

Experimental and Computational Investigation of High-Entropy Alloys (HEAs) for Elevated-Temperature Applications

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Outline of Presentation

- Potential significance
- Background and unique behavior of HEAs
- Project objectives
- Proposed work
- Results and discussion
- Future work
- Published papers and presentations
- Conclusions

Potential Significance

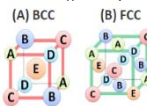
- Develop a suitable HEA for steam and gas turbine components operating at temperatures higher than 760° C.
- Increase the thermal efficiency of steam turbines and reduce the costs of fuel and emissions.
- Apply a computer-aided approach for designing new types of alloys applicable for the development of other high-temperature materials.



<http://middleeast.geblogs.com/en/lang-en/ge-steam-turbines-to-help-saudi-arabia-boost-power-output-and-efficiency-3/>

Background and Unique Behavior

- Most alloy systems are based on a single principal element to form the matrix.
- Multiple principal elements were expected to yield many intermetallic compounds creating a complex microstructure.
- However, simple face-centered cubic (FCC) and body-centered cubic (BCC) structures are possible and thermodynamically favorable.



[1] J. W. Yeh, S. K. Chen, S. J. Lin, J. Y. Gan, T. S. Chin, T. T. Shun, C. H. Tsau, and S. Y. Chang, *Advanced Engineering Materials* 6, 299 (2004).
 [2] K. Cantow, L. T. W. Chang, P. Knight, and A. J. B. Vincent, *Materials Science and Engineering A* 375-377, 213 (2004).
 [3] Y. Zhang, T. T. Zuo, Z. Tang, M. C. Gao, K. A. Dahmen, P. K. Liaw, and Z. P. Lu, *Progress in Materials Science* 61, 1 (2015).

Background and Unique Behavior (Cont'd)

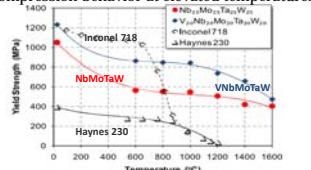
- In equimolar ratios,

$$\Delta S_{conf} = k_B \cdot \ln(\Omega) = \frac{R}{N_A} \ln(N)^N = R \cdot \ln(N)$$
 k: Boltzmann's constant
 Ω: Number of ways of mixing
 R: Gas Constant
 N: Number of elements
 N_A: Avogadro constant
- High entropy of mixing and sluggish diffusion yield stable FCC and BCC solid solutions.
- Stable phases have the lowest Gibbs Free Energy

$$\Delta G = \Delta H - T\Delta S$$
 G: Gibbs free energy
 H: Enthalpy
 T: Temperature
 S: Entropy
- At high temperatures, HEAs are stable and show great high-temperature strengths.

Background and Unique Behavior (Cont'd)

Compression behavior at elevated temperatures

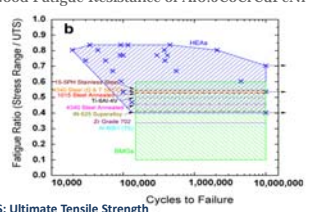


The yield stress of both alloys dropped by 30 - 40% between room temperature and 600 °C, but was relatively insensitive to temperature above 600 °C, comparing favorably with conventional superalloys.

O. N. Senkov, G. B. Wilks, J. M. Scott, and D. B. Miracle, *Intermetallics* 19, 698 (2011).
 O. N. Senkov, G. B. Wilks, D. B. Miracle, C. P. Chuang, and P. K. Liaw, *Intermetallics* 18, 1758 (2010).

Background and Unique Behavior (Cont'd)

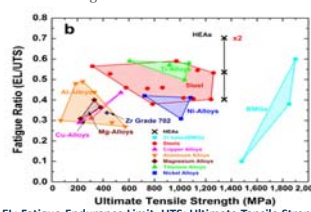
Good Fatigue Resistance of Al_{0.5}CoCrCuFeNi



UTS: Ultimate Tensile Strength
 M. A. Hemphill, T. Yuan, G. Y. Wang, J. W. Yeh, C. W. Tsai, A. Chuang, and P. K. Liaw, *Acta Materialia* 60, 5723 (2012).

Background and Unique Behavior (Cont'd)

Good Fatigue Resistance of Al_{0.5}CoCrCuFeNi



EL: Fatigue-Endurance Limit; UTS: Ultimate Tensile Strength
 M. A. Hemphill, T. Yuan, G. Y. Wang, J. W. Yeh, C. W. Tsai, A. Chuang, and P. K. Liaw, *Acta Materialia* 60, 5723 (2012).

Project Objectives

- Perform fundamental studies on the Al₁₀CrCuFeMnNi high-entropy alloy (HEA) system for use in boilers, and steam and gas turbines at temperatures above 760°C and a stress of 35 MPa.
- Develop the thermodynamic database and identify the potential HEAs within the Al-Cr-Cu-Fe-Mn-Ni system for further experimental investigation.
- Optimize microstructures to identify the best combined strength, ductility, and creep resistance.

Preliminary Results - Alloy Fabrication (Cont'd)

- Al₁₀CrCuFeMnNi (Al₁₀CrCuFeMnNi-HIP-899°C)
- Φ 25.4 mm (1 inch) × 762 mm (30 inch)
- Vacuum Cast → Hot Isostatic Pressing (HIP) 899 °C, 103 MPa, 2 h.
- Differential Thermal Analysis (DTA) 100 °C---1,600 °C
 - No thermal transitions: 100 to around 980 °C.
 - Three transitions occurred at 980 °C, 1,173 °C, and 1,358 °C.
 - Discernible liquidus is not observed up to 1,600 °C.

Preliminary Results - Alloy Fabrication (Cont'd)

- For the same chemical composition, Al₁₀CrCuFeMnNi, another specimen (Al₁₀CrCuFeMnNi-as cast) was prepared by arc-melting elemental Al, Cr, Cu, Fe, Mn, and Ni raw materials in a water-cooled copper hearth. Its dimension is 100 mm × 80 mm × 10 mm.
- To study the aluminum effect, two new chemical compositions, Al_{0.1}CrCuFeMnNi and Al_{0.3}CrCuFeMnNi, are casted.

Microstructures (Al₁₀CrCuFeMnNi-HIP-899°C)

Atomic %	Al	Cr	Mn	Fe	Ni	Cu
Nominal	13.79	17.24	17.24	17.24	17.24	17.24
Dark-grey area (I)	8.12	11.85	15.47	20.07	9.05	6.44
White-phase layer (II)	6.97	1.65	19.12	3.11	9.90	59.25
Light-grey area (III)	20.97	4.86	19.31	6.59	28.13	20.14

Microstructures (Al_{0.3}CrCuFeMnNi-as-cast)

Atomic %	Al	Cr	Mn	Fe	Ni	Cu
Nominal	13.79	17.24	17.24	17.24	17.24	17.24
Dark-grey area (I)	14.84	10.97	17.91	26.71	17.67	11.90
White-phase layer (II)	13.18	1.4	21.13	6.35	15.37	42.57

Microstructures

Alloy Composition: Al_{1.3}CoCrCuFeNi

High-resolution SEM Micrograph Color-coded according to X-ray EDS signals

- Al-Ni-enriched
- Cr-Fe-enriched
- Cu-rich

0.5 μm

Al Ni Co Fe Cr Cu

B2: Al_{1.3}NiCo_{0.7}Fe_{0.3} BCC: CrFeCu₂ FCC: Cu

Microstructures

Atomic concentrations across the B2-FCC interface

Atom-probe tomography (APT)

Alloy composition: Al_{1.3}CoCrCuFeNi

B2 phase Al_{1.3}NiFe_{1.3}Co_{0.7}

Cu-rich FCC precipitate

X-ray Diffraction Patterns

	FCC1	FCC2	BCC1	BCC2
Al _{10.1} CrCuFeMnNi	▲	▲	▲	▲
Al _{10.8} CrCuFeMnNi	▲	▲	▲	▲

In-situ neutron-diffraction experiments

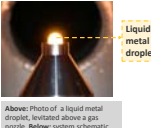
Spallation Neutron Source (SNS), Oak Ridge National Laboratory (ORNL)

The Nanoscale-Ordered Materials Diffractometer (NOMAD)

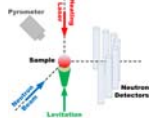
- Al_{1.3}CoCrCuFeNi and Al_{1.3}MnCrCuFeNi specimens were heated all the way to the molten state using the newly-developed gas levitation and laser heating system

Aerodynamic Levitator

- Uses aerodynamic forces and CO₂ laser-beam heating
 - Greater than 2,500 °C achieved
- Containerless measurement
 - Avoids contamination
 - Avoids heterogeneous nucleation
 - Accesses supercooled liquids
 - Accesses clean liquid surfaces
- Beamline system for *in-situ* studies of structure in extreme conditions
- Developed through the DOE Small Business Innovation Research (SBIR) Grant
 - P.I. Dr. Richard Weber (Materials Development, Inc.)
 - SNS Collaborators, Dr. Joerg Neuefeind and Mr. Louis J. Santodonato



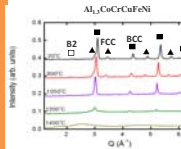
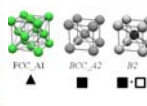
Above: Photo of a liquid metal droplet, levitated above a gas nozzle. Below: system schematic.



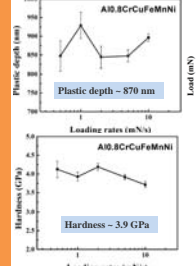
Neutron Scattering Results

SNS NOMAD Beamline

- Simple cubic structures
 - Face-centered-cubic (FCC)
 - Body-centered-cubic (BCC)
 - B2 (an ordered BCC variant)

Results-Nanoindentation



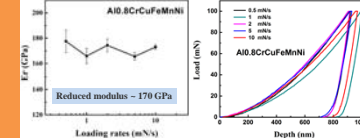
Plastic depth - 870 nm

Hardness - 3.9 GPa

$$H = \frac{P_{max}}{A}$$

where P_{max} is the peak load, and A is the project contact area at the peak load.

Results-Nanoindentation



Reduced modulus - 170 GPa

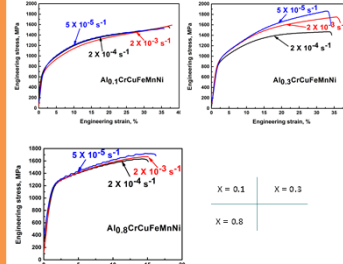
$$\frac{dP}{dh} = \frac{2}{\sqrt{\pi}} Er \sqrt{A}$$

$\frac{dP}{dh}$: the slope of the unloading load-displacement curve,
 $\frac{Er}{\sqrt{\pi}}$: the reduced modulus of a material
 A : the projected contact area of the indent.

The reduced modulus of Al_{0.8}CrCuFeMnNi is around 170 GPa, which is comparable to that of FeCoCrNiMn (179 ± 4 GPa).

P. Termonski, V. P. Alekshin, M. K. Shorshorov, M. M. Khrushchov, and V. N. Skvortsov. *Zvezdskaya Laboratoriya*, 20, 124(2)(1974).
 C. Zhu, Z. P. Lu, and T. G. Nieh, *Acta Materialia*, 2013, 61, 8 (2013).

Compression Tests



Engineering stress, MPa vs Engineering strain, %

Strain rates: $5 \times 10^{-5} s^{-1}$, $2 \times 10^{-3} s^{-1}$, $2 \times 10^{-4} s^{-1}$

Compression Tests (Cont'd):

The relationship between true stress (strain) and engineering stress (strain)

$$\epsilon_T = \ln(1 + \epsilon_E)$$

$$\sigma_T = \sigma_E (1 + \epsilon_E)$$

ϵ_T : true strain
 ϵ_E : engineering strain
 σ_T : true stress
 σ_E : engineering stress

Strain Hardening

$$\sigma_T = K \epsilon_T^n$$

K : constant
 n : strain hardening exponent

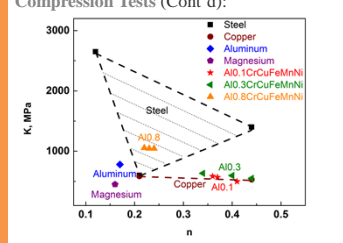
Al_{0.8}CrCuFeMnNi
 $n = 0.41$
 $K = e^{0.21} = 497 \text{ MPa}$

Callister, Jr., William D (2005), *Fundamentals of Materials Science and Engineering* (2nd ed.), pp. 152.

Compression Tests (Cont'd):

		$2 \times 10^{-3} s^{-1}$	$2 \times 10^{-4} s^{-1}$	$5 \times 10^{-5} s^{-1}$
Al _{0.8} CrCuFeMnNi	n	0.41	0.37	0.36
	K (MPa)	498	567	584
Al _{0.3} CrCuFeMnNi	n	0.40	0.34	0.44
	K (MPa)	596	633	550
Al _{0.1} CrCuFeMnNi	n	0.23	0.22	0.24
	K (MPa)	1043	1054	1043

Compression Tests (Cont'd):

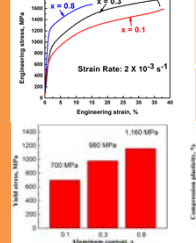


Yield strength, K, MPa vs n

Alloys: Steel, Copper, Aluminum, Magnesium, Al_{0.1}CrCuFeMnNi, Al_{0.3}CrCuFeMnNi, Al_{0.8}CrCuFeMnNi

Callister, Jr., William D (2005), *Fundamentals of Materials Science and Engineering* (2nd ed.), pp. 152.

Compression Tests



Engineering stress, MPa vs Engineering strain, %

Strain Rate: $2 \times 10^{-3} s^{-1}$

As the aluminum content increases,

- Yield stress increases
- Compression plasticity decreases

This phenomenon is due to the increased strengthening effect of lattice strains caused by the lattice-sites occupation of Al.

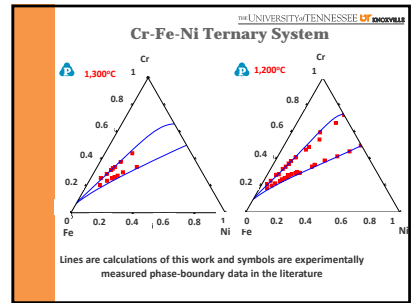
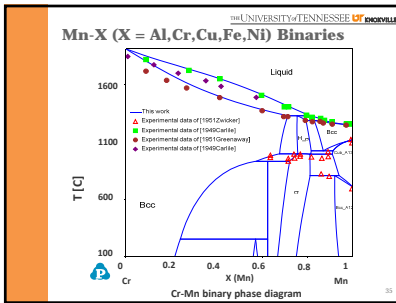
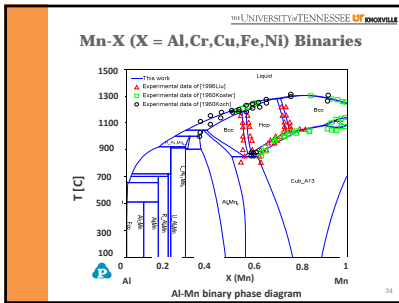
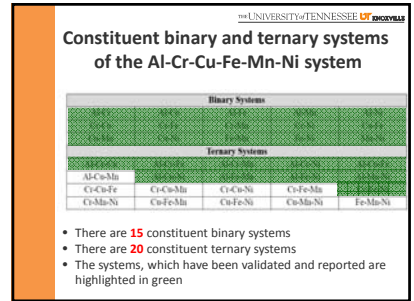
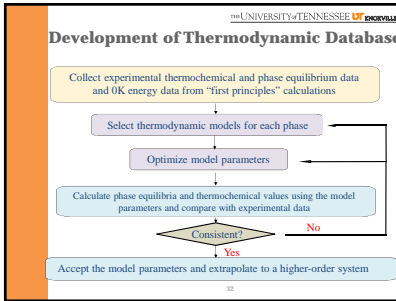
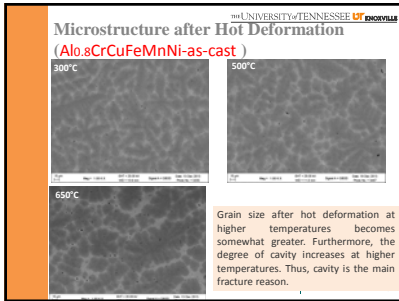
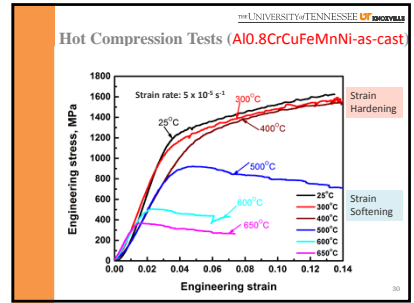
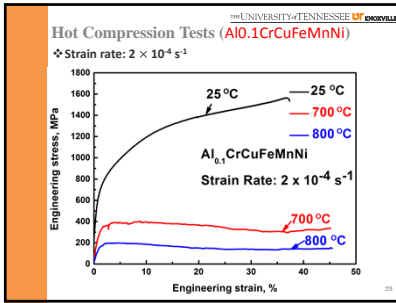
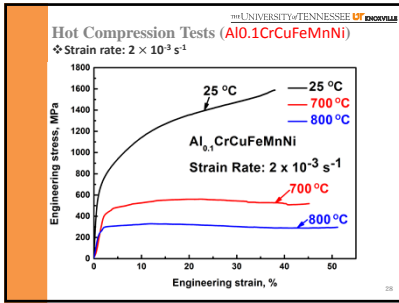
Yield strength, MPa vs Aluminum content, %

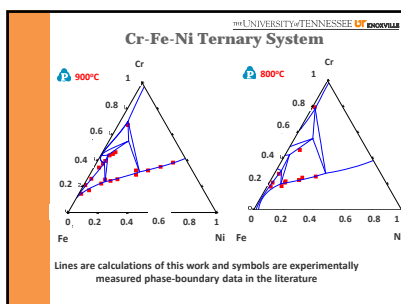
Aluminum content: 0.1, 0.3, 0.8


Yield strength: 700 MPa, 980 MPa, 1,160 MPa

Compression plasticity: 37%, 33%, 19%

Z. Tang, M. C. Gao, H. Diao, T. Yang, J. Liu, T. Zuo, Y. Zhang, Z. Lu, Y. Cheng, Y. Zhang, K. A. Dahmen, P. K. Liaw, and T. Egami, *JOM*, 65, 12 (2013).






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Future Work

- Fabricate high-quality HEA alloys
- Characterize the mechanical properties of the refined alloys
 - Conduct nanoindentation experiments
 - Perform compression tests
 - Analyze and model in-situ neutron levitation results
 - Carry out creep tests ex-situ as well as in-situ neutron diffraction
 - Verify creep behavior by crystal-plasticity finite-element simulations.
- Study the performance of HEAs after aging
- Conduct experimental validation of the developed thermodynamic database based on the experimental data from both literature and our team members.

38


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Published Papers and Presentations

Papers:

- [1] "Fatigue Behavior of Al0.5CoCrCuFeNi High Entropy Alloys", M. A. Hemphill, T. Yuan, G. Y. Wang, J. W. Yeh, C. W. Tsai, A. Chung, and P. K. Liaw, *Acta Materialia*, Vol. 60, No. 16, pp. 5723-5734 (2012).
- [2] "Microstructure and Compressive Properties of NbTiVTaAlx High Entropy Alloys", X. Yang, Y. Zhang, and P. K. Liaw, *Procedia Engineering*, Vol. 36, pp. 292-298 (2012).
- [3] "Alloy Design and Properties Optimization of High Entropy Alloys", Y. Zhang, X. Yang, and P. K. Liaw, *ISIJ*, Vol. 64, No. 7, pp. 810-818 (2012).
- [4] "Processing and Properties of High-Entropy Alloys and Micro- and Nano-Wires", Y. Zhang, T. T. Zuo, W. B. Liao, and P. K. Liaw, *ECS Transactions*, Vol. 41, No. 30, pp. 49-60 (2012).
- [5] "Local Atomic Structure of a High Entropy Alloy: An X-Ray and Neutron Scattering Study", W. Guo, W. Dmowski, J. Y. Noh, P. Rack, P. K. Liaw, and T. Egami, *Metallurgical and Materials Transactions A-Physical Metallurgy and Materials Science*, Vol. 44A, No. 5, pp. 1994-1997 (2013).
- [6] "Mechanical Properties of the High-Entropy Alloy Ag0.5CoCrCuFeNi at Temperatures of 4.2-300K", M. A. Laktionova, E. D. Tabachnikova, Z. Tang, and P. K. Liaw, *Low Temperature Physics*, Vol. 39, No. 7, pp. 630-632 (2013).
- [7] "A Successful Synthesis of the CoCrFeAl0.3 Single-Crystal, High-Entropy Alloy by Bridgman Solidification", S. G. Ma, S. F. Zhang, M. C. Gao, P. K. Liaw, and Y. Zhang, *ISIJ*, Vol. 65, No. 12, pp. 1751-1758 (2013).


39

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Papers (Cont'd):

- [8] "A Tensile Deformation Model for in-Situ Dendrite/Metallic Glass Matrix Composites", J. W. Qiao, T. Zhang, F. Q. Yang, P. K. Liaw, S. Pauliy, and B. S. Xu, *Scientific Reports*, Vol. 3, pp. 2816 (2013).
- [9] "Aluminum Alloying Effects on Lattice Types, Microstructures, and Mechanical Behavior of High-Entropy Alloys Systems", Z. Tang, M. C. Gao, H. Diao, T. Yang, J. Liu, T. Zuo, Y. Zhang, Z. Lu, Y. Cheng, K. A. Dahmen, P. K. Liaw, and T. Egami, *ISIJ*, Vol. 65, No. 12, pp. 1848-1858 (2013).
- [10] "Nature of the Interfaces between the Constituent Phases in the High Entropy Alloy CoCrFeNiAl", B. A. Welk, R. E. Williams, G. B. Viswanathan, M. A. Gibson, P. K. Liaw, and H. L. Fraser, *Intermetallics*, Vol. 134, pp. 193-9 (2013).
- [11] "High-Entropy Alloys with High Saturation Magnetization, Electrical Resistivity, and Malleability", Y. Zhang, T. T. Zuo, Y. Q. Cheng, and P. K. Liaw, *Scientific Reports*, Vol. 3, No. (2013).
- [12] "Processing Effects on the Magnetic and Mechanical Properties of FeCoNiAl26012 High Entropy Alloy", T. T. Zuo, S. B. Ren, P. K. Liaw, and Y. Zhang, *International Journal of Minerals Metallurgy and Materials*, Vol. 20, No. 6, pp. 549-555 (2013).
- [13] "Microstructures and Cracking Noise of AlNbTiMoV High Entropy Alloys", S. Chen, X. Yang, K. A. Dahmen, P. Liaw, and Y. Zhang, *Entropy*, Vol. 16, No. 2, pp. 870-884 (2014).
- [14] "Symposium on High-Entropy Alloys Foreword", P. K. Liaw, G. Y. Wang, M. C. Gao, and S. N. Mathaudu, *Metallurgical and Materials Transactions A-Physical Metallurgy and Materials Science*, Vol. 45A, No. 1, pp. 179-179 (2014).


40

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Papers (Cont'd):

- [15] "Alloying and Processing Effects on the Aqueous Corrosion Behavior of High-Entropy Alloys", Z. Tang, L. Huang, W. He, and P. K. Liaw, *Entropy*, Vol. 16, No. 2, pp. 895-911 (2014).
- [16] "Microstructures and Properties of High-Entropy Alloys", Y. Zhang, T. T. Zuo, Z. Tang, M. C. Gao, K. A. Dahmen, P. K. Liaw, and Z. P. Lu, *Progress in Materials Science*, Vol. 61, pp. 1-93 (2014).
- [17] "Microstructures and Cracking Noise of AlNbTiMoV High Entropy Alloys", S. Y. Chen, X. Yang, K. A. Dahmen, P. K. Liaw, and Y. Zhang, *Entropy*, Vol. 16, pp. 870-884 (2014).


41

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Presentation (Cont'd):

- ◆ The 9th International Conference on Bulk Metallic Glass (BMG-IX) 2012, Xiamen, China
 - **Computational Thermodynamics Aided High-Entropy Alloy Design**, C. Zhang, F. Zhang, S. L. Chen, W. S. Cao, Z. Tang, P. K. Liaw
- ◆ 2013 TMS Meeting, San Antonio, TX, USA, March 3-9, 2013
 - **Automatic Fabrication of High-Entropy Alloys and Their Properties**, Y. Yokoyama, X. Xie, J. Antonaglia, M. Hemphill, T. Zhu, T. Yuan, G. Wang, C. Tsai, J. Yeh, A. Chung, K. Dahmen, P. K. Liaw (invited)
 - **Extracting Materials Properties from Cracking Noise and Slip-Avalanche Statistics in Slowly-Sheared Materials**, K. Dahmen, X. Xie, J. Antonaglia, M. Laktionova, E. Tabachnikova, Z. Tang, J. Qiao, J. Gner, J. W. Yeh, J. U. P. Liaw
 - **Non-Equilibrium and Equilibrium Phases in AlCoCrFeNi High-Entropy Alloys**, Z. Tang, O. Senkov, C. Parish, L. Santodonato, D. Miracle, G. Wang, C. Zhang, F. Zhang, P. K. Liaw
 - **Ordering Behavior in the AlCoCrCuFeNi High-Entropy Alloys**, L. Santodonato, Y. Zhang, M. Gao, C. Parish, M. Feyyenson, Z. Tang, J. Neerfeld, R. Weber, P. K. Liaw
 - **Computational Modeling of High-Entropy Alloys**, M. Gao, D. Tafan, J. Hawk, Y. Wang, M. Widom, L. Santodonato, P. K. Liaw (invited)


42

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Presentation (Cont'd):

- **Mixing Phase and Defect Effects on Fatigue Behavior of Wrought Al0.5CoCrCuFeNi High-Entropy Alloys**, Z. Tang, M. Hemphill, T. Yuan, G. Wang, J. Yeh, C. Tsai, P. K. Liaw
- **Phase Separation and Intermetallic Formation in "High-Entropy" Alloys**, C. Parish, M. Miller, L. Santodonato, Z. Tang, P. K. Liaw
- **Computational Thermodynamics Aided High-Entropy Alloy Design**, C. Zhang, F. Zhang, S. Chen, W. Cao, J. Zhu, Z. Tang, P. K. Liaw
- **Statistical Fatigue-Life Modeling for High-Entropy Alloys**, T. Yuan, M. Hemphill, Z. Tang, G. Wang, A. Chung, C. Tsai, J. Yeh, P. K. Liaw (invited)
- **A Combinatorial Approach to the Investigation of Metal Systems that Form Both High-Entropy Alloys and Bulk Metallic Glasses**, B. Welk, P. K. Liaw, M. Gibson, H. Fraser
- ◆ 2014 TMS Meeting, San Diego, CA, USA, February 16-20, 2014
 - **Aluminum Alloying Effects on Lattice Types, Microstructures and Mechanical Behavior of High-Entropy Alloys Systems**, Z. Tang, M. Gao, H. Y. Diao, T. F. Yang, J. P. Liu, T. T. Zuo, Y. Zhang, Z. P. Lu, Y. Q. Cheng, Y. W. Zhang, K. Dahmen, P. K. Liaw, T. Egami
 - **The Influence of Cu and Al on the Microstructure, Mechanical Properties and Deformation Mechanisms in the High Entropy Alloys CoCrNiFeCuCoNiFeAl1.5 and CrCoNiFeCuAl1.5**, B. Welk, B. B. Viswanathan, M. Gibson, P. K. Liaw, and H. Fraser


43

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Presentation (Cont'd):

- **The Influence of Alloy Composition on the Interrelationship between Microstructure, Mechanical Properties of High Entropy Alloys with BCC/B2 Phase Mixtures**, B. Welk, D. Huber, J. Jensen, G. Viswanathan, R. Williams, P. K. Liaw, M. Gibson, D. Evans, and H. Fraser
- **The Oxidation Behavior of AlCoCrFeNi High-entropy Alloy at 1023, 1223K (750,1050°C)**, Wu Kai, W. S. Chen, C. C. Sung, Z. Tang, and P. K. Liaw
- **Strain-rate Effects on the Structure Evolution of High Entropy Alloys**, X. Xie, J. Antonaglia, J. P. Liu, Z. Tang, J. W. Qiao, G. Y. Wang, Y. Zhang, K. Dahmen, and P. K. Liaw
- **Neutron diffraction studies on creep deformation behavior in a high-entropy alloy CoCrFeMnNi under high temperature and low strain rate**, W. C. Woo, E. W. Huang, J. W. Yeh, P. K. Liaw, and H. Choo
- **The Hot Corrosion Resistance Properties of AlFeCoCrNi**, S. Z. Yang, M. Habibi, L. Wang, S. M. Guo, Z. Tang, P. K. Liaw, L. X. Tan, C. Guo, and M. Jackson
- **Using the Statistics of Strations in the Stress-Strain Curves to Extract Materials Properties of Slowly-sheared High Entropy Alloys**, Karim Dahmen, X. Xie, J. Antonaglia, M. Laktionova, E. Tabachnikova, J. W. Qiao, J. W. Yeh, C. W. Tsai, J. U. P. Liaw

44

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Presentation (Cont'd):

- **Environmental-temperature Effect on a Ductile High-entropy Alloy Investigated by In Situ Neutron-diffraction Measurements**, E. W. Huang, C. Lee, D. J. Yu, K. An, P. K. Liaw, and J. W. Yeh
- **Mechanical Behavior of an Al0.1CoCrFeNi High Entropy Alloy**, M. Komarsamy, N. Kumar, Z. Tang, R. Mishra, and P. K. Liaw
- **Characterizing Multi-component Solid Solutions Using Order Parameters and the Bragg-Williams Approximation**, L. Santodonato, and P. K. Liaw
- **Ultra Grain Refinement in High Entropy Alloys**, N. Tsuji, I. Watanabe, N. Park, D. Terada, A. Shibata, Y. Yokoyama, P. K. Liaw
- **Nanostructure Evolution through High-pressure Torsion and Recrystallization in a High-entropy CoNiFeCuNi Alloy**, N. Park, A. Shibata, D. Terada, Y. Yokoyama, P. K. Liaw, and N. Tsuji
- **Distinguished Work-hardening Capacity of a Ti-based Metallic Glass Matrix Composite upon Dynamic Loading**, J. W. Qiao, H. J. Yang, Z. H. Wang, and P. K. Liaw
- ◆ The 10th International Conference on Bulk Metallic Glass 2014, Shanghai, China, University of Science and Technology, Beijing, June 6-16, 2014
 - **Characterization of Serrated Flows in BMG and HEAs**, X. Xie, S. Y. Chen, J. Auto, J. P. Liu, J. W. Qiao, P. K. Liaw (invited)
- ◆ National Institute of Materials Science, Japan, 2014
 - **Fatigue Behavior of BMG and HEAs**, X. Xie, G. Y. Wang, P. K. Liaw

45

Conclusions

- ❖ We have fabricated the Al_{0.1}CrCuFeMnNi alloy, Al_{0.3}CrCuFeMnNi alloy, and Al_{0.8}CrCuFeMnNi alloys.
- ❖ The Al_{0.1}CrCuFeMnNi alloy has three phases, FCC1, FCC2, and BCC. The Al_{0.8}CrCuFeMnNi alloys have the structure, including three phases, disordered BCC1, BCC2, and FCC phases.

Conclusions (Cont'd)

- ❖ As the aluminum content increases, the yield stress increases and plasticity ductility decreases.
- ❖ Al_{0.1}CrCuFeMnNi shows good compression plasticity at 700°C and 800°C.
- ❖ Continue analyzing and modeling in-situ neutron levitation results.
- ❖ All binary and most of ternary systems have been thermodynamically modeled.

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Thank You!

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Hot Compression Tests

Background and Unique Behavior (Cont'd)
 Compression behavior at room temperature
 BCC + FCC matrix, heat-treatment effect

Compressive true stress-strain curves of Al_{0.1}CrCuFeNi after heat treatment for 5 h at different temperatures. The mechanical properties of this alloy have a strong correlation with the aging temperature.

L. H. Wen, H. C. Kou, J. S. Li, H. Chang, X. Y. Xue, and L. Zhou, *Intermetallics* 17, 266 (2009).

Background and Unique Behavior (Cont'd)

The Al_xCrCuFeMnNi system is good for high-temperature applications

The Al_xCrCuFeMnNi system is a mix of FCC and BCC phases.
 Al = 0, 0.3
 FCC1 + FCC2 + BCC
 Al = 0.5, 0.8
 FCC + BCC1 + BCC2
 Al = 1
 BCC1 + BCC2

H. Y. Chen, C. W. Tai, C. C. Tung, J. W. Yeh, T. T. Shun, C. C. Yang, and S. K. Chen, *Annals De Chimie - Science Des Materiaux* 31, 685 (2006).

Preliminary Results - Alloy Fabrication

- Fabricate multiple alloy compositions belonging to the Al_xCrCuFeMnNi system varying the Al content from 0 to 2.
- Alloys sensitive to impurities and defects introduced during casting and processing
- Reduce defects during fabrication

M. A. Hemphill, T. Yuan, G. Y. Wang, J. W. Yeh, C. W. Tai, A. Chuang, and P. K. Liaw, *Acta Materialia* 60, 5723 (2012).

PDF Studies at the SNS NOMAD

- Indications of a locally-strained crystal lattice

The neutron-pair-distribution-function (PDF) data and the calculated PDF agree well at larger distances. As shown by the difference curve, the agreement below 10 angstrom is worse. This trend is consistent with the expectation that HEAs are locally strained, and yet, possess long-range crystal order.

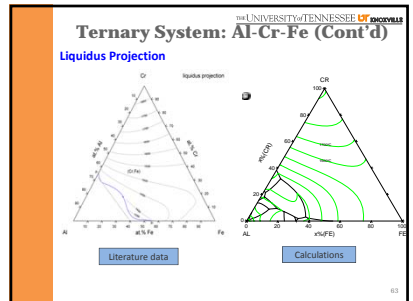
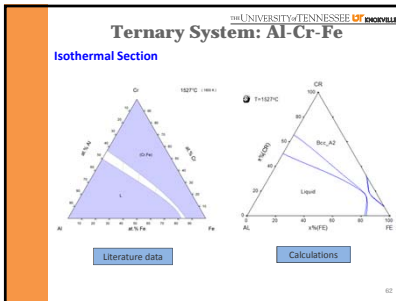
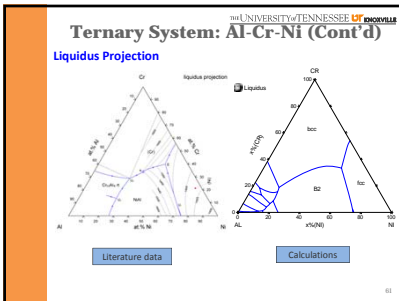
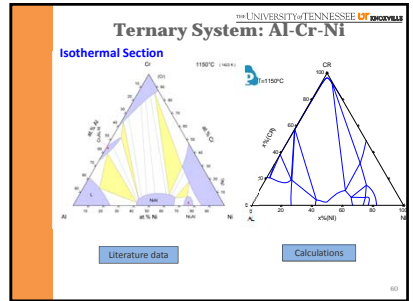
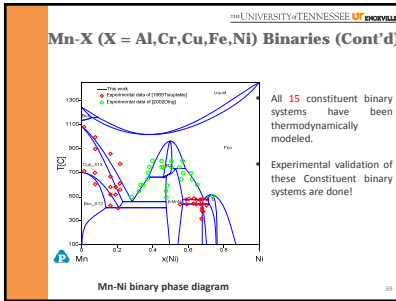
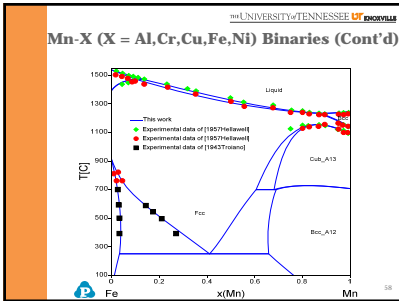
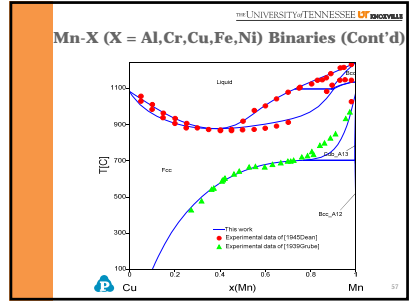
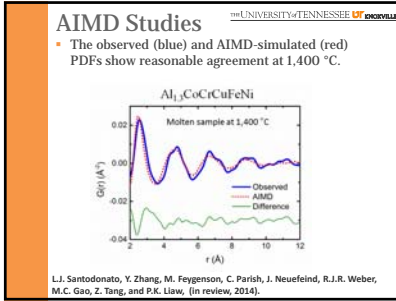
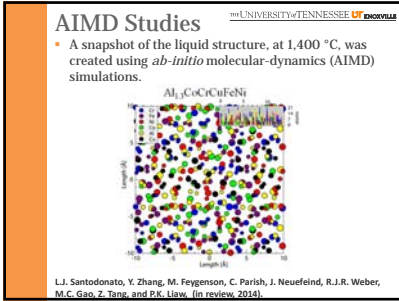
L.J. Santodonato, Y. Zhang, M. Feygenson, C. Parish, J. Neufeld, R.J.R. Weber, M.C. Gao, Z. Tang, and P.K. Liaw, (in review, 2014).

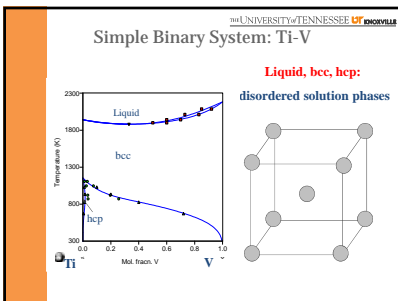
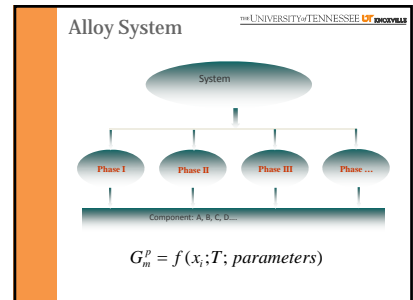
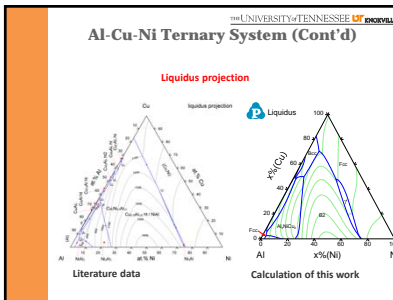
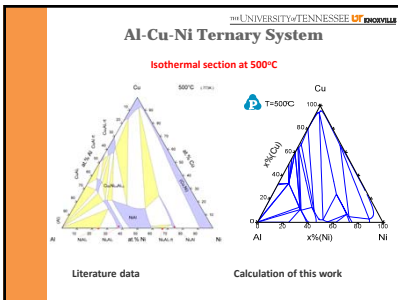
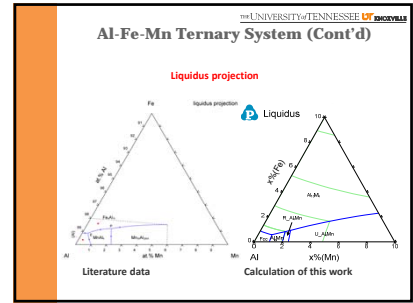
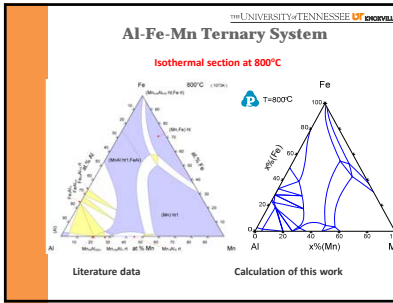
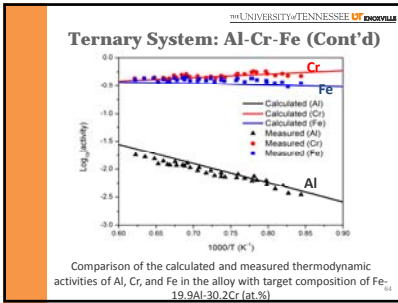
PDF Studies at the SNS NOMAD

- Molten sample

At 1,400 °C, the observed PDF shows short-range order, consistent with a liquid phase, and agrees roughly with a model based upon local structures (less than 20Å diameters) composed B2 and FCC cells

L.J. Santodonato, Y. Zhang, M. Feygenson, C. Parish, J. Neufeld, R.J.R. Weber, M.C. Gao, Z. Tang, and P.K. Liaw, (in review, 2014).





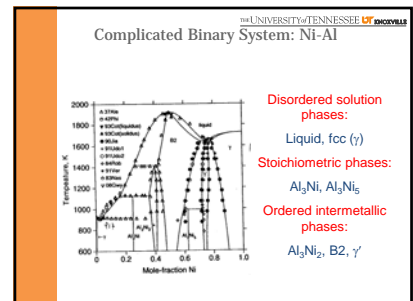
Gibbs Energy Function of Substitutional Solution Model

$$G = x_A \cdot G_A^{\phi, \rho} + x_B \cdot G_B^{\phi, \rho} + RT \cdot (x_A \ln x_A + x_B \ln x_B) + G^{\text{CS}}$$

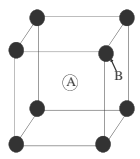
x_A, x_B --- mole fractions of component A and B.
 $G_A^{\phi, \rho}, G_B^{\phi, \rho}$ --- Gibbs energy of pure A, and B with ϕ structure.
 R --- Gas constant.
 T --- Temperature.
 $G^{\text{CS}} = x_A \cdot x_B \cdot \sum_{i,j} \lambda_{ij} \cdot (x_A - x_B)^j$

λ_{ij} --- Model parameters, obtained by optimization, using available experimental data

Regular solution model: $G^{\text{CS}} = \lambda_0 \cdot x_A \cdot x_B$



Other Model Types (I)



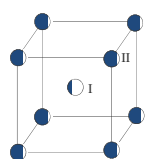
Stoichiometric Phase:

$$G = x_A C_A^{\phi} + x_B C_B^{\phi} + \Delta G_f$$

$$\Delta G_f = a + b \cdot T + c \cdot T \cdot \ln T + \dots$$

a, b, c, ...: parameters

Other Model Types (II)



Ordered Intermetallic Phase:

$$y_p^i$$

$$\sum_{p=A,B} y_p^i = 1 \quad (y_A^i + y_B^i = 1)$$

$$G = G(x, y, T, \text{parameters})$$

Other Model Types (II)

$$G_i^{\phi} = \sum_{i=A,B} \sum_{j=A,B} y_i^j y_j^i G_{ij}^{\phi} + RT \ln \frac{p}{p+q} \sum_{i=A,B} y_i^j \ln y_i^j + \frac{q}{p+q} \sum_{i=A,B} y_i^j \ln y_i^j$$

$$+ \sum_{i=A,B} \sum_{j=A,B} y_i^j y_j^i \sum_{l=A,B} (y_l^i - y_l^j) L_{i,l,j}^{\phi} + \sum_{i=A,B} \sum_{j=A,B} y_i^j y_j^i \sum_{l=A,B} (y_l^i - y_l^j) L_{i,l,j}^{\phi}$$

$$+ y_A^i y_B^j y_A^j y_B^i L_{A,B,A,B}^{\phi}$$
