

Exploring the Behavior of Shales as Seals and Storage Reservoirs for CO₂

Project Number 90210

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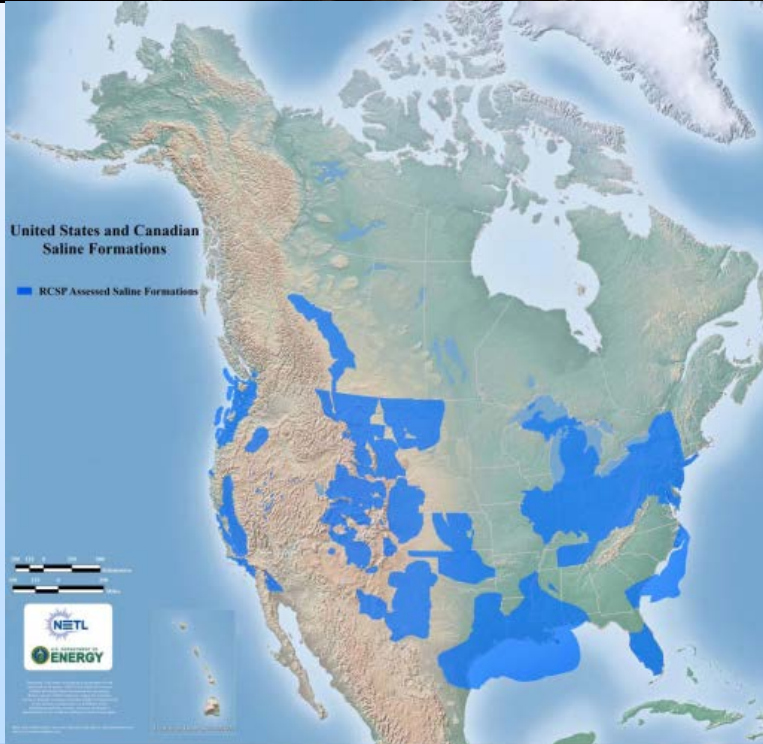
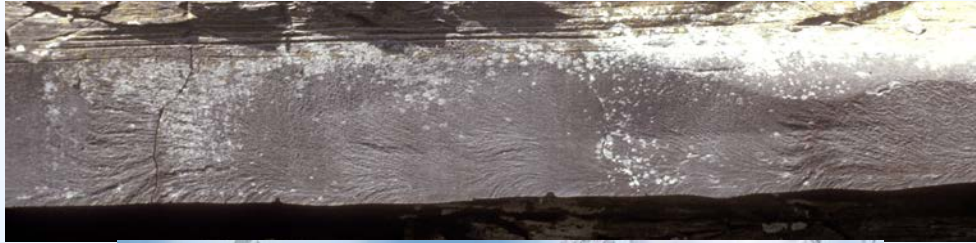
U.S. Department of Energy
National Energy Technology Laboratory
Carbon Storage R&D Project Review Meeting
Developing the Technologies and
Infrastructure for CCS
August 12-14, 2014

Presentation Outline

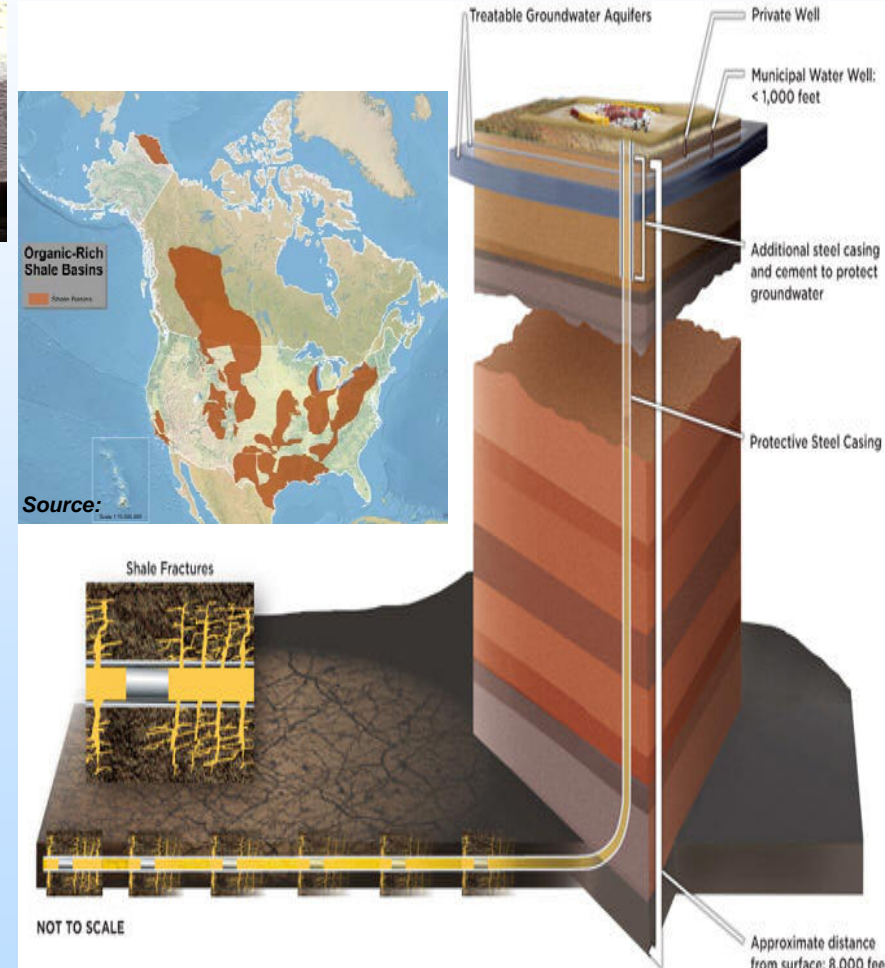
- Benefits to Program
- Project Goals and Objectives
- Technical Status
- Accomplishments to Date
- Summary

Technical Scope

Shales as Seals



Shales as Storage Reservoirs



Sources: HF illustration from National Energy Technology Laboratory (NETL), 2011), Micro CT images by Rebecca Rodriguez, ORISE; Shale image from Reference: Lacazette, A. and Engelder, T. (1992) Fluid-driven cyclic propagation of a joint in the Ithaca Siltstone, Appalachian Basin, New York: p. 297 - 323 in B. Evans and T.-F. Wong (editors): Fault Mechanics and Transport Properties of Rocks; a festschrift in honor of W. F. Brace: Academic Press, San Diego.; NETL Carbon Storage Atlas IV (2012)

Benefit to the Program

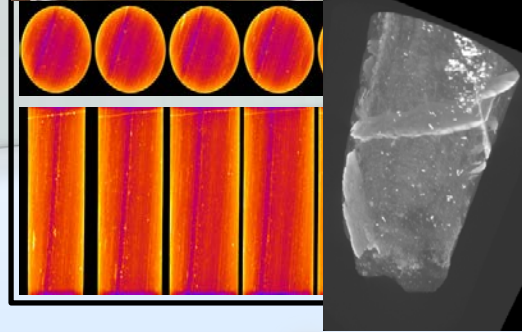
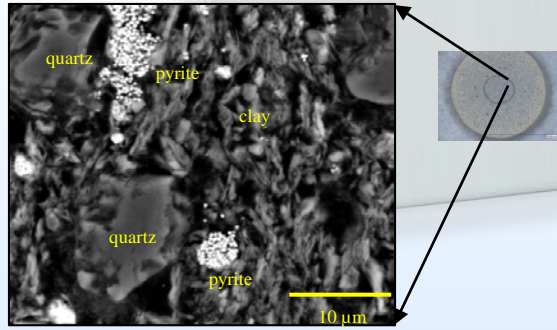
- Carbon Storage Program Goals Addressed:
 - Support industry's ability to predict CO₂ storage capacity in (*unconventional*) geologic formations to within ± 30 percent
 - Ensuring 99 percent storage permanence.
- Project Benefits:
 - Improve understanding of injection/storage performance of unconventional formations
 - Inform efficiency estimation for resource assessment
 - Insights feeding to seal characterization in integrated assessment of risk

Project Overview: Goals and Objectives

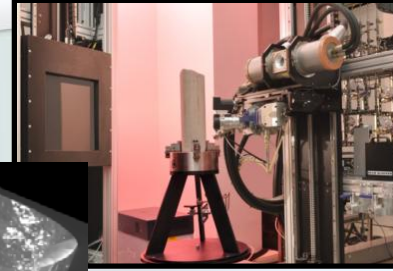
- Project Objectives
 - Evaluate matrix response to CO₂ exposure (sorption, swelling/shrinkage, geochemical interactions)
 - Characterize effective permeability and porosity of shale to CO₂
 - Experimental and simulation-based performance of CO₂ storage in/transport through shale with natural and engineered fractures
 - Reduced order characterization to improve resource estimation and quantitative risk assessment of geologic CO₂ storage

Science Base Feeding to Higher-Level Assessments

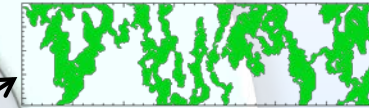
Micro-Scale Data Collection (CT, SEM, etc)



Core-Scale Flow and Imaging



Data Conversion & Computational Fluid Dynamics

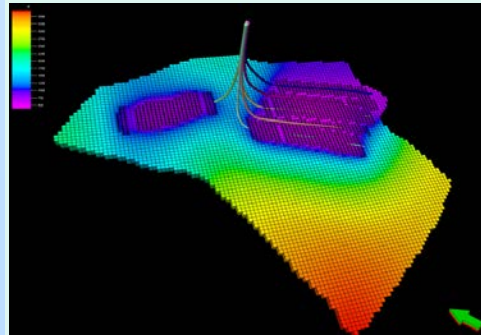


Gas/liquid flowing in rock fractures

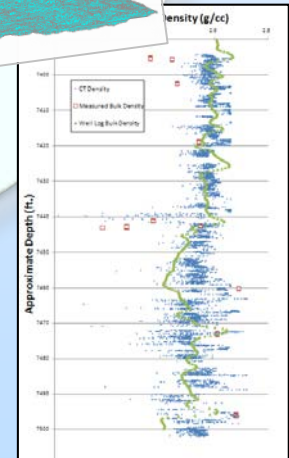
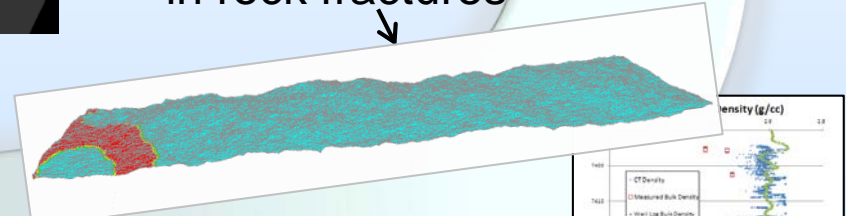
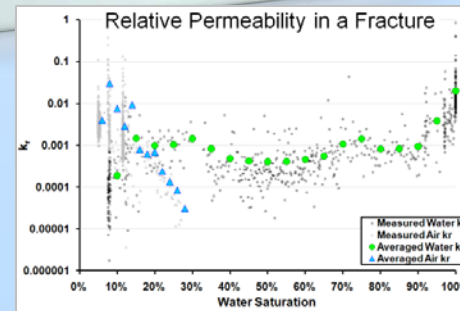
National-Scale Assessment



Well and Field-Scale Simulation



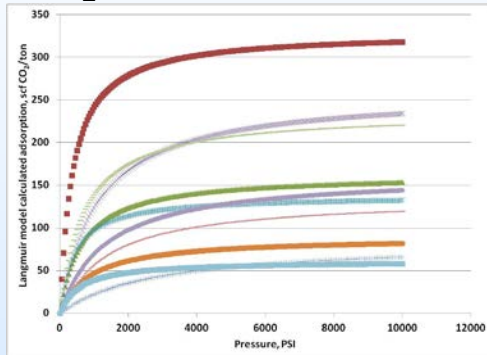
Multiscale Data Analysis



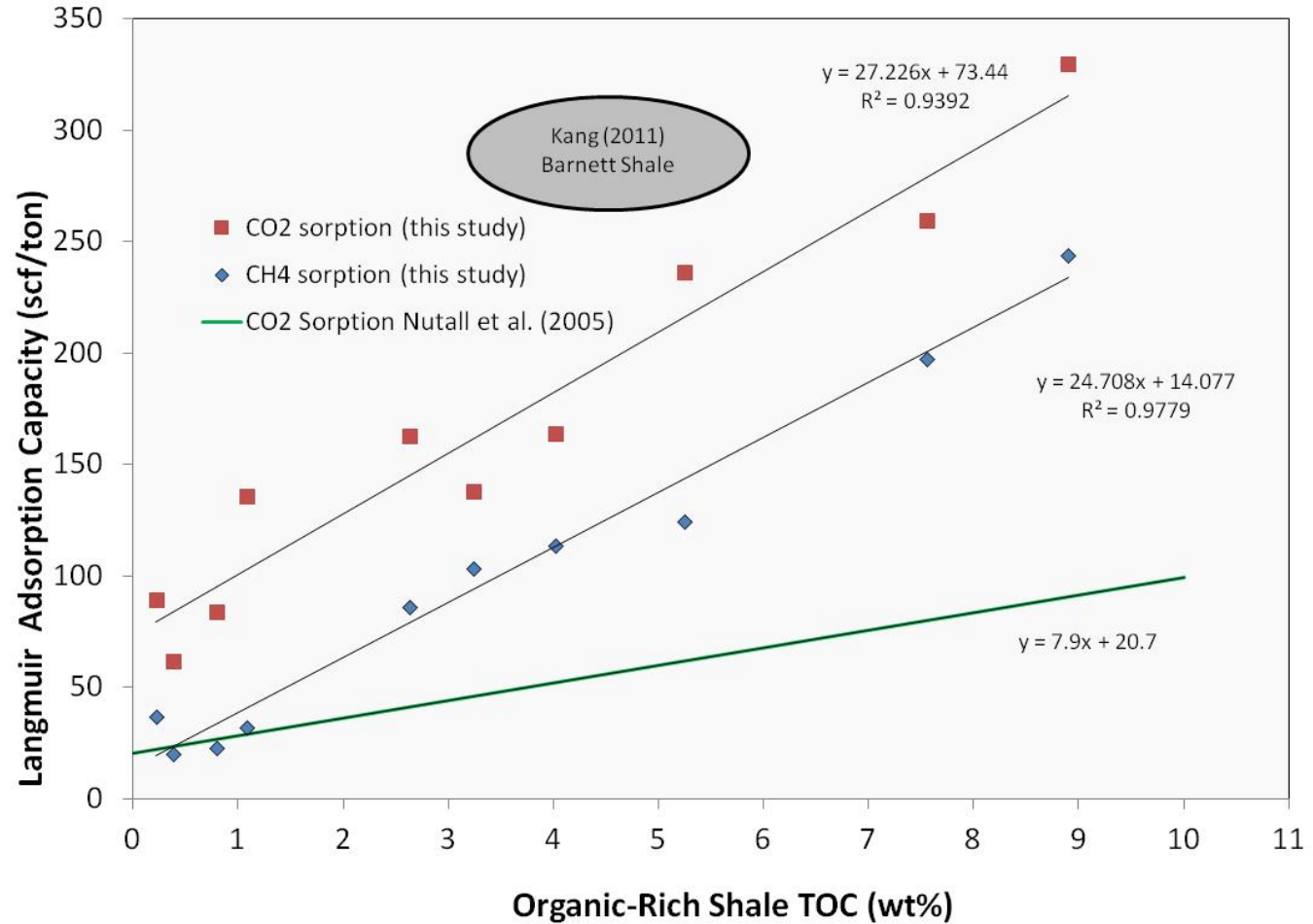
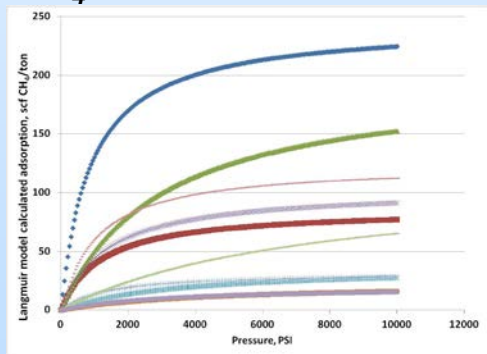
Comparison of Shale Density from CT Scans and Well Logs

CO₂ and CH₄ Sorption capacity as function of %TOC (single-fluid isotherms)

CO₂ Isotherms



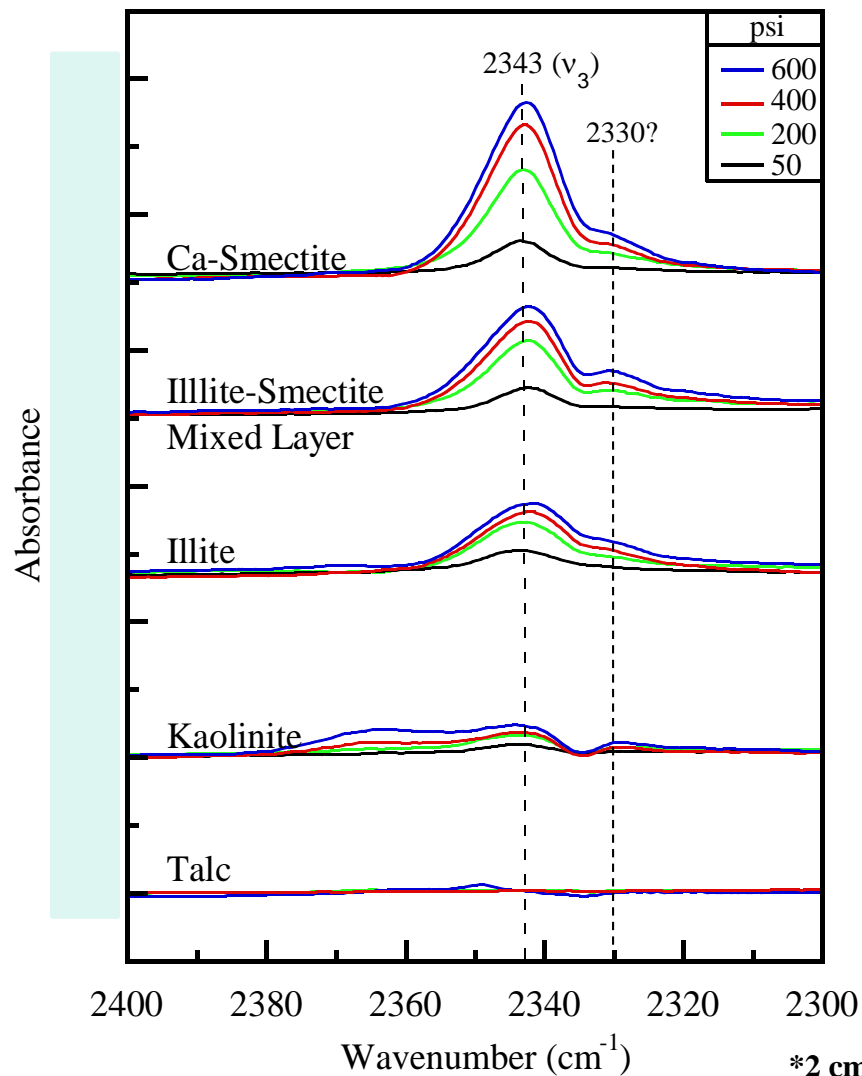
CH₄ Isotherms



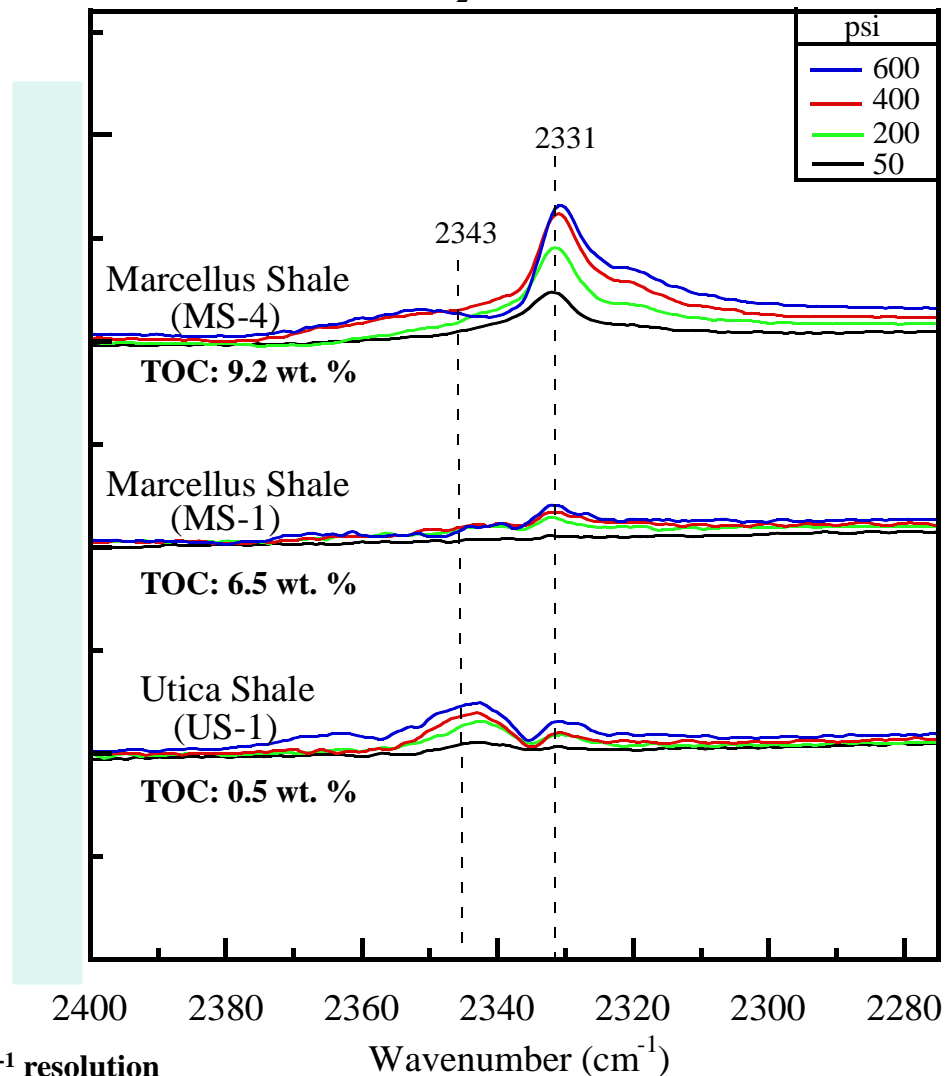
CO₂ Sorption Mechanisms: Fourier Transform-Infrared Spectroscopy (FT-IR)*

15 min CO₂ exposure at 40°C, 0-800 psi

Infrared Spectra of CO₂ Sorption on Clay Standards



Infrared Spectra of CO₂ Sorption onto Shale Samples



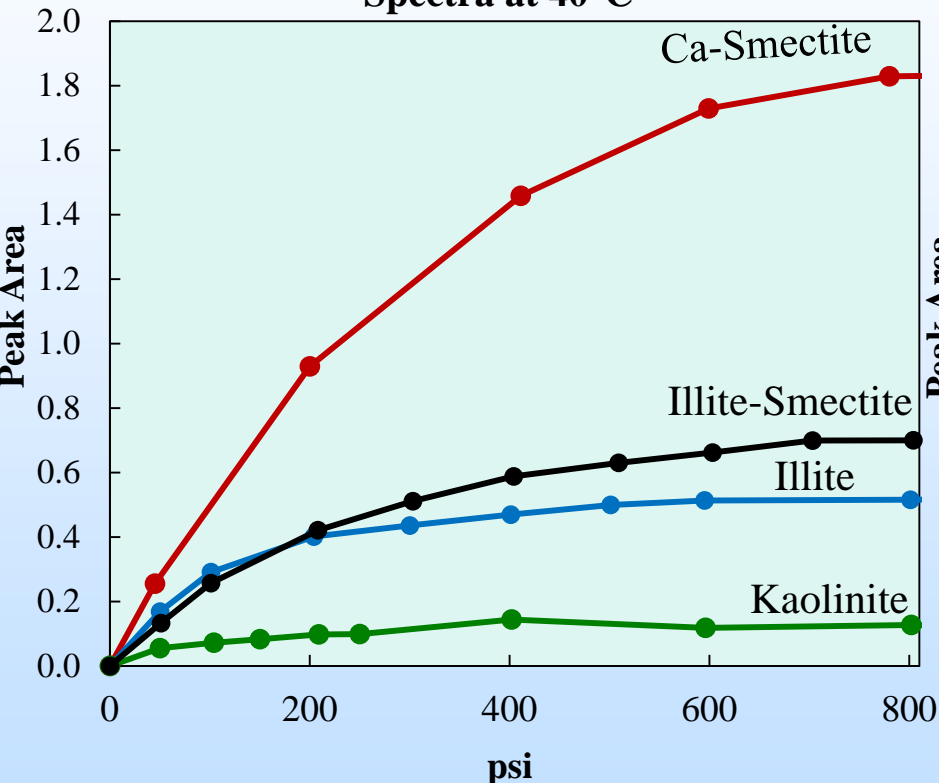
Physically Sorbed CO₂ IR Peaks: 2350-2330 cm⁻¹

CO₂ Sorption on Shale Samples

FT-IR Data:

Area of 2343 cm⁻¹ CO₂ Sorption Peaks

Peak Area vs Pressure for Clay Infrared Spectra at 40°C

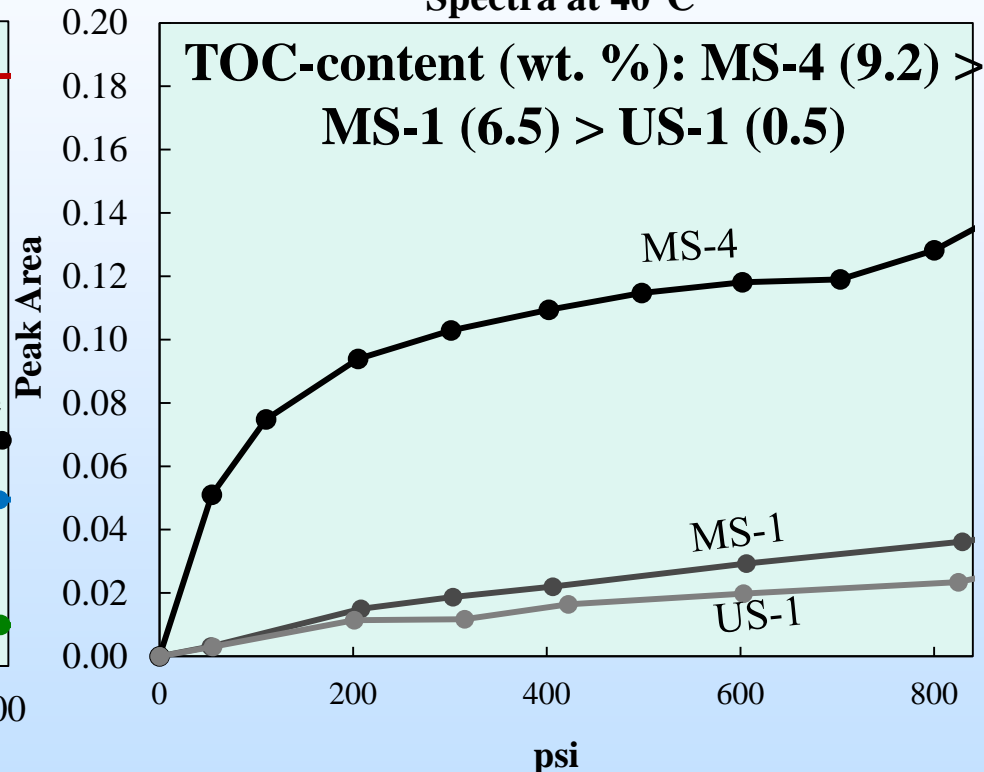


Integrated CO₂ peak area is not quantitative

FT-IR Data:

Area of 2331 cm⁻¹ CO₂ Sorption Peaks*

Peak Area vs Pressure of Shale Infrared Spectra at 40°C



*2343 cm⁻¹ peak not strong enough to obtain reliable area measurements

FT-IR trends compliment results of CO₂ isotherm measurements

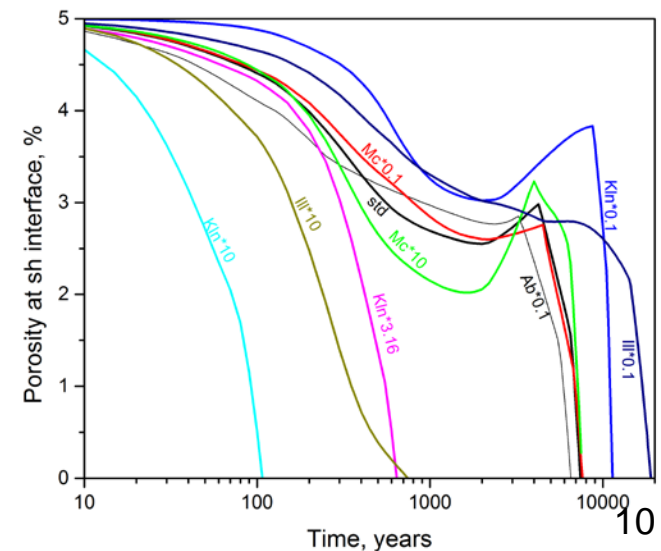
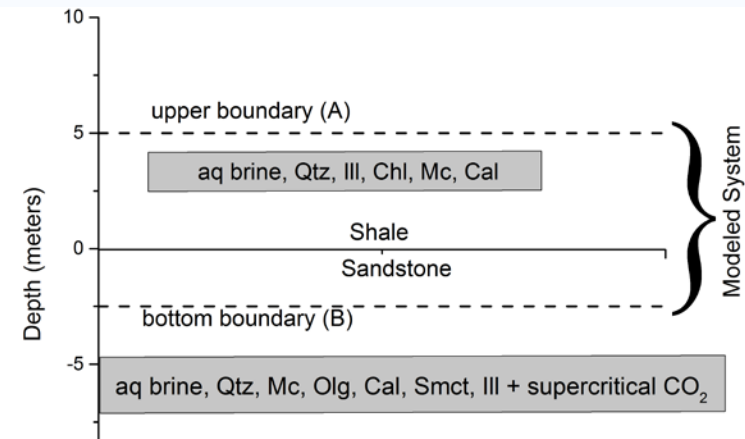


Geochemical Model Sensitivity and Caprock Interface

Study Problem: Geochemical calculations rely on uncertain thermodynamic & kinetic databases

Goal: Characterize the mineral precipitation and dissolution processes that are important at brine/aquifer/caprock interfaces.

Finding: The precipitation and dissolution processes for minerals Chlorite, and carbonates Cc, Dol, Ank contribute to autosealing at the brine/aquifer/caprock interfaces.

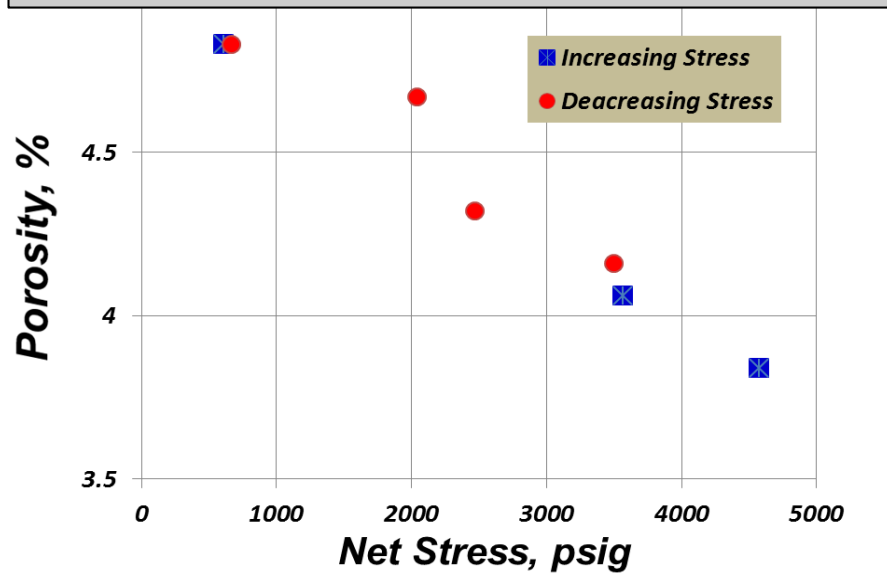


Steady-State Permeameter

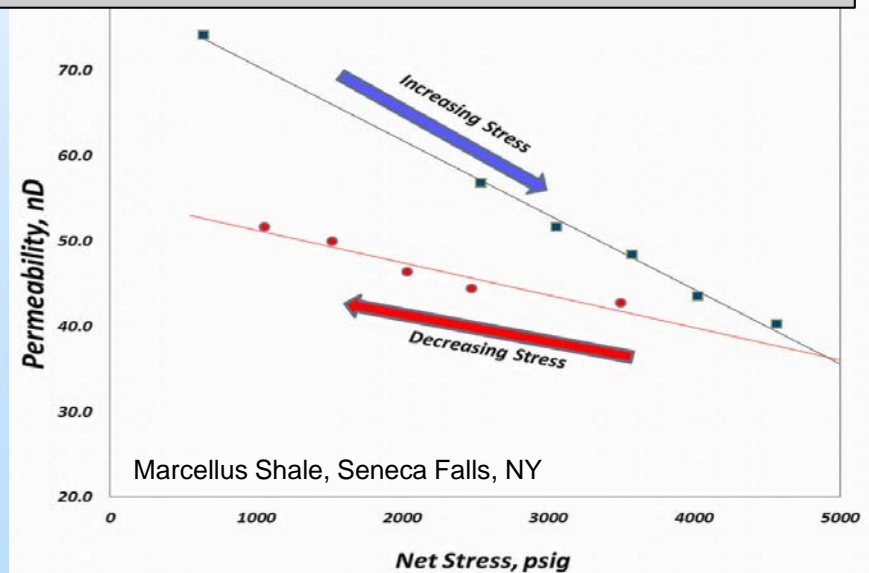
Capable of reproducing in-situ net stress, and measuring gas flow under partial liquid saturation



Effective porosity of shale as function of net stress



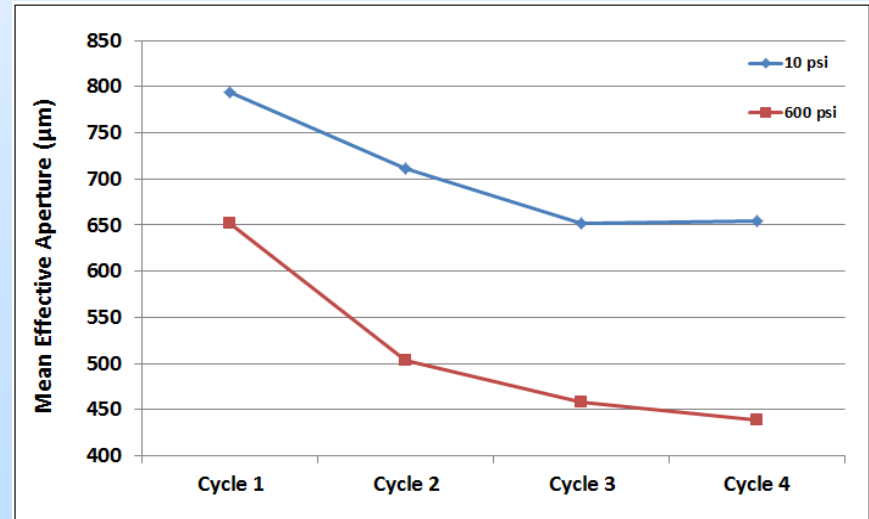
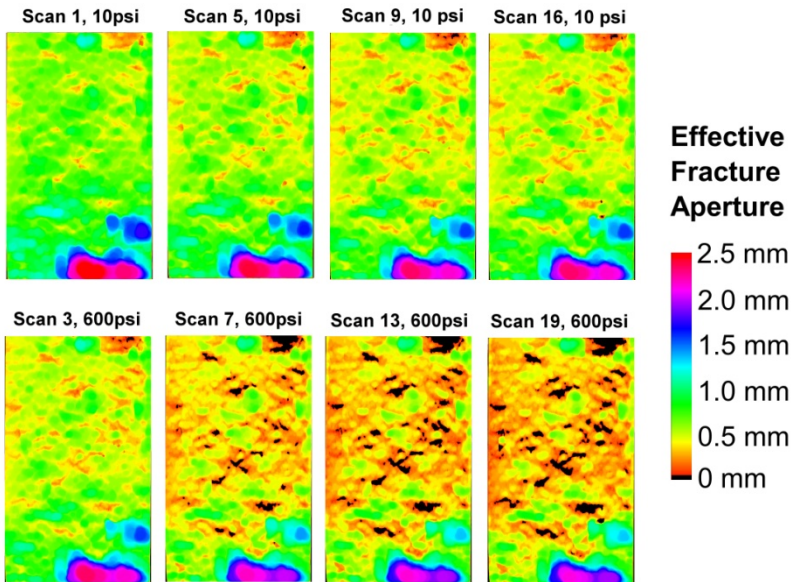
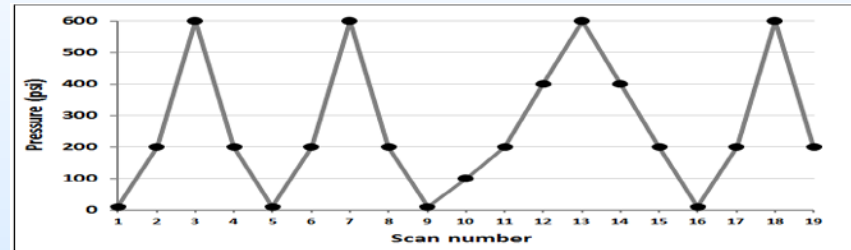
Effective permeability of shale as function of net stress



Coupling Mechanical Changes of Fractures to Hydraulic Changes

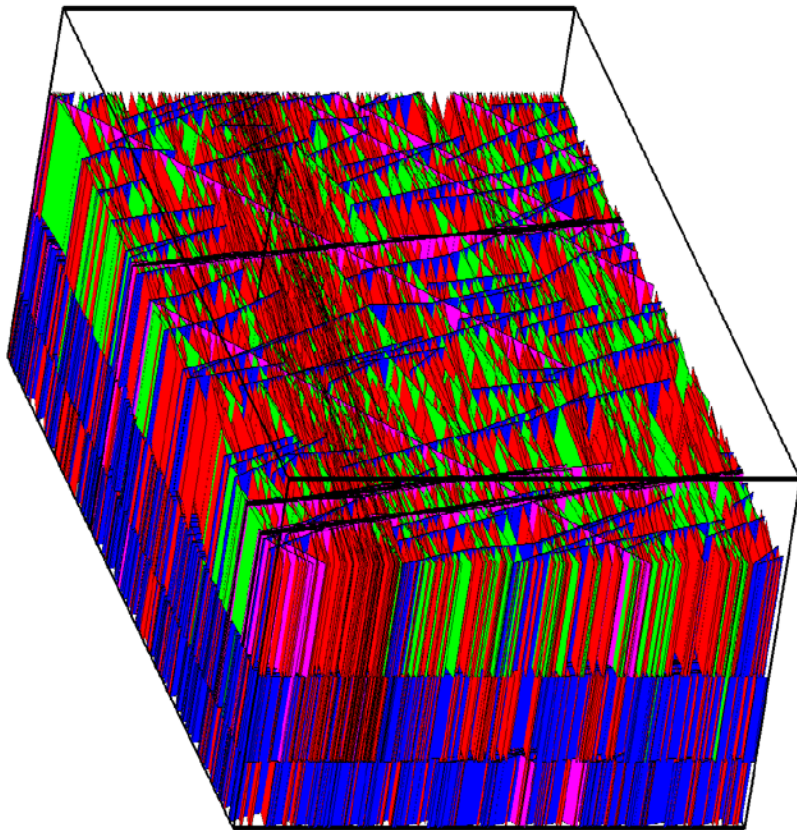


Cycling of confining pressure causes fracture asperities to break down, reducing effective fracture aperture

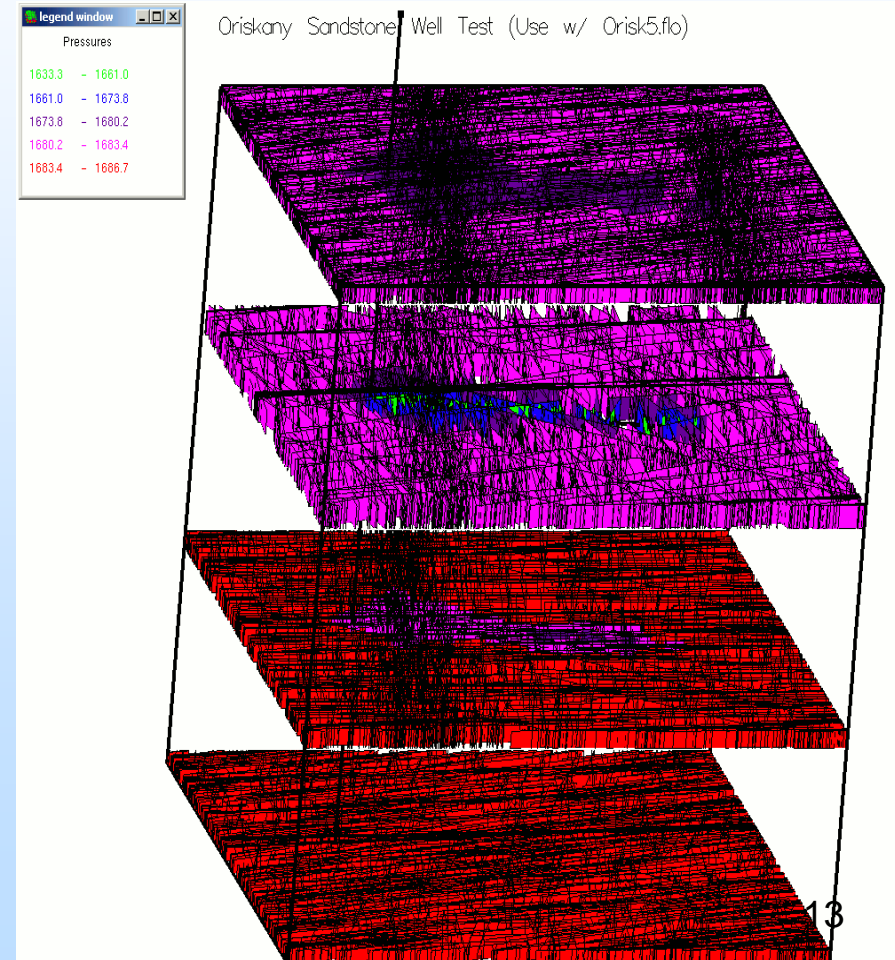


Modeling CO₂ Flow in Fractured Geologic Media

FRACGEN stochastically generates fracture networks



NFFLOW models flow in discrete fracture networks



CO₂ Storage in Depleted Shale Gas Formations

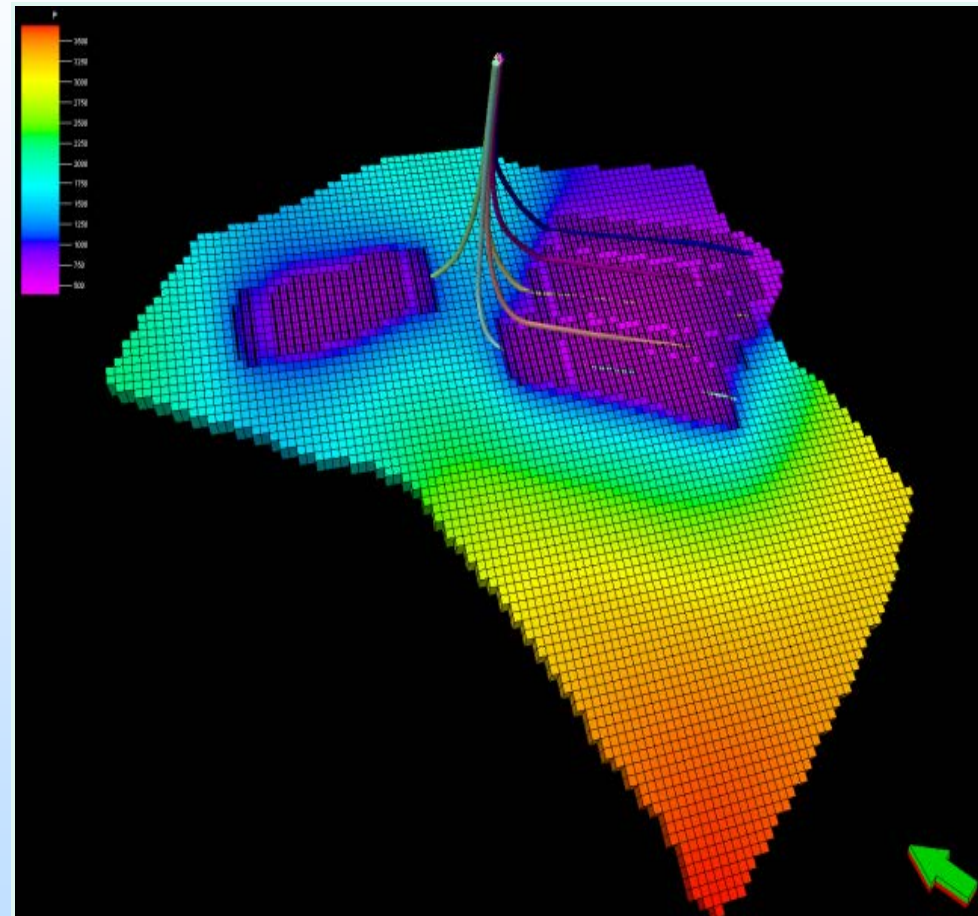


Goal: *Develop a robust characterization of site-scale CO₂ storage and EGR potential of gas-bearing shale formations*

Scenario: Dry gas window, Marcellus, SW PA, Depth of 6,700 ft (~ 2,000 m), gross interval thickness of 120 ft (37 m), 145°F (63°C), Initial pressure 4,000 psi (27.6 MPa), matrix permeability 0.1 -1 (μD)

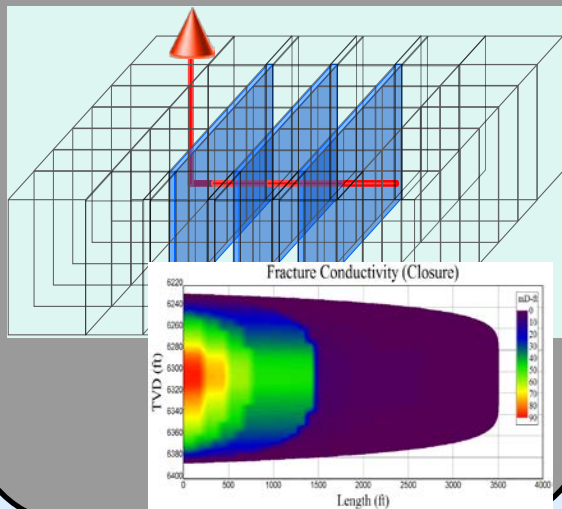
Sensitivity of CO₂ storage/EGR performance to:

- Fracture network characteristics
- Matrix CO₂ and CH₄ sorption characteristics
- Injector/producer distance
- Injection pressure
- Stress-dependent matrix perm.

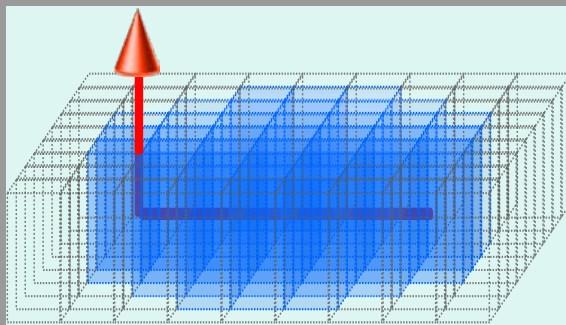


Representing Fracture Networks

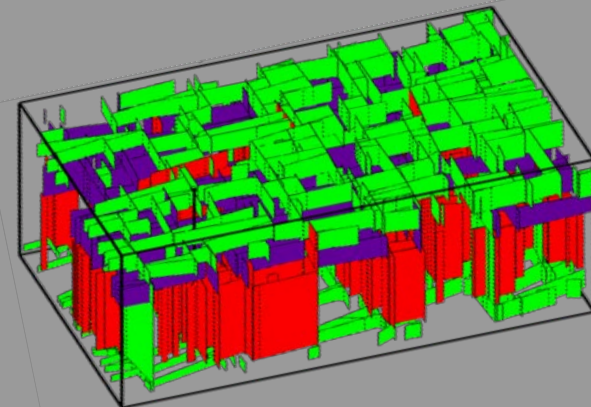
Discrete Transverse Fracture Planes



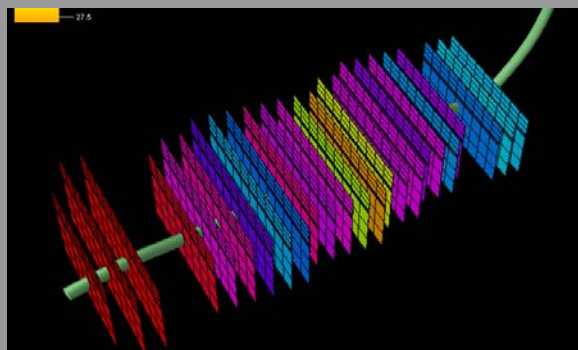
Crushed Zone Representation



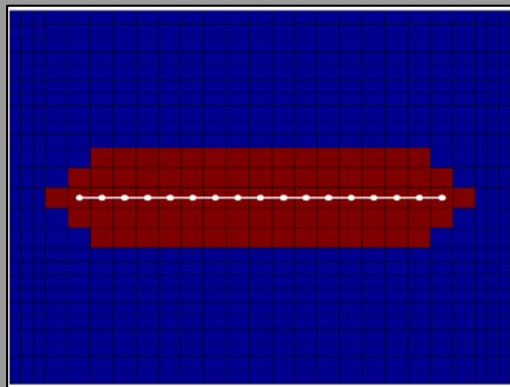
Semi-stochastic fracture Network



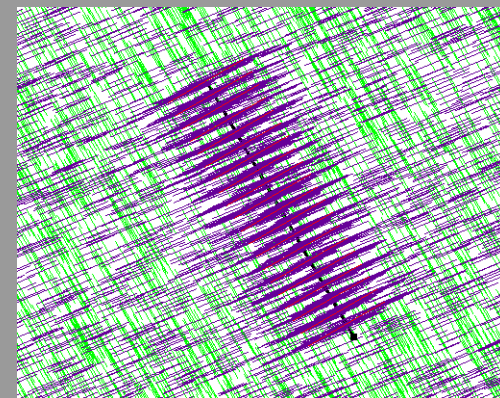
Discrete Fracture Modeling coupled with conventional reservoir simulation



Modified dual porosity, multiphase, compositional, multidimensional flow model



Semi-stochastic fracture network and flow modeling



Single Lateral CO₂ Storage Scenario

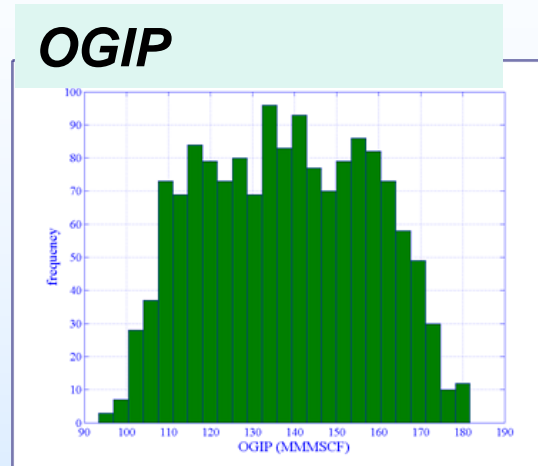
Scenario: Constant pressure at 5000 psi, single lateral

Uncertain Parameters:

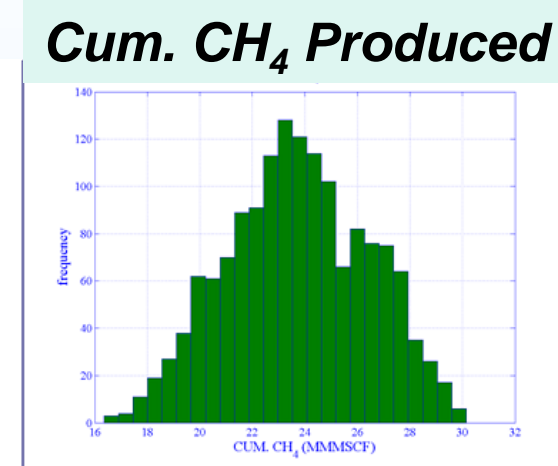
h_{net} , Φ_{matrix} , $\Phi_{fracture}$, k_{matrix} , $k_{fracture}$, fracture spacing, Langmuir constants

MC with 1000 realizations

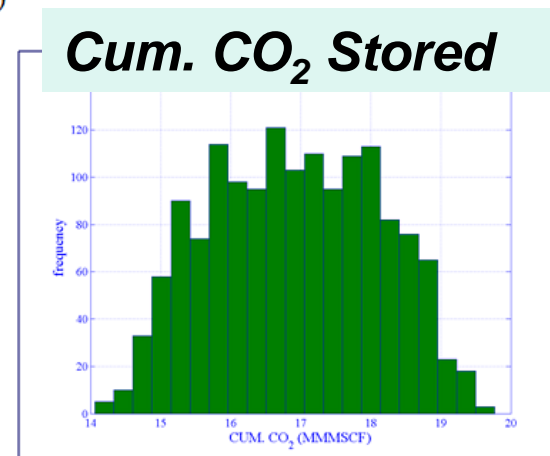
	P ₉₀	P ₅₀	P ₁₀
OGIP (BSCF)	111	138	165
CH ₄ Production over 30 Years (BSCF)	20.1	23.7	27.4
CO ₂ Stored after 30 Years (BSCF)	15.3	16.9	18.5



(a)



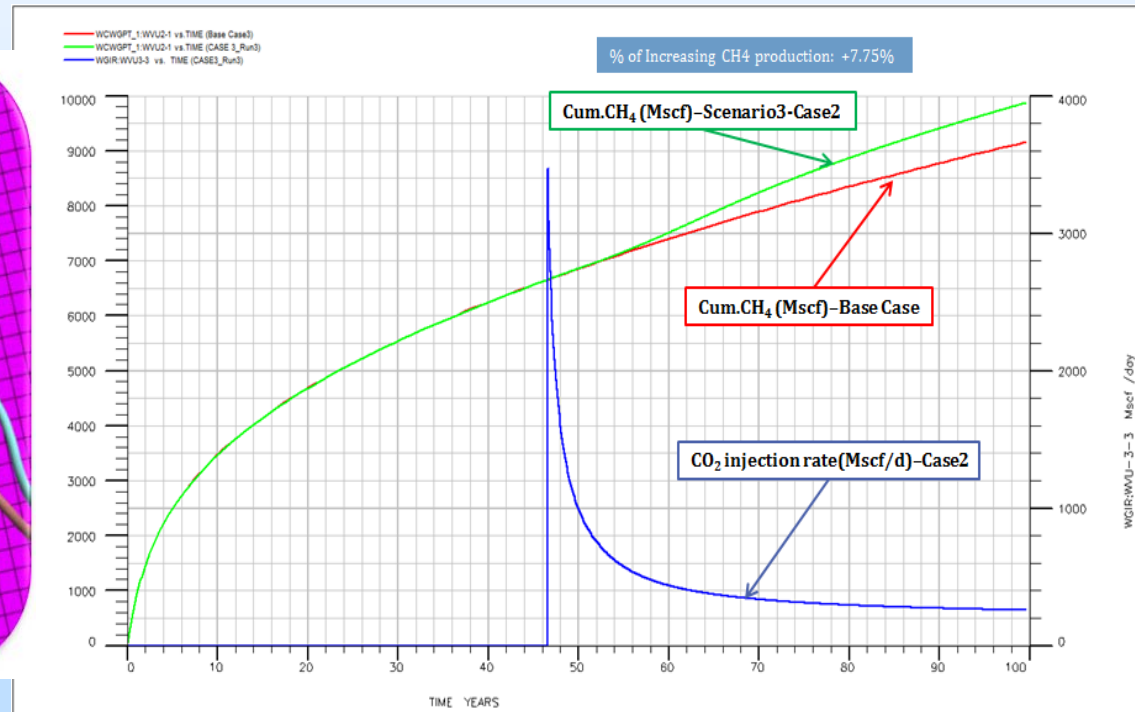
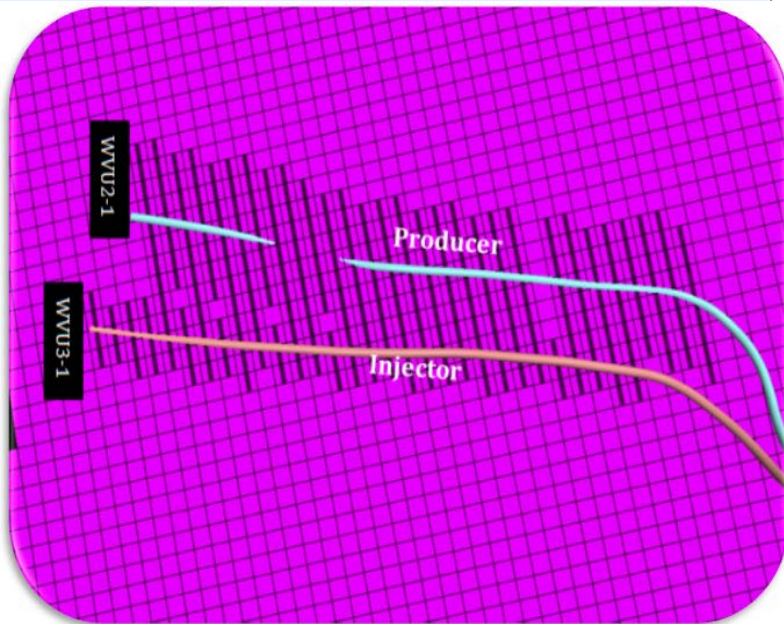
(b)



(c)

CO₂ Storage and Enhanced Gas Recovery Scenario

- CO₂ Injection for EGR not expected to start until primary production complete (nominally 40 years)
- Models predict EGR recovery (technical) potential between 0 and 11% (above primary production)
- Time to breakthrough of 10% mole fraction in produced stream decreases significantly as SRV overlap of adjacent laterals increases

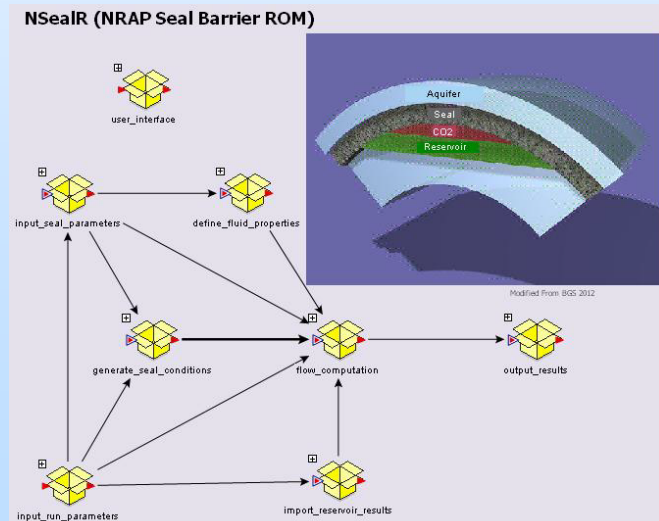
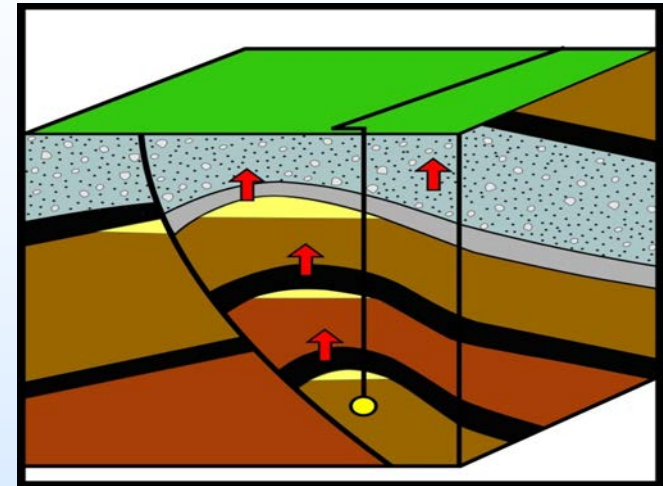


Flux through Fractured Seal ROM

NSEALR



- Assumes thin, relatively impermeable, fractured rock unit, initially saturated with a saline water.
- Two-phase, relative permeability approach and 1-D Darcy flow of carbon dioxide through the horizon in the vertical direction
- User defined or stochastically varying permeability, porosity, seal thickness
- Correction for in situ stress on aperture values generated by the fractured rock model, including shear stress options



Accomplishments to Date

- Well/pad-scale characterization of CO₂ storage and EGR performance in depleted shale gas formations
- Preliminary experimental characterization of:
 - Shale sorption characteristics
 - Mechanisms of CO₂/shale interactions
 - Matrix permeability
 - Fracture flow
 - Pore imaging
- Reduced physics model characterizing flux through fractured seal
- Contributing to methodology for CO₂ storage in shale

Summary

– Future Plans

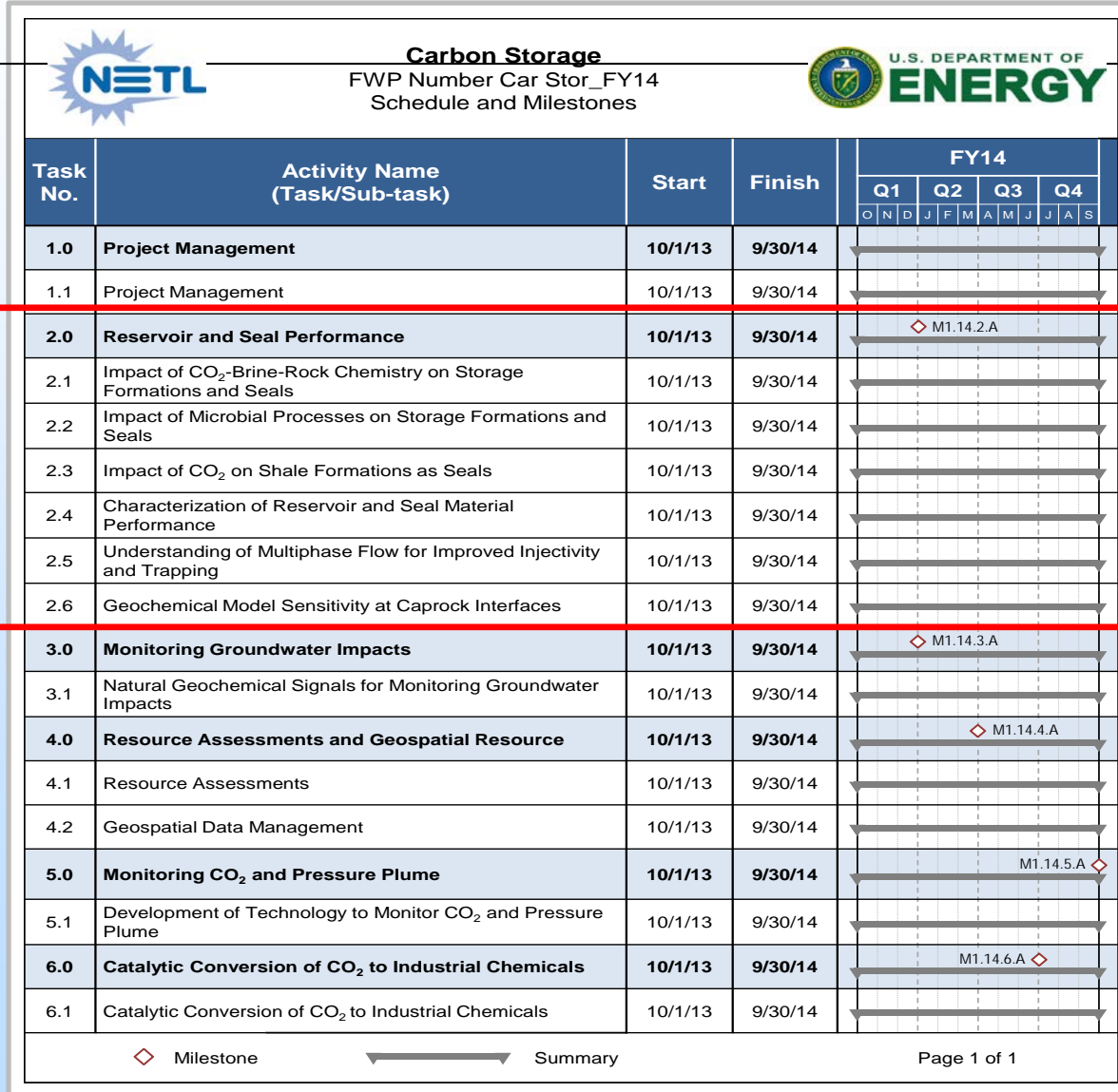
- Understanding shale pore type and structure
- Flow through nanopores on molecular scale
- Importance of pore effects at core-scale
- Matrix swelling/shrinkage effects
- Oil wet versus water wet (black shale vs. gray)
- Liquid and condensate reservoirs
- Simulation refinement and validation

Organization Chart

- NETL Office of Research & Development
 - Predictive Geosciences Division
 - Engineered Natural Systems Division
 - Material Characterization Division
- URS Corp.
- West Virginia University, Penn State University, Carnegie Mellon University

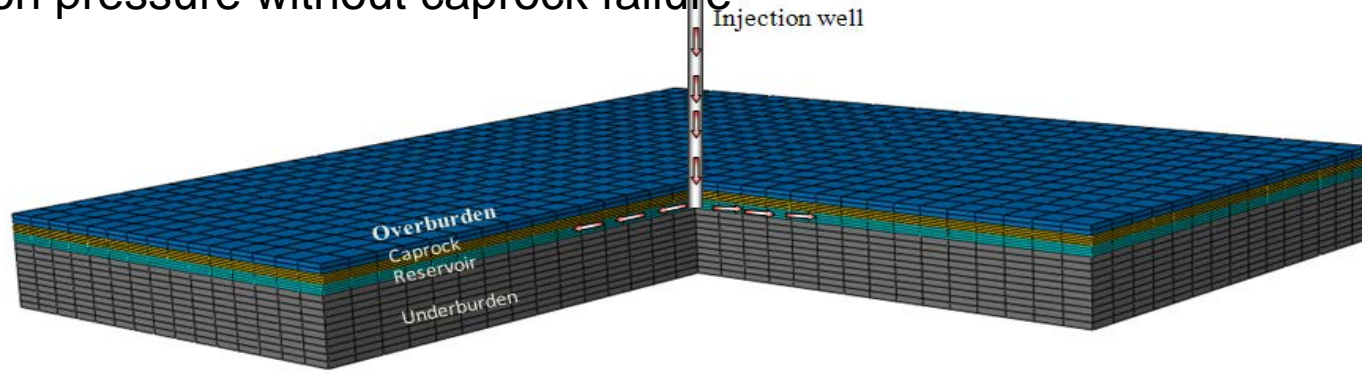


Gantt Chart



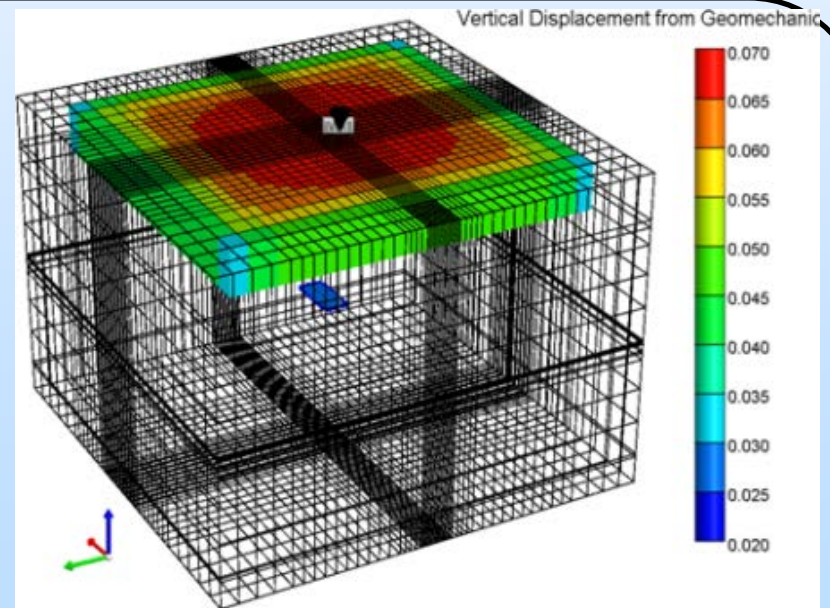
Coupled Fluid Flow and Geomechanical Modelling

3-D, single phase Finite Element Model to estimate maximum allowable injection pressure without caprock failure



Ground Deformation

- Maximum computed surface displacements are about 0.07 ft (21.3 mm).
- Can monitor with tiltmeter array



Vertical displacements above injection zone

Related Studies

- Nuttall et al., (2005) – Kentucky Geologic Survey
 - KGS developed the first volumetric estimates of CO₂ storage potential in the Carbonaceous (black) Devonian gas shales that underlie Kentucky, estimating that as much as 28 Gt could be stored there.
- Advanced Resources International (2013)
 - Basin-level assessment of CO₂ and EGR potential, reservoir simulation, novel monitoring, techno-economic assessment
- Tao & Clarens (2013) (U. Virginia)
 - Estimating CO₂ storage in Marcellus shale
- Zobak et al. (Stanford)
 - evaluate physical and chemical interactions between CO₂ and shale, imaging of fluid migration in shale
- Ripepi et al. (Virginia Tech)
 - Simulation and field demonstration in Central Appalachia

(2) Experimental Analysis of CO₂ Storage in Organic-rich Shale

Purpose:

Examine & quantify CO₂ sorption capacity of *individual* clay standards & shale samples

Determine relative roles of kerogen, clay, & clay type in CO₂ storage potential of shales

Analytical work conducted on shale samples and clay standards

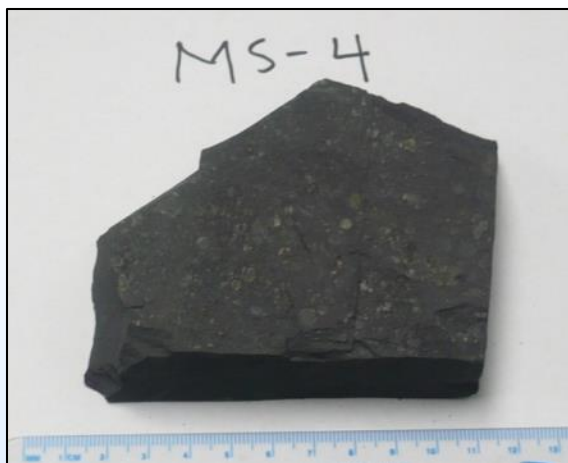
Sample	Description	He Pycnometry	E-SEM	FT-IR (std)	FT-IR (T&P)	CO ₂ Adsorption Isotherms	TOC	XRD
Shale Samples								
MS-1	Marcellus: <i>Oatka Creek</i>	Y	Y	Y	Y	Y	Y	Y
MS-4	Marcellus: <i>Union Springs</i>	Y	Y	Y	Y	Y	Y	Y
US-1	Utica: <i>Flat Creek</i>	Y	Y	Y	Y	Y	Y	Y
Clay Standards*								
STx-1	Ca-Smectite	Y	-	Y	Y	Y	-	-
IMt-2	Illite	Y	-	Y	Y	Y	-	-
KGa-1b	Kaolinite	Y	-	Y	Y	Y	-	-
ISCz-1	Illite-Smectite	Y	-	Y	Y	Y	-	-
Talc	control	-	-	Y	Y	-	-	-

*All clays are natural standards obtained from the Clay Mineral Society

“Y” indicates the procedure has been conducted on the sample

Organic-rich Shale Outcrop Samples

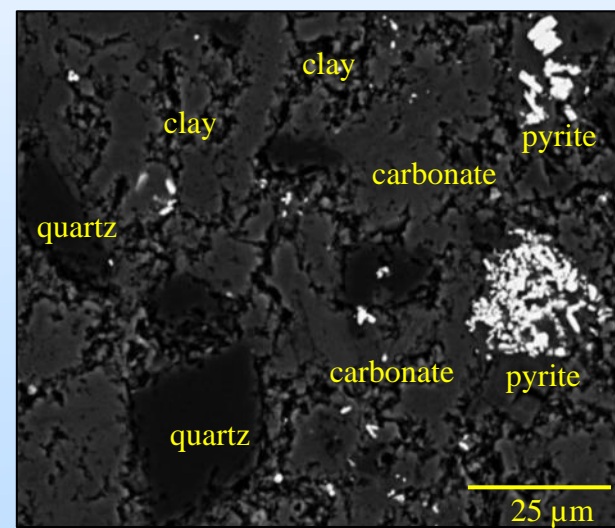
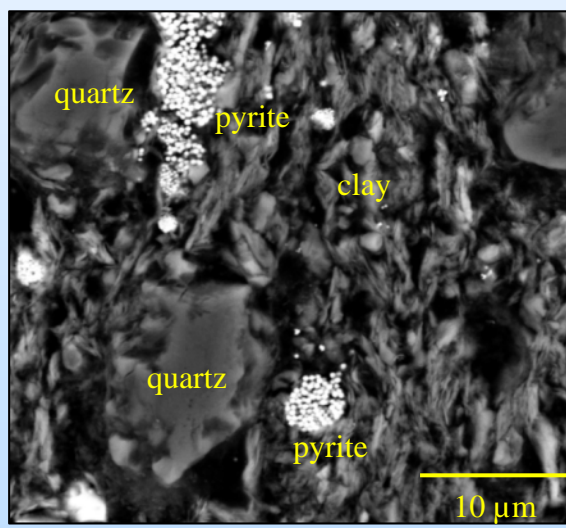
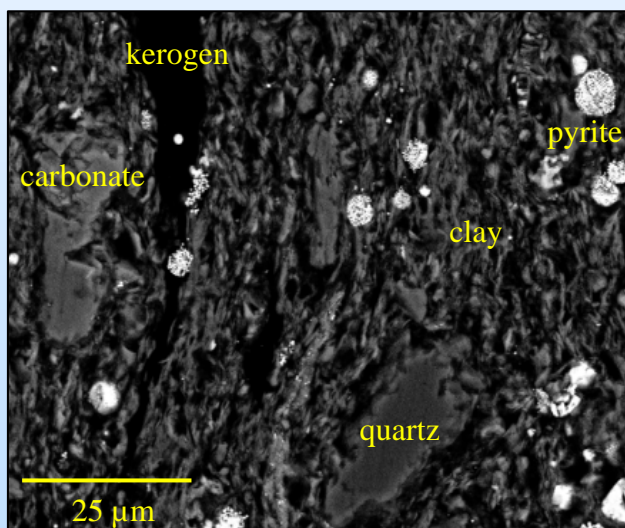
Marcellus - *Union Springs*



Marcellus - *Oatka Creek*



Utica - *Flat Creek*



TOC = 9.20 wt. % (σ 0.60)

TOC = 6.51 wt. % (σ 0.22)

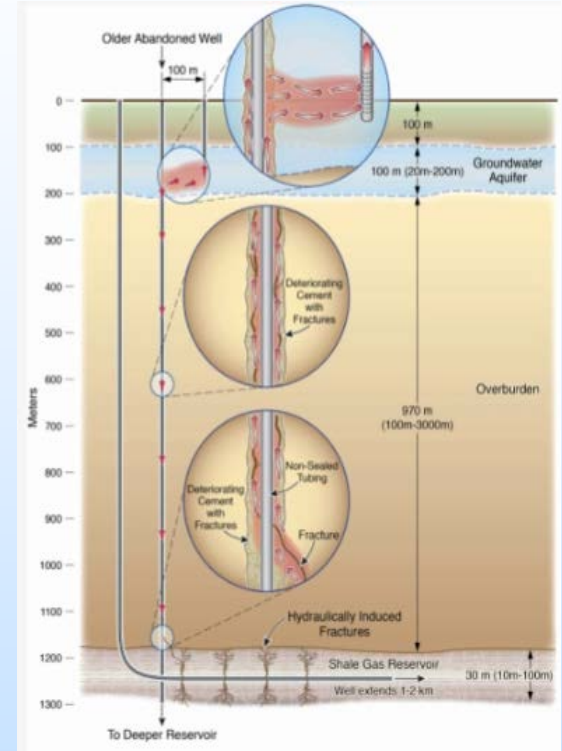
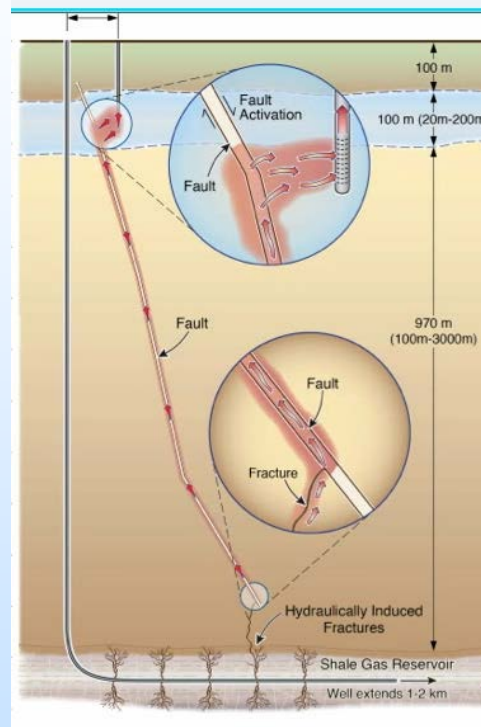
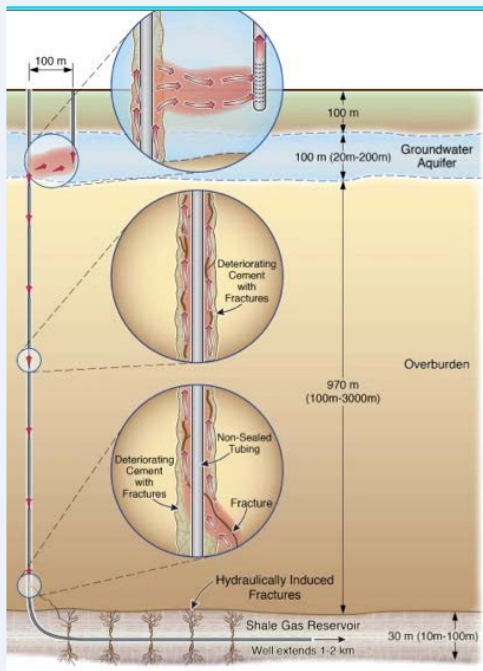
TOC = 0.45 wt. % (σ 0.17)

Quartz + Clay (e.g. illite, chlorite, kaolinite) + Carbonate + Pyrite + Kerogen \pm Feldspar 

Key Findings: CO₂ Storage in Shale

- Without HF and natural gas production, CO₂ can not be injected
- Storage predominantly as free-phase CO₂ in fractures – low permeability matrix limits amount of matrix available for sorption
- Favorable assumptions about Langmuir characteristics results in only a small increase in storage (sorbed phase)
- Storage ~ 50,000 tonnes per fractured stage
- CO₂ storage is not much greater in injector/producer scenario, and can be less in cases with significantly overlapping SRV

Potential Fluid Leakage Pathways from Unconventional HC Formations (US EPA, 2012)



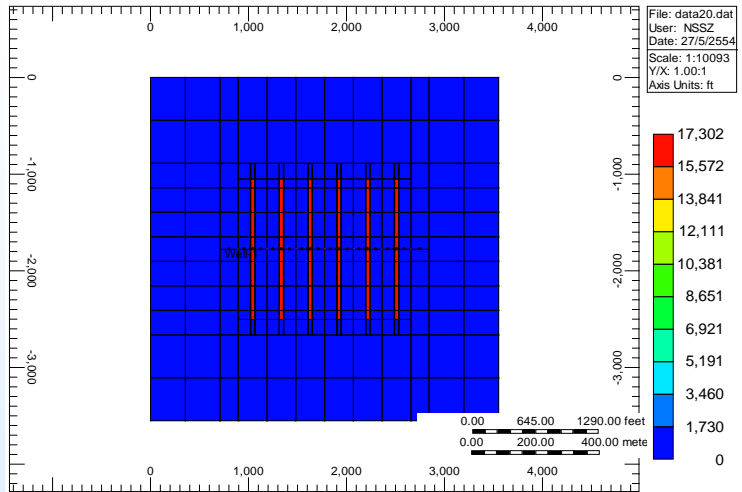
Leakage through the annuli of the vertical drilling well

Leakage through a natural fault

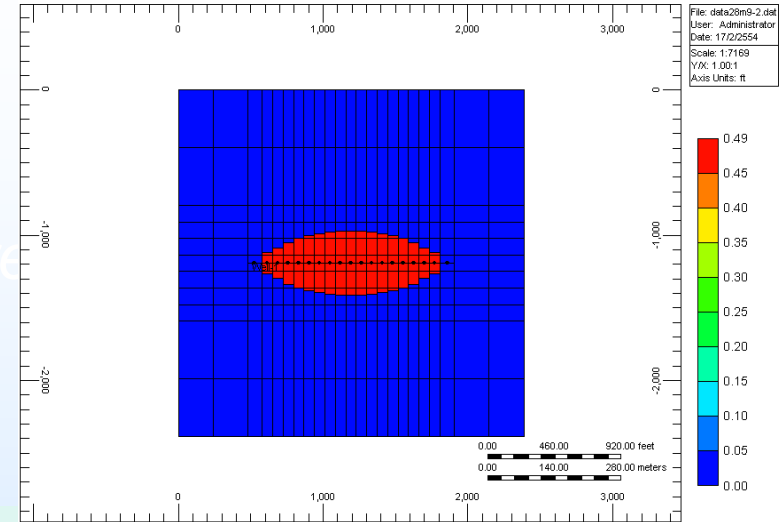
Leakage through an abandoned well

Representation of Horizontal Wells with Transvers Hydraulic Fractures Evaluating the potential viability of an Equivalency Network

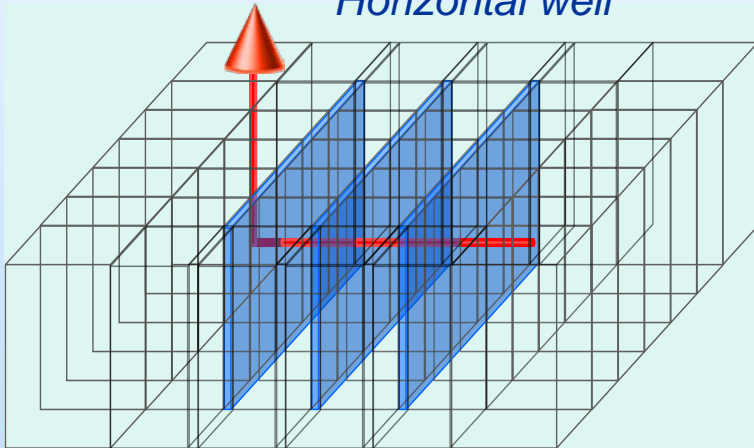
Permeability I - Fracture (md) 2010-01-01 K layer: 1



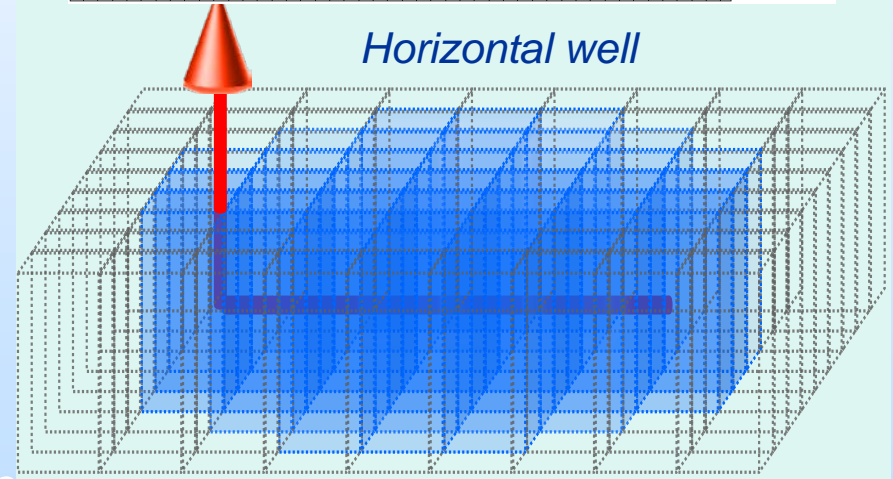
Permeability I - Fracture (md) 2010-01-01 K layer: 1



Horizontal well



Horizontal well

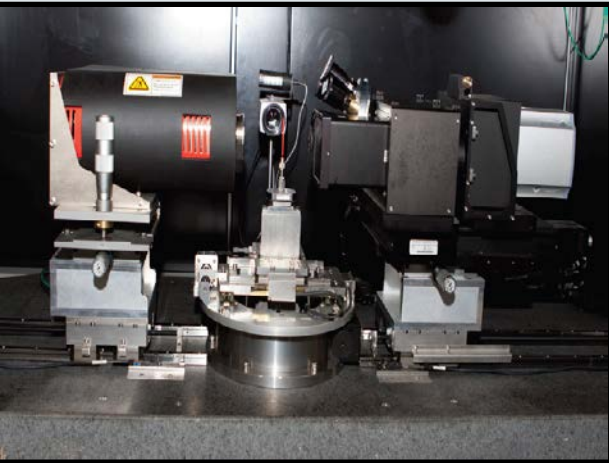


Discrete transverse fracture representation

Maximum cumulative production is with 5%

Crushed zone representation

NETL ORD Multi-Scale CT Flow and Imaging Facilities



Micro CT Scanner

- Resolution 10^{-6} to 10^{-5} m
- Pore scale



Industrial CT Scanner

- 10^{-6} to 10^{-3} m
- Pore & core scale
- Pressure & flow controls



Medical CT Scanner

- 10^{-4} to 10^{-2} m
- Core scale
- P, T, and flow controls

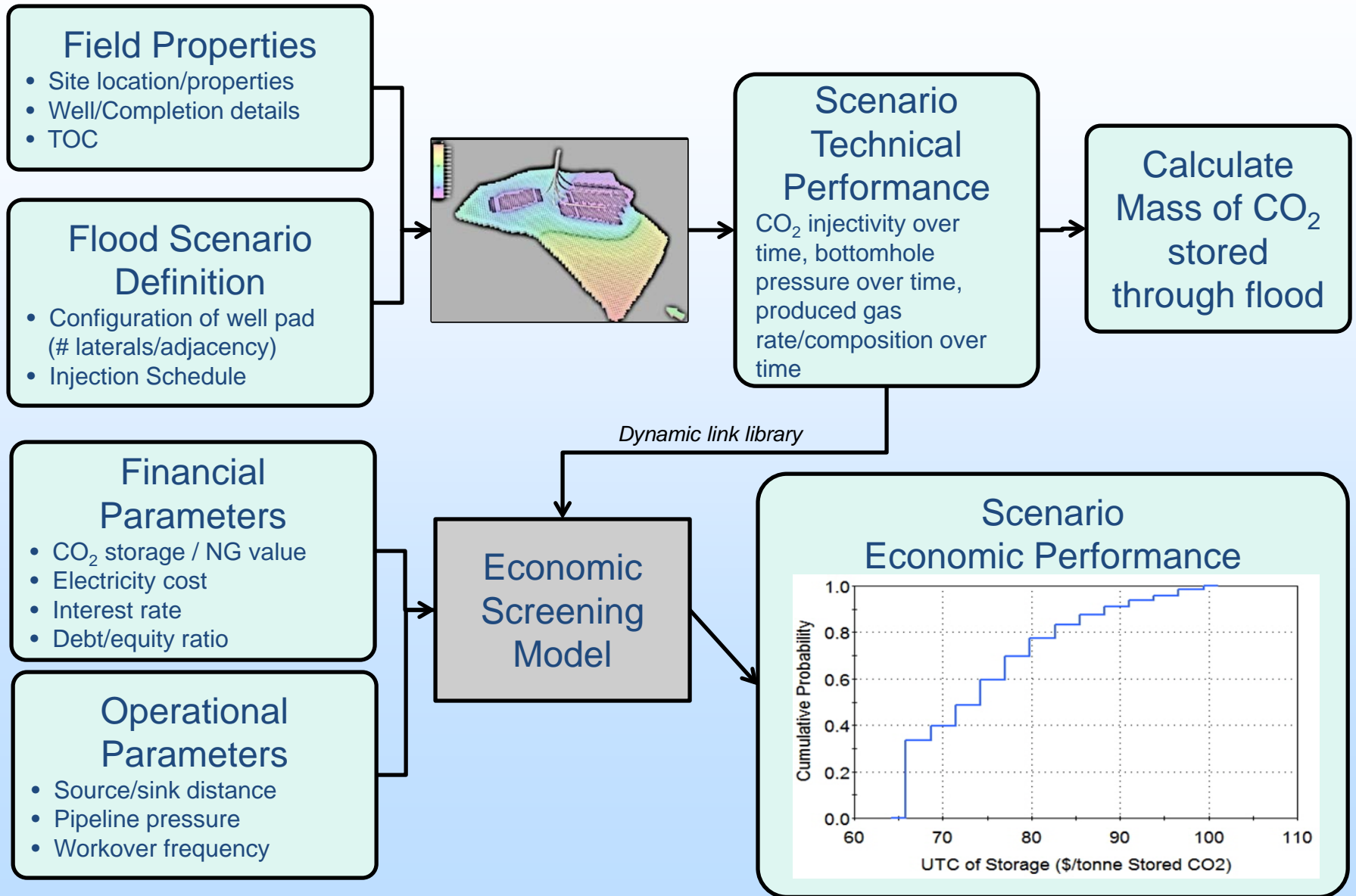
Precision Petrophysical Analysis Laboratory

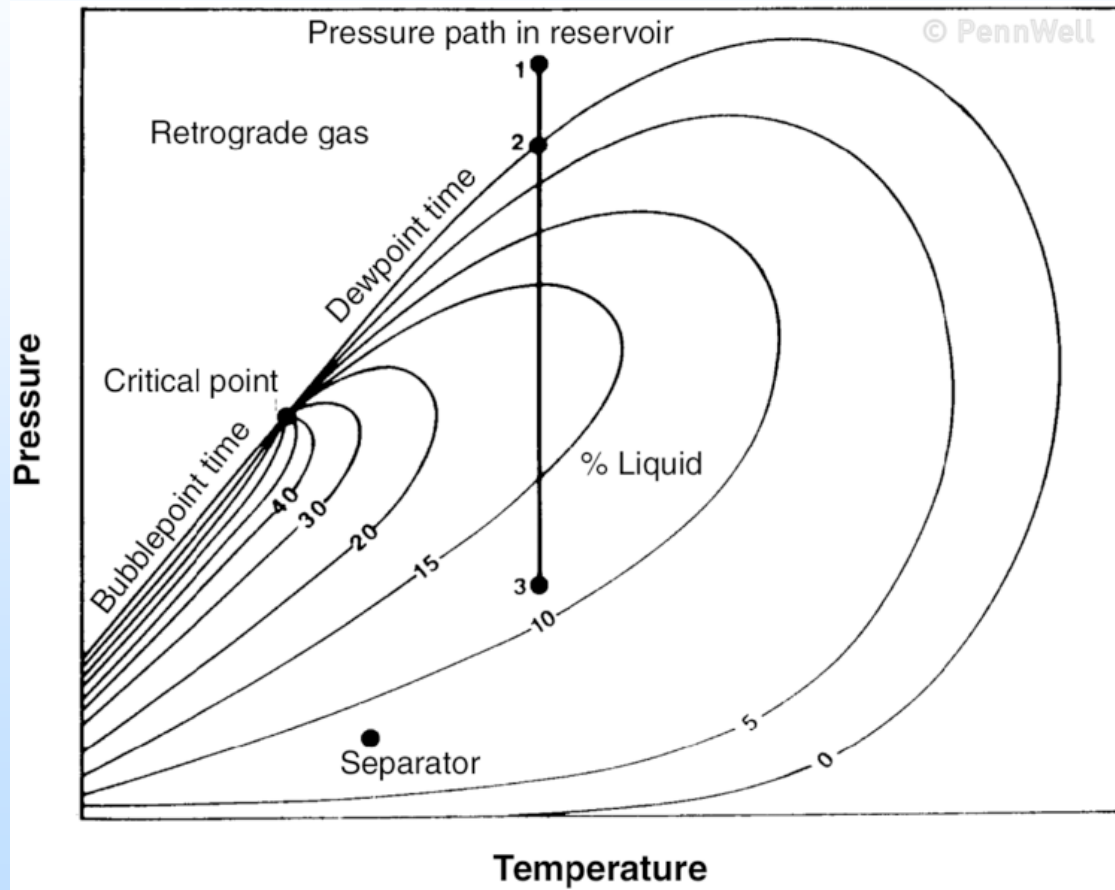
Effective porosity and permeability of shale to CO₂/CH₄ over range of effective stress, and characterization of hysteresis effects



- *Steady-state flow measurement, research quality data*
- *Capable of running different gases under different pressures, including nitrogen, methane and carbon dioxide.*
- *Capable of reproducing in-situ net stress, and measuring gas flow under partial liquid saturation.*
- *Can also measure pore volume to gas, adsorption isotherms and PV compressibility using N₂, CH₄ or CO₂*
- *Uses stable gas pressure as a reference for flow measurement*
 - *Temperature controlled*
 - *Stable to one part in 500,000*
 - *Target flow measurement is 10⁻⁶ standard cm³ per second*

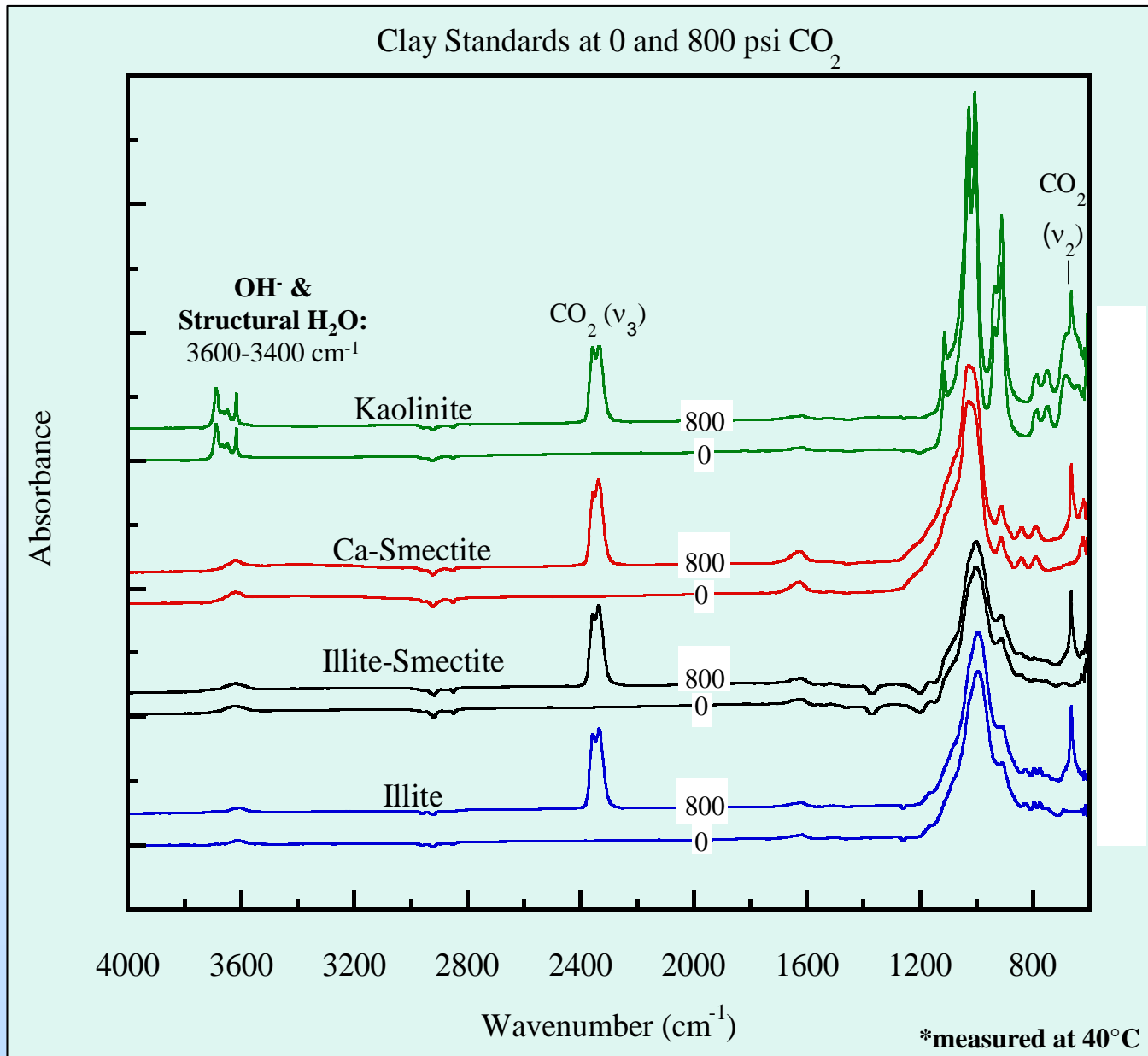
Linked SRM-Economic Screening Tool Modeling Approach





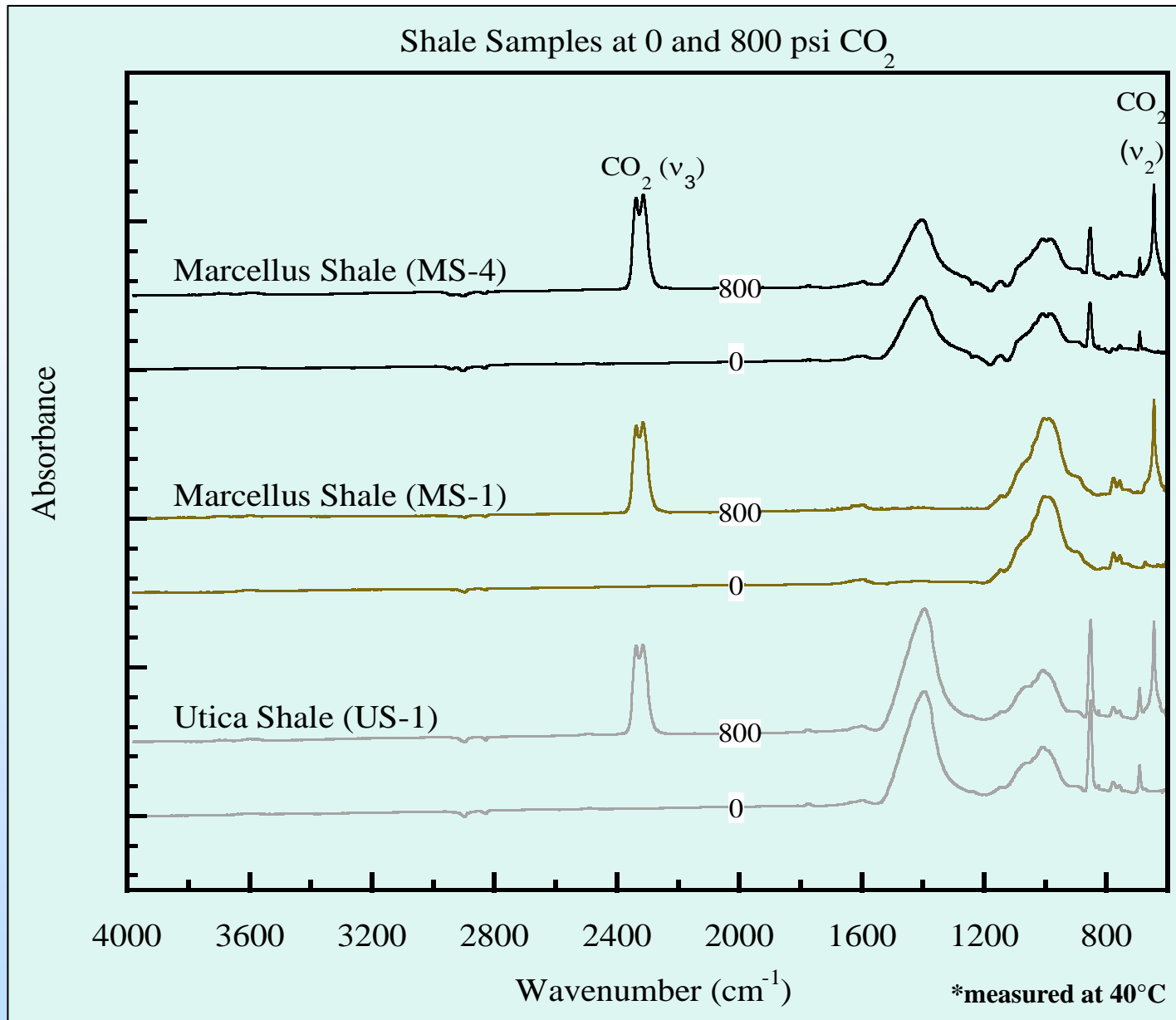
Source: *The Properties of Petroleum Fluids*, second edition, by William D. McCain Jr. Copyright Pennwell Books, 1990

CO₂-Clay Interactions: FT-IR Spectroscopy*:



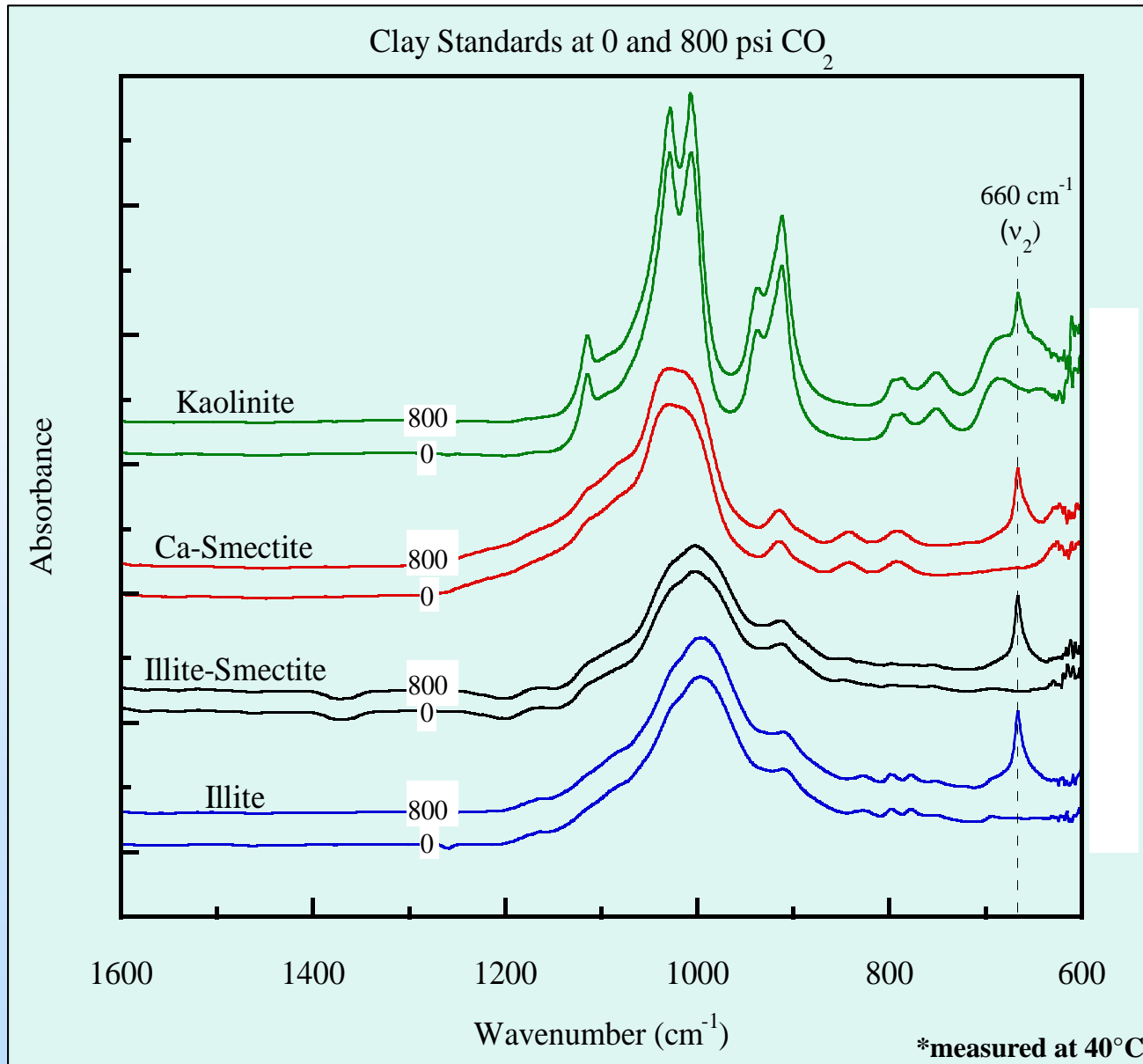
Chemically Sorbed CO₂ IR Peaks: 1400, 830, 720 cm⁻¹

CO₂-Shale Interactions: FT-IR Spectroscopy*:



No changes observed in IR spectra with addition of CO₂ and pressure

CO₂-Clay Interactions: FT-IR Spectroscopy*:



Chemically Sorbed CO₂ IR Peaks: 1400, 830, 720 cm⁻¹

CO₂ Sorption on Shale Samples

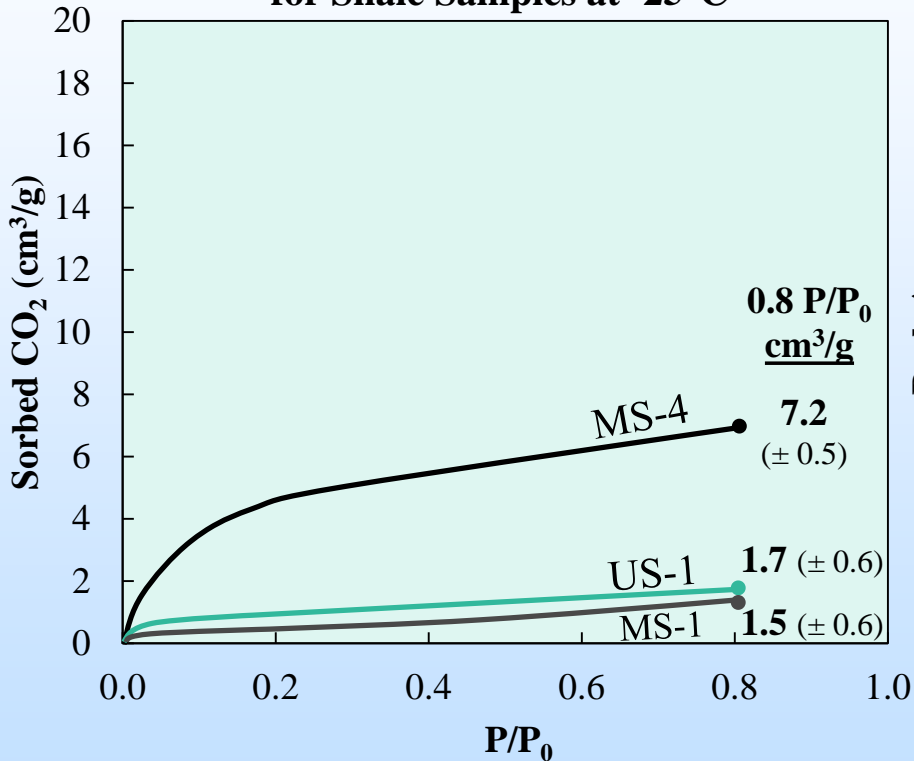
CO₂ Sorption Isotherms:

All Isotherm Data: 0-220 psi at -25, -15 & 0°C

FT-IR Data:

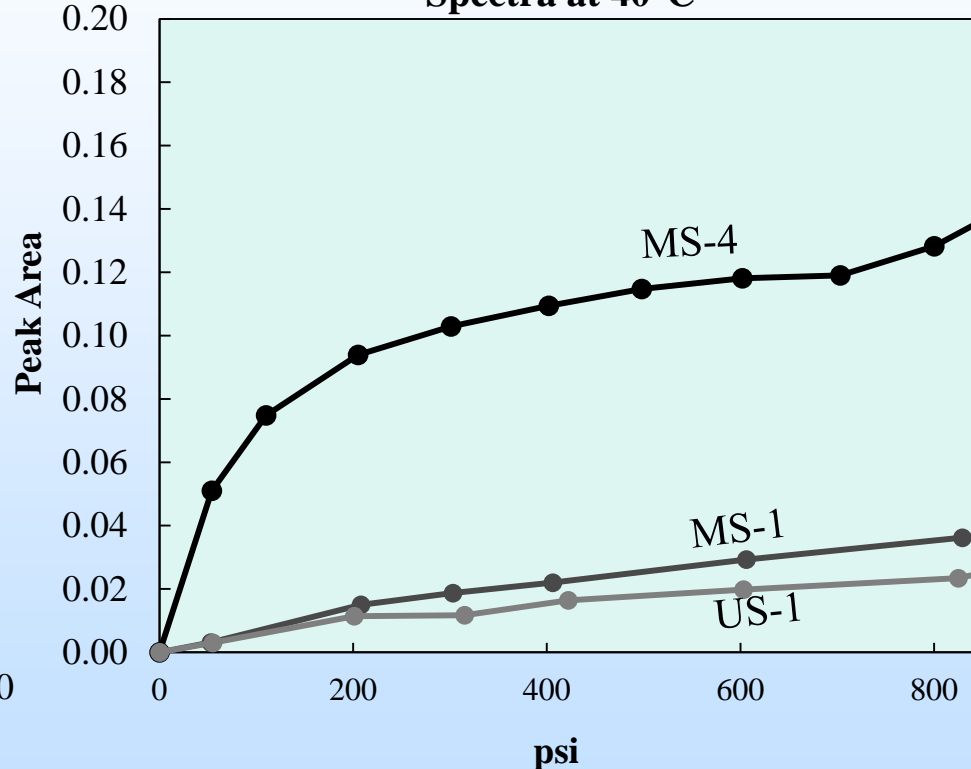
Area of 2331 cm⁻¹ CO₂ Sorption Peaks*

CO₂ Sorption Isotherms vs Relative Pressure for Shale Samples at -25°C



MS-4 > US-1 ≥ MS-1

Peak Area vs Pressure of Shale Infrared Spectra at 40°C



*2343 cm⁻¹ peak not strong enough to obtain reliable area measurements

TOC-content (wt. %): MS-4 (9.2) > MS-1 (6.5) > US-1 (0.5)

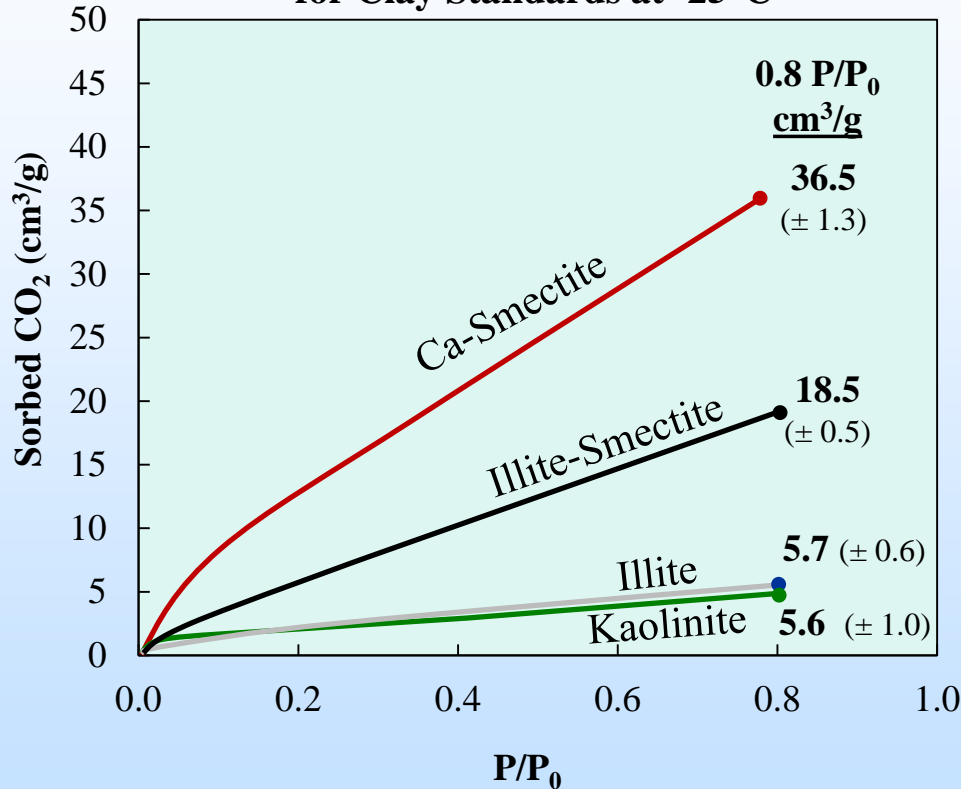


CO₂ Sorption on Clay Standards

CO₂ Sorption Isotherms:

All Isotherm Data: 0-220 psi at -25, -15 & 0°C

CO₂ Sorption Isotherms vs Relative Pressure for Clay Standards at -25°C

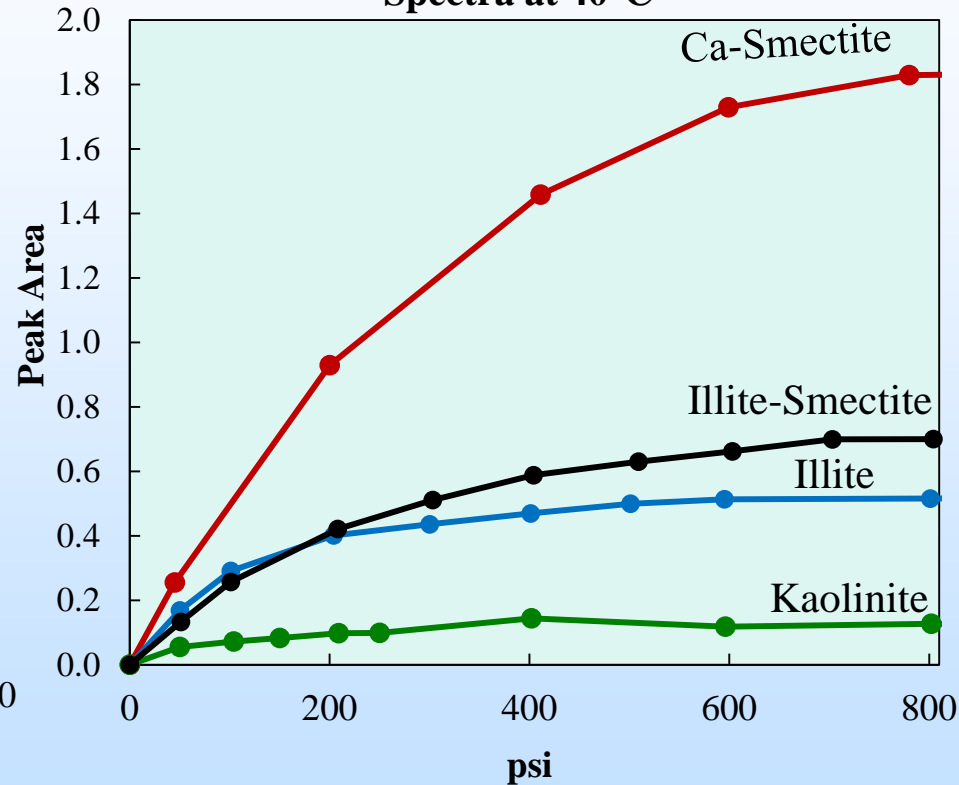


Smectite > Illite-Smectite > Illite ≥ Kaolinite

FT-IR Data:

Area of 2343 cm⁻¹ CO₂ Sorption Peaks

Peak Area vs Pressure for Clay Infrared Spectra at 40°C



Integrated CO₂ peak area is not quantitative

FT-IR trends compliment results of CO₂ isotherm measurements



Experimental Analysis of CO₂ Storage in Organic-rich Shale

Results:

- (1). Smectite > Illite-Smectite > MS-4 ≥ Illite ≥ Kaolinite > US-1 ≥ MS-1

Summary of CO₂ Sorption Isotherm Data at 0.8 P/P₀ & -25°C

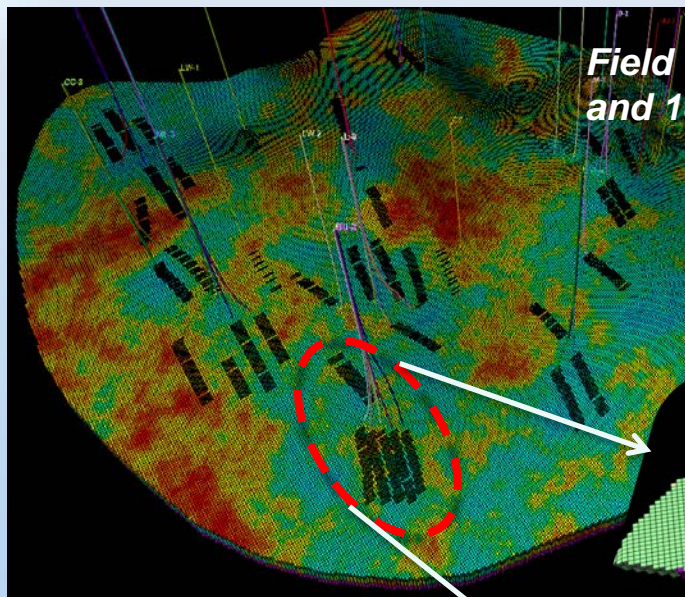
Sample	Smectite	Illite-Smectite	MS-4	Illite	Kaolinite	US-1	MS-1
cm ³ /g	36.5	18.5	7.2	5.7	5.6	1.7	1.5
error +/-	1.3	0.5	0.5	0.6	1.0	0.6	0.6

- (2). **Two CO₂ sorption peaks** observed at 2343 and 2331cm⁻¹ on IR spectra of the shale samples (possibly also clays)

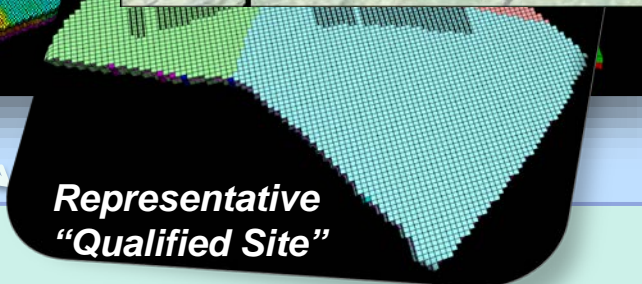
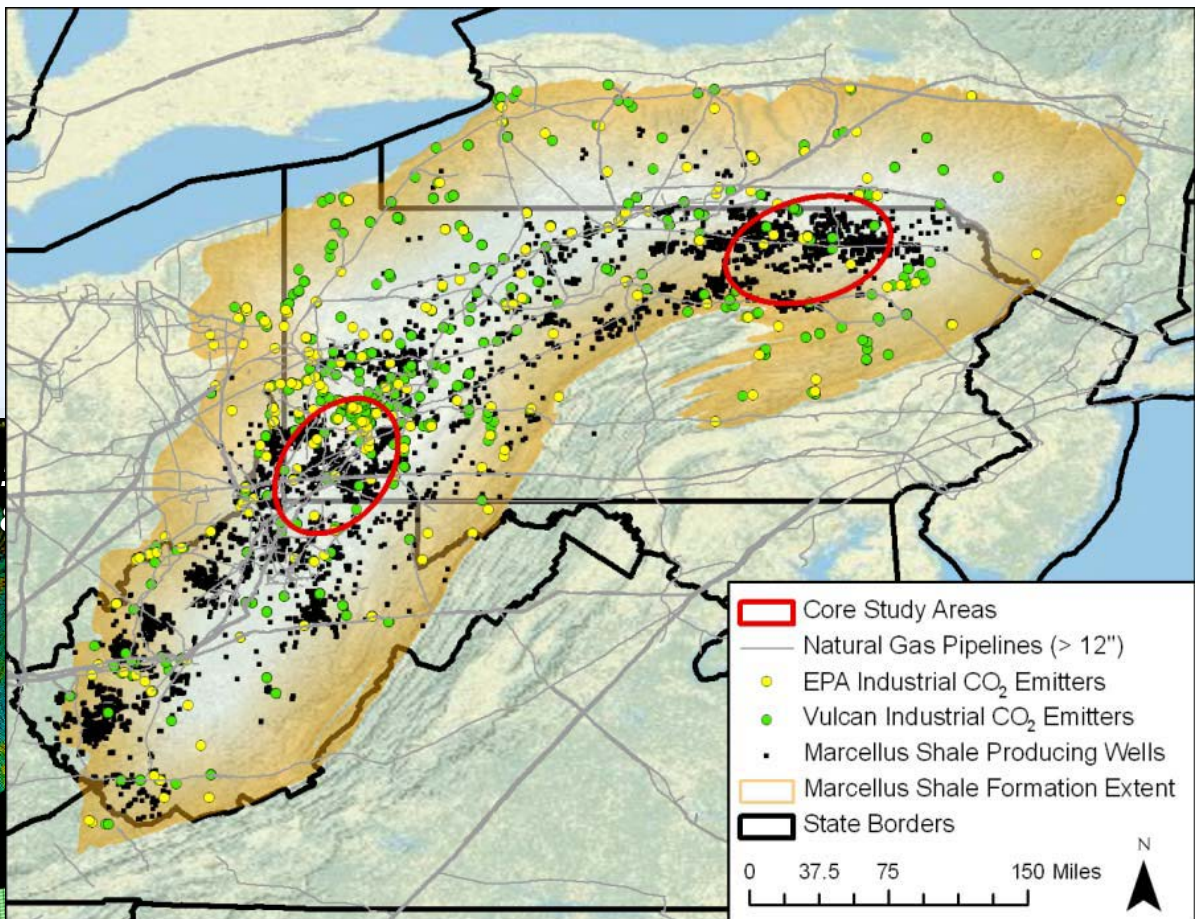
- (3). **No changes were observed in the IR spectra** of clays or shales after 15 min of exposure to CO₂ at pressures between 0-800 psi and 40°C.

Interpretations:

- (1). Shale formations with **high smectite, illite-smectite, and/or high TOC-content** may have high CO₂ storage potential (*e.g. Busch et al., 2008; Busch et al., 2009; Ross and Bustin, 2009*)
- (2). **There may be two CO₂ sorption sites in shales & clays:** in the interlayer* of clay structures & in the interpore space of minerals & kerogen. (**e.g. Rother et al., 2012; Geisting et al., 2012; Loring et al., 2012*)
- (3). At experimental conditions, **exposure to CO₂ does not induce chemical changes** in clays & shales of these compositions



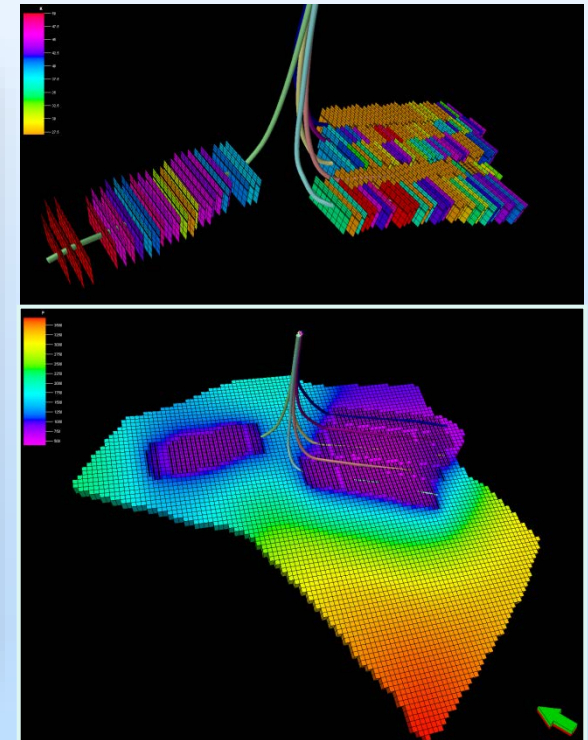
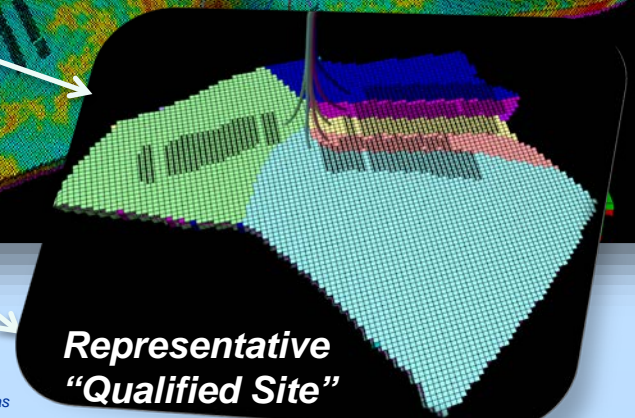
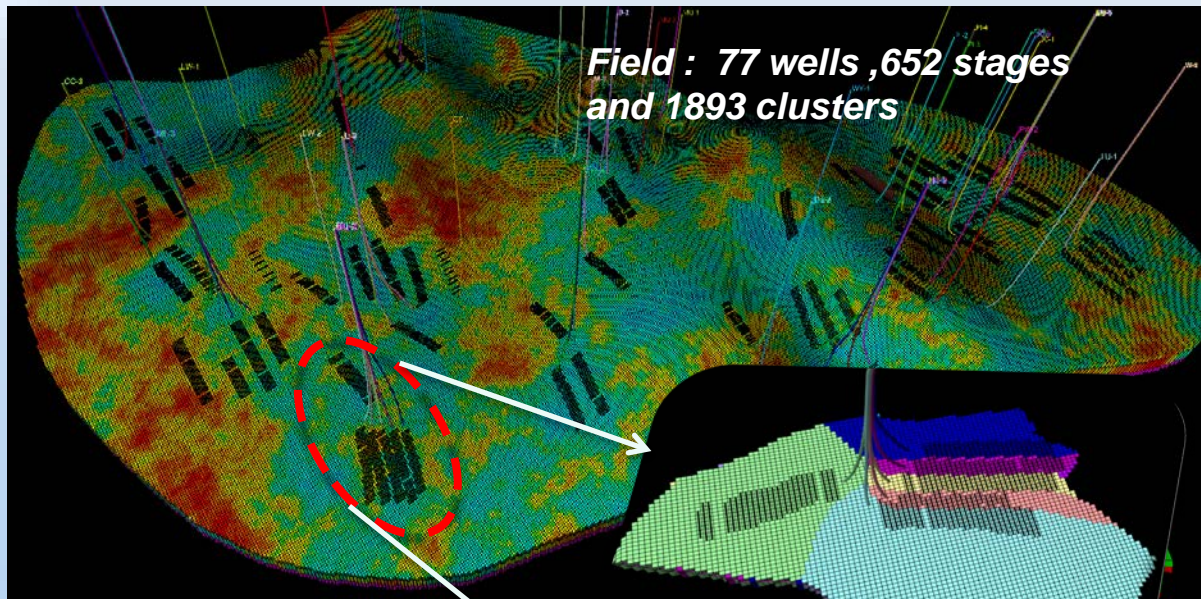
Field
and 10



Representative
"Qualified Site"

CO₂ Storage in Depleted Shale

- Acquire real-time gas production from a set of shale gas wells combining conventional AI-based reservoir modeling
- Use that set of data to develop population statistics
- Develop a history-matched model of shale gas production (29 month production history) using a conventional reservoir model
- Project forward to economic limit before initiating CO₂ injection
- Develop a surrogate reservoir model based on the history matched model to predict wellpad performance under CO₂ loading



CO₂ and CH₄ Sorption capacity as function of %TOC (single-fluid isotherms)

- Sorption capacity of CO₂ and CH₄ exhibit linear relationship with TOC

