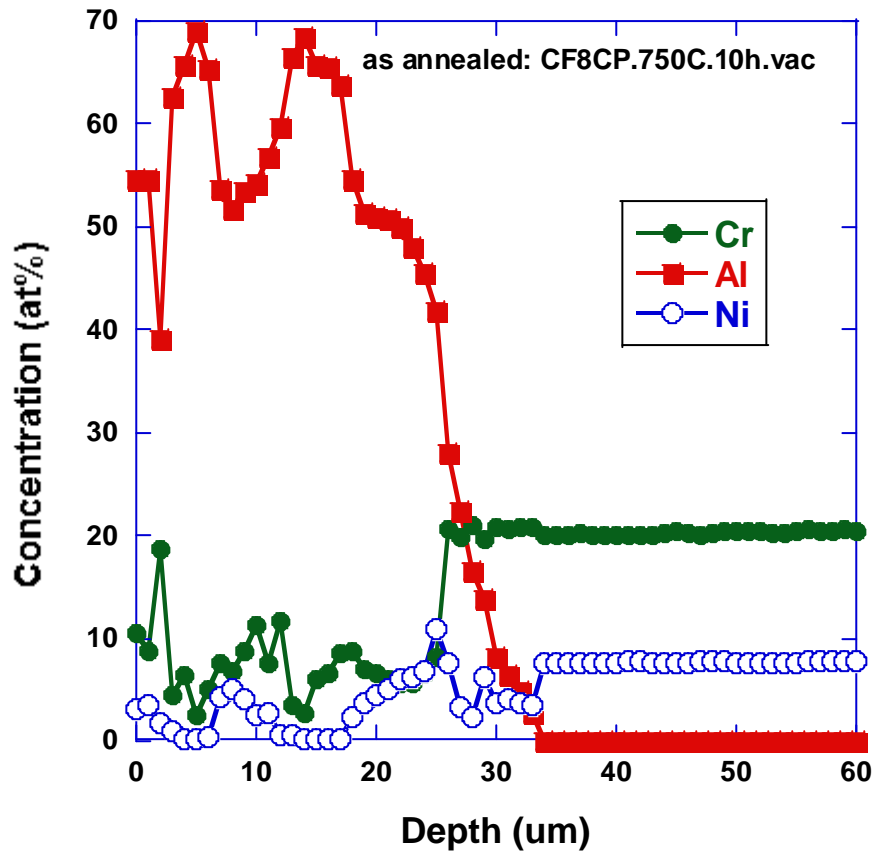
- FY14 slurry batch showed more rapid mass gains than best FY12 slurry coatings.
 - Variability: coating & surface preparation still not optimized

Substrate Cr content strongly influenced oxidation behavior of slurry aluminides at 650° C/air+10% H₂O



- Coatings on Gr.315 showed lower mass gains than the same slurry coating on Gr22.
- Slurries on Gr91 exhibited significant reductions in oxidation mass gain in short-term testing.
 - Suggests that Al-Cr slurry coatings may provide superior oxidation protection for Gr.22.
- Slurry aluminide on CF8C-Plus (austenitic) showed substantial further reduction in oxidation mass gain.
 - 19Cr, 13Ni

Composition profiles from a prior slurry coating batch on CF8C-Plus (750°C, 10h, 610 Torr)



Coating

Peak Al (inner layer): ~55 at%

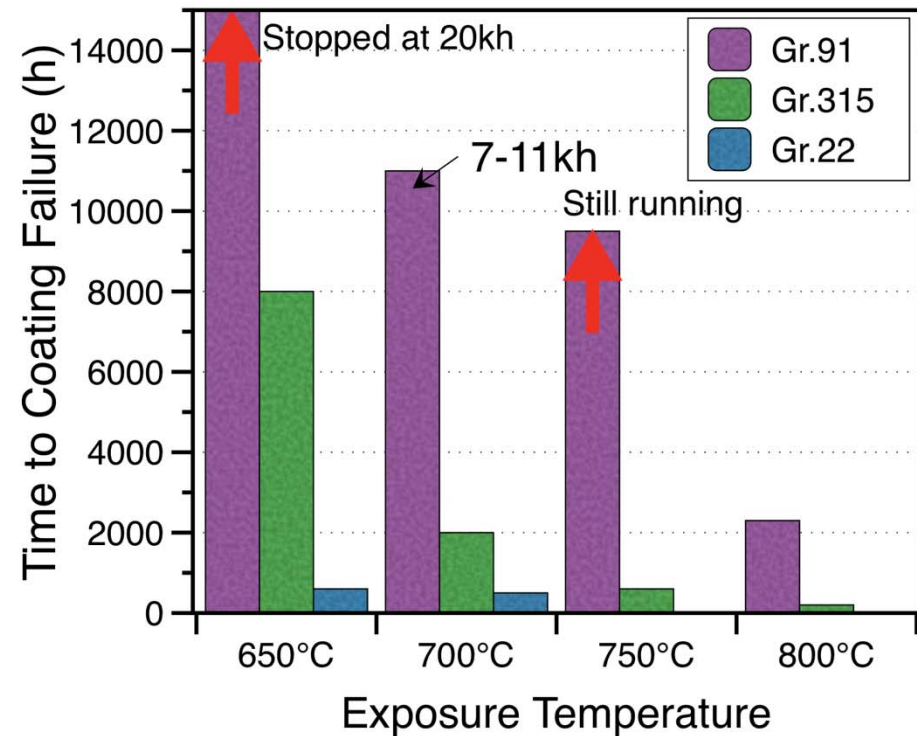
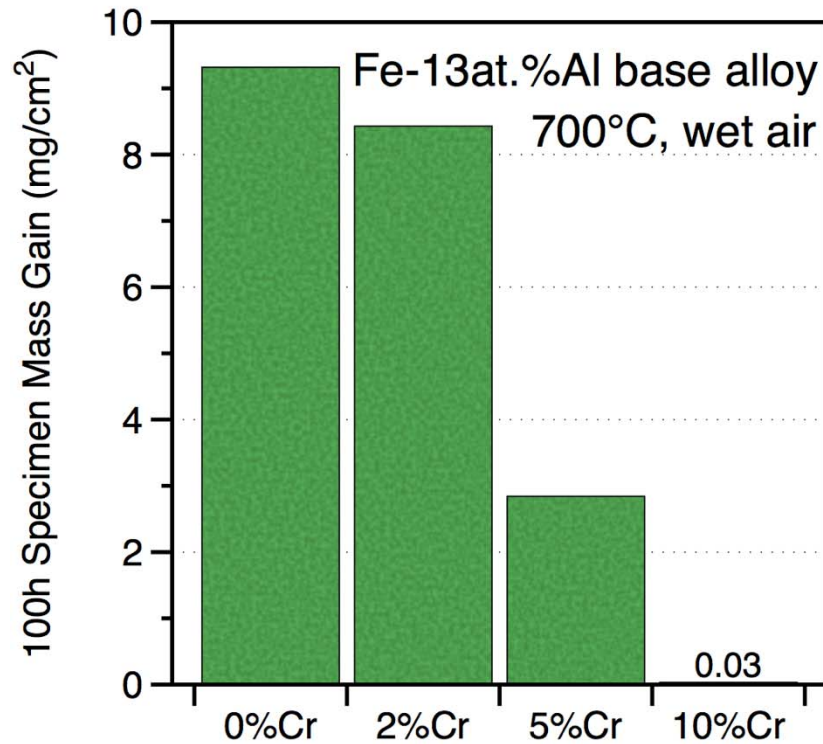
Avg Al (inner layer): 49.1 ± 4.0 at%

Avg Cr (inner layer): 6.8 ± 1.1 at%

Summary of Slurry Coating Results

- Simple slurry aluminides were demonstrated to provide a paint-on, low-cost route to oxidation protection of steels, but may not be appropriate for long-term protection on low-Cr steels such as Gr.22 at 650° C.
 - Unclear why Al content higher on 3Cr and 9Cr substrates
 - An Al-Cr slurry should be investigated for Gr.22.
 - Fe-Cr-Al likely superior to Fe-Al but may cause problems in sulfidizing environments (negative effect of Cr)
 - Slurry process needs to be optimized.
- Current simple slurry aluminides show promise for higher-Cr alloys at 650° C

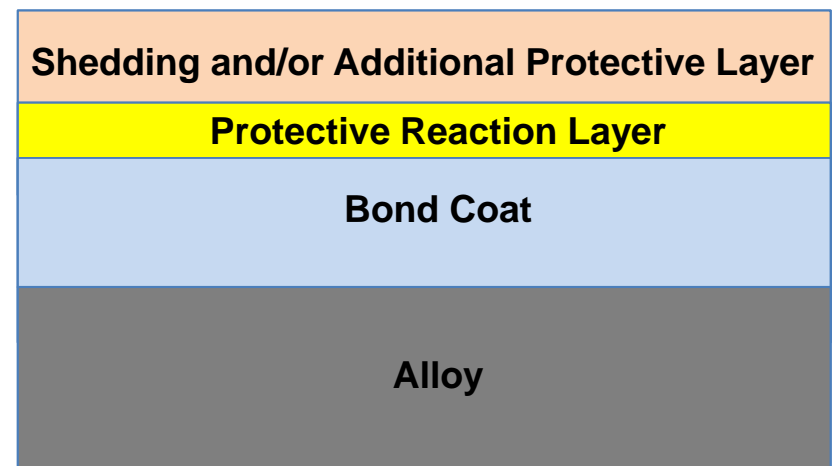
Effect of Cr has been studied previously



- Model alloys & CVD coatings both showed extreme effect of substrate Cr content at 650° -700° C using air+10%H₂O

Summary

- Modeling, synthesis, corrosion testing, and characterization
 - Project sidetracked by Gr.22 synthesis step
- Demonstrated that simple slurry aluminides can provide improvements in oxidation protection at 650° C vs. bare steel but may not be an adequate long term solution for low Cr alloys
 - Cr-Al coatings may be necessary for low-Cr steels
 - Processing & structure significantly affect efficacy of coating
- Conclude with proof of principle: A low-cost YSZ and/or mullite slurry coating deposited on a pre-oxidized FeCrAl rod for testing in coal-ash corrosion test.



Acknowledgements

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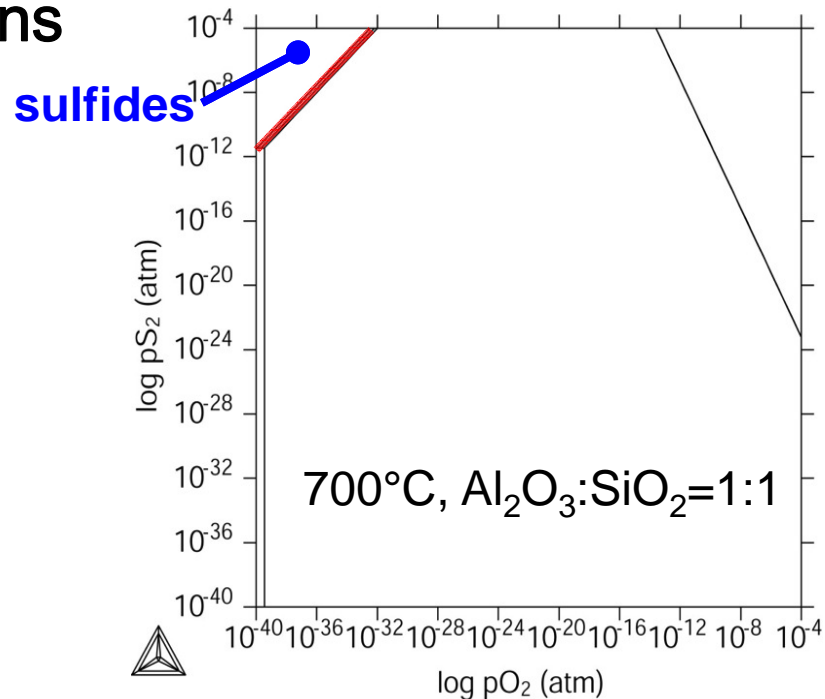
Backup Slides

Where Do We Go From Here?

- Synthesis

- Solution-based outer layer

- Examine/model incorporation of Cr, Mo, Si into existing coating for ash corrosion resistance
- Form additional protective oxide or sulfide scale
- Evaluate composite coatings (aluminide/silicide) and/or other compositions



Where Do We Go From Here?

- **Synthesis**

- **Solution-based outer layer**

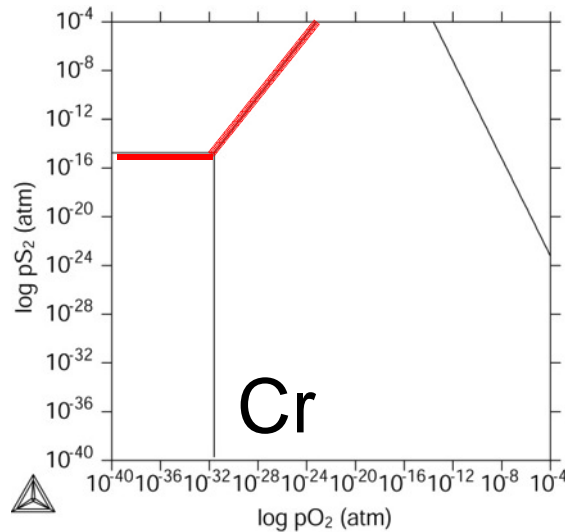
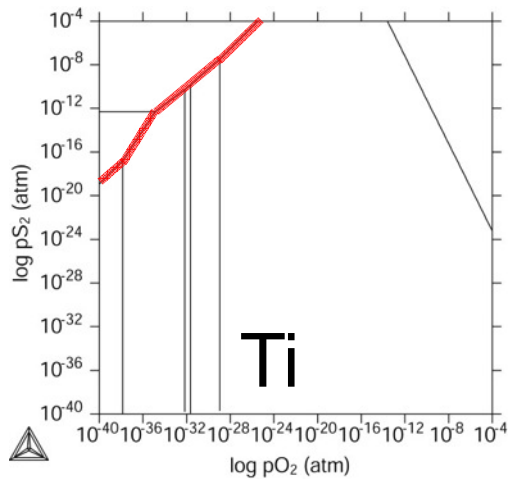
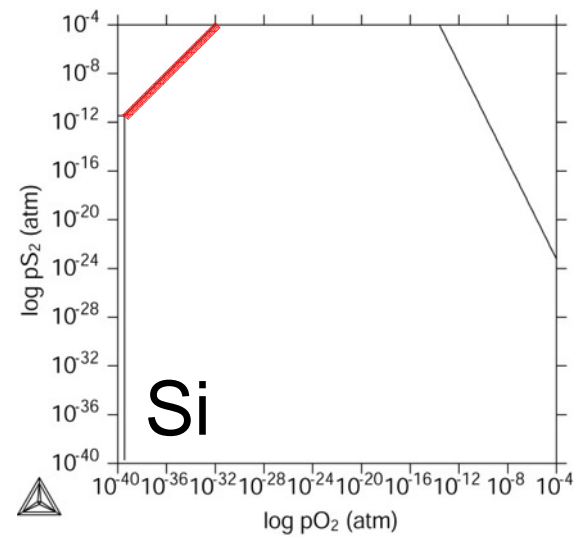
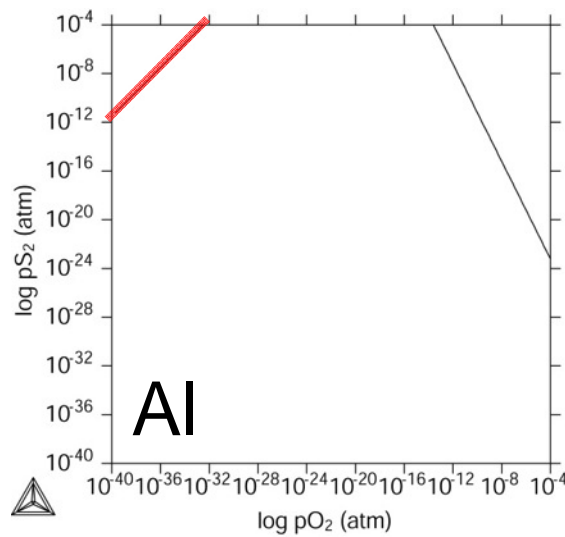
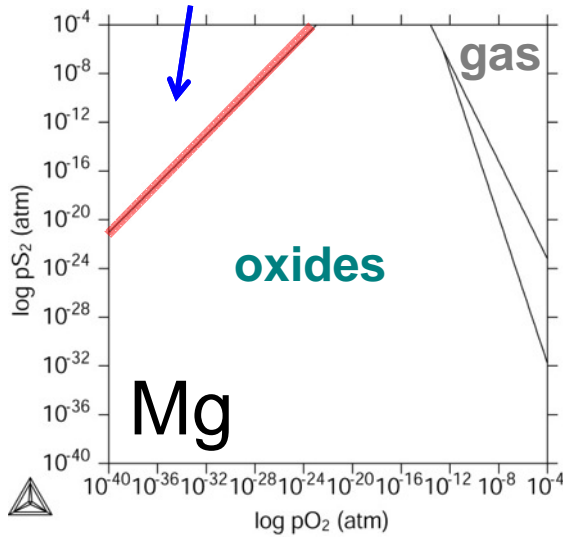
- **Examine/model incorporation of Cr, Mo, Si into existing coating for ash corrosion resistance**
 - **Form additional protective oxide or sulfide scale**
 - **Evaluate composite coatings (aluminide/silicide) and/or other compositions**

- **Examine surface structure/composition control as a non-stick approach, e.g., slag-shedding coatings**

- **Testing**

Calculation of Appropriate Metal-S-O Stability Diagrams (700° C)

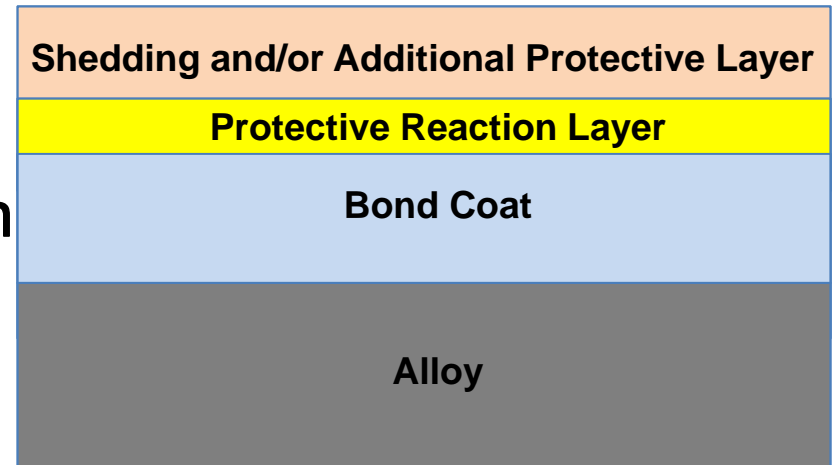
sulfides



- 19 binary metal oxides considered
- Diagonal line represents oxide/sulfides phase boundary
- Al_2O_3 and SiO_2 are thermochemically stable

Future Work – Building and Tailoring the Protection System

- Sulfidation testing in H₂/H₂S/Ar (4% H₂, 0.16 % H₂S and bal. Ar) environment at 650° C for 100 h, 500 h, 1000 h, and potentially 2000 h
 - Bare Gr22
 - Slurry aluminide coated Gr22
 - Preoxidized slurry aluminide Gr22
 - Al₂O₃-SiO₂ formers
- Ash corrosion and top-coat shedding layers

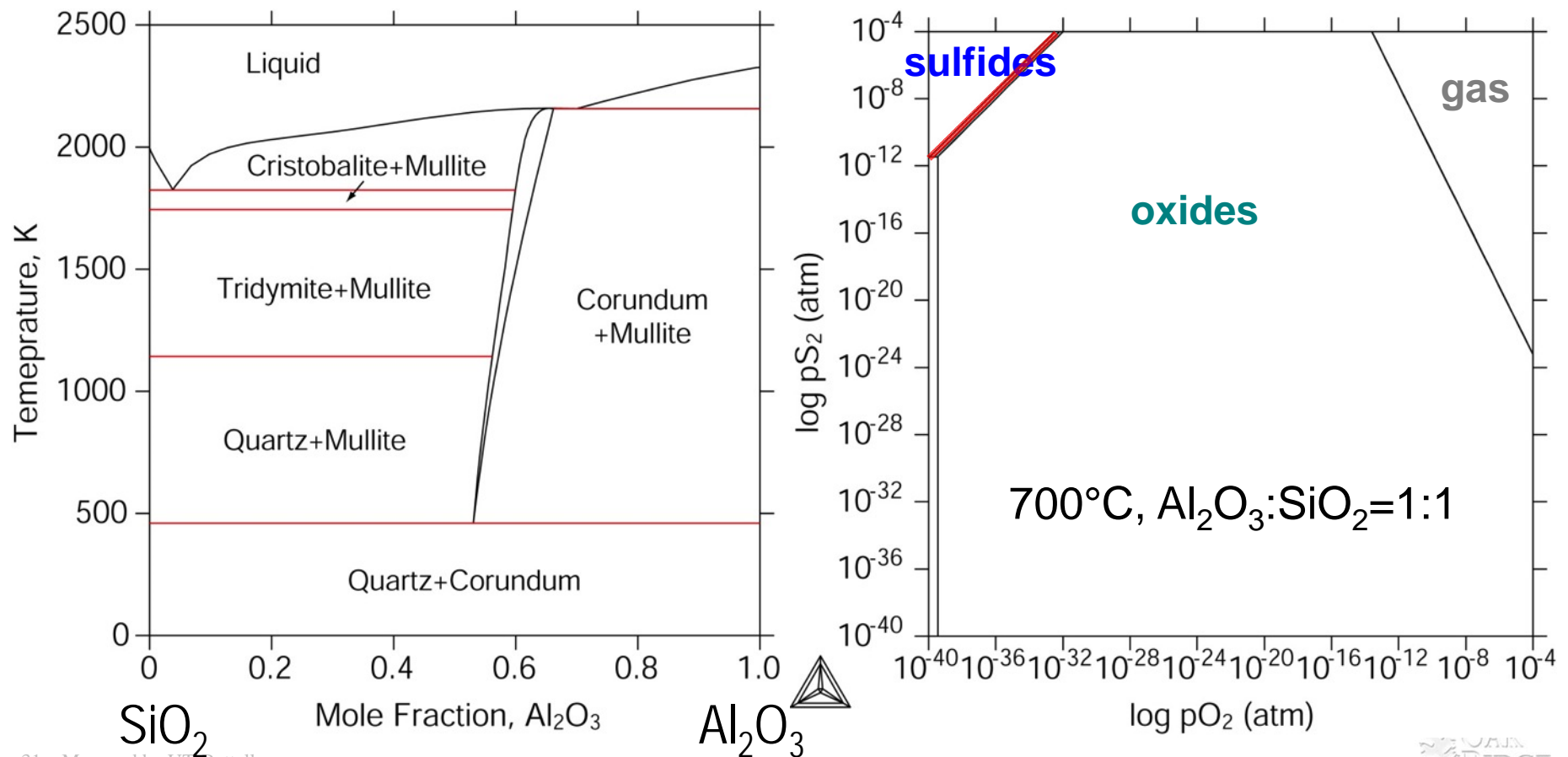


SiO₂ as the Sulfidation Resistant Layer

Thermochemical stability of Al₂O₃-SiO₂ was examined

Large phase field with SiO₂ polymorphs and mullite

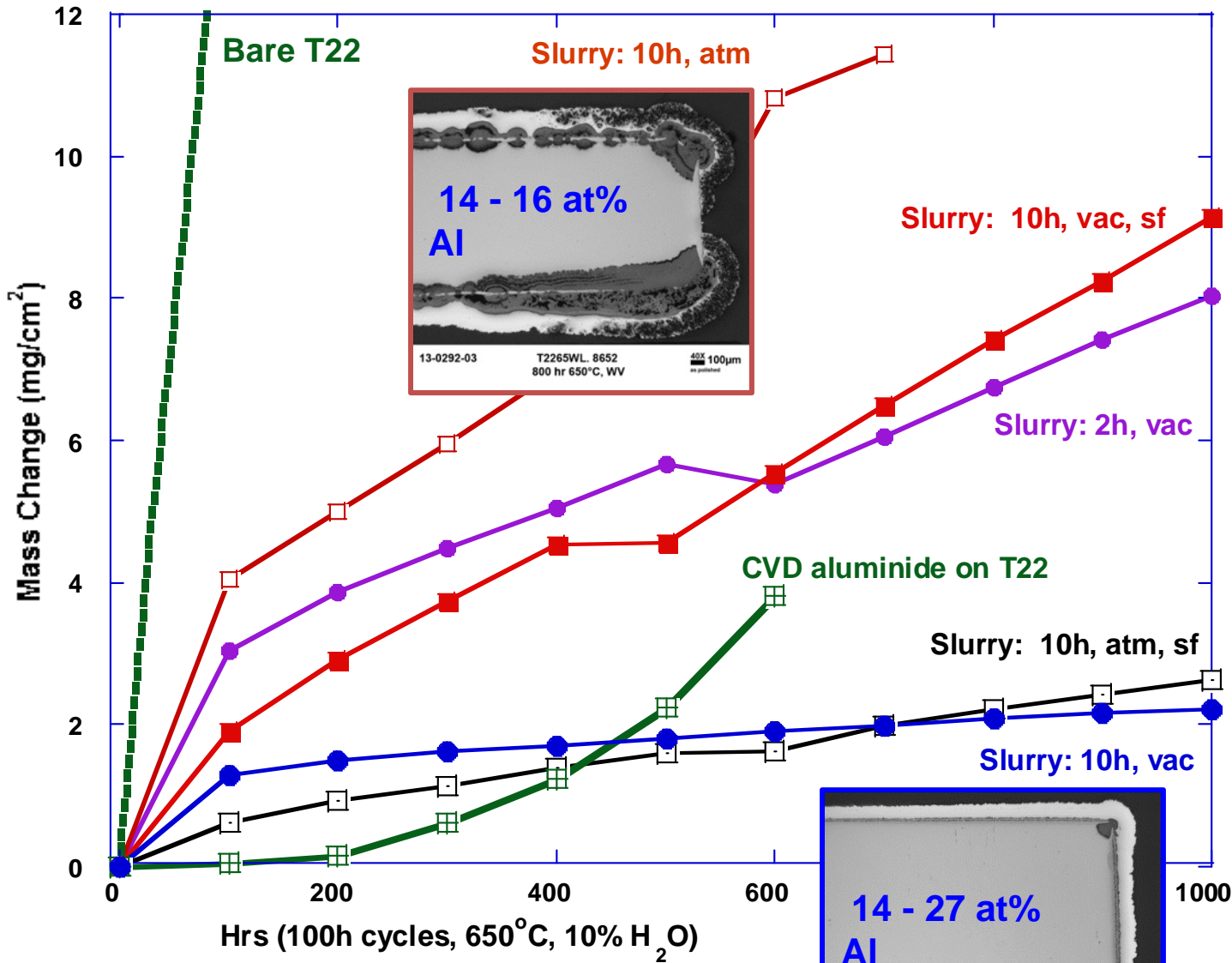
Phase stability is not affected significantly by different Al₂O₃ and SiO₂ ratios



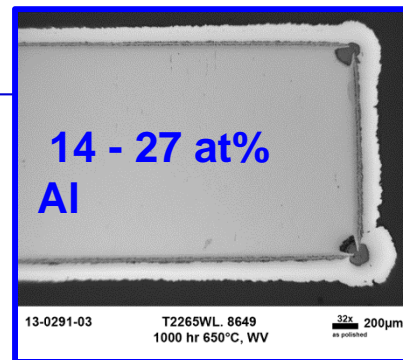
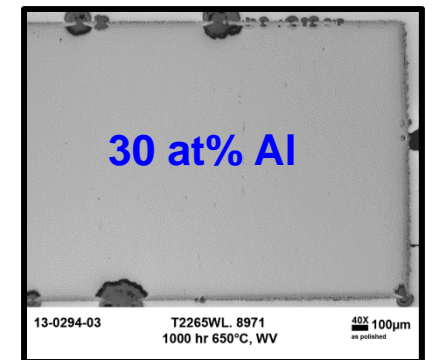
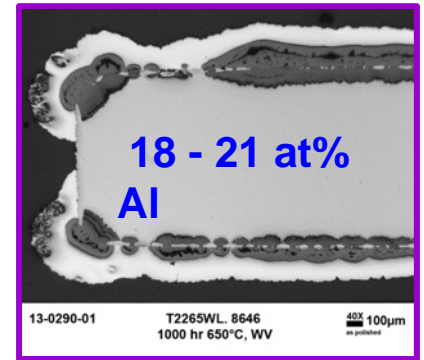
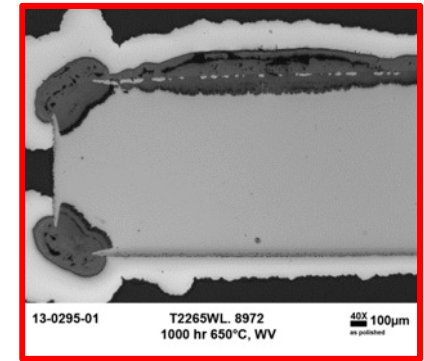
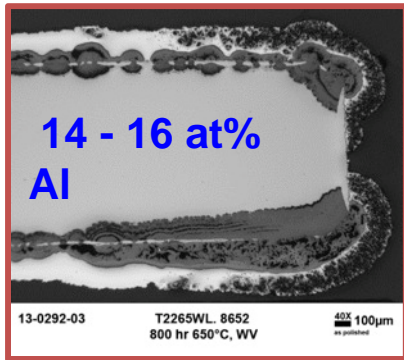
But We've Tried That Coating Before and It Didn't Work...

- **Coatings/surfaces are dependent upon**
 - Processing (resulting microstructure, density, contaminants...)
 - Substrate (composition, structure, CTE...)
- **Our surface treatment approach**
 - Slurry based
 - Robust, non-line of sight, industrially viable (spray, dip, paint...)
 - Can be adapted to multiple substrate and surface chemistries
 - Multiple layers feasible through diffusion, reaction and/or secondary applications
 - Control of processing variables are imperative for stability, i.e., success of resulting surfaces

Slurry Aluminide Coatings (annealed at 750°C) on T22



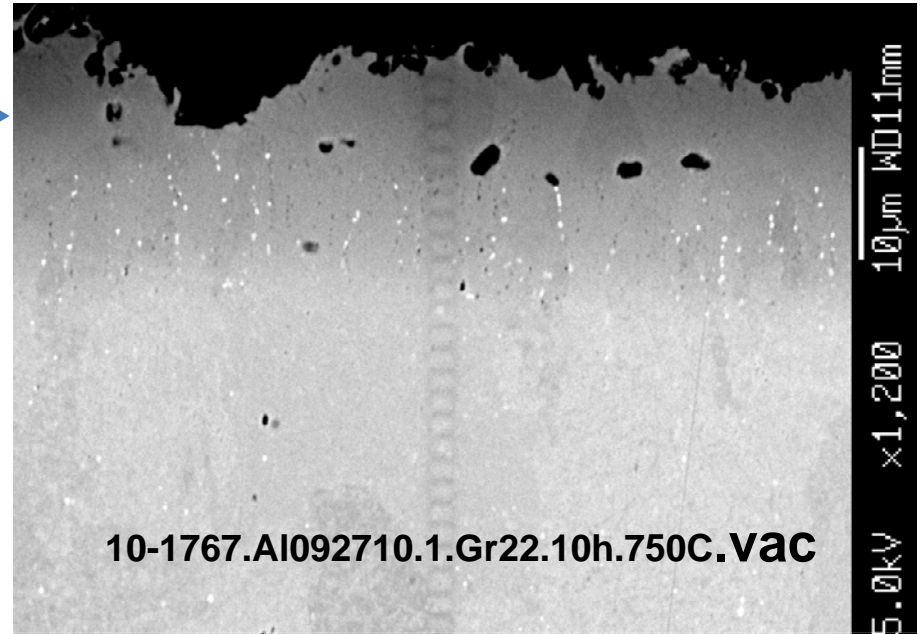
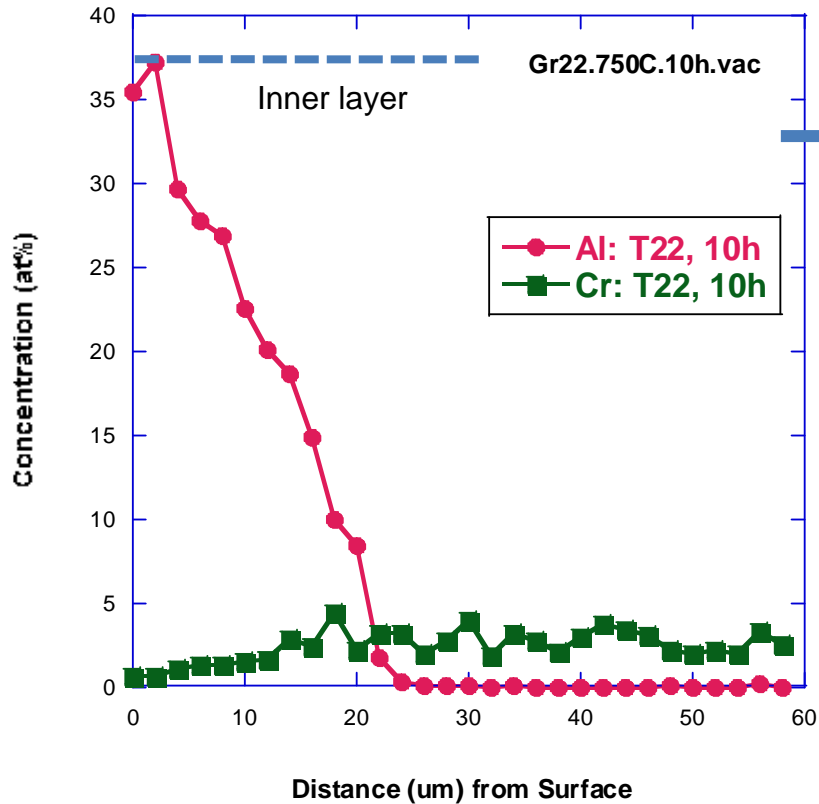
At% Al = Aluminum concentration (via EPMA) near coating surface after oxidation testing



Alloy Compositions

	Gr22	Gr315	Gr91	CF8CPlus
	wt %	wt %	wt %	wt %
Fe	95.5	93.4	89.7	61.8
Cr	2.3	2.9	8.3	19.2
Mo	0.9	0.7	0.9	0.33
Mn	0.6	0.3	0.3	4
Ni	0.2	0.1	0.1	12.8
W	<0.01	1.67	0.01	0.01
V	<0.01	0.21	0.3	0.05
Si	0.13	0.26	0.1	0.5
B	<0.001	<0.001	0.003	0.01
Nb	<0.01	<0.01	0.07	0.8
C	0.14	0.13	0.08	0.08
S	0.003	0.001	0.001	0.005
N	0.01	0.01	0.05	0.25
Cu	0.19	0.1	0.03	0.11
Al	0.02	0.01	<0.01	0.01

Composition profiles from prior slurry coating batch on Gr22 (750°C, 10h, 610 Torr)



BSE Image from FY11 slurry: high-Al outer layer spalled during annealing

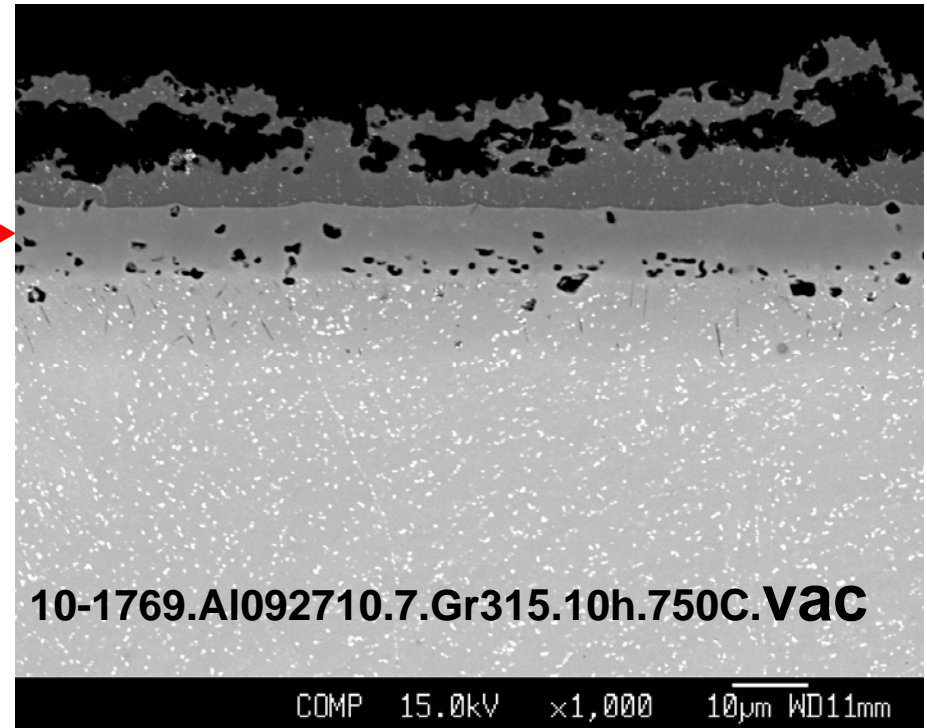
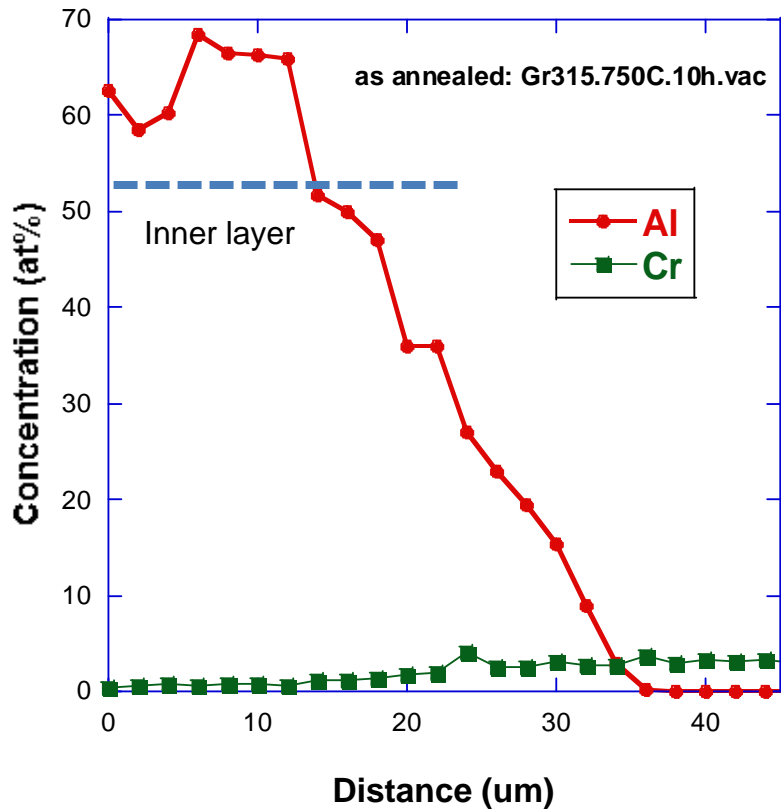
Coating

Peak Al: 38 at%

Avg Al: 29.7+7.5 at%

Avg Cr: 1.3+0.4 at%

Composition profiles from a prior slurry coating batch on Gr315 (750°C, 10h, 610 Torr)



BSE Image from FY11 slurry: initial optical microscopy indicates high-Al outer layer is not consistently present on FY14 batch

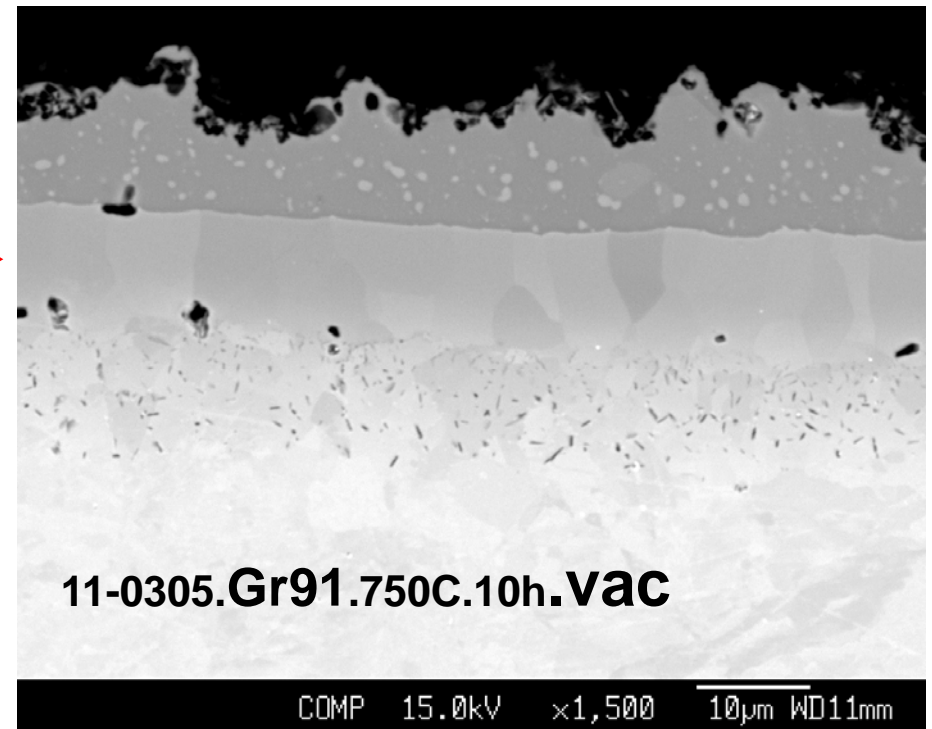
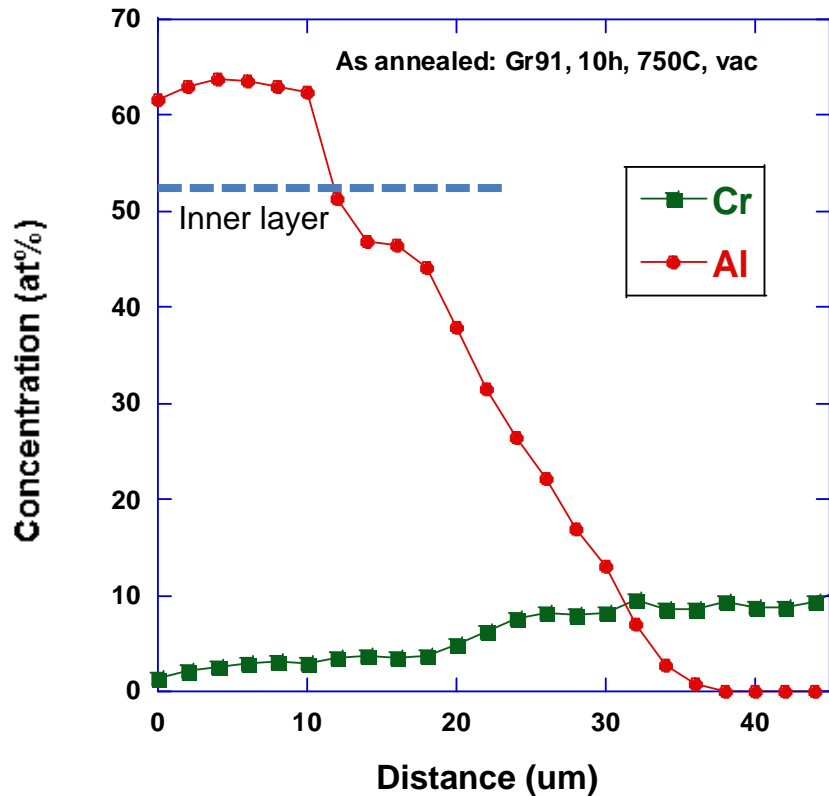
Coating

Peak Al (inner layer): 52 at%

Avg Al (inner layer): 41.3±9.8 at%

Avg Cr (inner layer): 1.9±1.1 at%

Composition profiles from a prior slurry coating batch on Gr91 (750°C, 10h, 610 Torr)



BSE image from FY11 slurry: initial optical microscopy indicates high-Al outer layer may not be consistently present on FY14 batch

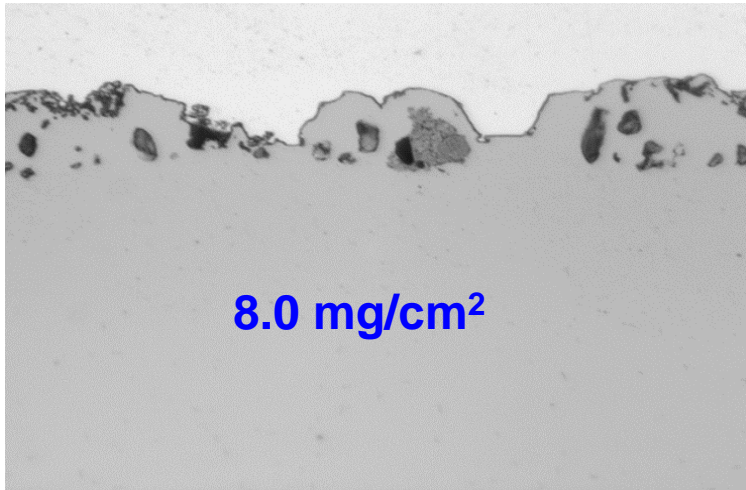
Coating

Peak Al (inner layer): 50 at%

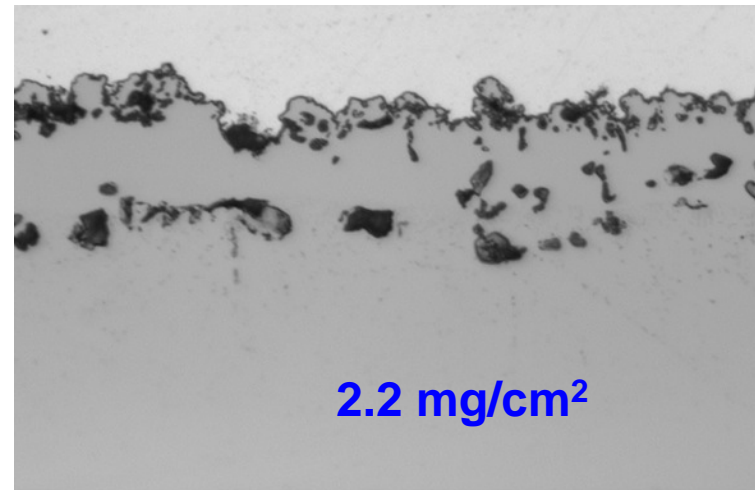
Avg Al (inner layer): 41.7 ± 7.6 at%

Avg Cr (inner layer): 4.7 ± 1.3 at%

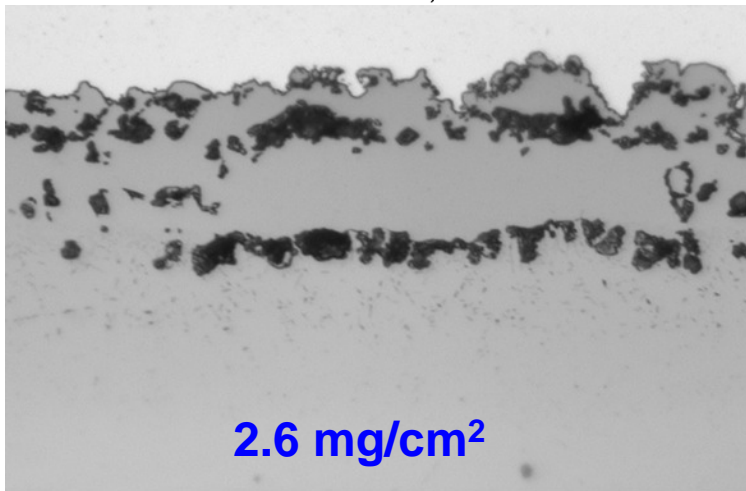
FY13: Intact slurry coating regions from various specimens after 1000h at 650°C



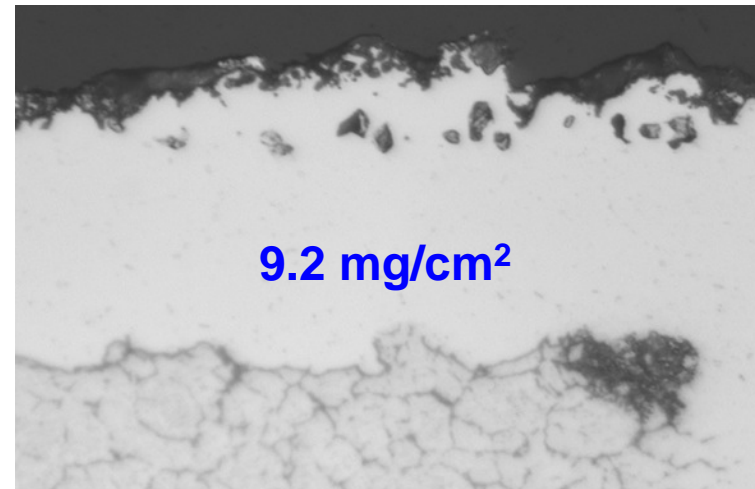
13-0290-08 T2265WL. 8646
1000 hr 650°C, WV 1000X 5µm
as polished



13-0291-08 T2265WL. 8649
1000 hr 650°C, WV 1000X 5µm
as polished

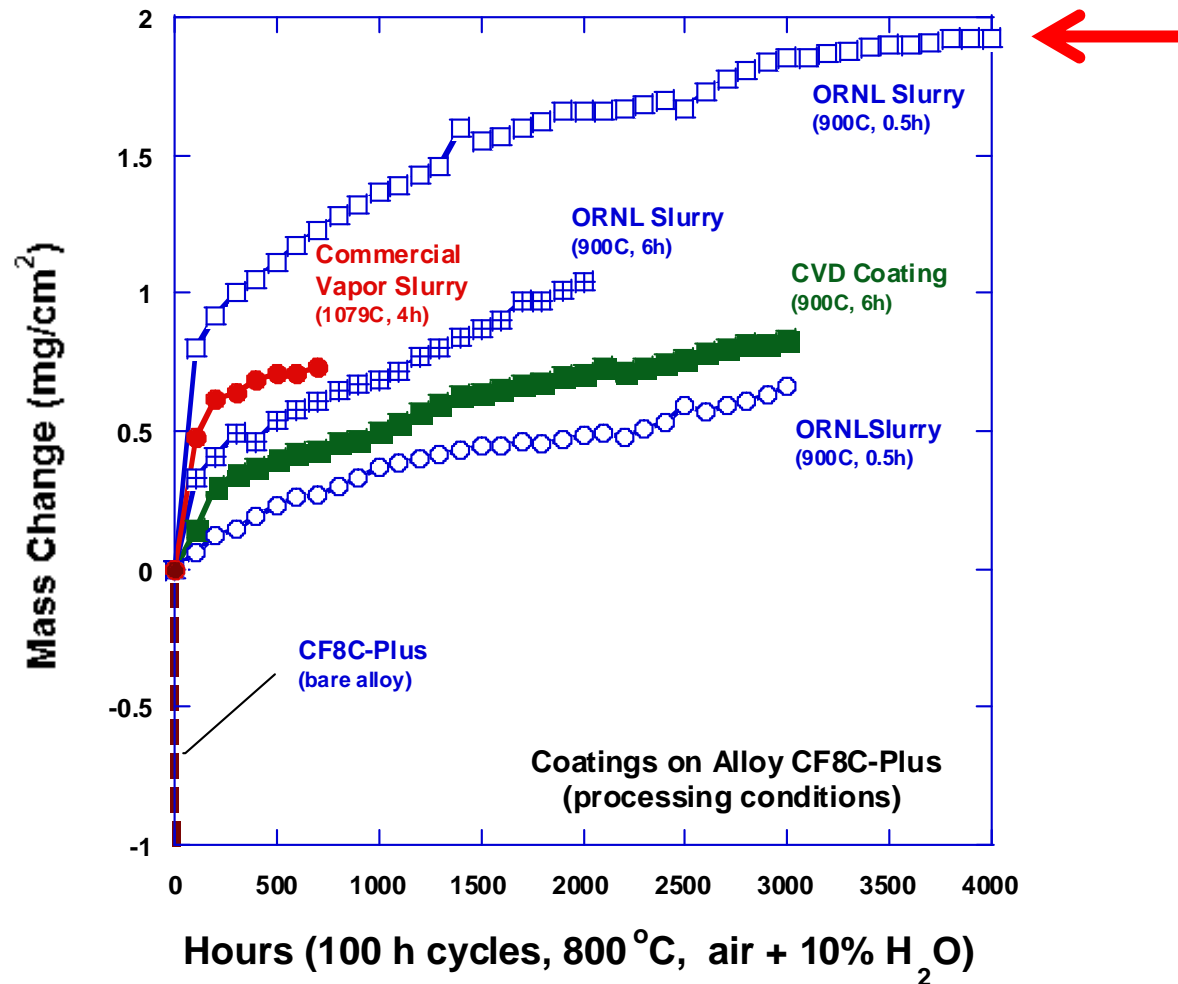


13-0294-07 T2265WL. 8971
1000 hr 650°C, WV 1000X 5µm
as polished



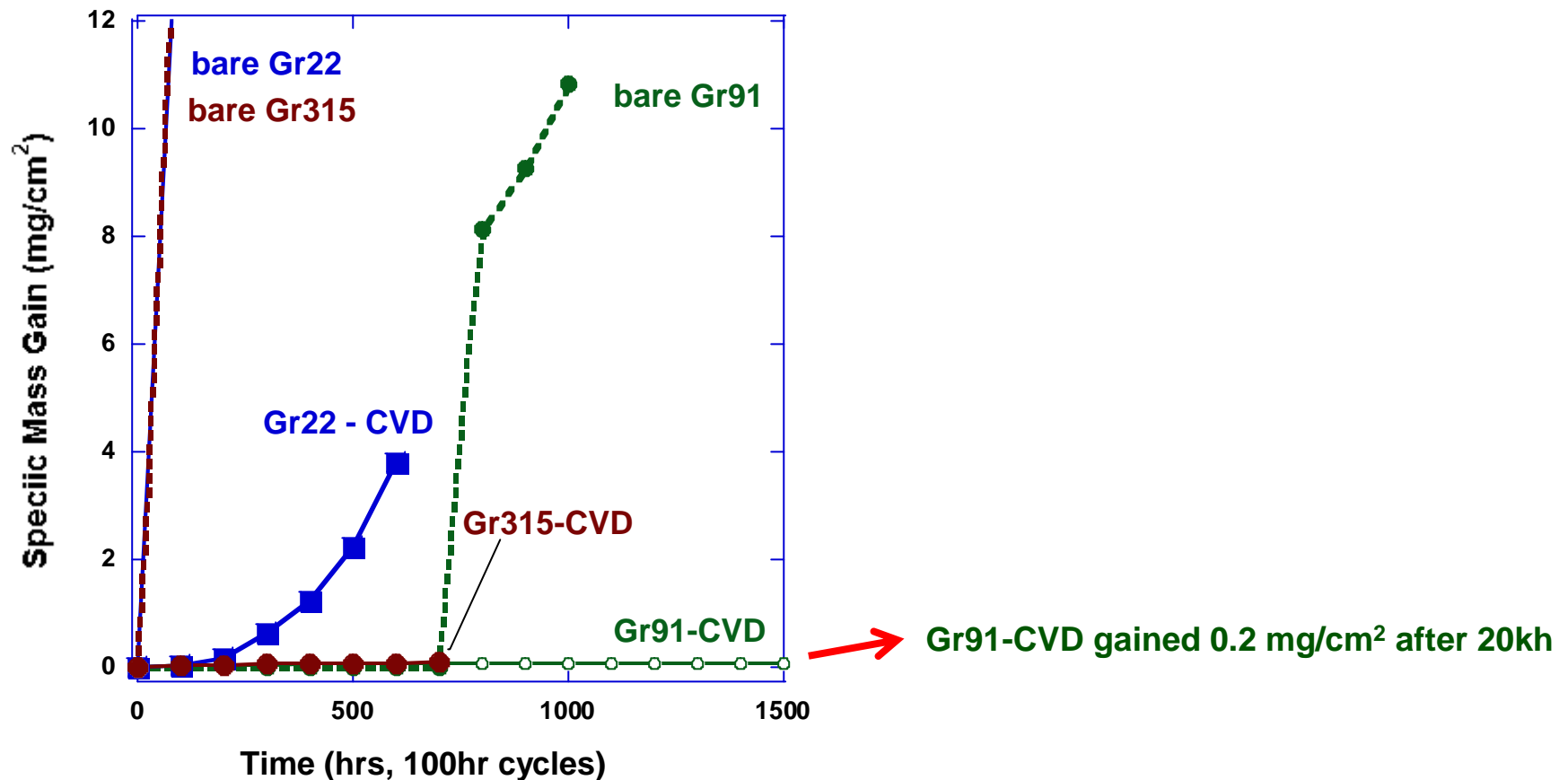
13-0295-07 T2265WL. 8972
1000 hr 650°C, WV 1000X 5µm
as polished

ORNL simple slurry aluminides on CF8C-Plus austenitic stainless steel exhibited lifetimes of >4,000h at 800°C in water vapor

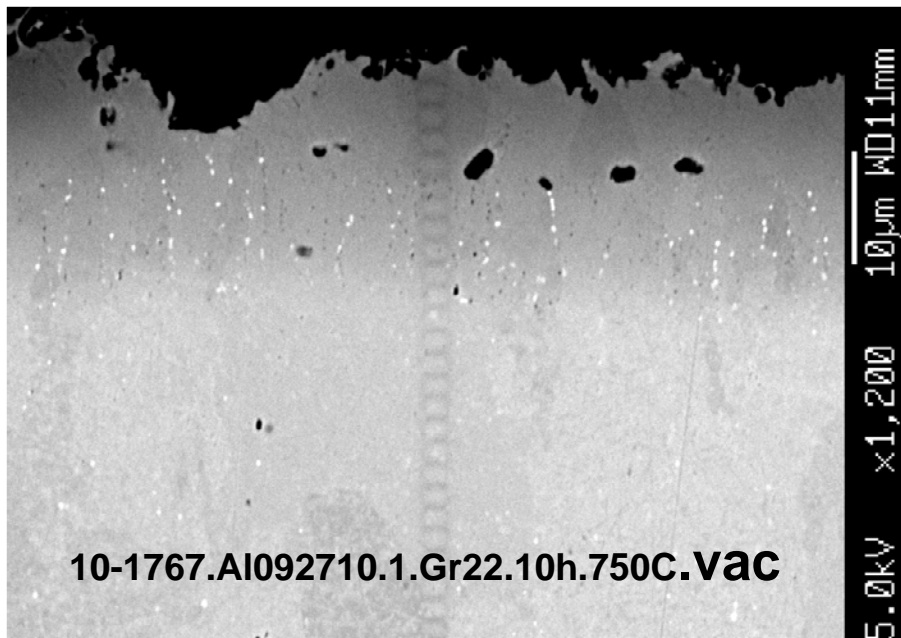


- CF8C-Plus: much higher Cr content, with Ni addition
- A variety of slurry processing conditions resulted in protective coatings.

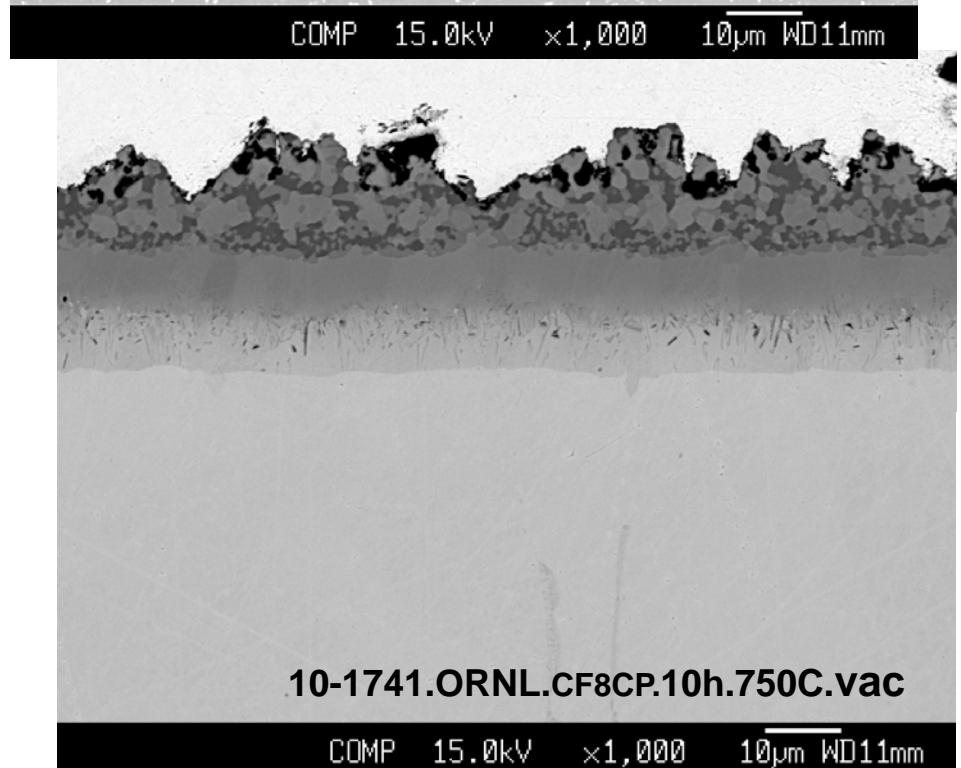
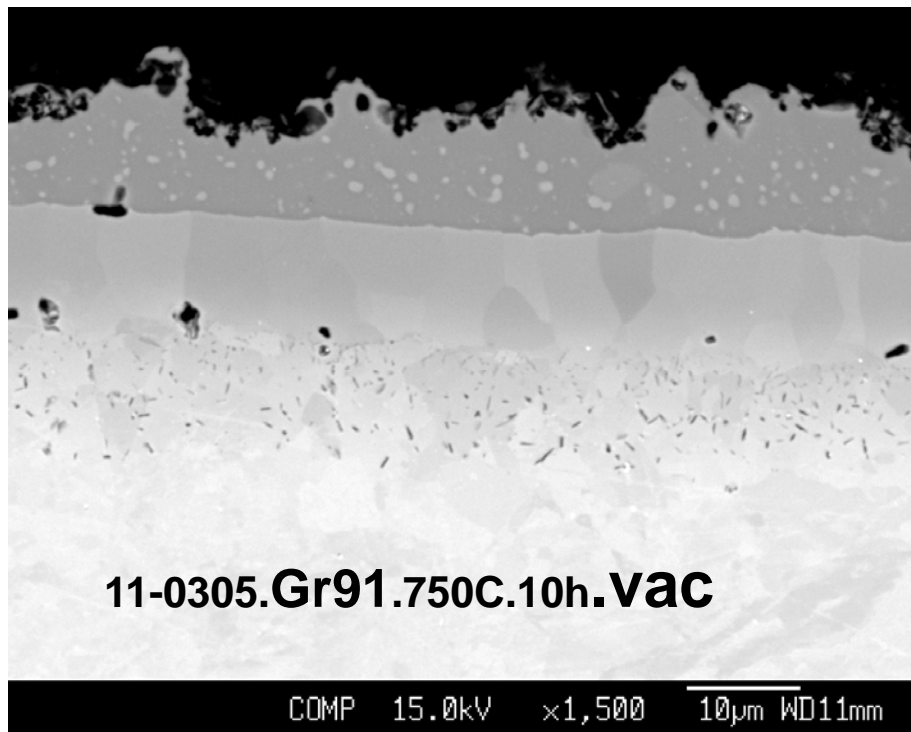
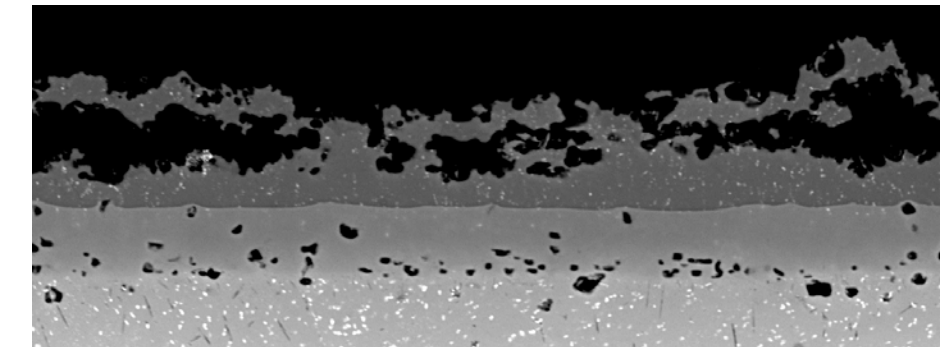
CVD aluminides: Substrate composition influenced coating oxidation at 650°C (air + 10% H₂O)



- Thin, high purity CVD aluminide on low-Cr Gr22 alloy provided minimal protection
 - Even a slight change (~0.6 wt.% increase) in Cr content of the substrate alloy exerted a noticeable effect on coating oxidation behavior.
- Early stage oxidation behavior improved substantially on Gr315
- CVD aluminide coating on 9-Cr alloy was stable to >20,000h



Optional...
-BSE & EPMA of FY11 slurry batch
-Note 315 and 91 have outer layers intact in
but those layers are not visible on most of
in the FY14 batch



What's next

- Thin slurry YSZ on FeCrAl (as a model alloy) for ash testing
 - Fired at 1325 – 1400C
 - June activity
- In application the YSZ might be fired via a plasma-arch lamp or other portable high temperature flash device...
 - Add schematics for flash lamp fired YSZ
- Finish slide